

## Supplementary Materials

### Photocatalytic Properties of $\text{PbMoO}_4$ Nanocrystals against Cationic and Anionic Dyes in Several Experimental Conditions

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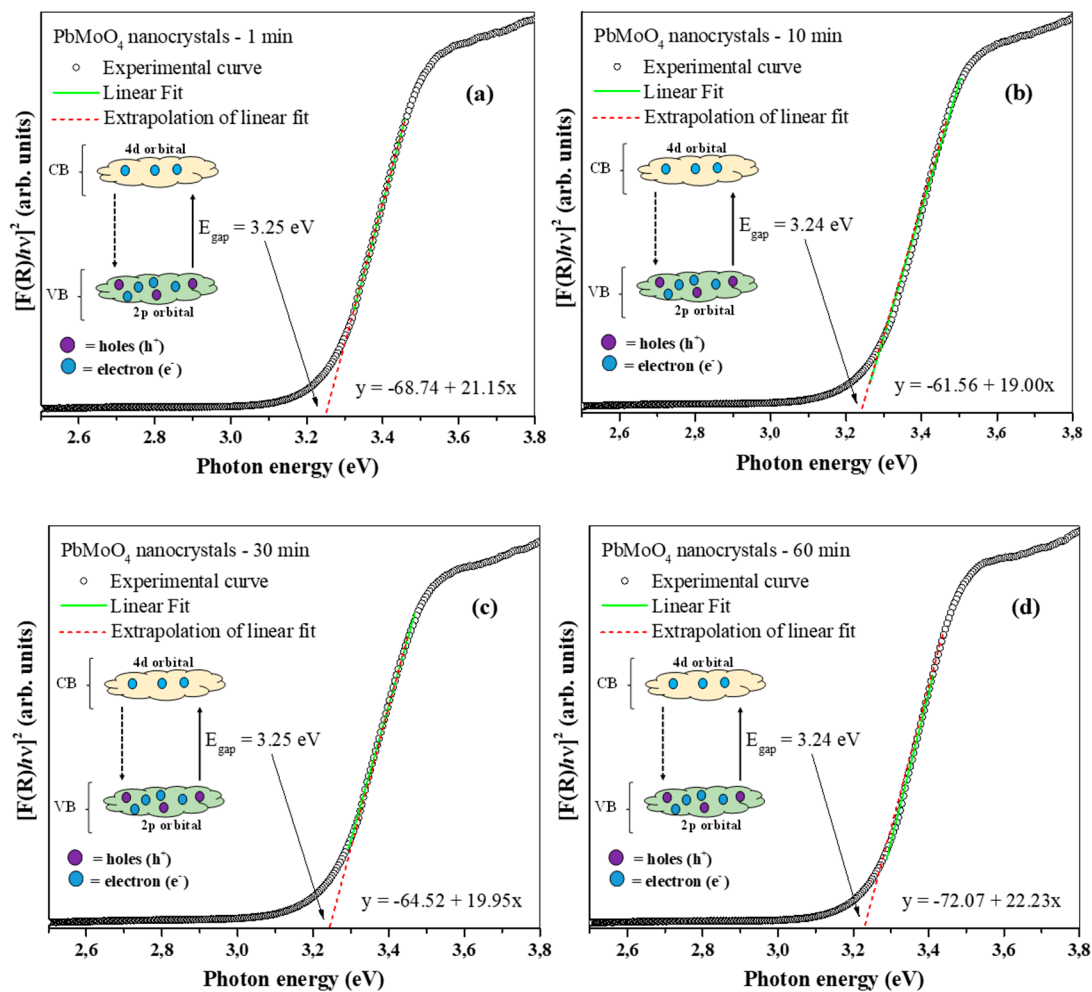
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## Chapter Figure



**Figure S1.** UV-vis spectra of PbMoO<sub>4</sub> nanocrystals synthesized at (a) 1 min, (b) 10 min, (c) 30 min and (d) 60 min by MH under 100 °C.



## Chapter Table

**Table S1** – Rietveld refinement results of PbMoO<sub>4</sub> nanocrystals.

Atoms	Wyckoff	Atomic coordinates		
		x	y	z
<b><u>PbMoO<sub>4</sub> – 1min</u></b>				
Pb	4b	0	0.25000	0.62500
Mo	4a	0	0.25000	0.12500
O	16f	0.23782	0.12102	0.04177
R <sub>p</sub> = 8.45; R <sub>wp</sub> = 11.0; R <sub>e</sub> = 6.16 and $\chi^2$ = 3.165 and GOF = 1.77.				
<b><u>PbMoO<sub>4</sub> – 10min</u></b>				
Pb	4b	0	0.25000	0.62500
Mo	4a	0	0.25000	0.12500
O	16f	0.23634	0.11447	0.04513
R <sub>p</sub> = 9.01; R <sub>wp</sub> = 11.5; R <sub>e</sub> = 6.97 and $\chi^2$ = 2.738 and GOF = 1.65.				
<b><u>PbMoO<sub>4</sub> – 30min</u></b>				
Pb	4b	0	0.25000	0.62500
Mo	4a	0	0.25000	0.12500
O	16f	0.23904	0.11805	0.04590
R <sub>p</sub> = 10.0; R <sub>wp</sub> = 12.5; R <sub>e</sub> = 7.28 and $\chi^2$ = 2.960 and GOF = 1.72.				
<b><u>PbMoO<sub>4</sub> – 60min</u></b>				
Pb	4b	0	0.25000	0.62500
Mo	4a	0	0.25000	0.12500
O	16f	0.23963	0.11430	0.05048
R <sub>p</sub> = 9.94; R <sub>wp</sub> = 12.1; R <sub>e</sub> = 7.36 and $\chi^2$ = 2.706 and GOF = 1.64.				

**Table S2.** Experimental and theoretical Raman active modes of PbMoO<sub>4</sub> nanocrystals synthesized and reported by literatures.

Assignment	This work		Moura et al. (2018) (cm <sup>-1</sup> )	Bomio et al. (2013) (cm <sup>-1</sup> )
	Experimental (cm <sup>-1</sup> )	Theoretical (cm <sup>-1</sup> )		
<i>E<sub>g</sub></i>	60	62.2	-	-
<i>B<sub>g</sub></i>		63.2	62	65
<i>B<sub>g</sub></i>	71	84.9	73	75
<i>E<sub>g</sub></i>	101	130.6	104	107
<i>A<sub>g</sub></i>	165	197.0	168	171
<i>E<sub>g</sub></i>	192	217.8	193	195
<i>B<sub>g</sub></i>	315	319.8	318	320
<i>A<sub>g</sub></i>		324.9	-	-
<i>E<sub>g</sub></i>	352	373.7	-	-
<i>B<sub>g</sub></i>		377.3	351	351
<i>E<sub>g</sub></i>	744	769.7	743	743
<i>B<sub>g</sub></i>	765	788.7	766	766
<i>A<sub>g</sub></i>	867	880.7	869	872

**Table S3.** Lattice parameters (Å and Å<sup>3</sup>), total energy differences per unit cell (eV), and electronic band gap (eV) of stoichiometric and reduced PbMoO<sub>4</sub>.

Structure / State <sup>(a)</sup>	<i>a</i>	<i>b</i>	<i>c</i>	<i>V</i>	ΔE/cell
Stoichiometric system					
Ground S0 (fully relaxed) - reference	5.4168	5.4168	11.8200	346.82	0.00
Triplet T1 (frozen from ground S0) - delocalized					0.33
Triplet T1 (atomic positions relaxed from S0 structure)					0.34
Triplet T1 (symmetry broken, lattice frozen from S0)					0.10
Triplet T1 (symmetry broken, fully relaxed) <sup>(b)</sup>	5.2567	5.2591	11.5729	319.94	0.09
Ground S0 (electronic structure from full relaxed T1)					0.03
Reduced system					
VO (S0, frozen from stoichiometric bulk) - reference	5.4168	5.4168	11.8200	346.82	0.00
VO (T1, frozen from stoichiometric bulk)					0.02
VO (T1, atomic positions and lattice fully relaxed) <sup>(c)</sup>	5.4996	5.3398	11.8164	346.95	-0.02

Experimental parameters					
Experimental, single crystal	5.4312	5.4312	12.1065	357.12	-
Experimental, this work	5.4332	5.4332	12.1027	357.28	-

**Legend:** (a) S for singlet (diamagnetic), T for triplet (ferromagnetic); (b)  $\alpha = \beta = 90^\circ$ ;  $\gamma = 89.87^\circ$ ; (c)  $\alpha = 90.65^\circ$ ;  $\beta = 90.24^\circ$ ;  $\gamma = 90.89^\circ$ .