## Miniaturized Bioaffinity Assessment Coupled to Mass Spectrometry for Guided Purification of Bioactives from Toad and Cone Snail

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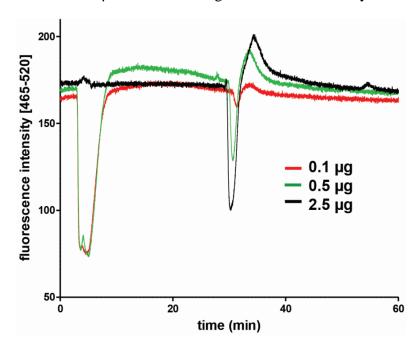
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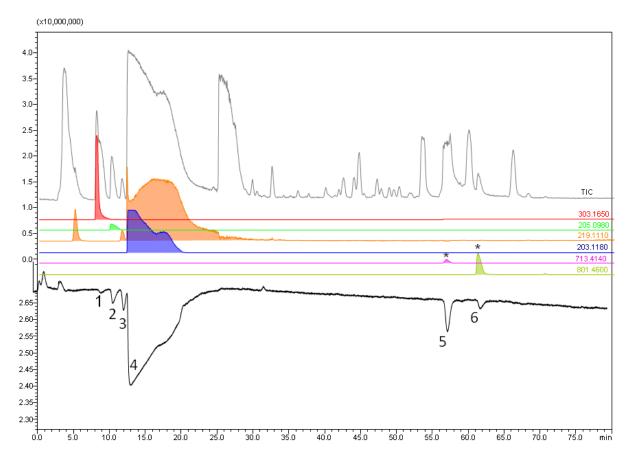
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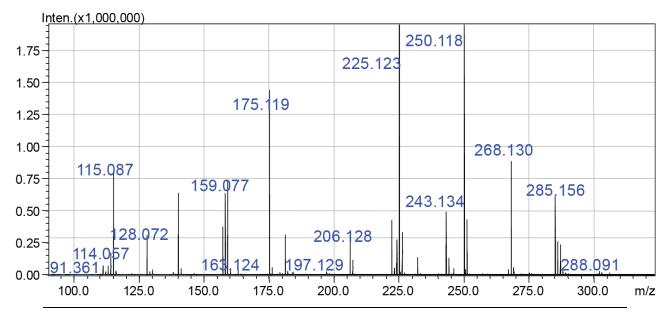
**Figure S1.** On-line bioassay analysis of three consecutive 500 nL sample injections containing 0.1, 0.5 and 2.5  $\mu$ g crude *Conus textile* venom. The 0.1 and 0.5  $\mu$ g venom injections also contained 40  $\mu$ M nicotine to align the MS and bioassay trace.



**Figure S2.** An analysis of a 500 nL sample containing 5 µg *Bufo alvarius* skin secretion extract obtained identical binding signals and attributed masses as the *Bufo marinus* extract analysis.

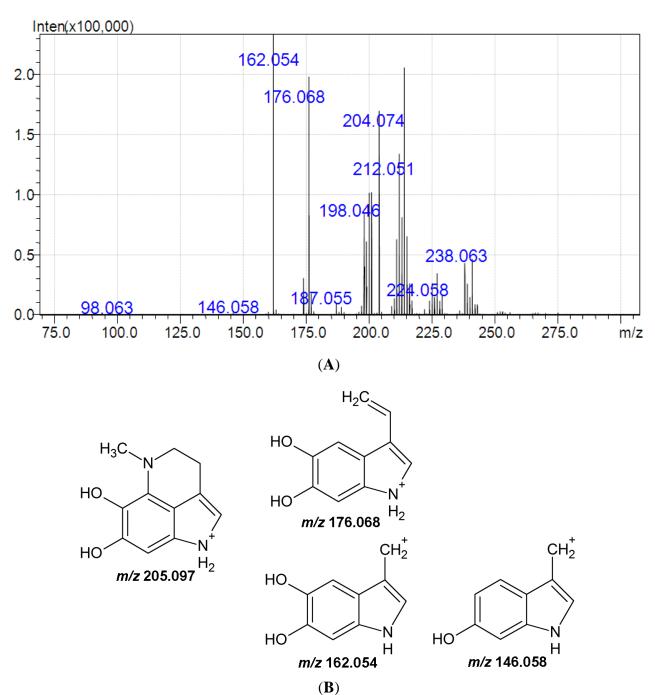


**Figure S3.**  $MS^2$  spectrum of the unknown bioactive compound with m/z 303.168, eluting at 8.5 minutes. In the table below Figure S3, the proposed molecular formula, and proposed fragmentation of the compound is shown.

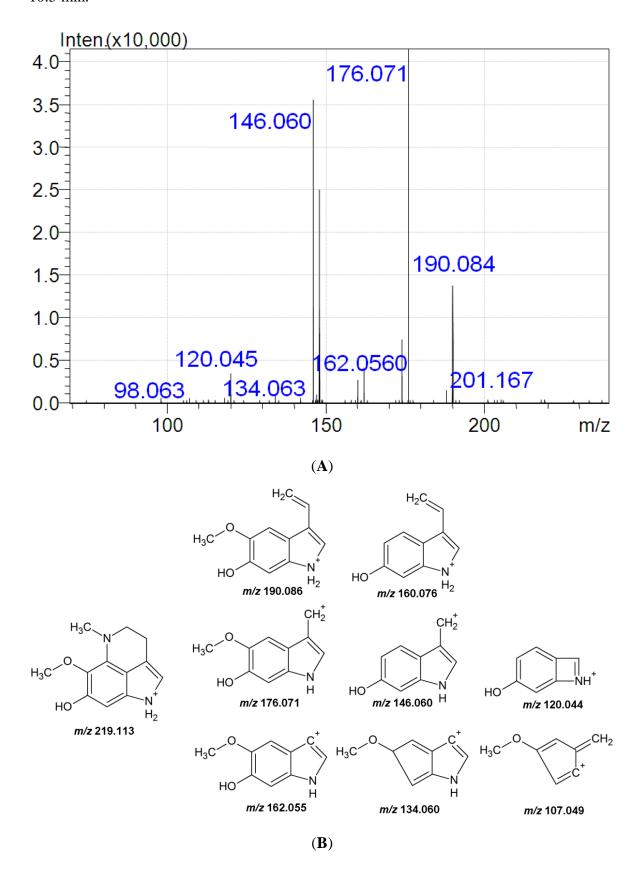


m/z	Formula	Interpretation
303.168	$C_{12}H_{23}N_4O_5^{}$	$[M+H]^+$
285.156	$C_{12}H_{21}N_4O_4$	Loss of H <sub>2</sub> O
268.130	$C_{12}H_{18}N_3O_4$	Loss of H <sub>2</sub> O and NH <sub>3</sub>
250.118	$C_{12}H_{16}N_3O_3$	Loss of 2×H <sub>2</sub> O and NH <sub>3</sub>
243.134	$C_{11}H_{19}N_2O_4$	Loss of CH <sub>4</sub> N <sub>2</sub> O, e.g., NH <sub>3</sub> and HNCO
225.123	$C_{11}H_{17}N_2O_3$	Loss of H <sub>2</sub> O from m/z 243
181.133	$C_{10}H_{17}N_2O$	Loss of C <sub>2</sub> H <sub>6</sub> N <sub>2</sub> O <sub>4</sub> , e.g., 2xH <sub>2</sub> O and 2xHNCO
175.119	$C_6H_{15}N_4O_2$	Loss of $C_6H_8O_3$
159.077	$C_6H_{11}N_2O_3$	Loss of $C_6H_{12}N_2O_2$
140.082	$C_6H_{10}N_3O$	Loss of C <sub>6</sub> H <sub>13</sub> NO <sub>4</sub>
128.072	$C_6H_{10}NO_2$	Loss of $C_6H_{13}N_3O_3$
115.087	$C_5H_{11}N_2O$	Loss of $C_7H_{12}N_2O_4$ or $C_6H_{12}N_2O_2$ and $CO_2$
91.361		Not consistent, too big mass defect

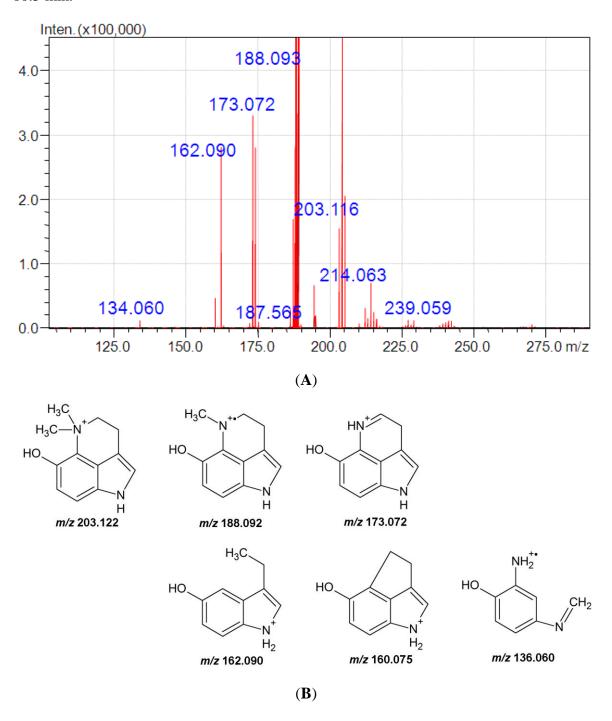
**Figure S4.** (A)  $MS^2$  spectra of the bioactive compound with m/z 205.098, eluting at 10.0 min. (B) The proposed structure, and proposed fragmentation scheme of the bioactive compound with m/z 205.098, eluting at 10.0 min.



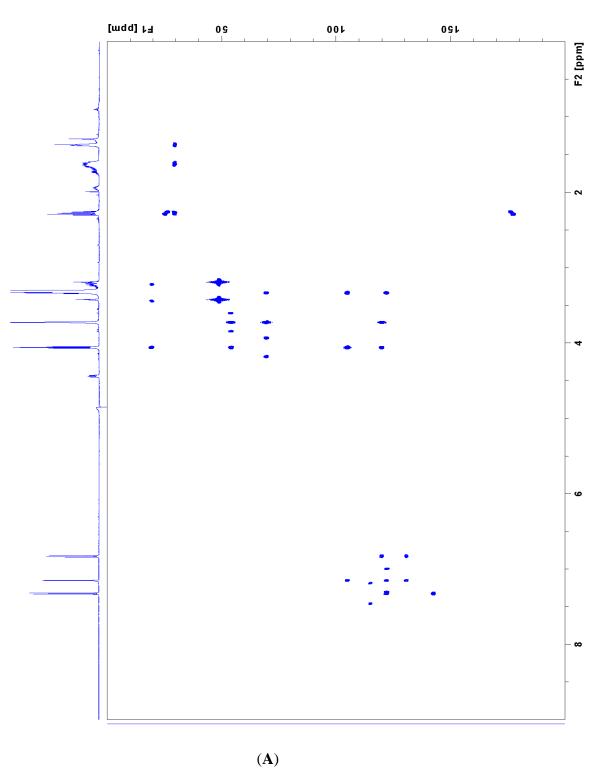
**Figure S5.** (A)  $MS^2$  spectrum, proposed structure, and proposed fragmentation scheme of the bioactive compound with m/z 219.112, eluting at 10.5 min. (B) Proposed structure and proposed fragmentation scheme of the bioactive compound with m/z 219.112, eluting at 10.5 min.

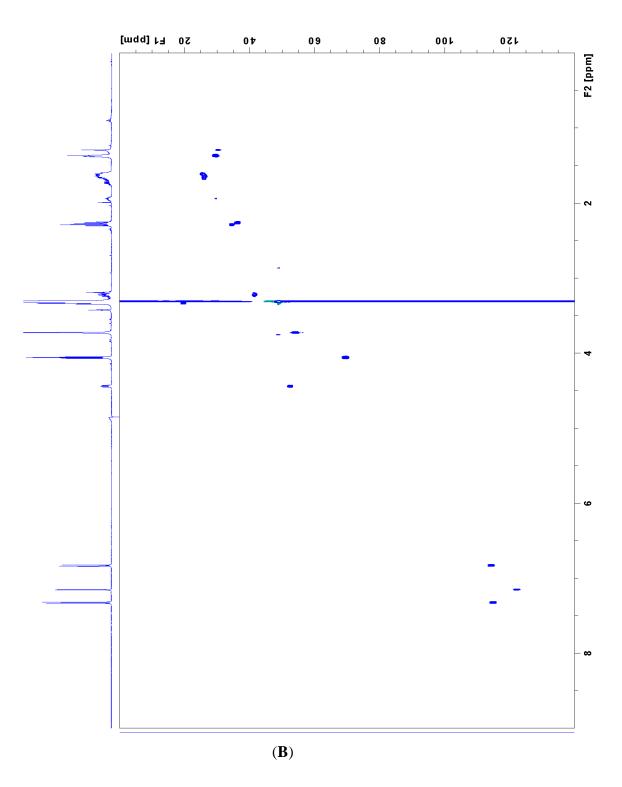


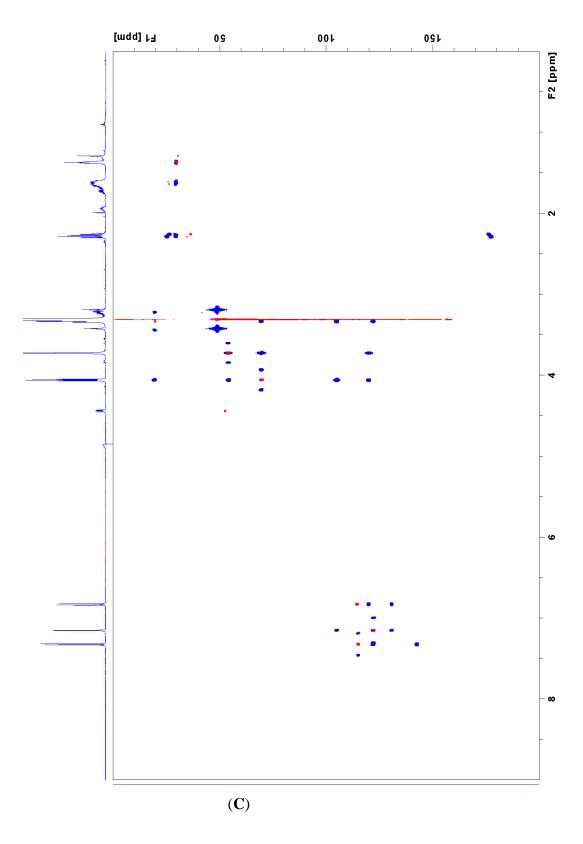
**Figure S6.** (A)  $MS^2$  spectrum, proposed structure, and proposed fragmentation scheme of the bioactive compound with m/z 203.118, eluting at 10.5 min. (B) Proposed structure, and proposed fragmentation scheme of the bioactive compound with m/z 203.118, eluting at 10.5 min.



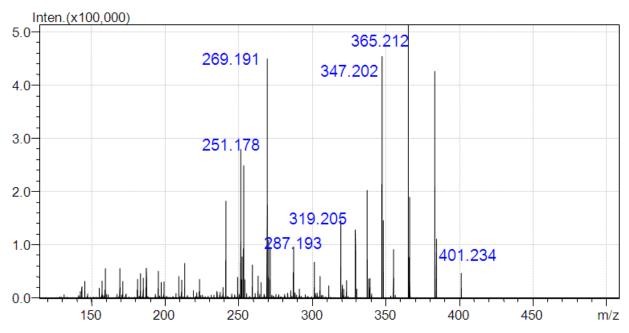
**Figure S7.** (A) HMBC-NMR spectrum of the bioactive compound with m/z 203.118, eluting at 10.5 min. (B) HSQC-NMR spectrum of the bioactive compound with m/z 203.118, eluting at 10.5 min. (C) HSQCHMBC-NMR spectrum of the bioactive compound with m/z 203.118, eluting at 10.5 min.





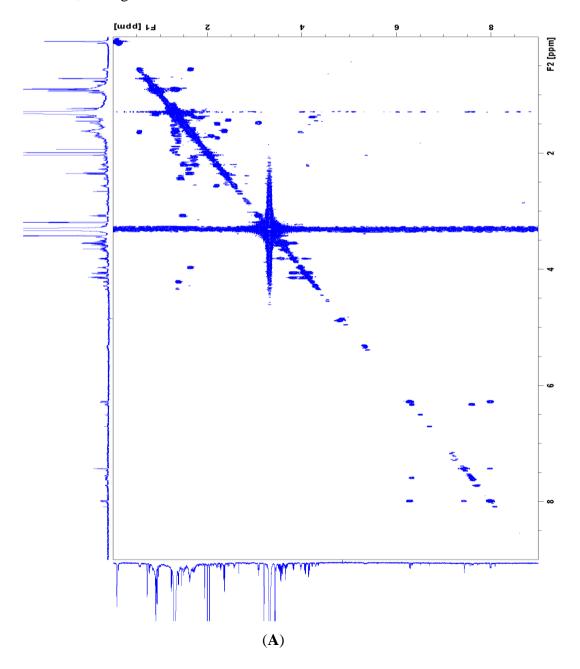


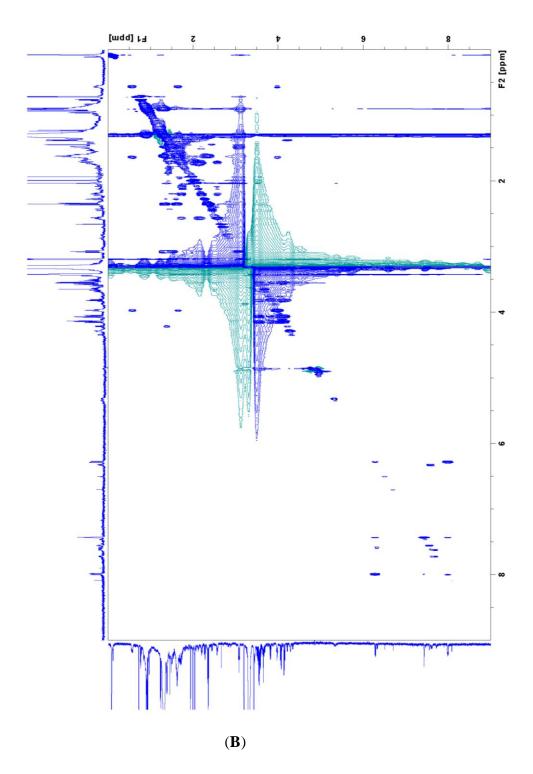
**Figure S8.**  $MS^2$  spectrum, proposed structure, and proposed fragmentation scheme of the bioactive compound with m/z 401.235, eluting at 58.0 min. Most of the peaks in the  $MS^2$  spectrum can be readily interpreted, e.g., in terms of water and CO losses, although it is generally unclear where exactly these losses occur.



m/z	Formula	Interpretation
401.234	$C_{24}H_{33}O_5^{}$	$[M+H]^+$
383.222	$C_{24}H_{31}O_4^{+}$	Loss of H <sub>2</sub> O
365.212	$C_{24}H_{29}O_3^{+}$	Loss of 2 $\times$ H <sub>2</sub> O; Loss of H <sub>2</sub> O from $m/z$ 383
355.228	$C_{23}H_{31}O_5^{}$	Loss of HCOOH or CO and H <sub>2</sub> O; Loss of CO from m/z 383
347.202	$C_{24}H_{27}O_2^{+}$	Loss of 3 $\times$ H <sub>2</sub> O; Loss of H <sub>2</sub> O from $m/z$ 365
337.218	$C_{23}H_{29}O_2^{+}$	Loss of 2 $\times$ H <sub>2</sub> O and CO; Loss of H <sub>2</sub> O from $m/z$ 355
329.191	$C_{24}H_{25}O^{+}$	Loss of 4 $\times$ H <sub>2</sub> O; Loss of H <sub>2</sub> O from $m/z$ 347
319.205	$C_{23}H_{27}O^{+}$	Loss of 3 $\times$ H <sub>2</sub> O and CO; Loss of H <sub>2</sub> O from $m/z$ 337
301.195	$C_{23}H_{25}^{+}$	Loss of $4 \times H_2O$ and CO; Loss of $H_2O$ from $m/z$ 319
287.193		No matches within 5 mDa
269.191	$C_{19}H_{25}O^{+}$	Loss of $C_5H_8O_4$ ; Loss of $C_5H_4O_2$ from $m/z$ 365
253.196	$C_{19}H_{25}^{^{+}}$	Loss of $C_4H_4O_2$ from $m/z$ 337
251.178	$C_{19}H_{23}^{+}$	Loss of $H_2O$ from $m/z$ 269
241.159	$C_{17}H_{21}O^{+}$	Loss of $C_2H_4$ from $m/z$ 269
213.164	$C_{16}H_{21}^{+}$	Loss of CO from $m/z$ 241

**Figure S9.** (A) COSY-NMR spectrum of the bioactive compound with m/z 401.235, eluting at 58.0 min. (B) TOCSY-NMR spectrum of the bioactive compound with m/z 401.235, eluting at 58.0 min.





**Figure S10.** (A) MS<sup>2</sup> spectrum of the bioactive compound with m/z 713.412, eluting at 62.0 min. (B) Proposed structure and proposed fragmentation scheme of the bioactive compound with m/z 713.412, eluting at 62.0 min. The minor fragments can be considered as secondary fragments of the ion with m/z 331 and are consistent with subsequent losses of H<sub>2</sub>O, CO and/or NH<sub>3</sub>, e.g., loss of water to the ion with m/z 313, loss of NH<sub>3</sub> to m/z 296, loss of H<sub>2</sub>O to m/z 278, and either loss of H<sub>2</sub>O to m/z 260 or CO to m/z 250. The ion with m/z 253 is consistent with the loss of HN=C=CH and 2 × H<sub>2</sub>O from the fragment ion with m/z 331.

