

Supplementary Material: Micromotors of MnO₂ for the Recovery of Microplastics

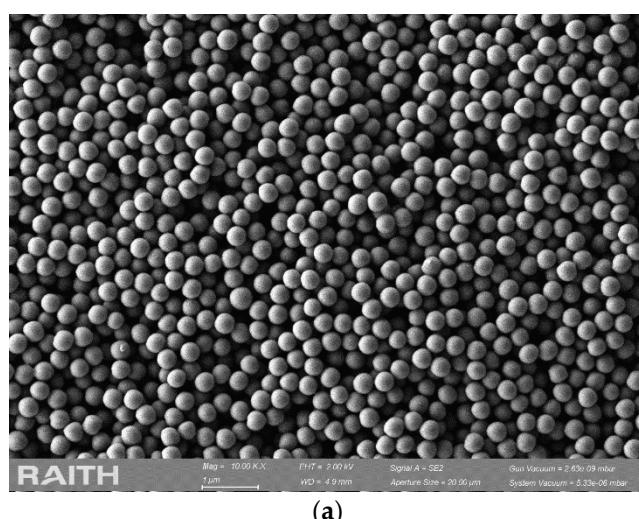
Oscar Cervantes ^{1,2}, Claudia Valtierra-Montiel ³, Laura Sampedro-Plata ¹, Norberto Casillas ², Nieves Menendez ¹ and Pilar Herrasti ^{1,*}

¹ Department of Applied Physical Chemistry, Faculty of Sciences, Autonomous University of Madrid, Francisco Tomás y Valiente 7, 28049 Madrid, Spain; ocervantesa@gmail.com (O.C.); laura.sampedrop@estudiante.uam.es (L.S.-P.); nieves.menendez@uam.es (N.M.)

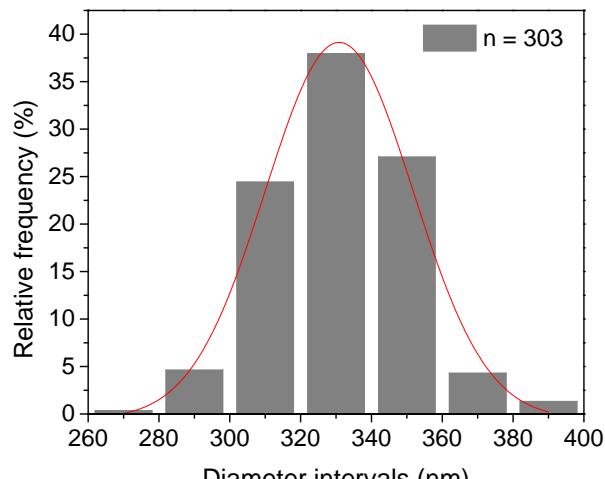
² Department of Chemistry, Center of Exact Sciences and Engineering (CUCEI), University of Guadalajara, Marcelino García Barragán 1421, Col. Olímpica, Guadalajara 44430, Jalisco, Mexico; norberto.casillas@academicos.udg.mx

³ Master's Program in Nanomaterials Science and Technology, Natural and Exact Sciences Division, University of Guanajuato, Noria Alta S/N, Guanajuato 36050, Guanajuato, Mexico; cd.valtierramontiel@hotmail.com

* Correspondence: pilar.herrasti@uam.es



(a)



(b)

Figure S1. FSEM images of the PS particles (a), and their size distribution (b).

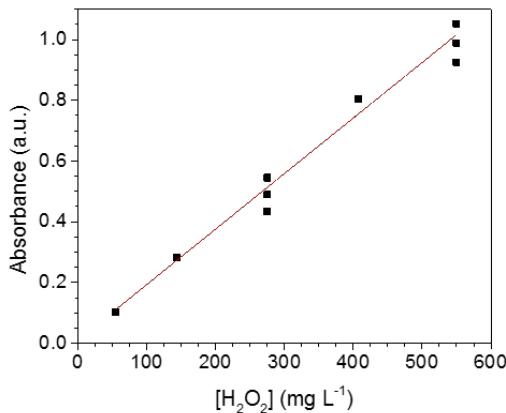


Figure S2. Simple regression and linear fit for Absorbance vs. [H₂O₂].

Table S1. Coefficients and Analysis of Variance for I vs. [H₂O₂].

Parameter	Least Squares Estimate	Standard Error	T Statistic	P-Value
Slope	0.00185488	0.000033572	55.2508	0.0000

Source	Sum of Squares	Df	Mean Square	F-Ratio	P-Value
Model	5.85499	1	5.85499	3052.65	0.0000
Residual	0.0268521	14	0.001918		
Total	5.88184	15			

Table S2. Analysis of Variance for %[H₂O₂] remaining in solution vs. %[H₂O₂]₀.

Source	Sum of Squares	Df	Mean Square	F-Ratio	P-Value
Between groups	27709.4	2	13854.7	39.43	0.0000
Within groups	8432.77	24	351.366		
Total (Corr.)	36142.2	26			

Table S3. Analysis of Variance for TOC/TOC₀ vs. [PS].

Source	Sum of Squares	Df	Mean Square	F-Ratio	P-Value
MAIN EFFECTS					
A:PS (mg L ⁻¹)	0.002951	2	0.0014755	0.16	0.8520
B:[H ₂ O ₂] ₀ (% in wt.)	0.0903406	2	0.0451703	4.95	0.0194
INTERACTIONS					
AB	0.0543381	4	0.0135845	1.49	0.2474
RESIDUAL	0.16432	18	0.00912891		
TOTAL (CORRECTED)	0.31195	26			

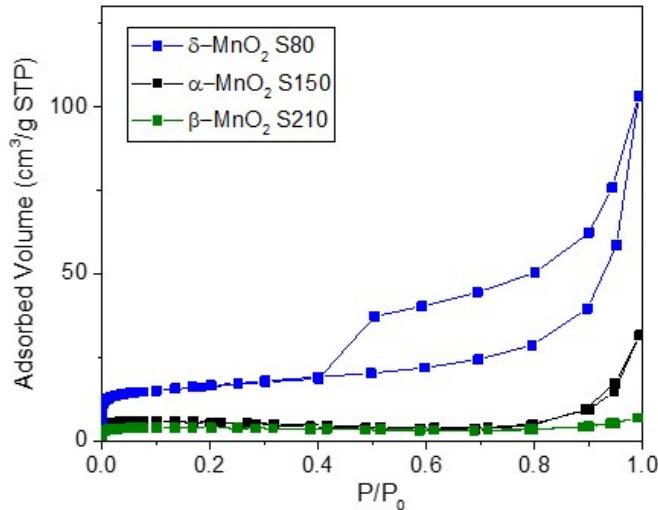


Figure S3. Nitrogen adsorption-desorption isotherms of the three MnO_2 synthesized samples.

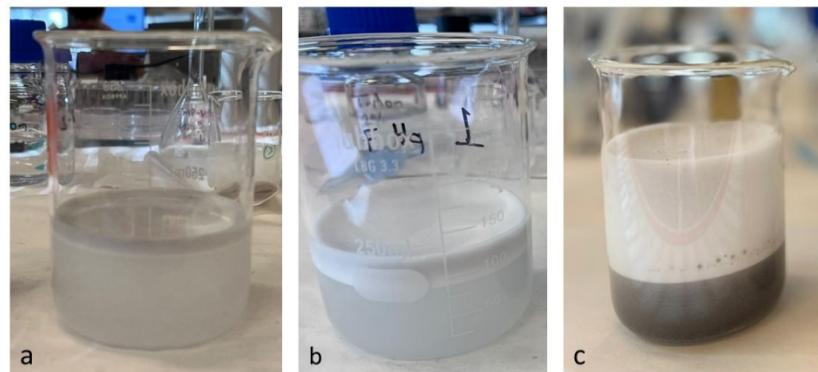


Figure S4. Effect of pH on the amount of foam formed after 5 minutes reaction time, with 0.2 g L^{-1} MnO_2 , 20 mg L^{-1} MP, 1.6% H_2O_2 and 0.01% Triton-X. (a) pH 3, (b) pH 7, and (c) pH 9.

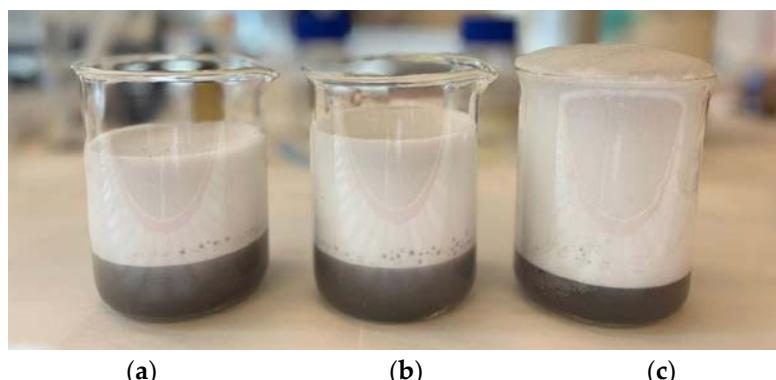


Figure S5. Effect of H_2O_2 concentration on the foam formation 5 min after the reaction started. 0.2 g L^{-1} MnO_2 , 20 mg L^{-1} MP, pH 9 and 0.01% Triton-X. (a) 1.6% H_2O_2 , (b) 3% H_2O_2 and (c) 6% H_2O_2 .