

Review

Solid-Phase Parallel Synthesis of Drug-Like Artificial 2*H*-Benzopyran Libraries

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Abstract: This review covers the construction of drug-like 2*H*-benzopyrans and related libraries using solid-phase parallel synthesis. In this context, the preparation of substituted benzopyrans such as mono-, di- and trisubstituted benzopyran derivatives and additional ring-fused benzopyrans such as benzopyranoisoxazoles, benzopyranopyrazoles, six-membered ring-fused benzopyrans, and polycyclic benzopyrans are highlighted.

Keywords: combinatorial chemistry; solid-phase synthesis; chemical library; drug-like molecules; 2*H*-benzopyran

1. Introduction

Combinatorial chemistry has become an extremely powerful technique for the generation of druglike and biologically active small organic molecule libraries in either the solution-phase or on solid supports [1–5]. In combinatorial synthesis, solid-phase organic synthesis (SPOS) is now routinely used to prepare a large number of small heterocyclic molecules and is especially useful in creating massive numbers of hit and lead compounds as part of high-throughput screening (HTS) technologies [6–10]. This is especially true for the privileged structures, which are core components of a large number of substances that possess a wide range of interesting biological activities and have been developed on solid-phase strategies [11–15].

Among biologically active heterocyclic scaffolds, the well-known privileged benzopyran structure frequently appears in many natural products and artificial bioactive molecules, that exhibit a wide range of biological activities [16–49]. Representative examples of benzopyran-containing natural products and artificial bioactive molecules are illustrated in Figure 1.

Figure 1. The representative examples of benzopyran-containing natural products and artificial bioactive molecules.

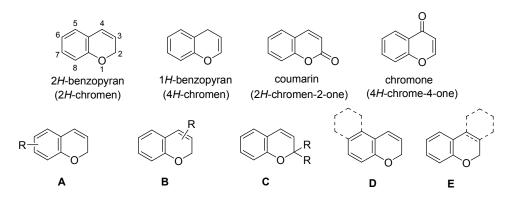
Hydroxylonchocarpin, lonchocarpin, and lonchocarpene exhibit anticancer activities [16–18], whereas moracin D exhibits antifungal activities [19,20]. Cromakalim [21–24] with its benzopyran moiety produces anti-hypertensive effects via potassium channel modulation, and SD8381 [25,26] with a 2*H*-benzopyran scaffold represents anti-inflammatory effects as a novel cyclooxygenase (COX)-2 inhibitor. Catechin and epigallocatechin gallate (EGCG) show both an antiallergic effects and anticancer action [27–29], whereas KRH-102140 with a 2*H*-benzopyran moiety is identified as a 5-lipoxygenase (5-LO) inhibitor [30,31]. Also, seselin and xanthylein with tricyclic benzopyrans exhibit anticancer activities [32–34].

(-)- Δ^9 -Tetrahydrocannabinol ((-)- Δ^9 -THC, dronabinol) and other cannabinoids have been used to treat the symptoms of cancer, pain relief, and spasticity in multiple sclerosis, or as appetite stimulants for acquired immunodeficiency syndrome (AIDS) patients [35–39]. Suksdorfin (khellactone ester) inhibits human immunodeficiency virus (HIV)-1 replication in H9 lymphocytes [40,41], whereas daleformis shows inhibitory activities against endothelin-converting enzyme [42]. Rotenone has been used as an antianaphylactic agent for the treatment of asthma [43–49].

This paper reviews the use of solid-phase parallel synthesis in the construction of 2H-benzopyran libraries that contain substituted 2H-benzopyrans (\mathbf{A} , \mathbf{B} , and \mathbf{C}) and additional cycle-fused

benzopyrans (**D** and **E**), except the derivatives of 1*H*-benzopyran, coumarin, and chromone moieties (Figure 2). In addition, the solid-phase synthesis of modified benzopyran derivatives using the 2H-benzopyran moiety as an intermediate is also discussed. The sections have been divided according to the number of substituents in the benzopyran core and the kinds of fused-benzopyran cores. Publications cited herein are mostly refereed journals and not patents.

Figure 2. The core structures of benzopyran.



2. Solid-Phase Synthesis of Substituted Benzopyran Compounds

2.1. Solid-Phase Synthesis of Monosubstituted Benzopyran Compounds

2.1.1. Solid-Phase Synthesis of 3-Substituted Benzopyran Compounds

Park and co-workers reported the solid-phase synthesis of 3-substituted 2*H*-benzopyran 1 (Scheme 1) and the fluorous-tag-based solution-phase synthesis of benzopyran derivatives with discrete core scaffolds to construct a 284-member library [50].

Scheme 1. Solid-phase synthesis of 3-substituted 2*H*-benzopyrans 1 by Park *et al.* [50].

3-EtS-Ph, 3,5-di-CF₃-Ph, 4-Me₂N-Ph, 4-Ac-Ph, 2-thiophenyl, etc

The library construction was started from four different chromanones 2 (2a-d, see Figure 3), which were subjected to α -bromination and subsequent silyl protection at the phenolic hydroxyl group. α -Bromoketones 3 were reduced to α -bromoalcohols 4 by NaBH₄, followed by acid-catalyzed dehydration and subsequent silyl deprotection, to yield four vinyl bromides containing 2H-benzopyran moiety 5. After the activation of (4-methoxyphenyl)-diisopropylsilylpropyl polystyrene resin 6 with TfOH, vinyl bromide intermediates 5 were immobilized on the activated resin 7 in the presence of 2,6-lutidine to afford polymer-bound intermediates 8.

Figure 3. Various structures of 2*H*-benzopyran precursor (chromanone) moieties 2.

The 3-substituted 2*H*-benzopyran resins **9** were introduced to various aryl and heteroaryl moieties via Suzuki coupling of boronic acids (24 commercially available aryl- and heteroaryl-boronic acids) with Pd(PPh₃)₄ and Na₂CO₃ in aqueous 1,4-dioxane with high yields and purity. After the standard cleavage protocol of silyl linkers using HF/pyridine and subsequent quenching with TMSOEt, the desired 3-substituted 2*H*-benzopyrans **1** (96 examples) were successfully prepared on a scale of 5–10 mg with an average purity of 86%.

2.1.2. Solid-Phase Synthesis of 4-Substituted Benzopyran Compounds

Park and co-workers also reported the solid-phase synthesis of 4-substituted 2*H*-benzopyran **10** using similar methods for the solid-phase synthesis of 3-substituted 2*H*-benzopyran **1** (see Scheme 1) except 3-triflated moiety in **11** as starting materials instead of 3-bromo-benzopyrans **5** (Scheme 2) [51].

Scheme 2. Solid-phase synthesis of 4-substituted 2*H*-benzopyrans 10 by Park *et al.* [51].

$$\begin{array}{c} \text{1. TMSCI} \\ \text{CH}_2\text{CI}_2, \text{ rt} \\ \text{OMe} \end{array} \begin{array}{c} \text{1. TMSCI} \\ \text{CH}_2\text{CI}_2, \text{ rt} \\ \text{OMe} \end{array} \begin{array}{c} \text{1. TMSCI} \\ \text{CH}_2\text{CI}_2, \text{ rt} \\ \text{CH}_2\text{CI}_2, \text{ rt} \end{array} \begin{array}{c} \text{OTf} \\ \text{2. TfOH} \\ \text{CH}_2\text{CI}_2, \text{ rt} \end{array} \begin{array}{c} \text{OTf} \\ \text{2. TfOH} \\ \text{CH}_2\text{CI}_2, \text{ rt} \end{array} \begin{array}{c} \text{OTf} \\ \text{CH}_2\text{CI}_2, \text{ 0 °C} \end{array} \begin{array}{c} \text{R}^2 \\ \text{CH}_2\text{CI}_2, \text{ 0 °C} \end{array} \begin{array}{c} \text{CH}_2\text{CI}_2, \text{$$

The synthetic route for 4-substituted 2*H*-benzopyran 10 was initiated from triflate resins 12, which were derived from eight different 2*H*-benzopyran moieties 2 (2a-h, see Figure 3), and the activated solid support 7. 4-Triflate-benzopyrans 11 were synthesized from four different hydroxyacetophenones through cyclization with acetone [R^2 = methyl] or cyclopentanone [R^2 = -(CH_2)4-] in the presence of pyrrolidine catalyst, and triflation of the resulting chromanones 2 using triflic anhydride in the presence of the proton sponge 2,6-di-*tert* butyl-4-methylpyridine (DTBMP). After activation of resin 6 by treating TfOH, eight different vinyl triflated intermediates 2 were immobilized on these activated resins 7 in the presence of 2,6-lutidine to afford intermediates 12 on solid supports (average loading level: ~0.9 mmol/g).

The various substituted aryl rings (R³) were introduced via palladium-mediated Suzuki coupling of aryl boronic acids. Among the many conditions tested in the solid-phase, the reaction condition with Pd(PPh₃)₄ and Na₂CO₃ in aqueous 1,4-dioxane displayed a robust chemical transformation of **3** with various substituted aryl boronic acids (18 different aryl boronic acids), resulting in high yields of the desired 3-substituted 2*H*-benzopyran resins **13**. Finally, the privileged benzopyrans **10** were produced by cleavage of resins **13** by using HF/pyridine in tetrahydrofuran (THF) and subsequent quenching with TMSOEt, and the resulting 144-member small-molecule collection was synthesized on a scale of 10–20 mg and their average purity was 87% without any purification steps.

Additionally, the solid-phase synthesis of the mono-substituted 2*H*-benzopyran-attaching 1,2,3-triazole ring at position 4 was described by Park and co-workers [51]. For the solid-phase synthesis of 1,2,3-triazole-substituted 2*H*-benzopyrans **14**, terminal alkyne resins **15** were introduced to the vinyl triflate intermediates **12** on the solid support through a palladium-mediated Negishi-type cross-coupling reaction (Scheme 3) [51]. After Negishi-type alkynylation with ethynylmagnesium bromide, Pd(PPh₃)₄, and ZnCl₂, the resulting terminal alkynyl moiety on the benzopyran core skeletons with azides **16** was subjected to regioselective Huisgen 1,3-dipolar [3 + 2] cycloaddition, namely, Click chemistry, in the presence of BrCu(PPh₃)₃ [52–54], a Cu-catalyst soluble in organic solvent, and *N*,*N*-diisopropylethylamine (DIPEA) to yield a new 1,2,3-triazole-substituted 2*H*-benzopyran resins **17** using solid-phase parallel synthesis. The eight different azides **16** were utilized for the Click chemistry and produced the novel 1,2,3-triazole-substituted 2*H*-benzopyrans **14** (64 examples) after HF/pyridine cleavage from the solid support and subsequent quenching with TMSOEt. The average purity, measured by liquid chromatography-mass spectrometry (LC-MS) analysis of the crude products, was 85%.

Scheme 3. Solid-phase synthesis of 1,2,3-triazole-substituted 2*H*-benzopyrans **14** by Park *et al.* [51].

Si) R
$$= \frac{R^2}{Pd(PPh_3)_4, ZnCl_2}$$
 $= \frac{R^2}{Pd(PPh_3)_4, ZnCl_2}$ $= \frac{R^2}{Pd(PPh_3)_4,$

Gong and Yoo reported the solid-phase synthesis of 3-hydroxy-4-amino-substituted benzopyrans 18 via epoxide opening in a two-phase solvent system [55]. The reaction of 4-nitrophenyl carbonate resin 19, which was formed by the reaction of Wang resin 20 and p-nitrophenyl chloroformate in CH₂Cl₂, with 6-amino-2,2-dimethyl chromene (21) and DIPEA in N,N-dimethylacetamide (DMA) afforded the carbamate resin 22 (Scheme 4).

Scheme 4. Solid-phase synthesis of 3-hydroxy-4-amino-substituted benzopyrans **18** by Gong and Yoo [55].

R' = Ph, 4-MeO-Ph, 2-MeO-Ph, 2,4-di-MeO-Ph, 4-Et-Ph, 3-NO₂-Ph, Bh, aliyi **R**' = H or Me

After various solvent systems and oxidizing agents were examined to avoid the formation of *m*-chlorobenzoic acid-added adduct resin in the case of oxidation of resin **22** with *m*-chloroperbenzoic acid (*m*CPBA), the two-phase solvent system comprised of chloroform and saturated aqueous NaHCO₃ with *m*CPBA afforded the epoxide resin **23**. The regioselective ring opening of the polymer-bounded epoxide **23** with nine amines produced the 3-hydroxy-4-amino-substituted benzopyran resins **24** in good overall yields without significant contamination of the by-products. The desired benzopyran derivatives **18** (9 examples) were finally liberated from the resin **24** using trifluoroacetic acid (TFA).

2.1.3. Solid-Phase Synthesis of 6-Substituted Benzopyran Compounds

Solid-phase synthesis of 6-amino-substituted 2*H*-benzopyrans **25** was reported for the enlargement of diverse points in the benzopyran moiety by Gong *et al.* [56]. The *N*-alkylation [57,58] of the carbamate resin 22 [55] with alkyl halides and lithium *t*-butoxide in dimethyl sulfoxide (DMSO) introduced subsequently to various alkyl substituents in the 6-amino moiety of resins **26**. The desired 6-amino-substituted 2*H*-benzopyran products **25** (16 examples, 87–71% yields) were liberated from the resins **26** with TFA (Scheme 5).

Scheme 5. Solid-phase synthesis of 6-amino-substituted 2*H*-benzopyrans **25** by Gong *et al.* [56].

2.1.4. Solid-Phase Synthesis of 8-Substituted Benzopyran Compounds

The new lead SD-8381 (see Figure 1) with 2*H*-benzopyran was identified as a novel COX-2 inhibitor from in-house HTS [25,26]. The synthesis of SD-8381 derivatives **27** (8-substituted 2*H*-benzopyrans) was carried out on solid-phase parallel synthetic approach to find more potent COX-2 inhibitors by Liao *et al.* (Scheme 6) [59].

Scheme 6. Solid-phase synthesis of 8-substituted 2*H*-benzopyrans 27 by Liao *et al.* [59].

 $\label{eq:Ar = Ph, 4-F-Ph, 4-Cl-Ph, 4-Me-Ph, 4-Et-Ph, 4-EtO-Ph, 4-CH_2CH(CH_3)_2-Ph, 3-F-4-Me-Ph, 3-Me-4-MeO-Ph 4-OH-Ph, 4-CH_2OH-Ph, 4-CO_2H-Ph, 4-NH_2-Ph, 4-NHSO_2CH_3-Ph, 4-SO_2CH_3-Ph, 4-SO_2NH_2-Ph }$

Aa a starting material, 3-bromo-5-chloro-2-hydroxybenzaldehyde (28) was reacted with ethyl trifluoro-methylcrotonate (29) in the presence of K₂CO₃ and *N*,*N*-dimethylformamide (DMF) under nitrogen at 70 °C (Scheme 6). The resulting 2*H*-benzopyran intermediate 30 was produced in 90% yield. After hydrolysis of ester 30 with aqueous lithium hydroxide in THF, acid 31 with an attached point was obtained in 80% yield. The benzopyran acid 31 was then loaded on bromo Wang resin 32 [4-(bromomethyl)phenoxymethyl-polystyrene] in the presence of cesium carbonate/DMA at 60 °C. Subsequently the bromo group on benzopyran 33 was converted into various aromatic substituents via palladium-mediated Suzuki coupling catalyzed by Pd(PPh₃)₄ with a diverse set of aromatic boronic acids [60]. The desired 8-substituted 2*H*-benzopyrnas 27 (17 examples) were cleaved from resin 34 with 95% TFA in CH₂Cl₂ using triisopropylsilane as scavenger. A few analogs of 8-substituted 2*H*-benzopyran 27 (Ar = 4-Et-Ph and 3-Me-4-MeO-Ph) were used for further investigation *in vivo*.

2.2. Solid-Phase Synthesis of Disubstituted Benzopyran Compounds

2.2.1. Solid-phase Synthesis of 2,3-Disubstituted Benzopyran Compounds

Takahashi and co-workers described the efficient solid-phase synthesis of EGCG (see Figure 1) and the combinatorial synthesis of protected methylated epicatechin derivatives **35** (2,3-disubstituted benzopyran derivatives) (Scheme 7) [61,62].

Scheme 7. Solid-phase synthesis of 2,3-disubstituted benzopyrans **35** by Takahashi *et al.* [61,62].

The solid-phase synthetic strategy of 2,3-disubstituted benzopyrans **35** began with the treatment of the aldehyde **36** with bromo Wang resin **32** (1.6 mmol/g) to provide the aldehyde resins **37** [63]. The treatment of the aldehyde resins **37** with the methyl ketones **38** under NaOMe basic conditions provided the solid-supported enone **39**, which underwent epoxidation with *t*BuOOH to give the solid-supported epoxide **40**. The regioselective epoxide-ring opening [64] of **40** with 1-dodecanethiol in the presence of Zn(OTf)₂ proceeded without cleavage of the Wang linker to afford the solid-supported α-hydroxyketone **41**, the acylation of which with benzoic acids (**42**) then gave the precursor **43** for reductive cyclization. The exposure of 15%TFA in CH₂Cl₂ in the presence of a PS-benzaldehyde resin followed by the addition of triethylsilane promoted the cleavage of the Wang linker, reduction of the sulfide and the bromide, and reductive etherification to provide the protected 2,3-disubstituted benzopyrans **44**. Finally, the 2,3-disubstituted benzopyran derivatives **44** were deprotected by conventional hydrogenolysis in the solution-phase by using a palladium catalyst to provide EGCG derivatives **35** (64 examples).

The two aldehydes **36**, six ketones **38**, and five carboxylic acids **42** were used as building blocks for the synthesis of 60 members of 2,3-disubstituted benzopyran library. The purity of the library was estimated by LC-MS analysis (58–15% purities). The growth-inhibitory effects of the resulting library compounds were examined [62]. Most of the 7-OMe derivatives exhibited biological activity comparable to that of the naturally occurring EGCG.

2.2.2. Solid-Phase Synthesis of 2,6-Disubstituted Benzopyran Compounds

Gong and co-workers reported the construction of a 2,6-difunctionalized 2*H*-benzopyran library of 1,200 analogues by using the solid-phase protocols [65]. An alternative linker-based synthetic strategy

was developed because of a restriction that a carbamate linker based solid-phase synthetic pathway to generate a substituted benzopyran library (18 and 25) could not be introduced at the 2-position of the benzopyran system by using strong bases. In the strategy, acid sensitive methoxy benzaldehyde (AMEBA) resin 45 [2-(4-formyl-3-methoxyphenoxy)ethyl polystyrene from Merrifield resin] [66] was selected as the polymer support since the secondary amino group, resulting from reductive amination, should be highly reactive towards various alkyl halides, acid halides, isocyanates, and sulfonyl chlorides (Scheme 8). Moreover, the final products should be readily cleaved from the support by using dilute TFA solutions [67,68]. In the first step of the sequence, 6-aminobenzopyran resin 46 was prepared by reaction of AMEBA resin 45 with 6-aminobenzopyran 47 [69] under reductive amination conditions [70] [NaBH(OAc)₃ in DMF containing 1% acetic acid].

Scheme 8. Solid-phase synthesis of 2,6-difunctionalized benzopyrans **49**, **53**, and **547** by Gong *et al.* [65].

 $\mathbf{R^1}$ = *n*-Bu, *neo*-amyl, -CH=C(CH₃)₂, cyclopropyl, cyclohexyl, 2-Cl-Ph, 4-Cl-Ph, 2-F-Ph, 4-F-Ph, 4-MeO-Ph, 2-EtO-Ph, 2-Et-Ph, 4-NO₂-Ph, 4-*t*Bu-Ph, 3,4-di-MeO-Ph, 2,6-di-Me-Ph, 2-thiophenyl, 2-naphthyl, -CH₂CO₂(CH₂)₃CH₃,

 R^2 = Bn, 2-Cl-Bn, 3-Cl-Bn, 3-F-Bn, 4-tBu-Bn, 4-Me-Bn, 4-CF₃-Bn, 4-CN-Bn, allyl, crotyl

R³ = Me, -CH=C(CH₃)₂, 2-Cl-Ph, 4-Cl-Ph, 2-F-Ph, 4-F-Ph, 2-CF₃-Ph, 2-furanyl, 2-thiophenyl

In the first-generation diversification step, the secondary amine group in **46** was transformed into the amide, sulfonamide, or urea groups in resin **48** by respective reactions with acid chlorides, sulfonyl chlorides, and isocyanates in the presence of triethylamine in DMF. To confirm the product formation, resin **48** was treated with 20% TFA in CH_2Cl_2 to give 6-amino-substuted 2*H*-benzopyran **49** (25 examples, 85–65% yields, and 99–72% purities).

For the purpose of second-generation diversity, resins **50** containing a free primary hydroxyl group were prepared by reaction of resins **48** with NaOMe in MeOH/THF at room temperature [71]. Functionalization of the hydroxyl groups in resins **50** was promoted by reactions with alkyl halides and acid chlorides to generate respective 2,6-diffuctionalized 2*H*-benzopyran resins **51** with an ether-substituent and **52** with an ester-substituent at position 2 in the 2*H*-benzopyran moiety. Alkylation reactions of **50** were carried out in the presence of lithium *tert*-butoxide in DMF and took place smoothly to yield the corresponding ethers. Subsequent treatment of the resins **51** with 20% TFA in CH₂Cl₂ produced the desired 2,6-difuctionalized 2*H*-benzopyran derivatives **53** with an ether-substituent (the representative 22 examples, 73–26% yields, and 99–83% purities) in high four-step overall yields from resin **46**. The ester-containing resins **52** were prepared by treatment of resins **50** with various acid chlorides in the presence of DBU and 4-dimethylaminopyridine (DMAP) in DMF. To confirm product formation, the resins **52** were treated with 20% TFA in CH₂Cl₂ to yield the desired 2,6-difuctionalized 2*H*-benzopyrans **54** with an ester-substituent (the representative 21 examples, 73–29% yields, and 99–81% purities).

Also, Gong and co-workers developed the solid-phase parallel synthesis of the additional 2,6-difunctionalized 2*H*-benzopyrans **55** and **56**, which provided a 2,000-member library of novel 6-alkylamino-2-(functionalized-aminomethyl)-2*H*-benzopyrans [72]. The overall synthetic strategy used to prepare the target 2*H*-benzopyran analogues **55** and **56** is outlined in Scheme 9. The Fmoc-protected 2*H*-benzopyran amine **57** was prepared from 2-dimethoxymethyl-2-methyl-2*H*-1-benzopyran **58** [31] using the reaction sequences by acetal deprotection, reductive amination with methylamine, secondary amine protection with Fmoc-Cl, and reduction of nitro-group.

In the parallel solid-phase protocol (Scheme 9), 6-aminobenzopyran resin **59** was prepared from the backbone amide linker (BAL) resin **60** [73–75] by reaction with 6-aminobenzopyran **57** under reductive amination conditions with NaBH(OAc)₃ [70].

In the first-generation diversification step, resin **61**, containing a secondary amine group, was reacted with alkyl halides in the presence of diisopropylethylamine (DIEA) in CH₂Cl₂. For second-generation diversification, resins **61** containing a secondary amino group were prepared by removal of Fmoc on resins **61** with 20% piperidine in DMF. Functionalization of the secondary amine groups on resins **62** is promoted by reaction with various electrophiles, including acid chlorides, sulfonyl chlorides, isocyanates, and isothiocyanates. This leads to the generation of the respective amide, sulfonamide, urea and thiourea derivatives. Further confirmation of product formation was accomplished by treatment of resins **63** and **64** with 20% TFA in CH₂Cl₂ and by characterization of the liberated 2,6-difunctionalized 2*H*-benzopyrans **11** and **12** with 2-functionalized-aminomethyl group.

Lipinski's rule [76] and similar formulations [77,78] serve as guidelines to estimate the physicochemical properties of the 2,000-member library of 6-alkylamino-2-(functionalized-aminomethyl)-2*H*-1-benzopyran derivatives **55** and **56**. Most of the key parameters for members of the library fall within the range of those predicted for reasonable oral bioavailable drugs by using the commonly known guidelines.

The drug-like small molecule library with variously substituted benzopyrans was constructed by Gong and co-workers [79,80]. An orally active 5-LO [81–87] inhibitor KRH-102140 [30,31] with the benzopyran moiety (see Figure 1) was discovered as a lead compound on the drug discovery program

via HTS of in-house small molecule library by Gong *et al.* [88]. Also, KRH-102140 showed hypoxia-inducible factor (HIF)- 1α [89–93] inhibitory activities [94].

Scheme 9. Solid-phase synthesis of 6-alkylamino-2-(functionalized-aminomethyl)-2*H*-benzopyrans **55** and **56** by Gong *et al.* [72].

 \mathbf{R}^1 = Ph, 3-Cl-Ph, 4-F-Ph, 4-Me-Ph, 3-MeO-Ph, 4-NO₂-Ph, 4-CF₃-Ph, 2-thiophenyl, 5-Me-2-thiophenyl, 2-pyridyl \mathbf{R}^2 = n-Pr, n-Bu, cyclopropyl, Bn, Ph, 4-Me-Ph, 4-fBu-Ph, 4-F-Ph, 4-MeO-Ph, 2-thiophenyl \mathbf{R}^3 = n-Bu, cyclohexyl, Ph, 2-Cl-Ph, 4-F-Ph, 4-Me-Ph, 4-Et-Ph, 4-MeO-Ph, 4-NO₂-Ph, 2,6-di-Me-Ph, -CH₂CO₂Et

On the basis of the biological activities of the benzopyran moiety with KRH-102140, Gong and co-workers demonstrated the solid-phase synthesis of a 222-number library of 2,6-difunctionalized 2*H*-benzopyran **65** and **66** for the lead optimization as a 5-LO inhibitor [80].

As shown in Scheme 10, resin-bounded spirobenzopyran 67 was prepared by reaction of BAL resin 60 with *N*-[ethylcarbamate-spiro(2*H*-1-benzopyran-2,4-piperidine)-6-yl]amine 68, which was synthesized by general manipulations, under reductive amination conditions. In the first generation diversification step, the secondary amine group on resin 67 was transformed into the tertiary amine and amide on benzopyran resins 68 and 69 at position 6 by the reactions with various acid chlorides and alkyl halides in the presence of bases, respectively.

After the carbamate deprotection in a spiro-ring of benzopyran resins **68** and **69** by hydrolysis reactions, the secondary piperidine amines on the 6-amino- or 6-amido-substituted 2*H*-benzopyran resins **70** and **71** were converted by reactions with various sulfonyl chlorides to generate respective

6-amino-substituted amide resins **72** and 6-amino-substituted resins **73** for the introduction of second generation diversity.

Scheme 10. Solid-phase synthesis of 6-amido- and 6-amino-substituted-2-functionalized benzopyrans **65** and **66** by Gong *et al.* [80].

 R^1 = cyclohexyl, 2-Cl-Ph, 2-F-Ph, 3-MeO-Ph, 4-MeO-Ph, 3-NO₂-Ph, 4-NO₂-Ph, 4-CN-Ph, 2-CF₃-Ph, 4-Ph-Ph, 2-furanyl - Cl-Ph, 2-F-Ph, 3-MeO-Ph, 3-MeO-Ph, 3-NO₂-Ph, 4-NO₂-Ph, 4-CN-Ph, 2-CF₃-Ph, 4-Ph-Ph, 2-Furanyl - Cl-Ph, 2-CF₃-Ph, 4-Ph-Ph, 2-Furanyl - Cl-Ph, 2-CF₃-Ph, 4-Ph-Ph, 2-Furanyl - Cl-Ph, 3-MeO-Ph, 3-MeO-Ph, 3-MeO-Ph, 3-NO₂-Ph, 4-NO₂-Ph, 4-CN-Ph, 2-CF₃-Ph, 4-Ph-Ph, 2-Furanyl - Cl-Ph, 3-MeO-Ph, 3-MeO-Ph, 3-MeO-Ph, 3-NO₂-Ph, 4-NO₂-Ph, 4-CN-Ph, 3-Ph, 4-Ph-Ph, 2-Furanyl - Cl-Ph, 3-MeO-Ph, 3-MeO-Ph, 3-NO₂-Ph, 4-NO₂-Ph, 4-CN-Ph, 3-NO₂-Ph, 4-Ph-Ph, 3-Ph, 4-Ph-Ph, 3-Ph-Ph, 3-NO₂-Ph, 4-NO₂-Ph, 4-NO₂-Ph, 4-NO₂-Ph, 4-NO₂-Ph, 4-NO₂-Ph, 4-Ph-Ph, 3-Ph-Ph, 3-Ph-Ph, 3-NO₂-Ph, 4-NO₂-Ph, 4-NO

R² = *n*-Pr, *i*-Bu, Bn, 3-Cl-Bn, 4-Cl-Bn, 2-F-Bn, 3-F-Bn, 4-F-Bn, 2-MeO-Bn, 4-MeO-Bn, 2-NO₂-Bn, 4-NO₂-Bn, 4-CN-Bn, 2-pyridinomethyl, 4-pyridinomethyl, 2-thiophenomethyl

 R^3 = Me, n-Pr, i-Pr, 4-Cl-Ph, 4-F-Ph, 4-Me-Ph, 4-MeO-Ph, 4-NO₂-Ph, 2-thiophenyl

Finally, the liberation of solid support on the resins **72** and **73** with 20% TFA in CH₂Cl₂ gave the desired drug-like 2-functionalized-6-amido-substituted 2*H*-benzopyran derivatives **65** (the representative 25 examples) and 2-functionalized-6-amino-substituted 2*H*-benzopyran derivatives **66** (the representative 25 examples), respectively.

In general, the goal of a drug discovery process is to synthesize chemical entities which are orally bioavailable; *i.e.* they possess physiological properties that allow them to be absorbed into the gastrointestinal system. Lipinski's Rule [76] and similar formulations [77,78] served as guidelines to estimate the physicochemical properties of the synthesized 222-member library of 6-amido- and 6-amino-substituted-2-functionalized benzopyrans **65** and **66**, respectively [80].

2.3. Solid-Phase Synthesis of Trisubstituted Benzopyran Compounds

2.3.1. Solid-Phase Synthesis of 2,4,6-Trisubstituted Benzopyran Compounds

Breitenbucher and Hui developed the titanium-mediated reductive amination procedure for the practical solid-phase synthesis of 2,4,6-trisubstituted benzopyran derivatives **74** (2,4,6-trisubstituted chromane) [95]. The three different 6-carboxybenzopyran-4-one scaffolds **75** [96] was attached to Merrifield/hydroxythiophenol resin **76** via diisopropylcarbodiimide (DIC) coupling, to provide benzopyranone resins **77** (Scheme 11). Treatment of resin **77** with Ti(O*i*Pr)₄ and a primary amine

(eight R¹NH₂) in toluene provided a tetrahedral intermediate resin **78**. Na(OAc)₃BH was then added to the reaction to effect reduction to the amine on resin, followed by washings to afford 4-amino-benzpyran resins **79**.

Scheme 11. Solid-phase synthesis of 2,4,6-trisubstituted benzopyrans **74** by Breitenbucher and Hui [95].

Acylation (16 acylating agents) of the 24 different resins **79** was then performed by addition of either an isocyanate, or acid chlodde to afford 4-disubstituted benzopyran resins **80**. Cleavage from support was accomplished by treatment with 4 equivalents of an amine (22 amines) in pyridine at room temperature. The resulting library products **74** (8,448 spatially separated benzopyrans) were then concentrated and subjected to supported liquid extraction (SLE) [97,98] to remove the excess cleaving amines from the products. Their average purity of samples measured by LC-MS analysis of the crude products was around 73%.

2.3.2. Solid-Phase Synthesis of 3,4,6-Trisubstituted Benzopyran Compounds

Gong and co-workers described the construction of a 3,4,6-trisubstituted benzopyran library of 2,000 analogues using consecutive nucleophilic addition *via m*-CPBA epoxidation on solid support [99,100]. Various reaction conditions were examined to find a condition of the nucleophilic alcohol addition at an epoxide on resin and completion of 3-hydroxy-4-alkoxy-6-amino-substituted benzopyrans **81** on solid support. The desired 3-hydroxy-4-alkoxy-benzopyran resins **82** were obtained by the consecutive nucleophilic alcohol addition reactions of resins **26** with nucleophiles, immediately followed by *m*-CPBA epoxidation. Finally, the cleavage of resins **82** with 25% TFA in CH₂Cl₂ produced the target 3-hydroxy-4-alkoxy-6-amino-substituted benzopyrans **81** (26 examples) without significant contamination of by-products (Scheme 12).

Scheme 12. Solid-phase synthesis of 3,4,6-trisubstituted benzopyrans 84 and 86 by Gong et al. [99].

R¹ = H, Bn, 4-MeO-Bn, 4-Me-Bn, 4-t-Bu-Bn, 4-F-Bn, 3-F-Bn, Me, pentyl

 \mathbf{R}^2 = Me, Et, *i*-Pr, *n*-Bu, Bn, -(CH₂)₂Ph, -(CH₂)₂C₆H₁₁

 \mathbf{R}^3 = Me, pentyl, allyl, propagyl, Bn, 2-Cl-Bn, 3-Cl-Bn, 3-F-Bn, 4-F-Bn, 4-Me-Bn, 2-naphthyl-CH₂-, 3,5-di-CF₃-Bn

R⁴ = Me, neo-amyl, cyclohexyl, acryl, Ph, 4-F-Ph, 4-Me-Ph, 2-furanyl, 2-thiophenyl

After confirmation of the consecutive regioselective nucleophilic addition *via m*-CPBA epoxidation on solid support, the 3-hydroxyl benzopyran resins **82** were scrutinized for enlargement of diverse points in the benzopyran moiety. The reactions with alkyl halides and acid chloride were examined to generate ethers and esters in 3-hydroxyl benzopyran resins **82**. To prepare the ether-type at position 4 in resins **83**, the resins **82** were treated with various alkyl and benzyl halides in the presence of lithium *t*-butoxide in DMF. The reaction proceeded nicely to provide polymer-bounded benzopyrans **83** with an ether-substituent, and subsequent treatment of the resins **83** with 25% TFA in CH₂Cl₂ produced the desired 3,4,6-trisubstituted benzopyrans **84** with an ether-substituent (the representative 26 examples) in good four-step overall yields.

For the preparation of esters at position 4 in the benzopyran moiety, the resins **82** were treated with various acid chlorides with pyridine and DMAP as bases in CH₂Cl₂, to produce the benzopyran resins **85** with an ester-substituent, which were again treated with 25% TFA in CH₂Cl₂ h to give the desired 3,4,6-trisubstituted benzopyrans **86** with ester-substituent (26 representative examples) in good four-step overall yields. The obtained 3,4,6-trisubstituted benzopyrans **84** with an ether-substituent were identified as prolyl 4-hydroxylase inhibitors via a screening process using HSC-T6 and LI 90 cells that express an immortalized rat hepatic stellate cell line and as part of a test of the type I collagen contents employing the ELISA method [100].

3. Solid-Phase Synthesis of Additional Cycle-Fused Benzopyran Compounds

3.1. Solid-Phase Synthesis of Tricyclic Benzopyran Compounds

3.1.1. Solid-Phase Synthesis of Isoxazole-Fused Benzopyran Compounds

Collins and co-workers developed the solid-phase synthesis of isoxazole-fused benzopyrans (benzopyranoisoxazoles) **87** as potential steroid mimetic templates using intramolecular 1,3-dipolar cycloaddition with a tethered alkyne [101].

The reductive amination reaction of the commercially available AMEBA [66] resin **45** (2-(4-formyl-3-methoxyphenoxy)ethyl polystyrene) and butylamine with sodium triacetoxyborohydride gave

polymer-bounded secondary amine **88** (Scheme 13). 3-Formyl-4-hydroxybenzoyl moiety loaded resin **89** was introduced by amide formation of amine resin **88** and 3-formyl-4-hydroxybenzoyl chloride (**90**) in the presence of 2,6-lutidine. The various Mitsunobu reactions [102–106] were explored for a polymer-bounded aldoxime containing a phenol-tethered alkyne **94**. A Mitsunobu reaction between polymer-bounded phenol **89** and propargyl alcohol **91** with R² and R³ using sulfonamide betaine **92** gave disubstituted resins **93**. The isoxzole precursor resins **94** with aldoxime and alkyne were obtained upon treatment of aldehyde resins **93** with hydroxylamine hydrochloride and triethylamine.

Scheme 13. Solid-phase synthesis of isoxazole-fused benzopyrans 87 by Collins and co-workers [101].

An intramolecular 1,3-dipolar cycloaddition of **94** with NBS and Et₃N in DMF generated polymer-bounded benzopyranoisoxazoles **95**. After cleavage with 15% TFA in CH₂Cl₂, the target benzopyranoisoxazoles **87** (9 examples) were obtained.

3.1.2. Solid-Phase Synthesis of Pyrazole-Fused Benzopyran Compounds

Park and co-workers developed the solid-phase construction of a pyrazole-fused benzopyran (benzopyranopyrazole) library with the recombination of privileged benzopyran and pyrazole structures [107]. The solid-phase synthesis was started from the functionalized benzopyranopyrazoles **96** (Scheme 14). A 7-fluoro-2,2-dimethyl-2,3-dihydrochromen-4-one **97** underwent nitration with potassium nitrate to introduce the aniline moiety in a masked form, followed by nucleophilic aromatic substitution of monoprotected Cbz-piperazine at the fluoride position because of its rich body of biological effects. The resulting chromenone **98** was converted to s-*cis* enone **99** by the sequent reactions of acetal formation [108–110] with triethyl orthoformate in high yield (93%) and I₂-catalyzed acetal deprotection [111–117] (84% yield). In the case of an electron-deficient chromenone **98**, the acidic proton at the C-3 position was readily removed by treatment with sodium methoxide (the reaction condition for hydroxysubstituted s-*cis* enone **99**), which led to the decomposition of benzopyran itself.

Scheme 14. Solid-phase synthesis of pyrazole-fused benzopyrans **104** by Park *et al.* [107].

The regioselective synthesis of benzopyranopyrazole derivatives **100** was achieved by the condensation of a β-keto aldehyde with mono-substituted hydrazine (R¹BocNNH₂) in AcOH [108,109]. The Cbz protection group on the piperazine moiety was removed from benzopyranopyrazole **96** by 40% KOH or dimethyl sulfate and BF₃·OEt₂ [118] for the immobilization of the piperazinyl secondary amine **96** on a solid support.

As shown in Scheme 14, Wang resin was activated with *p*-nitrophenylchloroformate in the presence of DIPEA, followed by the loading of the piperazinyl secondary amine **96** on the solid support. The nitro group on polymer-bounded intermediates **101** was reduced with tin chloride dihydrate in DMF. The resulting aniline resins **102** were subsequently diversified with a set of 12 building blocks (six carboxylic acids, one isothiocyanate, two isocyanates, and three sulfonyl chlorides) identical to that used for the modification at the R² position. The final cleavage step with resins **103** was performed under 50% TFA in dichloromethane to liberate various benzopyranopyrazoles **104** (96 examples). Overall, the average purity of the final 96 benzopyranopyrazoles **104** with R¹ and R² diversification was 84%.

3.1.3. Solid-Phase Synthesis of Six-Membered Ring-Fused Benzopyran Compounds

A novel solid-phase synthetic approach toward tricyclic benzopyrans **105–109** (cannabinoid derivatives) is described by Bräse and Kapeller [119]. The synthetic approach for the solid-phase synthesis of tricyclic benzopyrans **105–109** was to immobilize the commercially available 4-hydroxysalicylaldehyde (4-HSA) with Ellman's acid-labile DHP-linker **110** [120,121], which was prepared by etherification of a 3,4-dihydro-2*H*-pyran-2-methanol (**111**) and Merrifield resin **112** [122] (0.99 mmol/g). At first, polymer **110** was treated with 4-HSA/PPTS (pyridinium *p*-toluenesulfonate) to give polymer-bounded aldehyde **113** (Scheme 15).

Scheme 15. Solid-phase synthesis of tricyclic benzopyrans 105–109 by Bräse and Kapeller [119].

To generate the benzopyran core structure, the domino oxa-Michael-aldol (DOMA) condensation [123–125] for the resin 113 and α,β-unsaturated carbonyl moieties 114 as Michael acceptors gave polymer-bounded benzopyrans 115 by employing K₂CO₃ in 1,4-dioxane at 80 °C. The next step for a diene moiety was either Wittig reaction with CH₃PPh₃Br to 116 or TBS-enol ether formation yielding 117. For the Diels-Alder reaction, a thermal condition at 80 °C was chosen, as Lewis or Brønsted acid catalysis would lead to concomitant cleavage from the resin. The Diels-Alder reaction of diene resins 116 and various dienophiles (see Figure 4) gave tricyclic benzopyran resins 118 and 119 in this thermal condition. After cleavage of resins 118 and 119 with PPTS in DCE/EtOH, the desired tricyclic benzopyran derivatives 105 and 106 (14 examples, 95–11% yields) were obtained.

Also, enol ether resin 117 was subjected to the Diels-Alder reaction. The resulting tricyclic benzopyran resins 120 with enol ether were at first cleaved with PPTS in DCE/EtOH giving ketones 107 directly under concomitant loss of the TBS-group. When switching the solvent to THF/water, but the TBS group was only partially cleaved-off, mainly giving enol-ethers 121. Further treatment with TBAF in acetonitrile finally liberated tricycles 107 (4 examples, 51-10% yields) in all cases except for 108 ($R^5 = CO_2iPr$), where side-product 109 was formed exclusively.

Figure 4. Diverse dienophile reagents for the Diels-Alder reaction.

Lee and co-workers described the asymmetric solid-phase parallel synthesis of (3'R,4'R)-di-*O-cis*-acyl 3-carboxyl khellactones (tricyclic benzopyrans) as potent anti-HIV agents [126,127]. The strategy was begun with ethyl malonate bound resin **122**, which was introduced by a reaction of Wang resin **20** and ethyl potassium malonate (Scheme 16). The resin-bounded tricyclic benzopyran **123** was prepared by a Knoevenagel condensation [128–131] between ethyl malonate resin **122** and *o*-hydroxy-arylaldehyde **124** in pyridine and piperidine. The *o*-hydroxyarylaldehyde **124** was prepared from an acetyl acetaldehyde dimethyl acetal using the reaction sequences by Grignard addition, nucleophilic substitution of 2,4-dihydroxybenzoylaldehyde, and regiospecific aromatic cyclization.

Scheme 16. Asymmetric solid-phase synthesis of tricyclic benzopyrans **127** by Lee *et al.* [126,127].

The polymer-bounded 3,4-diacyl-substitued tricyclic benzopyrans **125** were obtained by acylation with various carboxylic acids and an optically active *cis*-diol resin **126** from the Sharpless asymmetric dihydroxylation (AD) reaction of resin **123**. The treatment of diacyl resins **125** with TFA in CH₂Cl₂ gave the desired tricyclic benzopyran derivatives **127** (6 examples, 44–24% yields, and >90% purities).

3.2. Solid-Phase Synthesis of Polycyclic Benzopyran Compounds

Novel polycyclic scaffolds **128–130** containing the benzopyran moiety with variable substituents were described by Park *et al.* [51]. The excellent *endo*-selective Diels-Alder reaction with dienophiles **131** (17 substituted maleimides) and solid-supported dienes **132** (8 resins from **2**, see Scheme 2 and Figure 3), which were derived from palladium-mediated Stille-type vinylation on vinyl triflate intermediate **12**, gave benzopyran-containing polycycloheterocyclic resins **133** (Scheme 17). After HF/pyridine cleavage of resins **133** and subsequent quenching with TMSOEt, diastereomerically enriched novel tricyclic benzopyran derivatives **128** were obtained on a scale of 10 mg each (136 examples). Their average purity measured by LC-MS analysis of the crude products, was around 85%.

Scheme 17. Solid-phase synthesis of polycyclic benzopyrans 128–130 by Park et al. [51].

To expand the molecular diversity of the small-molecule library, novel polycyclic benzopyran derivatives 128 were transformed to discrete core skeletons 129 and 130, using chemical transformations such as Pd/C-based diastereoselective hydrogenation of 128 by the library-to-library approach and the sequence reaction of 2,3-dichloro-5,6-dicyanobenzoquinone (DDQ)-mediated aromatization of resins 133 and liberation of resins 134, respectively. The average purity of 129 (54 examples), which obtained from 12 without further purification, was about 81%, and that of 130 (36 examples) measured by LC-MS analysis of the crude products, was 84%.

The shape of the resulting core skeleton 129 is structurally discrete and more concave than that of its precursor 128 because of the conversion at the monoene site of sp^2 carbon to sp^3 carbon. Compared to heterogeneous hydrogenation, which introduces the sp^3 carbon center in an asymmetric fashion, the aromatization using DDQ can remove the existing stereogenic carbon centers of the monoene precursor 128 and provide a new flatter core skeleton 130.

4. Solid-Phase Synthesis of Miscellaneous Benzopyran Compounds

The selenium-mediated solid-phase syntheses of various 2,2-dimethylbenzopyran derivatives 131 were published by Nicolau *et al.* [132–138]. The solid-phase synthetic strategy started from the treatment of selenyl bromide resin 132 [139,140] with a three-fold excess of various *ortho*-prenylated phenols 133 in CH_2Cl_2 . The dihydrobenzopyran resins 135 were produced via a 5-*endo-trig* cycloaddition of 134. The desired 2*H*-benzopyrans 131 (45 examples) were obtained in high yields with high purities by oxidation to selenium oxide with H_2O_2 and *syn*-elimination of intermediate resin 136 in a traceless manner (Scheme 18).

Scheme 18. Solid-phase synthesis of benzopyrans **131** by Nicolau *et al.* [132–138].

PS SeBr + R
OH

$$CH_2Cl_2$$
, rt

 CH_2Cl_2 , r

After preliminary studies, their utility was demonstrated for the syntheses of the benzopyran-based privileged structures **137a–l** (natural benzopyrans, multi-substituted and polycyclic benzopyrans, and so on) from polymer-bounded 2-selenodihydrobenzopyrans **135** (Scheme 19).

Scheme 19. Benzopyran-based privileged structures from 2-selenodihydrobenzopyran resins **135** by Nicolau *et al.* [132–138].

The acquired polymer-bounded benzopyrans **135** were subsequently used as scaffolds in the synthesis of several benzopyran-based combinatorial libraries (for example, a 52-member library [137] aimed at the development of new NADH/ubiquinone oxidoreductase inhibitors, and a 10,000-member library [135] constructed by directed split-and-pool chemistry).

5. Summary

The combinatorial synthesis of drug-like small organic molecules plays a significant role in the area of drug discovery. Especially, the various natural and artificial benzopyran compounds as bioactive molecules have proven to be broadly useful as therapeutic agents because of their high degree of structural diversity. In this respect, many synthetic methods have been developed for fabricating the privileged benzopyran structures with drug-like properties by using solid-phase synthetic strategies. In this article, we have introduced the preparation of diverse and drug-like benzopyrans as substituted benzopyrans, additional cycle-fused benzopyrans, and their related compounds. Further studies in this area are underway and the various strategies for syntheses of benzopyran derivatives on solid support will be reported for medicinal chemistry and drug discovery.

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