



# Article Effect of Temperature on the Adhesion and Bactericidal Activities of Ag<sup>+</sup>-Doped BiVO<sub>4</sub> Ceramic Tiles

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**Abstract:** The aim of this research was to study the effect of temperature on the adhesion and disinfection activities of an Ag<sup>+</sup>-doped BiVO<sub>4</sub> (Ag<sup>+</sup>/BiVO<sub>4</sub>) coating. Ag<sup>+</sup>/BiVO<sub>4</sub> was prepared by a sol–gel method, and spraying was used as the deposition method of coating. X-ray diffraction patterns showed that the monoclinic scheelite phase of the samples was unchanged by annealing at 450–650 °C. Scanning electron microscopy results showed that, at high temperatures, the particles melted and formed a dense coating, and the roughness of the coating decreased after initially increasing. The adhesion and disinfection activities were evaluated by ASTM D3359-08 and disinfection experiments. The results showed that the samples modified by silver had a good disinfection activity when annealed in the range of 450–650 °C. The adhesion increased upon increasing the annealing temperature. The sample annealed at 650 °C showed the best coating adhesion and completely killed *Escherichia coli*, *Staphylococcus aureus*, *Shigella*, and *Salmonella* after 2 h of visible-light irradiation.

Keywords: glazed ceramic tiles; disinfection activity; bismuth vanadate; coating adhesion

## 1. Introduction

The production of photocatalytic ceramic tiles with special functions, such as selfcleaning abilities, super-hydrophilicity, and antibacterial properties, has received attention in recent years [1]. TiO<sub>2</sub> is the most common photocatalyst due to its high chemical stability, good environmental friendliness, and low cost [2–4]. Unfortunately, TiO<sub>2</sub> is prone to an anatase-to-rutile transformation (ART) at high temperatures, significantly reducing its photocatalytic activity. At low temperatures (<950 °C), TiO<sub>2</sub> has extremely poor adhesion to ceramic tiles, which limits its immobilization [1,5]. BiVO<sub>4</sub> is non-toxic, environmentally friendly, and has excellent visible-light photocatalytic activity because the bandgap of monoclinic scheelite BiVO<sub>4</sub> is about 2.4 eV [6,7]. Unfortunately, it suffers from rapid recombination of photogenerated charge carriers, which greatly reduces its photocatalytic activity [8]. This can be overcome by doping or compounding it with other materials, such as Ag, BiOI, BiOCl, or CdS [9–11]. However, no studies have used BiVO<sub>4</sub> coatings on ceramic tiles.

Silver can form a Schottky barrier at the interface of photocatalytic materials, and the high conductivity of Ag can also improve the charge transfer rate of photocatalytic materials, which can improve electron–hole separation. In addition, silver has a better surface plasmon resonance (SPR) effect than other noble metals, which gives materials modified by silver a higher electron–hole separation efficiency and, thus, better photocatalytic activity [12,13]. Silver (Ag<sup>+</sup>) is also an excellent antibacterial agent [14].



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**Copyright:** © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). Temperature has a significant effect on the photocatalytic activity and adhesion of coatings [1]. According to a previous study about  $BiVO_4$ , temperature affected the photocatalytic performance by changing the purity and particle size of  $BiVO_4$ . When the temperature is too high, grains will agglomerate, causing the particle size to increase. When the temperature is too low, other substances, such as  $Bi_4V_2O_{11}$ , can be found in materials that also decrease photocatalytic activity [15]. With regards to adhesion, current research has focused on  $TiO_2$ , and there is little research on the use of  $BiVO_4$  coatings on ceramic tiles. Therefore, in this study, we studied the effect of temperature on the adhesion of coatings.

In this work, we prepared  $Ag^+$ -doped  $BiVO_4$  ( $Ag^+/BiVO_4$ ) by a sol-gel method and then immobilized it onto glazed ceramic tiles by spraying to prepare  $Ag/BiVO_4$ functionalized ceramic tiles. Then, the effects of the annealing temperature on the phase, adhesion, morphology, roughness, and disinfection activities of the coatings were studied. This work may provide a material to replace  $TiO_2$  in functionalized ceramic tiles.

## 2. Materials and Methods

#### 2.1. *Materials*

All chemicals were used as received without further purification. Bismuth nitrate pentahydrate ( $Bi(NO_3)_3 \cdot 5H_2O$ ), ammonium vanadate ( $NH_4VO_3$ ), and silver nitrate ( $AgNO_3$ ) were all purchased from Chengdu Kelong Chemical Plant (AR purity, Chengdu, China). Glacial acetic acid ( $CH_3COOH$ ), absolute ethanol ( $CH_3CH_2OH$ ), and ammonium hydroxide ( $NH_4OH$ ) were purchased from Chongqing Chuandong Chemical Co., Ltd. (AR purity, Chongqing, China). Nitric acid ( $HNO_3$ ) was purchased from Chongqing Huihuang Chemical Co., Ltd. (AR purity, Chongqing, China). Glazed ceramic tiles were purchased from Guangdong Dongpeng Holdings Co., Ltd. (Guangdong, China) (The basic characteristics of the glazed ceramic tiles can be found in Supplementary Materials Table S1). All water was purified with a Millipore Ultra-purification System (Kun Shan Ultrasonic Instruments Co., Ltd., Kun Shan, China).

#### 2.2. Ag<sup>+</sup>-Doped BiVO<sub>4</sub> Functionalized Ceramic Tiles Preparation

 $Ag^+/BiVO_4$  was prepared by a sol–gel method. The optimal Ag content was determined by disinfection experiments (as detailed in Supplementary Materials Figure S1). First, 0.728 g (Bi(NO<sub>3</sub>)<sub>3</sub>·5H<sub>2</sub>O) and 0.046 g AgNO<sub>3</sub> were dissolved in 50 mL of 4 M HNO<sub>3</sub>, and 0.175 g NH<sub>4</sub>VO<sub>3</sub> was dissolved in 50 mL of 4 M NH<sub>4</sub>OH. Then, the two solutions were mixed and magnetically stirred for 30 min to obtain a yellow solution. Ethanol (100 mL) was added to the yellow solution and heated at 70 °C with stirring for 60 min to obtain a yellow sol. Then, 50 mL deionized water and 5 mL 1 M CH<sub>3</sub>COOH were added to the yellow sol and stirred for 30 min to obtain yellow sol–gel. The ceramic tiles were ultrasonically cleaned for 60 min and then sprayed with the sol–gel (the spring-coating procedure was adopted from K. Murugan et al. [16]). The tiles were then dried at 100 °C for 30 min and finally annealed at 450 °C, 550 °C, or 650 °C for 2 h to obtain Ag<sup>+</sup>/BiVO<sub>4</sub> functionalized ceramic tiles (named CSB, for ceramic–silver–bismuth vanadate) and unmodified BiVO<sub>4</sub> functionalized ceramic tiles (named CB, for ceramic–bismuth vanadate).

## 2.3. Disinfection Experiment

All instruments used in the disinfection experiments were sterilized in a steam sterilizer at 125 °C for 20 min.

The visible-light photocatalytic inactivation of *Escherichia coli* [ATCC25922], *Staphylococcus aureus* [ATCC6538], *Shigella* [CMCC (B) 51592], and *Salmonella* [CMCC (B) 50093] was conducted using a xenon lamp (1000 W;  $\lambda \ge 420$  nm) with a 420 nm cut-off filter. All bacteria were cultured in a nutrient agar medium for 16 h. The bacterial stock solution was prepared by dispersing bacteria in 0.9% sterile saline to obtain a concentration of 1 × 10<sup>8</sup> CFU/mL (determined via a standard curve as detailed in the Supplementary Materials Figure S2). Next, 1 mL of the prepared bacterial solution was added to 99 mL of sterilized saline water, to which the functionalized ceramic tiles or uncoated ceramic tiles were added. A final sample group received

no further addition and served as a blank control group. Next, the prepared bacterial suspension was placed in a photoreactor under a xenon lamp for disinfection experiments at 25 °C. Samples were collected after irradiation for 5, 15, 30, 60, and 120 min. Three parallel samples were tested for each experiment.

After diluting the samples collected after disinfection to a certain concentration gradient according to continuous dilution, 0.1 mL of each sample was collected and evenly spread on nutrient agar for cultivation. Three gradient ranges were set for each sample, and four parallel samples were used for each gradient. The coated samples were placed in an incubator set to 37 °C and incubated upside-down for 24 h before counting.

# 2.4. Characterization

The crystal structure of the samples was determined by X-ray diffraction (XRD, X'Pert PRO, Rigaku, Japan). Cu-K $\alpha$  was the X-ray source ( $\lambda = 0.15406$  nm), the power was 3 kW, and the 2 $\theta$  scanning range was 10–80°. Ultraviolet-visible spectrophotometry (UV-Vis/DRS, UV3600, Shimadzu, Japan) was used to characterize the optical properties of the material. The surface roughness of the coatings was analyzed by atomic force microscopy (AFM, Dimension ICON, Bruker, Germany). The scanning area was 20  $\mu$ m × 20  $\mu$ m. The morphology of the coatings was analyzed by field-emission scanning electron microscopy (SEM, S4800, Hitachi, Japan). Before tests, the surface of the sample was sputtered with gold. In addition, the Ag<sup>+</sup> concentration in the disinfection experiments was measured by using inductively coupled plasma mass spectrometry (ICP-MS, PerkinElmer NexION 300X, USA).

The Standard Test Methods for Measuring Adhesion by Tape Test (D 3359-08) [17] was used to test the adhesion of the coatings. To verify the adhesion of the coatings to the glazed tiles, scratch tests (Micro-Combi tester, CSM, MCT, Switzerland) with a linearly increasing load (0.1 N to 30 N, scratch speed of 1 mm/min) were performed on the samples using a Rockwell indenter with a spherical tip (100  $\mu$ m radius). Three scratches were created on each coating, with the minimum distance between two scratches set at 4 mm to achieve a representative average response over greater surfaces. The critical load  $L_{c1}$  (first crack) and  $L_{c2}$  (edge spallation) were determined by optical microscopy.

## 3. Results and Discussion

# 3.1. Phase, Chemical and Optical Characterization

Figure 1 shows the XRD patterns of BiVO<sub>4</sub> and Ag/BiVO<sub>4</sub> coatings annealed at 450  $^{\circ}$ C, 550  $^{\circ}$ C, and 650  $^{\circ}$ C.

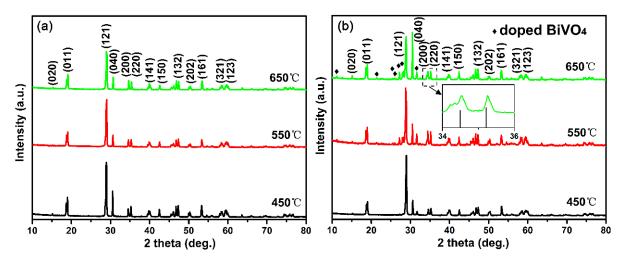


Figure 1. (a) XRD patterns of BiVO<sub>4</sub> and (b) Ag/BiVO<sub>4</sub> coatings obtained at different annealed temperatures.

Comparing  $BiVO_4$  and  $Ag/BiVO_4$  coatings with the corresponding standard diffraction data (JCPDS 14-0688) revealed that  $BiVO_4$  existed as monoclinic scheelite. The diffraction peaks at 15.14°, 18.98°, 28.94°, 30.54°, 34.49°, 37.86°, 39.54°, and 42.33° correspond to the (020), (011), (121), (040), (200), (220), (141) and (150) facets of BiVO<sub>4</sub>, and no additional peaks were observed. These results indicate that BiVO<sub>4</sub> and Ag/BiVO<sub>4</sub> phases were unaffected by temperature within the range of 550–650 °C, revealing their good thermal stability. Figure 1b shows that major characteristic peaks at 11.3°, 28.0°, 21.5°, 25.5°, 26.0°, 28.3°, and 31.70° were observed, which were due to the formation of hybrid metal oxides on the surface of the material. [18,19]. These findings implied that sliver and temperature did not change the lattice structure of BiVO<sub>4</sub>. Interestingly, Figure 1b shows that when the temperature increased to 650 °C, the XRD peaks of Ag/BiVO<sub>4</sub> shifted slightly to the right due to the smaller ionic radius of Bi<sup>3+</sup> (103 pm) than Ag<sup>+</sup> (115 pm), showing that Ag+ replaced bi3+ during the preparation process [20]. Upon increasing the temperature, the intensities of the peaks representing the (040) and (020) planes of Ag/BiVO<sub>4</sub> increased significantly because a high temperature led to the rapid decomposition of vanadate and an increase in the pH, which promoted the formation of BiVO<sub>4</sub> [21].

Figure S3 shows the XPS spectra of BiVO<sub>4</sub> and Ag/BiVO<sub>4</sub>. Early studies about Ag, Ag<sub>2</sub>O, and AgO showed that the order of binding energy of Ag3d<sub>5/2</sub> was Ag (368.2 eV) > Ag<sub>2</sub>O (367.8 eV) > AgO (367.4 eV) [19]. Therefore, Ag was in the +1 valence state, showing that the composite material was Ag<sup>+</sup>/BiVO<sub>4</sub>. Figure S3c is the characteristic peak of oxygen. It can be seen that there were two diffraction peaks of O at binding energies of 529.8 eV and 531.9 eV, indicating that at least two kinds of oxygen are present. The first peak belongs to completely oxidized oxygen ions, and the latter peak corresponds to external OH groups or water molecules adsorbed on the surface of the single-bond sample. Figure S2d shows that V produced a set of double peaks at 516.5 eV and 523.8 eV, which were attributed to V2p<sub>5/2</sub> and V2p<sub>3/2</sub>, respectively. Figure S3e shows that Bi produced peaks at 158.9 eV and 164.2 eV, which were attributed to Bi4f<sub>7/2</sub> and Bi4f<sub>5/2</sub> [9,18].

The UV-Vis spectra of BiVO<sub>4</sub> and Ag/BiVO<sub>4</sub> are shown in Figure S4a. The absorption edge of pure BiVO<sub>4</sub> is at about 527 nm, which is slightly higher than the reported absorption edge (504 nm).

This may be due to the sample's large particle size and surface defects. Compared with  $BiVO_4$ , the absorption capacity of  $Ag/BiVO_4$  increased to 500–850 nm due to the charge transfer between Ag and  $BiVO_4$  [22]. This indicates that  $Ag/BiVO_4$  has better optical properties than pure  $BiVO_4$ . The bandgap of a semiconductor is another important factor that determines its photocatalytic performance; therefore, we also calculated the bandgap energy (*Eg*) of  $BiVO_4$  and  $Ag/BiVO_4$  according to the reported equation [23]:

$$\alpha = A(hv - E)g^{n/2}/hv \tag{1}$$

where  $\alpha$ , *h*, *v*, *A*, and *Eg* are the absorption coefficient, Planck's constant, incident light frequency, constant, and bandgap energy, respectively. *n* depends on the characteristics of the optical transition of the semiconductors, where *n* = 1 for direct and *n* = 4 for indirect bandgap semiconductors. BiVO<sub>4</sub> is a typical direct bandgap semiconductor. From the  $(\alpha hv)^2$  versus *Eg* plot shown in Figure S4b, the bandgaps of BiVO<sub>4</sub> and Ag/BiVO<sub>4</sub> were 2.35 eV and 2.15 eV, respectively. The bandgap of Ag/BiVO<sub>4</sub> was smaller, so all samples should have a bandgap suitable for sterilization under visible light irradiation [24].

## 3.2. Adhesion of Coating on Ceramic Tiles

Adhesion is the most critical variable determining whether a photocatalytic coating can be applied to ceramic tiles, but few studies in the literature have evaluated this [16,25–27]. In this study, adhesion tests were performed according to ASTM D 3359-08 [17], and scratch tests were also performed. The results were classified into six grades: 0B, 1B, 2B, 3B, 4B, and 5B. The adhesion of these grades successively increased.

Table 1 shows the coating adhesion test results of functionalized ceramic tiles annealed at different temperatures. Both  $BiVO_4$  and  $Ag/BiVO_4$  coatings achieved 3B at 450 °C. The coating was greatly affected by temperature, showing increasing adhesion upon increasing the temperature. At 650 °C, the coating adhesion was the best. In addition, the Ag species

increased the photocatalytic coating's adhesion to the ceramic tiles, indicating that Ag species helped immobilize the photocatalytic materials on the surface of ceramic tiles, but their influence was small.

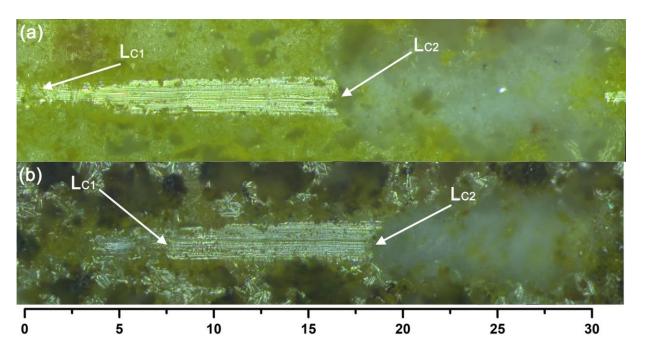
Commlo	Tommorroturo/0C	Percent of Area Removed			
Sample	Temperature/°C —	1st/%	2nd/%	3rd/%	— Classification
СВ	450	5	5	10	3B
CSB	450	5	10	3	3B
CB	550	3	2	6	3B
CSB	550	0	1	1	4B
CB	650	0	0	0	5B
CSB		0	0	0	5B

Table 1. The tape test results of functionalized ceramic tiles were sintered at different temperatures.

In addition, scratch tests with a linearly increasing load were performed on the samples to characterize the adhesion of the coated surface (the). In Table 2, the first (coating failure) and second (coating detachment) critical load values are reported. Compared with the reported results of titanium dioxide [26,27], Ag/BiVO<sub>4</sub>-functionalized ceramic tiles annealed at 650 °C have good adhesion. The scratch track appeared at a higher load with BiVO<sub>4</sub> and Ag/BiVO<sub>4</sub>, as shown in the optical images in Figure 2. Obviously, the CSB samples presented higher critical loads because they failed at 8.0 N and were removed at 18.8 N (Table 2, Figure S5). This phenomenon was explained using SEM data.

**Table 2.** First critical load ( $L_{c1}$ ) and second critical load ( $L_{c2}$ ) of the samples.

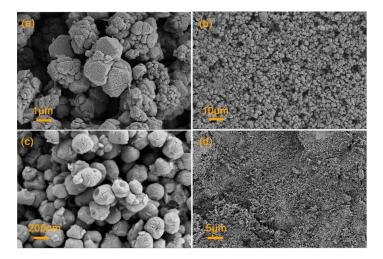
Sample	<i>L</i> <sub>c1</sub> (N)	<i>L</i> <sub>c2</sub> (N)
СВ	$2.1\pm0.3$	$17.0 \pm 0.5$
CSB	$8.0\pm0.9$	$18.8 \pm 1$



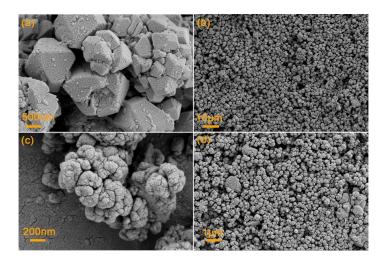
**Figure 2.** Optical images of the samples (annealed at 650 °C) after the scratch tests. (**a**) BiVO<sub>4</sub> photocatalytic ceramic tiles; (**b**) Ag/BiVO<sub>4</sub> photocatalytic ceramic tiles.

# 3.3. Coating Microstructure and Morphology

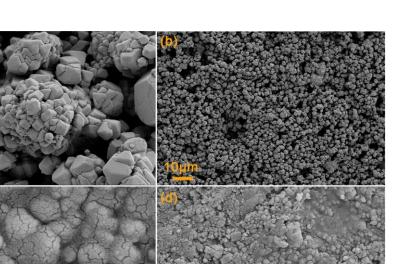
SEM was used to investigate the adhesion and morphology of the coatings obtained at different temperatures. Figures 3-5 and S6 show the SEM images of samples obtained at different annealing temperatures. As shown in Figures 4 and 5, there were large voids between the photocatalyst particles on the surface of the coating, which led to poor adhesion of the coating, causing the photocatalyst particles to agglomerate. At 550 °C (Figure 4), the voids between the photocatalyst particles on the surface of the ceramic tile decreased. This phenomenon was more evident in BiVO<sub>4</sub> functionalized ceramic tiles, while a small amount of photocatalyst melted and fused to form a dense coating on the surface of CSB. When the temperature increased to 650 °C (Figure 5), the photocatalytic materials immobilized on the surface of CB agglomerated into larger particles. This phenomenon was more obvious in CSB, in which the particles melted to form a dense coating structure, giving the coating on the surface of CSB the best adhesion. In addition to the temperature, silver significantly affected the morphology of the coating. Ag<sup>+</sup> promoted the mutual melting of BiVO<sub>4</sub> particles, so the voids between catalyst particles on the surface of CSB were much smaller, which improved the adhesion. Figure S6 shows that the thickness of the Ag/BiVO<sub>4</sub> coating annealed at 650  $^{\circ}$ C was 17.64  $\mu$ m, which was attributed to the large particle size.



**Figure 3.** SEM images of (**a**,**b**) BiVO<sub>4</sub> and (**c**,**d**) Ag/BiVO<sub>4</sub> functionalized ceramic tiles obtained at 450 °C.



**Figure 4.** SEM images of (**a**,**b**) BiVO<sub>4</sub> and (**c**,**d**) Ag/BiVO<sub>4</sub> functionalized ceramic tiles obtained at 550 °C.



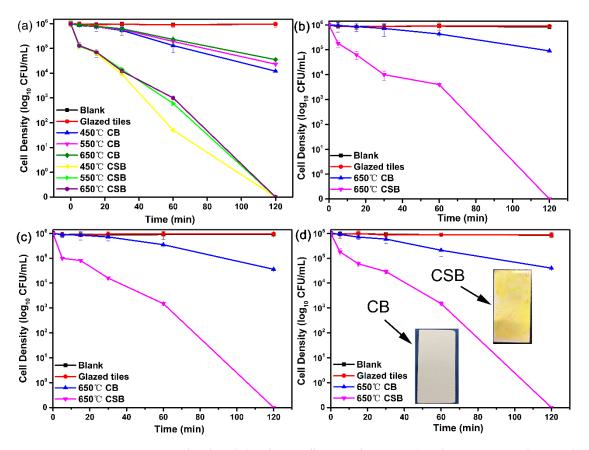
**Figure 5.** SEM images of (**a**,**b**) BiVO<sub>4</sub> and (**c**,**d**) Ag/BiVO<sub>4</sub> functionalized ceramic tiles obtained at

650 °C.

Figures S7–S9 show the AFM images of the samples obtained at different temperatures. At 450 °C, the roughness (estimated by arithmetical mean deviation of the profile (Ra)) of CB and CSB was 70.2 nm and 141 nm, respectively. When the temperature increased to 550 °C, the roughness of the two samples increased to 117 nm and 162 nm, respectively. When the temperature increased to 650 °C, the roughness decreased to 100 nm and 117 nm, respectively. Studies have shown that the agglomeration and particle size of photocatalytic materials affect their parameters. When a material agglomerates or melts into larger particles, the roughness of the coating will increase [28]. The SEM results show that when the temperature increased from 450 °C to 550 °C, the photocatalytic particles agglomerated, which increased the surface roughness of the coated ceramic tiles. As the temperature increased to 650 °C, the roughness decreased. The reason is that the photocatalytic particles on the surface of the ceramic tiles melted at high temperatures, and the voids between the materials gradually decreased, which decreased the roughness of the coating. Compared with CB, the CSB coating had a much higher roughness for the hybrid metal oxides on the surface of BiVO<sub>4</sub> (determined from SEM, as detailed in Figures S10 and S11). Studies have shown that surfaces with a higher roughness have a greater surface area, which increases the photocatalytic reaction rate [29]. Therefore, CSB displayed higher disinfection efficiency.

#### 3.4. Disinfection Results

The sterilization effects of the samples at different temperatures were evaluated by disinfection experiments, and the results are shown in Figure 6a. The bacterial concentration using glazed ceramic tiles was almost unchanged, indicating that unmodified ceramic tiles cannot kill bacteria. Upon increasing the temperature, the disinfection efficiency of CB and CSB decreased. The reason can be attributed to particle melting and agglomeration at high temperatures, which increased the particle size and decreased the available surface area, so the bactericidal effect was worse. This increased the particle size of the material, which decreased its disinfection efficiency [30,31]. Compared with CB, CSB had a higher disinfection efficiency, and the CSB annealed within the range 450–650 °C killed 100% of bacteria after disinfection under visible light irradiation for 2 h.



**Figure 6.** *Escherichia coli* disinfection efficiency of BiVO<sub>4</sub> and Ag/BiVO<sub>4</sub> ceramic tiles annealed at different temperatures (**a**) and ceramic tiles disinfection efficiency under different bacterial species. (**b**) *Staphylococcus aureus*, (**c**) *Salmonella*, and (**d**) *Shigella* (bacterial suspension concentration:  $1 \times 10^6$  CFU/mL; common anions: 0.9% Cl<sup>-</sup>).

To illustrate that the functionalized ceramic tiles have good application prospects, the tiles annealed at 650 °C were selected for disinfection experiments using different bacteria species, and the results are shown in Figure 6a–d. The experimental results confirmed that CSB had a high disinfection efficiency against gram-negative bacteria such as *E. coli* and gram-positive bacteria such as *Staphylococcus aureus*, *Shigella*, and *Salmonella*. In addition, CSB showed a high disinfection efficiency, and it completely killed four common pathogenic bacteria after visible light irradiation for 2 h. Compared with TiO<sub>2</sub>-coated ceramic tiles [27,32], Ag<sup>+</sup>/BiVO<sub>4</sub>-functionalized ceramic tiles annealed at 650 °C displayed the highest disinfection efficiency. In addition, the concentrations of released Ag<sup>+</sup> (Figure S12) remained under 100  $\mu$ g/L. A previous study [33] showed that the inactivation effect of 100  $\mu$ g/L Ag<sup>+</sup> was only 0.9 log CFU/mL in 3 h; therefore, the contribution of Ag<sup>+</sup> from Ag<sup>+</sup>/BiVO<sub>4</sub> functionalized ceramic tiles was negligible.

# 4. Conclusions

BiVO<sub>4</sub> and Ag/BiVO<sub>4</sub>-coated ceramic tiles were prepared using a sol–gel and spraying method. A disinfection experiment and subsequent characterization investigated the effect of temperature. It was found that temperature had little effect on the phases of BiVO<sub>4</sub> and Ag/BiVO<sub>4</sub>, but it greatly influenced the adhesion and disinfection activities of the coatings. This was attributed to the melting and agglomeration of photocatalytic particles at 550 °C, which then formed a dense coating structure at 650 °C. Moreover, the experimental results demonstrated that the Ag/BiVO<sub>4</sub>-functionalized ceramic tiles annealed at 650 °C displayed the best adhesion and killed various bacteria after 2 h of visible-light irradiation. Therefore, Ag/BiVO<sub>4</sub>-functionalized ceramic tiles are promising candidates for practical applications.

Supplementary Materials: The following supporting information can be downloaded at: https:// www.mdpi.com/article/10.3390/inorganics10050061/s1, Figure S1: The effect of Ag dosage on the sterilization effect of Ag/BiVO<sub>4</sub> obtained at 450 °C (bacterial suspension concentration:  $1 \times 10^{6}$  CFU/mL; bacterial species: *Escherichia coli*; common anions: 0.9% Cl<sup>-</sup>); Figure S2: Standard curve of bacteria suspension concentration vs. absorbance (wavelength: 600 nm); Figure S3: XPS spectra of BiVO<sub>4</sub> and Ag/BiVO<sub>4</sub> powder (prepared temperature: 450 °C): (a) Full-scan spectra of BiVO<sub>4</sub> and Ag/BiVO<sub>4</sub>; (b–e) peak positions of Ag, O, V, and Bi species; Figure S4: (a) UV-Vis spectra of samples prepared at 450 °C; (b) band gap widths of the samples prepared at 450 °C; Figure S5: Escherichia disinfection efficiency of Ag/BiVO<sub>4</sub> ceramic tiles after scratch tests (bacterial suspension concentration:  $1 \times 10^6$  CFU/mL; common anions: 0.9% Cl<sup>-</sup>); Figure S6: SEM cross-section image of Ag/BiVO<sub>4</sub> coating obtained at 650 °C; Figure S7: AFM images of (a-b) BiVO<sub>4</sub> and (c-d) Ag/BiVO<sub>4</sub>functionalized ceramic tiles annealed at 450 °C; Figure S8: AFM images of (a-b) BiVO<sub>4</sub> and (c-d) Ag/BiVO<sub>4</sub>-functionalized ceramic tiles annealed at 550 °C; Figure S9: AFM images of (a-b) BiVO<sub>4</sub> and (c-d) Ag/BiVO<sub>4</sub>-functionalized ceramic tiles annealed at 650 °C; Figure S10: (a-c) SEM images of BiVO<sub>4</sub> and (d-f) Ag/BiVO<sub>4</sub> powders at 450 °C; Figure S11: Grain size distribution of BiVO<sub>4</sub> and (a) Ag/BiVO<sub>4</sub> (b) powder obtained at 450  $^{\circ}$ C; Figure S12: Concentration of Ag+ released from of Ag/BiVO4-functionalized ceramic tiles annealed at 650 °C; Table S1: The basic characteristics of glazed ceramic tiles (model number: 630ELN52005-A).

**Author Contributions:** Y.Z.: conceptualization; data curation; investigation; methodology; resources; writing—original draft; writing—review and editing. X.Z., H.W., S.F., X.L., R.L. and Q.H.: writing—review and editing; project administration; supervision. F.J. and X.X.: resources; writing—review and editing. All authors have read and agreed to the published version of the manuscript.

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