Supporting Information

Iron(III) azadiphenolate compounds in a new family of spin crossover iron(II)-iron(III) mixed-valent complexes

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Table of Contents

Table S1 Crystallographic data and structure refinement for 1, 2, and 3

Table S2 Selected bond length and octahedral distortion parameters for 1-ref[1], 1, 2, and 3

Table S3 Selected intermolecular interactions in compound 2

 Table S4 Selected intermolecular interactions in compound 3

 Table S5 Selected intermolecular interactions in compound 4

 Table S6 Selected intermolecular interactions in compound 5

Table S7 Selected intermolecular interactions in compound 6

Figure S1 a) A dimer of $[Fe(azp)_2]^-$ moieties linked via K⁺ cations in **1-ref [1]**, and b) the K⁺ ions forming η_4 interactions to aromatic rings of **1**, and c) {K-OH₂···O(phenolate)-Fe} hydrogen-bonding pathways in **1**.

Figure S2 a) Crystal packing of a) **1-ref**, viewing through the *a*-axis. and b) **1**, viewing through the *b*-axis. Red and blue broken lines represent $K^+ \cdots \pi$ and $O \cdots H$ interactions, respectively. Hydrogen atoms omitted for clarity

Figure S3 C-H···O interactions between Fe^{II} moieties and ClO₄⁻ in compound **2** that link Fe^{II} cationic molecules in a) a sheet along the ac plane and b) a pseudo-3D network, c) showing ClO_4^- anions around the $[Fe^{II}{(pz)_3CH}_2]^{2+}$ molecule

Figure S4 Crystal packing in compound **3** showing the C-H····Cl interactions that connect $[Fe^{II}(TPPZ)_2]^{2+}$ in a) a chain along the c axis, b) a 2D sheet on an *ac* plane and c) 3 types of P4AE interactions that connect $[Fe^{II}(TPPZ)_2]^{2+}$ along an *ab* plane (P4AE-A: red, P4AE-B: orange and P4AE-C: blue)

Figure S5 π - π (red broken lines) and C-H···O/N (blue broken lines) interactions connecting the anionic Fe^{II} (orange polyhedra) and Fe^{III} (yellow polyhedra) molecules in compound **4** along the *c* axis.

Figure S6 π ··· π interactions between HS-[Fe^{III}(azp)₂]⁻; Fe1 and LS-[Fe^{III}(TPPZ)₂]²⁺; Fe3 in compound **5** a) Type A and b) Type B of π ··· π interactions yielding a 1D chain along the *b* axis. These 1D chains further connect through other two types of π ··· π interactions *i.e.* c) Type C and D) Type D yielding a 2D layer of HS-[Fe^{III}(azp)₂]⁻; Fe1 and LS-[Fe^{III}(TPPZ)₂]²⁺; Fe3 on an *ab* plane

Figure S7 a) the crystal packing in **5**, the red rectangle highlights the area of discussion b) $\pi \cdots \pi$ interactions between HS-[Fe^{III}(azp)₂]⁻; Fe1 (blue molecules) and LS-[Fe^{III}(azp)₂]⁻; Fe2, viewing along the *a* axis and c) a view along the *b* axis of the same chain as b) to clearly show the $\pi \cdots \pi$ stacks between two types of [Fe^{III}(azp)₂]⁻.

Figure S8 Representation of crystal packing and selected intermolecular interaction in **6** a) C-H··· π interactions between LS-Fe^{II} forming a 1D chain along the *a* axis, b) *pseudo*-2D sheet on an *ab* plane, and c) C-H···N22 interactions relating to MeCN solvent that link sheets of Fe molecules in *pseudo*-3D network

Figure S9 Water solvation occupation in the pocket of Fe molecules in **5** and **6**. Light yellow, yellow and orange polyhedra are Fe1(HS-Fe^{III}), Fe2(LS-Fe^{III}) and Fe3(LS-Fe^{III}), respectively.

Figure S10 Comparison of PXRD patterns between simulated and experimental PXRD patterns of compound **5** and **6**.

Figure S11 Comparison of PXRD patterns between simulated and experimental PXRD patterns of compound a) **3** and b) **4**



Figure S1 a) A dimer of $[Fe(azp)_2]^-$ moieties linking *via* K⁺ cations in **1-ref** [1], and b) the K⁺ ions forming η_4 interactions to aromatic rings of **1**, and c) {K-OH₂···O(phenolate)-Fe} hydrogen-bonding pathways in **1**.

Fable S1 Crystallographic data a	nd structure refinement	for 1, 2, and 3
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	1	2	:	3
	123 K	123 K	100 K	300 K
Formula	$C_{25}H_{26}FeKN_4O_8$	$C_{20}H_{20}Cl_2FeN_{12}O_8$	$C_{52}H_{38}Cl_6Fe_3N_{14}O$	$C_{52}H_{38}Cl_6Fe_3N_{14}O$
Molecular weight / gmol-1	605.45	683.23	1255.21	1255.21
Crystal system	Monoclinic	Monoclinic	Triclinic	Triclinic
Space group	P21/n	P21/n	$P\overline{1}$	ΡĪ
<i>a /</i> Å	9.8080 (1)	10.2987 (2)	13.060 (3)	13.230 (3)
b/Å	16.7440 (3)	7.5907 (2)	14.650 (3)	14.730 (3)
c / Å	15.6654 (2)	17.3764 (4)	15.310 (3)	15.370 (3)
α/°	90	90	78.29 (3)	78.12 (3)
β / °	93.098 (1)	103.506 (2)	78.75 (3)	78.98 (3)
γ / °	90	90	70.90 (3)	71.00 (3)
Cell volume / ų	2568.89 (6)	1320.82 (5)	2684.0 (11)	2746.3 (11)
Ζ	4	2	2	2
Absorption coefficient / mm ⁻¹	6.663	7.086	1.153	1.126
Reflections collected	26214	13697	56946	59154
Independent reflections, R _{int}	5320, 0.1088	2738, 0.0680	15032, 0.0641	15489, 0.0235
Max. and min. transmission	1.00000 and 0.24243	1.00000 and 0.54615	0.966 and 0.933	0.967 and 0.935
Restraints/parameters	3/338	0/196	0/687	0/687
Final R indices [/>2σ(I)]: R ₁ , wR ₂	0.0610, 0.1734	0.0456, 0.1270	0.0666, 0.1915	0.0503, 0.1483
CCDC number	1905259	1905257	1905260	1905261

	1-ref [1]		1
	90 K		123 K
Fe-O1/Å	1.971(5)	Fe1-01/Å	1.928 (3)
Fe-O2/Å	1.890(4)	Fe1-O2/Å	1.878 (2)
Fe-O3/Å	1.928(8)	Fe1-O3/Å	1.925 (2)
Fe-O4/Å	1.949(8)	Fe1-O4 /Å	1.873 (2)
Fe-N1/Å	2.079(6)	Fe1-N1/Å	1.917 (3)
Fe-N2/Å	2.096(8)	Fe1-N3/Å	1.908 (3)
Σ/° (Fe1)	94	Σ/° (Fe1)	36
Θ/° (Fe1)	235	Θ/° (Fe1)	48

Table S2 Selected bond length and octahedral distortion parameters for 1-ref [1], 1, 2, and 3

2		3		
	123 K		100 K	300 K
Fe1-N1/Å	1.9736 (19)	Fe1-N1/Å	1.966 (2)	1.9711 (18)
Fe1-N1 ⁱ /Å	1.9736 (19)	Fe1-N2/Å	1.877 (2)	1.8771 (17)
Fe1-N2/Å	1.972 (2)	Fe1-N3/Å	1.956 (2)	1.9618 (18)
Fe1-N2 ⁱ /Å	1.972 (2)	Fe1-N7/Å	1.968 (2)	1.9683 (18)
Fe1-N3/Å	1.970 (2)	Fe1-N8/Å	1.878 (2)	1.8800 (16)
Fe1-N3 ⁱ /Å	1.970 (2)	Fe1-N9/Å	1.972 (2)	1.9794 (18)
Σ/° (Fe1)	29	Σ/° (Fe1)	80	81
$\Theta/^{\circ}$ (Fe1)	35	$\Theta/^{\circ}$ (Fe1)	240	252

Symmetry code: (i) -x+1, -y+1, -z+1.



Figure S2 Crystal packing of a) **1-ref**, viewing through the *a*-axis. and b) **1**, viewing through the *b*-axis. Red and blue broken lines represent $K^+ \cdots \pi$ and $O \cdots H$ interactions, respectively. Hydrogen atoms omitted for clarity





Figure S3 C-H···O interactions between Fe^{II} moieties and CIO_4^- in compound **2** that link Fe^{II} cationic molecules in a) a sheet on *ac* plane and b) a *pseudo*-3D network, c) showing CIO_4^- anions around the $[Fe^{II}{(pz)_3CH}_2]^{2+}$ molecule

Fe ^{II} molecules with O from ClO₄ on <i>ac</i> sheet	Distance/Å
С3-Н3…О1	2.7019(0)
С10-Н10…О2	2.2773(0)
С10-Н10…ОЗ	2.3585(1)
C4-H4…O1	2.4496(0)
С7-Н7…О4	2.3435(0)
Connect the <i>ac</i> sheet along the <i>b</i> axis	
С2-Н2…О2	2.3722(0)
C8-H8····O4	2.4555(0)
С5-Н5…О1	2.6629(0)
С6-Н6…О3	2.6617(1)

Table S3 Selected intermolecular interactions in compound 2. (Å)



b



Figure S4 Crystal packing in compound **3** showing the C-H····Cl interactions that connect $[Fe^{II}(TPPZ)_2]^{2+}$ in a) a chain along the *c* axis, b) a 2D sheet on an *ac* plane and c) 3 types of P4AE interactions that connect $[Fe^{II}(TPPZ)_2]^{2+}$ along an *ab* plane (P4AE-A: red, P4AE-B: orange and P4AE-C: blue)

Table S4 Selected intermolecular interactions in compound 3. (Å)

	100 K	300 K
Chain along c axis		
C46-H46····N6	2.6725(11)	2.7163(11)
C34-H34…Cl1	2.9321(10)	-
C25-H25…Cl2	2.7432(10)	2.7661(10)
C33-H33···Cl3	2.8920(7)	2.9404(7)
Chain along a axis		
C26-H26····Cl5	2.9461(8)	-
C27-H27…Cl5	2.6863(6)	2.7557(6)
Sheet on ab plane		
P4AE-A		
C15-H15…π	2.514	2.592
π…π	3.377	3.419
P4AE-B		
С3-Н3…π	2.618	2.748
π…π	3.281	3.341
P4AE-C		
С39-Н39…π	2.554	2.645
π•••π	3.443	3.465

	100 K	300 K
A sheet along <i>ab</i> plane		
С3-Н3…π	2.745	2.881
C22-H22…O1	2.7026(8)	-
π-π	3.407	3.496
C10-H10…π	2.901	-
C14-H14…π	2.701	2.775
A chain along the <i>c</i> axis		
C27-H27…N2	2.6158(13)	2.6687(1)
C31-H31…N4	2.6479(9)	2.6970(1)
C32-H32…O1	2.2972(13)	2.3835(1)
π-π	3.916	-
С30-Н30…О2	2.3102(10)	2.3439(1)

Table S5 Selected intermolecular interactions in compound 4. (Å)



Figure S5 π - π (red broken lines) and C-H···O/N (blue broken lines) interactions connecting the anionic Fe^{II} (orange polyhedra) and Fe^{III} (yellow polyhedra) molecules in compound **4** along the *c* axis.





Figure S6 π ··· π interactions between HS-[Fe^{III}(azp)₂]⁻; Fe1 and LS-[Fe^{III}(TPPZ)₂]²⁺; Fe3 in compound **5** a) Type A and b) Type B of π ··· π interactions yielding a 1D chain along the *b* axis. These 1D chains further connect through other two types of π ··· π interactions *i.e.* c) Type C and d) Type D yielding a 2D layer of HS-[Fe^{III}(azp)₂]⁻; Fe1 and LS-[Fe^{III}(TPPZ)₂]²⁺; Fe3 on an *ab* plane





Figure S7 a) the crystal packing of **5**, the red rectangle highlights the area of discussion b) $\pi \cdots \pi$ interactions between HS-[Fe^{III}(azp)₂]⁻; Fe1 (blue molecules) and LS-[Fe^{III}(azp)₂]⁻; Fe2, viewing along the *a* axis and c) a view along the *b* axis of the same chain as b) to clearly show the $\pi \cdots \pi$ stackes between two types of [Fe^{III}(azp)₂]⁻.

Table S6 Selected intermolecular interactions in compound 5. (Å)

	100 K	300 K
Anionic sheet of LS-Fe ^{III} on an <i>ab</i> plane		
C41-H41…N6	2.6327(4)	2.6663(4)
C46-H46…O9	2.5706(5)	2.6390(6)
О9-Н9В…О7	2.0276(4)	1.9941(4)
A chain along the <i>b</i> axis, HS-Fe ^{III} -LS-Fe ^{II}		
Type A		
С84-Н84…π	2.762	2.802
$\pi \cdots \pi$	3.598	3.631
Туре В		
С2-Н2…π	2.806	(2.908)
$\pi \cdots \pi$	3.740	3.784
Between chains of HS-Fe ^{III} -LS-Fe ^{II}		
Туре С		
$\pi \cdots \pi$	3.942	3.926
Type D		
$\pi \cdots \pi$	3.740	3.782
Along the c axis, HS-Fe ^{III} -LS-Fe ^{III}		
$\pi \cdots \pi$	3.580	3.607

 $\pi{\cdots}\pi$ is centroid to centroid distance,

Note for Type C at 100 K, plane to plane distance is 3.62 Å





Figure S8 Representation of crystal packing and selected intermolecular interaction in **6**. a) C-H··· π interactions between LS-Fe^{II} forming a 1D chain along the *a* axis, b) *pseudo*-2D sheet on an *ab* plane, and c) C-H···N22 interactions relating to MeCN solvent that link sheets of Fe molecules in forming a *pseudo*-3D network



Compound 5

Compound 6

Figure S9 Water solvation occupation in the pocket of Fe molecules in **5** and **6**. Light yellow, yellow and orange polyhedra are Fe1(HS-Fe^{III}), Fe2(LS-Fe^{III}) and Fe3(LS-Fe^{III}), respectively.

Table S7 Selected intermolecular interactions in compound 6 (Å)

	100 K	300 K
chains along the <i>a</i> axis		
C94-H94…π	2.667	2.751
C93-H93…N15	2.5698(5)	2.6131(6)
C74-H74…O6	2.3161(4)	2.3932(4)
C75-H75…π	2.964	(3.060)
С9-Н9…π	2.944	3.000
С90-Н90…О4	2.5927(4)	2.6633(4)
chains along <i>b</i> axis		
С50-Н50…π	2.640	2.740
С70-Н70…π	2.864	2.949
C65-H65…π	2.638	2.702
Fe2-Fe2		
C38-H38…O8	2.4982(5)	2.6003(4)
C39-H39…N6	2.6899(6)	2.7467(6)
Fe1-Fe1		
C11-H11…O2	2.5522(5)	2.5398(5)
MeCN solvate		
C76-H76…N22	2.6025(7)	2.7375(8)
C58-H58…N22	2.7025(6)	2.7831(7)

Powder diffraction data

All the experiments were performed at room temperature. The PXRD plot for the bulk sample of **5** does not agrees with that simulated from single crystal data, suggesting the bulk sample is not a pure phase. Bulk sample of compound **5** was re-crystallized in MeCN and yielded compound **6**. Experimental PXRD result of **6** is different from that of **5** but agree very well with the simulated PXRD pattern from the single crystal structure of **6**. This suggests the bulk phase of **6** is pure.



Figure S10 Comparison of PXRD patterns between simulated and experimental PXRD patterns of compound **5** and **6**.



Figure S11 Comparison of PXRD patterns between simulated and experimental PXRD patterns of compound a) 3 and b) 4

References

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