

New radical cation salts based on BDH-TTP donor: two stable molecular metals with a magnetic anion $[\text{ReF}_6]^{2-}$ and a semiconductor with an anion $[\text{ReO}_4]^-$

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1. X-ray Crystallography

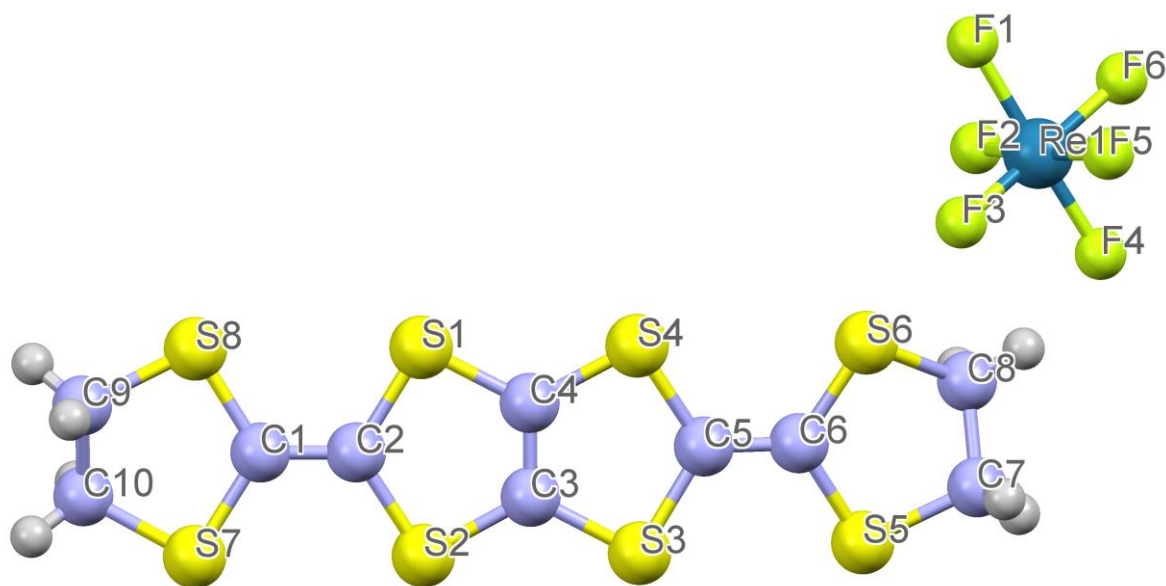


Figure S1. Asymmetric unit and the designations of the radical cation and anion in the crystals κ -(BDH-TTP)₄ReF₆ salt (**1**).

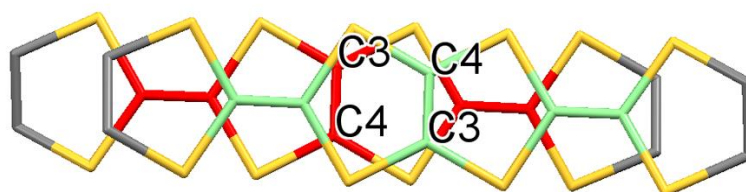


Figure S2. Mutual arrangement of the radical cations in the dimer and the designations of central C=C bond in the molecules κ -(BDH-TTP)₄ReF₆ salt (**1**).

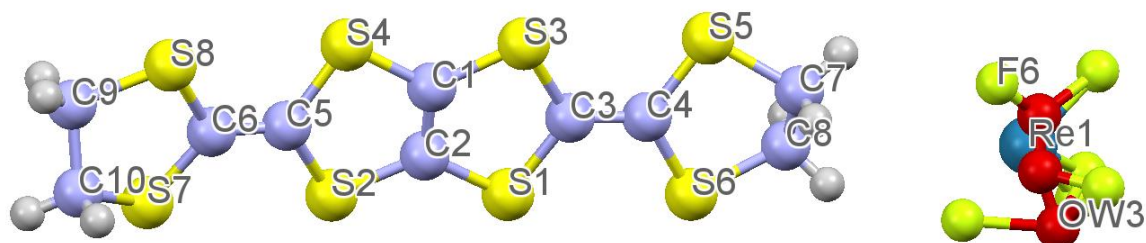


Figure S3. Asymmetric unit and the designations of the radical cation and anion in the crystals κ -(BDH-TTP)₄ReF₆·4.8H₂O (**2**).

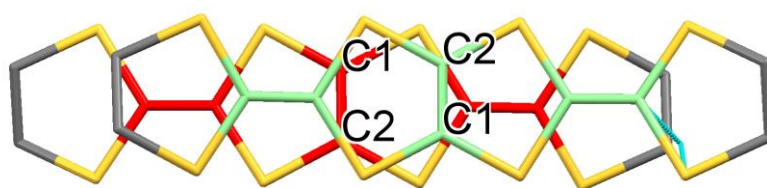


Figure S4. Mutual arrangement of the radical cations in the dimer and the designations of central C=C bond in the molecules κ -(BDH-TTP)₄ReF₆·4.8H₂O salt (**2**).

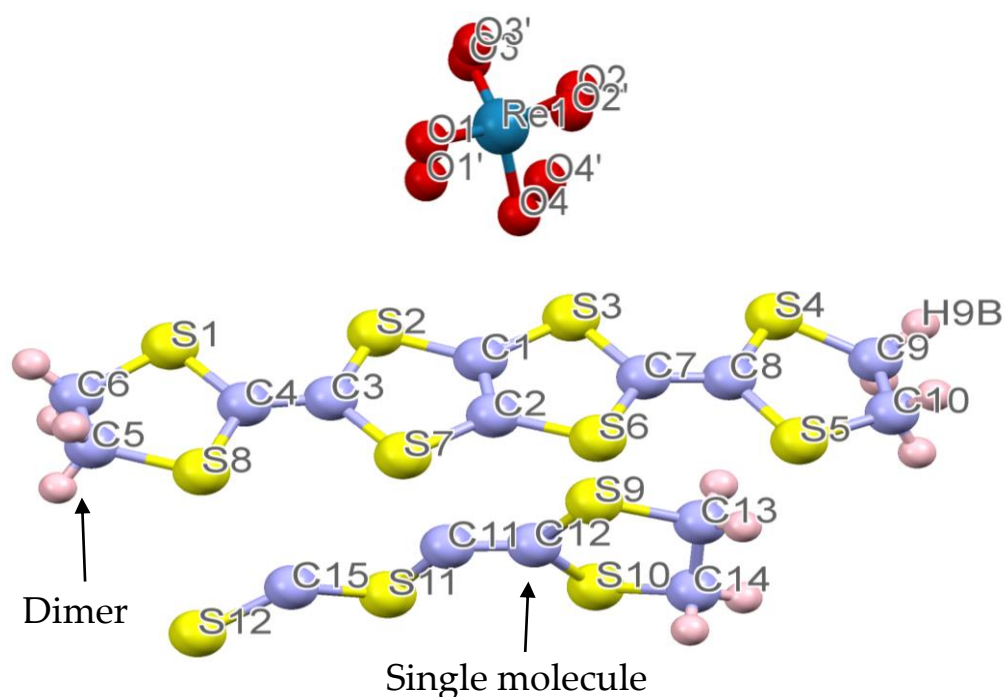


Figure S5. Asymmetric unit and the designations of the radical cations and anion in the crystals pseudo- κ'' -(BDH-TTP)₃(ReO₄)₂ (**3**).

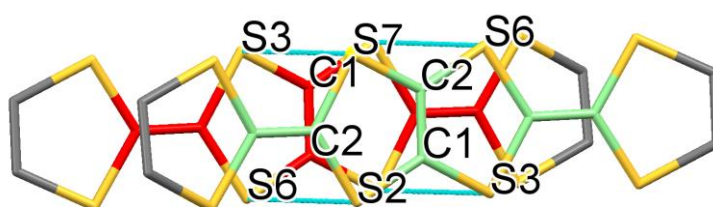


Figure S6. Mutual arrangement of the radical cations in the dimer and the designations of central C=C bond in the molecules of the pseudo- κ'' -(BDH-TTP)₃(ReO₄)₂ salt (**3**). Dashed lines show shortened contacts designated as S2...S6 and S3...S7 between the radical cations in the dimer.

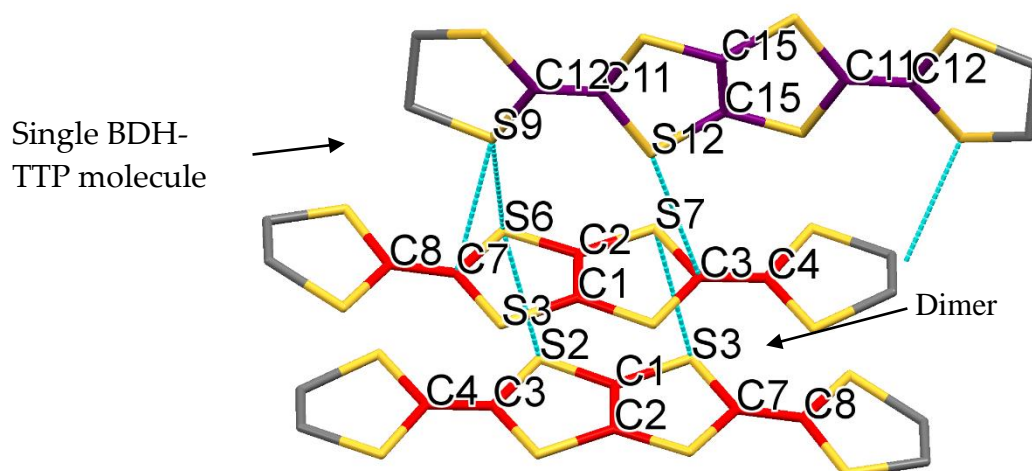


Figure S7. The short contacts between the radical cations inside the dimer and between the dimer and single BDH-TTP molecule.

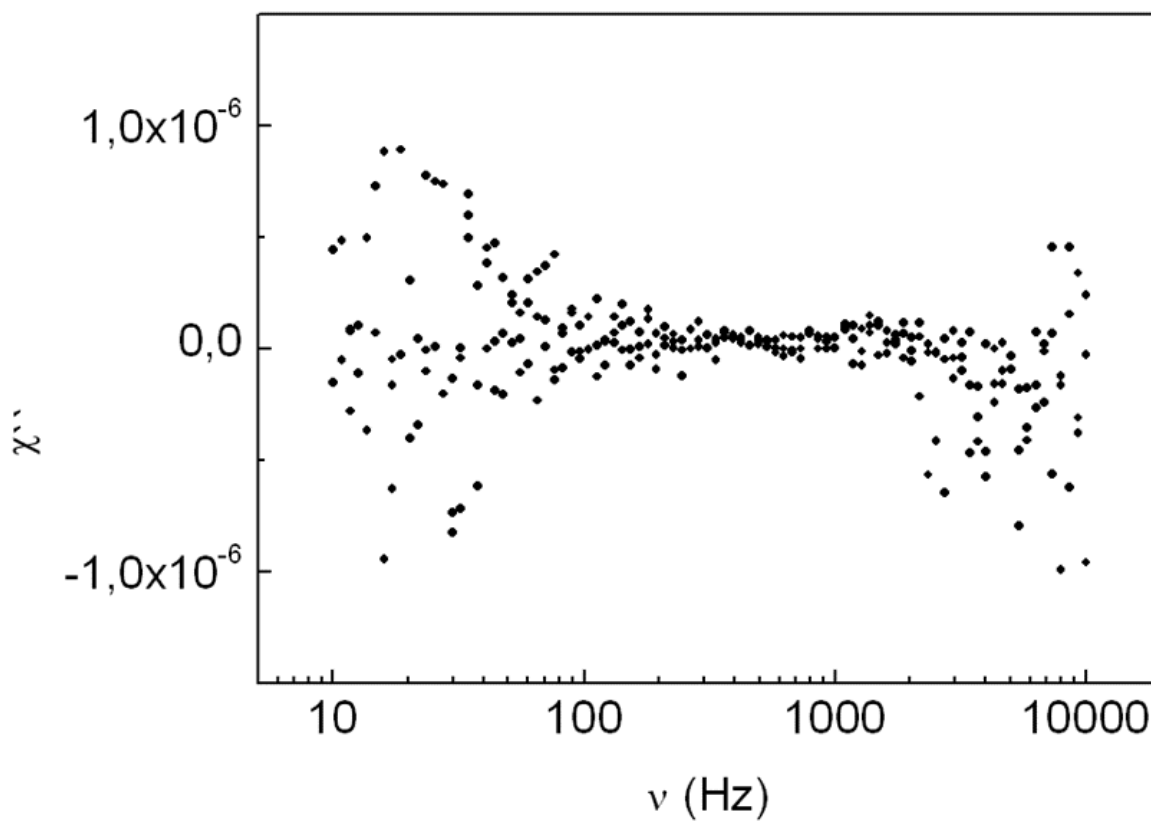


Figure S8. Frequency dependences of the out-of phase (χ'') ac susceptibility for κ -(BDH-TTP)₄ReF₆ salt (**1**) at temperature of 2.0 K in a dc field of 0.4 Tesla

Table S1. Crystallographic data and refined structural parameters for the crystals **1-3**

Identification code	1 salt	2 salt	3 salt
Empirical formula	C ₄₀ H ₃₂ F ₆ Re S ₃₂	C ₄₀ H ₃₂ F ₆ O _{4.80} Re S ₃₂	C ₁₅ H ₁₂ O ₄ Re S ₁₂
Formula weight	1838.78	1915.58	827.17
Temperature	150 K	150 K	100 K
Wavelength, Å	0.71073	0.71073	0.71073
Crystal system, space group	Monoclinic, P2 ₁ /c	Monoclinic, C2/c	Triclinic, P/1
a, Å	20.0098(11)	35.837(2)	8.0239(4)
b, Å	8.0568(4)	7.9677(5)	9.3028(5)
c, Å	10.8367(5)	10.8502(7)	17.3930(10)
α, deg.	90.00	90.00	96.935(5)
β, deg.	92.795(5)	99.028(9)	97.999(5)
γ, deg.	90.00	90.00	110.093(5)
Volume, Å ³	1744.96(15)	3059.8(3)	1187.41(11)
Calculated density, Mg/m ³	1.750	2.079	2.314
Z	4	2	2
Absorption coefficient, mm ⁻¹	2.746	3.142	6.210
Crystal size, mm	0.20×0.15×0.07	0.25 x 0.20 x 0.10	0.15 x 0.07 x 0.05
Theta range for data collection, deg.	3.15 to 34.07	3.09 to 26.07	2.76 to 29.07
Reflections collected /unique	27424 /7158/ [R(int) = 0.0583]	10393 / 3028 [R(int) = 0.0568]	10285 / 6338 [R(int) = 0.0404]
Data / restraints / parameters	7158 / 7 / 226	3028 / 18 / 207	6338 / 44 / 302
GOF	1.100	1.005	1.047
Final R indices [I>2sigma(I)]	R1 = 0.0929, wR2 = 0.2540	R1 = 0.0702, wR2 = 0.1876	R1 = 0.0468, wR2 = 0.0958
R indices (All data)	R1 = 0.1182, wR2 = 0.2699	R1 = 0.0783, wR2 = 0.1947	R1 = 0.0620, wR2 = 0.1081

Table S2. Bond length of Re=O in a tetrahedron [ReO₄]⁻

Bond	Bond length, Å	Bond	Bond length, Å
Re-O1	1.786(7)	Re-O'1	1.700(9)
Re-O2	1.716(6)	Re-O'2	1.730(9)
Re-O3	1.732(7)	Re-O'3	1.734(10)
Re-O4	1.717(7)	Re-O4'	1.771(9)