

Review

# Use of Non-Chlorine Sanitizers in Improving Quality and Safety of Marketed Fresh Salad Vegetables

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**Abstract:** The safety of vegetable food is compromised by various factors, including the inefficient or excessive use of sanitizers. Instances of individuals falling ill after consuming raw vegetables have been reported, with outbreaks of diseases caused by pathogens on fresh vegetables becoming increasingly prevalent globally, attracting significant media coverage and impacting the economic viability of vegetable cultivation. Measures to enhance food safety in postharvest horticultural produce involve controlling microbial proliferation and minimizing cross-contamination. Sanitizers were utilized in the food safety arsenal for a variety of purposes, including pathogen elimination and microbe reduction, hand, tool, and vegetable contact surface cleaning, and produce shelf-life extension. Choosing an appropriate sanitizer for all vegetables is difficult due to a lack of knowledge on which sanitizers are ideal for the many types of vegetables grown on farms under different environmental circumstances. Although chlorine-based sanitizers, such as sodium or calcium hypochlorite, have been widely used for the past 50 years, recent research has revealed that chlorine reacts with an organic compound in fresh vegetables to produce trihalomethane, a carcinogen precursor, and as a result, many countries have prohibited the use of chlorine in all foods. As a result, horticulture research groups worldwide are exploring non-chlorine, ecologically friendly sanitizers for the vegetable industry. They also want to understand more about the present procedures in the vegetable business for employing alternative sanitizers, as well as the efficacy and potential dangers to the food safety of fresh salad vegetables. This review paper presents detailed information on non-chlorine sanitizers, such as their efficacy, benefits, drawbacks, regulatory requirements, and the need for additional research to lower the risk of marketed salad vegetable food safety.

**Keywords:** non-chlorine sanitizer; food safety; common fresh salad vegetables

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## 1. Introduction

Vegetables are becoming a more essential “healthy” group than animal proteins, complex carbohydrates, or sweeter alternatives like fruits in today’s diets, as consumers demand safer, fresher, and healthier options [1–3]. Fresh vegetables, including cucumber, tomato, carrot, capsicum, lettuce, green chili, and onion, are frequently used as salad items in Asian countries and are an essential part of a healthy diet. The demand for ready-to-eat fresh produce also means that cutting, peeling, and dicing vegetables is an increasingly common practice, significantly raising food safety risks. Bacteria (such as *Aeromonas* spp., *Bacillus cereus*, *Clostridium botulinum*, *Clostridium perfringens*, *E. coli* (pathogenic and non-pathogenic), *L. monocytogenes*, *Pseudomonas* spp., *Salmonella* spp., *Shigella* spp., *S. aureus*, and *Yersinia enterocolitica*), viruses (such as Hepatitis A, Rotavirus and Norovirus), and parasites (such as *Cryptosporidium parvum*, *Cyclospora cayetanensis*, *Giardia lamblia*, and *Toxoplasma gondii*) are the food safety concern microorganisms. Table 1 provides

a list of the microorganisms responsible and the specific vegetables that contributed to foodborne outbreaks.

**Table 1.** Selected vegetable pathogens linked with outbreaks.

Bacteria	Selected Vegetables	References
Toxin producing Bacteria		
<i>Clostridium botulinum</i>	Carrots	[4]
<i>Escherichia coli</i>	Lettuce, tomato	[5,6]
<i>Staphylococcus aureus</i>	Lettuce, tomato, and carrot	[7–9]
Non-toxin producing Bacteria		
<i>Listeria monocytogenes</i>	Lettuce, tomato	[10,11]
<i>Salmonella</i> spp.	Lettuce, tomato	[12,13]
<i>Shigella</i> spp.	Lettuce, salad vegetables	[14]
<i>Yersinia enterocolitica</i>	Carrots, cucumber, lettuce, and tomatoes	[15–17]
Viruses		
Hepatitis A and Norovirus	Lettuce	[18,19]
Protozoa		
<i>Cryptosporidium</i> spp. and <i>Cyclospora</i> spp.	Lettuce	[20,21]

Plant pathogenic fungi are mainly responsible for the spoilage of vegetables, known as quality deterioration. The initial microbial loads and diversity of microorganisms present on fresh vegetables at harvest vary depending on factors such as horticultural practices, employee health and hygiene, field or farm settings, farm management regimens, and so on [22,23]. In addition, not all microorganisms are capable of proliferating on produce commodities. Several microbial species, however, can break the protective barriers plants possess, grow, and cause spoilage of these products, while others can enter the plant tissue only through wounds, but once the protective cover has been compromised, the chances of spoilage increase exponentially [24]. Common spoilage bacteria including *Aeromonas* spp., *Pectobacterium* spp., *Xanthomonas* spp., *Pseudomonas* spp., and *Erwinia* spp. are mainly responsible for bacterial soft rot, while *Bacillus* spp. and lactic acid bacteria cause the spoilage of vegetables, known as quality deterioration. Common fungi species, including *Rhizopus stolonifer*, *Fusarium oxysporum*, *Aspergillus niger*, *Fusarium solani*, and *Geotrichum candidum*, were found to be associated with the spoilage of horticultural crops [25,26]. Table 2 summarizes the microorganisms causing postharvest deterioration of various fresh vegetables.

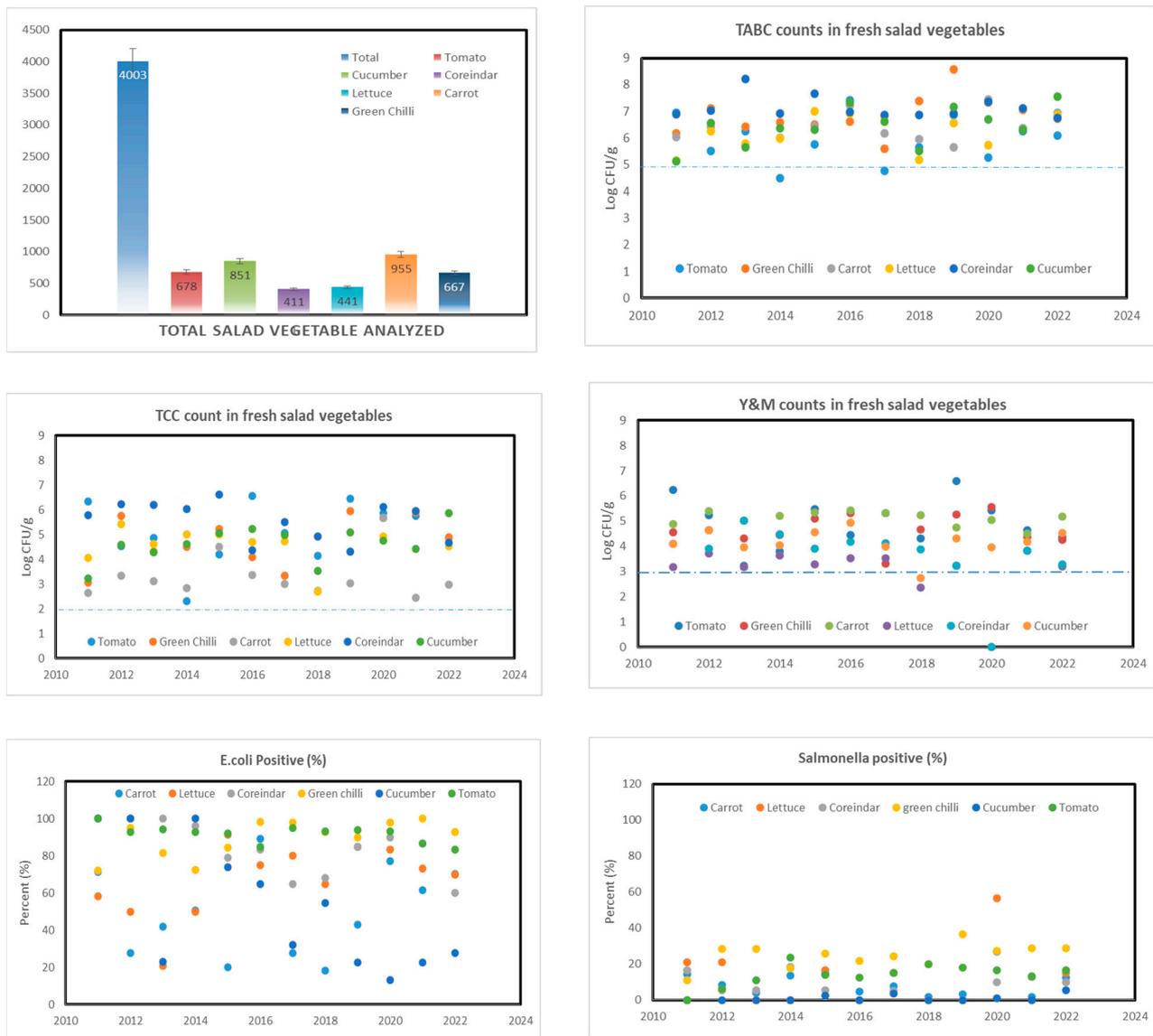
**Table 2.** Summary of microorganisms causing postharvest spoilage of selected fresh produce.

Bacteria/Fungi	Type of Spoilage	Selected Vegetables	References
<i>Aeromonas</i> and <i>Pectobacterium</i> spp.	bacterial soft rot	Lettuce	[27]
Lactic acid bacteria	spoilage	Tomato	[28]
<i>Xanthomonas</i>	leaf spot and bacterial spot on tomato	Lettuce and Tomato	[29]
<i>Pseudomonas</i>	bacterial spot	Carrot, Lettuce, and Tomato	[29]
<i>Erwinia</i>	soft rot	Carrot, Cucumber, Lettuce, and Onion	[29]
<i>Bacillus coagulans</i>	spoilage	Cucumber, Onion, and Tomato	[29]

Table 2. Cont.

Bacteria/Fungi	Type of Spoilage	Selected Vegetables	References
<i>Thielaviopsis basicola</i>	black root rot	Carrot	[30]
<i>Pythium</i>	cottony rot	Cucumber	[29]
<i>Phytophthora</i>	Buckeye rot	Tomato	[31]
<i>Penicillium (blue mold) and Rhizopus</i>	spoilage	Cucumber and Tomato	[29]
<i>Aspergillus niger</i>	black rot	Onion and Tomato	[32]
<i>Sclerotinia</i>	white rot, white mold	Carrot, Lettuce, and Tomato	[24]
<i>Geotrichum</i>	sour rot	Carrot, Lettuce, Onion, and Tomato	[33]
<i>Collectotrichum</i>	Anthraxnose	Capsicum, Cucumber, Onion, and Tomato	[34]
<i>Rhizopus</i> spp.	storage rot, rhizopus rot	Capsicum, Carrot, Cucumber, and Tomato	[35]
<i>Botrytis</i> spp.	neck rot, grey mold	Carrot, Cucumber, Lettuce, Onion, and Tomato	[29]
<i>Fusarium and Alternaria</i> spp.	soft rot, dry rot, and black rot	Capsicum, Carrot, Cucumber, Onion, and Tomato,	[27]

All of these vegetables have the potential to pick up pathogenic microorganisms while being harvested on the field, handled afterward, processed and packaged in the pack house, or transported and distributed to stores. The agroecology of the geographical locations, genetic diversity, agronomic practices, and environmental responses in different farm production stages all influence the level of microbial contamination. Following the postharvest process, human activity and the environmental responses of the vegetable-packing plants can also increase the risk of contamination. The overwhelming majority of published studies stated that contamination occurs primarily prior to harvesting, either from animals, domestic or wild; through contaminated manure, sewage, irrigation water, wastewater from livestock operations; during harvesting, transport, processing, distribution, pack house, and marketing; or even at home [36–39]. Vegetable food safety risks in connection to agroecology during farm production and the environment of vegetable-packaging facilities during postharvest operations can be significant, and this field of study is just beginning. As mentioned in the literature, vegetables can also become infected at the retail or consumer levels, and this can happen through direct contamination, contact with contaminated soil or water, symptomatic and asymptomatic employees, or cross-contamination with other foods [40]. This risk, however, is completely determined by regional food safety culture, which is also a new area of study. Figure 1 depicts the typical annual average microbiological quality and safety indicator microorganisms discovered in different salad vegetables in Bangladesh from 2010 to 2022, where almost all the parameters are above the acceptable limit set by the regulatory agencies.



**Figure 1.** Indicators of microbiological quality (total aerobic bacteria, total coliform bacteria, total yeast, and mold) and safety (*E. coli* and *Salmonella*) were distributed on an annual basis in salad vegetables such as tomato, cucumber, carrot, coriander leaf, green chili, and lettuce in Bangladesh between 2011 and 2022 [41].

## 2. Use of Sanitizers

The most common reasons to use sanitizer are: (1) to eliminate any bacteria or pathogens and lower microbial load; (2) to control microbiological hazards from human sources; (3) to maintain a safe environment and control product cross-contamination; (4) to reduce fungi's ability to propagate; (5) to disinfect and/or clean the processing facility, machinery, and any surfaces that come into contact with the vegetable crop; (6) to make sterile water; (7) to sterilize water-holding vessels; (8) to maintain the product's cleanliness; and (9) to increase shelf life [42].

Although various microbial agents are available for sanitizing fresh produce, their efficacies differ, and none can eliminate pathogens without compromising sensory quality. Recent research has also demonstrated that chlorine is ineffective at reducing pathogens, and the creation of chlorine byproducts harms human health [43]. As a consequence, there is considerable interest in developing a chlorine substitute that is both safer and better for the environment. On the other hand, various non-chlorine sanitizers including, calcined

calcium, hydrogen peroxide, ozone, peroxyacetic acid, organic acid, natural antimicrobial agents, or combinations of these and other physical sanitization techniques have been seen in the literature for washing fresh salad vegetables to improve the quality and safety of these vegetables; however, it is vital to remember that the efficiency of any sanitizer, chlorine or non-chlorine, depends on factors such as concentration, contact time, and application method. Proper handling and sanitation practices are crucial regardless of the type of sanitizer used to ensure the safety and quality of fresh salad vegetables. Additionally, regulatory guidelines and specific industry requirements should be considered when selecting and implementing sanitization methods. The disadvantages and advantages of chlorine-based sanitizers and non-chlorine sanitizers are summarized in Table 3.

**Table 3.** The disadvantages and advantages of chlorine-based sanitizers and non-chlorine sanitizers.

Factors	Chlorine-Based Sanitizers	Non-Chlorine Sanitizers
Reduced Chemical Residue	Can leave behind chemical residues that may affect the taste and safety of vegetables	Can effectively sanitize vegetables without leaving harmful residues
Gentler on Produce	Can sometimes be harsh on delicate salad vegetables, potentially causing discoloration or off-flavors	Often gentler and less likely to affect the appearance or taste of the vegetables
Effective Pathogen Reduction	Can be effective in pathogen reduction but the creation of chlorine byproducts harms human health	Can effectively reduce pathogens and no known byproducts which harm human health are produced
Organic Compliance	Does not adhere to organic standards	Align more closely with organic certification requirements
Versatility	May not be suitable for certain sensitive vegetables	Often be used across a wider range of vegetables without adverse effects
Reduced Environmental Impact	Can have environmental implications, including the formation of harmful disinfection byproducts	No environmental implications or impact and no formation of harmful byproducts
Consumer Preferences	May not be preferred by some consumers because they cause harm to the environment	May be preferred due to perceived health and environmental benefits

Generally, the ideal sanitizing agent should possess two key qualities: an adequate level of antimicrobial activity and little to no impact on the product's sensory quality. Consumer demands for safe and ecologically friendly products would increasingly influence sanitizer selection for both domestic and international use. There is a lack of information on suitable alternatives, more environmentally friendly sanitizers, their efficacy on various products, and their market acceptance. The US Environmental Protection Agency (EPA) has registered and authorized nearly 4000 antimicrobial sanitizers, including 275 distinct active components [44]. Choosing the best sanitizer for specific vegetables (cucumber, tomato, carrot, capsicum, lettuce, green chili, and onion) might be challenging, given the lengthy list. The most commonly used sanitizers in the food business are chlorine-based, which include sodium or calcium hypochlorite, acidified chlorite, electrolyzed water, and others; they have been tested on a variety of fresh produce and are effective against a wide range of pathogens [45]. However, chlorine can interact with organic matter in the natural environment to produce halogenated byproducts like trihalomethanes or haloacetic acids [46]. Because these byproducts are carcinogenic and unfriendly to the environment, using chlorine to wash fresh vegetables has been prohibited in countries such as Belgium, Switzerland, and the Netherlands for safety reasons [47]. As a result, Table 4 includes a number of substitute sanitizers proven to improve the safety and quality of salad vegetables, such as calcinated calcium, aqueous chlorine dioxide (ClO<sub>2</sub>), ozone, hydrogen peroxide, peroxyacetic acid, organic acid, and plant bioactive compounds or essential oils. Table 4 describes each non-chlorine sanitizer, their dose level, advantages, and disadvantages.

**Table 4.** List of various applicable non-chlorine sanitizers for washing fresh salad vegetables.

Sanitizer	Allowable Levels	Advantages	Disadvantages	Rinse Step	Additional Comments	References
Calcinated calcium (CCa)	Use 1 g powder/10 L water (giving a concentration of 0.01% and pH about 11) and 40 s wash.	Calcinated calcium (CCa) is safe and eco-friendly; produced from marine waste (scallop shells).	If not dissolved properly, may contribute to residues on the vegetable surface. Can be affected by organic load in wash water.	Yes	Effective at lower dose, high pH, and less contact time. Available and cheap.	[48,49]
Hydrogen peroxide (H <sub>2</sub> O <sub>2</sub> )	Typical concentration used: (0.04–2%). Environmentally friendly. Declared GRAS by FDA.	Breaks down easily, no harmful by-products. Higher temperatures could produce better reduction.	Higher concentrations can cause browning or bleaching in certain products and can be corrosive and irritating. Unstable and degrades fast.	No	Commercially available at 31–70% but 30–50% is most common. Dilute (3%) solutions are available to consumers.	[50,51]
Ozone (O <sub>3</sub> )	No regulatory limit but typically used at 2–10 ppm for up to 5 min. Activity reduced in presence of organic load.	Declared GRAS by FDA. Environmentally friendly. Effective at low concentrations. No harmful end products.	Has to be generated on-site; unstable and highly reactive. Corrosive to equipment. OSHA requirements on employee exposure.	No	Solubility in water increases at lower temperatures and pH. Does not work as well at higher pH.	[52–56]
Organic acids (acetic acid, citric acid, lactic acid, tartaric acid, oxalic acid, ascorbic acid, and phytic acid)	1% oxalic acid, 0.03% phytic acid, 0.5% CA, 0.5% lactic for 2 min or 2% acetic acid for 15 min.	Organic acids have been used as sanitizers for fresh produce. The FDA recognizes organic acids as GRAS.	Their usefulness against microorganisms is generally low and requires high concentrations for long periods. Sensory quality might also be affected with 5–15 min treatment.	No	Effective at higher conc., depends on water quality and costly. Antimicrobial efficacy is dependent on the microorganism strain and acid type.	[57–59]
Aqueous chlorine dioxide (ClO <sub>2</sub> )	ClO <sub>2</sub> 5 mg/L, 60 s overhead spray and brush roller system at 25 °C.	Easy to handle; inexpensive. It can be used in the form of a spray, or by immersion or washing. Concentration and contact can be maintained. Easy to adopt in industrial washing lines.	Produce surface properties can affect ClO <sub>2</sub> accessibility to microbes. Residual moisture after the water rinsing can promote microbial growth.	Yes	Not suitable for dried foods. Relatively less effect on microbial Internalization.	[60,61]
Peroxyacetic acid	Strong oxidizing agent. Use 80 ppm–150 ppm; 2 min on fruits and vegetables. Can work well in cooler temperatures.	Environmentally friendly and less corrosive to equipment. Works at a wide range of pH values and temperatures. Effective against biofilms. Not as sensitive to organic load as chlorine.	Costs more than chlorine. Vinegar odor. Loses its effectiveness in the presence of metals (copper). High concentrations damage produce and can shorten shelf life.	No	Store in a well-ventilated area. Concentrated peroxyacetic acid is a safety hazard.	[62–66]
Plant extracts (bioactive compounds)	Grape stem extract, 2.5% solution; 2 min, and dried for 30 min could reduce pathogens by 2.0–4.0 log CFU/g in lettuce. Oregano aqueous extract for 2 min. Green tea extract 60% GTE for 5 min. Essential oil (Cypriot oregano), 0.1% for 10 min.	The antibacterial activity could be due to the damage of cytoplasmic membranes, inhibition of synthesis of nucleic acids, cell wall components, and cell membranes [67,68].	Fewer effects than chemical sanitizers and non-economic efficiency. Unpleasant aroma. Longer durations.	No	Store in dark place. Since sensitive to light, volatile nature. Higher concentration (0.5%) results in softer fruits.	[69–72]

GRAS = generally recognized as safe; GTE = green tea extract.

### 2.1. Calcinated Calcium (Green Agrowash®)

Calcinated calcium is a pyrolysis product of waste shell aggregate, which is baked into a fine, odorless, natural, environmentally beneficial, and biodegradable powder. Japan's Specifications and Standards for Food Additives (JFSA) has approved using this shell powder with microparticles in food. This powder is sparingly soluble in water and possesses antibacterial and antifungal activities, which has proven effective in killing bacteria and fungi and removing contaminants from the fruit and vegetable surfaces [73–75]. One gram of powder in 10 L tap water, or (0.01% solution), is recommended for a 40–60 s wash. Followed by a clean water wash, this was able to eliminate pathogens and remove other contaminants from the surface of the vegetables. However, using more than the suggested concentration can leave a white stain on stainless steel or glass surfaces. Since calcinated calcium is made from natural ingredients, it is readily accessible, inexpensive, does not harm the environment, and is not toxic to humans or animals.

### 2.2. Hydrogen Peroxide

Hydrogen peroxide, commonly known as hydrogen dioxide, can be used to sterilize fruits and vegetables as a liquid or gas. It is considered “generally recognized as safe” or GRAS by the FDA and EPA and is environmentally friendly because it breaks down to oxygen and water. The organic load in the wash water has an impact on it, but pH has no effect. Fresh vegetables can be sanitized with hydrogen peroxide at concentrations no more than 59 ppm, according to FDA approval. The usage of hydrogen peroxide in conjunction with acetic acid (PAA) has increased recently, as opposed to using it alone. Hydrogen peroxide causes cells to die by altering osmotic pressure, leading to the loss of cell wall integrity. This compound is cheap, easy to prepare, fast acting for bacteria, and can kill spores. Hydrogen peroxide must be used cautiously because it is unstable in water, highly allergenic, and loses potency if not stored properly.

### 2.3. Ozone

Ozone is also thought to have high antimicrobial activity and has a high level of reactivity and penetrability [76]. Ozone generation/production has lower operating costs, and it is GRAS. It was investigated how the microbial population and fresh-cut produce's qualitative features were affected by ozonated water at varying concentrations and contact times [77]. Ozone breaks down into non-toxic compounds and does not produce any dangerous disinfection byproducts. Aqueous ozone is less effective than gaseous ozone against both pathogenic and non-pathogenic bacteria. However, gaseous ozone may be harmful, poisonous, and reactive in this state [78–80].

### 2.4. Organic Acids

Fresh fruit has been sanitized using organic acids like citric acid, acetic acid, and lactic acid as well as combinations with phosphoric acid and sulfuric acid. These substances cannot stain or emit odors and are not corrosive to stainless steel, making them more natural ingredients in food. On the negative side, yeasts, fungi, Gram-positive bacteria, and others are not destroyed by organic acids. Although organic acids are deemed GRAS by the FDA, their effectiveness against microorganisms is typically low and they are needed at high concentrations for extended periods of time. Fresh and freshly cut vegetables have been sanitized using organic acids and acid compound sanitizers. In order to maintain the physical and chemical properties of many fresh-cut products, and to stop microbial development, organic acids are crucial sanitizers. According to [81], decontaminating leaves of some particular vegetables with 5% citric acid resulted in a noticeably lower microbial count than washing them with water. The ideal form of organic acid had no negative effects on flavor or taste, and it had no negative effects on the ecosystem. Fresh-cut fruit can have their shelf life extended by citric acid because it prevents the food quality from deteriorating and the spread of disease. However, using these acids at greater concentrations may lead to a quality loss in some freshly cut leafy vegetables due to off-odors and texture damage.

It has also been investigated how the order of citric acid and ethanol treatment affects the quality and microbial reduction in organic vegetables. As a result, an organic acid-based disinfectant has been developed using a combination of technologies in place of chlorine.

### 2.5. Chlorine Dioxide

Fresh fruit can be effectively protected from bacterial, fungal, and viral contamination by using the oxidizing gas  $\text{ClO}_2$  [82,83].  $\text{ClO}_2$  is effective across a wide pH range (pH 3–8) and does not create any toxic byproducts or change the nutritive or olfactory qualities of food products. However, using  $\text{ClO}_2$  to wash fresh products in gaseous and aqueous forms has benefits and drawbacks, which are listed in Table 5.

**Table 5.** Chlorine dioxide ( $\text{ClO}_2$ ) application in the aqueous and gaseous form: advantages and disadvantages.

<b>Aqueous <math>\text{ClO}_2</math> [45]</b>	
<b>Advantages</b>	<b>Disadvantages</b>
Easy to handle, inexpensive	Produce surface properties can affect $\text{ClO}_2$ accessibility to microbes
It can be used in the form of a spray, by immersion or washing	Cross-contamination of wash water
Concentration and contact can be maintained	Water rinsing is required after the treatment
Easy to adopt in industrial washing lines	Residual moisture after the water rinsing can promote microbial growth
	Not suitable for dried foods
	Relatively less effective on microbial internalization
<b>Gaseous <math>\text{ClO}_2</math> [77]</b>	
<b>Advantages</b>	<b>Disadvantages</b>
Higher antimicrobial activity	Needs onsite generation
It can be applied as a batch treatment or a continuous treatment	Needs technical knowledge
High accessibility to microbes, irrespective of surface barriers	Laborious to perform and expensive
No water rinsing is required after the treatment	Explosive at higher concentrations
It can impact microbial internalization	Challenging to maintain concentration and contact time
No issue of cross-contamination of wash water	Challenging to implement at the industry scale

### 2.6. Natural Plant Extracts

Natural goods are increasingly being looked into as alternatives to conventional sanitizing agents in the washing processes for fresh produce. Essential oils (EOs) and hydrosols from aromatic plants are examples of natural plant extracts that are generally accepted as safe (GRAS) for use in the food industry and are also covered by EC Regulation No. 1334/2008 on flavorings and certain food ingredients with flavoring properties for use in and on foods [84,85]. Numerous EOs and other natural extracts, such as sage, Greek oregano, eucalyptus, and rosemary, have been used to preserve fresh produce and barely processed vegetables [86–89]. Furthermore, no variations in lowering *E. coli* O157: H7 and total coliforms in lettuce and spinach were discovered after washing with water and tannin solutions [90]. Washing spinach and lettuce samples in aqueous oregano extract for two minutes reduced *E. coli* O157:H7 counts by 2.1 log CFU  $\text{g}^{-1}$  and 3.7–4.0 log CFU  $\text{g}^{-1}$ , respectively, when coupled with Citrox<sup>®</sup> (a product containing citric acid and phenolic compounds) [91]. These findings suggested that plant extracts can successfully decrease the pathogenic load in fresh vegetables. Edible coatings with natural antimicrobial agents are becoming more popular as possible treatments to lessen the adverse effects of processing fresh vegetables. However, using natural edible coatings for freshly cut

vegetables has not attracted attention, and vegetable businesses have yet to wash or preserve fresh-cut vegetables using a natural antimicrobial agent due to fewer side effects than chemical sanitizers and non-economic efficiency. Plant extracts should also be kept in the dark because they are typically volatile and light-sensitive. Vegetables may become softer when plant extracts are used in greater concentrations (0.5%). The washing of fresh organic produce offered at a higher price can be achieved with natural detergents. The need for natural food preservation techniques, such as using natural antimicrobials and their combination with other obstacles, without adverse effects on the consumer or the environment, has been brought on by consumers' increasing demand for fresh and freshly cut produce [92]. Essential oils are natural antimicrobial agents; however, it is practically difficult to use these oils because of their hydrophobic, volatile, and unstable nature [93].

## 2.7. Green Tea Extract

Green tea extract (GTE; 60%) was shown by [94] to exhibit a rise in antiviral activity with increasing pH. The cytoplasmic membrane damage, nucleic acid synthesis suppression, cell wall component inhibition, and cell membrane damage could contribute to the antibacterial activity [95,96]. Temperature, concentration, and contact time all impacted GTE's reaction. For lettuce and spinach, using 60% GTE also successfully lowered the bacterial count by 1.5 logs after 30 min of exposure. A non-economic and occasionally greater dosage of GTE may result in an unpleasant odor and soften the vegetables because it requires longer times, less effectiveness, and a higher concentration than chemical sanitizers (Table 6).

**Table 6.** Efficacy of non-chlorine-based sanitizers in reducing bacterial pathogens from fresh salad vegetables' surfaces.

Vegetables	Non-Chlorine Sanitizers (Conc. and Contact Time)	Microorganisms	Maximum Reduction (log CFU/g)	Complete Reduction/Number of Samples	References
Lettuce	Peracetic acid (PAA) (100 mg/L; 5 min at 65 rpm)	<i>Escherichia coli</i> O157:H7,	2.2	0/6	[89]
		<i>Salmonella typhimurium</i> DT104	6.8	6/6	
		<i>Listeria monocytogenes</i> ,	2.4	0/6	
	Lactic acid (2%; 5 min at 65 rpm)	<i>Escherichia coli</i> O157:H7,	1.7	0/6	[89]
		<i>Listeria monocytogenes</i> ,	1.7	0/6	
	Calcinated calcium (0.01% for 40–60 s)	<i>Escherichia coli</i>	2.1	3/3	[48]
	Hydrogen peroxide (H <sub>2</sub> O <sub>2</sub> ) (2% for 90 s)	<i>Escherichia coli</i> O157:H7	4.3	0/3	[97]
		<i>Salmonella enteritidis</i>	4.3	0/3	
	Aqueous ozone (O <sub>3</sub> ) (3 ppm for 5 min)	<i>Escherichia coli</i> O157:H7,	5.9	0/5	[56]
		<i>Listeria monocytogenes</i>	5.9		
	ClO <sub>2</sub> (3 ppm, 5 min)	<i>Escherichia coli</i> O157:H7,	5.8	0/5	[56]
		<i>Listeria monocytogenes</i>	6.0		
	Plant extract (grape stem extract, 25 mg/mL)	<i>Escherichia coli</i> O157:H7	0.7	0/5	[98]
		<i>Salmonella enterica</i>	1.0	0/5	
		<i>Listeria monocytogenes</i>	0.8	0/5	

Table 6. Cont.

Vegetables	Non-Chlorine Sanitizers (Conc. and Contact Time)	Microorganisms	Maximum Reduction (log CFU/g)	Complete Reduction/Number of Samples	References	
Tomato	PAA at 100 mg/L (5 min@65 rpm) (laboratory scale)	<i>Escherichia coli</i> O157:H7,	5.5	3/6	[69]	
		<i>Salmonella typhimurium</i> DT104	6.8	6/6		
		<i>Listeria monocytogenes</i>	2.4	0/6		
	Lactic acid (2%) (5 min@65 rpm)	<i>Escherichia coli</i> O157:H7,	2.4	0/6		
		<i>Salmonella typhimurium</i> DT104	4.8	0/6		
		<i>Listeria monocytogenes</i>	2.3	0/6		
	ClO <sub>2</sub> (5 mg/L, 60 s) (commercial scale)	<i>Salmonella</i> spp.	4.9	0/15	[77]	
		PAA (80 mg/L, 60 s) (commercial scale)	<i>Salmonella</i> spp.	5.5	15/15	[89]
	Calcinated calcium (1 min, 0.01%)		<i>Escherichia coli</i> O157:H7	7.6	0/3	[48]
			<i>Salmonella</i> spp.	7.4		
			<i>Listeria monocytogenes</i>	7.5		
		H <sub>2</sub> O <sub>2</sub> (5% for 2 min, 60 °C)	<i>Salmonella</i> spp.	2.6	0/3	[97]
<i>Escherichia coli</i>			1.4			
<i>Listeria monocytogenes</i>			2.5			
Aqueous O <sub>3</sub> (0.45 ppm for 10 min)	<i>Salmonella</i> spp.	4.5	0/6	[99]		
Green tea extract (60%; 5 min)	<i>Escherichia coli</i>	5.66 ± 0.1	3/3	[68]		
	<i>Salmonella enteritidis</i>	5.23 ± 0.12	0/3			
Cucumber	Peracetic acid (PAA) (0.5%)	<i>Salmonella typhimurium</i>	2.66 ± 0.20	0/12	[74]	
		<i>Listeria monocytogenes</i>	1.28 ± 0.35			
	Lactic acid (2%)	<i>Salmonella typhimurium</i>	2.14 ± 0.26	0/12	[74]	
		<i>Listeria monocytogenes</i>	0.75 ± 0.43			
	Calcinated calcium (0.01% for 1 min)	<i>Escherichia coli</i>	3.62 ± 0.1	3/3	[48]	
		H <sub>2</sub> O <sub>2</sub> (0.5% for 2 min)	<i>Salmonella typhimurium</i>	2.63 ± 0.19	0/12	[100]
	<i>Listeria monocytogenes</i>		1.16 ± 0.40			
	Aqueous O <sub>3</sub> (2% for 5 min)	-	-	-	-	
	ClO <sub>2</sub> (100 ppm)	<i>Escherichia coli</i>	2.61 ± 0.1	0/5	[101]	
	Green tea extract (60%; 5 min)	<i>Salmonella enterica</i>	2.0 ± 0.1	0/4	[52]	
<i>Listeria monocytogenes</i>		2.07 ± 0.1				

Table 6. Cont.

Vegetables	Non-Chlorine Sanitizers (Conc. and Contact Time)	Microorganisms	Maximum Reduction (log CFU/g)	Complete Reduction/Number of Samples	References
Carrot	Peracetic acid (PAA) (40 ppm, 1 min)	<i>Escherichia coli</i>	0.5	0/4	[102]
		<i>Salmonella</i> spp.	1.5		
		<i>Listeria monocytogenes</i>	0.5		
	Lactic acid (0.1%; 5 min)	<i>Escherichia coli</i> O157:H7	0.4	0/5	[103]
	Calcinated calcium (0.01% for 1 min)	<i>Escherichia coli</i>	0.5	3/3	[48]
		<i>Salmonella</i> spp.	0.5	3/3	
	H <sub>2</sub> O <sub>2</sub> (1.5% for 90 s)	<i>Escherichia coli</i>	0.8	0/3	[104]
	Aqueous O <sub>3</sub> (16.5 mg/L)	<i>Escherichia coli</i> O157:H7	1.85	0/3	[105,106]
	ClO <sub>2</sub> (20 mg/L)	<i>Escherichia coli</i> O157:H7	3.0	0/3	
	Plant extract	-	-	-	-
Green chili	Peracetic acid (PAA)	-	-	-	-
	Lactic acid	-	-	-	-
	Calcinated calcium (0.01% for 1 min)	<i>Escherichia coli</i>	0.5	0/3	[48]
		<i>Salmonella</i> spp.	0.5	0/3	
	H <sub>2</sub> O <sub>2</sub> (0.5% for 2 min)	<i>Escherichia coli</i>	0.5	0/3	[107]
		<i>Salmonella</i> spp.	0.5	0/3	
	Aqueous O <sub>3</sub>	-	-	-	-
	ClO <sub>2</sub>	-	-	-	-
	Plant extract	-	-	-	-
	Coriander leaf	Peracetic acid (PAA)	-	-	-
Lactic acid		-	-	-	-
Calcinated calcium (0.01% for 1 min)		<i>Escherichia coli</i>	1.6 ± 0.1	3/3	[48]
H <sub>2</sub> O <sub>2</sub>		-	-	-	-
Aqueous ozone (O <sub>3</sub> ) (6% for 30 min)		<i>Escherichiacoli</i>	2.5	0/3	[108]
		<i>Salmonella typhimurium</i>	2.7	0/3	
Aqueous ClO <sub>2</sub>		-	-	-	-
Plant extract		-	-	-	-

### 3. Conclusions

The majority of research has found that irrigation or wash water poses the greatest threat to vegetable food safety, followed by how the crop is treated during the postharvest process. Microbial contamination is a major cause of postharvest losses in fresh vegetables and washing them in water reduces the quantity of microbes on their surface by one or two logs (one log reduction being a 10-fold reduction). The quantity of total dissolved solids (such as soil, dirt, and debris) in the water, the water temperature, the quality of the incoming water (such as pH and mineral content), the contact time with the produce, and the texture of the produce are all variables that influence the effectiveness of the

sanitizer in vegetables (smooth or rough surface). As chlorine-based sanitizers pose risks to human health, non-chlorine alternatives have emerged as viable options. Washing fresh vegetables with non-chlorine sanitizers has shown significant reductions in resident aerobic bacteria, total coliform bacteria, yeast, and mold populations. Moreover, these sanitizers effectively reduce pathogenic bacterial populations to below detectable limits. Research findings suggest that all non-chlorine sanitizers can enhance the safety and quality of fresh vegetables. However, among them, 0.01% calcinated calcium emerges as the optimal and safest choice due to its biodegradable nature, organic characteristics, and environmentally friendly properties. Additionally, the majority of sanitizer effectiveness studies have only been conducted in laboratories; it is crucial to validate sanitizers in actual workplace settings. Furthermore, when determining the precise procedure (amount and mode of sanitizer delivery) to handle the vegetables, appropriate sanitation should consider the aforementioned factors.

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