

Supplementary Materials

Iron Oxide-Silica Core-Shell Nanoparticles Functionalized with Essential Oils for Antimicrobial Therapies

Figures S1 and S2 represent the size distribution and the correlation graphs, respectively, for the samples synthesized through the co-precipitation method, i.e., $\text{Fe}_3\text{O}_4@\text{SiO}_2_CP$, $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{thyme_CP}$, $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{rosemary_CP}$, and $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{basil_CP}$.

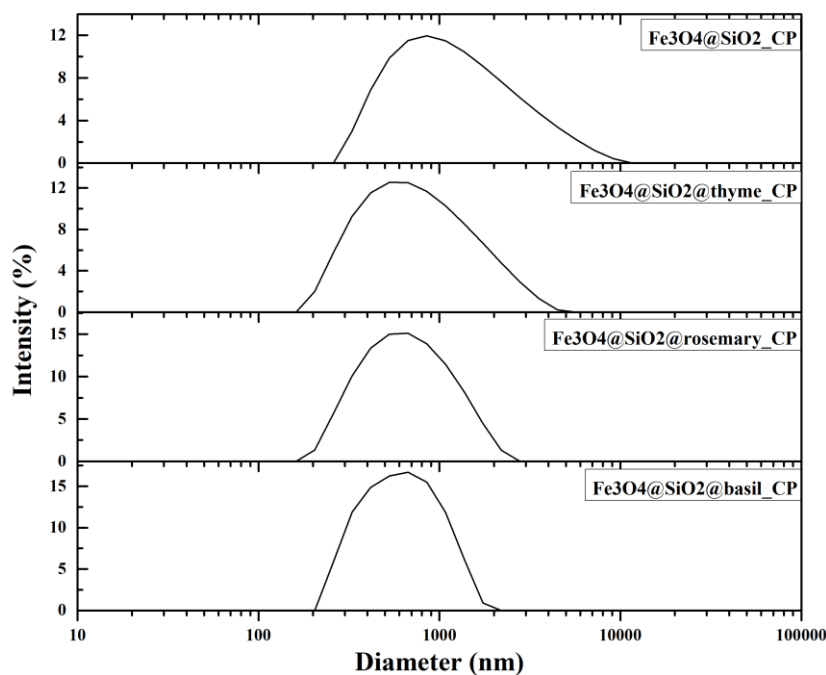


Figure S1: Size distribution for samples $\text{Fe}_3\text{O}_4@\text{SiO}_2_CP$, $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{thyme_CP}$, $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{rosemary_CP}$, and $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{basil_CP}$.

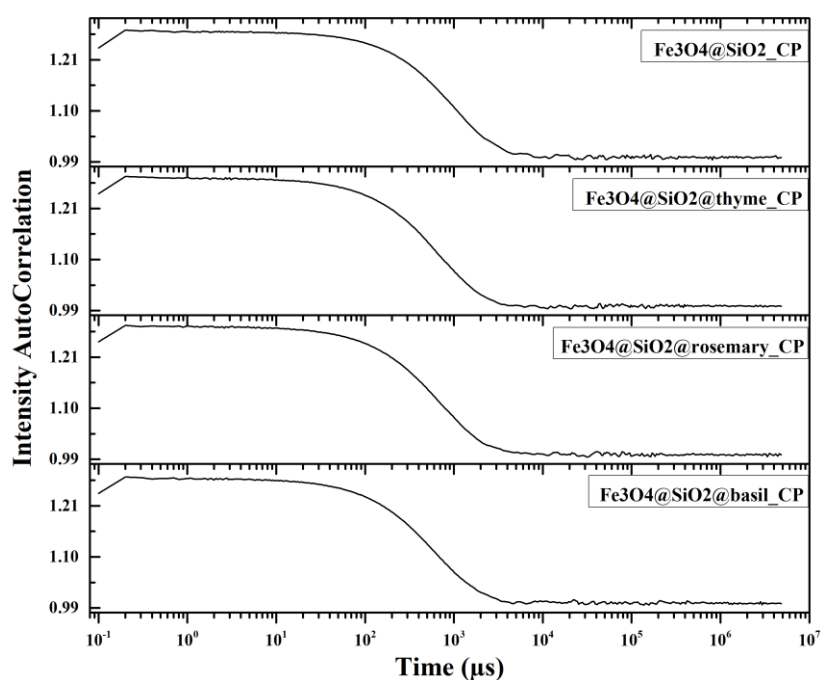


Figure S2: Correlation graphs for samples $\text{Fe}_3\text{O}_4@\text{SiO}_2\text{-CP}$, $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{thyme-CP}$, $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{rosemary-CP}$, and $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{basil-CP}$.

Figures 3 and 4 represent the size distribution and the correlation graphs, respectively, for the samples synthesized through the co-precipitation method, i.e., $\text{Fe}_3\text{O}_4@\text{SiO}_2\text{-SW}$, $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{thyme-SW}$, $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{rosemary-SW}$, and $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{basil-SW}$.

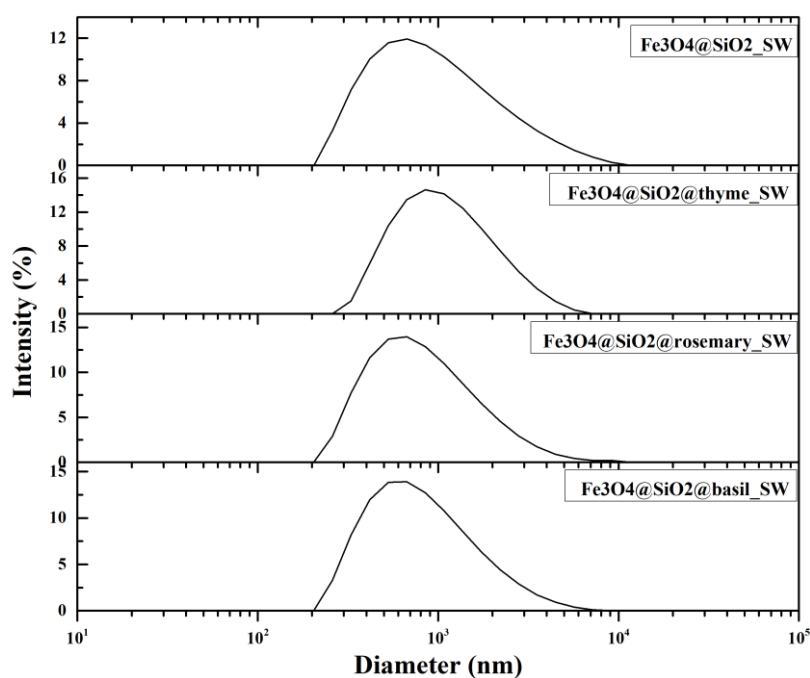


Figure S3: Size distribution for samples $\text{Fe}_3\text{O}_4@\text{SiO}_2\text{-CP}$, $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{thyme-CP}$, $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{rosemary-CP}$, and $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{basil-CP}$.

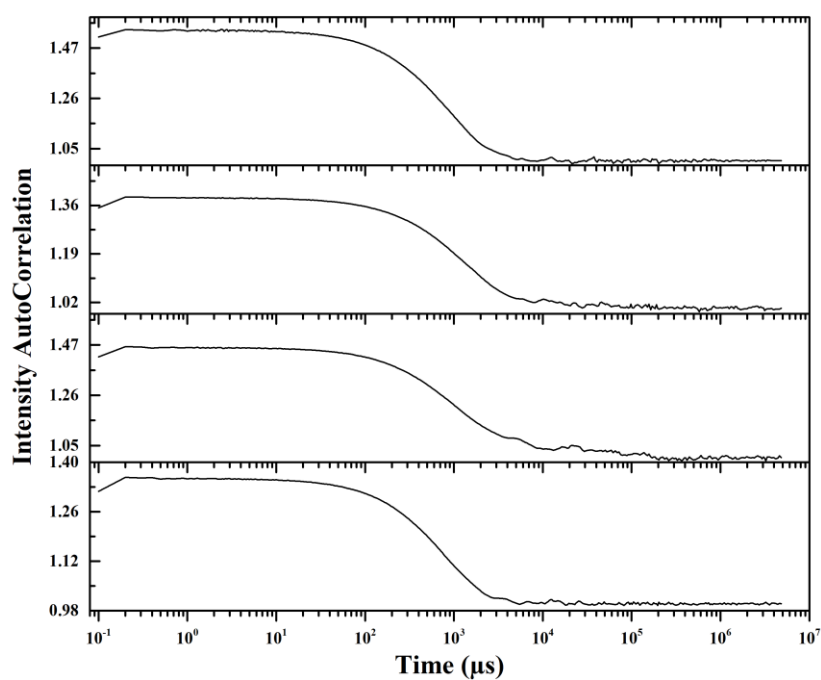


Figure S4: Correlation graphs for samples Fe₃O₄@SiO₂_CP, Fe₃O₄@SiO₂@thyme_CP, Fe₃O₄@SiO₂@rosemary_CP, and Fe₃O₄@SiO₂@basil_CP.

The compounds identified within EOs and their retention time were summarized in Table S1.

Table S1: The compounds identified within thyme, rosemary, and basil EOs, and their retention time.

Sample	Compound	Retention time
thyme EO	α -pinene	9.467
	camphene	10.050
	sabinene	11.051
	β -pinene	12.520
	α -phellandrene	12.114
	1,4-cineol	12.447
	p-mentha-1,5,8-triene	12.780
	d-limonene	12.978
	eucalyptol	13.083
	γ -terpinene	14.052
	terpinolene	15.062
	linalool	15.656
	fenchol	16.323
	1-terpineol	19.147
	endo-borneol	18.272
	4-terpineol	18.595
	α -copaene	25.305
	caryophyllene	26.681
	m-camphorene	41.436
rosemary EO	camphene	10.050
	fenchene	11.050
	β -myrcene	11.509
	o-cymene	12.801
	eucalyptol	13.061
	3-thujene	14.041
	linalool	15.604
	(+)-2-bornanone	17.323
	endo-borneol	18.261
	α -terpineol	19.095
	(-)-bornyl acetate	22.231
	α -copaene	25.305
	caryophyllene	26.722
	humulene	27.848
	caryophyllene oxide	31.641
basil EO	α -pinene	9.467

camphene	10.050
sabinene	11.051
β -pinene	12.520
α -phellandrene	12.114
1,4-cineol	12.447
p-mentha-1,5,8-triene	12.780
d-limonene	12.978
eucalyptol	13.083
γ -terpinene	14.052
terpinolene	15.062
linalool	15.656
fenchol	16.323
1-terpineol	19.147
endo-borneol	18.272
4-terpineol	18.595
α -copaene	25.305
caryophyllene	26.681
m-camphorene	41.436

The chromatogram profiles for each EO and EOs-functionalized core-shell nanoparticles and the mass spectra for each of the major compounds identified within the EOs can be found in the Supplementary Material.

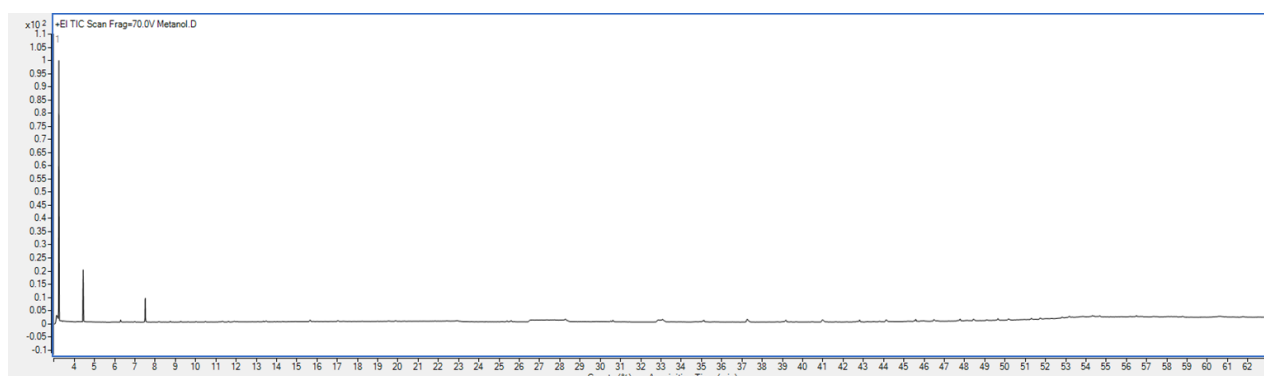


Figure S5: Chromatographic profile of the extraction solvent (methanol).

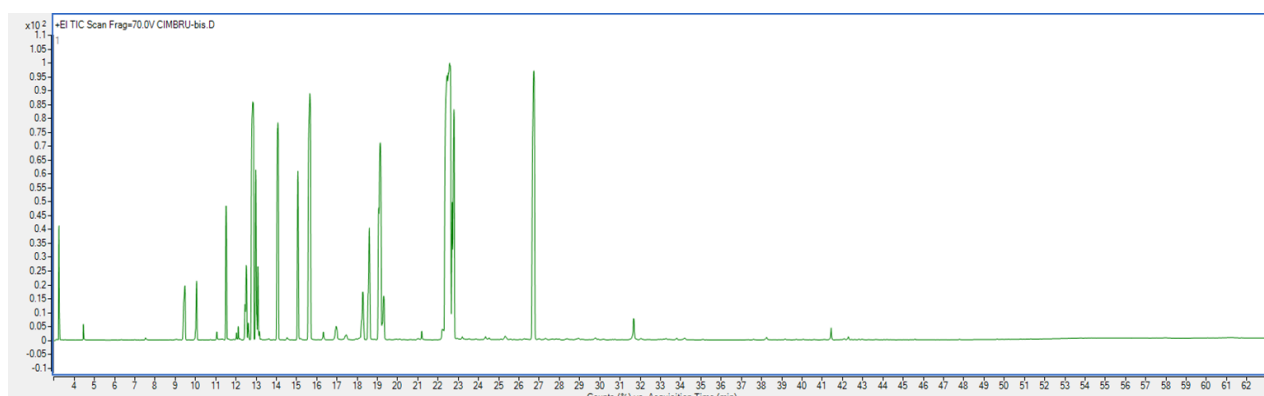


Figure S6: Chromatographic profile of the thyme EO.

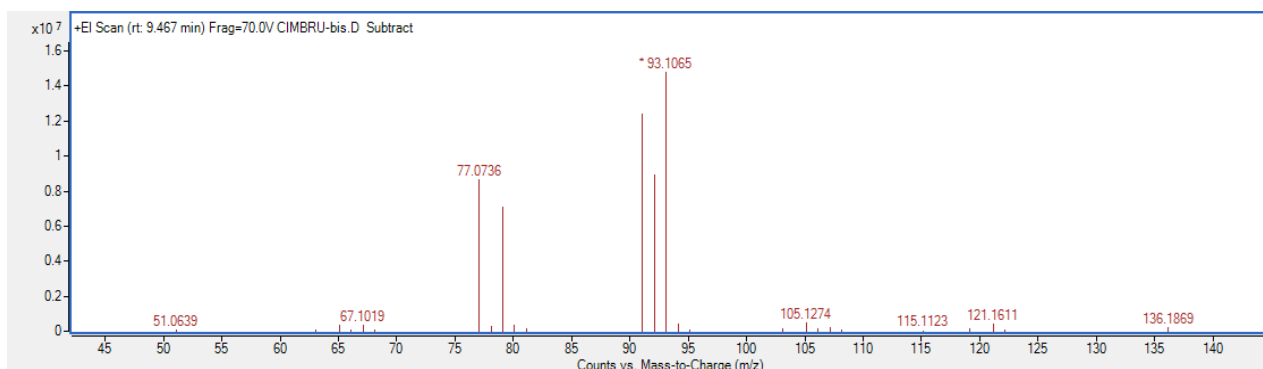


Figure S7.1: Mass spectrum for α -pinene.

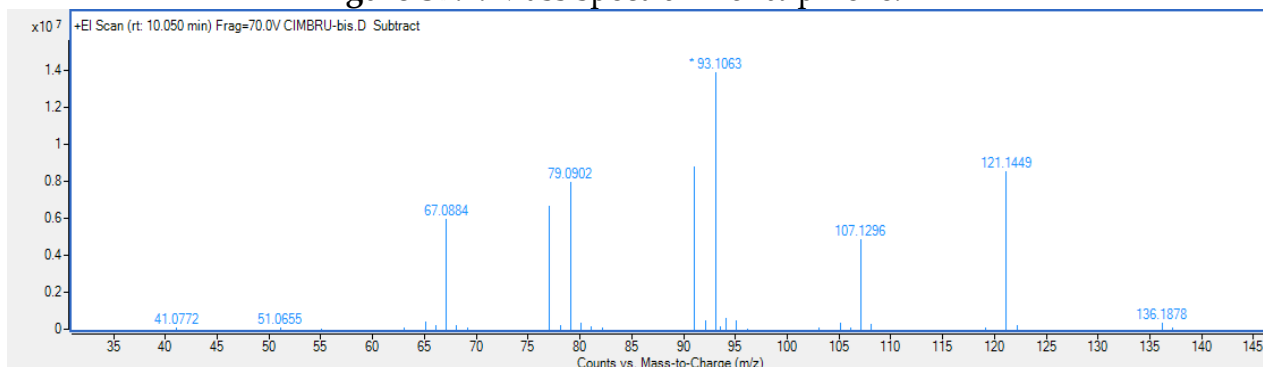


Figure S7.2: Mass spectrum for camphene.

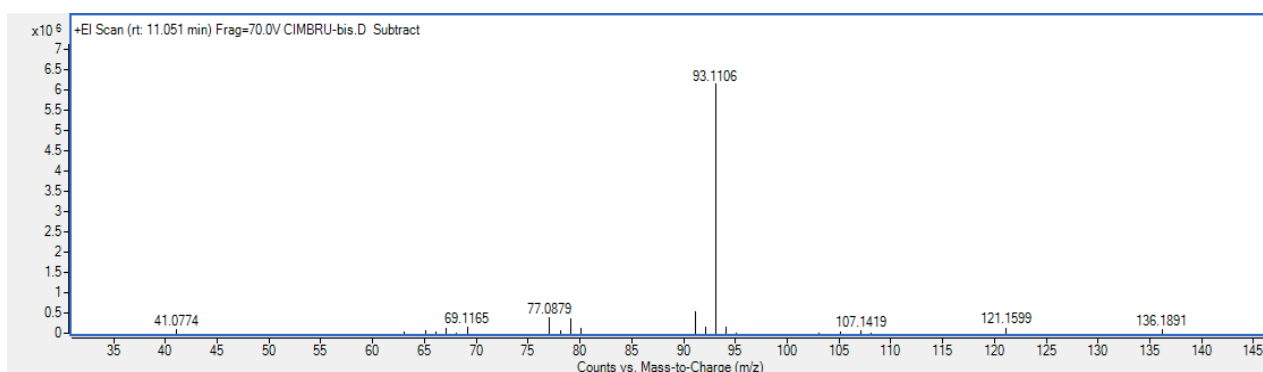


Figure S7.3: Mass spectrum for sabinene.

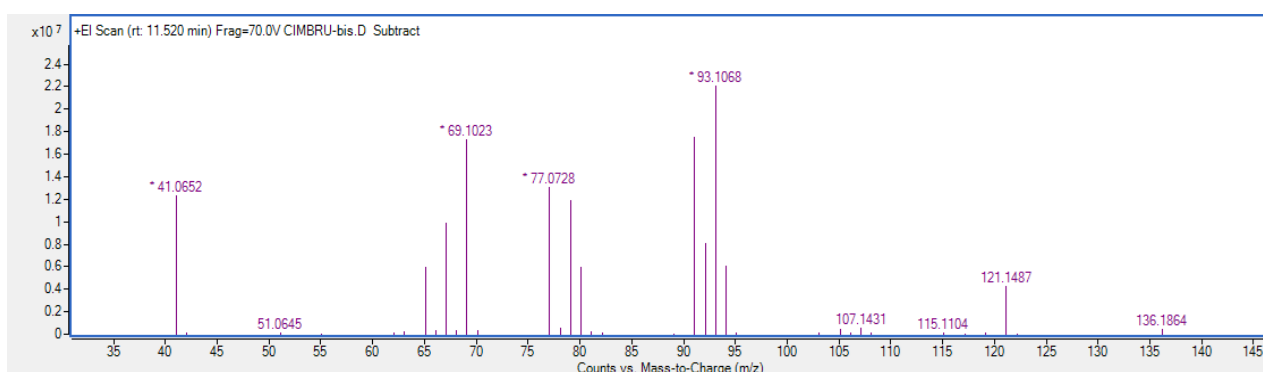


Figure S7.4: Mass spectrum for β -pinene.

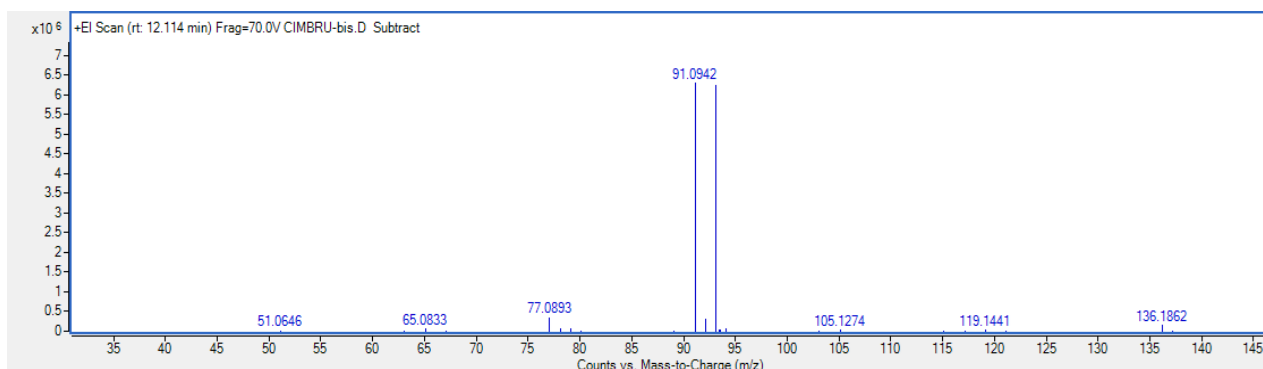


Figure S7.5: Mass spectrum for α -phellandrene.

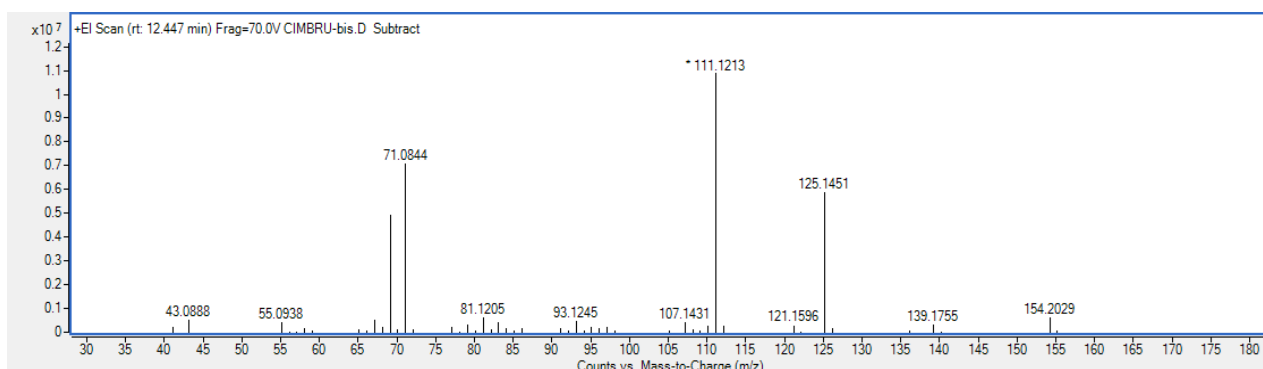


Figure S7.6: Mass spectrum for 1,4-cineol.

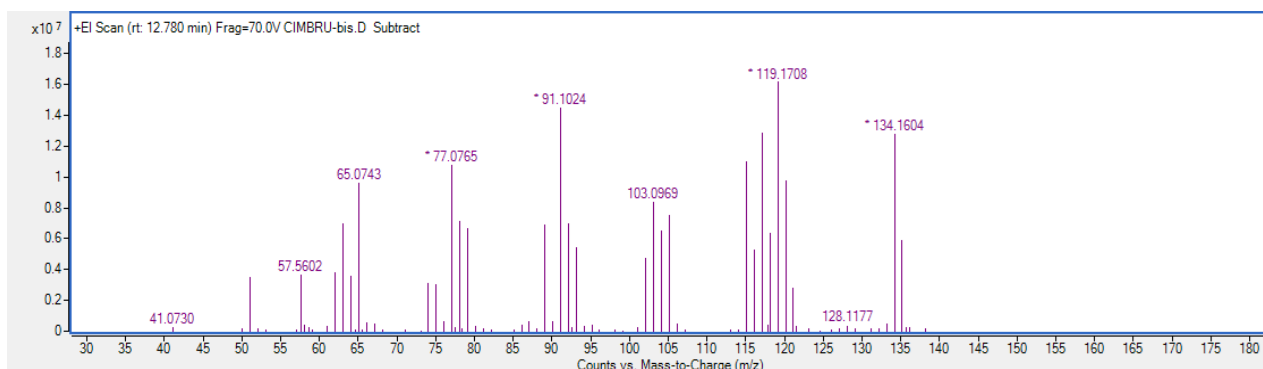


Figure S7.7: Mass spectrum for p-mentha-1,5,8-triene.

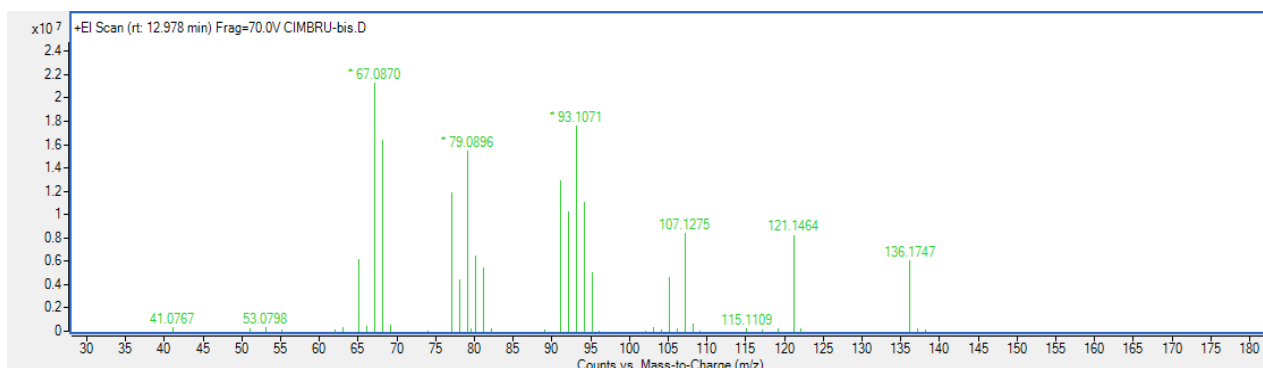


Figure S7.8: Mass spectrum for D-limonene.

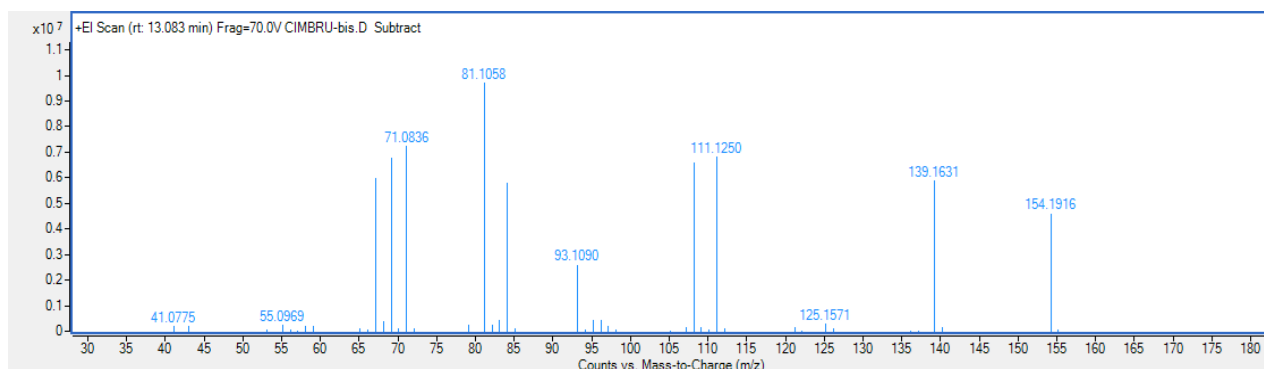


Figure S7.9: Mass spectrum for eucalyptol.

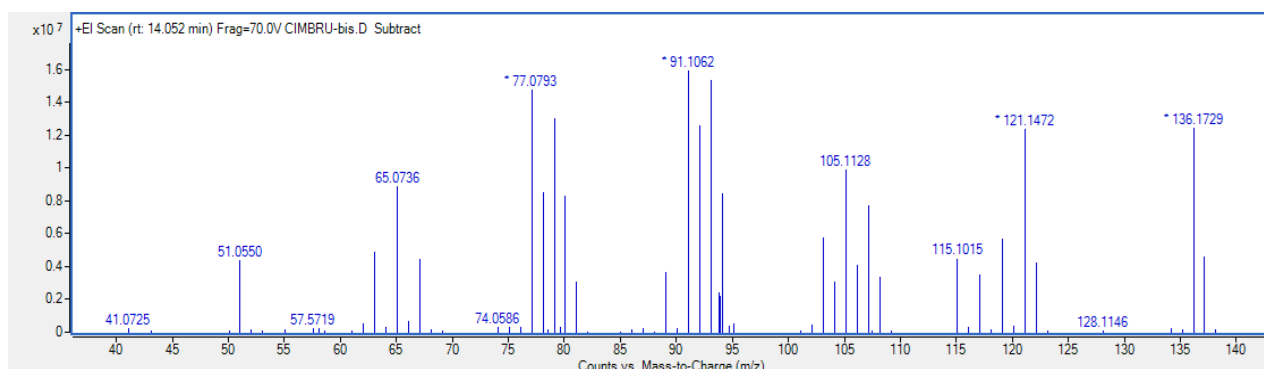


Figure S7.10: Mass spectrum for γ -terpinene.

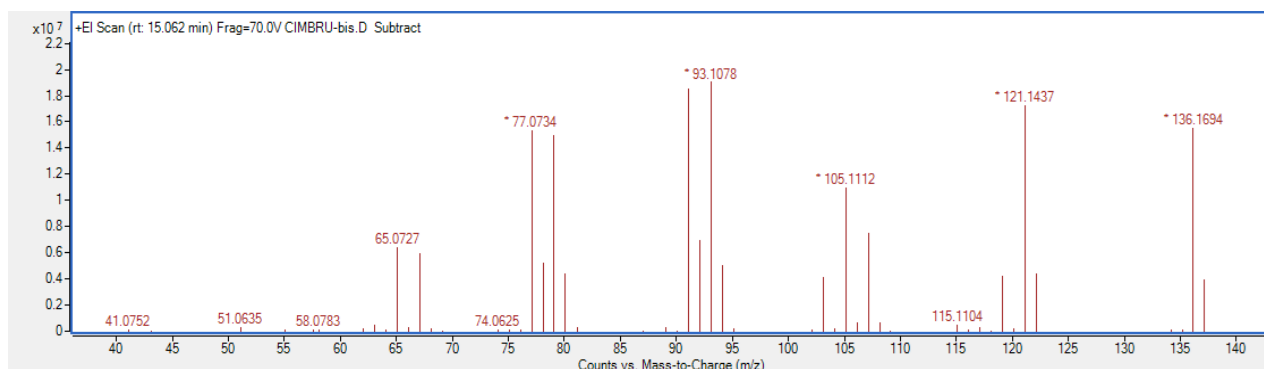


Figure S7.11: Mass spectrum for terpinolene.

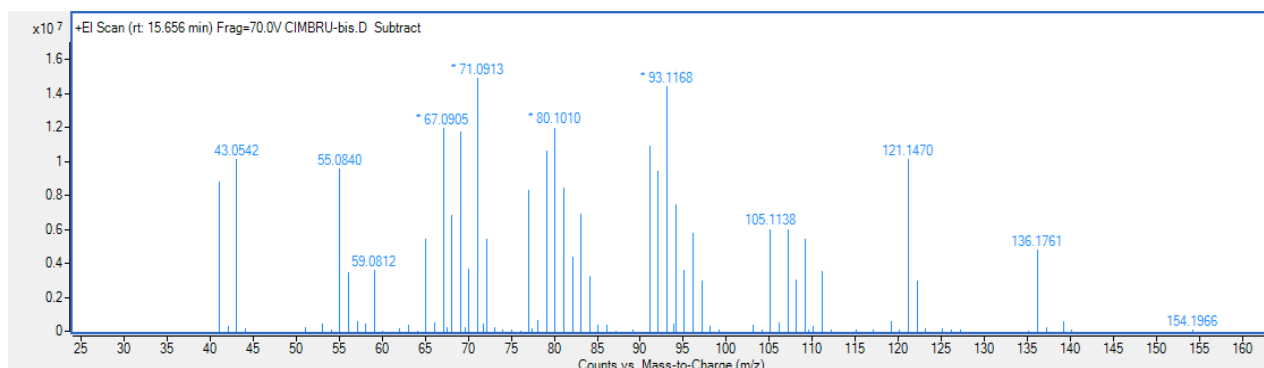


Figure S7.12: Mass spectrum for linalool.

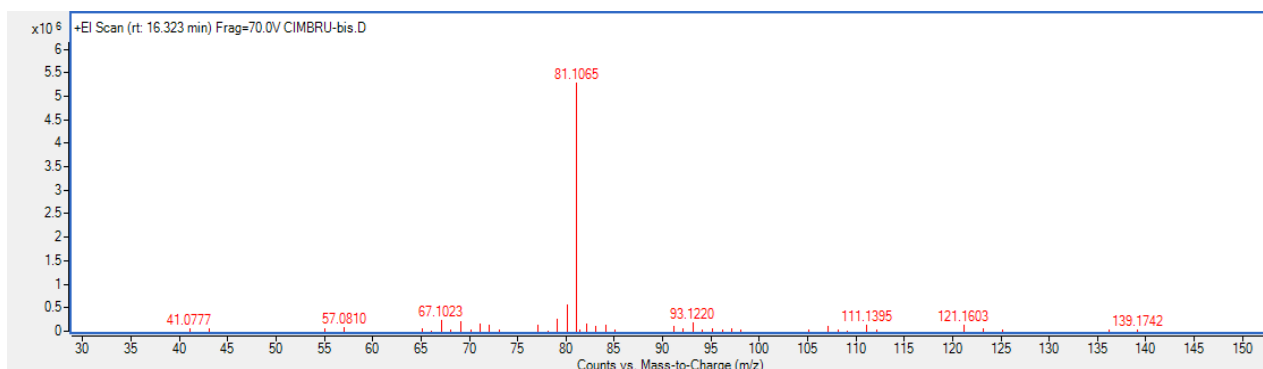


Figure S7.13: Mass spectrum for fenchol.

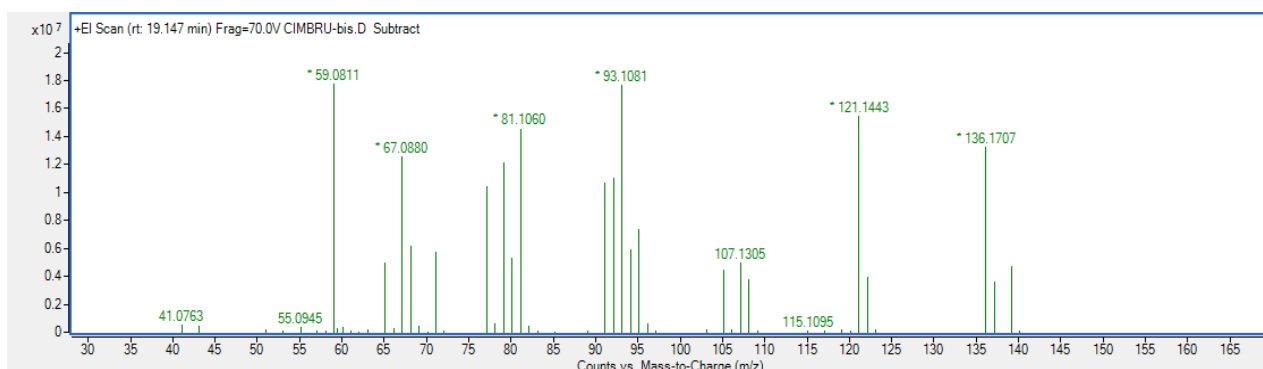


Figure S7.14: Mass spectrum for 1-terpineol.

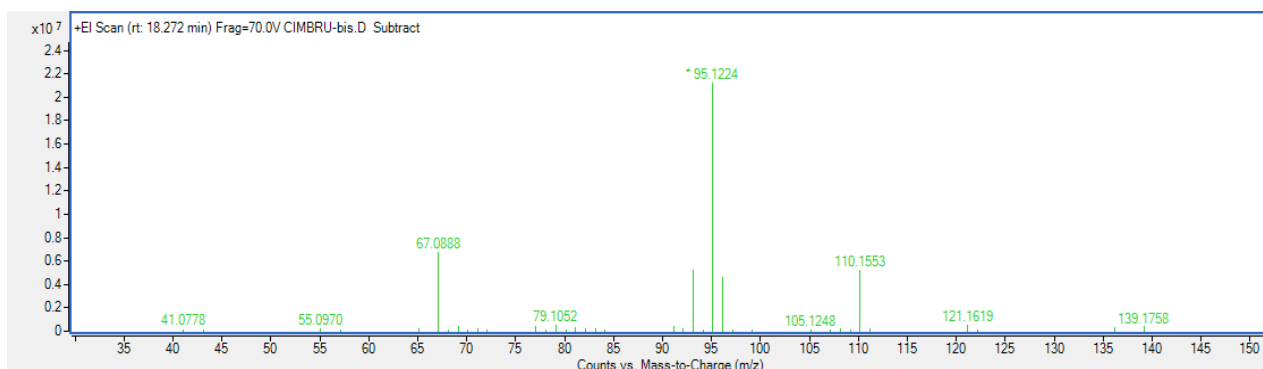


Figure S7.15: Mass spectrum for endo-borneol.

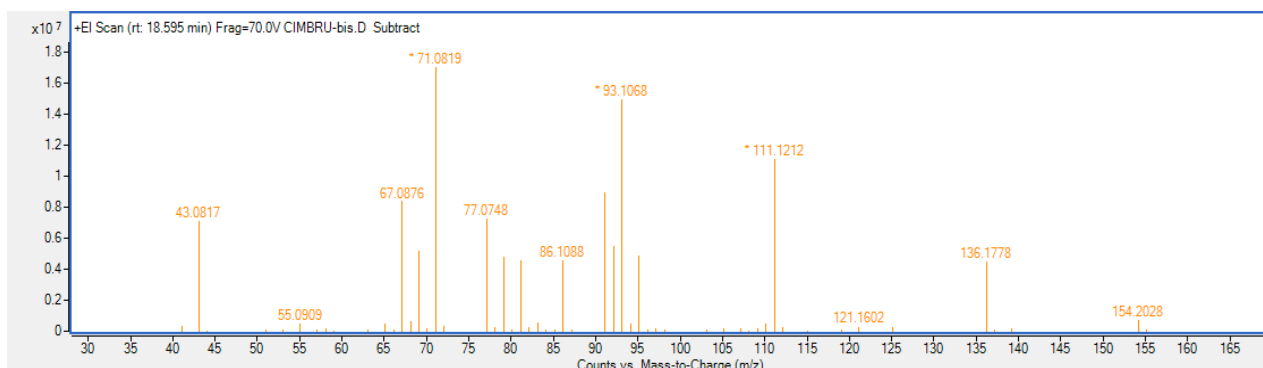


Figure S7.16: Mass spectrum for 4-terpineol.

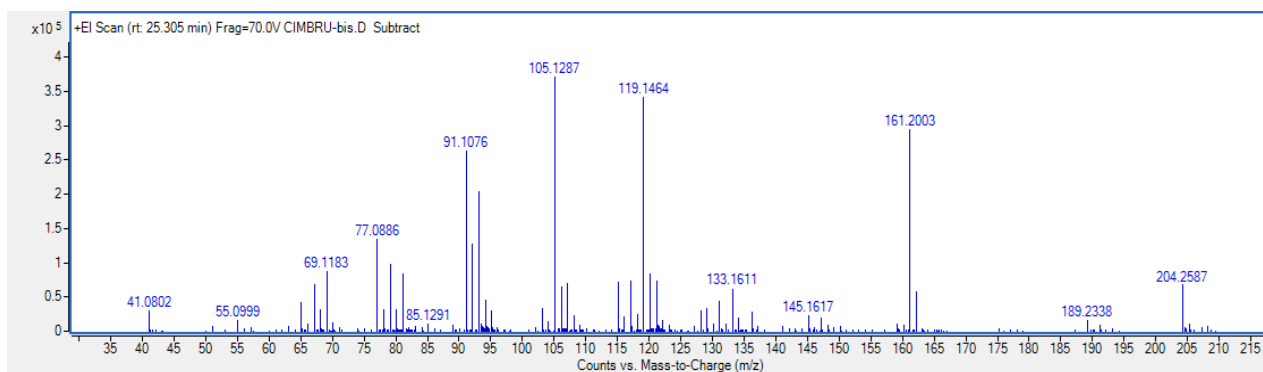


Figure S7.17: Mass spectrum for α -copaene.

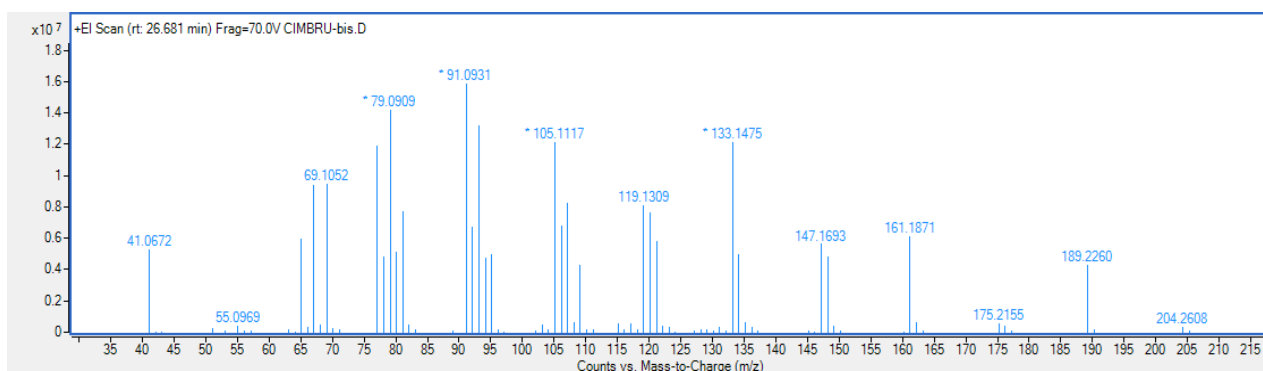


Figure S7.18: Mass spectrum for caryophyllene.

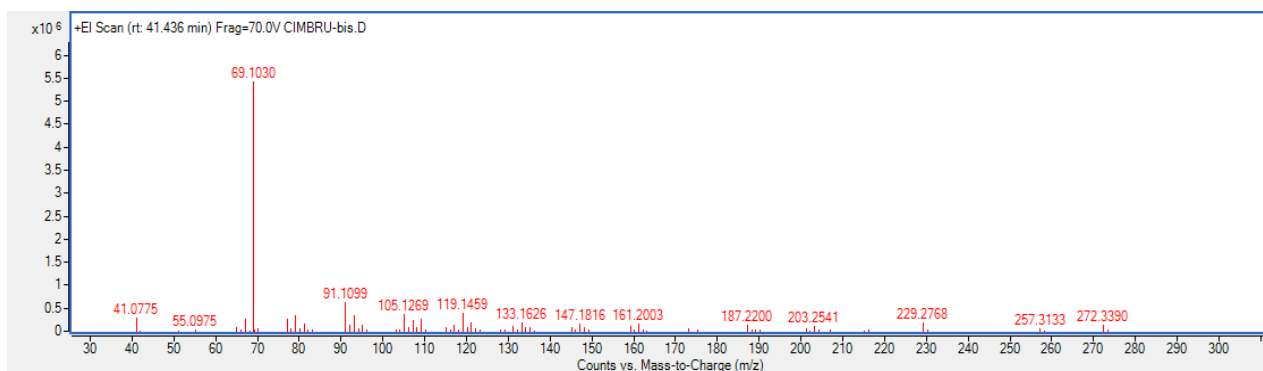


Figure S7.19: Mass spectrum for m-camphorene.

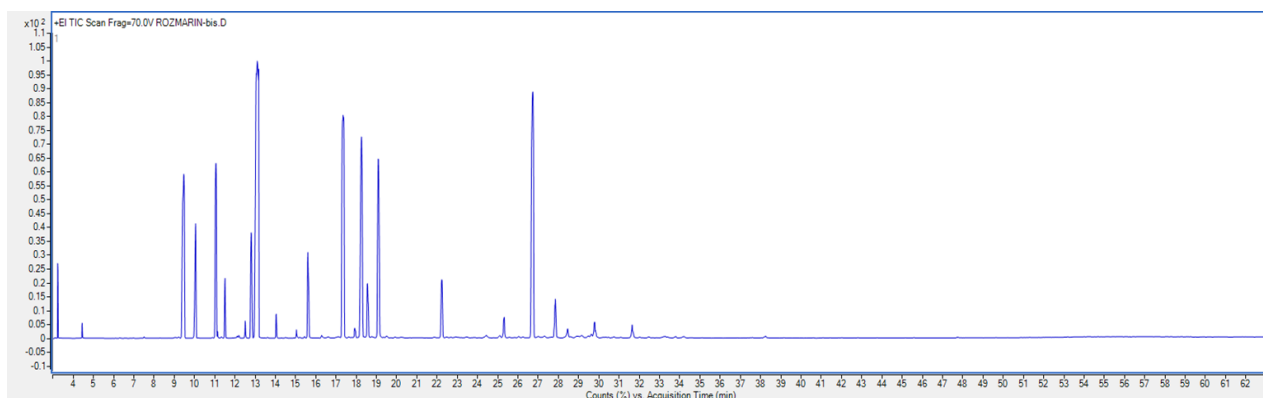


Figure S8: Chromatographic profile of the rosemary EO.

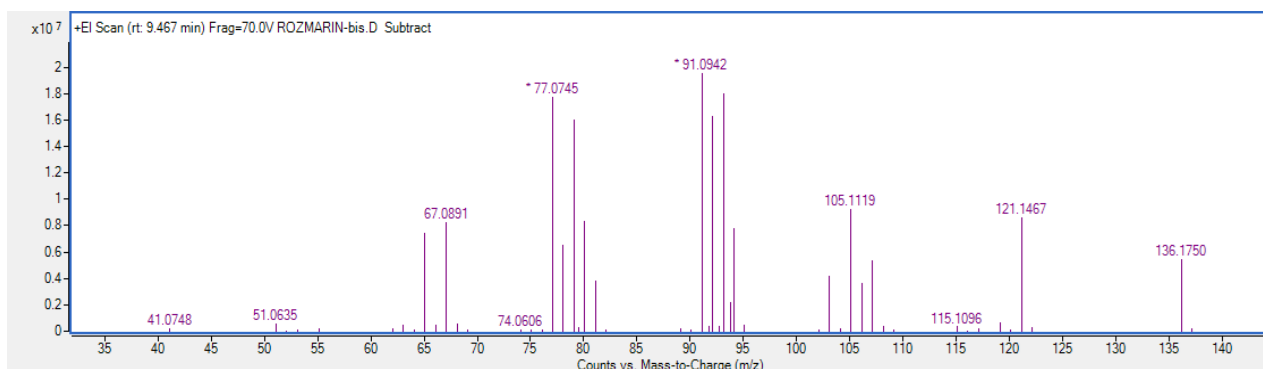


Figure S9.1: Mass spectrum for 3-carene.

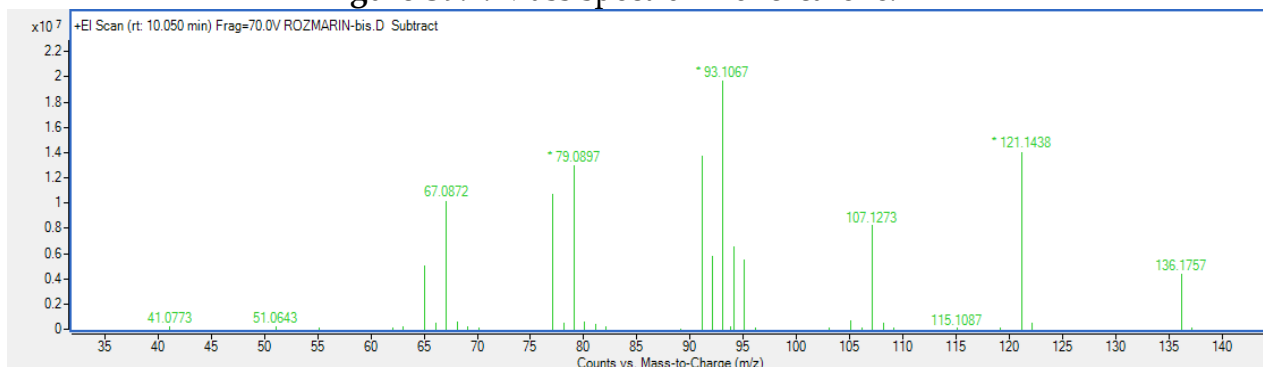


Figure S9.2: Mass spectrum for camphene.

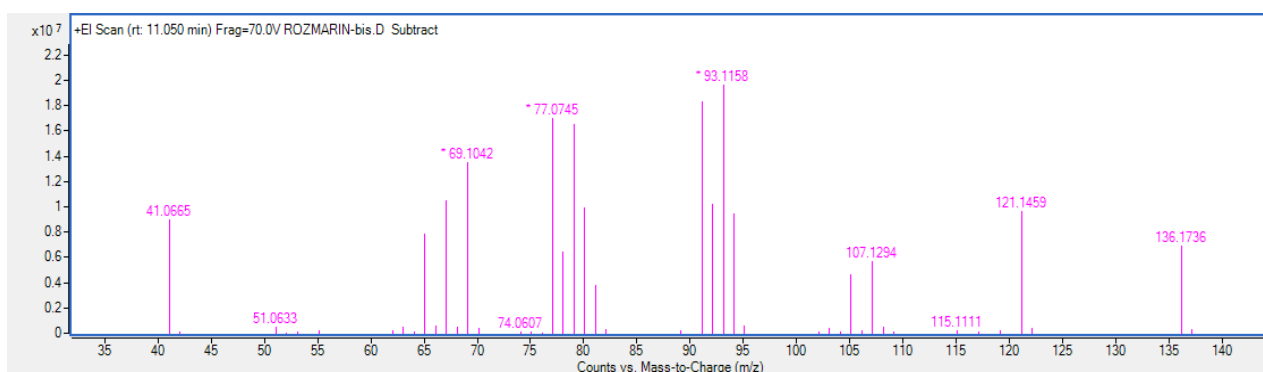


Figure S9.3: Mass spectrum for fenchene.

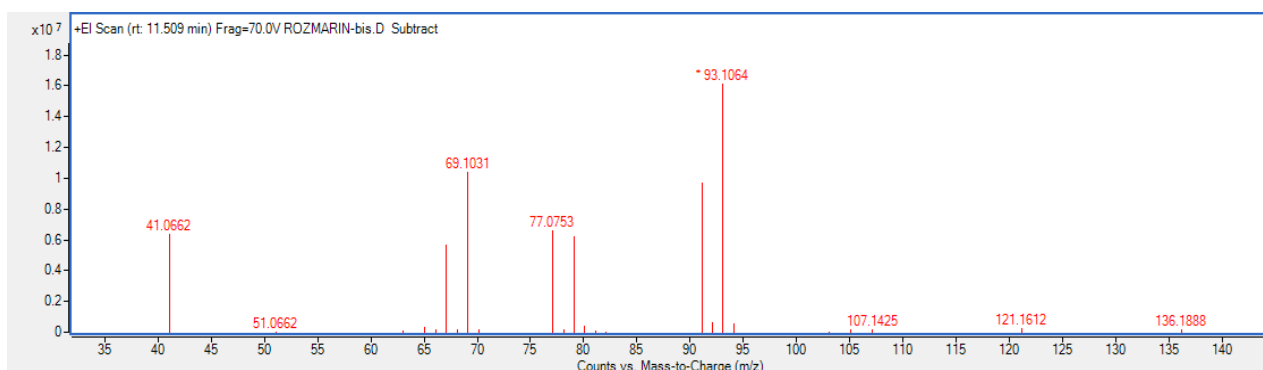


Figure S9.4: Mass spectrum for β -myrcene.

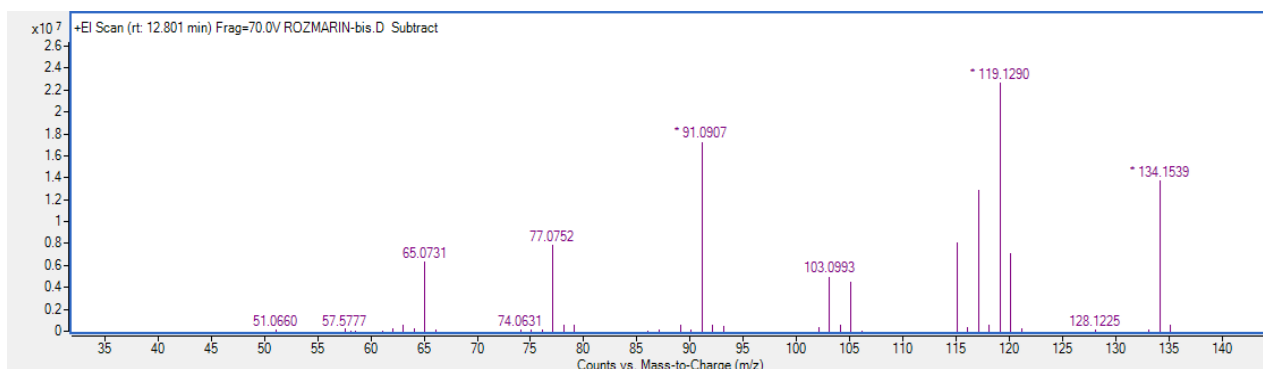


Figure S9.5: Mass spectrum for o-cymene.

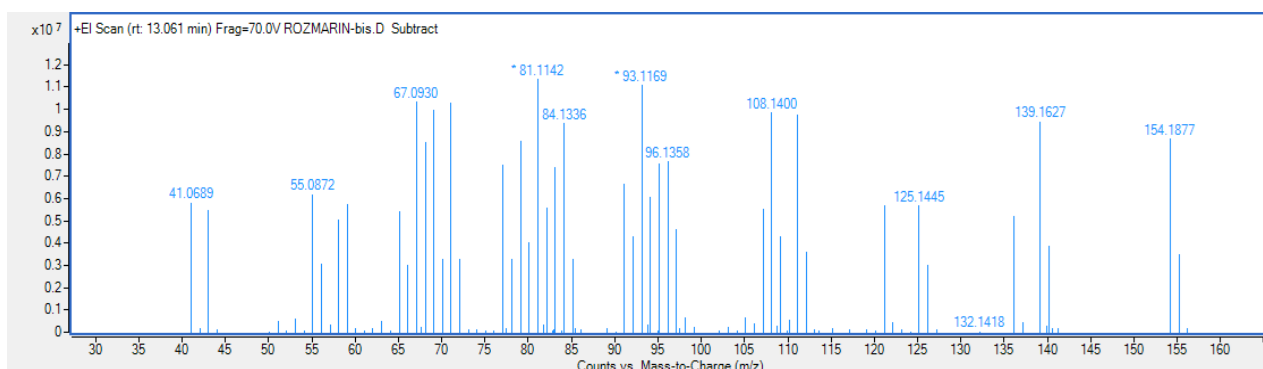


Figure S9.6: Mass spectrum for eucalyptol.

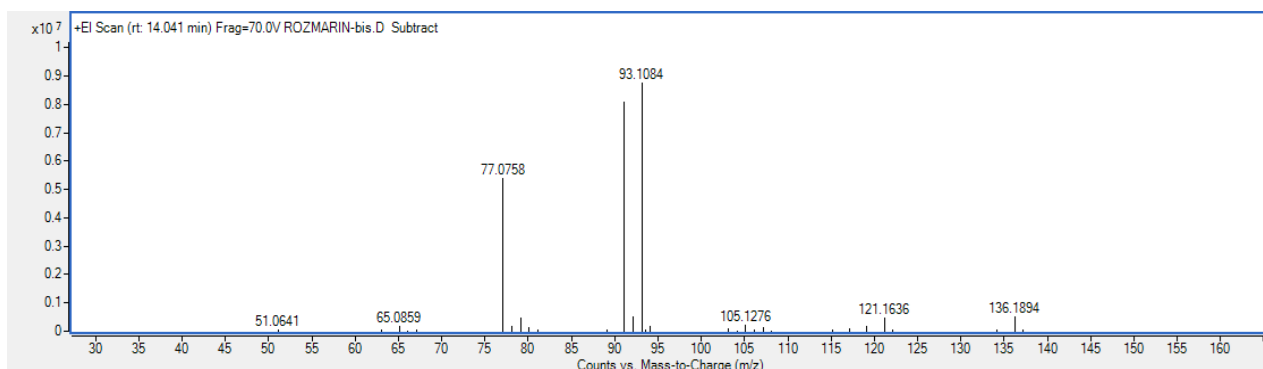


Figure S9.7: Mass spectrum for 3-thujene.

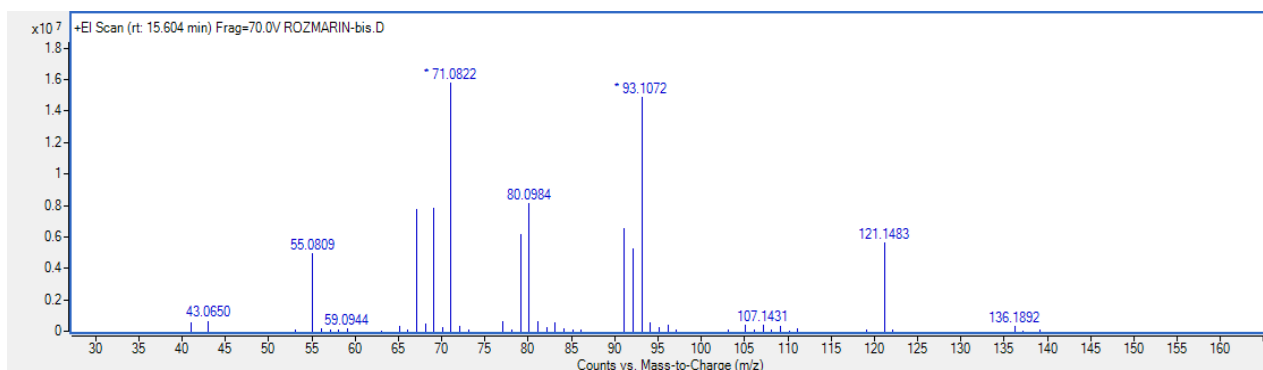


Figure S9.8: Mass spectrum for linalool.

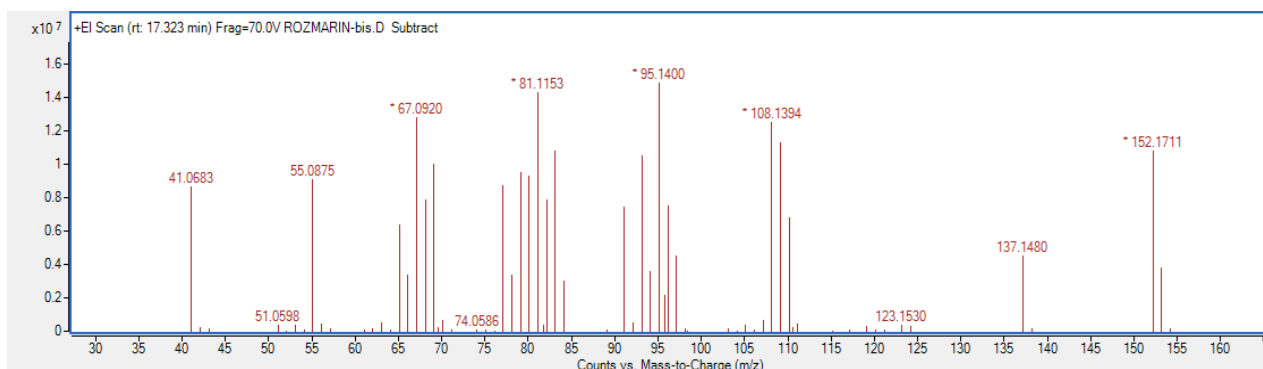


Figure S9.9: Mass spectrum for (+)-2-bornanone.

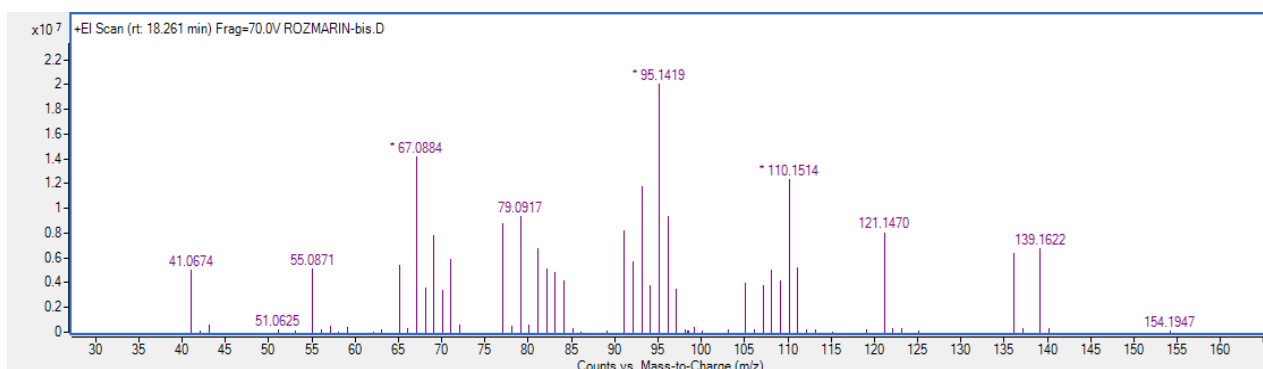


Figure S9.10: Mass spectrum for endo-borneol.

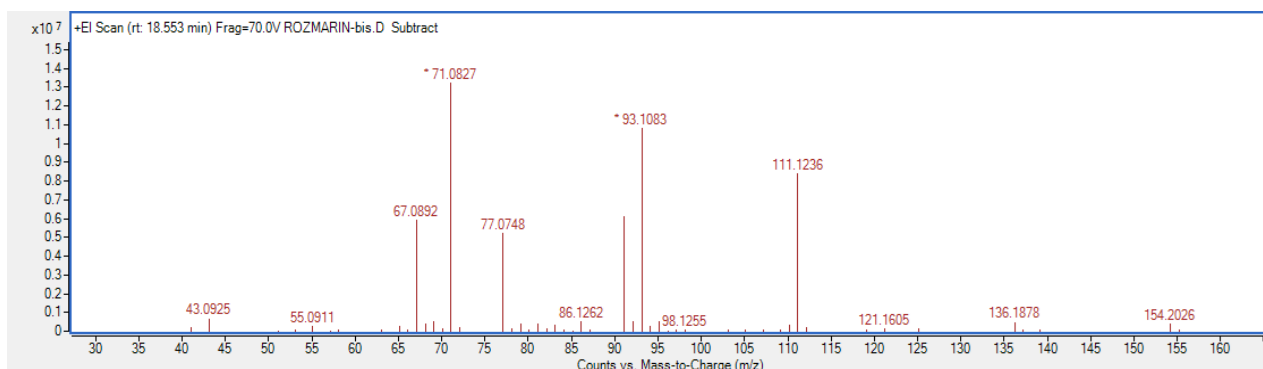


Figure S9.11: Mass spectrum for (-)-4-terpineol.

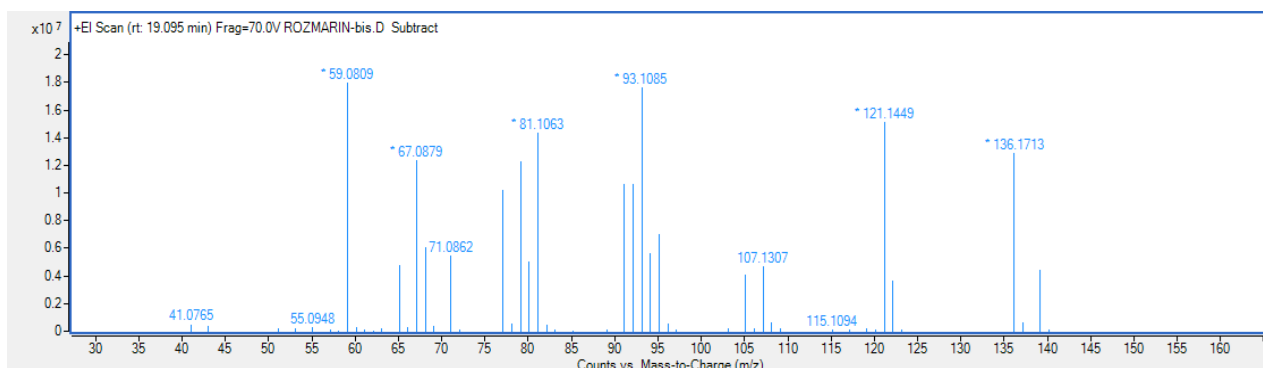


Figure S9.12: Mass spectrum for α-terpineol.

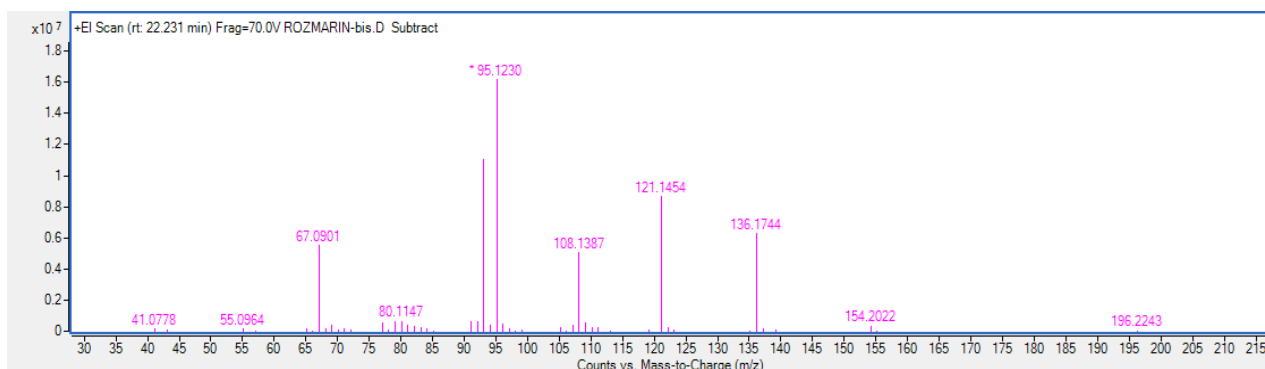


Figure S9.13: Mass spectrum for (-)-bornyl acetate.

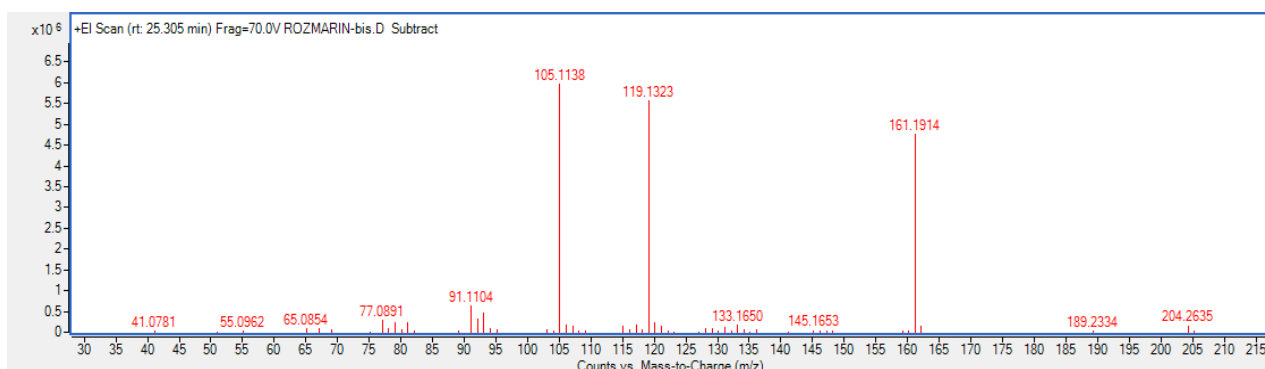


Figure S9.14: Mass spectrum for α-copaene.

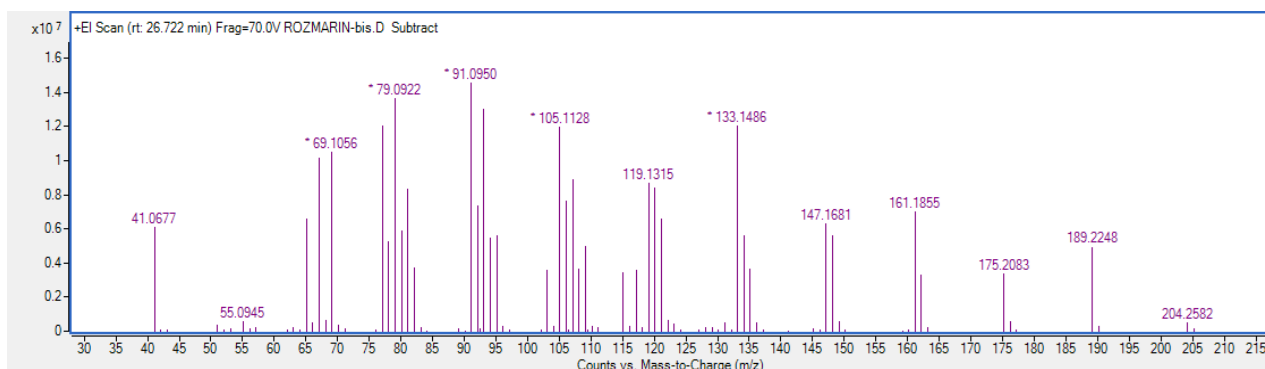


Figure S9.15: Mass spectrum for caryophyllene.

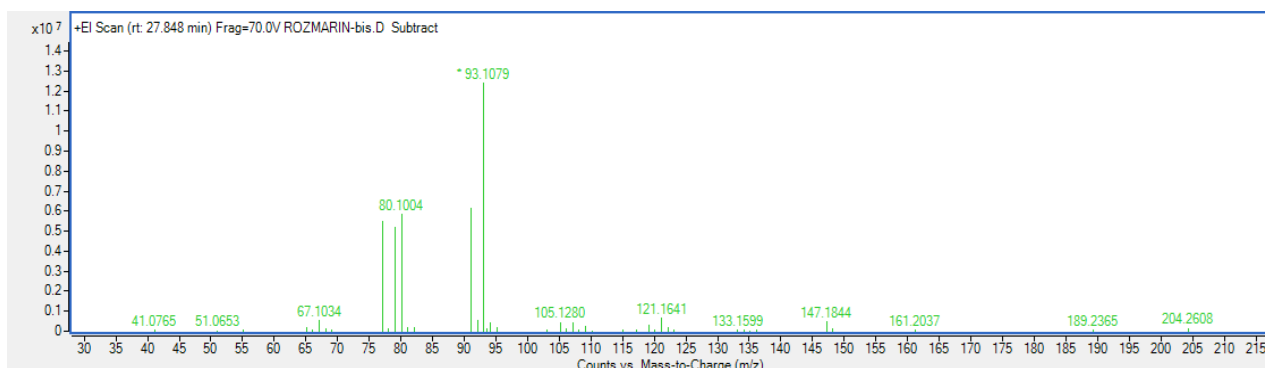


Figure S9.16: Mass spectrum for humulene.

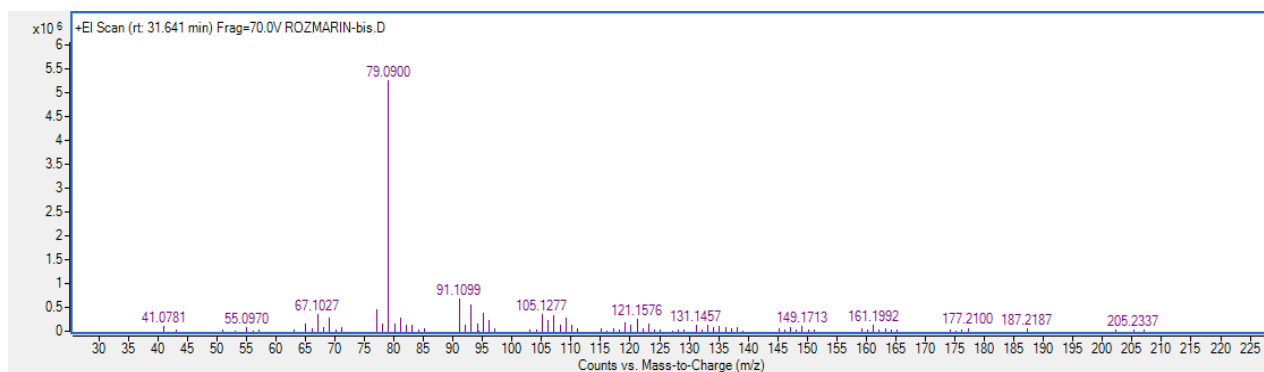


Figure S9.17: Mass spectrum for caryophyllene oxide.

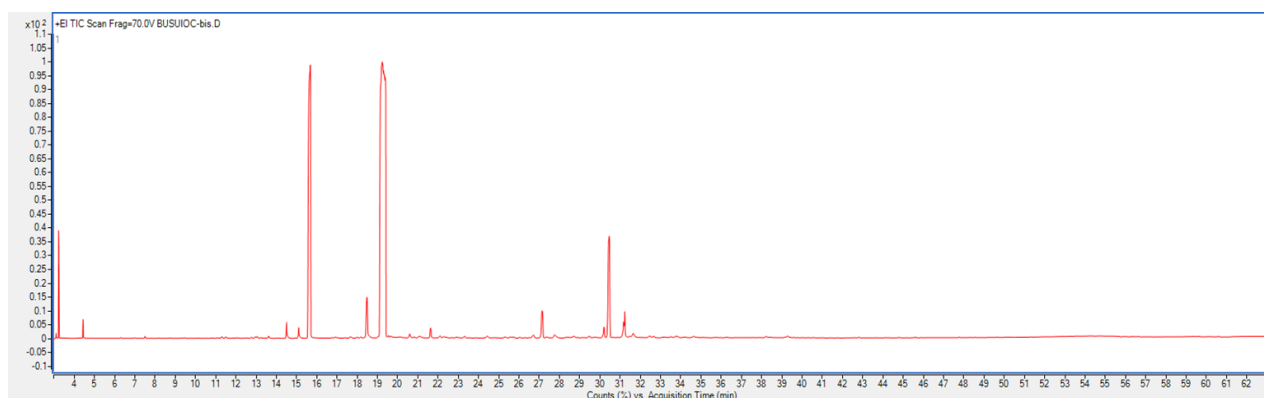


Figure S10: Chromatographic profile of the basil EO.

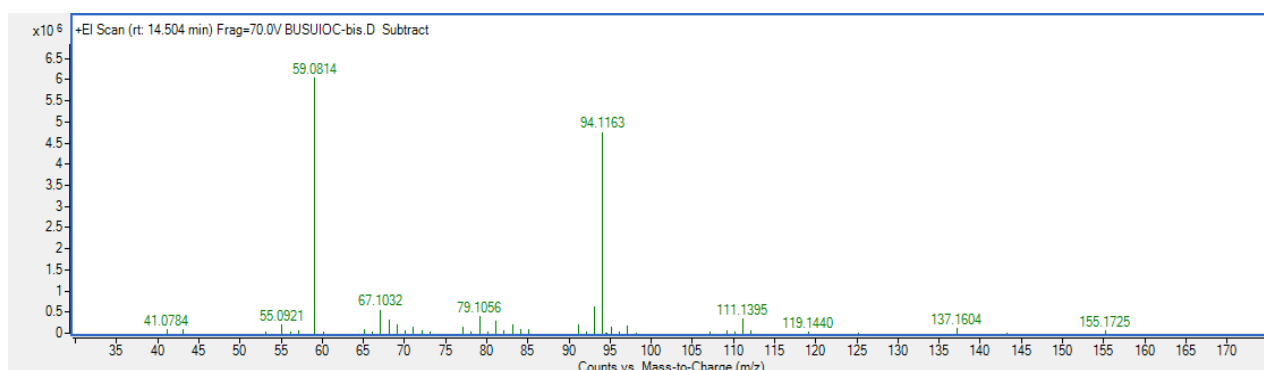


Figure S11.1: Mass spectrum for cis-linalool oxide.

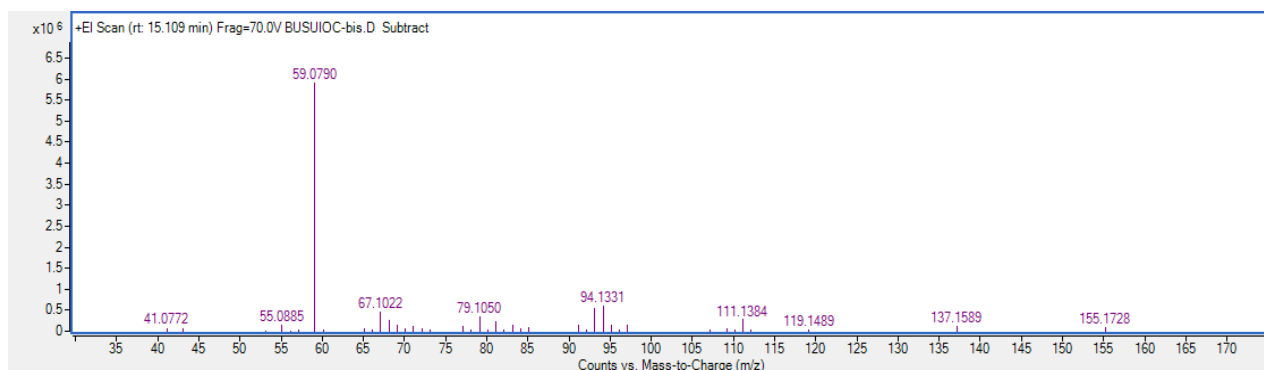


Figure S11.2: Mass spectrum for trans-linalool oxide.

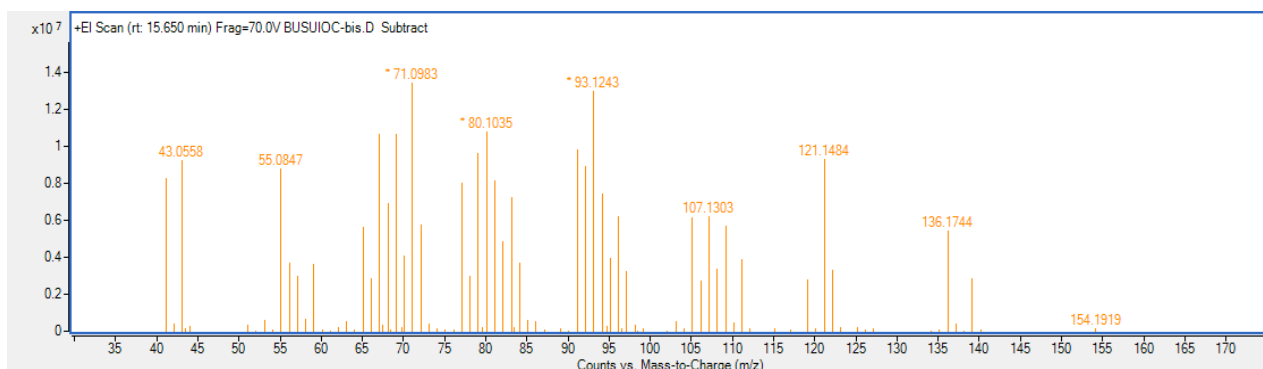


Figure S11.3: Mass spectrum for linalool.

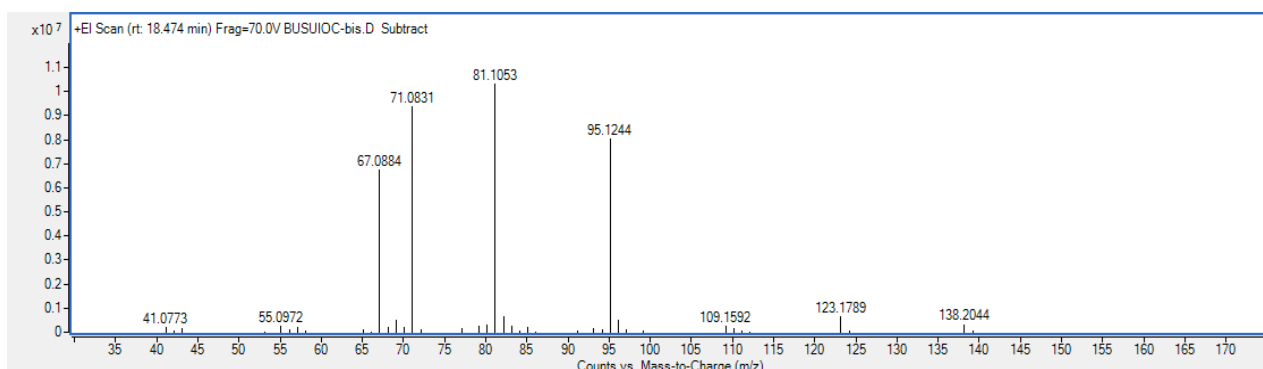


Figure S11.4: Mass spectrum for DL-menthol.

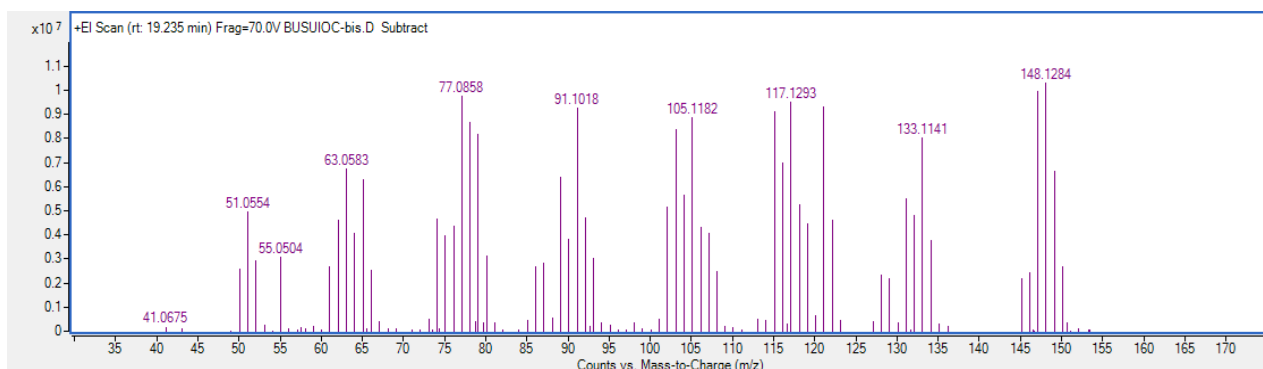


Figure S11.5: Mass spectrum for estragole.

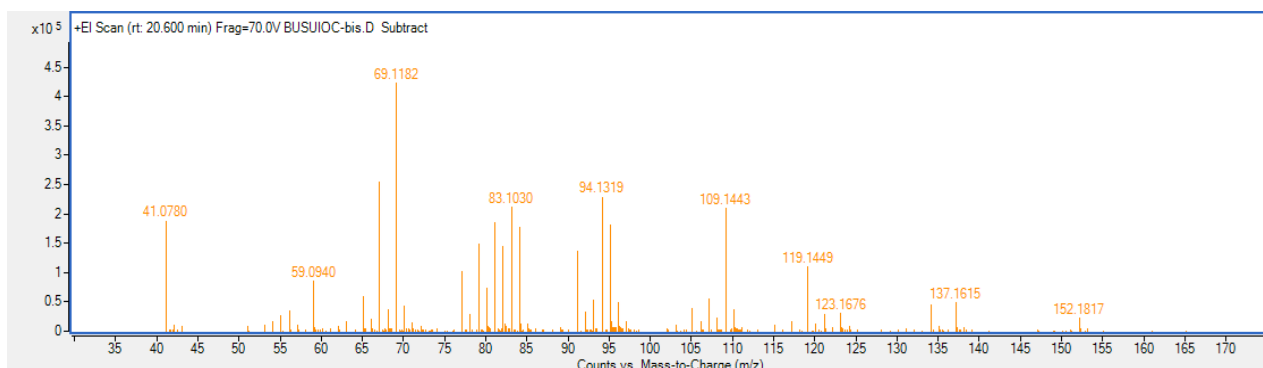


Figure S11.6: Mass spectrum for β -citral.

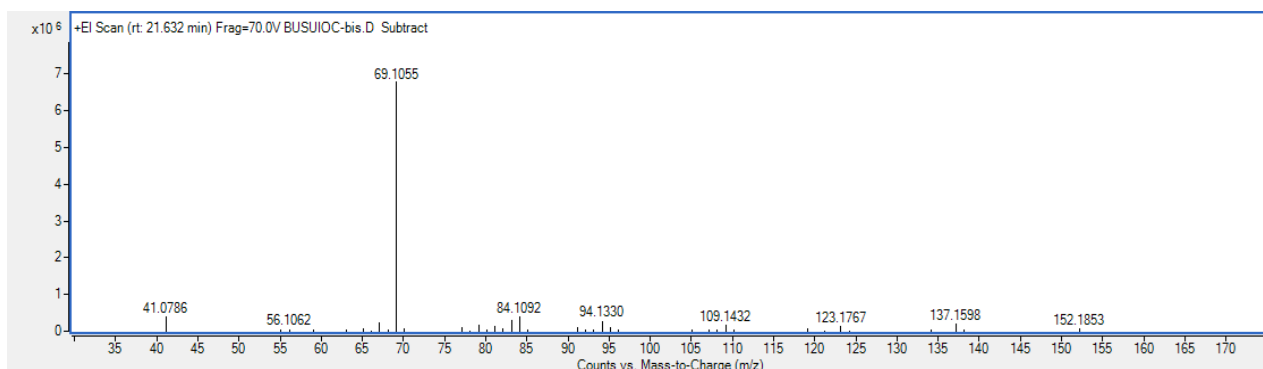


Figure S11.7: Mass spectrum for citral.

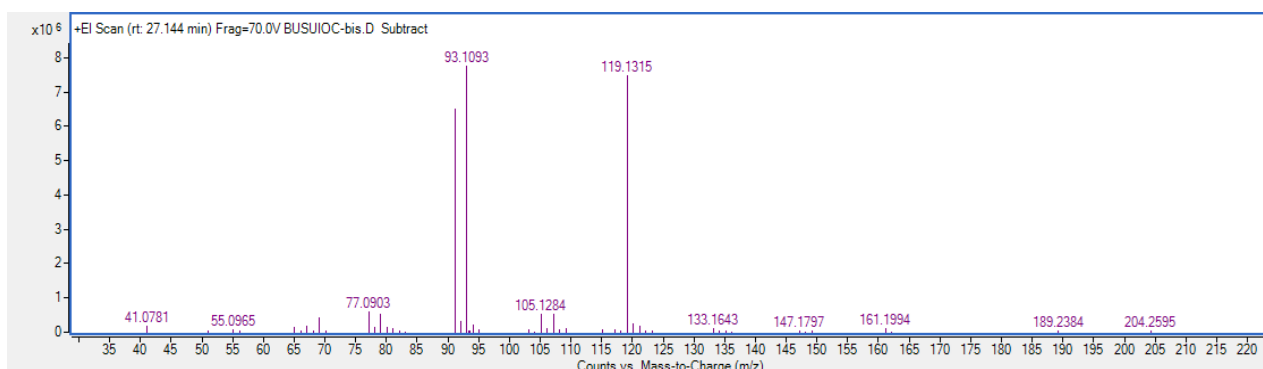


Figure S11.8: Mass spectrum for trans- α -bergamotene.

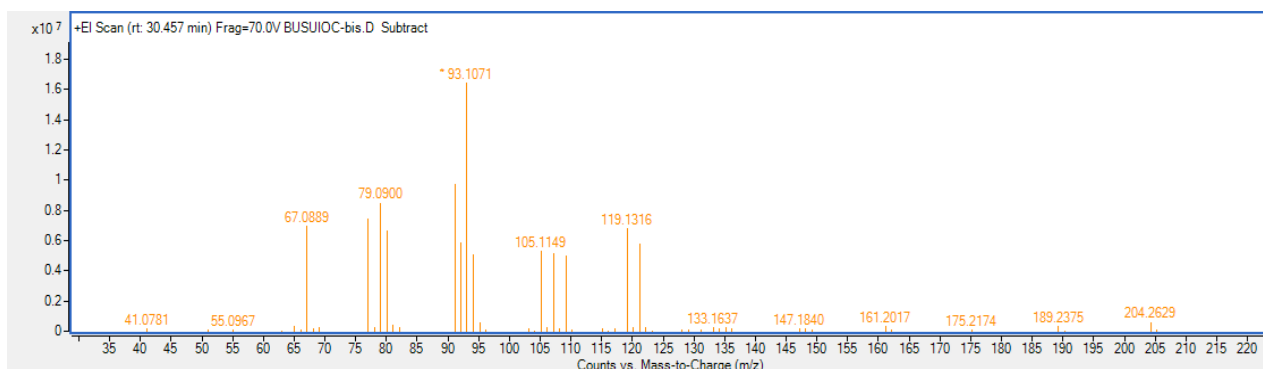


Figure S11.9: Mass spectrum for trans- α -bisabolene.

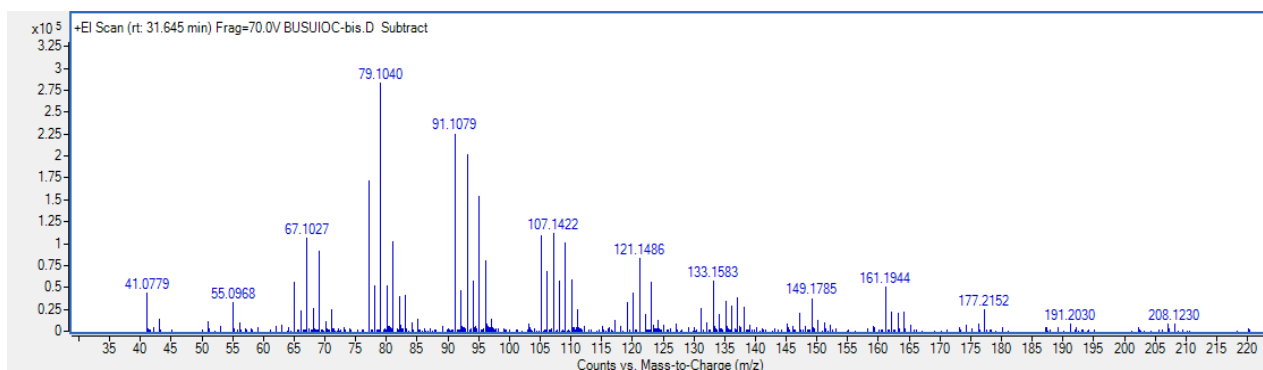


Figure S11.10: Mass spectrum for caryophyllene oxide.

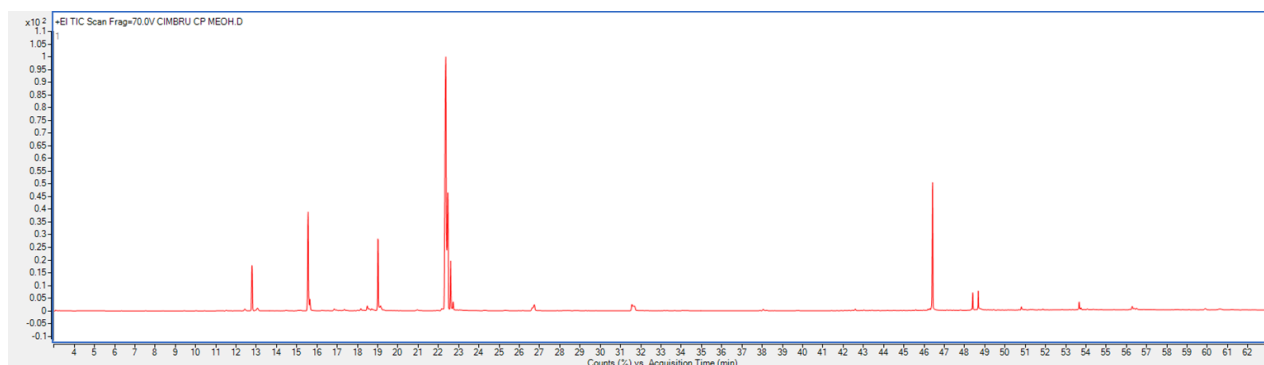


Figure S12: Chromatographic profile for Fe₃O₄@SiO₂@thyme_CP nanosystem (methanol).

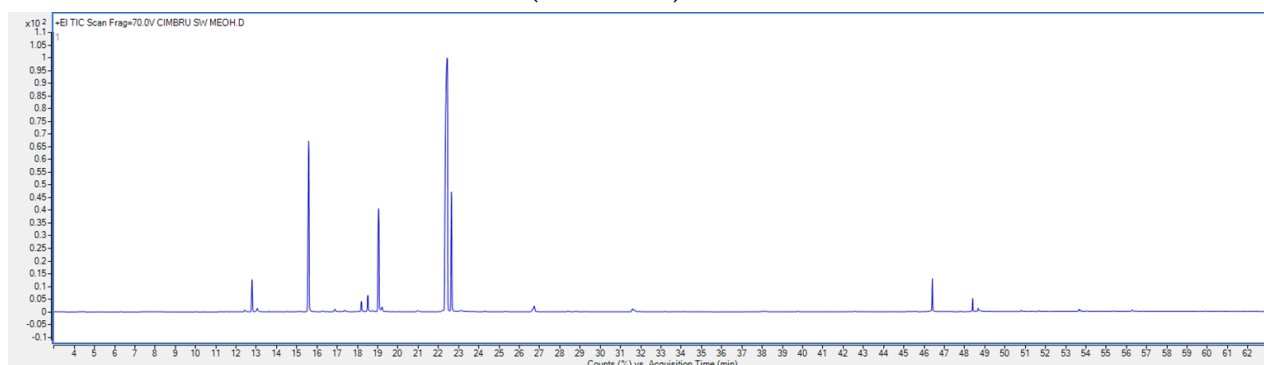


Figure S13: Chromatographic profile for Fe₃O₄@SiO₂@thyme_SW nanosystem (methanol).

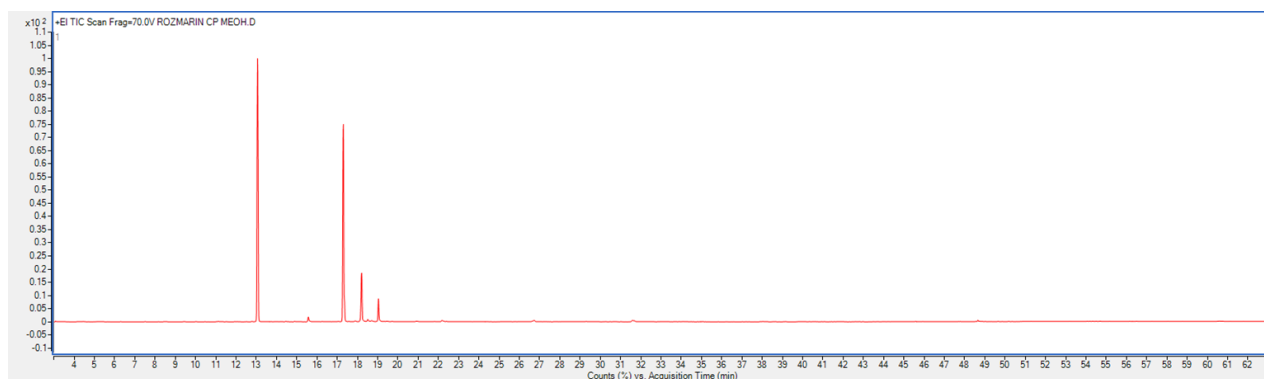


Figure S14: Chromatographic profile for Fe₃O₄@SiO₂@rosemary_CP nanosystem (methanol).

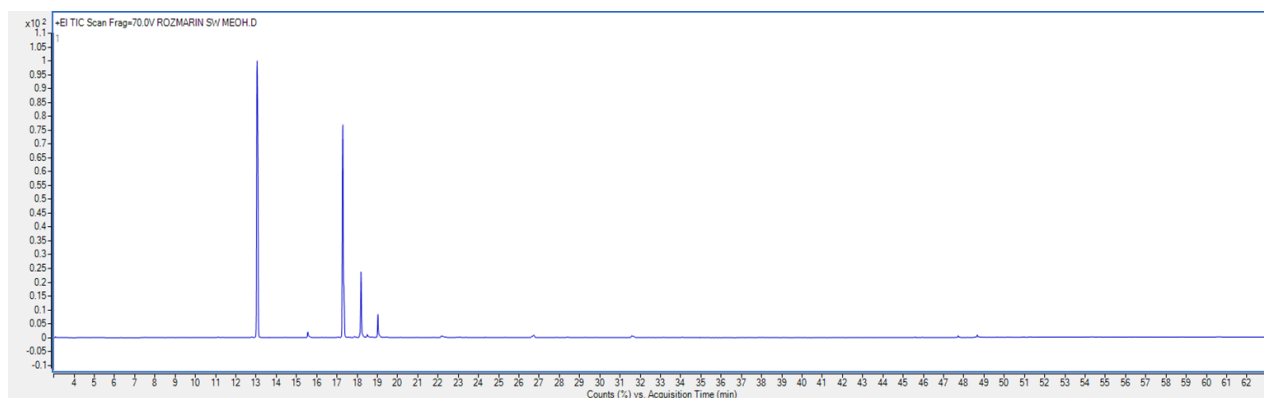


Figure S15: Chromatographic profile for Fe₃O₄@SiO₂@rosemary_SW nanosystem (methanol).

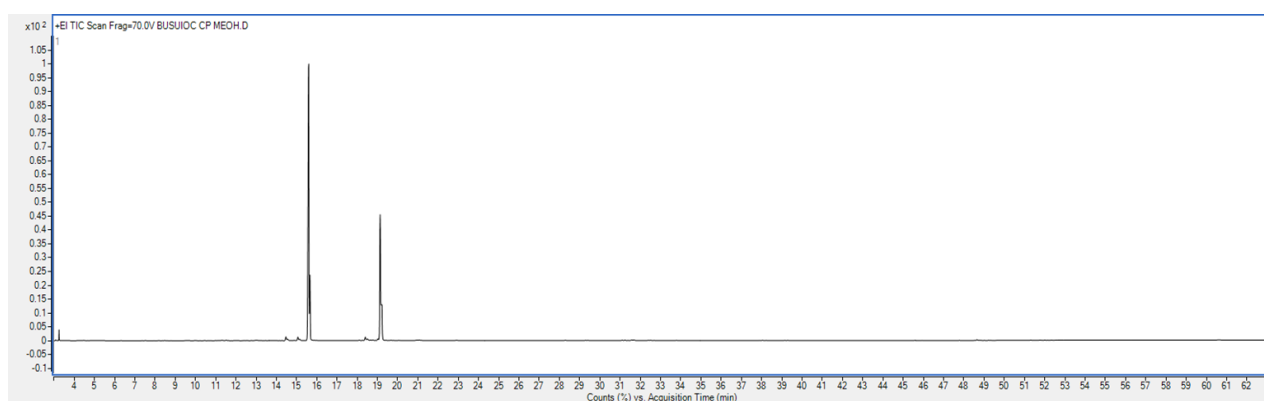


Figure S16: Chromatographic profile for Fe₃O₄@SiO₂@basil_CP nanosystem (methanol).

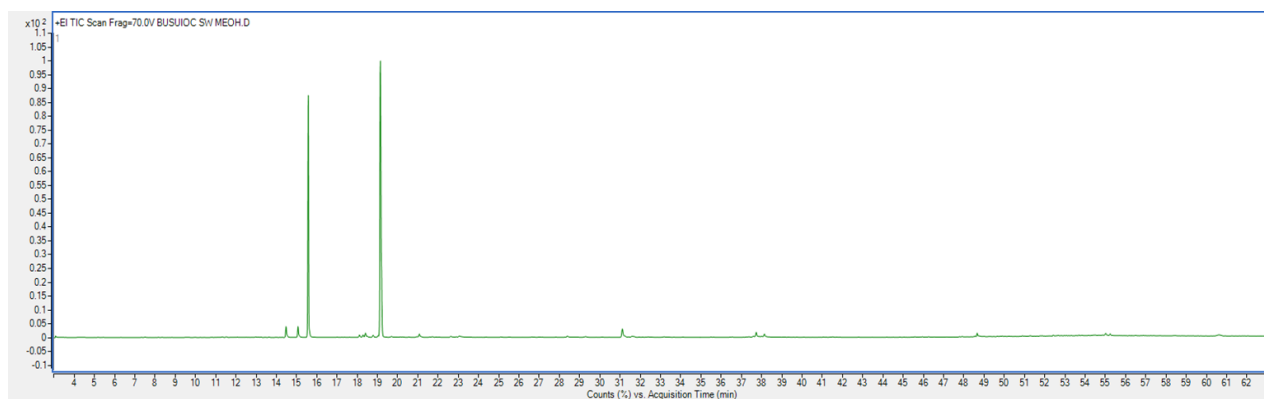


Figure S17: Chromatographic profile for Fe₃O₄@SiO₂@basil_SW nanosystem (methanol).