

Supplementary Information

Ammonium and Phosphonium Salts Containing Monoanionic Iron(II) Half-Sandwich Complexes [Fe(η^5 -Cp^{*})X₂][−] (X = Cl – I)

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Table S1. X-ray crystallographic details.

	NnPr ₄ [Fe(η^5 -Cp*)Cl ₂]	NnPr ₄ [Fe(η^5 -Cp*)BrCl]	NnPr ₄ [Fe(η^5 -Cp*)Br ₂]	PPh ₄ [Fe(η^5 -Cp*)Cl ₂]	PPh ₄ [Fe(η^5 -Cp*)Br ₂]	PPh ₄ [Fe(η^5 -Cp*)I ₂]	[Fe(η^5 -Cp*)Cl(CO) ₂]
Empirical formula	C ₂₂ H ₄₃ Cl ₂ FeN	C ₂₂ H ₄₃ BrClFeN	C ₂₂ H ₄₃ Br ₂ FeN	C ₃₄ H ₃₅ Cl ₂ FeP	C ₃₄ H ₃₅ Br ₂ FeP	C ₃₄ H ₃₅ FeI ₂ P	C ₁₂ H ₁₅ ClFeO ₂
Formula weight	448.32	492.78	537.24	601.34	690.26	784.24	282.54
Crystal system	triclinic	triclinic	orthorhombic	orthorhombic	orthorhombic	orthorhombic	triclinic
Space group	<i>P</i> -1	<i>P</i> -1	<i>Pca</i> 2 ₁	<i>Pna</i> 2 ₁	<i>Pnma</i>	<i>Pnma</i>	<i>P</i> -1
<i>a</i> /Å	8.5399(4)	8.526(2)	34.9997(10)	29.237(7)	29.7108(15)	30.170(2)	7.4528(16)
<i>b</i> /Å	12.3334(5)	12.334(2)	12.6237(4)	7.5015(9)	14.0744(7)	14.3281(14)	7.6886(16)
<i>c</i> /Å	14.2796(6)	14.310(3)	23.2128(7)	14.075(4)	7.4002(3)	7.4587(5)	11.905(4)
α /°	69.433(3)	69.577(13)	90	90	90	90	79.38(2)
β /°	79.375(3)	79.840(17)	90	90	90	90	78.75(2)
γ /°	86.175(3)	85.489(18)	90	90	90	90	69.59 0(16)
Volume/Å ³	1384.00(11)	1388.0(5)	10256.0(5)	3087.0(13)	3094.5(3)	3224.3(4)	622.0(3)
<i>Z</i>	2	2	16	4	4	4	2
ρ_{calcd} /g cm ⁻³	1.076	1.179	1.392	1.294	1.482	1.616	1.509
μ /mm ⁻¹	6.165	2.085	8.390	6.147	3.143	2.453	1.408
<i>F</i> (000)	484.0	520.0	4448.0	1256.0	1400.0	1544.0	292.0
Crystal size/mm ³	0.12 × 0.09 × 0.02	0.18 × 0.13 × 0.05	0.15 × 0.07 × 0.02	0.23 × 0.14 × 0.04	0.45 × 0.21 × 0.07	0.22 × 0.11 × 0.04	0.15 × 0.08 × 0.03
Radiation used	Cu K α (λ = 1.54186)	Mo K α (λ = 0.71073)	Cu K α (λ = 1.54186)	Cu K α (λ = 1.54186)	Mo K α (λ = 0.71073)	Mo K α (λ = 0.71073)	Mo K α (λ = 0.71073)
2 θ range for data collection/°	6.712 to 143.33	3.076 to 51.752	5.05 to 143.418	6.046 to 140.404	3.986 to 51.134	3.92 to 51.61	3.516 to 51.478
Index ranges	−5 ≤ <i>h</i> ≤ 10	−10 ≤ <i>h</i> ≤ 10	−22 ≤ <i>h</i> ≤ 41	−31 ≤ <i>h</i> ≤ 35	−30 ≤ <i>h</i> ≤ 35	−36 ≤ <i>h</i> ≤ 32	−7 ≤ <i>h</i> ≤ 8
	−14 ≤ <i>k</i> ≤ 15	−13 ≤ <i>k</i> ≤ 15	−11 ≤ <i>k</i> ≤ 15	−9 ≤ <i>k</i> ≤ 4	−17 ≤ <i>k</i> ≤ 14	−15 ≤ <i>k</i> ≤ 17	−9 ≤ <i>k</i> ≤ 9
	−16 ≤ <i>l</i> ≤ 17	−17 ≤ <i>l</i> ≤ 17	−28 ≤ <i>l</i> ≤ 27	−16 ≤ <i>l</i> ≤ 16	−8 ≤ <i>l</i> ≤ 7	−7 ≤ <i>l</i> ≤ 9	−14 ≤ <i>l</i> ≤ 14
Refl. collected	11462	9633	60397	14812	6839	8493	4139
Independent refl.	5213 [<i>R</i> _{int} = 0.0403]	5258 [<i>R</i> _{int} = 0.0561]	19487 [<i>R</i> _{int} = 0.0534]	5553 [<i>R</i> _{int} = 0.0412]	2951 [<i>R</i> _{int} = 0.0354]	3182 [<i>R</i> _{int} = 0.0261]	2312 [<i>R</i> _{int} = 0.0653]
Data/restr./param.	5213/0/244	5258/2/251	19487/135/936	5553/1/348	2951/2/197	3182/2/199	2312/0/150
Goodness-of-fit on <i>F</i> ²	1.045	1.020	1.566	1.052	1.144	1.078	1.028
Final <i>R</i> indexes	<i>R</i> ₁ = 0.0488	<i>R</i> ₁ = 0.0626	<i>R</i> ₁ = 0.1434	<i>R</i> ₁ = 0.0453	<i>R</i> ₁ = 0.0518	<i>R</i> ₁ = 0.0348	<i>R</i> ₁ = 0.0665
[<i>I</i> > 2 σ (<i>I</i>)]	<i>wR</i> ₂ = 0.1261	<i>wR</i> ₂ = 0.1278	<i>wR</i> ₂ = 0.3794	<i>wR</i> ₂ = 0.1047	<i>wR</i> ₂ = 0.1344	<i>wR</i> ₂ = 0.0949	<i>wR</i> ₂ = 0.1597
Final <i>R</i> indexes	<i>R</i> ₁ = 0.0618	<i>R</i> ₁ = 0.1001	<i>R</i> ₁ = 0.1974	<i>R</i> ₁ = 0.0584	<i>R</i> ₁ = 0.0626	<i>R</i> ₁ = 0.0417	<i>R</i> ₁ = 0.0947
[all data]	<i>wR</i> ₂ = 0.1355	<i>wR</i> ₂ = 0.1459	<i>wR</i> ₂ = 0.4676	<i>wR</i> ₂ = 0.1132	<i>wR</i> ₂ = 0.1425	<i>wR</i> ₂ = 0.0980	<i>wR</i> ₂ = 0.1815
Largest diff. peak/hole / e Å ⁻³	0.34/−0.52	0.50/−0.51	2.65/−0.99	0.40/−0.36	1.14/−0.58	1.35/−1.03	0.52/−1.31
Flack parameter			0.29(3)	−0.003(8)			
CCDC No.	2300615	2300616	2300617	2300618	2300619	2300620	2300621

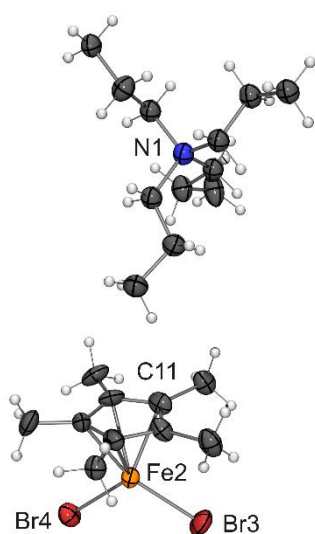


Figure S1. Molecular structure of $\text{NnPr}_4[\text{Fe}(\eta^5\text{-Cp}^*)\text{Br}_2]$ in the crystal (ORTEP with 50% probability ellipsoids). The asymmetric unit contains four anions and four cations. Only one each is shown.

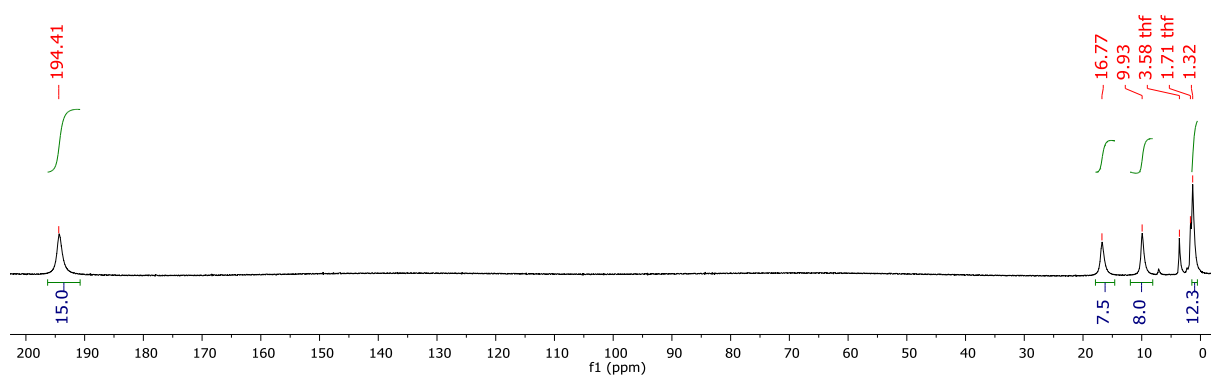


Figure S2. ^1H NMR spectrum (400 MHz, $\text{THF-}d_8$) of $\text{NnPr}_4[\text{Fe}(\eta^5\text{-Cp}^*)\text{Cl}_2]$.

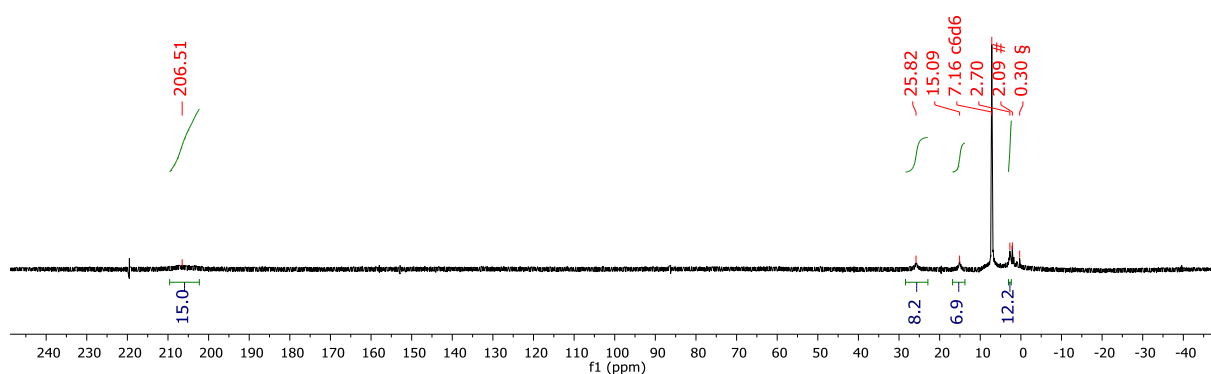


Figure S3. ^1H NMR spectrum (400 MHz, $\text{THF-}d_8$) of $\text{NnPr}_4[\text{Fe}(\eta^5\text{-Cp}^*)\text{BrCl}]$. Signals marked belong to small amounts of decamethylferrocene (#) and silicone grease (§).

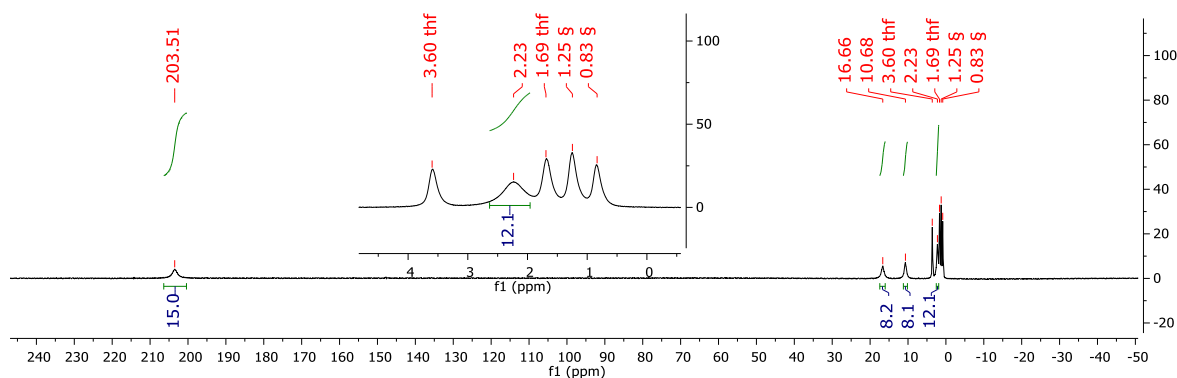


Figure S4. ^1H NMR spectrum (400 MHz, $\text{THF-}d_8$) of $\text{NnPr}_4[\text{Fe}(\eta^5\text{-Cp}^*)\text{Br}_2]$. Signals marked belong to residual *n*-hexane (§).

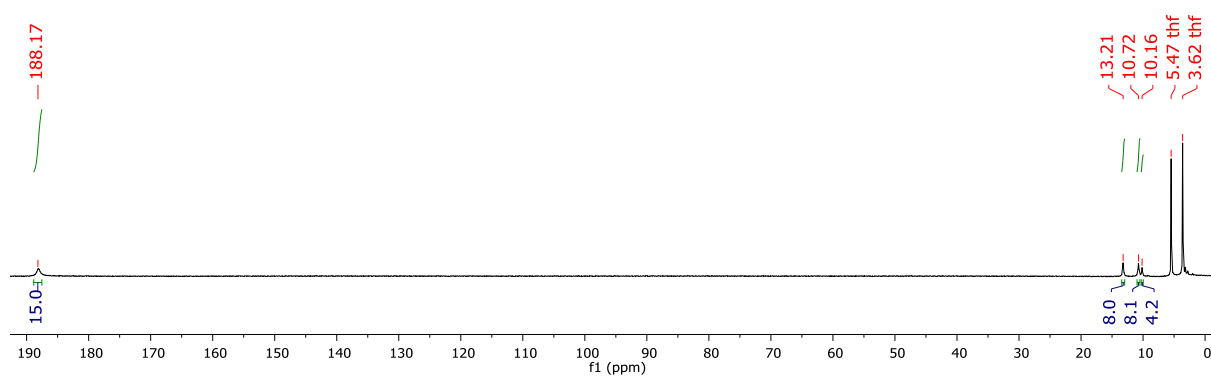


Figure S5. ¹H NMR spectrum (500 MHz, THF-*d*₈) of PPh₄[Fe(η⁵-Cp*)Cl₂].

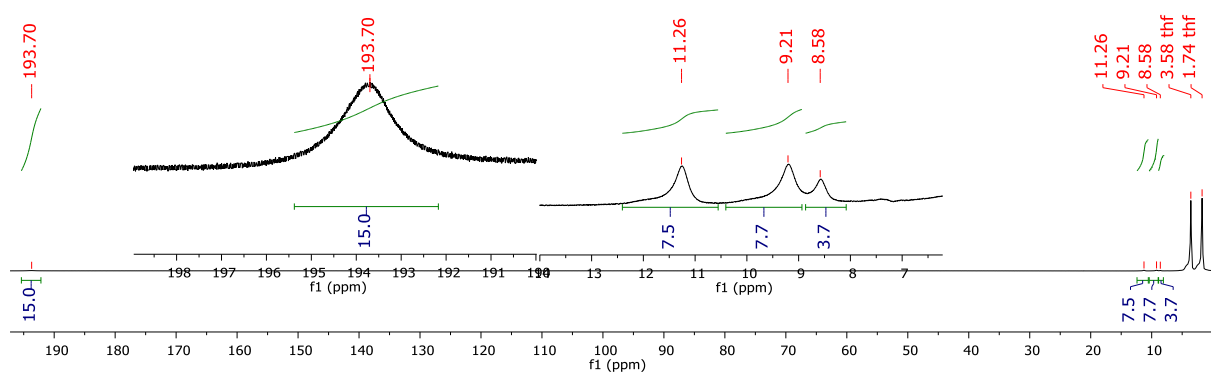


Figure S6. ¹H NMR spectrum (500 MHz, THF-*d*₈) of PPh₄[Fe(η⁵-Cp*)Br₂].

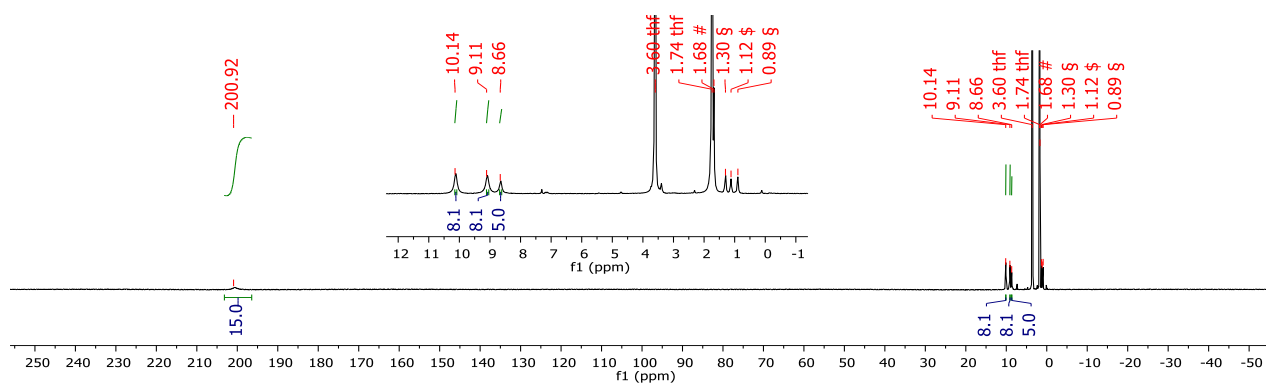


Figure S7. ¹H NMR spectrum (500 MHz, THF-*d*₈) of PPh₄[Fe(η⁵-Cp*)I₂]. Signals marked belong to residual n-hexane (\$), decamethylferrocene (#) and an unknown impurity (\$).

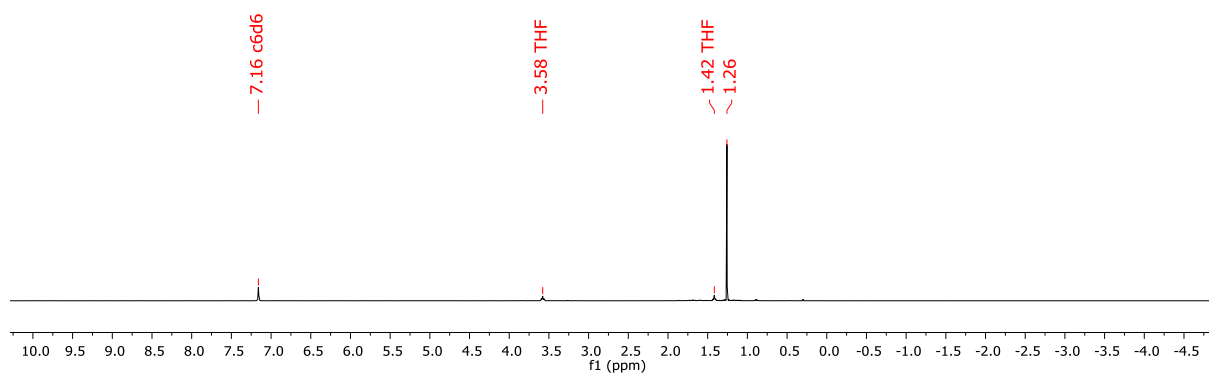


Figure S8. ¹H NMR spectrum (400 MHz, C₆D₆) of [Fe(η⁵-Cp*)Cl(CO)₂].

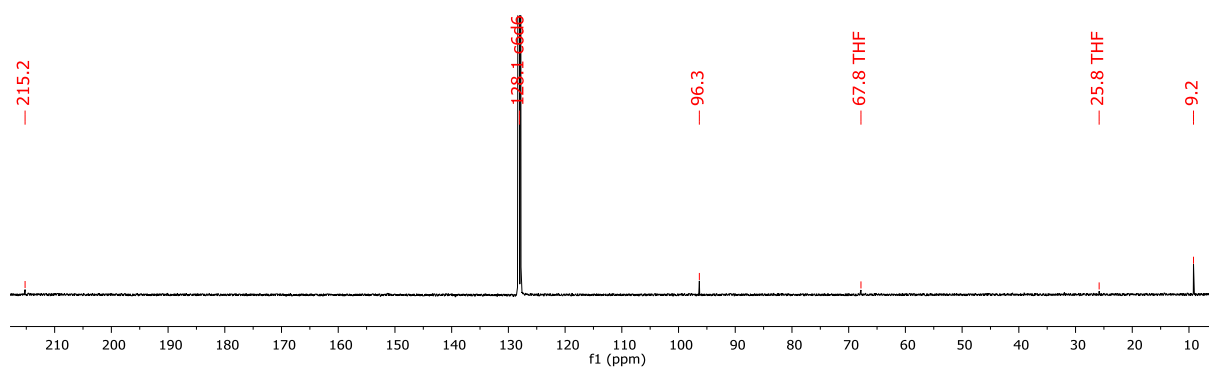


Figure S9. ¹³C NMR spectrum (101 MHz, C₆D₆) of [Fe(η⁵-Cp*)Cl(CO)₂].