

Supporting Information

Surface Passivation of Organic-Inorganic Hybrid Perovskites with Methylhydrazine Iodide for Enhanced Photovoltaic Device Performance

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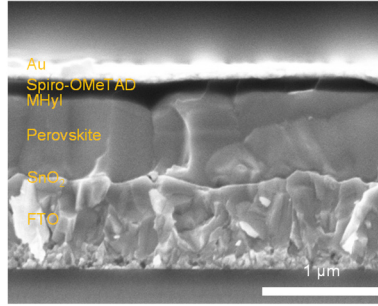


Figure S1. Cross-sectional SEM image of target device.

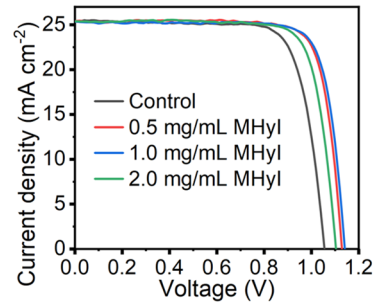


Figure S2. J - V curves of the PSCs treated with different concentrations of MHyI solution.

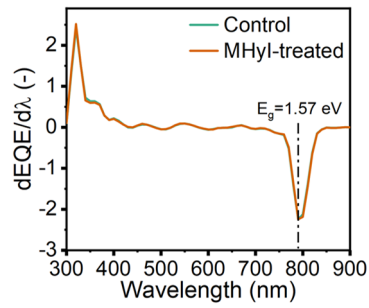


Figure S3. Determination of the absorption of the perovskite absorber by the differential of EQE curves.

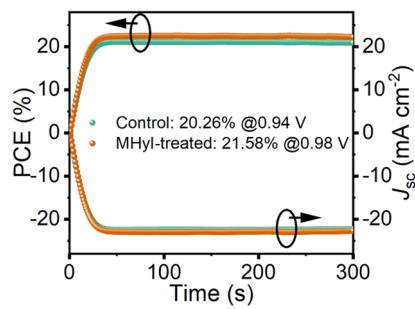


Figure S4. The SPO of the control and MHyI-treated devices.

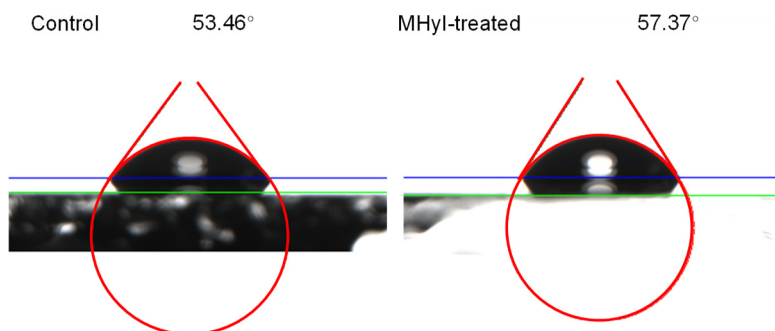


Figure S5. The static water contacts angle of the standard film and MHyl-treated film.

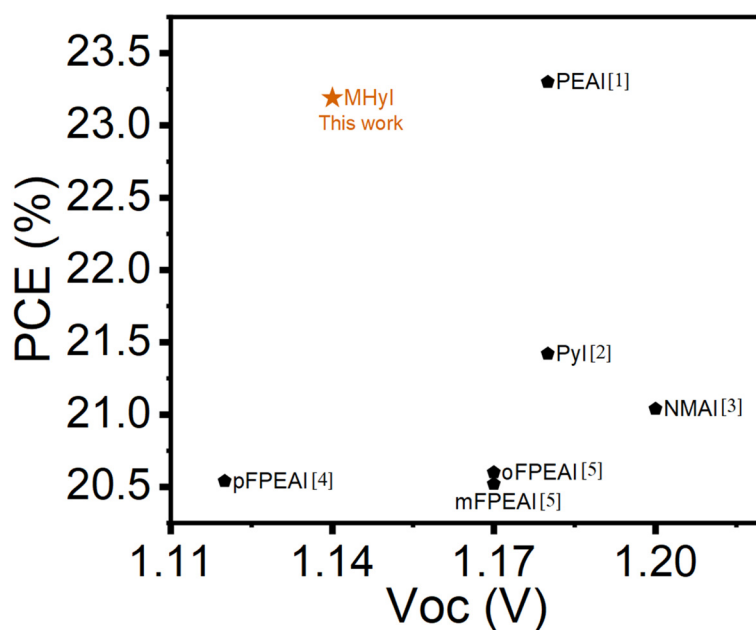


Figure S6. Comparative performance of devices with different post-treatment ammonium salts (PEAI [1], PyI [2], NMAI [3], pFPEAI [4], oFPEAI [5], mFPEAI [5]).

Table S1. Fitted results of TRPL curves of perovskite films with and without MHyl-modified and average PL lifetime τ_{ave} was calculated using $\tau_{ave}=(A_1\tau_1^2+A_2\tau_2^2)/(A_1\tau_1+A_2\tau_2)$.

	A_1	τ_1 (ns)	A_2	τ_2 (ns)	τ_{ave} (ns)
Control	0.36	31.67	0.57	131.53	118.09
MHyl-treated	0.19	35.46	0.73	354.56	346.37

Table S2. Fitting parameters of the EIS measurement based on control and MHyl-treated perovskite solar cells.

	R_s (Ω)	C_{tr} (F)	R_{tr} (Ω)	C_{rec} (F)	R_{rec} (Ω)
Control	10.90	2.115×10^{-6}	1.125×10^5	1.092×10^{-8}	2.189×10^4
MHyl-treated	10.16	1.796×10^{-6}	1.504×10^5	1.101×10^{-8}	3.932×10^4

References:

1. Jiang, Q.; Zhao, Y.; Zhang, X.; Yang, X.; Chen, Y.; Chu, Z.; Ye, Q.; Li, X.; Yin, Z.; You, J. Surface Passivation of Perovskite Film for Efficient Solar Cells. *Nature Photonics* **2019**, *13*, 460–466, doi:10.1038/s41566-019-0398-2.
2. Du, Y.; Wu, J.; Zhang, X.; Zhu, Q.; Zhang, M.; Liu, X.; Zou, Y.; Wang, S.; Sun, W. Surface Passivation Using Pyridinium Iodide for Highly Efficient Planar Perovskite Solar Cells. *Journal of Energy Chemistry* **2021**, *52*, 84–91, doi:10.1016/j.jechem.2020.04.049.
3. Liang, L.; Luo, H.; Hu, J.; Li, H.; Gao, P. Efficient Perovskite Solar Cells by Reducing Interface-Mediated Recombination: A Bulky Amine Approach. *Advanced Energy Materials* **2020**, *10*, doi:10.1002/aenm.202000197.
4. Zhou, Q.; Liang, L.; Hu, J.; Cao, B.; Yang, L.; Wu, T.; Li, X.; Zhang, B.; Gao, P. High-Performance Perovskite Solar Cells with Enhanced Environmental Stability Based on a (p-FC 6 H 4 C 2 H 4 NH 3) 2 [PbI 4] Capping Layer. *Advanced Energy Materials* **2019**, *9*, doi:10.1002/aenm.201802595.
5. Zhou, Q.; Xiong, Q.; Zhang, Z.; Hu, J.; Lin, F.; Liang, L.; Wu, T.; Wang, X.; Wu, J.; Zhang, B.; et al. Fluoroaromatic Cation-Assisted Planar Junction Perovskite Solar Cells with Improved VOC and Stability: The Role of Fluorination Position. *Solar RRL* **2020**, *4*, doi:10.1002/solr.202000107.