

Supporting information

Aliphatic-Bridged Early Lanthanide MOFs: Topological Polymorphism and Excitation-Dependent Luminescence

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Single crystal X-ray diffraction details

Table S1. Single crystal XRD experiment and refinement details for **1**, **2_{Ce}**, **2_{Pr}**.

	1	2_{Ce}	2_{Pr}
Chemical formula	C ₃₀ H ₅₄ La ₂ N ₂ O ₁₉	C _{50.25} H _{51.25} Ce ₂ N _{4.75} O _{12.75}	C _{50.25} H _{51.25} N _{4.75} O _{12.75} Pr ₂
<i>M_r</i> , g/mol	1024.57	1205.95	1207.53
Crystal system	Triclinic	Monoclinic	Monoclinic
Space group	<i>P</i> $\bar{1}$	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> 2 ₁ / <i>n</i>
Temperature, K	170	292	170
<i>a</i> , Å	9.0209(6)	10.5829(3)	10.5217(4)
<i>b</i> , Å	10.5442(7)	17.9877(5)	17.9353(7)
<i>c</i> , Å	11.8081(6)	13.2388(4)	13.0839(5)
α , °	81.253(5)	90	90
β , °	68.994(6)	98.833(3)	99.054(4)
γ , °	72.993(6)	90	90
<i>V</i> , Å ³	1001.27(12)	2490.28(13)	2438.30(16)
<i>Z</i>	1	2	2
<i>F</i> (000)	514	1208	1212
<i>D</i> (calc.), g/cm ³	1.699	1.608	1.645
μ , mm ⁻¹	2.18	1.87	2.04
Crystal size, mm	0.24 × 0.09 × 0.07	0.51 × 0.11 × 0.10	0.35 × 0.10 × 0.07
θ range for data collection, °	1.9 < θ < 25.4	1.9 < θ < 25.3	2.3 < θ < 25.4
No. of reflections:	6947	11900	11095
measured/independent/observed	3667	4560	4467
[<i>I</i> > 2 σ (<i>I</i>)]	3268	4021	3777
<i>R</i> _{int}	0.0278	0.0206	0.0232
Index ranges	−10 < <i>h</i> < 10 −12 < <i>k</i> < 8 −14 < <i>l</i> < 14	−12 < <i>h</i> < 12 −21 < <i>k</i> < 19 −15 < <i>l</i> < 11	−12 < <i>h</i> < 12 −21 < <i>k</i> < 20 −15 < <i>l</i> < 15
Final <i>R</i> indices [<i>I</i> > 2 σ (<i>I</i>)]	0.0331 0.0790	0.0234 0.0545	0.0235 0.0495
<i>R</i> indices (all data)	0.0396 0.0815	0.0285 0.0569	0.0325 0.0521
Goodness-of-fit on <i>F</i> ²	1.050	1.056	1.027
Largest diff. peak/hole, e/Å ³	2.16/−0.89	0.53/−0.37	0.77/−0.45

Table S2. Single crystal XRD experiment and refinement details for **2_{Nd}**, **2_{Sm}**, **3**.

	2_{Nd}	2_{Sm}	3
Chemical formula	C _{50.25} H _{51.25} N _{4.75} Nd ₂ O _{12.75}	C _{50.25} H _{51.25} N _{4.75} O _{12.75} Sm ₂	C ₅₄ H _{61.33} N ₆ Nd ₂ O _{14.67}
<i>M_r</i> , g/mol	1214.19	1226.41	1317.57
Crystal system	Monoclinic	Monoclinic	Trigonal
Space group	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> 2 ₁ / <i>n</i>	<i>R</i> [−] 3
Temperature, K	170	170	100
<i>a</i> , Å	10.5234(4)	10.4751(4)	37.790(5)
<i>b</i> , Å	17.9568(6)	17.8590(6)	37.790(5)
<i>c</i> , Å	13.0405(5)	12.9892(5)	10.303(2)
α , °	90	90	90
β , °	98.916(3)	98.735(3)	90
γ , °	90	90	120
<i>V</i> , Å ³	2434.44(16)	2401.77(15)	12742(4)
<i>Z</i>	2	2	9
<i>F</i> (000)	1216	1224	5982
<i>D</i> (calc.), g/cm ³	1.656	1.696	1.545
μ , mm ^{−1}	2.18	2.49	2.12
Crystal size, mm	0.79 × 0.14 × 0.11	0.33 × 0.11 × 0.09	0.70 × 0.20 × 0.20
θ range for data collection, °	2.0 < θ < 25.4	2.0 < θ < 25.3	2.7 < θ < 26.7
No. of reflections: measured/independent/observed [<i>I</i> > 2 σ (<i>I</i>)]	8388	10347	13340
	4471	4404	5107
	3895	3813	4819
<i>R</i> _{int}	0.0343	0.0251	0.0323
Index ranges	−12 < <i>h</i> < 7	−12 < <i>h</i> < 12	−45 < <i>h</i> < 45
	−21 < <i>k</i> < 13	−21 < <i>k</i> < 17	−25 < <i>k</i> < 45
	−15 < <i>l</i> < 14	−10 < <i>l</i> < 15	−12 < <i>l</i> < 10
Final <i>R</i> indices [<i>I</i> > 2 σ (<i>I</i>)]	0.0317	0.0253	0.0393
	0.0743	0.0551	0.0939
<i>R</i> indices (all data)	0.0382	0.0326	0.0417
	0.0794	0.0575	0.0950
Goodness-of-fit on <i>F</i> ²	1.072	1.051	1.093
Largest diff. peak/hole, e/Å ³	1.09/−1.10	0.68/−0.58	1.19/−1.35

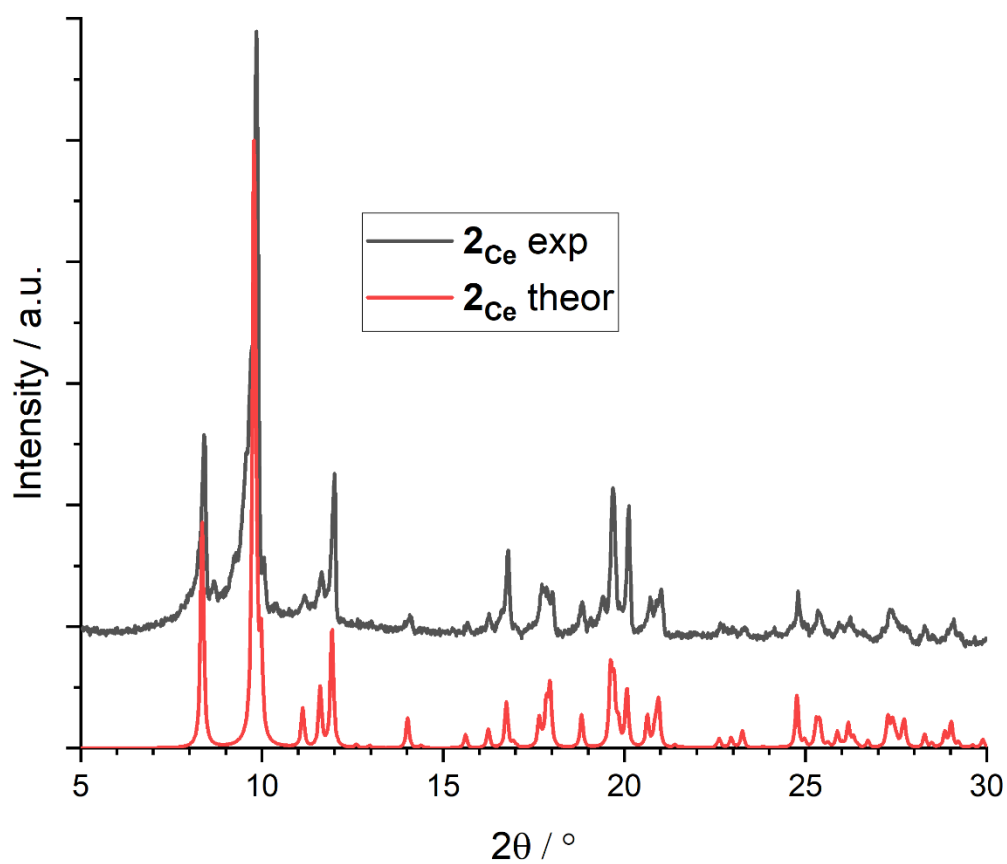


Figure S1. Experimental (black) and theoretical (red) PXRD patterns for **2_{ce}**.

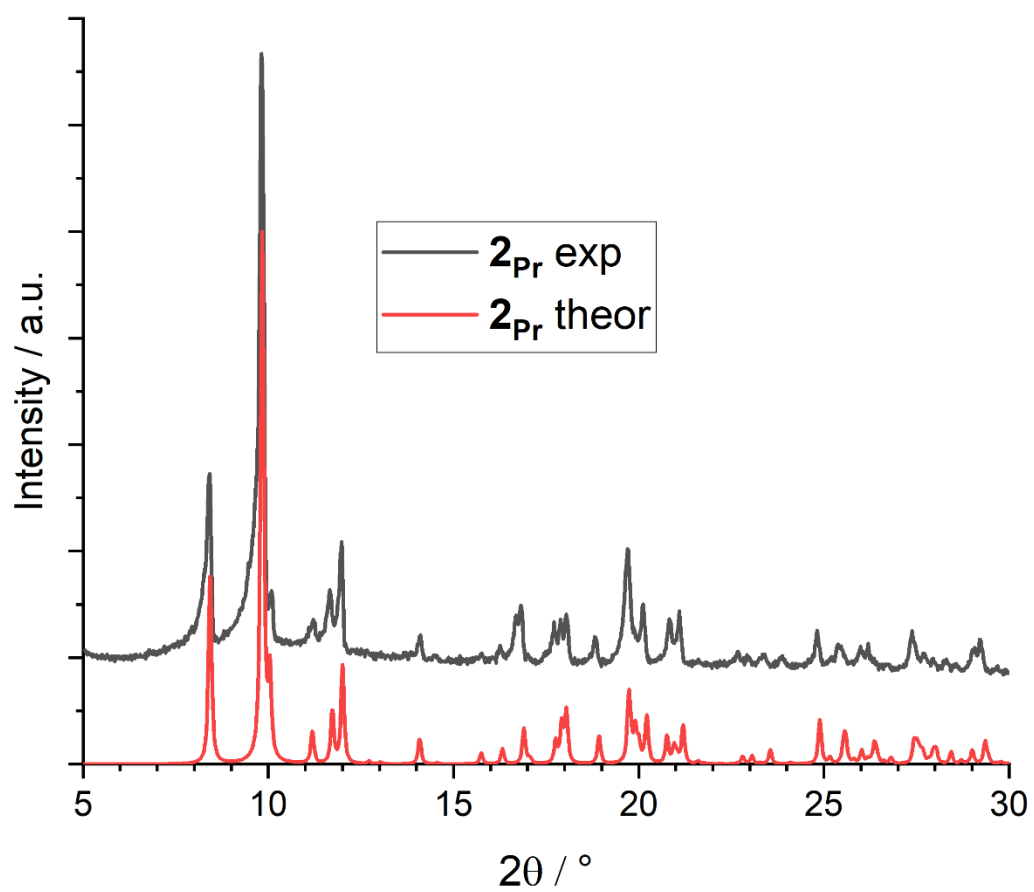


Figure S2. Experimental (black) and theoretical (red) PXRD patterns for 2_{Pr} .

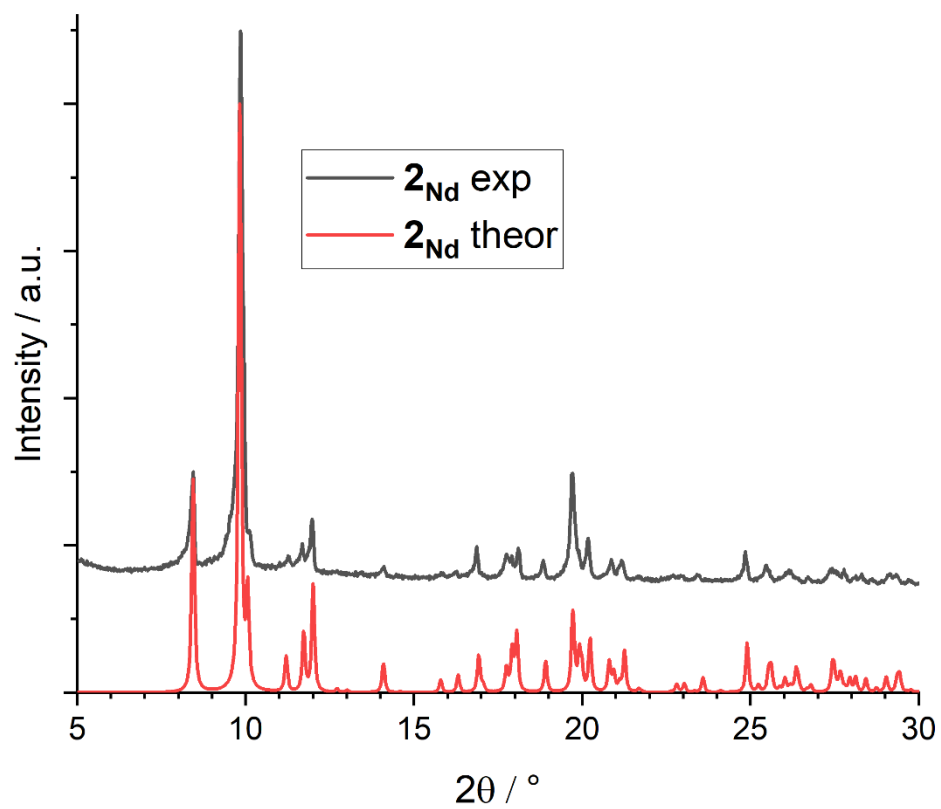


Figure S3. Experimental (black) and theoretical (red) PXRD patterns for 2Nd .

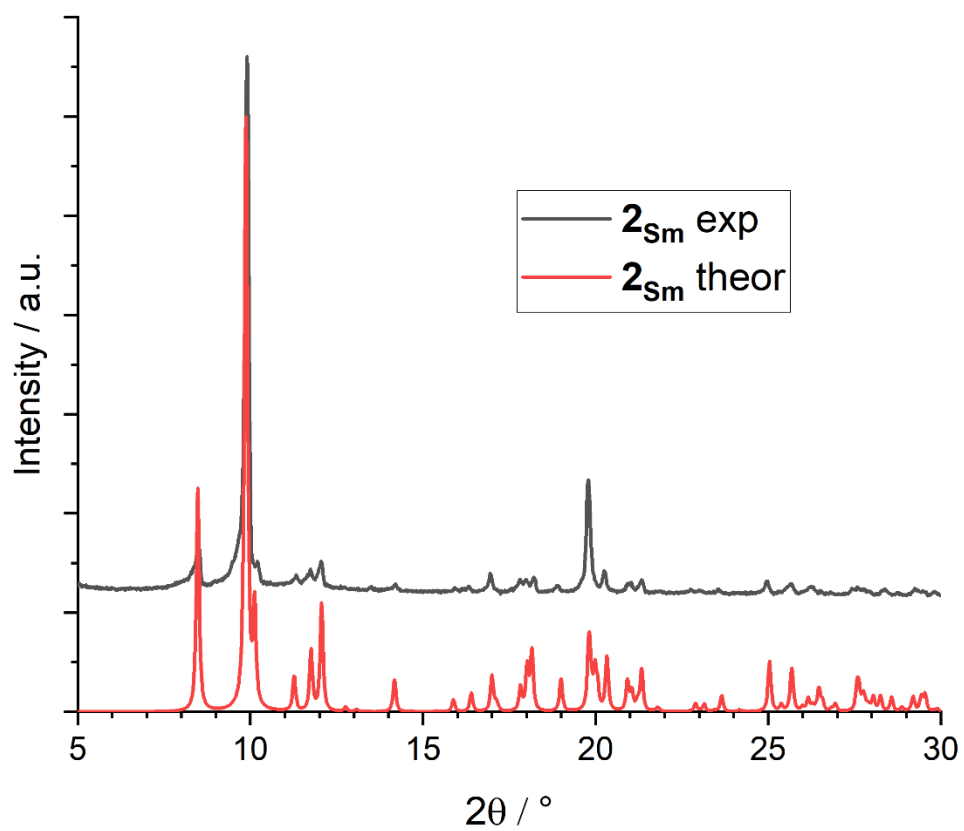


Figure S4. Experimental (black) and theoretical (red) PXRD patterns for 2Sm .

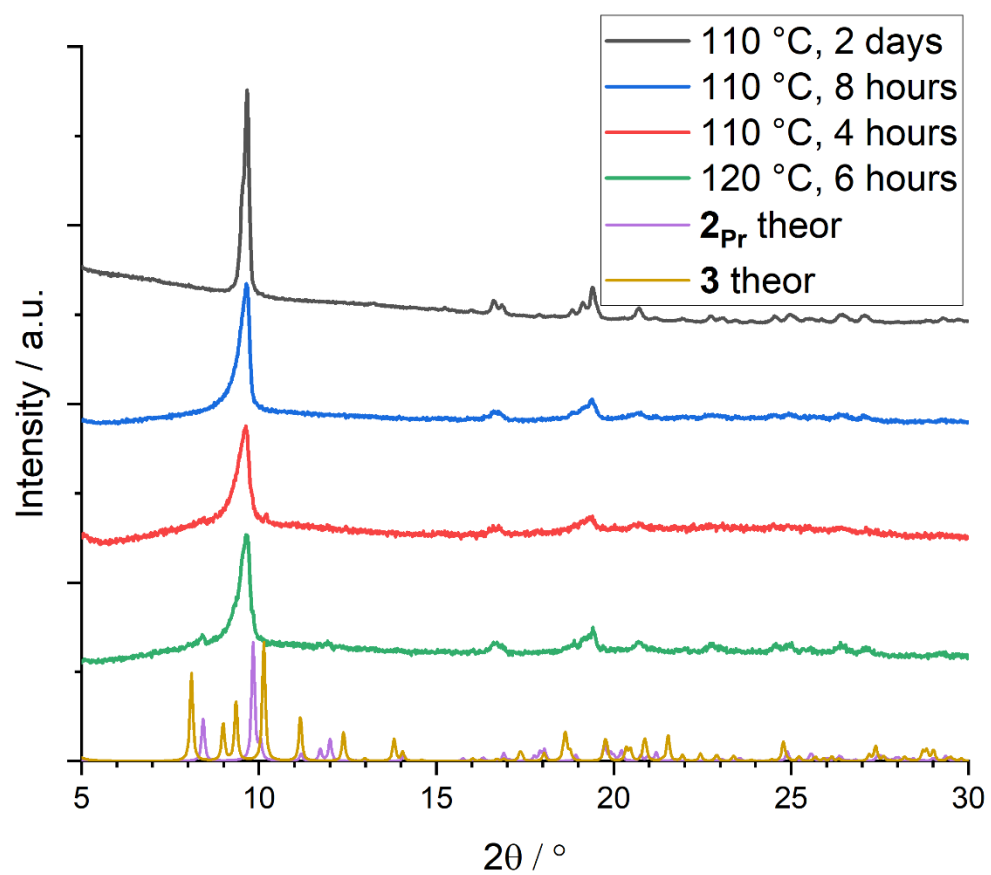


Figure S5. PXRD patterns for **2_{Pr}** temperature- and time-screening syntheses.

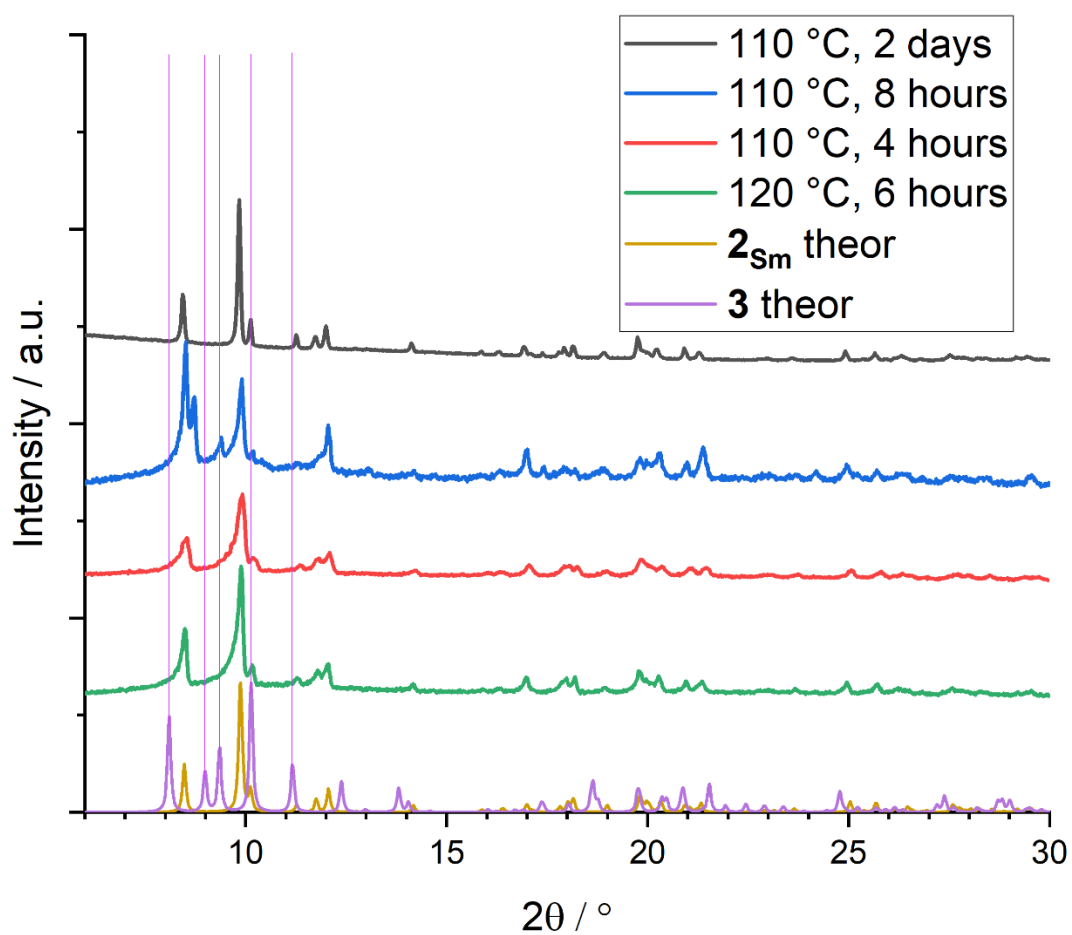


Figure S6. PXRD patterns for **2_{sm}** temperature- and time-screening syntheses. Magenta lines show positions for lowest-angle reflections in theoretical **3** structure.

Description of 1' compound

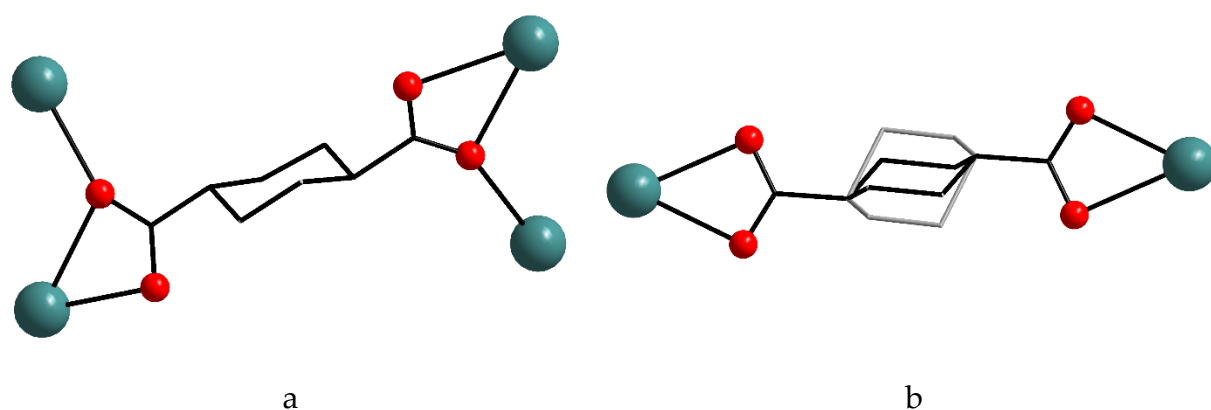


Figure S7. Coordination modes of $(\kappa^2, \kappa^1; \kappa^2, \kappa^1)$ -chdc ligand (a) and $(\kappa^2; \kappa^2)$ -chdc ligand (b). Cyclohexane ring in (b) is disordered over two positions, the second position is shown grey.

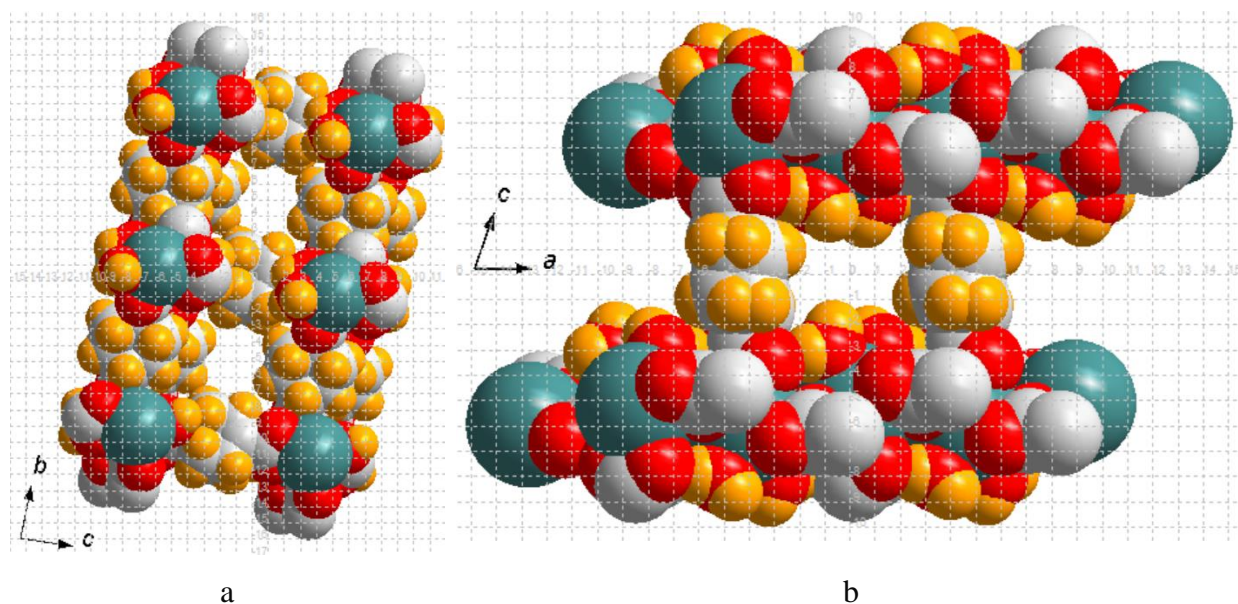


Figure S8. Space-filling representation of channels in the coordination framework of 1. Grids with 1 Å step are imposed on both the figures.

PXRD pattern obtained for the filtered sample of 1' shows the disappearance of most of the peaks presented in the theoretical pattern. However, a detailed analysis of the indexed diffractogram (see figure S1a) gives necessary insights into the nature of molecular mobility in the crystal structure of 1. First two reflections appearing at *ca.* 8.0° and 8.8° on the theoretical diffractogram contain the reflections of crystallographic planes with (0 0 1) and (0 1 0) Miller indexes, respectively. The corresponding interplanar distances are mainly controlled by $(\kappa^2; \kappa^2)$ -chdc ligand (fig. 2c) and $(\kappa^2, \kappa^1; \kappa^2, \kappa^1)$ -chdc ligand, respectively. On the experimental diffractogram, the intensity of (0 0 1) reflection is significantly decreased and the peak is shifted to $2\theta \approx 8.1^\circ$, while (0 1 0) reflection remains intact. At larger angles, the only intensive peak at $2\theta \approx 17.6^\circ$ presents, which corresponds to (0 2 0) plane. Such analysis suggests that desolvation of 1 when stored in air leads to a contraction of $[\text{La}_2(\text{H}_2\text{O})_4(\text{chdc})_3]$ 3D framework along *c* crystallographic axis, which is attributed to the $(\kappa^2; \kappa^2)$ -chdc ligand. On the contrary, the framework structure retains along *b* axis, which corresponds to more robustly coordinated $(\kappa^2, \kappa^1; \kappa^2, \kappa^1)$ -chdc. These observations are also supported by a larger space which is accessible for $(\kappa^2; \kappa^2)$ -chdc molecular motions (see figure 2c) instead of $(\kappa^2, \kappa^1; \kappa^2, \kappa^1)$ -chdc ligands, closely packed to each other (see figure S9b).

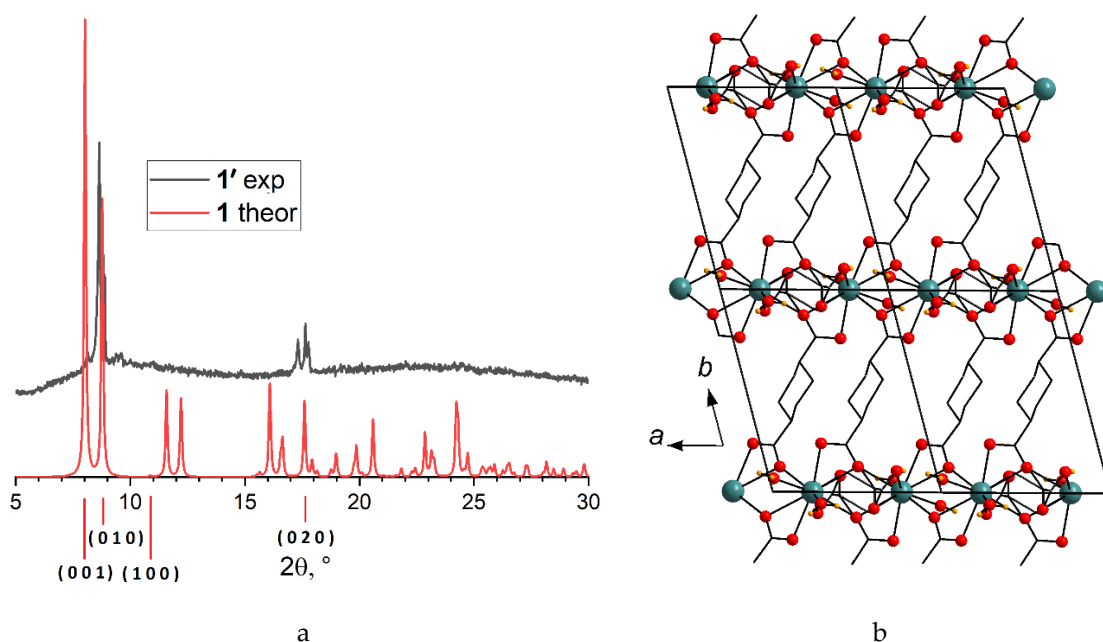


Figure S9. Experimental (black) PXRD pattern for **1'** and theoretical (red) one for **1** (a). View along *c* crystallographic axis in **1** (b). Atom colors correspond to figure 3 in the manuscript, H atoms of chdc ligand and guest molecules are not shown.

IR spectrum of **1'** (Fig. S11) contains characteristic absorption bands for water O–H vibrations, cyclohexane ring and DMF methyl group C(sp³)–H vibrations, DMF C=O vibrations and COO group antisymmetric and symmetric stretchings. The absence of aromatic C(sp²)–H absorption bands in 3075–3060 cm^{−1} region, as well as the absence of strong bands at *ca.* 1583, 1534 and 850 cm^{−1} confirm the absence of phenanthroline in **1'** in any (coordinated or guest) form, fully matching crystal structure data obtained for its parent compound **1**.

According to TG (fig. S10), first 13 % weight loss step for **1'** occurs at *ca.* 280 °C, which corresponds well to the evaporation of the residual coordinated solvents. The second weight loss step occurs at *ca.* 505 °C and is apparently attributed to the decomposition of La-chdc coordination framework.

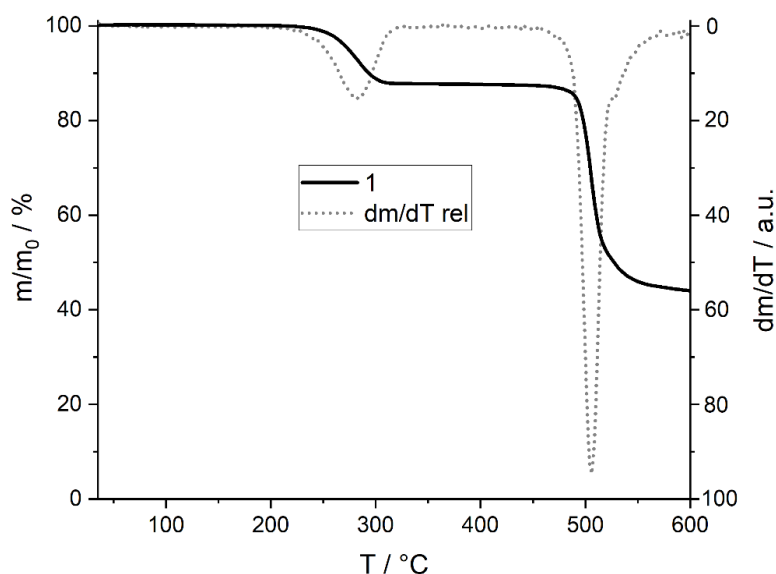


Figure S10. TGA pattern and relative dm/dT curve for the sample **1'**.

Bond lengths in metal environment

Table S3. Selected bond lengths in **2_{Ln}** and **3**.

Bond	2_{Ce}	2_{Pr}	2_{Nd}	2_{Sm}	3 (Nd)
Ln–O(κ^2 -COO)	2.530(2)	2.508(2)	2.502(3)	2.478(2)	2.471(3)
	2.553(2)	2.526(2)	2.516(3)	2.481(3)	2.490(3)
Ln–O(κ^1, κ^1 -COO)	2.437(2)	2.418(2)	2.402(3)	2.374(2)	2.414(3)
	2.487(2)	2.4706(19)	2.456(3)	2.422(2)	2.437(3)
Ln–O(κ^2, κ^1 -COO)	2.4172(18)	2.3965(19)	2.383(2)	2.351(2)	2.413(3)
	2.572(2)	2.553(2)	2.540(3)	2.495(2)	2.561(3)
	2.6151(17)	2.5803(18)	2.573(2)	2.551(2)	2.565(3)
Ln–N(phen)	2.639(14)	2.600(12)	2.581(15)	2.552(12)	2.617(4)

	2.70(3)	2.70(3)	2.67(3)	2.61(2)	2.71(4)

Infrared spectra and TGA details

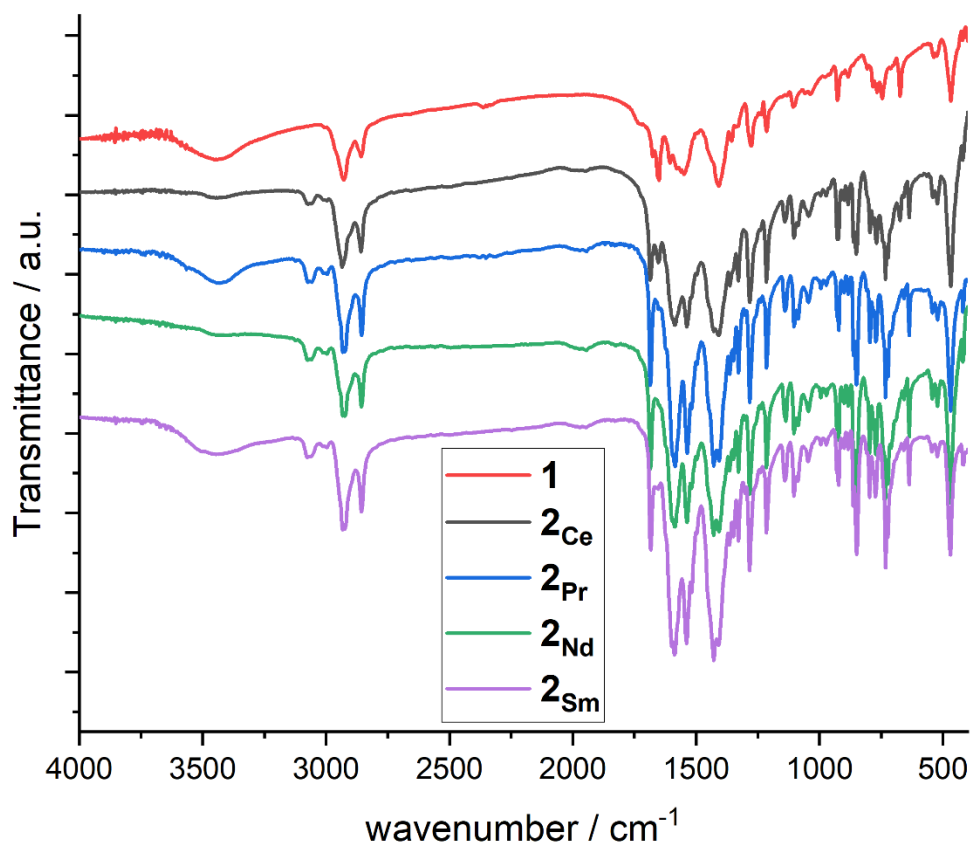


Figure S11. Infrared spectra for 1' and 2_{Ln}.

Table S4. Details on TG decomposition steps of 1' and 2_{Ln}.

Compound	Solvent loss			Framework decomposition			Ln ₂ O ₃ calcd. residue / %
	T range / °	Max.loss T / °	Weight residue / %	T start / °	Max.loss T / °	Weight residue / %	
1'	220 – 320	280	87	~ 460	505	44	36
2 _{Ce}	25 – 400	-	95	~ 400	470, 495	43	27 (28 for CeO ₂)
2 _{Pr}	230 – 430	-	95.5	~ 410	470, 485, 505	34	27
2 _{Nd}	250 – 440	-	96	~ 415	495, 510	38	28
2 _{Sm}	250 – 450	-	96	~ 415	490, 505	39	28

Luminescence supplementary data

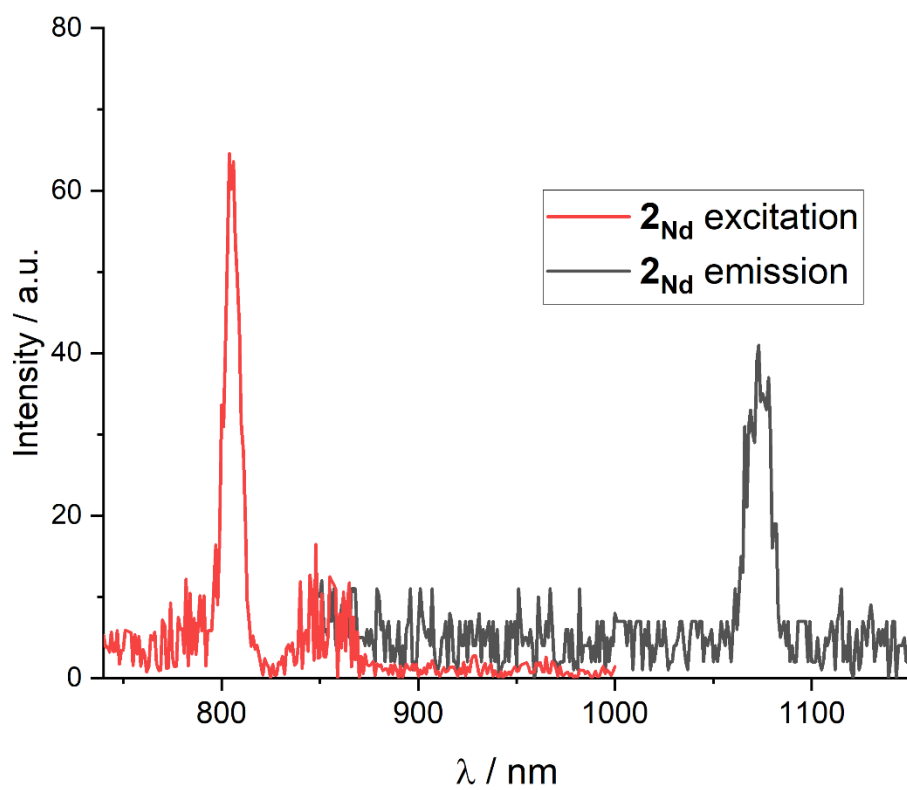


Figure S12. Excitation (red) spectrum for 2_{Nd} at $\lambda_{em} = 1077$ nm and emission (black) spectrum for 2_{Nd} at $\lambda_{ex} = 804$ nm.

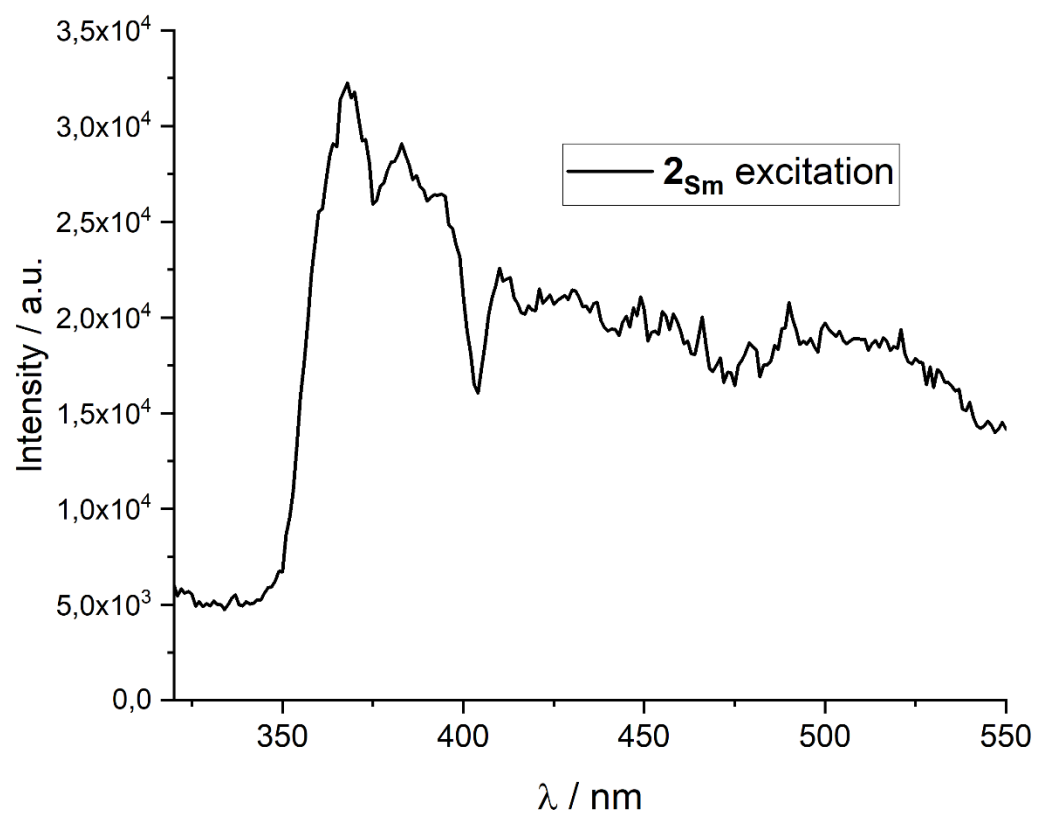


Figure S13. Excitation spectrum for 2_{Sm} at $\lambda_{em} = 580$ nm