



## **Supplemental Materials**

## Interfacial Stabilization of a Graphene-Wrapped Cu<sub>2</sub>S Anode for High-Performance Sodium-Ion

## **Batteries via Atomic Layer Deposition**

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**Table S1.** Summary of electrode loading and battery performance in all previously reported literatures and this study<sup>†</sup>

Materials (Composite)	Active loading (%)	Voltage (V)/ Current Density (mA·g <sup>-1</sup> )	Cu2S Capacity (mAh·g <sup>-1</sup> )	CE/Cycles	Electrode Capacity (mAh·g <sup>-1</sup> )	Ref.
Cu <sub>2</sub> S	60.0	0.4-2.6/50	220	~100%/20	132	[1]
Cu <sub>2</sub> S	80.0	0.2-2.5/337	270	~100%/400	216	[2]
Cu2S@ NSC	51.5	0.01-3/100	182.3	~100%/50	93.9	[3]
Cu <sub>2</sub> S@C	70.0	0.01-2.5/5000	290	~100%/5000	203	[4]
Cu9S5@ NSC	48.2	0.01-3.0/100	344.3	~100%/200	166	[5]
Cu <sub>9</sub> S <sub>5</sub> @NC	65.0	0.4-2.6/300	300	99%/500	194.9	[6]
Cu <sub>9</sub> S5-AHP (S impurity)	62.2	0.3–3.2/100	386	94.3%/200	239.9	[7]
Cu <sub>1.8</sub> S	80.0	0.5-2.2/84	250	~100%/1000	200	[8]
Cu2S@ NG	80	0.2–3.0/100	314	99.9%/1000	252	[9]
ALD- Cu2S@NG	80	0.01-3.0/100	374	99.9%/100	300	This work

+ Our electrodes show the highest electrode capacity, regarding to the active material loading.



**Figure S1**. High-magnification SEM images of (**a**) bare and (**b**) Al<sub>2</sub>O<sub>3</sub>-coated (80 cycles) NG and (**c**) bare and (**d**) Al<sub>2</sub>O<sub>3</sub>-coated (80 cycles) Cu<sub>2</sub>S demonstrate the uniformity and conformity of ALD coating on electrode materials. EDX element mapping of Al<sub>2</sub>O<sub>3</sub>-coated (**e**) NG and (**f**) Cu<sub>2</sub>S reveals the content of Al as Al<sub>2</sub>O<sub>3</sub> coating.



Figure S2. Cycling performance of a Na-NG half-cell, delivering 84 mAh g<sup>-1</sup> capacity at 100th cycle.



**Figure S3.** Nyquist plots from EIS measurements of ALD-coated Cu<sub>2</sub>S@NG electrodes with (a) 1, (b) 2, (c) 4, (d) 6, and (e) 8 nm Al<sub>2</sub>O<sub>3</sub> coating during 5 cycles. (f) The physical model used for fitting EIS data.

Electrode	Cycle #	Rs (ohm)	R <sub>SEI</sub> (ohm)	Rct (ohm)	Csei (F)	C <sub>ct</sub> (F)
	1	6.896	21.54	3.132	7.33E-07	7.42E-05
Cu <sub>2</sub> S@NG	2	5.948	177.4	18.56	4.18E-06	0.01408
	5	6.269	1017	89.01	2.21E-05	0.02195
ALD costed (6	1	5.202	14.54	1.454	1.015E-6	0.01569
nm Al <sub>2</sub> O <sub>3</sub> )	2	4.366	81.11	21.48	1.945E-6	0.03081
Cuzowing	5	4.702	420	116.5	3.372e-5	0.02195

Table S2. The fitted EIS data for Cu<sub>2</sub>S@NG and ALD-coated Cu<sub>2</sub>S@NG at 0.01-3 V

## References

[1] J Kim, D Kim, G Cho, T Nam, K Kim, H Ryu, et al., J. Power Sources. 189 (2009) 864-868.

[2] MG Boebinger, M Xu, X Ma, H Chen, RR Unocic, MT McDowell, Journal of Materials Chemistry A. 5 (2017) 11701-11709.

[3] Q Chen, M Ren, H Xu, W Liu, J Hei, L Su, et al., ChemElectroChem. 5 (2018) 2135-2141.

[4] H Li, K Wang, S Cheng, K Jiang, ACS applied materials & interfaces. 10 (2018) 8016-8025.

[5] M Jing, F Li, M Chen, J Zhang, F Long, L Jing, et al., J.Alloys Compounds. 762 (2018) 473-479.

[6] Y Fang, X Yu, XW Lou, Angewandte Chemie. 131 (2019) 7826-7830.

[7] A Qin, H Wu, J Chen, T Li, S Chen, D Zhang, et al., Nanoscale. 11 (2019) 7188-7198.

[8] H Park, J Kwon, H Choi, D Shin, T Song, XWD Lou, ACS nano. 12 (2018) 2827-2837.

[9] J Cai, B Reinhart, P Eng, Y Liu, C Sun, H Zhou, et al., Carbon. 170 (2020) 430-438.