

Editorial

# CO<sub>2</sub> Capture and Sequestration

Diganta Bhusan Das 

Department of Chemical Engineering, Loughborough University, Loughborough LE11 3TU, Leicestershire, UK; d.b.das@lboro.ac.uk; Tel.: +44-1509222509

CO<sub>2</sub> capture and sequestration (CCS) aims to capture carbon dioxide (CO<sub>2</sub>) from CO<sub>2</sub> sources (e.g., fossil fuel power plants), separate the CO<sub>2</sub>, and store it in suitable media. CO<sub>2</sub> can be captured using various technologies, including absorption, adsorption, cryogenic processes, and membrane gas separation [1]. Therefore, accurate selection, design, modelling and optimisation of the processes for CO<sub>2</sub> capture and the tuning of the material properties are essential. There are different methods used for CO<sub>2</sub> sequestration, e.g., (i) geological sequestration that injects different phases of CO<sub>2</sub> into the subsurface [2], (ii) oceanic storage that dissolves CO<sub>2</sub> into an ocean at different depths [3], (iii) the solid-phase reaction of CO<sub>2</sub> with metal oxides to produce stable carbonates with no risk of CO<sub>2</sub> release to the atmosphere [4], and others. The flow, transport, and reaction of CO<sub>2</sub> during CCS and other related matters, such as monitoring critical parameters, are also essential [5].

To address these points, a Special Issue (SI) of *Clean Technologies* has been organised to highlight the recent trends and innovative developments in CCS [6]. Thirteen (13) submissions were received, which underwent a rigorous peer review process. Two papers were declined at the peer review stage, and the remaining eleven papers [7–17] have now been published [6]. The published papers are also being compiled as an edited e-book to be published by MDPI. The papers [7–17] highlight several common and important issues.

Issues related to CCS project development and deployment have been considered by Marshall [7] and Veloso et al. [8]. Marshall [7] has identified that although CCS projects are essential to lower gas emissions, they have not achieved their desired objectives in Australia. To investigate the reasons for this failure, Marshall [7] undertook a historical and social study of the Gorgon gas project in Western Australia, considered one of the world's most significant CCS projects. The study has rightly concluded that CCS's social dynamics must be included in CCS project projections to enhance the accuracy of their expectations, without which the project projections are likely to miss their targets. Veloso et al. [8] emphasised that there are few commercial-scale CCS projects worldwide, and almost all are in the USA and China. Despite the many CCS pilot-scale projects planned in Europe, only two commercial-scale projects operate today. To help improve this situation, the authors have proposed a 'multicriteria regional-scale approach' that can help select the most promising locations in France to deploy CCS pilot-scale projects. Subsequently, the authors have assessed different aspects of CCS technology at the regional scale, including the key economic performance indicators of the CCS project. The authors have rightly concluded that the CCS projects should be located strategically close to potential CO<sub>2</sub> sources in case of the confirmation of proven resources.

Several fundamental issues concerning CCS have also been addressed in the SI. Pfennig and Kranzmann [9] considered cases where CO<sub>2</sub> is compressed to sequester it into deep geological formations. In this process, the corrosion of injection steel pipes can occur due to the contact of the metal with CO<sub>2</sub> and saline water in the geological formation. The published work is supported by the authors' laboratory experiments, which have evaluated corrosion kinetics on stainless steels X<sub>35</sub>CrMo<sub>17</sub> and X<sub>5</sub>CrNiCuNb<sub>16-4</sub> with approximately 17% Cr. The relationship between the corrosion rate and ionic species diffusion into the metal has been studied to determine the longevity of the chosen steels



**Citation:** Das, D.B. CO<sub>2</sub> Capture and Sequestration. *Clean Technol.* **2024**, *6*, 494–496. <https://doi.org/10.3390/cleantechnol6020025>

Received: 27 December 2023

Accepted: 12 March 2024

Published: 16 April 2024



**Copyright:** © 2024 by the author. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (<https://creativecommons.org/licenses/by/4.0/>).

in a CCS environment. In the paper by Abidoye and Das [10], the effects of particle size, carbonation time, curing time and pressure on the efficiency of carbon storage in Portland cement mortar as the media for CCS have been investigated. The authors have shown how carbonation efficiency increases with decreased particle size using data generated in pressure chamber experiments. Overall, these authors show that carbonation efficiency increases with smaller-sized particles or higher-surface areas, carbonation time and higher pressure, but it decreases with hydration/curing time. Quaid and Reza [11] analysed deep eutectic solvents (DESs) for their carbon capture and biogas upgrade applications. In particular, they analysed how the presence of contaminants in biogas may affect the carbon capture by DESs. The behaviour of DESs under different temperatures, pressures, and influences from pollutants has been studied, which suggests that a complex interplay of variables must be understood when choosing DESs for CO<sub>2</sub> absorption for biogas uplifting.

This Special Issue also highlights how the captured CO<sub>2</sub> may be further used to synthesise value-added chemicals. Khokarale et al. [12] demonstrate that industrially important solvents, namely, dimethyl carbonate (DMC) and glycidol, could be synthesised in a combined process using glycerol-derived 1,3-dichloro-2-propanol and captured CO<sub>2</sub> via a metal-free reaction route under mild conditions.

The mathematical modelling applications in CCS have been demonstrated by Deschamps et al. [13] and Khudaida and Das [14]. Deschamps et al. [13] used conservation of mass and energy principles and equations of states to evaluate the performance of a vacuum temperature swing adsorption (VTSA) process for direct CO<sub>2</sub> capture from the air at an industrial scale. A parametric study on the effects of the main operating conditions has been undertaken to assess the performance and energy consumption of the VSTA. The developed approach considers how the lab-scale process could be upscaled to a larger industrial scale. In contrast to lab- or industrial-scale processes, Khudaida and Das [14] attempted to conduct a numerical study on the significance of injecting CO<sub>2</sub> into deep saline aquifers at the scale of geological formations. Several CO<sub>2</sub> injection scenarios and aquifer characteristics have been investigated to enhance current knowledge on the effects of the residual and solubility trapping of CO<sub>2</sub> on the sequestration mechanisms. For example, it was shown how the extent of subsurface heterogeneity increases the residual trapping of CO<sub>2</sub> in geological formations.

Finally, this Special Issue highlighted the critical issues relating to the techno-economic costing of CCS projects. Pieri and Angelis-Dimakis [15] reviewed the current approaches used to quantify CO<sub>2</sub> capture costs. It has been shown that with the existing knowledge in the literature, one can estimate capture costs based on the amount of CO<sub>2</sub> captured and the technologies used in CO<sub>2</sub> capture technology. In the paper by Szima et al. [16], it has been pointed out that increased levelized electricity costs within CCS projects are associated with significant energy penalties involved in CO<sub>2</sub> capture. Consequently, Szima et al. evaluated three CCS approaches that rely on integrated gasification combined cycles: (i) gas switching combustion (GSC), (ii) GSC with added natural gas firing to increase the turbine inlet temperature, and (iii) oxygen production pre-combustion that replaces the air separation unit with more efficient gas switching oxygen production reactors. This comparison has enabled the authors to identify the most promising solution for further development and exploitation in CCS. Reeve et al. [17] carried out a techno-economic analysis of three processes for hydrogen production from advanced steam reforming (SR) of bio-oil as an alternative route to hydrogen with bioenergy with carbon capture and storage (BECCS): conventional steam reforming (C-SR), C-SR with CO<sub>2</sub> capture (C-SR-CCS), and sorption-enhanced chemical looping (SE-CLSR). The analysis concluded that SE-CLSR is comparable to C-SR-CCS in terms of the levelized cost of hydrogen (LCOH).

Overall, it is evident that this Special Issue and the forthcoming e-book cover a diverse range of topics, including some of the most pressing concerns for CCS. I envisage that the authors of the published papers and I, as the guest editor of the SI, can motivate future directions and progress in CCS.

I appreciate the efforts of the authors and referees of all the accepted and declined papers. These contributions have made this Special Issue a true success. Finally, I acknowledge the Editorial Office for supporting this Special Issue and the edited e-book.

**Conflicts of Interest:** The author declares no conflict of interest.

## References

1. Chowdhury, S.; Kumar, Y.; Shrivastava, S.; Patel, S.K.; Sangwai, J.S. A Review on the Recent Scientific and Commercial Progress on the Direct Air Capture Technology to Manage Atmospheric CO<sub>2</sub> Concentrations and Future Perspectives. *Energy Fuels* **2023**, *37*, 10733–10757. [CrossRef]
2. Abidoeye, L.K.; Khudaida, K.J.; Das, D.B. Geological Carbon Sequestration in the Context of Two-Phase Flow in Porous Media: A Review. *Crit. Rev. Environ. Sci. Technol.* **2015**, *45*, 1105–1147. [CrossRef]
3. Ho, H.-J.; Iizuka, A. Mineral carbonation using seawater for CO<sub>2</sub> sequestration and utilization: A review. *Sep. Purif. Technol.* **2023**, *307*, 122855. [CrossRef]
4. Tyagi, P.; Singh, S.; Malik, N.; Kumar, S.; Malik, R.S. Metal catalyst for CO<sub>2</sub> capture and conversion into cyclic carbonate: Progress and challenges. *Mater. Today* **2023**, *65*, 133–165. [CrossRef]
5. Arellano, Y.; Tjugum, S.-A.; Pedersen, O.B.; Breivik, M.; Jukes, E.; Marstein, M. Measurement technologies for pipeline transport of carbon dioxide-rich mixtures for CCS. *Flow Meas. Instrum.* **2024**, *95*, 102515. [CrossRef]
6. Available online: [https://www.mdpi.com/journal/cleantechnol/special\\_issues/CO2\\_capture\\_sequestration](https://www.mdpi.com/journal/cleantechnol/special_issues/CO2_capture_sequestration) (accessed on 19 January 2024).
7. Marshall, J.P. A Social Exploration of the West Australian Gorgon Gas, Carbon Capture and Storage Project. *Clean Technol.* **2022**, *4*, 6. [CrossRef]
8. Veloso, F.M.L.; Gravaud, I.; Mathurin, F.A.; Rhouma, S.B. Planning a Notable CCS Pilot-Scale Project: A Case Study in France, Paris Basin—Ile-de-France. *Clean Technol.* **2022**, *4*, 28. [CrossRef]
9. Pfennig, A.; Kranzmann, A. Understanding the Anomalous Corrosion Behaviour of 17% Chromium Martensitic Stainless Steel in Laboratory CCS-Environment—A Descriptive Approach. *Clean Technol.* **2022**, *4*, 14. [CrossRef]
10. Abidoeye, L.K.; Das, D.B. Carbon Storage in Portland Cement Mortar: Influences of Hydration Stage, Carbonation Time and Aggregate Characteristics. *Clean Technol.* **2021**, *3*, 34. [CrossRef]
11. Quaid, T.; Reza, M.T. Carbon Capture from Biogas by Deep Eutectic Solvents: A COSMO Study to Evaluate the Effect of Impurities on Solubility and Selectivity. *Clean Technol.* **2021**, *3*, 29. [CrossRef]
12. Khokarale, S.; Shelke, G.; Mikkola, J.P. Integrated and Metal Free Synthesis of Dimethyl Carbonate and Glycidol from Glycerol Derived 1,3-Dichloro-2-propanol via CO<sub>2</sub> Capture. *Clean Technol.* **2021**, *3*, 41. [CrossRef]
13. Deschamps, T.; Kanniche, M.; Grandjean, L.; Authier, O. Modeling of Vacuum Temperature Swing Adsorption for Direct Air Capture Using Aspen Adsorption. *Clean Technol.* **2022**, *4*, 15. [CrossRef]
14. Khudaida, K.J.; Das, D.B. A Numerical Analysis of the Effects of Supercritical CO<sub>2</sub> Injection on CO<sub>2</sub> Storage Capacities of Geological Formations. *Clean Technol.* **2020**, *2*, 21. [CrossRef]
15. Pieri, T.; Angelis-Dimakis, A. Model Development for Carbon Capture Cost Estimation. *Clean Technol.* **2021**, *3*, 46. [CrossRef]
16. Szima, S.; del Pozo, C.A.; Cloete, S.; Fogarasi, S.; Álvaro, Á.J.; Cormos, A.; Cormos, C.; Amini, S. Techno-Economic Assessment of IGCC Power Plants Using Gas Switching Technology to Minimize the Energy Penalty of CO<sub>2</sub> Capture. *Clean Technol.* **2021**, *3*, 36. [CrossRef]
17. Reeve, J.; Grasham, O.; Mahmud, T.; Dupont, V. Advanced Steam Reforming of Bio-Oil with Carbon Capture: A Techno-Economic and CO<sub>2</sub> Emissions Analysis. *Clean Technol.* **2022**, *4*, 309–328. [CrossRef]

**Disclaimer/Publisher’s Note:** The statements, opinions and data contained in all publications are solely those of the individual author(s) and contributor(s) and not of MDPI and/or the editor(s). MDPI and/or the editor(s) disclaim responsibility for any injury to people or property resulting from any ideas, methods, instructions or products referred to in the content.