

Supplementary Materials

An Electrode Based on the Manganese Dioxide Nanorods and Hexadecylpyridinium Bromide for the Rosmarinic Acid Voltammetric Assay

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Table S1. Comparison of rosmarinic acid analytical characteristics on various electrodes.

Electrode	Method	Detection limit, M	Linear dynamic range, M	Ref.
Carbon paste electrode/Carbon nanotubes dispersed in chitosan/ DNA	SWV*	1.4×10^{-8}	4.0×10^{-8} – 1.5×10^{-6}	[3]
Carbon nanotube paste electrode with <i>n</i> -octyl-pyridinium hexafluorophosphate	DPV**	1.5×10^{-8}	0.0 – 6.8×10^{-4}	[4]
Carbon paste electrode with [Fe ^{III} Z-n ^{II} (μ-OH)(2-[bis(2-pyridylmethyl)aminomethyl]-6-[(2-hydroxy-5-methylbenzyl)(2-pyridyl-methyl)aminomethyl]-4-methylphenol)](ClO ₄) ₂	SWV	2.3×10^{-6}	2.98×10^{-5} – 3.83×10^{-4}	[5]
Glassy carbon electrode/poly(<i>o</i> -phenylenediamine)/Pt nanoparticles	CA***	5×10^{-7}	$(1-55) \times 10^{-6}$	[6]
Carbon paste electrode/MMIP**** Fe ₃ O ₄ @SiO ₂ @NH ₂ nanoparticles	DPV	8.5×10^{-8}	1×10^{-7} – 1×10^{-4} and 1×10^{-4} – 5×10^{-4}	[7]
GCE/MnO ₂ nanorods–hexadecylpyridinium bromide	DPV	9.7×10^{-9}	2.5×10^{-8} – 1.0×10^{-6} and 1.0×10^{-6} – 1.0×10^{-5}	This work

* Square-wave voltammetry. ** Differential pulse voltammetry. *** Chronoamperometry.

**** Magnetic functionalized molecularly imprinted polymer.