



Article Polysaccharides, Total Phenolic, and Flavonoid Content from Different Kenaf (*Hibiscus cannabinus* L.) Genotypes and Their Antioxidants and Antibacterial Properties

Ziggiju Mesenbet Birhanie⁺, Aiping Xiao⁺, Dawei Yang, Siqi Huang, Chao Zhang, Lining Zhao, Liangliang Liu⁰, Jianjun Li, Anguo Chen, Huijuan Tang, Li Chang, Gen Pan, Cuiping Zhang, Ashok Biswas, Susmita Dey, Defang Li * and Yong Deng *¹⁰

Institute of Bast Fiber Crops, Chinese Academy of Agricultural Sciences, Changsha 410000, China; zegje23@gmail.com (Z.M.B.); xiaoaiping@caas.cn (A.X.); 18317826602@163.com (D.Y.); huangsiqi@caas.cn (S.H.); zhangchao_20180604@163.com (C.Z.); csbtzln@163.com (L.Z.); liuliangliang@caas.cn (L.L.); lijianjun006@163.com (J.L.); CAGIBFC@126.com (A.C.); tanghuijuan@caas.cn (H.T.); changli519@163.com (L.C.); pangen@caas.cn (G.P.); hncszcp@126.com (C.Z.); ashok.ag1sau@gmail.com (A.B.); susmita.ag4sau@gmail.com (S.D.)

- * Correspondence: lidefang@caas.cn (D.L.); dengyong@caas.cn (Y.D.); Tel.: +86-13873129468 (D.L.); +86-13808498640 (Y.D.)
- + These authors contributed equally to this work.

Abstract: Kenaf (*Hibiscus cannabinus* L.) is a valuable plant with a potential health benefit because of its extensive bioactive compounds. Leaf extracts of 33 kenaf genotypes were investigated for their polysaccharide, total phenolic, and flavonoid content. The antioxidant properties were evaluated by 2,2-Diphenyl-1-picrylhydrazyl (DPPH), 2,2'-azinobis (3-ethylbenzothiazoline-6-sulfonic acid (ABTS), and ferric reducing antioxidant potential (FRAP) assays. Antimicrobial capacity was also assessed against *Staphylococcus aureus* and *Escherichia coli* using a disc diffusion assay. The polysaccharide content varied from 6.45–16.12 mg glucose per g DW. Total phenolic and flavonoid content ranged from 6.03–21.15 mg GAE/g DW and 1.55–9.24 mg RE/g DW, respectively. Similarly, varied values in the range 20.55–79.99% of inhibition by DPPH, 56.28–88.30% of inhibition by ABTS and 1.26–5.08 mmol Fe²⁺/g DW by FRAP assays were obtained for antioxidants of the genotype extracts. Extracts from CS4 and CS2 genotypes had the highest antioxidant activities. Kenaf leaves exhibited antibacterial activity against *Staphylococcus aureus* and *Escherichia coli*. Strong correlation was found between antioxidant activity with polysaccharide (DPPH, r = 0.893; ABTS, r = 0.819; FRAP, r = 0.864) and total phenolic content (DPPH, r = 0.850; ABTS, r = 0.959; FRAP, r = 0.953). The results suggested that the kenaf leaves could be used as a natural antioxidants and antimicrobial in food industries.

Keywords: *Hibiscus cannabinus;* antioxidant activity; antibacterial activity; polysaccharide; phenolic content; flavonoid content

1. Introduction

Even with the improvement in living standards, large numbers of people suffer from non-infectious diseases associated with free radicals which could be controlled by antioxidants [1]. Antioxidants play a significant role in preventing degenerative diseases, including cancer, cardiovascular, and neurological triggered by free radicals generated during cellular metabolism [2–4].

Plant bioactive constituents have been a major area of interest to scientists in recent years due to their great potential in human health. Phenolic and flavonoid compounds are part of the plant bioactive substances exhibiting antioxidant and antibacterial properties [5–7] and are widely distributed in vegetables, fruits, and herbaceous medicinal plants [8]. Similarly, polysaccharides from plants have been proven to possess a



Citation: Birhanie, Z.M.; Xiao, A.; Yang, D.; Huang, S.; Zhang, C.; Zhao, L.; Liu, L.; Li, J.; Chen, A.; Tang, H.; et al. Polysaccharides, Total Phenolic, and Flavonoid Content from Different Kenaf (*Hibiscus cannabinus* L.) Genotypes and Their Antioxidants and Antibacterial Properties. *Plants* **2021**, *10*, 1900. https://doi.org/10.3390/ plants10091900

Academic Editor: Davide Barreca

Received: 27 July 2021 Accepted: 26 August 2021 Published: 14 September 2021

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Copyright: © 2021 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). good biological activity, such as immunity regulation, antitumor activity [9], antioxidant activity [10], and antibacterial activity [11].

Kenaf (Hibiscus cannabinus L.) is a herbaceous, dicotyledonous plant originated from Africa with a wide distribution encouraged by climate adaptation [12]. It has been cultivated for years for textiles, paper, and cordage. Kenaf leaves have many important medicinal properties, including antioxidants, anti-inflammatory, anticancer, aphrodisiacs, analgesic, and hepatoprotective activities [13–15]. Kenaf leaf is used in traditional African medicine to treat anemia and Guinea worms disease [16]. Additionally, in ayurvedic medicine, researchers have found kenaf leaf helpful against fatigue, blood, bilious, diabetes, and throat diseases [13,17]. There is also evidence that kenaf extract is one of the keys in healing scurvy and jaundice, exciting the stomach and improving its action [17]. It is also used in the pharmaceuticals and food industry as a value-added ingredient [18–20]. The kenaf leaves can be used to make healthy beverages as well as functional additives for bakery products [21]. Additionally, tea prepared from kenaf leaves has been recognized as a supplement to the human diet [18]. These medicinal benefits are revealed due to the presence of polysaccharide, flavonoid [13,22], and phenolic [14,17] compounds with antioxidant and antibacterial activities in the kenaf plant [8,12]. A previous study reported that the polysaccharides extracted from kenaf seed have a cholesterol lowering effect in rats [23].

To the best of our knowledge, no information has been published about polysaccharides from kenaf leaves and a few scientific reports have been found on the variation of total phenolic, flavonoid content, antioxidant, and antibacterial activities of leaves' extracts from different kenaf genotypes. This research was therefore aimed at determining the polysaccharides, total phenolic, and flavonoid contents in the leaves of 33 kenaf genotypes and evaluate their antioxidant and antibacterial properties.

2. Results and Discussion

2.1. Polysaccharide Content of Kenaf Leaf Extracts

The polysaccharide content varied from 6.45–16.12 mg glucose/g DW as showed in Table 1. CS4 (16.12 mg glucose/g DW) had the highest polysaccharide contents, followed by CS2 (16.12 mg glucose/g DW) and CS33 (14.57 mg glucose/g DW); however, CS18 had the lowest polysaccharide content among the tested kenaf genotypes.

Polysaccharides are abundant in plants as bioactive substances, but little is known about the polysaccharide content in kenaf. When comparing values in this study with other medicinal plants, kenaf leaves have a comparable amount of polysaccharide. As reported in a previous study, the polysaccharide content in 49 edible macro-fungi ranging from 2.440 to 14.145 mg glucose/g [24] and 8.13–22.56 mg/g dry weight g in *Callerya speciosa* Champ [25]. To the best of our knowledge, this report is the first to determine the polysaccharide content in kenaf leaf. The results showed that kenaf has a high protentional for use in the food and pharmaceutical industries.

2.2. Total Phenolic Content of Kenaf Leaf Extracts

Total phenolic content of each kenaf genotype were determined, and the results are summarized in (Table 1). Significant differences in the range of 6.03–21.15 mg GAE/g DW were observed among the genotypes. The highest total phenolic content (21.15 mg GAE/g DW) was found in CS4 extract followed by CS2 extract (19.40 mg GAE/g DW), CS27 extract (16.81 mg GAE/g DW), CS33 extract (16.63 mg GAE/g DW), and CS16 extract (16.34 mg GAE/g DW). The lowest total phenolic content (6.03 mg GAE/g DW) was obtained in the CS18 extract.

As Ryu et al. reported, total phenolic content of three kenaf varieties ranged from 17.4–28 mg GAE/g dry weight [26]. In the other study, the total phenolic content varied from 19 to 26 mg GAE/g dry weight for two varieties [17]. Similarly, Zhen et al. reported that total phenolic content in the leaves of 22 roselle (*H. sabdariffa*) accessions varied from 19.0–30.0 mg GAE /g depending on the cultivar [27]. The total phenol content obtained

from this study (6.03–21.15 mg GAE/g DW) is comparable to the results found in extracts of 56 commonly consumed vegetables (4.99–23.27 mg GAE/g) [28]. When the results of this study are compared to those of known medicinal plants, there is comparable variation with the reported total phenolic content in the leaves of 19 mulberry varieties, with values ranging from 8.76–20.26 mg RE/g dry weight [29].

Table 1. Polysaccharide, total phenolic, and flavonoid content of leaf extracts from 33 kenaf genotypes (mean value \pm standard deviation of three replicates).

| Genotypes | Polysaccharide Content (mg Glucose/g DW) | Total Phenolic Content (mg GAE/g DW) | Flavonoid Content (mg RE/g DW) |
|-----------|---|---|------------------------------------|
| CS1 | $12.12\pm0.41\mathrm{k}$ | $10.91\pm0.46 \mathrm{ki}$ | 4.03 ± 0.49 ki |
| CS2 | $15.08\pm0.05\mathrm{b}$ | $19.40\pm0.73\mathrm{b}$ | $5.99 \pm 0.34 \mathrm{ef}$ |
| CS3 | 14.58 ±0.03c | $14.24\pm1.84\mathrm{de}$ | $7.08 \pm 0.62 \mathrm{c}$ |
| CS4 | $16.12\pm0.07\mathrm{a}$ | $21.15\pm0.98a$ | 4.74 ± 0.45 h–j |
| CS5 | 12.63 ± 0.01 j | 11.70 ± 1.02 jk | $6.24\pm0.68~\mathrm{de}$ |
| CS6 | $10.74\pm0.01{\rm o}$ | 10.72 ± 0.21 ki | 3.32 ± 0.27 mn |
| CS7 | 11.03 ± 0.06 n | $9.05\pm0.10\mathrm{m}$ | $5.92\pm0.13\mathrm{e}\mathrm{-g}$ |
| CS8 | $12.91 \pm 0.03i$ | $13.46 \pm 0.15e-h$ | $5.96 \pm 0.21e-g$ |
| CS9 | $14.12\pm0.06\mathrm{de}$ | $13.99\pm0.15\mathrm{de}$ | 2.87 ± 0.13 no |
| CS10 | 11.58 ± 0.061 | $13.52 \pm 0.42e-g$ | 2.00 ± 0.20 pq |
| CS11 | 12.48 ± 0.11 j | $12.83\pm0.27\mathrm{f}\text{-i}$ | 2.35 ± 0.13 op |
| CS12 | $11.26\pm0.04\mathrm{m}$ | $12.80\pm0.60\mathrm{f}{\text{-i}}$ | 3.32 ± 0.19 mn |
| CS13 | $10.88\pm0.05 \mathrm{no}$ | 8.55 ± 0.75 nm | 3.47 ± 0.13 l-n |
| CS14 | $11.05\pm0.01\mathrm{n}$ | 7.70 ± 0.67 no | 4.37 ± 0.23 jk |
| CS15 | $10.78\pm0.02\mathrm{o}$ | $7.41\pm0.41\mathrm{op}$ | $4.65\pm0.37i$ –k |
| CS16 | 12.65 ± 0.03 j | $16.34\pm0.60\mathrm{c}$ | $5.88\pm0.64\mathrm{e}	ext{-g}$ |
| CS17 | $13.55\pm0.01\mathrm{g}$ | 11.00 ± 0.33 ki | $8.81\pm0.56a$ |
| CS18 | $6.45\pm0.02r$ | $6.03\pm0.10\mathrm{q}$ | $5.96\pm0.32\mathrm{e}\mathrm{-g}$ |
| CS19 | $13.39\pm0.05{ m gh}$ | 12.40 ± 0.56 ij | $9.24\pm0.59a$ |
| CS20 | 12.60 ± 0.06 j | 12.60 ± 0.30 g–j | $7.06 \pm 0.19c$ |
| CS21 | $9.79\pm0.09\mathrm{p}$ | 6.64 ± 0.44 pq | $3.25\pm0.10mn$ |
| CS22 | $5.16\pm0.02\mathrm{s}$ | 6.45 ± 0.46 pq | $1.55\pm0.10q$ |
| CS23 | $13.87\pm0.02\mathrm{f}$ | $10.12\pm0.17\mathrm{i}$ | 4.37 ± 0.06 jk |
| CS24 | $13.34\pm0.02h$ | 12.52 ± 1.02 h-j | 3.27 ± 0.11 mn |
| CS25 | $13.02\pm0.14\mathrm{i}$ | 12.06 ± 0.51 ij | 3.55 ± 0.13 lm |
| CS26 | $14.00 \pm 0.05 \text{ef}$ | $12.49\pm0.33h	ext{-j}$ | 5.56 ± 0.04 fg |
| CS27 | $14.57\pm0.20\mathrm{c}$ | $16.81 \pm 1.47 \mathrm{c}$ | 4.52 ± 0.26 jk |
| CS28 | $12.17\pm0.01\mathrm{k}$ | $10.12 \pm 0.79i$ | 6.72 ± 0.46 cd |
| CS29 | $9.19\pm0.20\mathrm{q}$ | 8.02 ± 0.23 no | $4.03 \pm 1.78 kl$ |
| CS30 | $14.20\pm0.02\mathrm{d}$ | $14.54\pm0.40\mathrm{d}$ | 5.30 ± 0.38 ghi |
| CS31 | 12.52 ± 0.01 j | 13.87 ± 0.24 de | 5.36 ± 050 fgh |
| CS32 | $13.02 \pm 0.14i$ | $13.59\pm0.63df$ | 4.09 ± 0.48 j–l |
| CS33 | 14.57±0.20c | $16.63\pm0.55c$ | $7.82\pm0.15b$ |

GAE: Gallic acid equivalent; RE: rutin equivalent. Means followed by a different letter within a column indicates a significant difference (p < 0.05) in Duncan's multiple range test.

The total phenolic content is greatly influenced by environmental, climatic environments, and genetic factors [30–32]. In this research, the genotypes were grown under similar conditions and the samples were collected at the same time, so any differences in the total phenolic content are likely to be genotype dependent. This result showed that genotype plays an important role in the total phenolic content of kenaf leaves.

2.3. Total Flavonoid Content of Kenaf Leaf Extracts

Total flavonoid content for the 33 kenaf genotypes are presented in Table 1. The values ranged from 1.55–9.24 mg RE/g DW for the genotypes. Extracts from CS19 genotype showed the highest total flavonoid content (9.24 mg RE/g DW) followed by CS17 (8.81 mg RE/g DW), CS33 (7.82 mg RE/g DW), CS3 extract (7.08 mg RE/g DW), CS20 extract (7.06 mg RE/g DW), and CS28 extract (6.72 mg RE/g DW). CS22 genotype extract had the lowest total flavonoid content (1.55 mg RE/g DW).

The results of total flavonoid content of the kenaf genotypes were similar to that of a well-known beverage such as tea (*Camellia sinensis*) (3.45–5.40 RE/g DW) [33]. Although previous study found that genotype had no significant impact on the total flavonoid content [26], statistical differences in the total flavonoid content were detected among genotypes in this study. Variations in total flavonoid contents among these kenaf genotypes enable us to ease selection of genotypes that might be used as medicinal applications.

2.4. Antioxidant Capacities of Kenaf Leaf Extracts

The antioxidant activities of different kenaf genotypes were assessed by DPPH, ABTS, and FRAP assays, and the results are presented in Table 2.

Table 2. Antioxidant activities of leaf extracts from 33 kenaf genotypes (mean value \pm standard deviation of three replicates).

| Genotypes | DPPH Inhibition (%) | ABTS Inhibition (%) | FRAP (mmol Fe ²⁺ /g) |
|-----------|------------------------------|-------------------------------------|---------------------------------|
| CS1 | 65.81 ± 0.64 hi | $69.61\pm0.69\mathrm{i}$ | $3.41\pm0.06\mathrm{s}$ |
| CS2 | $79.77 \pm 1.93a$ | $86.67\pm0.56\mathrm{b}$ | $4.89\pm0.08\mathrm{b}$ |
| CS3 | $71.78\pm0.74\mathrm{e}$ | $80.77\pm0.96ef$ | $4.43\pm0.10\mathrm{f}$ |
| CS4 | $79.77 \pm 1.93a$ | $88.30\pm0.88a$ | $5.08\pm0.13a$ |
| CS5 | $66.35 \pm 1.48 \mathrm{h}$ | 69.29 ± 1.68 ij | $3.48\pm0.02r$ |
| CS6 | 64.22 ± 0.88 jk | $68.72\pm0.60\mathrm{i-l}$ | $3.22\pm0.07t$ |
| CS7 | 60.38 ± 1.151 | $67.98\pm0.88\text{k-n}$ | $2.97\pm0.07v$ |
| CS8 | $69.76 \pm 1.30 \mathrm{fg}$ | 68.21 ± 1.00 j $-$ m | $3.98\pm0.02k$ |
| CS9 | $75.01\pm0.65\mathrm{c}$ | 67.95 ± 1.73 k–n | $4.40\pm0.08{ m g}$ |
| CS10 | $69.76\pm0.89\mathrm{fg}$ | 67.77 ± 1.00 k-n | $4.01\pm0.03 \mathrm{j}$ |
| CS11 | $66.35\pm2.03h$ | 76.52 ± 1.72 g | $3.92\pm0.10\mathrm{l}$ |
| CS12 | 65.50 ± 0.88 h–j | $76.34\pm0.55g$ | $3.88\pm0.02m$ |
| CS13 | $63.37\pm0.96k$ | 67.38 ± 1.04 m $-$ o | $2.94\pm0.02\mathrm{v}$ |
| CS14 | $46.22\pm0.80\mathrm{mn}$ | $66.99\pm0.74 no$ | $2.29\pm0.02x$ |
| CS15 | $45.05\pm1.19\mathrm{n}$ | $65.63 \pm 0.75 \mathrm{p}$ | $2.22\pm0.02\mathrm{y}$ |
| CS16 | $73.27 \pm 0.96d$ | $84.17 \pm 1.00 d$ | $4.74\pm0.14d$ |
| CS17 | $68.48\pm0.67\mathrm{g}$ | $68.96\pm0.78\mathrm{i}\mathrm{-k}$ | $3.42\pm0.02\mathrm{s}$ |
| CS18 | $20.55\pm1.33\mathrm{p}$ | $56.28\pm0.51s$ | $1.26\pm0.01z$ |
| CS19 | 65.39 ± 0.92 h $-$ j | $74.46\pm0.76h$ | $3.58\pm0.10\mathrm{p}$ |
| CS20 | $66.77\pm0.32h$ | $76.04 \pm 1.34 \mathrm{g}$ | 3.87 ± 0.03 mn |
| CS21 | $28.86\pm0.49\mathrm{o}$ | $63.81\pm0.29q$ | $2.23\pm0.02y$ |
| CS22 | $28.75\pm0.64\mathrm{o}$ | $58.78\pm0.61\mathrm{r}$ | $2.21\pm0.02y$ |
| CS23 | $64.54\pm0.55\mathrm{i-k}$ | 67.68 ± 0.60 l-n | $2.97\pm0.02\mathrm{v}$ |
| CS24 | $65.81\pm0.68 \mathrm{hi}$ | $74.82\pm0.60\mathrm{h}$ | $3.85\pm0.11n$ |
| CS25 | 68.90 ± 1.18 g | $74.43\pm0.56h$ | $3.59\pm0.02q$ |
| CS26 | $69.12 \pm 1.29 \mathrm{g}$ | $74.70 \pm 0.68 h$ | 3.69 ± 0.060 |
| CS27 | $79.31\pm0.76\mathrm{ab}$ | $85.39 \pm 1.18 \mathrm{c}$ | $4.80\pm0.08\mathrm{c}$ |
| CS28 | 64.86 ± 0.68 ij | $68.42\pm0.83\mathrm{i-m}$ | 3.17 ± 0.04 u |
| CS29 | $47.60 \pm 0.55 m$ | $66.31 \pm 1.78 \text{op}$ | $2.47\pm0.02w$ |
| CS30 | $75.83\pm0.74\mathrm{c}$ | $81.73 \pm 1.27 \mathrm{e}$ | $4.67\pm0.07\mathrm{e}$ |
| CS31 | $70.93 \pm 1.56 \mathrm{ef}$ | $80.60\pm0.67 ef$ | $4.29\pm0.06h$ |
| CS32 | $68.58 \pm 1.79 \mathrm{g}$ | $80.18 \pm 1.32 g$ | $4.13\pm0.04\mathrm{i}$ |
| CS33 | $78.27 \pm 1.69 \mathrm{b}$ | $85.09\pm0.51cd$ | $4.74\pm0.07d$ |

DPPH: 2,2-Diphenyl-1-picrylhydrazyl; ABTS: 2,2'-azino-bis (3-ethylbenzothiazoline-6-sulfonic acid; FRAP: Ferric Reducing Antioxidant Potential. Means followed by the different letters within a column indicates significant difference (p < 0.05) in Duncan's multiple range test.

DPPH radical scavenging activities of the extracts ranged from 20.55–79.77%, which reflects about 4-fold variation. CS2 and CS4 extracts exhibited the highest antioxidant activity with each having 79.77% of DPPH inhibition, followed by CS27 extract (79.31%), CS33 extract (78.27%), CS30 extract (75.83%), and CS9 extract (75.0%). CS18 extract exhibited the lowest antioxidant activity (20.55%). Previous study reported that the DPPH radical scavenging activity of kenaf leaves' extracts exhibited about 65.35% of inhibition [34]. Among the 33 kenaf genotypes analyzed, 22 exhibited higher DPPH scavenging activity than the scavenging capacity of *H. sabdariffa* leaf extract reported in a previous study with a

value of 65.19% [35]. Although DPPH radical scavenging activity of plant extracts depend on the solvent [36] and cultivar [37], phenolic compounds were generally considered as the source of scavenging activities of the extracts because of their hydroxyl group [38,39]. The variation in the radical scavenging activities obtained in this study can only be attributed to genotypes as the same solvent and procedure were followed in the extraction process.

The ABTS assay values varied from 56.28–88.30%, representing a lower difference than the value of DPPH assay about 2-fold. CS4 extract had the highest antioxidant activity (88.30% ABTS inhibition) followed by the CS2 extract (86.67%), CS27 extract (85.39%), CS33 (85.09), and CS16 extract (84.17%). As noted in the DPPH assay, the CS18 extract exhibited the lowest antioxidant activity (56.28%). The DPPH and ABTS assays exhibited similar results for the antioxidant activities determined in extracts of 33 kenaf genotypes. In this analysis, the values of ABTS radical inhibitions are higher than the value stated by previous study for unspecified kenaf variety [40], which had inhibition value of 25.96%. The antioxidant activity of kenaf leaf extract by the ABTS radical scavenging activity assay was also studied by other study [22] and reported an inhibition of 65.55%. The total phenolic contents may be correlated with these variations.

In FRAP assay, the pattern for reducing capacities of the 33 genotype extracts examined did not differ appreciably from their scavenging activities for DPPH and ABTS. Comparable to the results found for DPPH and ABTS assays, CS4, CS2, CS27, and CS33 extracts exhibited strong ferric-reducing capacities (5.08, 4.89, 4.80, and 4.74 mmol Fe²⁺/g DW respectively). CS18 extract had the lowest ferric reducing capacity (1.26 mmol Fe²⁺/g DW) in this study. In comparison to tea, which is a well-known source of antioxidants, kenaf extracts had higher ferric reducing capacity than white, green, and black teas [41]. In this study, the values of ferric reducing capacity of 27 genotypes are higher than the values stated by a previous study for eggplant (Purple) and marrow squash [42]. Previous studies have shown that phenolic compounds play a significant role in the reduction capacity of plant extracts [43,44]. Thus, kenaf genotypes with high levels of phenolic contents might show greater ferric reducing capacity.

2.5. Antibacterial Activities of Kenaf Leaf Extracts

The antibacterial activity of each extract was tested against Gram- positive (*Staphylococcus aureus*) and Gram negative (*Escherichia coli*) bacteria. As shown in Table 3, the extracts of the 33 kenaf genotypes showed antibacterial activity against two bacteria, the zones of inhibition ranging from 9.40–13.55 mm (*Staphylococcus aureus*) and from 6.15–12.67 mm (*Escherichia coli*). The Gram-negative bacteria exhibited a wider range of levels of antibacterial activity. CS5 extract exhibited the highest bacterial inhibition zone against *Staphylococcus aureus* (13.55 mm), followed by CS3 extracts (12.60 mm). CS2 extract exhibited the maximum zone of inhibition against *Escherichia coli* (12.67 mm), followed by CS3 extract (11.85 mm).

These findings suggest that the antibacterial effects of kenaf leaf extracts vary depending on genotype. Moreover, phenolic, flavonoid, and polysaccharide composition of the extracts are the primary inhibitors of bacteria by the alteration of membrane integrity and permeability [6,34,45,46]. The antibacterial properties of kenaf leaf extract as obtained in this study exhibited great potential as an antibacterial agent for the preservation of food.

2.6. Correlation between Antioxidant Capacities with Polysaccharide, Total Phenolic, and Flavonoid Content

The Pearson's correlation analysis between antioxidant capacities with polysaccharide content, total phenolic content, and total flavonoid content is shown in Table 4. DPPH antioxidant capacities had a significant positive correlation with polysaccharide content (r = 0.893, p < 0.01) and total phenolic content (r = 0.850, p < 0.01). Strong correlation was found between ABTS antioxidant capacities with polysaccharide (r = 0.819, p < 0.01) and total phenolic content (r = 0.959, p < 0.01). Similarly, FRAP reducing capacities had a significance positive correlation with polysaccharide content (r = 0.864, p < 0.01) and total

phenolic content (r = 0.953, p < 0.01). In contrast, DPPH, ABTS, and FRAP antioxidant capacities had no correlation with total flavonoid content.

Table 3. Antibacterial activity of leaf extracts from 33 kenaf genotypes against Gram positive (*Staphylococcus aureus*) and Gram-negative bacteria (*Escherichia coli*). The values are presented as mean inhibition zone (mm) \pm standard deviation of three replicates including diameter of disk (6 mm).

| | Zone of Inhibition (mm) | | |
|----------|---------------------------|-------------------------------------|--|
| Genotype | Bacterial Strains | | |
| | Staphylococcus aureus | Escherichia coli | |
| CS1 | $11.42\pm0.04\mathrm{d}$ | 10.06 ± 0.05 jk | |
| CS2 | $12.05\pm0.06\mathrm{c}$ | $12.67\pm0.06a$ | |
| CS3 | $12.60\pm0.14\mathrm{b}$ | $11.85\pm0.13\mathrm{b}$ | |
| CS4 | $12.43\pm0.13b$ | $11.18\pm0.11\text{c-e}$ | |
| CS5 | $13.55\pm0.12a$ | $6.85\pm0.07\mathrm{o}$ | |
| CS6 | $12.42\pm0.17\mathrm{b}$ | $10.20\pm0.12\mathrm{i}\mathrm{-k}$ | |
| CS7 | $10.54\pm0.10{ m gh}$ | $11.04\pm0.61\mathrm{d}\mathrm{-g}$ | |
| CS8 | 9.05 ± 0.04 lm | $11.47\pm0.54\mathrm{b-d}$ | |
| CS9 | 11.01 ± 0.49 ef | $11.32\pm0.21b-e$ | |
| CS10 | $10.04\pm0.19\mathrm{i}$ | $7.15\pm0.28\mathrm{o}$ | |
| CS11 | 9.28 ± 0.421 | $6.63\pm0.05 \mathrm{op}$ | |
| CS12 | 11.09 ± 0.10 de | 9.19 ± 0.20 mn | |
| CS13 | $12.05\pm0.33\mathrm{c}$ | 8.89 ± 0.05 mn | |
| CS14 | $10.54\pm0.10{ m gh}$ | 8.74 ± 0.94 n | |
| CS15 | 9.62 ± 0.22 jk | $9.77\pm0.26\mathrm{ki}$ | |
| CS16 | 10.68 ± 0.30 fg | $6.84\pm0.64\mathrm{o}$ | |
| CS17 | 10.21 ± 0.13 hi | 11.22 ± 0.02 c-e | |
| CS18 | 9.40 ± 0.16 j $-$ l | $6.15\pm0.20\mathrm{p}$ | |
| CS19 | 10.54 ± 0.04 gh | $10.54 \pm 0.14 { m f}$ –j | |
| CS20 | $10.60\pm0.36 m{g}$ | 8.74 ± 0.16 mn | |
| CS21 | 8.60 ± 0.24 n | 11.22 ± 0.27 c-e | |
| CS22 | $7.28\pm0.05\mathrm{o}$ | 6.61 ± 0.18 op | |
| CS23 | $8.78\pm0.27\mathrm{mn}$ | $10.76 \pm 0.30 e^{-h}$ | |
| CS24 | 8.73 ± 0.11 mn | 10.49 ± 0.15 g–j | |
| CS25 | 9.31 ± 0.08 kl | $11.15 \pm 0.07 c-f$ | |
| CS26 | 10.39 ± 0.08 g–i | 11.12 ± 0.15 d-f | |
| CS27 | 11.27 ± 0.17 de | 10.39 ± 0.15 h $-$ j | |
| CS28 | 11.3 ± 0.24 de | $11.77\pm0.37\mathrm{bc}$ | |
| CS29 | 11.09 ± 0.32 de | $10.99\pm0.78\mathrm{d}	extrm{-g}$ | |
| CS30 | 9.70 ± 0.23 j | $10.30\pm0.06\mathrm{i}-\mathrm{k}$ | |
| CS31 | 9.34 ± 0.14 j $-$ l | 10.56 ± 0.22 f-j | |
| CS32 | 11.36 ± 0.18 de | $9.36\pm0.52 \mathrm{lm}$ | |
| CS33 | $10.66\pm0.27\mathrm{fg}$ | $11.59\pm0.04bd$ | |

Means followed by the different letter within a column indicate significant difference (p < 0.05) in Duncan's multiple range test.

Table 4. Pearson's correlation coefficients between polysaccharide, total phenolic content, flavonoid content, and antioxidant capacities.

| Antioxidant Activities | Polysaccharide Content | Total Phenolic Content | Total Flavonoid Content |
|---------------------------|---------------------------|---------------------------|----------------------------|
| DPPH | 0.893 ** | 0.850 ** | 0.253 |
| ABTS | 0.819 ** | 0.959 ** | 0.164 |
| FRAP | 0.864 ** | 0.953 ** | 0.215 |

** Correlation significant at p < 0.01.

According to literature, polysaccharides can significantly improve antioxidant activity and prolonged exercise performance [47]. The strong correlation between polysaccharide content and antioxidant capacity has been previously reported for other antioxidants [24,48].

7 of 13

In recent studies, the antioxidant capacities of kenaf leaf tea determined by DPPH and ABTS assays exhibited strong correlations with phenolic content [22,40]. Likewise, the strong correlation between FRAP assay and total phenolic of other antioxidants has been reported [44]. The number and arrangement of hydroxyl groups, as well as the existence of electron donating and electron withdrawing substituents in the ring structure of phenolics, decide their antioxidant capacity [49,50]. These results showed that polysaccharides and total phenolic content play an important role in the measured antioxidant properties of kenaf leaves, although the degree of correlation suggests that other factors could also contribute. The genotypes with the highest polysaccharide and total phenolic content also had the highest antioxidant activities determined by DPPH, ABTS, and FRAP assays.

2.7. Hierarchical Cluster Analysis

The hierarchical cluster analysis clustered the 33 genotypes on the basis of their bioactive compounds and antioxidant capacities' similarities. The samples were clustered into five major groups using Ward's method without taking into consideration the detail about the class of genotype. The results are presented in Figure 1. Cluster I contained 24.24% of the total kenaf genotypes analyzed and was characterized by high content of flavonoids. Cluster II included 45.45% of the genotypes with intermediate antioxidant activities (DPPH, ABTS, and FRAP). Cluster III accounted for 12.12% of the genotypes studied. This cluster was characterized by the highest content of total phenolic, polysaccharide, with greater antioxidant and antibacterial activities. Cluster IV accounted for 9.09% of the genotypes with intermediate content of flavonoid and low content of FRAP reducing capacity. Cluster V also accounted for 9.09% of the genotypes with the lowest content of flavonoid, phenolic, and antioxidant capacities. The analysis indicated that genotypes from cluster III could serve as potential sources of natural antioxidants.



Figure 1. Dendrogram of hierarchical cluster analysis of the 33 kenaf genotypes according to their polysaccharide content, total phenolic content, total flavonoid content, antioxidant, and antibacterial activities using the Ward and Euclidean distance.

3. Materials and Methods

3.1. Chemicals and Reagents

The 2,2'-azinobis (3-ethylbenzothiazoline-6-sulfonic acid (ABTS), 2,2-Diphenyl-1picrylhydrazyl (DPPH), gallic acid, and rutin were obtained from Solar Bio-Science and Technology Co., Ltd. (Beijing, China). In addition, 2,4,6-tripyridyl-s-triazine (TPTZ), the regent of Folin–Ciocalteau, was purchased from Coolaber Science and Technology (Beijing, China). The highest analytical grade was used for all reagents in the analysis.

3.2. Plant Materials

Leaf samples from 33 kenaf genotypes were collected from Innovation experimental base of Institute of Bast Fiber Crops, Chinese Academy of Agricultural Sciences, Changsha, China, on 8 September 2020. The names of the 33 kenaf genotypes are shown in Table 5. The leaves were air dried and grinded using a small crusher (HX-200 A, Yongkang City, Xian Hardware and Pharmacy Co., Ltd., Xi'an China) and sieved using a 60- mesh sieve, labeled, and stored at room temperature for further use.

| Genotypes Code | Cultivar/Variety Name | |
|----------------|-----------------------|--|
| CS1 | 19Sanya357 | |
| CS2 | Fu Hong992 | |
| CS3 | 18Sanya538 | |
| CS4 | 18Sanya536 | |
| CS5 | 19Sanya599 | |
| CS6 | 19Sanya355 | |
| CS7 | 19Sanya598 | |
| CS8 | WJZ1903-1 | |
| CS9 | WJZ1903-2 | |
| CS10 | WJZ1903-3 | |
| CS11 | WJZ1903-4 | |
| CS12 | WJZ-1 | |
| CS13 | 19WJZ-2 | |
| CS14 | 19WJZ-3 | |
| CS15 | 19WJZ-4 | |
| CS16 | WMS1803 | |
| CS17 | WMS1903 | |
| CS18 | 19BiJiA | |
| CS19 | 19BiJiB | |
| CS20 | 19MLM-1 | |
| CS21 | 19MLM-2 | |
| CS22 | 19MLM-3 | |
| CS23 | 19MLM-4 | |
| CS24 | China Kenaf18 | |
| CS25 | H1701 | |
| CS26 | Sanya362 | |
| CS27 | Sanya355 | |
| CS28 | Sanya354 | |
| CS29 | Sanya360 | |
| CS30 | Sanya348 | |
| CS31 | Sanya361 | |
| CS32 | 19Xin-2 | |
| CS33 | 19XJ-1-2 | |

Table 5. Name of the 33 kenaf genotypes used in this study.

3.3. Preparation of Leaf Extracts

Five grams of each sample were extracted with 90% ethanol (10% w/v) by the reflux method for 2 h at 85 °C. The extracted samples were centrifuged at $4000 \times g$ for 15 min. The supernatants were moved to 50 mL volumetric flasks, and the procedure was repeated. Extract samples stored at -20 °C for analysis.

3.4. Determination of Polysaccrides

The polysaccharide content was measured following the phenol-sulfuric acid colorimetric method [51]. The sample solution was diluted 10 times with distilled water and 2 mL from which it was mixed with 1 mL of 5% phenol (w/v) and 5 mL of concentrated

ubated in a boiling water bath for 15 min. The

sulfuric acid, shaken and subsequently incubated in a boiling water bath for 15 min. The mixture was allowed to cool down to room temperature; the absorbance was then recorded at 490 nm. The standard curve was prepared with glucose (10–80 mg/L), and the results expressed as mg glucose per grams dry weight of the sample (mg glucose/g DW).

3.5. Determination of Total Phenolic Content

The total phenolic content was determined following the Folin–Ciocalteau colorimetric method [52]. Furthermore, 1.5 mL of Folin–Ciocalteau (20% v/v) reagent was added to 0.2 mL of each sample extract separately, mixed properly and kept for 5 min. Four milliliters of Na₂CO₃ (7%) were then added, and final volume made up to 10 mL with distilled water followed by incubation for 90 min in dark at room temperature. Absorbance was measured by microplate spectrophotometer at 760 nm. All determinations were calculated in triplicate. The standard curve of gallic acid (20–800 mg/L) was prepared and the results expressed as milligrams of gallic acid equivalents per grams of dry weight of the sample (mg GAE/g DW).

3.6. Determination of Total Flavonoid Content

The total flavonoid content of each extract was determined according to the aluminum chloride method [13]. In addition, 0.2 mL ethanolic sample extract was mixed with 4 mL of distilled water in a flask and then 0.3 mL NaNO₂ (5%) were added to the flask and reacted for 5 min. Then, 0.3 mL AlCl₃ (10%) was added and kept for 6 min for reaction. Then, 2 mL NaOH (4%) was added and filled with up to 10 mL with distilled water. The absorbance was measured at 510 nm. A rutin standard curve (50–400 mg/L) was prepared, and the results were reported as milligrams of rutin equivalents per grams of dry weight of the sample (mg RE/g DW). The analysis was executed in triplicate.

3.7. Antioxidant Capacity Determinations

3.7.1. DPPH Free Radical-Scavenging Potential

The free radical-scavenging potential of the extract was determined using the Brand–Williams methods with minor modification [53]. Three milliliters of freshly prepared methanol DPPH solution (6×10^{-5} M) were added to 100 µL of each sample separately and the mixtures incubated in dark for 15 min at room temperature, and then the sample absorption (A sample) was recorded at 515 nm. A blank sample absorbance (A blank) containing 100 µL of methanol was also measured. Vitamin C was used as positive controls and methanol as negative control. The triplicate experiment was executed and their scavenging capacity was determined using the following equation:

inhibition (%) = [(A blank – A sample)/A blank]
$$\times$$
 100 (1)

where A blank = absorbance of the blank, and A sample = absorbance of the sample extract.

3.7.2. ABTS Radical Scavenging Activity

The ABTS radical scavenging potential was determined on the basis of measuring the degree of absorbance reduction of the radical ABTS+• ion by the tested sample extracts [43]. In addition, 10 μ L of sample extract and 20 μ L solution of peroxidase were mixed with 170 μ L solution of ABTS and allowed to react for 6 min in the dark, after which the absorption was measured using a 405 nm UV-visible spectrophotometer (A sample). By mixing 10 μ L of distilled water with 20 μ L solution of peroxidase and 170 μ L solution of ABTS, a blank sample was prepared, and the absorbance of the blank sample was measured (A blank). Vitamin C was used as positive controls and methanol as negative. Determinations were performed in triplicate. The ABTS+• inhibition percentage was determined using the above Equation (1).

3.7.3. Ferric Reducing Antioxidant Potential (FRAP)

Ferric reduction capability of extracts was determined using a modified version of the FRAP assay [54]. By adding 20 mL of acetate buffer (300 mM, pH 3.6,) and 2 mL of TPTZ (10 mM) in 40 mM hydrochloric acid and 2 mL of ferric chloride (20 mM), the working solution of the FRAP reagent was prepared. Then, the mixture was allowed to incubate at 37 °C for 30 min in a water bath. In addition, 180 μ L of FRAP reagent were added to 5 μ L of sample extract, and the sample absorption was estimated at 593 nm. Absorbance of a blank without a sample extract was also measured, and analyses were executed in triplicate. In order to measure the FRAP value, the difference was calculated between the sample absorbance and the blank absorbance. On the day of preparation, all solutions were used. A standard curve of ferrous sulphate (0.15–1.5 mmol/L) was used, and the results were reported in mmol ferrous ion equivalents per gram dry weight of the sample (Fe²⁺/g DW).

3.8. Antibacterial Activities

Antibacterial activity was determined using an agar diffusion method [55]. *Staphylococcus aureus*—A Gram-positive (ATCC 6538) and *Escherichia coli*—Gram-negative (ATCC 8739) microorganisms were used. The strains were cultivated on a Nutrient Broth medium for 24 h at 30 °C. In addition, 15 mL of nutrient agar medium were dispensed into petri dishes in a laminar airflow and allowed to solidify; then, 100 μ L of bacterial suspension (10⁷ CFU/mL) were spread on the surface of the agar plates, and sterile filter paper disk (6 mm diameter) was placed on the agar plate. Furthermore, 50 mg/mL of each extract were loaded on the discs and incubated for 24 h at 37 °C. The diameter of the inhibition zones was measured and expressed in millimeters. Ethanol (80%) was used as a control [6].

3.9. Statistical Analysis

The values of three measurements were reported as mean \pm standard deviation. Statistical analysis was conducted using SAS[®] 9.4 software (SAS Inc., Cary, NC, USA). Duncan's multiple comparison test was used for means separation and differences considered significant at p < 0.05. Hierarchical cluster analysis was performed using OriginPro[®] version 9 software (Northampton, MA, USA). Correlation analysis was determined using SigmaPlot[®] version 12.5 software (San Jose, CA, USA).

4. Conclusions

In this study, kenaf genotypes with higher phenolic and polysaccharide contents could be a significant source of natural antioxidants. The antibacterial test revealed that extracts of kenaf leaves exhibited antibacterial effects against both Gram-positive and Gram-negative bacteria. Therefore, the kenaf extracts might have the potential to serve as natural antioxidants and bio-preservatives in food industry. The results could be useful for the selection of kenaf genotypes for future genetic engineering purposes to enhance a particular bioactive compound.

Author Contributions: Conceptualization, Y.D. and D.L.; Data curation, S.H., C.Z. (Chao Zhang), L.Z., L.L., J.L., A.C., D.Y. and Z.M.B.; Formal analysis, A.X., A.B., S.D., D.Y. and Z.M.B.; Investigation, Z.M.B.; Methodology, A.X., Y.D., D.L. and Z.M.B.; Supervision, A.X., S.H., Y.D. and D.L; Validation, A.X., C.Z. (Cuiping Zhang), L.Z., L.L., J.L., A.C., H.T., L.C., G.P., C.Z. (Chao Zhang) and C.Z. (Cuiping Zhang). All authors have read and agreed to the published version of the manuscript.

Funding: This work was supported by the Chinese Agriculture Technology Research System (CARS-16) and the Agricultural Science and Technology Innovation Program (ASTIP-IBFC03).

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: Not applicable.

Conflicts of Interest: The authors declare no conflict of interest.

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