



## Article

# Quality Assessment of Loquat under Different Preservation Methods Based on Physicochemical Indicators, GC–MS and Intelligent Senses

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**Abstract:** To explore the effects of different preservation methods on the quality of loquat after fresh-keeping treatment, various preservation techniques were employed. These included natural preservation (NP), vacuum freezing preservation (VFP), vacuum at room temperature preservation (VP) and freezing preservation (FP). The quality assessment involved analyzing the effects of these preservation methods using physicochemical indexes, a colorimeter, an electronic nose (E-nose), an electronic tongue (E-tongue) and gas chromatography–mass spectrometry (GC–MS). The results showed minor differences in loquat quality under different preservation methods, with sensory scores ranging from 55 to 78 and  $\Delta E$  values ranging from 11.92 to 18.59. Significant variations were observed in moisture content (ranging from 53.20 g/100 g to 87.20 g/100 g), calorie content (ranging from 42.55 Kcal/100 g to 87.30 Kcal/100 g), adhesion (ranging from 0.92 to 1.84 mJ) and hardness (ranging from 2.97 to 4.19 N) ( $p < 0.05$ ). Additionally, the free amino acid content varied from 22.47 mg/g to 65.42 mg/g. GC–MS analysis identified a total of 47 volatile flavor substances in varieties of loquats, including 13 aldehydes, 9 esters, 6 ketones, 2 acids, 3 alcohols, 2 phenols, 3 pyrazines, 1 furan and 8 other substances. The relative content of aldehydes was significantly higher than that of other chemicals. The VFP and FP samples exhibited higher aldehyde content compared to the NP and VP samples. Moreover, Orthogonal Partial Least Squares-Discriminant Analysis (OPLS-DA) revealed 18 marked compounds that could differentiate between 5 loquat species. Analysis using E-nose and E-tongue indicated significant changes in the olfactory and gustatory senses of loquats following preservation. The VFP samples demonstrated the most effective preservation of loquat quality with minimal impact. This study provides some theoretical guidance for the home preservation of loquats.

**Keywords:** loquat; preservation method; physicochemical indicators; GC–MS; intelligent senses



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## 1. Introduction

Loquat (*Eriobotrya japonica* Lindl.), a perennial fruit tree belonging to the *Eriobotrya* genus in the Rosaceae family, possesses both medicinal and edible properties [1–3]. However, it was not resistant to storage after picking and was highly perishable and deteriorated, attributed to its thin peel, tender flesh and fragile stalk [4,5]. Loquats undergo damage to a certain extent during transportation and storage, greatly reducing their quality, and in order to address these problems, the selection and research of preservation methods were crucial [6].

In recent years, research on loquats has deepened, highlighting the importance of understanding the effect of commonly used preservation methods on fruit quality to ensure long-term preservation. For example, Chen et al. [7] investigated that ultrasonic treatment combined with peroxyacetic acid (PA) treatment could reduce decay and maintain the quality of loquat fruits. They concluded that the combined treatment of UT and PA was a useful method to reduce the decay and browning of loquat fruits stored at room temperature. Yu et al. [8] investigated the effect of the phytosulfonic acid  $\alpha$ -gene on sugar, proline and polyamine metabolism during the cold storage of loquat to explore the susceptibility of loquat to cold damage under low stress. Peng et al. [9] used low-temperature regulation to mitigate the cold damage in loquats, regulating the glycine content and energy status of loquats, and prolonging the shelf life of loquats. After picking loquat, complex physicochemical changes will occur in the storage process, and preservation methods on its quality characteristics have a great impact. Therefore, appropriate preservation methods can reduce the changes in loquat quality brought about by prolonged storage time [10].

Currently, analytical methods for food flavor evaluation include a human sensory evaluation and intelligent sensory evaluation. An intelligent sensory evaluation, such as gas chromatography–mass spectrometry (GC–MS), E-nose and E-tongue, etc., offers advanced approaches. GC–MS was a widespread and effective method for analyzing volatile compounds in food samples based on solid-phase microextraction, gas-phase separation and mass spectrometry [11–14]. E-nose technology was widely used in the food industry, effectively addressing the shortcomings of traditional manual evaluation, such as subjectivity and repeatability [15,16]. E-tongue technology was a new detection technology that is efficient and convenient to identify and analyze the taste of samples [17,18]. The physicochemical parameters were combined with intelligent senses to analyze the similarities and differences of loquat samples under different preservation methods [19]. The integration of various technologies provides more comprehensive and scientific information about the aroma of food [20].

Therefore, in this study, different preservation methods commonly used in home storage such as natural preservation (NP), vacuum frozen preservation (VFP), vacuum preservation (VP) and frozen preservation (FP) were applied to loquats. The sensory evaluation served as the primary method, supplemented by colorimeter analysis, textural property assessments, amino acid and organic acid analyses, GC–MS, E-nose and E-tongue evaluations, to comprehensively assess the changes and flavor differences in loquats. The study aimed to compare the effects of different preservation methods on the quality of loquats, offering theoretical guidance for identifying optimal home preservation methods for loquats.

## 2. Materials and Methods

### 2.1. Material and Sample Handling

The “Longquan No.1” loquats used for the experiment were purchased from the local fruit supermarket in Chengdu City, Sichuan Province, China, and these loquats were selected based on their uniform size, consistent appearance and fresh undamaged condition. About 18 (1 kg) ripe loquats were used for the preservation treatments at a time. The average diameter and weight of the loquat were measured at  $4.56 \pm 0.18$  cm and  $56.34 \pm 2.71$  g, respectively.

Fresh loquats of a relatively uniform size, weight and freshness were chosen for the study. Four different preservation methods were used: natural preservation (NP) at  $20 \pm 5$  °C, vacuum freezing preservation (VFP) at 4 °C, vacuum preservation (VP) at  $20 \pm 5$  °C and frozen preservation (FP) at 4 °C for seven days, with fresh loquat serving as the control group. (The temperature range during the ripening time of loquat was  $20 \pm 5$  °C).

To prepare the samples, the loquats were peeled, split, cored and diced. Approximately 1 kg of loquat pieces were then placed in a wall-breaker and pulped for 2 min at a speed of 32,000 r/min to obtain a uniformly colored and flavored slurry sample. The resulting

loquat slurry could be used for physicochemical indicators, GC–MS and intelligent sensory determination, or stored at  $-20\text{ }^{\circ}\text{C}$  for later use [21].

## 2.2. Determination of Physicochemical Indicators

### 2.2.1. Sensory Evaluation

An evaluation panel consisting of 15 experienced food sensory evaluators was selected to assess the color, aroma, taste and completeness of fresh and preserved treated whole loquat fruits, with a maximum score of 100 points. (7 males and 8 females; Age: 20–30: 5, 31–40: 5, 41–50: 5). The sensory evaluation criteria are detailed in Table 1 [22].

**Table 1.** Loquat sensory evaluation scoring criteria (100 points).

Project	Grading Criteria	Sensory Scores/Points
Color (30 points)	Bright, and the peel was golden yellow	21–30
	Relatively bright, and the peel was golden yellow	11–20
	Dull and the epidermis was yellowish-brown	1–10
Aroma (30 points)	Bouquet, and the sweet and sour were moderate	21–30
	Bouquet, and the sweet and sour were better	11–20
	Faint scent, and the sweet and sour were poor	1–10
Taste (20 points)	Soft and juicy	16–20
	Soft and dry	6–15
	Shriveled	1–5
Completeness (20 points)	Round and complete	16–20
	Full and relatively complete	6–15
	Damaged	1–5

### 2.2.2. Colorimetric Analysis

A new fully automatic colorimeter (C-P3, Zhejiang Guangnian Zhixin Instrument Co., Ltd., ShaoXing, China) was used to determine the color values of different loquat samples. Healthy loquat flesh was selected randomly for measurements, three loquats were chosen for each sample and three replicate experiments were conducted to obtain  $L^*$ ,  $a^*$  and  $b^*$  data and take the average value [23]. Ensure that the sample covers the probe completely and no excess light passes through.  $\Delta E$  represents the difference between two hues, and the larger the value, the greater the color difference between the experimental group and the control group. The formula for  $\Delta E$  is shown in Equation (1).

$$\Delta E = [(L_0^* - L^*) + (a_0^* - a^*) + (b_0^* - b^*)]^2]^{(1/2)} \quad (1)$$

### 2.2.3. Calories Analysis

Calory Answer (CA-HM, JWP, Tokyo, Japan) was used to determine the energy, carbohydrates, proteins and fat contents in loquat samples subjected to different preservation methods. First, a homogeneous and fine loquat slurry sample was selected, and then 5 g of the same mass of loquat slurry sample was sent for determination each time. The measurements were repeated three times for each sample [24].

### 2.2.4. Textural Properties Analysis

Texture profile analysis (TPA) was conducted using a texture analyzer (TMS-Pro, FTC, Los Angeles, CA, USA). The pre-test speed was set to  $1.00\text{ mm s}^{-1}$ , the test speed was  $2\text{ mm s}^{-1}$ , the post-test speed was  $10\text{ mm s}^{-1}$ , the measurement distance was 10 mm and the trigger value was 10 gf [25]. The hardness, toughness, chewability, adhesion and resilience of loquats were determined and analyzed by the P2 probe under the above measurement conditions. The measurements were repeated five times for each sample and the results were averaged.

### 2.2.5. Determination of Free Amino Acids (FAAs)

The free amino acid content was determined using an automated amino acid analyzer (S433D, Sykam, Munich, Germany). The samples were pretreated by hydrolysis with sulfosalicylic acid (Sinopharm Chemical Reagent Co., Ltd., Shanghai, China). A 20 g sample of loquat pulp was mixed with ultrapure water (resistivity  $\geq 18.2 \Omega$ ) at a mass ratio of 1:1. After homogenization, filtration and centrifugation, 1 mL of the supernatant was filtered through a 0.22  $\mu\text{m}$  microporous membrane (Sigma-Aldrich Trading Co., Ltd., Shanghai, China) into a sample bottle for testing. The analytical conditions were as follows: a PEEK analytical column (4.6 mm  $\times$  150 mm, 7  $\mu\text{m}$ ) with 10% cross-linking; column temperature ranging from 20 to 99  $^{\circ}\text{C}$ ; reactor temperature of 130  $^{\circ}\text{C}$ ; detection wavelengths of 570 and 440 nm; analytical time of 57 min; ninhydrin reagent (Merrill Biochemicals and Technology Co., Ltd., Shanghai, China) at a flow rate of 0.25 mL/min; and an injection volume of 40  $\mu\text{L}$  [26]. The measurements were repeated three times for each sample and the results were averaged.

### 2.2.6. Determination of Organic Acids

HPLC (E2695, Waters, Milford, CT, USA) was used to determine the organic acids. The instrument was equipped with a 717+ autosampler, which injected 20  $\mu\text{L}$  at a time, and a detector (PDA) set at a UV wavelength of 210 nm. The separation was carried out at 25  $^{\circ}\text{C}$  using an Agilent Zorbax SB-C18 column (5  $\mu\text{m}$ , 4.6 mm  $\times$  250 mm). The mobile phase consisted of 20 mmol/L  $\text{NaH}_2\text{PO}_4$  (prepared with 0.01 mol/L of potassium dihydrogen phosphate, pH 2.60) at a flow rate of 1.00 mL/min [27]. The measurements were repeated three times for each sample and the results were averaged.

## 2.3. Analysis of Volatile Compounds by GC–MS

Volatile compounds were analyzed by gas chromatography–mass spectrometry (GC–MS) (SQ680, PerkinElmer Instrument Co., Ltd., Waltham, MA, USA); 2.0 g of the slurry samples were placed in 15 mL vials, sealed and numbered, and each sample was subjected to three parallel experiments. The automated headspace sampling method with an injection volume of 100  $\mu\text{L}$ , an incubation time of 15 min, an incubation temperature of 60  $^{\circ}\text{C}$ , an injection needle temperature of 85  $^{\circ}\text{C}$  and an incubation speed of 500 rpm was used [28]. The GC conditions included a DB-WAX capillary column (30 m  $\times$  0.25 mm, 0.25  $\mu\text{m}$ ), with an initial column temperature of 30  $^{\circ}\text{C}$ , ramped at 4  $^{\circ}\text{C}/\text{min}$  to 150  $^{\circ}\text{C}$ , followed by 5  $^{\circ}\text{C}/\text{min}$  to 240  $^{\circ}\text{C}$ , holding for 5 min. The inlet temperature was set at 250  $^{\circ}\text{C}$ , with helium as the carrier gas at a flow rate of 1.0 mL/min, and the injection was carried out in a non-split mode. The MS conditions comprised a mass spectrometry interface temperature of 250  $^{\circ}\text{C}$ , ion source temperature of 230  $^{\circ}\text{C}$ , electron ionization source with an energy of 70 eV, detector voltage of 0.2 kV, mass spectral scanning range  $m/z$  of 35–500 and a solvent delay time of 3 min [29].

## 2.4. Analysis of Intelligent Senses

### 2.4.1. Analysis of E-Nose

Olfactory analysis was conducted using the E-nose odor analyzer (FOX4000, Alpha MOS, Toulouse, France). A 2.0 g slurry sample was placed in a 20 mL headspace injection vial, sealed and numbered, and then tested using a PEN 3.5 E-nose. The E-nose operated with a carrier gas flow rate and injection flow rate of 0.30 L/min, a sensor cleaning time of 180 s, a waiting time of 15 s before sampling, and a sample testing time of 60 s [30].

### 2.4.2. Analysis of E-Tongue

Taste analysis was performed using the  $\alpha$ -ASTREE E-tongue (Alpha MOS, Toulouse, France). Processed loquat pulp samples, weighing 50 g, were placed in a multifunctional wall-breaker. Then, 200 g of ultrapure water was added, crushed and mixed for 2 min. The mixture was filtered through four layers of gauze, and the filtrate was used for the

E-tongue test [31]. The E-tongue was activated, calibrated and diagnosed under conditions that ensured the reliability and stability of the collected data [32].

### 2.5. Statistical Analysis

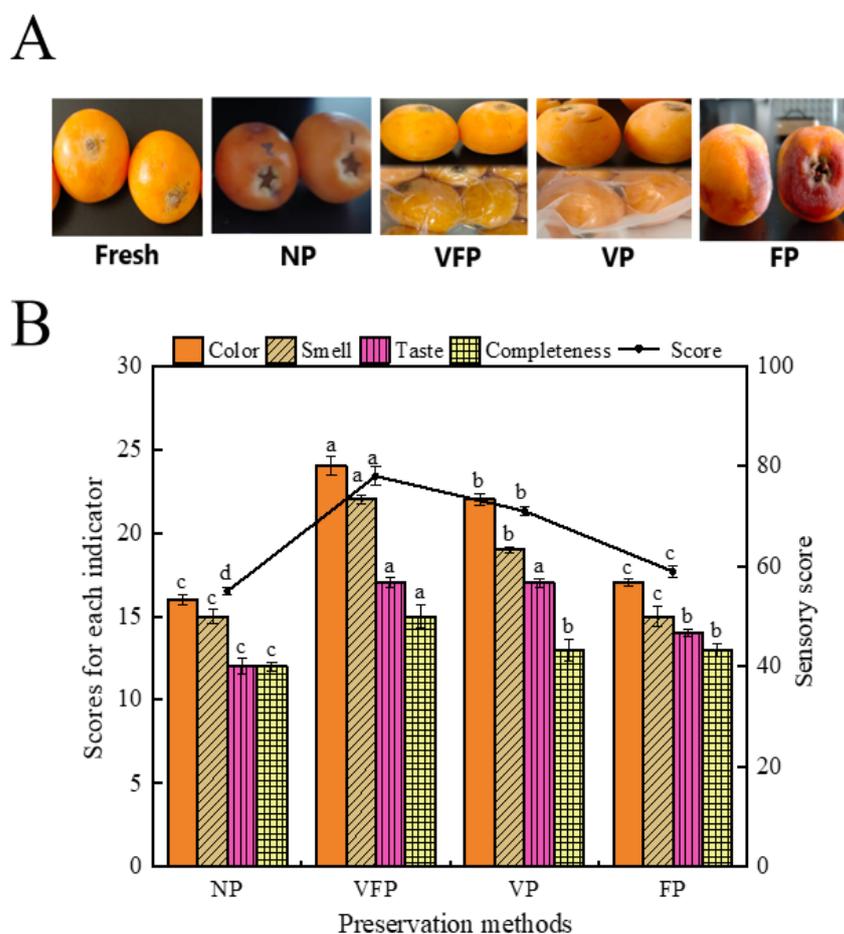
All data were analyzed using IBM SPSS Statistics 27.0 (SPSS Inc., Chicago, IL, USA), and  $p < 0.05$  was considered statistically significant. The data are expressed as the mean  $\pm$  standard deviation of multiple measurements. Origin 2021 (Origin Lab Corporation, Northampton, MA, USA) was used for radar plot analysis and histogram analysis. PCA analysis was performed by GenesCloud Tools ([www.genescloud.cn](http://www.genescloud.cn). Access date: 1 April 2024). Correlation analyses such as OPLS-DA and VIP were conducted by SIMCA 14.1 (MKS Umetrics, Umea, Sweden).

## 3. Results and Discussion

### 3.1. Physicochemical Properties Analysis

#### 3.1.1. Sensory Analysis of Loquat Samples

Figure 1A shows the treatment of loquat under the fresh, NP, VFP, VP and FP conditions. As depicted in Figure 1, VFP better preserved the color and appearance of the loquat. Under this condition, the color of the treated loquats closely resembled that of fresh ones, with minimal morphological changes and little damage [33]. The FP specimens exhibited a darker color, due to the surface oxidation of loquats upon exposure to air after freezing [34].

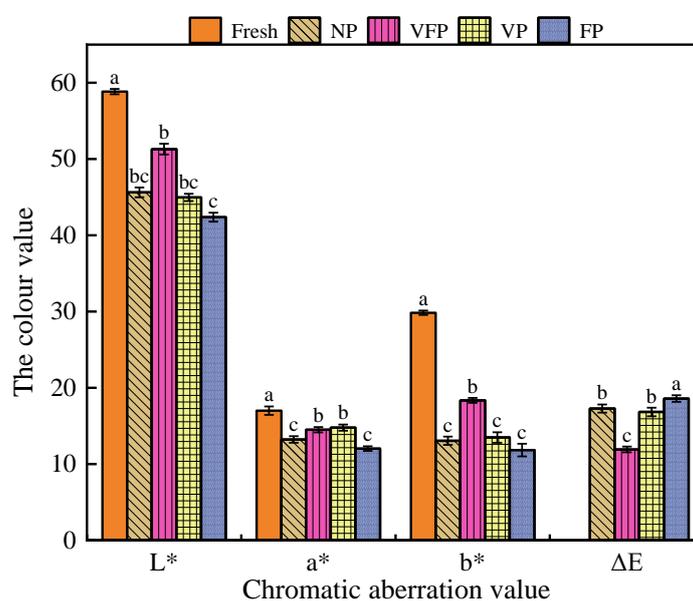


**Figure 1.** Sensory analysis of loquat samples. (A) Loquat samples treated with different preservation methods. (B) Histogram of sensory evaluation results. Values marked with different letters in the histogram were significantly different ( $p < 0.05$ ,  $n = 15$ ).

As illustrated in Figure 1B, the sensory scores ranged from 55 to 78, with the VFP samples achieving the highest sensory scores. This indicates that the VFP samples exhibited few differences in color, flavor and loss after picking compared to fresh loquats [35]. The NP and FP samples show minimal difference, consistent with the results of the color difference principal component analysis and the E-nose principal component analysis. The lowest scores have been observed previously for the NP samples, indicating significant changes in color, flavor and integrity considerably during home preservation, likely influenced by the air temperature and humidity during preservation. VFP used a combination of vacuum packaging and freezing to better separate the loquats from the outside environment and reduce air oxidation, while reducing the activity of the loquat's internal tissues when the temperature was 4 °C, preserving the freshness for a longer period of time [36].

### 3.1.2. Colorimetric Analysis

A new automatic colorimeter was used to determine the color values of the loquat samples. Therefore, the trichromatic values, chromaticity and hue of loquats were analyzed, and the data were plotted as histograms (Figure 2). The results indicated that  $L^*$  (from 58.85 to 42.37),  $a^*$  (from 17.01 to 12.03) and  $b^*$  (from 29.84 to 11.84) were significantly lower ( $p < 0.05$ ) compared to the fresh loquat samples in the four groups of NP, VFP, VP and FP samples. Among them, the FP samples showed the most significant change in values. This phenomenon could be explained by the increase in moisture content on the surface of loquats after freezing treatment, which accelerated the rate of darkening. The color difference values indicated that the VFP treatment had the least effect on the loquat color ( $\Delta E$  of 11.92) while the FP treatment had the greatest effect on the loquat color ( $\Delta E$  of 18.59). It is clear that vacuum packaging and a temperature of 4 °C reduced the possibility of browning [37].



**Figure 2.** Histogram of color aberration analysis of loquat. Values marked with different letters in the histogram were significantly different ( $p < 0.05$ ,  $n = 3$ ).

### 3.1.3. Nutritional Analysis

A physico-chemical analysis of loquat was carried out using four different methods, and the data were plotted as histograms (Table 2). The results indicated significant variations in energy (ranging from 39.40 kcal/100 g to 87.30 kcal/100 g), protein content (ranging from 0.80 g/100 g to 3.30 g/100 g) and carbohydrate content (increasing from 9.30 g/100 g to 17.8 g/100 g) with the varying preservation methods ( $p < 0.05$ ). Conversely, moisture content decreased from 87.20 g/100 g to 53.20 g/100 g ( $p < 0.05$ ), while fat content increased

marginally (from 0.010 g/100 g to 1.40 g/100 g) ( $p < 0.05$ ). Notably, the preservation effect of VFP on the protein, carbohydrate and water of loquat was obviously better than that of other preservation methods. The moisture content of fresh loquat gradually declined with the increase in the preservation time, and different preservation methods exhibited distinct effects on loquat water retention [38]. The results showed that the use of two methods of vacuum packaging treatment (VFP and VP) can better reduce the loss of water inside the loquats.

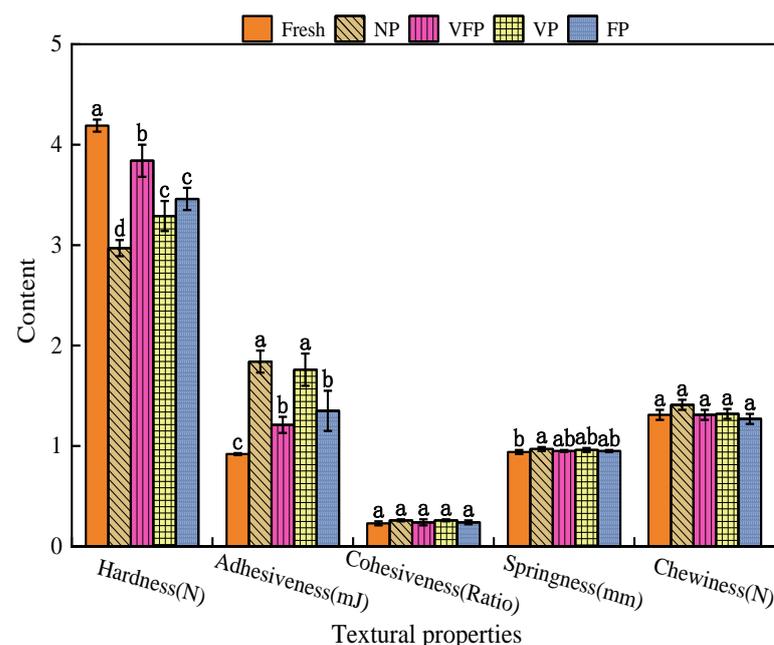
**Table 2.** Calories analysis of loquat under different preservation methods.

Preservation Method	Energy/(kcal/100 g)	Protein/(g/100 g)	Fat/(g/100 g)	Carbohydrates/(g/100 g)	Water/(g/100 g)
Fresh	39.40 ± 0.35 <sup>c</sup>	0.80 ± 0.10 <sup>c</sup>	0.10 ± 0.00 <sup>c</sup>	9.30 ± 0.35 <sup>c</sup>	87.20 ± 0.75 <sup>a</sup>
NP	87.30 ± 0.50 <sup>a</sup>	3.30 ± 0.25 <sup>a</sup>	0.90 ± 0.10 <sup>b</sup>	17.80 ± 0.20 <sup>a</sup>	53.20 ± 0.50 <sup>c</sup>
VFP	42.55 ± 1.10 <sup>c</sup>	1.20 ± 0.05 <sup>c</sup>	1.40 ± 0.08 <sup>a</sup>	12.15 ± 0.80 <sup>b</sup>	81.40 ± 0.60 <sup>a</sup>
VP	58.65 ± 0.20 <sup>b</sup>	2.15 ± 0.13 <sup>b</sup>	0.20 ± 0.02 <sup>c</sup>	15.50 ± 0.30 <sup>ab</sup>	76.70 ± 0.20 <sup>b</sup>
FP	61.70 ± 0.70 <sup>b</sup>	1.60 ± 0.15 <sup>bc</sup>	0.30 ± 0.10 <sup>c</sup>	13.40 ± 0.55 <sup>b</sup>	68.50 ± 0.30 <sup>b</sup>

Note: Different letters denote significant differences between the means of each parameter ( $p < 0.05$ ,  $n = 3$ ).

### 3.1.4. Textural Properties Analysis

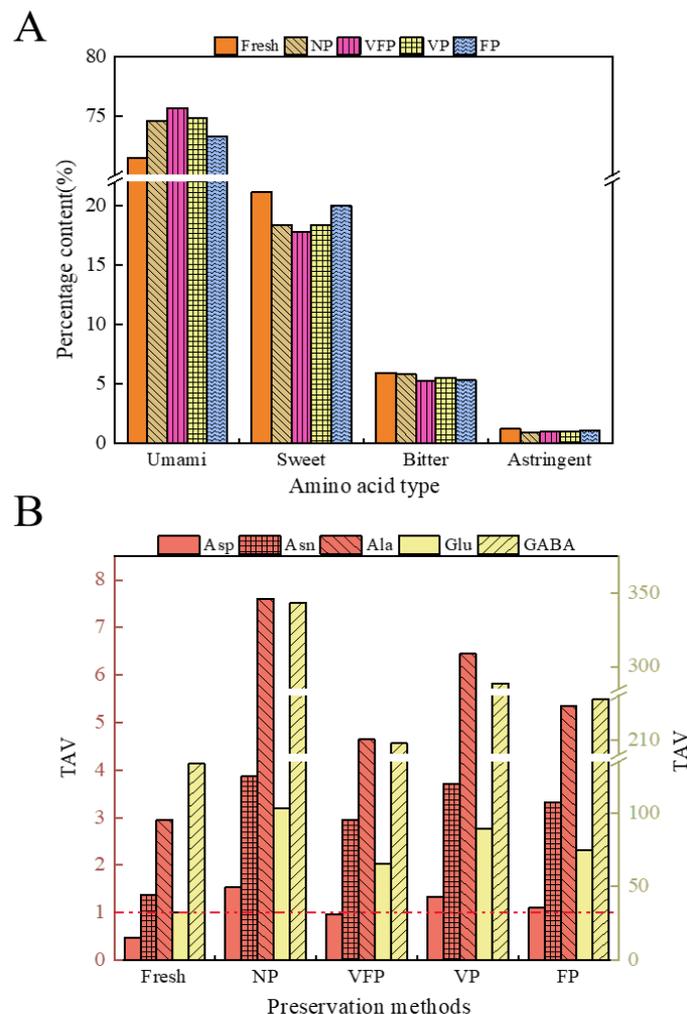
The textural quality of food is an organoleptic characterization of the textural and structural properties of food, and the adhesion and hardness of the loquat varied under different preservation methods (Figure 3). The results showed significant increases in the adhesion (ranging from 0.92 to 1.84 mJ) and hardness (ranging from 2.97 to 4.19 N) of loquats with changes in preservation methods ( $p < 0.05$ ), which may be attributed to enhanced intermolecular interactions due to water evaporation and the prolonged shelf life of loquats. However, the cohesion (ranging from a 0.23 to 0.26 ratio), elasticity (ranging from 0.94 to 0.97 mm) and chewiness (ranging from 1.27 to 1.41 N) of loquats did not change significantly with alterations in the preservation method ( $p < 0.05$ ), suggesting that different preservation methods altered the cohesion of loquats. Furthermore, there was no significant effect on elasticity and chewability [39]. The results showed that the two methods, VFP and FP, used at a temperature of 4 °C, better retained the original hardness of loquats.



**Figure 3.** Histogram of the qualitative analysis of loquat, values marked with different letters in the same histogram are significantly different ( $p < 0.05$ ,  $n = 5$ ).

### 3.1.5. Free Amino Acids Analysis

As shown in Table 3, the content of FAAs in loquats varied with different preservation methods. The highest total free amino acid content has been observed previously in NP (65.42 mg/g) and the lowest in fresh (22.47 mg/g). The free amino acids can be classified into fresh flavor amino acids (including Asp, Asn and Glu) (16.06–47.93 mg/g), sweet amino acids (including Ala, Gly, Pro, Thr and Ser) (4.74–13.07 mg/g), bitter amino acids (including His, Lys, Arg, Ile, Leu and Val) (1.33–3.49 mg/g) and astringent amino acids (including GABA) (0.27–0.69 mg/g). The loquat treated with five different preservation methods had the highest fresh amino acid content (Figure 4A), accounting for 71.48, 74.59, 75.67, 74.78 and 73.27% of the total free amino acids, respectively. The TAVs of FAAs were calculated to evaluate the contribution of FAAs to the taste of loquat (Figure 4B). The TAVs of Asn, Glu, Ala and GABA in loquats treated with five different preservation methods were all above 1.0, indicating their potential role in imparting freshness and sweetness to the loquat. The TAVs of Asp in NP, VP and FP were also greater than 1.0, suggesting variations in the taste of loquat among the different preservation methods. This indicates that different preservation methods bring about a greater effect on the amino acid content of loquats [40]. The increase in freshness amino acids and sweetness amino acids indicated that, within a certain time frame, the content of these two types of amino acids in loquats increased with the extension of storage time, which gave loquats a good flavor and texture.



**Figure 4.** (A) Percentage of different classes of free amino acids in loquat. (B) Taste Activity Value (TAV) of free amino acids. (Red dashed line: TAV equals 1). TAVs of free amino acids were calculated by the ratio of the concentration of a compound to its taste threshold.

**Table 3.** Free amino acid content of different loquat samples.

Form	Name	Fresh	NP	VFP	VP	FP
Umami	Asp	1.26 ± 0.22 <sup>d</sup>	4.11 ± 0.11 <sup>a</sup>	2.58 ± 0.35 <sup>c</sup>	3.54 ± 0.28 <sup>b</sup>	2.93 ± 0.19 <sup>b</sup>
	Asn	9.06 ± 0.89 <sup>c</sup>	25.61 ± 0.76 <sup>a</sup>	19.56 ± 0.66 <sup>b</sup>	24.54 ± 0.95 <sup>a</sup>	21.91 ± 0.58 <sup>b</sup>
	Glu	5.74 ± 0.36 <sup>d</sup>	18.22 ± 0.25 <sup>a</sup>	11.54 ± 0.39 <sup>c</sup>	15.83 ± 0.25 <sup>b</sup>	13.20 ± 0.41 <sup>b</sup>
	Ala	3.17 ± 0.41 <sup>c</sup>	8.13 ± 0.88 <sup>a</sup>	4.97 ± 0.36 <sup>c</sup>	6.89 ± 0.29 <sup>b</sup>	5.71 ± 0.62 <sup>b</sup>
Sweet	Gly	0.18 ± 0.21 <sup>d</sup>	0.59 ± 0.05 <sup>a</sup>	0.35 ± 0.18 <sup>c</sup>	0.49 ± 0.15 <sup>b</sup>	0.41 ± 0.03 <sup>b</sup>
	Pro	0.37 ± 0.02 <sup>c</sup>	1.23 ± 0.31 <sup>a</sup>	1.09 ± 0.28 <sup>a</sup>	0.81 ± 0.08 <sup>b</sup>	0.67 ± 0.23 <sup>b</sup>
	Thr	0.16 ± 0.02 <sup>c</sup>	0.47 ± 0.09 <sup>a</sup>	0.29 ± 0.02 <sup>b</sup>	0.40 ± 0.05 <sup>a</sup>	0.33 ± 0.02 <sup>b</sup>
	Ser	0.87 ± 0.18 <sup>c</sup>	2.65 ± 0.38 <sup>a</sup>	1.60 ± 0.10 <sup>b</sup>	2.21 ± 0.16 <sup>a</sup>	1.84 ± 0.24 <sup>b</sup>
Bitter	His	0.33 ± 0.05 <sup>d</sup>	0.81 ± 0.15 <sup>a</sup>	0.51 ± 0.15 <sup>c</sup>	0.70 ± 0.15 <sup>b</sup>	0.59 ± 0.24 <sup>c</sup>
	Lys	0.33 ± 0.05 <sup>c</sup>	0.58 ± 0.20 <sup>b</sup>	0.65 ± 0.21 <sup>a</sup>	0.64 ± 0.18 <sup>a</sup>	0.48 ± 0.15 <sup>b</sup>
	Arg	0.31 ± 0.10 <sup>c</sup>	0.92 ± 0.18 <sup>a</sup>	0.67 ± 0.11 <sup>b</sup>	0.89 ± 0.18 <sup>a</sup>	0.68 ± 0.17 <sup>b</sup>
	Ile	0.08 ± 0.01 <sup>c</sup>	0.25 ± 0.02 <sup>a</sup>	0.17 ± 0.01 <sup>b</sup>	0.22 ± 0.01 <sup>a</sup>	0.18 ± 0.01 <sup>b</sup>
Tasteless	Leu	0.06 ± 0.01 <sup>c</sup>	0.19 ± 0.01 <sup>a</sup>	0.13 ± 0.01 <sup>b</sup>	0.15 ± 0.01 <sup>b</sup>	0.13 ± 0.01 <sup>b</sup>
	Val	0.23 ± 0.10 <sup>d</sup>	0.73 ± 0.30 <sup>a</sup>	0.47 ± 0.20 <sup>c</sup>	0.63 ± 0.20 <sup>b</sup>	0.56 ± 0.20 <sup>b</sup>
	β-Ala	0.07 ± 0.01 <sup>d</sup>	0.24 ± 0.02 <sup>a</sup>	0.14 ± 0.01 <sup>c</sup>	0.12 ± 0.01 <sup>c</sup>	0.17 ± 0.01 <sup>b</sup>
Astringent	GABA	0.27 ± 0.02 <sup>d</sup>	0.69 ± 0.05 <sup>a</sup>	0.42 ± 0.03 <sup>c</sup>	0.58 ± 0.02 <sup>b</sup>	0.48 ± 0.05 <sup>c</sup>

Note: Different letters denote significant differences between the means of each parameter ( $p < 0.05$ ,  $n = 3$ ).

### 3.1.6. Organic Acids Analysis

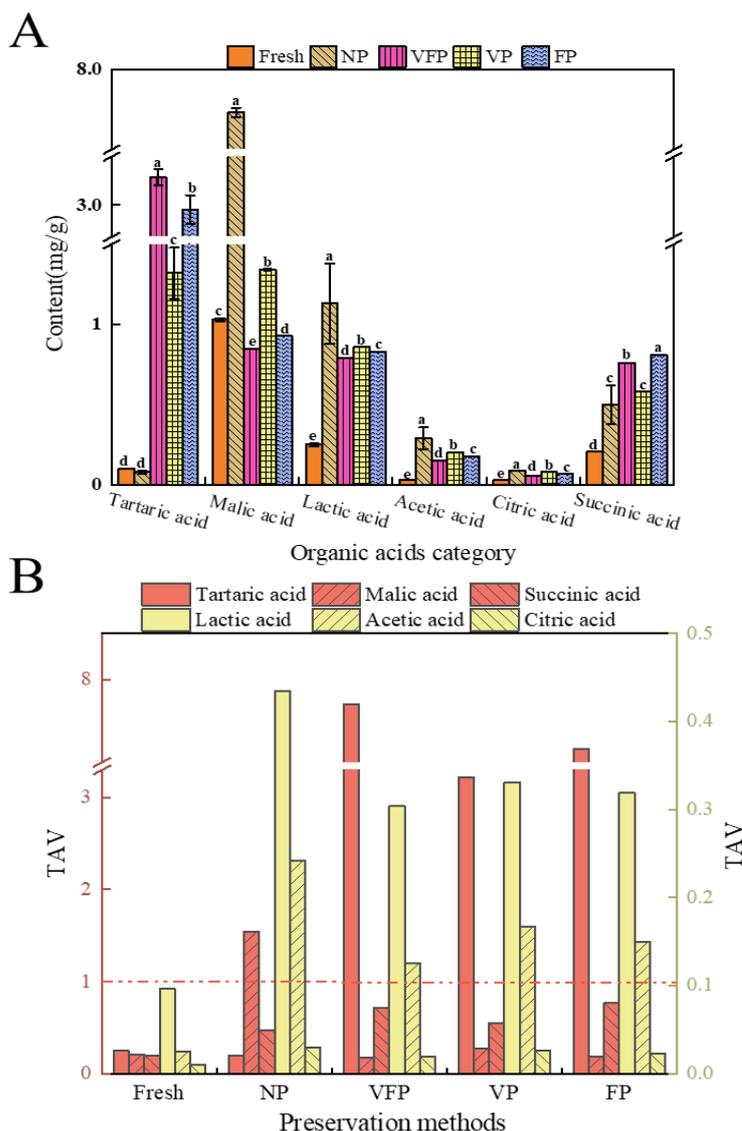
As shown in Figure 5A, the type of organic acids did not change with the change in the preservation method, but the content changed. The highest organic acid content has been found previously in VFP (9.07 mg/g), followed by FP (8.68 mg/g), VP (4.56 mg/g) and NP (2.91 mg/g), with fresh having the lowest (0.78 mg/g). To assess the contribution of organic acids to the flavor of loquats, the TAV of organic acids was calculated. As depicted in Figure 5B, the TAV of lactic and acetic acids was greater than 1.0 for all four loquats except fresh. The TAV of tartaric acid was greater than 1.0 for VFP, VP and FP, while the TAV of malic acid was greater than 1.0 for NP. Succinic acid and citric acid had lower TAVs than 1.0 for the five loquats. The results showed that tartaric, malic, lactic and acetic acids play significant roles in the taste of loquat, and are influenced by different preservation methods [41].

## 3.2. Volatile Profile of Loquat Characterized by GC–MS

### 3.2.1. Differences in Flavor Substances in Loquat with Different Treatment Methods

The total concentration of volatile compounds varies depending on the preservation method. Among the flavor substances analyzed by GC–MS (Table 4), 47 peaks (including monomers and dimers) were identified for loquat volatile substances, including 13 aldehydes, 9 esters, 6 ketones, 2 acids, 3 alcohols, 2 phenols, 3 pyrazines, 1 furan, 1 pyrrole, 1 aromatic, 1 olefin, 1 thioether, 2 amines and 2 alkanes.

To better illustrate the differences between loquat treatments under different preservation methods, an approximate percentage of odorant species in loquat under different preservation methods was obtained (Figure 6A). The results showed that the volatile flavor compounds of the five loquat samples were mainly composed of aldehydes, esters and acids, accounting for 59.71–64%, 18.13–20.28% and 4.08–5.43%, respectively. Among the aldehydes, compared with fresh loquat, VFP and FP exhibited higher aldehyde content in the other four loquat samples. Moreover, the OAV value of VFP hexenal was significantly higher than that of other treatment methods, followed by VP, NP and FP, aligning with the conclusions in the table. During the preservation process, a substantial amount of esters was produced, significantly higher in loquat treated with NP, VFP, VP and FP than in fresh, due to the oxidation reaction of oxidase in loquat pulp upon contact with air [42]. Also, the reduction of Aldehydes may be due to the microbial growth of loquats during the preservation process, and it has been shown that the proliferation of *Pseudomonas aeruginosa* leads to significant changes in hexenal content [43].

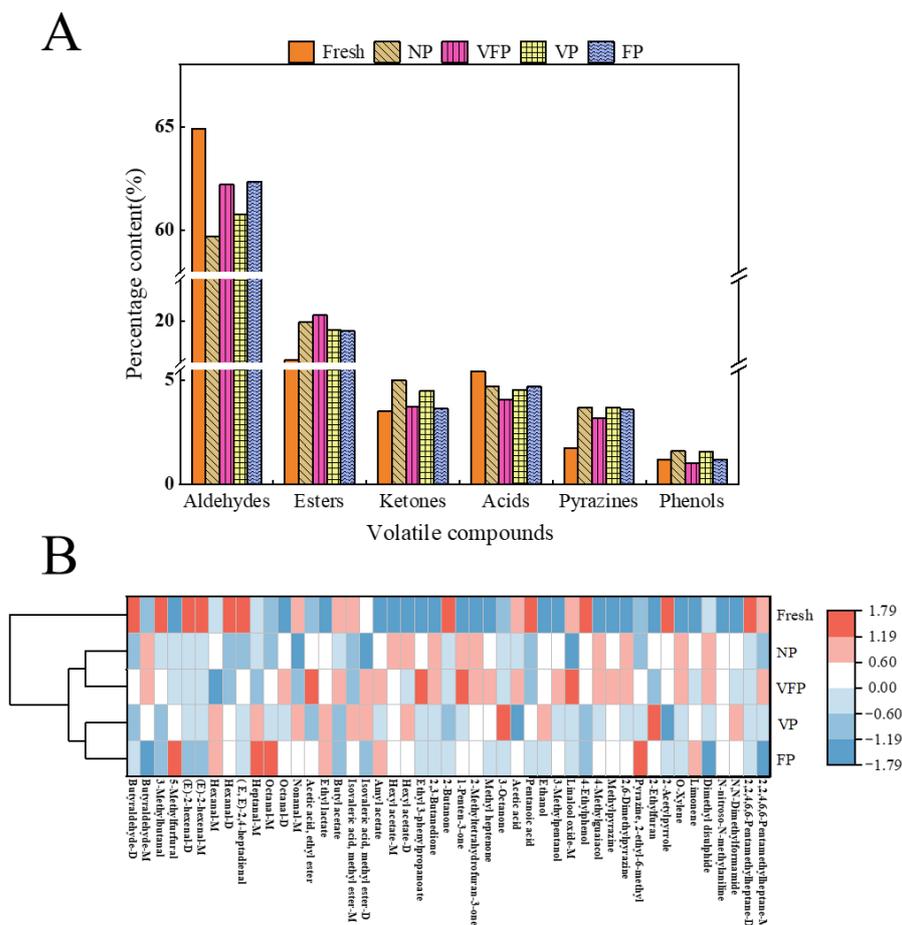


**Figure 5.** Analysis of organic acids in loquat. (A) Concentrations of organic acids, values marked with different letters in the same histogram are significantly different ( $p < 0.05$ ,  $n = 3$ ) (B) TAVs of organic acids. (Red dashed line: TAV equals 1). TAVs of organic acids were calculated by the ratio of the concentration of a compound to its taste threshold.

To further understand the differences in the volatile components of loquat under different preservation methods, a cluster analysis was conducted using heat maps (Figure 6B). The relative percentages of characteristic flavor compounds such as Butyraldehyde-D, 3-Methylbutanal, (E)-2-hexenal, Hexanal-D, 2-Butanone, pentanoic acid and 4-Ethylphenol in fresh are relatively high. Meanwhile, the concentrations of characteristic flavor substances such as acetic acid, ethyl ester, ethyl-3-phenylpropanoate, 1-Penten-3-one and Linalool oxide-M in NP were relatively high. The contents of characteristic flavor compounds such as 3-Octanone and 2-Ethylfuran were relatively high in VFP. The concentrations of 5-methylfurfural, octanal-M, heptanal-M, Pyrazine and 2-ethyl-6-methyl in FP were relatively high. Compared with fresh, the concentrations of many compounds in NP, VFP, VP and FP increased or decreased significantly, which can be intuitively reflected in the heat map of volatile compound concentrations. Based on the above results, it can be inferred that the volatile composition of loquat varies significantly due to different treatment methods, and NP, VFP and VP have a great influence on loquat volatility [44].

**Table 4.** Volatile compounds found in loquat from five different preservation methods.

No.	Component Name	Molecular Formula	Relative Amount/%				
			Fresh	NP	VFP	VP	FP
1	Butyraldehyde-D	C <sub>4</sub> H <sub>8</sub> O	1.68	1.41	1.30	1.30	1.33
2	Butyraldehyde-M	C <sub>4</sub> H <sub>8</sub> O	2.70	3.22	2.97	3.36	2.50
3	3-Methylbutanal	C <sub>5</sub> H <sub>10</sub> O	1.50	1.12	0.77	0.86	0.82
4	5-Methylfurfural	C <sub>6</sub> H <sub>6</sub> O <sub>2</sub>	9.02	10.70	11.59	10.61	12.41
5	(E)-2-hexenal-D	C <sub>6</sub> H <sub>10</sub> O	14.70	11.24	10.31	10.53	10.21
6	(E)-2-hexenal-M	C <sub>6</sub> H <sub>10</sub> O	13.90	12.26	12.55	12.48	12.04
7	Hexanal-M	C <sub>6</sub> H <sub>12</sub> O	6.27	5.73	6.82	6.73	6.81
8	Hexanal-D	C <sub>6</sub> H <sub>12</sub> O	8.19	5.37	6.87	5.47	6.81
9	(E,E)-2,4-heptadienal	C <sub>7</sub> H <sub>10</sub> O	3.19	2.58	2.27	2.08	2.24
10	Heptanal-M	C <sub>7</sub> H <sub>14</sub> O	0.50	0.34	0.79	0.50	0.97
11	Octanal-M	C <sub>8</sub> H <sub>16</sub> O	1.04	1.41	1.23	0.96	1.96
12	Octanal-D	C <sub>8</sub> H <sub>16</sub> O	0.30	1.28	0.89	1.17	1.20
13	Nonanal-M	C <sub>9</sub> H <sub>18</sub> O	1.89	1.26	1.98	1.08	1.79
Total (Aldehydes)			64.88	57.92	60.34	57.13	61.09
14	Acetic acid, ethyl ester	C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>	0.01	1.09	0.12	0.55	0.57
15	Ethyl lactate	C <sub>5</sub> H <sub>10</sub> O <sub>3</sub>	0.25	0.41	0.43	0.40	0.43
16	Butyl acetate	C <sub>6</sub> H <sub>12</sub> O <sub>2</sub>	0.13	0.12	0.05	0.08	0.06
17	Isovaleric acid, methyl ester-M	C <sub>6</sub> H <sub>12</sub> O <sub>2</sub>	5.36	3.08	4.95	3.36	4.22
18	Isovaleric acid, methyl ester-D	C <sub>6</sub> H <sub>12</sub> O <sub>2</sub>	11.45	11.51	11.49	11.18	11.23
19	Amyl acetate	C <sub>7</sub> H <sub>14</sub> O <sub>2</sub>	0.06	0.28	0.19	0.22	0.27
20	Hexyl acetate-M	C <sub>8</sub> H <sub>16</sub> O <sub>2</sub>	0.12	0.21	0.22	0.23	0.21
21	Hexyl acetate-D	C <sub>8</sub> H <sub>16</sub> O <sub>2</sub>	0.17	0.82	1.12	1.11	0.99
22	Ethyl 3-phenylpropanoate	C <sub>11</sub> H <sub>14</sub> O <sub>2</sub>	0.58	1.82	1.10	1.25	1.15
Total (Esters)			18.13	19.34	19.67	18.38	19.13
23	2,3-Butanedione	C <sub>4</sub> H <sub>6</sub> O <sub>2</sub>	0.07	0.35	0.24	0.31	0.24
24	2-Butanone	C <sub>4</sub> H <sub>8</sub> O	2.40	0.95	0.60	0.76	0.63
25	1-Penten-3-one	C <sub>5</sub> H <sub>8</sub> O	0.46	0.93	0.62	0.82	0.71
26	2-Methyltetrahydrofuran-3-one	C <sub>5</sub> H <sub>8</sub> O <sub>2</sub>	0.26	1.52	1.11	1.38	1.16
27	Methyl heptenone	C <sub>8</sub> H <sub>14</sub> O	0.24	1.01	0.82	0.86	0.71
28	3-Octanone	C <sub>8</sub> H <sub>16</sub> O	0.08	0.09	0.21	0.10	0.11
Total (Ketones)			3.51	4.85	3.60	4.23	3.56
29	Acetic acid	CH <sub>3</sub> COOH	3.97	4.00	3.06	3.71	3.93
30	Pentanoic acid	C <sub>5</sub> H <sub>10</sub> O <sub>2</sub>	1.46	0.56	0.90	0.56	0.67
Total (Acids)			5.43	4.56	3.96	4.27	4.60
31	Ethanol	C <sub>2</sub> H <sub>6</sub> O	0.28	0.44	0.51	0.47	0.42
32	3-Methylpentanol	C <sub>6</sub> H <sub>14</sub> O	0.09	0.32	0.22	0.25	0.25
33	Linalool oxide-M	C <sub>10</sub> H <sub>18</sub> O <sub>2</sub>	0.72	0.76	0.66	0.56	0.62
Total (Alcohols)			1.09	1.52	1.39	1.28	1.29
34	4-Ethylphenol	C <sub>8</sub> H <sub>10</sub> O	1.03	0.82	0.56	0.77	0.64
35	4-Methylguaiaicol	C <sub>8</sub> H <sub>10</sub> O <sub>2</sub>	0.14	0.72	0.44	0.71	0.53
Total (Phenols)			1.17	1.54	1.00	1.48	1.17
36	Methylpyrazine	C <sub>5</sub> H <sub>6</sub> N <sub>2</sub>	0.93	1.77	1.62	1.73	1.67
37	2,6-Dimethylpyrazine	C <sub>6</sub> H <sub>8</sub> N <sub>2</sub>	0.28	1.07	0.82	1.16	0.97
38	Pyrazine, 2-ethyl-6-methyl	C <sub>7</sub> H <sub>10</sub> N <sub>2</sub>	0.52	0.74	0.64	0.58	0.87
Total (Pyrazines)			1.73	3.58	3.08	3.47	3.51
39	2-Ethylfuran	C <sub>6</sub> H <sub>8</sub> O	0.32	0.46	0.83	0.61	0.68
Total (Furan)			0.32	0.46	0.83	0.61	0.68
41	O-xylene	C <sub>8</sub> H <sub>10</sub>	0.27	0.79	0.54	0.75	0.48
Total (Aromatic hydrocarbon)			0.27	0.79	0.54	0.75	0.48
42	Limonene	C <sub>10</sub> H <sub>16</sub>	0.06	0.34	0.44	0.42	0.52
Total (Olefin)			0.06	0.34	0.44	0.42	0.52
43	Dimethyl disulphide	C <sub>2</sub> H <sub>6</sub> S <sub>2</sub>	0.14	0.16	0.14	0.16	0.12
Total (Thioether)			0.14	0.16	0.14	0.16	0.12
44	N-nitroso-N-methylaniline	C <sub>7</sub> H <sub>8</sub> N <sub>2</sub> O	0.09	0.50	0.48	0.50	0.49
45	N,N-Dimethylformamide	C <sub>3</sub> H <sub>7</sub> NO	0.15	0.56	0.85	0.67	0.61
Total (Amines)			0.24	1.06	1.33	1.17	1.10
46	2,2,4,6,6-Pentamethylheptane-D	C <sub>12</sub> H <sub>26</sub>	2.81	0.68	0.52	0.46	0.59
47	2,2,4,6,6-Pentamethylheptane-M	C <sub>12</sub> H <sub>26</sub>	0.11	0.11	0.09	0.08	0.07
Total (Alkanes)			2.92	0.79	0.61	0.54	0.66

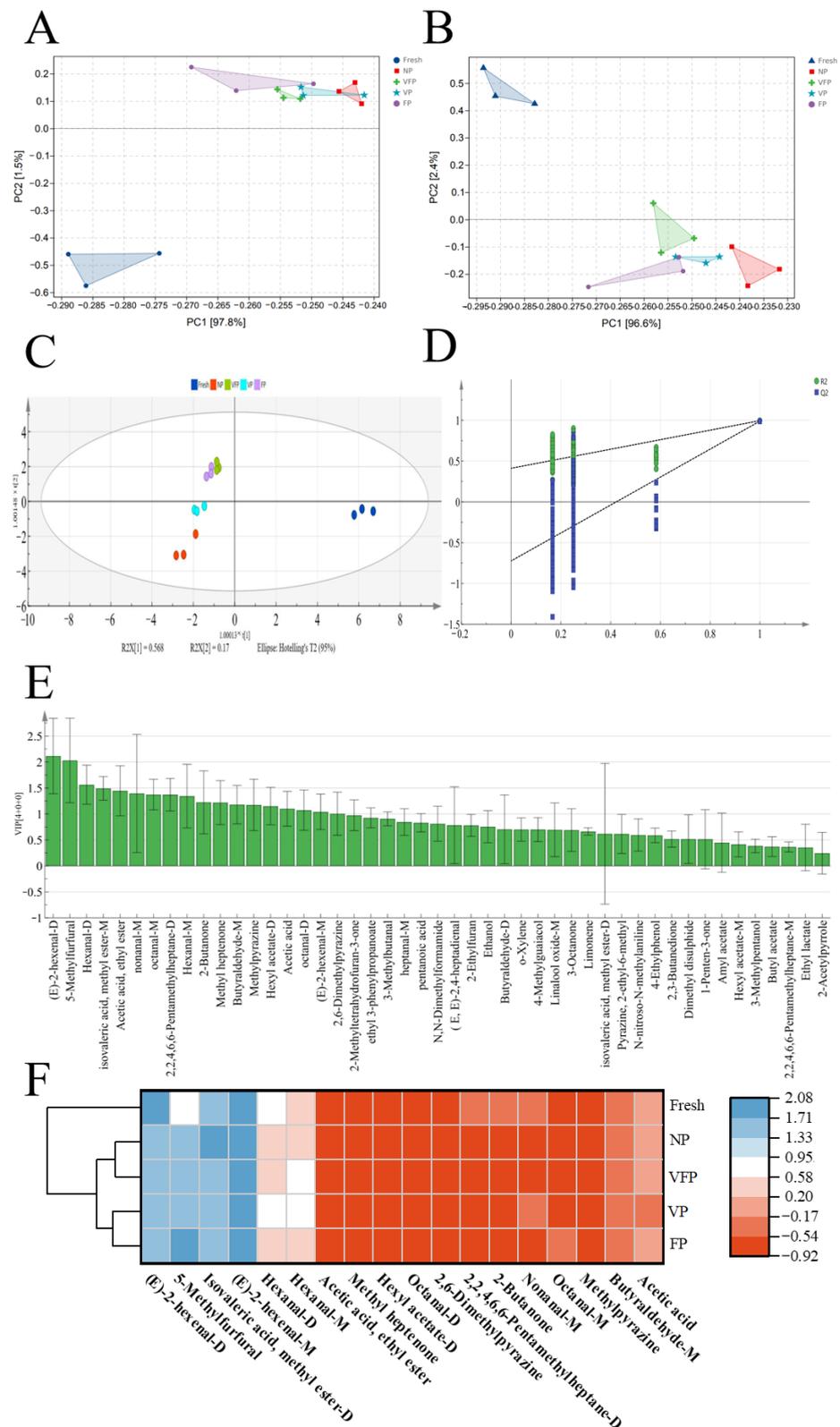


**Figure 6.** Analysis of volatile compounds of the loquat identified by GC–MS. **(A)** Proportions of each group of volatiles. **(B)** Concentrations of volatile compounds and clustering results.

### 3.2.2. Principal Component Analysis and Orthogonal Partial Least Squares-Discriminant Analysis with Cross-Validation

Five loquat samples were quantified by GC–MS through principal component analysis (PCA). As shown in Figure 7A, NP is positioned farther away from the rest of the samples on PC1, whereas fresh was closer to VP and FP. Similarly, in PC2, the VFP was notably distant from the other samples. In the clustered heat map (Figure 7E), it can be seen that the volatile spectrum of fresh was significantly different from that of the other four loquats, while the volatile spectra of VFP and FP were highly similar and were grouped as NP and VP. Therefore, principal component analysis (PCA) and cluster similarity analysis (clustering) were used to distinguish the odor characteristics of the five loquat cultivars.

In Figure 7D,  $R^2X$  and  $R^2Y$  represent the rate at which the simulation interprets the  $X$  and  $Y$  vectors, respectively, and  $Q^2$  represents the predictive capability in the actual simulation. Parameters with  $R^2$  and  $Q^2$  above 0.5 and close to 1.0 were considered definitive results. As shown in Figure 7C, the model encapsulates a significant portion of the information on the volatile components in the five loquats. The results suggested that most of the loquat samples can be classified by the OPLS-DA diffusion plot, with the classification response closely resembling the PCA plot (Figure 7A). To mitigate overestimation, the accuracy of the OPLS-DA model was confirmed through rearrangement experiments, as depicted in Figure 7. Subsequently, 200 cross-validations were conducted, simulating the backtrace fringes of  $Q^2$  across the horizontal ordinates, with node increments representing negativity. All rearrangement experiments  $R^2$  and  $Q^2$  were smaller than the original data, indicating that the simulation equations were not over-adapted. Therefore, the results obtained from the established OPLS-DA simulations were consistent and reasonable [45].



**Figure 7.** (A) PCA scatter map of 47 volatile substances. (B) PCA scatter map of 18 differential volatile substances. OPLS-DA scatter plot (C) and trans-verifications by a rearrangement trial (D) of odor profiles in five loquats. VIP distribution (E) and clustering heatmap (F) of the differential volatile compounds screened from five loquat samples.

After detecting 47 odor compounds in loquat samples treated with five different preservation methods using GC–MS, the VIP values (Figure 7E) obtained from establishing a reliable OPLS-DA match were used to estimate the involvement of each odorant compound in loquat. The study revealed 18 odor chemicals, including 5-methylfurfural, acetic acid, nonanal-M, (E)-2-hexenal, octanal-M, methyl isovalerate, hexyl acetate-D, butyraldehyde-M, 2,6-dimethylpyrazine, heptanal-M, hexanal-D, 2-ethyl-6-methylpyrazine and ethyl acetate. PCA was performed using these discriminatory indicator chemicals (Figure 7B). Moreover, the study confirmed that the combination of 18 indicator odor chemicals and PCA could effectively distinguish the variation among the five loquats [46].

### 3.3. Intelligent Senses Analysis

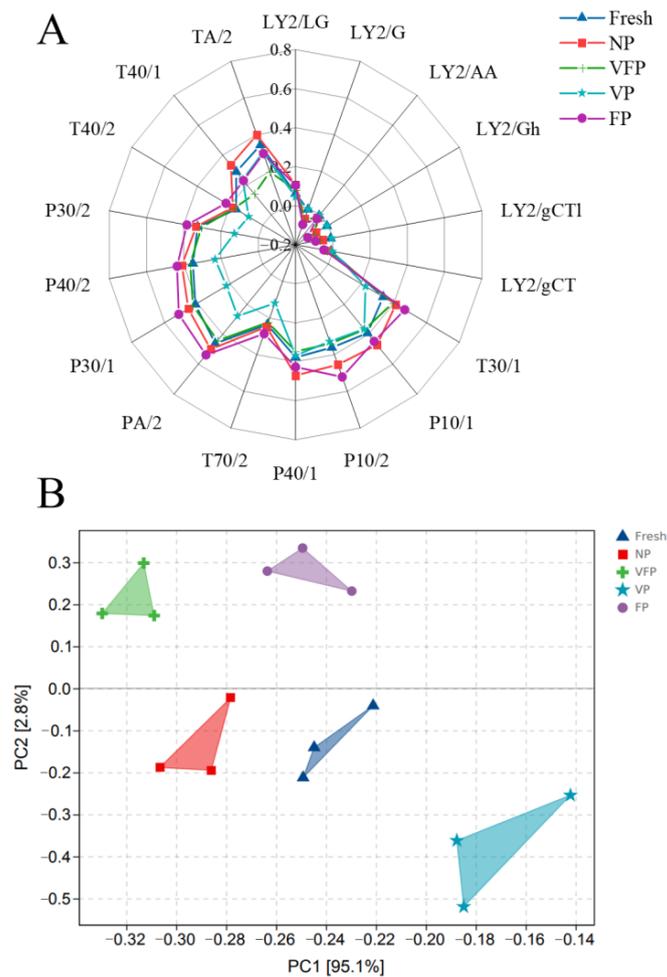
#### 3.3.1. E-Nose Analysis

E-nose analysis was conducted on loquat samples preserved by five different methods, with the response values of the sensors at 120 s depicted as radargrams. In Figure 8A, significant variations in the response values of the dried samples were observed across most sensors, except for the LY2/LG sensors. The results indicated that the different preservation methods affected the aroma of loquat, and that the aroma of the VP loquat samples changed the most significantly. Comparing the fresh and VFP samples, the response values closely aligned with those of the T30/1, P10/1, P10/2, T70/2, P40/1, PA/2, P30/1, P40/2, P30/2 and T40/2 sensors, which were consistent with the chromatographic results. Notably, the response values of the fresh, NP, VFP, VP and FP samples were consistent with the LY2/G, LY2/AA, LY2/Gh, LY2/gCT1 and LY2/gCT sensors, indicating a minimal change in the aroma before and after preservation. Significant differences were observed in the response values of VP and FP on the E-nose sensor. The method of preservation significantly affected the flavor of loquats, with VFP and fresh loquats exhibiting similar flavor qualities, indicating that the flavor profiles were highly similar within these groups.

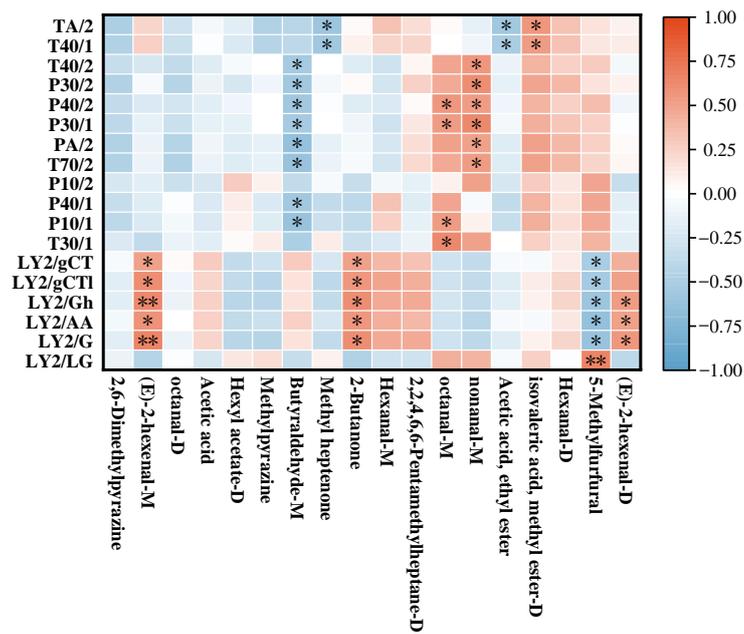
This phenomenon was reflected in the PCA results (Figure 8B). VP was distributed in the first and second quadrants, while FP was in the fourth quadrant, indicating significant differences in an olfactory sense between the two preservation methods. The contribution of the first principal component (PC1) was 95.10% and the contribution of the second principal component (PC2) was 2.80%. The total contribution of the two principal components was 97.90%, indicating a better effect of dimensionality reduction. The E-nose effectively distinguished the odor of loquats treated with different preservation methods, further confirming that the electronic nose combined with PCA analysis could distinguish loquat samples treated by various preservation methods [47].

#### 3.3.2. E-Nose and GC–MS Correlation Analysis

In the E-nose and GC–MS correlation analysis (Figure 9), the response values of the electronic nose sensor were compared with the differential volatile compounds detected by GC–MS. Moreover, compounds like (E)-2-hexanal-D (monomer and dimer) and 2-butanone showed positive correlations with the LY2/G, LY2/AA and LY2/LG sensors, while 5-methylfurfural exhibited a negative correlation. (E)-2-hexanal-M (monomer and dimer) was extremely negatively correlated with the LY2/G and LY2/LG sensors. Additionally, octanal correlated positively with the P10/1 sensor, while butanal showed a negative correlation. The GC–MS results indicated that these compounds were highly expressed in fresh. On the contrary, 5-methylfurfural was positively correlated with the LY2/gCT, LY2/gCT1, LY2/Gh, LY2/AA and LY2/G sensors and negatively correlated with the LY2/LG sensor. Thus, combining E-nose and GC–MS could distinguish loquats treated with five different preservation methods based on olfaction.



**Figure 8.** E-nose analysis of the loquat. **(A)** Radar graph for the E-nose analysis. **(B)** PCA graph for the E-nose analysis.

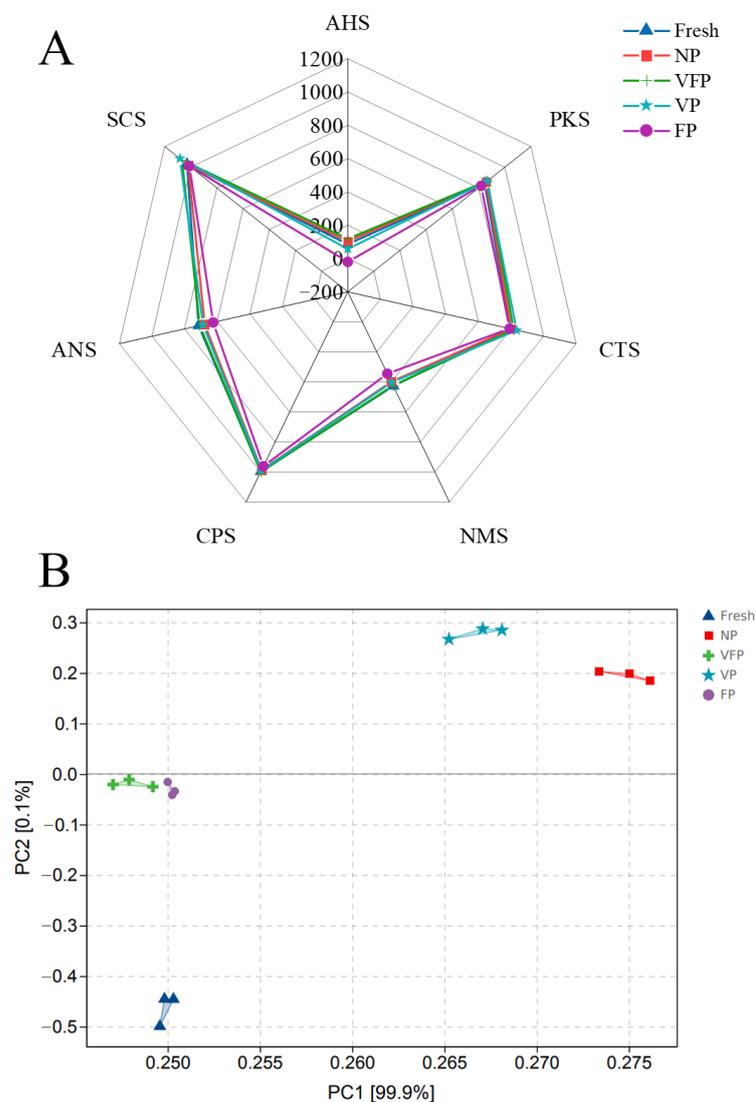


\* p<0.05 \*\* p<0.01

**Figure 9.** Heat map of the correlation between the response values of the E-nose sensors and the levels of differential volatile compounds.

### 3.3.3. E-Tongue Analysis

The E-tongue method was used to evaluate the differences in the taste attributes of loquat under different preservation methods [48]. The basic sensory taste indices were presented as radar images (Figure 10A). The greatest differences in response values were observed between FP and the other four sensors for loquat acidity (AHS), sweetness (ANS) and freshness (NMS). The highest values were observed for NP (118.67) in AHS and freshness (714.92) in ANS. The bitter (SCS) and savory (CTS) flavors showed little variability. The PCA results (Figure 10B) depicted NP, VFP, VP and FP as relatively independent and far away from fresh. The results indicated that the intensity of loquat flavor gradually changed due to the change in preservation methods, and fresh was significantly different from the other four loquat samples. The results suggested the effect of different preservation methods on loquats' texture.



**Figure 10.** E-tongue analysis of the loquat. (A) Radar graph for the E-tongue analysis. (B) PCA graph for the E-tongue analysis.

## 4. Conclusions

To address the issue of loquats being prone to oxidation and blackening after picking, and being damaged by microorganisms, resulting in a serious loss of quality such as color, aroma, flavor and shape, different preservation methods were adopted to treat loquat from the perspective of feasibility analysis of loquat preservation. The effects of different

preservation methods on the quality of loquat were analyzed using physical and chemical indicators, such as a colorimeter, E-nose, E-tongue and GC–MS. The sensory scores ranged from 55 to 78, and  $\Delta E$  ranged from 11.92 to 18.59. Moisture content, heat, adhesion and hardness varied significantly ( $p < 0.05$ ) with the change in preservation methods. Free amino acid and organic acid contents also varied. Moreover, a total of 47 volatile flavor substances were detected by the GC–MS method, including 13 aldehydes, 9 esters, 6 ketones, 2 acids, 3 alcohols, 2 phenols, 3 pyrazines, 1 furan and 8 other substances. Aldehydes were found to be the most abundant, significantly contributing to the flavor substances of loquat. Furthermore, the OPLS-DA model analysis identified 18 marker compounds capable of distinguishing between the five loquat species. Both the E-nose and E-tongue PCA results showed that the VFP samples were the closest to fresh loquat in the olfactory and gustatory senses. When the preservation method was FP, both the olfactory and gustatory sensations of loquat changed significantly. This indicates that the FP preservation treatment at 4 °C alone had a bigger effect on loquats compared to the combination of vacuum packaging and low temperature treatment at 4 °C. Based on physicochemical indexes, color differences, E-nose, E-tongue, GC–MS and intelligent sensory analyses, when the preservation temperature was 4 °C, and also after using vacuum packaging, loquats were the most similar to fresh loquats in color, aroma, taste and shape, with less loss of volatiles. Therefore, VFP could be the best method of home loquat preservation.

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**Data Availability Statement:** Data are contained within the article.

**Conflicts of Interest:** The authors declare that they have no conflicts of interest.

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