



Microwave Accelerates the Synthesis of 1-Amidoalkyl-2-Naphthols Using Cu₂(NH₂-BDC)₂(DABCO)-Sal-Co(II) as a Heterogeneous Catalyst ⁺

Leila Panahi and M. Reza Naimi-Jamal *

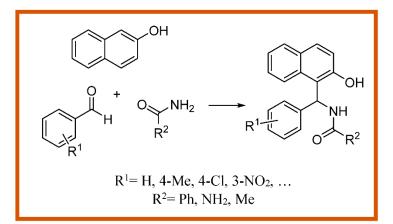
Research Laboratory of Green Organic Synthesis & Polymers, Department of Chemistry, Iran University of Science and Technology, P.O. Box 16846-13114 Tehran, Iran

- * Correspondence: naimi@iust.ac.ir
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Multicomponent reactions have attracted significant attention since they are performed without the need to isolate any intermediate during their process; this diminishes the time and saves both energy and raw materials [1,2]. Derivatives of 1-Amidoalkyl-2-naphthol are of importance, as they can be easily converted to biologically active compounds through amide hydrolysis reaction [3]. Recently, various catalysts have been used to prepare these compounds through a multicomponent reaction. In this research, metal–organic frameworks (MOFs) were utilized as heterogeneous catalysts. They are often crystalline solids consisting of metal ions or clusters coordinated to mostly rigid organic linkers such as aromatic polycarboxylates or polyamines to form one-, two-, or three-dimensional porous structures [4]. We report a green and convenient method for the synthesis of 1-amidoalkyl-2-naphthols from the reaction between β -naphthol, aromatic aldehydes, and amide derivatives under microwave irradiation in the presence of Cu₂(NH₂-BDC)₂(DABCO)-Sal-Co(II) as a modified catalyst through a post-synthesis method. (Scheme 1).

Cu₂(NH₂-BDC)₂(DABCO) was synthesized and then modified with salicylaldehyde and cobalt acetate (II) salt by the ball-milling technique.



Scheme 1. Synthesis of 1-Amidoalkyl-2-naphthol derivatives catalyzed by metal–organic frameworks (MOFs).

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