

Article

Reaction of bis[(2-chlorocarbonyl)phenyl] Diselenide with Phenols, Aminophenols, and Other Amines towards Diphenyl Diselenides with Antimicrobial and Antiviral Properties †

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Abstract: A reaction of bis[(2-chlorocarbonyl)phenyl] diselenide with various mono and bisnucleophiles such as aminophenols, phenols, and amines have been studied as a convenient general route to a series of new antimicrobial and antiviral diphenyl diselenides. The compounds, particularly bis[2-(hydroxyphenylcarbonyl)]phenyl diselenides and reference benzoselenazol-3(2*H*)-ones, exhibited high antimicrobial activity against Gram-positive bacterial species (*Enterococcus* spp., *Staphylococcus* spp.), and some compounds were also active against Gram-negative *E. coli* and fungi (*Candida* spp., *A. niger*). The majority of compounds demonstrated high activity against human herpes virus type 1 (HHV-1) and moderate activity against encephalomyocarditis virus (EMCV), while they were generally inactive against vesicular stomatitis virus (VSV).

Keywords: ebselen; organoselenium compounds; diselenides; benzoselenazol-3(2*H*)-ones; antimicrobial activity; antiviral activity; morpholine; benzohydrazide; 2-hydroxyacetanilide; antimycin A

1. Introduction

Organoselenium chemistry and biology have been areas of continuous research over the last few decades [1–17]. Selenium-containing molecules have been of interest as reagents in new synthetic routes [1], promising catalysts of various oxidation reactions [1–7,18–21], and bioactive agents [8–16,22,23]. In the context of biological applications, organoselenium compounds have received particular attention as glutathione peroxidase (GPx) mimics [9,11,14,24], antioxidant, and anti-inflammatory agents [10,11,13,23]. Furthermore, a number of selenium-containing compounds have been studied as anticancer [12,22,25], anti-angiogenic [26], antiviral [27–35], antimicrobial [32–46], and antileishmanial agents [47].

The development of novel anti-infective agents is an area of intense activity due to the constant appearance of new infections, the emergence of multidrug resistance in common pathogens, as well as the need to eliminate undesirable side effects of available drugs. The urgent need for the next generation of antimicrobials and antivirals perpetuates research into designing novel molecules and investigating their mechanisms of action. The tremendous efforts in this area has led to the identification of many bioactive structures [27–46]. Several enzymes, among them thioredoxin reductase [36,48], bacterial urease [49], and plasma membrane H⁺-ATPase [50–52], were suggested as targets for organoselenium compounds with antimicrobial properties. Furthermore, it has been reported that the antiviral effects of organoselenium compounds might be related to the inhibition of hepatitis C virus NS3 helicase [28], HIV-1 capsid dimerization [29], and nucleocapsid protein 7 [30].

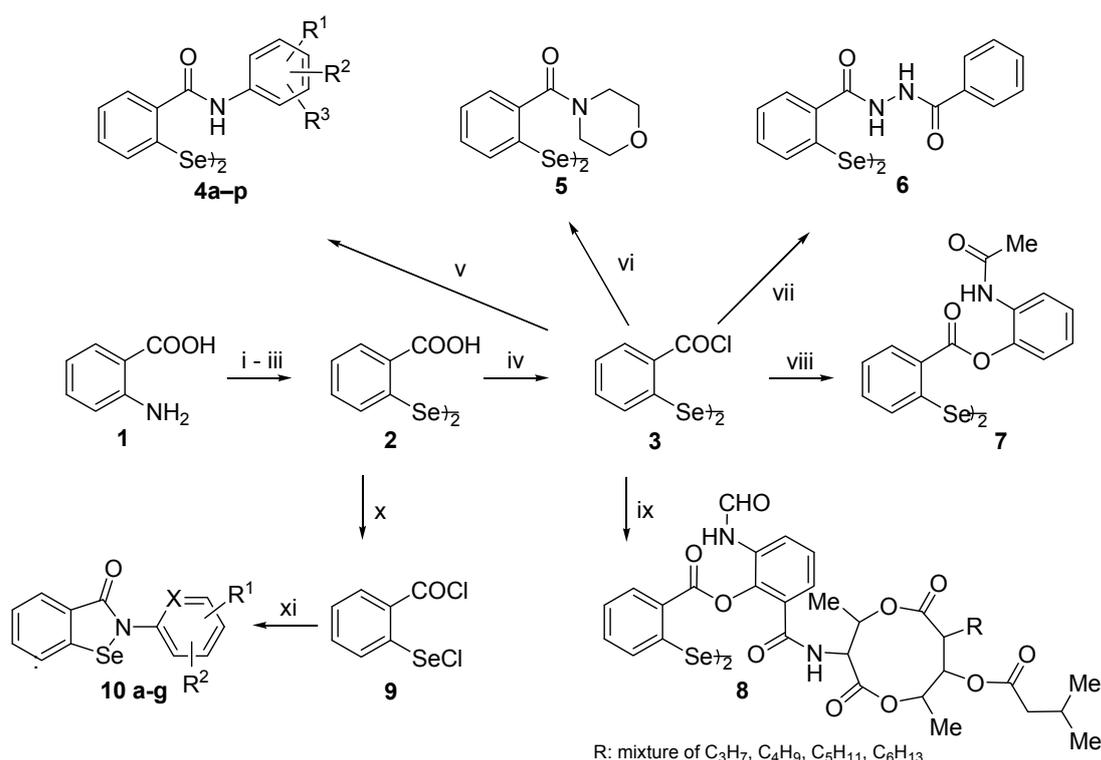
Inspired by the encouraging findings in the area of antimicrobial and antiviral properties of organoselenium compounds and as a continuation of our interest in the chemistry and biology of this class of compounds, we aimed to develop a series of new diphenyl diselenides and investigate their activity against various microbial strains and viruses, along with selected cyclic analogues. It seemed interesting to focus our attention on a group of 2,2'-dicarbamoylphenyl diselenides **4** as the most reduced analogues of ebselen, with a phenyl ring containing hydroxy and/or methoxy moieties. However, the rationale behind studying these compounds was based not only on the fact that the structural modifications in the diphenyl diselenide scaffold seemed to be interesting from the medicinal chemistry perspective; we also aimed to investigate a synthetic route to reduced forms of ebselenols (2-(hydroxyphenyl)-1,2-benzisoselenazol-3(2*H*)-ones) and its derivatives via the reaction of bis[(2-chlorocarbonyl)phenyl] diselenide **3** with various mono and bisnucleophiles such as aminophenols, phenols, and amines.

Up to the present time, several methods have been applied for the preparation of bis(2-arylcarbonyl)phenyl diselenides **4**. One path is the reductive benzisoselenazolone ring opening of ebselenols and its derivatives [53], which can be obtained from 2-(chloroseleno)benzoyl chloride with moderate to excellent yields [54–56]. An alternative route to ebselenols requires metalation-selenenylation of appropriate benzanilides while protecting the hydroxyl group in the first step [57,58]. Although the reductive ring opening reaction has been successfully applied for the reduction of ebselen and its various alkyl analogues [59], in the case of ebselenols, yields were less satisfactory [53]. Only bis[2-(4-hydroxyphenylcarbonyl)]phenyl diselenide was prepared in 46% yield while bis[2-(2-hydroxyphenylcarbonyl)]phenyl diselenide and 2-(3-hydroxyphenylcarbonyl)]phenyl diselenide were obtained in very low yields of 8.3% and 12%, respectively [53]. Another, general route to ebselenol-derived diselenides which we present in this study is the reaction of bis[(2-chlorocarbonyl)phenyl] diselenide **3** with the corresponding anilines. In the past, this reaction was successfully applied to obtain various 2,2'-dicarbamoyldiphenyl diselenides [60]. However, to our best knowledge, there is only one example of a reaction with an electron-rich arene system bearing a hydroxyl group [26] but with no synthetic details. Thus, we aimed to investigate the behaviour of bis[(2-chlorocarbonyl)phenyl] diselenide **3** in the reaction with *N* and *O* nucleophiles, such as aminophenols, phenols, and amines. The aminophenols were of particular interest as the ones containing two nucleophilic centers in the molecule, with both amine nitrogen and hydroxide oxygen potentially susceptible toward acylation. The reaction of various *O* and *N* nucleophiles with another selenium-containing chloride, 2-(chloroseleno)benzoyl chloride **9**, showed the preference of a primary amino group toward selenenylation-acylation [55]. As chloride **9** contains both hard and soft electrophilic centers localized on the carbonyl carbon and selenium atoms, respectively, the results of the reaction with *O*-nucleophiles strongly depended on the nucleophile structure. However, in the absence of the soft electrophilic center, no side reactions should occur. Thus, we expected that the reaction of bis[(2-chlorocarbonyl)phenyl] diselenide **3** with aminophenols could be considered as a convenient alternative for low-efficient reductive ring opening of ebselenols [53] and the preparation from 2,2'-dicarboxydiphenyl diselenide using active ester coupling agents [23,61], where the protection of the hydroxyl group would be recommended.

2. Results and Discussion

2.1. Synthesis

Herein we present our investigation of synthetic route to diselenides **4–8** via a reaction of (bis[(2-chlorocarbonyl)phenyl] diselenide), hereafter called chloride **3**, with *N* and *O* nucleophiles, such as a series of various aminophenols and their derivatives as well as other specific reactants such as morpholine, benzohydrazide, 2-hydroxyacetanilide, and antimycin A (Scheme 1, Table 1). The key chloride **3** was prepared in a three-step synthesis starting from anthranilic acid **1**, which was diazotized and reacted with disodium diselenide in a methanolic alkaline medium to give, after acidification, 2,2'-diselenobisbenzoic acid **2** which was further reacted with thionyl chloride [59,60,62].



Reagents: i: NaNO₂, HCl, -7 - +7°C, ii: Na₂Se₂, MeOH, NaOH, -7 - +5°C, iii: HCl, iv: 3,5 eq. SOCl₂, benzene, Δ, v: Ar-NH₂, CH₂Cl₂, K₂CO₃ or Ar-NH₂, MeCN, Et₃N, vi: morpholine (2.5 eq.), CH₂Cl₂, vii: benzohydrazide, Na₂CO₃, CH₂Cl₂, viii: 2-hydroxyacetanilide, Na₂CO₃, CH₂Cl₂, ix: Antimycin A, Na₂CO₃, CH₂Cl₂, x: SOCl₂ (excess), Δ, xi: Ar-NH₂, Na₂CO₃, CH₂Cl₂ or Ar-NH₂, Et₃N, MeCN.

Scheme 1. Synthesis of bis(2-carbamoy)diphenyl diselenides **4a–p**, **5** and **6**, phenolic esters **7–8**, and benzenoselenazol-3(2*H*)-ones **10a–g**.

In a first step, reaction of chloride **3** was tested with various free and protected aminophenols, such as monohydroxyanilines, monomethoxyanilines, and their derivatives containing additional substituents in the benzene ring, such as electron-donating methyl and methoxy groups and electron-withdrawing chloro and carboxymethyl groups. As in the case of 2-(chloroseleno)benzoyl chloride **9**, which was intensively studied with various bisnucleophiles [55], the reaction occurred on the amine group of both protected (**4k–p**) and free phenols (**4a–j**). In the IR spectra, there were observed signals from primary amide bonds at ca. 3300–3500 cm⁻¹ (N–H), carbonyl groups at ca. 1600–1650 cm⁻¹ (C=O), amide groups at ca. 1500–1550 cm⁻¹ (CONH), and in ¹H-NMR there were present protons of N–H group at ca. 9.6–10.5 ppm (see the Supplementary Materials for details). All compounds were prepared with very good to excellent yields, significantly higher than those achieved by the reductive ring opening method [53]. The simplest three isomeric aminophenols reacted

with 73–92% yield towards **4a–c**. Reaction with 2-aminophenols containing electron-withdrawing substituents (Cl and COOMe) at the vicinity carbon atom generally resulted in slightly lower yields (**4g–i**; 57–63%), when compared to those containing electron-donating ones (**4d–f**; 60–89%). As expected [35], methoxy derivatives **4j–p** were also prepared with very good yields (72–99%). As a result of our studies, seven new open-chain analogues of ebselenols **4d–j** and two new methoxy derivatives **4n** and **4p** were developed.

Table 1. Characteristics of compounds **4a–p**, **5–8**, and **10a–g**.

Compound	R ¹	R ²	R ³	X	Rx Time (h)	Yield (%)	m.p. (°C)	m.p. lit (°C)
4a	2-OH	-	-	-	96	92	233.5–236.5	235–238 [53]
4b	3-OH	-	-	-	60	84	285.5–287.0	285–286 [53]
4c	4-OH	-	-	-	120	73	271.0–272.0	282–284 [53]
4d	2-OH	4-Me	-	-	24	60	221.0–223.5	-
4e	2-OH	5-Me	-	-	96	89	191.5–194.5	-
4f	2-OH	6-Me	-	-	168	88	264.0–267.0	-
4g	2-OH	4-Cl	-	-	96	59	230.0–234.0	-
4h	2-OH	5-Cl	-	-	48	63	219.5–221.5	-
4i	2-OH	5-COOMe	-	-	72	57	281.0–282.0	-
4j	3-OH	4-OMe	-	-	120	94	232.5–236.5	-
4k	2-OMe	-	-	-	168	94	211.0–214.0	216–217 [53]
4l	3-OMe	-	-	-	72	98	221.0–223.0	221–223 [35]
4m	4-OMe	-	-	-	48	94	285.5–287.5	290–292 [53]
4n	2-OMe	4-Ome	-	-	168	99	211.0–213.5	-
4o	3-OMe	4-OMe	-	-	96	72	231.5–234.5	235 [53]
4p	3-OMe	4-OMe	5-OMe	-	96	76	193.0–195.0	-
5	-	-	-	-	120	85	149.0–152.0	-
6	-	-	-	-	120	100	296.5 dec.	-
7	-	-	-	-	96	36	178.0–181.0	-
8	-	-	-	-	240	97	>300	-
10a	2-OH	-	-	CH	20	40	195.5–197.5	194–196 [54]
10b	6-OH	-	-	N	2	65	229.0 dec.	199–201 [55] ^a
10c	2-OMe	-	-	CH	24	71	189.0–191.5	172–174 [63]
10d	3-OMe	-	-	CH	2	65	166.0–168.0	165.5–167.5 [35]
10e	4-OMe	-	-	CH	20	70	180.0–181.0	180.5–181.5 [35]
10f	2-OMe	4-Ome	-	CH	6	65	240.5–242.5	-
10g	3-OMe	4-Ome	-	CH	48	68	159.0–161.0	-

^a Phase transfer was observed at 199 °C.

Reaction of benzoyl chloride **3** with secondary amines such as morpholine also resulted in secondary amide bond formation to produce compound **5** with a very good yield of 85%. A similar compound containing piperazine moiety was prepared with a comparable yield [64].

Another highly nucleophilic benzohydrazide reacted quantitatively with chloride **3** on its terminal nitrogen atom to form bishydrazide **6**. Encouraged by this result, we continued our studies with other *N*-nucleophiles, focusing on hydrazine, ethylene diamine, tri(hydroxymethyl)aminomethane, and isoniazide (isonicotinic hydrazide). The resulting oligomeric products were, however, difficult to identify.

Further efforts were directed into the investigation of the behavior of aminophenol derivatives containing blocked amine groups such as 2-hydroxyacetanilide and natural antimycin A. In this case, reaction occurred on the free hydroxyl group only, resulting in phenolic ester derivatives **7** and **8**, however, highly sterically hindered antimycin A required an extremely long reaction time. In the IR spectrum of compound **7**, a characteristic phenolic ester bond was observed at 1760 cm⁻¹. Antimycin A is a mixture of at least four closely related compounds, containing alkyl or acyl substituents of different lengths or degrees of branching (usually C₃–C₆) in the dilactone portion of the molecule [65]. Thus, the spectroscopic analysis is complicated and only diagnostic signals could be reported in ¹³C-NMR spectra. Fortunately, in HRMS spectra, seven diagnostic molecular species (M + Na⁺)

were observed with $R=C_5H_{11}$ being a predominant one, which was consistent with the results of the elemental analysis.

Ebselen and several other benzisosenazol-3(2H)-ones **10a–g**, including two new molecules, **10f** and **10g**, were prepared as a reference for biological tests. The 2,2'-diselenobisbenzoic acid **2** was converted to the 2-(chloroseleno)benzoyl chloride **9** that, upon reaction with corresponding anilines, formed ebselen and benzisosenazol-3(2H)-ones **10a–g** through a tandem selenenylation-acylation sequence (Scheme 1, Table 1) [55,59]. To our best knowledge, compounds **4d–j**, **4n**, **4p**, **5–8**, **10f–g** are new structures and were not previously described in the literature.

2.2. Antimicrobial Activity

The in vitro antimicrobial activity of compounds **4–7**, **10** was screened against four Gram-positive bacterial strains (*Staphylococcus aureus*, *Staphylococcus epidermidis*, *Enterococcus faecalis*, *Enterococcus hirae*), two Gram-negative bacterial strains (*Escherichia coli*, *Pseudomonas aeruginosa*), two yeast-like fungi (*Candida albicans*, *Candida glabrata*), and filamentous fungus (*Aspergillus niger*) in a range of 0.125–128 $\mu\text{g/mL}$. The minimum inhibitory concentration values (MIC) for organoselenium compounds are summarized in Tables 2 and 3, along with reference substances.

Table 2. Antibacterial activity of compounds **4–7**, and **10**.

Compound	Gram-Positive Bacteria ^a				Gram-Negative Bacteria ^a	
	<i>S. aureus</i>	<i>S. epidermidis</i>	<i>E. faecalis</i>	<i>E. hirae</i>	<i>E. coli</i>	<i>P. aeruginosa</i>
4a	16	8	4	8	>128	>128
4b	16	8	2	8	>128	>128
4c	16	16	2	8	>128	>128
4d	>128	128	2	4	>128	>128
4e	16	16	1	2	64	>128
4f	>128	>128	4	8	>128	>128
4g	128	64	2	4	>128	>128
4h	>128	128	>128	>128	>128	>128
4i	>128	128	>128	>128	>128	>128
4j	>128	128	>128	>128	>128	>128
4k	>128	>128	>128	>128	>128	>128
4l	>128	128	>128	>128	>128	>128
4m	>128	>128	8	>128	>128	>128
4n	>128	128	4	64	>128	>128
4o	>128	128	16	64	>128	>128
4p	>128	128	4	32	>128	>128
5	>128	128	>128	>128	>128	>128
6	>128	128	4	8	>128	>128
7	>128	>128	>128	>128	>128	>128
10a	8	4	2	4	32	>128
10b	8	4	1	2	>128	>128
10c	32	32	2	8	128	>128
10d	>128	128	16	32	>128	>128
10e	>128	128	4	32	>128	>128
10f	64	64	2	4	>128	>128
10g	64	32	2	8	>128	>128
Ebs	64	32	2	8	128	>128
AMC	>128	>128	>128	>128	>128	>128
AMP	4	>128	16	>128	16	>128
CHK	>128	>128	128	128	2	>128
ERM	>128	>128	>128	>128	64	>128
STR	32	128	>128	>128	32	128
NAT	>128	>128	>128	>128	16	32

^a MIC—Minimum Inhibitory Concentration ($\mu\text{g/mL}$); AMC—antimycin A; AMP—ampicillin; CHK—chloramphenicol; ERM—erythromycin; STR—streptomycin; NAT—nourseothricin.

Table 3. Antifungal activity of compounds 4–7, and 10.

Compound	Yeasts ^a		Fungi ^a
	<i>C. albicans</i>	<i>C. glabrata</i>	<i>A. niger</i>
4a	>128	>128	>128
4b	>128	>128	>128
4c	>128	>128	>128
4d	>128	>128	>128
4e	32	32	128
4f	>128	>128	>128
4g	>128	>128	>128
4h	>128	>128	>128
4i	>128	>128	>128
4j	>128	>128	>128
4k	>128	>128	>128
4l	>128	>128	>128
4m	>128	>128	>128
4n	>128	>128	>128
4o	>128	>128	>128
4p	>128	>128	>128
5	>128	>128	-
6	>128	>128	-
7	>128	>128	-
10a	16	16	>128
10b	>128	>128	>128
10c	16	32	64
10d	>128	>128	-
10e	64	>128	-
10f	16	64	64
10g	32	64	-
Ebselen	32	64	16
AMC	>128	>128	>128
KLK	>128	>128	-
ITC	>128	>128	-
NAT	64	64	-
AMB	1	1	-

^a MIC—Minimum Inhibitory Concentration ($\mu\text{g}/\text{mL}$); AMC—antimycin A; KLK—ketoconazole; ITC—itraconazole; NAT—nourseothricin; AMB—amphotericin B.

Generally, there is a noticeable tendency that the compounds tested have been more active against Gram-positive bacterial strains than against Gram-negative ones. These results are in line with our previous observations [33–35] and might indicate the possible mode of action of the whole group. Gram-positive bacteria are more susceptible to plasma membrane-binding agents, absorbing them directly into the plasma membrane, whereas a Gram-negative outer membrane may capture such agents before they penetrate through a cell wall and bind into the inner membrane, which usually results in subsequent cell permeabilization [66]. Using propidium iodide, no significant plasma membrane permeabilization was observed in *S. aureus* after treating with ebselen [67], however, membrane binding is not always sufficient for its permeabilization [68]. Membrane binding agents alter plasma membrane biophysics, which may be toxic itself.

The antibacterial activity towards Gram-positive species varied in a range of 1 to >128 $\mu\text{g}/\text{mL}$. Particularly interesting is the strong activity towards *E. faecalis* and *E. hirae*. These enterococci display highest vulnerability to most of the tested compounds and only compounds 4h–l, 5 and 7 (as well as 4m in the case of *E. hirae*) exhibit no activity in the tested range of concentrations. Enterococci seem to be significantly more susceptible towards organoselenium compounds than reference antibiotics, with the exception of 4o and 10d, displaying same MIC as ampicillin towards *E. faecalis*. This makes organoselenium compounds promising for the treatment of enterococci infections, which although

being low virulent bacteria, often cause infections in immunocompromised patients [69]. In the case of staphylococci, it can be noticed that compounds **4a–c** and **10a–b**, containing hydroxyl substituent in the phenyl ring, were the most active against both *S. aureus* and *S. epidermidis*. Benzisosenazol-3(2*H*)-ones **10a** and **10b** were slightly more active than their open-chain analogues **4a–c**. Only two compounds, **4e** and **10a**, showed inhibitory effect against *E. coli*, similar to commonly known antibiotics. None of the compounds showed activity towards *P. aeruginosa* in the screening range. However, it is worth pointing out that this microorganism was also not susceptible toward the majority of reference antibiotics in the tested range.

Contrary to their cyclic analogues, diselenides appeared to be practically inactive towards both *Candida* spp., with the exception of compound **4e**, which also displayed a high inhibitory effect against Gram-positive bacteria and moderate effects against Gram-negative *E. coli* (Table 1). The probable lack of inhibitory effect of the majority of compounds tested might be due to the fact that the antifungal activity of diphenyl diselenides is dependent on the presence of functional groups [40]. However, diphenyl diselenides may still display high activity in reducing germ tube formation, one of the crucial virulent factors of *C. albicans* [38]. Azoles (ketoconazole and itraconazole), commonly used in antifungal therapy, also did not show activity under the test conditions. However, these two compounds are considered static drugs [70], therefore it is difficult to obtain MIC₉₀ values for them. Thus, it is possible that organoselenium compounds which lack inhibitory effects may still display synergistic effects. Benzisosenazol-3(2*H*)-ones **10a**, **10c**, **10f–g**, and **10e** (in the case of *C. albicans* only) inhibited fungal growth in a range of 16–64 µg/mL (Table 2), comparable to nourseothricin. Amphotericin B exhibited a much higher activity than all the benzisosenazol-3(2*H*)-ones tested, however, this highly cytotoxic compound is considered as a drug of choice for the most severe fungal infections, when the patients are close to death [71]. The majority of diselenides have shown no activity against *A. niger* in the tested range. Only compound **4e** showed weak activity against this filamentous fungus. Benzisosenazol-3(2*H*)-ones **4c** and **4f** showed moderate activity towards *A. niger*, comparable to other previously tested cyclic analogues of ebselen [33,34].

The compound **5**, prepared by reaction with a secondary amine, as well as phenolic ester **7** were inactive against all tested microorganisms.

2.3. Antiviral Activity

The antiviral activity of compounds **4**, **5**, **7**, and **10** has been evaluated in vitro towards HHV-1 (human herpes virus type 1, *Herpesviridae*, enveloped virus), EMCV (encephalomyocarditis virus, *Picornaviridae*, non-enveloped virus), and VSV (vesicular stomatitis virus, *Rhabdoviridae*, enveloped virus) in the human cell line A549. The minimum inhibitory concentrations (MICs) for the abovementioned viruses along with cytotoxicity against the testing cell line are summarized in Table 4.

The compounds exhibited high antiviral activity against HHV-1 and moderate to high activity against EMCV, but in almost all cases were inactive towards VSV (Table 4). A similar tendency was observed in our previous studies [32–35]. The anti-HHV-1 activity of most diselenides was in a range of 2–40 µg/mL; only the one containing morpholine moiety showed lower activity at 200 µg/mL. The benzisosenazol-3(2*H*)-ones also showed high activity against HHV-1, in a range of 2–4 µg/mL, comparable to ebselen. The most promising anti-HHV-1 agents would be compounds **4k**, **4n**, and **10f** with higher chemotherapeutic indices (I), indicating that the concentration active against the virus was lower than that showing a cytotoxic effect. Up to the present time, there were only several organoselenium compounds identified with such high chemotherapeutic indices [33]. Generally, the compounds were less active against EMCV and only compounds **4e**, **4o–p**, and **10c** showed activity below 10 µg/mL.

Table 4. Antiviral activity and cytotoxicity of compounds **4**, **5**, **7**, and **10**.

Compound	TCCD ₅₀ ^a	HHV-1		EMCV		VSV	
		MIC ^b	I ^c	MIC ^b	I ^c	MIC ^b	I ^c
4a	4.9	40	0.12	10	0.49	>1000	<0.005
4c	3.7	2	1.8	60	0.06	>1000	<0.004
4d	4.9	2	2.5	>1000	<0.005	>1000	<0.005
4e	4.9	2	2.5	6	0.82	>1000	<0.005
4f	9.8	4	2.5	>1000	<0.01	>1000	<0.01
4g	4.9	2	2.5	40	0.12	>1000	<0.005
4h	4.9	6	0.82	>1000	<0.005	>1000	<0.005
4i	7.3	20	0.37	20	0.37	>1000	<0.07
4j	4.9	4	1.2	20	0.25	>1000	<0.005
4k	312	6	52	400	0.78	>1000	<0.31
4n	58	2	29	600	0.10	>1000	<0.06
4o	4.9	2	2.5	8	0.61	>1000	<0.005
4p	2.4	4	0.61	8	0.30	200	0.01
5	4.9	200	0.02	10	0.49	>1000	<0.005
7	15	20	0.75	400	0.04	600	0.03
10c	9.8	4	2.5	8	1.2	>1000	<0.01
10f	156	2	78	200	0.78	>1000	<0.16
Ebselen	15	2	7.5	10	1.5	>1000	<0.02

^a TCCD₅₀—Tissue Culture Cytotoxic Dose (µg/mL); ^b MIC—Minimum Inhibitory Concentration (µg/mL);

^c I = TCCD₅₀/MIC.

3. Materials and Methods

3.1. Synthesis

General Information

Melting points were determined on an Electrothermal IA 91100 digital melting-point apparatus using the standard open capillary method. ¹H- and ¹³C-NMR spectra were recorded in CDCl₃ or DMSO-*d*₆ on a Bruker DRX 300 Spectrometer (Bruker, Rheinstetten, Germany) (¹H: 300.1 MHz, ¹³C: 75.4 MHz), on a Joel 400 Spectrometer (Joel, Tokyo, Japan) (¹H: 399.8 MHz, ¹³C: 100.5 MHz) or on a Bruker Avance 600 Spectrometer (¹H: 600.6 MHz, ¹³C: 151.0 MHz) at 295 K. Chemical shifts (δ) are given in parts per million (ppm) downfield relative to tetramethylsilane (TMS, Si(CH₃)₄), and coupling constants (*J*) are in Hz. Residual solvent central signals were recorded as follows: CDCl₃, δ_H = 7.263, δ_C = 77.00; DMSO-*d*₆, δ_H = 2.50, δ_C = 39.43. FT-IR spectra were recorded between 4000 and 400 cm⁻¹ on a Perkin-Elmer 2000 FT-IR spectrometer. High-resolution mass spectra (HRMS) were recorded on a Waters LCD Premier XE instrument (Waters, Manchester, UK), and only the [M + H]⁺ or [M + Na]⁺ diagnostic molecular species are reported. Preparative column chromatography was performed on Merck Si60 silica gel (63–200 µm) (Sigma-Aldrich, Saint Louis, MO, USA) as the stationary phase. Analytical thin-layer chromatography (TLC) was performed on PET foils precoated with silica gel (Merck silica gel, 60 F254) (Sigma-Aldrich, Saint Louis, MO, USA), and visualised using UV light (λ_{max} = 254 nm), or by staining with iodine vapours. Selenium powder (100 mesh) (Sigma-Aldrich, Saint Louis, MO, USA) used for Na₂Se₂ preparation had a purity of ≥99.5%. Acetonitrile (MeCN) (POCH, Gliwice, Poland) and methylene chloride (CH₂Cl₂) (POCH, Gliwice, Poland) were distilled over P₂O₅, and CH₂Cl₂ was stored over sodium hydride (NaH) pills. Triethylamine (Et₃N) (POCH, Gliwice, Poland), distilled over NaOH, was stored over NaOH pellets. Methanol (MeOH) (POCH, Gliwice, Poland) was refluxed over magnesium shavings (Mg) in the presence of elemental iodine (I₂) and distilled over Mg(OMe)₂ was formed and stored over 3A molecular sieves (Sigma-Aldrich, Saint Louis, MO, USA). Diethyl ether (DEE) (POCH, Gliwice, Poland) was distilled over a CaH₂ and LiAlH₄ mixture from water batch at ca. 45 °C. Morpholine (POCH, Gliwice, Poland) was distilled over NaOH pellets (POCH, Gliwice, Poland) before use. Anhydrous sodium carbonate (Na₂CO₃) (POCH, Gliwice, Poland) was ground in a mortar before use. Antimycin A natural product (Sigma-Aldrich,

Saint Louis, MO, USA) was used as a mixture of co-crystallized *n*-alkylated compounds with R = C₃–C₆ predominantly. 2-Hydroxy-5-carboxymethylaniline was prepared from 3-amino-4-hydroxybenzoic acid (Sigma-Aldrich, Saint Louis, MO, USA) by direct metoksylation in a hot methanol environment and in the presence of H₂SO₄ (POCH, Gliwice, Poland), as mentioned below. Ebselen was prepared according to the procedure described earlier [59]. Other reagents and starting materials were directly used as obtained commercially. Purity of the products was confirmed by the comparison of their melting point with data given in literature and by spectroscopic methods. The new compounds were fully characterized.

2-Hydroxy-5-carboxymethylaniline. To 2-amino-4-hydroxybenzoic acid (1.53 g, 10 mmol) in dry MeOH (25 mL) was added H₂SO₄ (1.0 mL, 1.8 g, 19 mmol) portionwise during stirring, and the solution formed was stirred in solvent reflux (+65 °C) for 20 h. The reaction mixture was cooled and poured into 0.5% NaOH (200 mL) and kept in an ice/water bath, NaHCO₃ (3.6 g, 44 mmol) was added and the mixture was washed with DEE (100 mL) followed by the addition of 3.5% HCl to a pH of 8.0, and was then extracted with degased DEE. The collected DEE solution was washed with water, dried over Na₂SO₄, and the solvent was distilled off from a water bath to obtain crude 2-hydroxy-5-carboxymethylaniline. Yield: (1.60 g, 96%); m.p.: 114.5–115.0 °C (110 °C [72]); as prisms (benzene, 40 mL/g). IR (KBr, cm⁻¹) ν 3390, 3300 (NH), 2500–3200 (br OH), 1702 (C=O), 1610, 1429, 1306, 1294, 1284 (C–N), 1201 and 1112 (C–O), 858, 764, 421. ¹H-NMR (300 MHz, CDCl₃:DMSO-*d*₆, 5:1, *v/v*): δ 3.81 (s, 3H, OCH₃), 4.24 (br s, 2H, NH₂), 6.75 (d, *J* = 8.2 Hz, 1H, Ar-H), 7.24 (dd, *J* = 8.2 Hz, *J* = 2.0 Hz, 1H, Ar-H), 7.33 (d, *J* = 2.0 Hz, 1H, Ar-H), 9.50 (br s, 1H, OH).

2,2'-Diselenobisbenzoic acid 2. The compound was prepared from anthranilic acid **1** using disodium diselenide with a modified literature procedure [59,60,62]. The anthranilic acid **1** (20.5 g, 0.15 mol) was dissolved in a solution of 37% aq. HCl (30 mL) in distilled water (90 mL) at ca. 75 °C. The solution was cooled in an ice-water bath (0–5 °C) to form anthranilic acid hydrochloride precipitate. To the resulting thick suspension, a cold solution of NaNO₂ (11.4 g, 0.165 mol) in distilled water (90 mL) at ca. –5 °C was added while stirring for 0.5 h, and the reaction was continued for an additional 15 min. The prepared cold solution of anthranilic acid diazonium chloride was added dropwise for 2.5 h to the freshly prepared methanolic solution of disodium diselenide (NaSeSeNa) at –10–7 °C during stirring with nitrogen evolution accompanied by the formation of red selenium. The reaction mixture was warmed to RT (room temperature) and stored overnight. (NaSeSeNa solution preparation: to the suspension of selenium element (11.2 g, 0.15 mol) in NaOH (18 g, 0.45 mol) solution with MeOH (300 mL) in a round-bottom flask equipped with an oil closure, hydrazine monohydrate (1.9 mL, 0.040 mol) was added and gently stirred at RT for 60 h.) The formed grey selenium was filtered, washed with water, and air-dried to obtain selenium powder (6.20 g, 0.083 mol) with 55% yield, which can be reused. The collected filtrates were acidified with 37% aq. HCl (25 mL), gently heated with mixing and left at RT overnight. The solid was filtered off, washed with hot water (ca. 1 L) (until no odor of salicylic acid in the filtrate was detected), and air-dried. The crude product was recrystallized from hot 1,4-dioxane (400 mL) by slow concentration under ca. 200 mmHg to give acid **2**. Yield: (11.1 g, 37%); m.p.: 298–300 °C (300–303 °C [59]); as a pale brick powder (1,4-dioxane).

Bis[(2-chlorocarbonyl)phenyl] diselenide 3. To 2,2'-diselenobisbenzoic acid (**2**) (20 g, 50 mmol) suspended in benzene (300 mL) in a round-bottom flask equipped with an oil closure and gas trapping, thionyl chloride (SOCl₂) (15.5 g, 0.13 mol) was added. The mixture was gently refluxed at ca. +80 °C with stirring for 4 h and the evolution of gaseous HCl and SO₂, following the method previously reported [60]. Benzene and excess of SOCl₂ were removed under reduced pressure, then the residue was washed with dry *n*-hexane (3 × 25 mL) and crystallized from dry CH₂Cl₂ (15 mL/g) to obtain **3**. Yield: (16.4 g, 75%); m.p.: 174–175 °C (m.p.: 174–175 °C [60]); as green yellow prisms.

General procedure for the synthesis of bis(2-carbamoylaryl)phenyldiselenides 4a–p. To a stirred mixture of substituted aniline (1.05 mmol) and sodium carbonate (Na₂CO₃) powdered in a mortar (0.265 g,

2.5 mmol) in dry CH_2Cl_2 (5 mL), the solution of bis[(2-chlorocarbonyl)phenyl] diselenide (**3**) (0.218 g, 0.50 mmol) in dry CH_2Cl_2 was added dropwise for 30 min at RT. The reaction progress was monitored using TLC (CHCl_3 :EtOAc, 4:1, *v/v*). The stirring was continued for 24–168 h. Over a 3-day period, chloride **3** (10.9 mg, 25.0 μmol) was added. When the reaction was completed, the product was filtered off and washed with CH_2Cl_2 (2×5 mL), water (5 mL), and 3.5% aq. HCl (3×5 mL), and air-dried to obtain products **4a–p**.

Bis[2-(2-hydroxyphenylcarbamoyl)]phenyl Diselenide **4a**. The general procedure starting from 2-hydroxyaniline (0.115 g, 1.05 mmol) at RT was employed, with a 96-h reaction time to obtain **4a**. Yield: (0.28 g, 92% calculated on 2-hydroxyaniline used); m.p.: 233.5–236.5 °C (235–238 °C [53]); as a white solid (MeOH, 0.45 g/L). IR (KBr, cm^{-1}) ν 3403 (NH), 3176 (br OH), 1636 (C=O), 1549 (CONH), 1456, 1351, 1286 (C–N), 1235 (C–O), 1194, 1029, 757, 729, 622 (C–Se), 594. $^1\text{H-NMR}$ (300 MHz, DMSO- d_6): δ 6.86 (ddd, $J = 7.8$ Hz, $J = 7.3$ Hz, $J = 1.0$ Hz, 2H, Ar-H), 6.95 (dd, $J = 7.5$ Hz, $J = 1.0$ Hz, 2H, Ar-H), 7.09 (ddd, $J = 7.5$ Hz, $J = 7.3$ Hz, $J = 1.5$ Hz, 2H, Ar-H), 7.40 (ddd, $J = 7.3$ Hz, $J = 6.5$ Hz, $J = 1.5$ Hz, 2H, Ph-H), 7.46 (ddd, $J = 7.8$ Hz, $J = 6.5$ Hz, $J = 1.5$ Hz, 2H, Ph-H), 7.63 (dd, $J = 7.8$ Hz, $J = 1.5$ Hz, 2H, Ar-H), 7.79 (dd, $J = 7.8$ Hz, $J = 1.5$ Hz, 2H, Ph-H), 8.03 (dd, $J = 7.3$ Hz, $J = 1.5$ Hz, 2H, Ph-H), 9.76 (s, 2H, NH), 9.78 (s, 2H, OH). $^{13}\text{C-NMR}$ (75 MHz, DMSO- d_6): δ 115.8 (2 \times CH), 118.9 (2 \times CH), 125.1 (2 \times CH), 126.2 (2 \times CH), 126.3 (2 \times CH), 128.4 (2 \times CH), 130.1 (2 \times CH), 132.0 (2 \times C), 132.1 (2 \times CH), 132.2 (2 \times C), 133.0 (2 \times C), 150.0 (2 \times C), 166.1 (2 \times C=O). HRMS (TOF MS ESI): m/z for $\text{C}_{26}\text{H}_{20}\text{N}_2\text{O}_4\text{Se}_2 + \text{Na}^+$ calculated: 604.9654; found: 604.9662.

Bis[2-(3-hydroxyphenylcarbamoyl)]phenyl Diselenide **4b**. The general procedure starting from 3-hydroxyaniline (109 mg, 1.00 mmol) in anhydrous MeCN (5 mL) and benzoyl chloride **3** in anhydrous MeCN (10 mL) at RT was employed, with a 60-h reaction time followed by the dropwise addition of H_2O (10 mL) during stirring. Solid precipitate was filtered off, washed with aq. HCl and EtOH: H_2O (19:1, *v/v*), and air-dried to obtain **4b**. Yield: (244 mg, 84%); m.p.: 285.5–287.0 °C (m.p.: 285–286 °C [53]); as a pale yellow powder. $^1\text{H-NMR}$ (400 MHz, DMSO- d_6): δ 6.51–6.59 (m, 2H, Ar-H), 7.11–7.19 (m, 4H, Ar-H), 7.37–7.39 (m, 2H, Ar-H), 7.40 (ddd, $J = 7.5$ Hz, $J = 7.4$ Hz, $J = 1.0$ Hz, 2H, Ph-H), 7.46 (ddd, $J = 7.9$ Hz, $J = 7.4$ Hz, $J = 1.4$ Hz, 2H, Ph-H), 7.79 (dd, $J = 7.9$ Hz, $J = 1.0$ Hz, 2H, Ph-H), 7.93 (dd, $J = 7.5$ Hz, $J = 1.4$ Hz, 2H, Ph-H), 9.50 (s, 2H, OH), 10.45 (s, 2H, NH). $^{13}\text{C-NMR}$ (101 MHz, DMSO- d_6): δ 107.56 (2 \times CH), 111.23 (4 \times CH), 126.38 (2 \times CH), 128.60 (2 \times CH), 129.35 (2 \times CH), 130.15 (2 \times CH), 131.93 (2 \times CH), 131.95 (2 \times C), 133.93 (2 \times C), 139.65 (2 \times C), 157.55 (2 \times C), 166.23 (2 \times C=O). HRMS (TOF MS ESI): m/z for $\text{C}_{26}\text{H}_{20}\text{N}_2\text{O}_4\text{Se}_2 + \text{Na}^+$ calculated: 604.9654; found: 604.9660. Anal. Calcd. for $\text{C}_{26}\text{H}_{20}\text{N}_2\text{O}_4\text{Se}_2$ (582.37): C, 53.62; H, 3.46; N, 4.81. Found: C, 53.87; H, 3.25; N, 4.89.

Bis[2-(4-hydroxyphenylcarbamoyl)]phenyl Diselenide **4c**. The general procedure starting from 4-hydroxyaniline (0.115 g, 1.05 mmol) at RT was employed, with a 120-h reaction time to obtain **4c** as a beige powder. Yield: (222 mg, 73% calculated on the 4-hydroxyaniline used). An analytically pure sample was obtained by column chromatography on silica gel using a mixture of CHCl_3 and EtOAc (4:1, *v/v*) as an eluent to obtain pure **4c**; m.p.: 271–272 °C (282–284 °C [53]); as a pale solid (CH_2Cl_2). IR (KBr, cm^{-1}) ν 3260 (br NH), 3071 (br OH), 1640 (C=O), 1619, 1589, 1510 (CONH), 1444, 1352, 1280 (C–N), 1235 (C–O), 828, 732, 670 (C–Se), 517. $^1\text{H-NMR}$ (300 MHz, DMSO- d_6): δ 6.83 (d, $J = 8.7$ Hz, 4H, Ar-H), 7.35 (d, $J = 8.7$ Hz, 4H, Ar-H), 7.46 (dd, $J = 7.5$ Hz, $J = 7.2$ Hz, 2H, Ph-H), 7.66 (dd, $J = 8.0$ Hz, $J = 7.2$ Hz, 2H, Ph-H), 7.87 (d, $J = 7.5$ Hz, 2H, Ph-H), 8.06 (d, $J = 8.0$ Hz, 2H, Ph-H), 9.60 (s, 4H, 2 \times NH and 2 \times OH). $^{13}\text{C-NMR}$ (75 MHz, DMSO- d_6): δ 115.49 (4 \times CH), 125.67 (2 \times CH), 126.04 (2 \times CH), 126.70 (4 \times CH), 127.74 (2 \times CH), 128.21 (2 \times C), 130.53 (2 \times C), 131.84 (2 \times CH), 138.91 (2 \times C), 155.66 (2 \times C), 164.82 (2 \times C=O). HRMS (TOF MS ESI): m/z for $\text{C}_{26}\text{H}_{20}\text{N}_2\text{O}_4\text{Se}_2 + \text{Na}^+$ calculated: 606.9646; found: 606.9676.

Bis[2-(2-hydroxy-4-methylphenylcarbamoyl)]phenyl Diselenide **4d**. The general procedure starting from 2-hydroxy-4-methylaniline (129 mg, 1.05 mmol) at RT was employed, with a 24-h reaction time to obtain **4d**. Yield: (183 mg, 60% calculated on benzoyl chloride **3** used); m.p.: 221.0–223.5 °C; as a beige

powder (CH₂Cl₂). IR (KBr, cm⁻¹) ν 3394 (NH), 3250 (br OH), 1640 (C=O), 1603, 1541 (CONH), 1288 (C–N), 1228 (C–O), 809, 736, 647 (C–Se), 589. ¹H-NMR (300 MHz, DMSO-*d*₆): δ 2.25 (s, 6H, CH₃), 6.66 (dd, *J* = 8.2 Hz, *J* = 1.2 Hz, 2H, Ar-H), 6.76 (d, *J* = 1.2 Hz, 2H, Ar-H), 7.39 (ddd, *J* = 7.2 Hz, *J* = 7.0 Hz, *J* = 1.0 Hz, 2H, Ph-H), 7.43 (d, *J* = 8.2 Hz, 2H, Ar-H), 7.44 (ddd, *J* = 7.8 Hz, *J* = 7.0 Hz, *J* = 1.6 Hz, 2H, Ph-H), 7.77 (dd, *J* = 7.8 Hz, *J* = 1.0 Hz, 2H, Ph-H), 8.00 (dd, *J* = 7.2 Hz, *J* = 1.6 Hz, 2H, Ph-H), 9.60 (s, 2H, OH), 9.72 (s, 2H, NH). ¹³C-NMR (75 MHz, DMSO-*d*₆): δ 20.7 (2 × CH₃), 116.4 (2 × CH), 119.5 (2 × CH), 122.5 (2 × C), 125.0 (2 × CH), 126.3 (2 × CH), 128.4 (2 × CH), 130.0 (2 × CH), 131.9 (2 × CH), 132.2 (2 × C), 133.0 (2 × C), 135.7 (2 × C), 149.9 (2 × C), 166.1 (2 × C=O). HRMS (TOF MS ESI): *m/z* for C₂₈H₂₄N₂O₄Se₂ + Na⁺ calculated: 634.9959; found: 634.9949. Anal. Calcd. for C₂₈H₂₄N₂O₄Se₂ (610.42): C, 55.09; H, 3.96; N, 4.59. Found: C, 55.18; H, 3.82; N, 4.52.

Bis[2-(2-hydroxy-5-methylphenylcarbamoyl)]phenyl Diselenide **4e**. The general procedure starting from 2-hydroxy-5-methylaniline (129 mg, 1.05 mmol) and benzyl chloride **3** (240 mg, 0.55 mmol) at RT was employed, with a 96-h reaction time to obtain **4e**. Yield: (285 mg, 89% calculated on 2-hydroxy-5-methylaniline); m.p.: 191.5–194.5 °C; as a beige powder (CH₂Cl₂). IR (KBr, cm⁻¹) ν 3409 (NH), 3235 (br OH), 1640 (C=O), 1599, 1547 (CONH), 1504, 1279 (C–N), 1236 (C–O), 1200, 1027, 809, 740, 619 (C–Se), 452. ¹H-NMR (300 MHz, DMSO-*d*₆): δ 2.25 (s, 6H, CH₃), 6.83 (d, *J* = 8.2 Hz, 2H, Ar-H), 6.89 (dd, *J* = 8.2 Hz, *J* = 1.7 Hz, 2H, Ar-H), 7.39 (ddd, *J* = 7.5 Hz, *J* = 7.1 Hz, *J* = 1.0 Hz, 2H, Ph-H), 7.46 (d, *J* = 1.7 Hz, 2H, Ar-H), 7.47 (ddd, *J* = 7.8 Hz, *J* = 7.1 Hz, *J* = 1.4 Hz, 2H, Ph-H), 7.78 (dd, *J* = 7.8 Hz, *J* = 1.0 Hz, 2H, Ph-H), 8.00 (dd, *J* = 7.5 Hz, *J* = 1.4 Hz, 2H, Ph-H), 9.49 (s, 2H, NH), 9.71 (s, 2H, OH). ¹³C-NMR (75 MHz, DMSO-*d*₆): δ 20.2 (2 × CH₃), 115.7 (2 × CH), 124.8 (2 × C), 125.2 (2 × CH), 126.4 (2 × CH), 126.5 (2 × CH), 127.5 (2 × C), 128.4 (2 × CH), 130.1 (2 × CH), 132.0 (2 × CH), 132.2 (2 × C), 133.0 (2 × C), 147.5 (2 × C), 166.1 (2 × C=O). HRMS (TOF MS ESI): *m/z* for C₂₈H₂₄N₂O₄Se₂ + Na⁺ calculated: 632.9967; found: 632.9987. Anal. Calcd. for C₂₈H₂₄N₂O₄Se₂ (610.42): C, 55.09; H, 3.96; N, 4.59. Found: C, 55.01; H, 3.88; N, 4.64.

Bis[2-(2-hydroxy-6-methylphenylcarbamoyl)]phenyl Diselenide **4f**. The general procedure starting from 2-hydroxy-6-methylaniline (129 mg, 1.05 mmol) at RT was employed, with a 168-h reaction time to obtain **4f**. Yield: (282 mg, 88% calculated on 2-hydroxy-6-methylaniline used); m.p.: 264–267 °C; as a pale beige powder (CH₂Cl₂). IR (KBr, cm⁻¹) ν 3360 (NH), 3251 (OH), 1622 (C=O), 1592, 1515 (CONH), 1472, 1294 (C–N), 1191 (C–O), 1026, 781, 737, 549 (C–Se). ¹H-NMR (300 MHz, DMSO-*d*₆): δ 2.19 (s, 6H, CH₃), 6.74 (d, *J* = 7.7 Hz, 2H, Ar-H), 6.78 (d, *J* = 8.2 Hz, 2H, Ar-H), 7.05 (dd, *J* = 8.2 Hz, *J* = 7.7 Hz, 2H, Ar-H), 7.39 (dd, *J* = 7.6 Hz, *J* = 6.0 Hz, 2H, Ph-H), 7.45 (dd, *J* = 6.8 Hz, *J* = 6.0 Hz, 2H, Ph-H), 7.77 (d, *J* = 7.6 Hz, 2H, Ph-H), 8.09 (d, *J* = 6.8 Hz, 2H, Ph-H), 9.39 (s, 2H, NH), 9.78 (s, 2H, OH). ¹³C-NMR (75 MHz, DMSO-*d*₆): δ 17.9 (2 × CH₃), 113.5 (2 × CH), 120.4 (2 × CH), 123.4 (2 × C), 126.1 (2 × CH), 127.3 (2 × CH), 128.5 (2 × CH), 129.9 (2 × CH), 131.7 (2 × CH), 132.4 (2 × C), 133.0 (2 × C), 136.8 (2 × C), 153.3 (2 × C), 166.5 (2 × C=O). HRMS (TOF MS ESI): *m/z* for C₂₈H₂₄N₂O₄Se₂ + Na⁺ calculated: 632.9967; found: 632.9971. Anal. Calcd. for C₂₈H₂₄N₂O₄Se₂ (610.42): C, 55.09; H, 3.96; N, 4.59. Found: C, 55.16; H, 4.09; N, 4.45.

Bis[2-(2-hydroxy-4-chlorophenylcarbamoyl)]phenyl Diselenide **4g**. The general procedure starting from 2-hydroxy-4-chloroaniline (151 mg, 1.05 mmol) at RT was employed, with a 96-h reaction time to obtain **4g**. Yield: (202 mg, 59% calculated on 2-hydroxy-4-chloroaniline used); m.p.: 230–234 °C; as a brownish gray powder (CH₂Cl₂). IR (KBr, cm⁻¹) ν 3398 (NH), 3176 (br OH), 1639 (C=O), 1532 (CONH), 1414, 1342, 1265 (C–N), 1226 (C–O), 911, 734, 648, 590 (C–Se), 452. ¹H-NMR (300 MHz, DMSO-*d*₆): δ 6.92 (dd, *J* = 8.5 Hz, *J* = 2.2 Hz, 2H, Ar-H), 6.98 (d, *J* = 2.2 Hz, 2H, Ar-H), 7.40 (dd, *J* = 7.2 Hz, *J* = 7.0 Hz, 2H, Ph-H), 7.46 (ddd, *J* = 7.9 Hz, *J* = 7.0 Hz, *J* = 1.2 Hz, 2H, Ph-H), 7.63 (d, *J* = 8.5 Hz, 2H, Ar-H), 7.76 (d, *J* = 7.9 Hz, 2H, Ph-H), 8.02 (dd, *J* = 7.2 Hz, *J* = 1.2 Hz, 2H, Ph-H), 9.86 (s, 2H, NH), 10.37 (s, 2H, OH). ¹³C-NMR (75 MHz, DMSO-*d*₆): δ 115.5 (2 × CH), 118.7 (2 × CH), 124.3 (2 × C), 126.4 (2 × CH), 126.6 (2 × CH), 128.6 (2 × CH), 129.6 (2 × C), 130.0 (2 × CH), 132.1 (2 × CH), 132.3 (2 × C), 132.7 (2 × C), 151.3 (2 × C), 166.2 (2 × C=O). HRMS (TOF MS ESI): *m/z* for C₂₆H₁₈Cl₂N₂O₄Se₂ + Na⁺ calculated:

672.8874; 674.8866; found: 672.8876; 674.8884. Anal. Calcd. for $C_{26}H_{18}Cl_2N_2O_4Se_2$ (651.26): C, 47.95; H, 2.79; N, 4.30. Found: C, 48.08; H, 2.54; N, 4.50.

Bis[2-(2-hydroxy-5-chlorophenylcarbamoyl)]phenyl Diselenide 4h. The general procedure starting from 2-hydroxy-5-chloroaniline (151 mg, 1.05 mmol) at RT was employed, with a 48-h reaction time to obtain **4h**. Yield: (206 mg, 63% calculated on benzoyl chloride **3**); m.p.: 219.5–221.5 °C; as a gray-brown powder (CH_2Cl_2). IR (KBr, cm^{-1}) ν 3400 (NH), 3218 (br OH), 1638 (C=O), 1596, 1543 (CONH), 1428, 1270, 1239 (C–N), 1195 (C–O), 1027, 731 (C–Cl), 655 (C–Se), 614. 1H -NMR (400 MHz, $DMSO-d_6$): δ 6.96 (d, $J = 8.7$ Hz, 2H, Ar-H), 7.13 (dd, $J = 8.7$ Hz, $J = 2.6$ Hz, 2H, Ar-H), 7.41 (dd, $J = 7.6$ Hz, $J = 7.3$ Hz, 2H, Ph-H), 7.47 (dd, $J = 7.9$ Hz, $J = 7.3$ Hz, 2H, Ph-H), 7.77 (d, $J = 2.6$ Hz, 2H, Ar-H), 7.78 (d, $J = 7.6$ Hz, 2H, Ph-H), 8.01 (d, $J = 7.9$ Hz, 2H, Ph-H), 9.82 (s, 2H, OH), 10.18 (s, 2H, NH). ^{13}C -NMR (101 MHz, $DMSO-d_6$): δ 116.89 (2 \times CH), 121.93 (2 \times C), 124.06 (2 \times CH), 125.56 (2 \times CH), 126.38 (2 \times C), 126.41 (2 \times CH), 128.61 (2 \times CH), 130.10 (2 \times CH), 132.18 (2 \times CH), 132.30 (2 \times C), 132.68 (2 \times C), 148.76 (2 \times C), 166.19 (2 \times C=O). HRMS (TOF MS ESI): m/z for $C_{26}H_{18}Cl_2N_2O_4Se_2 + Na^+$ calculated: 674.8866; found: 674.8862. Anal. Calcd. for $C_{26}H_{18}Cl_2N_2O_4Se_2$ (651.26): C, 47.95; H, 2.79; N, 4.30. Found: C, 48.03; H, 2.56; N, 4.25.

Bis[2-(2-hydroxy-5-carboxymethylphenylcarbamoyl)]phenyl Diselenide 4i. The general procedure starting from 2-hydroxy-5-carboxymethylaniline (175 mg, 1.05 mmol) at RT was employed, with a 72-h reaction time to obtain **4i**. Yield: (209 mg, 57% calculated on 2-hydroxy-5-carboxymethylaniline used); m.p.: 281–282 °C; as a cream powder (CH_2Cl_2). IR (KBr, cm^{-1}) ν 3323 (br NH and OH), 1709, 1691 (OC=O), 1639 (NC=O), 1604, 1537 (CONH), 1430, 1298 (C–N), 1236 (C–O), 1120 (C–O), 737, 606 (C–Se). 1H -NMR (400 MHz, $DMSO-d_6$): δ 3.83 (s, 6H, OCH_3), 7.05 (d, $J = 8.5$ Hz, 2H, Ar-H), 7.41 (ddd, $J = 7.6$ Hz, $J = 7.3$ Hz, $J = 1.2$ Hz, 2H, Ph-H), 7.48 (ddd, $J = 7.9$ Hz, $J = 7.3$ Hz, $J = 1.2$ Hz, 2H, Ph-H), 7.74 (dd, $J = 8.5$ Hz, $J = 2.2$ Hz, 2H, Ar-H), 7.79 (dd, $J = 7.9$ Hz, $J = 1.2$ Hz, 2H, Ph-H), 8.03 (dd, $J = 7.6$ Hz, $J = 1.2$ Hz, 2H, Ph-H), 8.31 (d, $J = 2.2$ Hz, 2H, Ar-H), 9.89 (s, 2H, NH), 10.86 (s, 2H, OH). ^{13}C -NMR (101 MHz, $DMSO-d_6$): δ 51.76 (2 \times OCH_3), 115.58 (2 \times CH), 120.14 (2 \times C), 125.09 (2 \times C), 126.27 (2 \times CH), 126.42 (2 \times CH), 127.95 (2 \times CH), 128.61 (2 \times CH), 130.10 (2 \times CH), 132.16 (2 \times CH), 132.30 (2 \times C), 132.74 (2 \times C), 154.64 (2 \times C), 165.83 (2 \times C=O), 166.25 (2 \times C=O). HRMS (TOF MS ESI): m/z for $C_{30}H_{24}N_2O_8Se_2 + Na^+$ calculated: 720.9763; found: 720.9755. Anal. Calcd. for $C_{30}H_{24}N_2O_8Se_2$ (698.44): C, 51.59; H, 3.46; N, 4.01. Found: C, 51.43; H, 3.60; N, 3.95.

Bis[2-(3-hydroxy-4-methoxyphenylcarbamoyl)]phenyl Diselenide 4j. The general procedure starting from 3-hydroxy-4-methoxyaniline (146 mg, 1.05 mmol) at RT was employed, with a 120-h reaction time to obtain **4j**. Yield: (317 mg, 94% calculated on 3-hydroxy-4-methoxyaniline used); m.p.: 232.5–236.5 °C; as a white powder (CH_2Cl_2). IR (KBr, cm^{-1}) ν 3500 (NH), 3270 (OH), 1670, 1636 (C=O), 1513 (CONH), 1270 (C–N), 1239 (C–O), 1175 (C–O), 1024 (C–O), 738, 565 (C–Se). 1H -NMR (300 MHz, $DMSO-d_6$): δ 3.76 (s, 6H, $2OCH_3$), 6.91 (d, $J = 8.8$ Hz, 2H, Ar-H), 7.13 (dd, $J = 8.8$ Hz, $J = 2.5$ Hz, 2H, Ar-H), 7.37 (d, $J = 2.5$ Hz, 2H, Ar-H), 7.36–7.46 (m, 4H, Ph-H), 7.78 (dd, $J = 7.7$ Hz, $J = 1.2$ Hz, 2H, Ph-H), 7.93 (dd, $J = 7.3$ Hz, $J = 1.4$ Hz, 2H, Ph-H), 7.06 (br s, 2H, OH), 10.32 (s, 2H, NH). ^{13}C -NMR (75 MHz, $DMSO-d_6$): δ 55.8 (2 \times OCH_3), 108.9 (2 \times CH), 111.4 (2 \times CH), 112.3 (2 \times CH), 126.3 (2 \times CH), 128.4 (2 \times CH), 130.0 (2 \times CH), 131.7 (2 \times CH), 131.9 (2 \times C), 132.1 (2 \times C), 133.9 (2 \times C), 144.4 (2 \times C), 146.3 (2 \times C), 165.8 (2 \times C=O). HRMS (TOF MS ESI): m/z for $C_{28}H_{24}N_2O_6Se_2 + Na^+$ calculated: 664.9865; 666.9857; found: 664.9854; 666.9868. Anal. Calcd. for $C_{28}H_{24}N_2O_6Se_2$ (642.42): C, 52.35; H, 3.77; N, 4.36. Found: C, 52.30; H, 3.52; N, 4.45.

Bis[2-(2-methoxyphenylcarbamoyl)]phenyl Diselenide 4k. The general procedure starting from 2-methoxyaniline (129 mg, 1.05 mmol) at RT was employed, with a 168-h reaction time to obtain **4k**. Yield: (0.301 g, 94% calculated on 2-methoxyaniline used); m.p.: 214–215 °C (m.p.: 216–217 °C [53]); as yellow crystals (MeOH, 2 L/g). IR (KBr, cm^{-1}) ν 3411 (NH), 1646 (C=O), 1602, 1528 (CONH), 1462, 1434, 1340, 1289 (C–N), 1251 and 1026 (C–O), 745, 629 (C–Se), 591, 561, 457. 1H -NMR (300 MHz, $DMSO-d_6$): δ 3.84 (s, 6H, OCH_3), 7.00 (ddd, $J = 7.7$ Hz, $J = 7.4$ Hz, $J = 1.0$ Hz, 2H, Ar-H), 7.13 (dd,

$J = 8.3$ Hz, $J = 1.0$ Hz, 2H, Ar-H), 7.24 (ddd, $J = 8.3$ Hz, $J = 7.4$ Hz, $J = 1.6$ Hz, 2H, Ar-H), 7.40 (dd, $J = 7.4$ Hz, $J = 7.4$ Hz, 2H, Ph-H), 7.47 (dd, $J = 7.9$ Hz, $J = 7.4$ Hz, 2H, Ph-H), 7.73 (dd, $J = 7.7$ Hz, $J = 1.6$ Hz, 2H Ar-H), 7.77 (d, $J = 7.9$ Hz, 2H, Ph-H), 8.01 (d, $J = 7.4$ Hz, 2H, Ph-H), 9.83 (s, 2H, NH). $^{13}\text{C-NMR}$ (75 MHz, DMSO- d_6): δ 55.56 (2 \times OCH $_3$), 111.56 (2 \times C), 120.17 (2 \times C), 125.15 (2 \times C), 126.16 (2 \times C), 126.39 (2 \times C), 126.42 (2 \times C), 128.49 (2 \times C), 130.06 (2 \times C), 132.04 (2 \times C), 132.30 (2 \times C), 132.96 (2 \times C), 152.04 (2 \times C), 166.12 (2 \times C=O). The HRMS (TOF MS ESI) is in agreement with literature data [26].

Bis[2-(3-methoxyphenylcarbamoyl)]phenyl Diselenide 4l. Preparation according to Reference [35]. The general procedure starting from 3-methoxyaniline (129 mg, 1.05 mmol) at RT was employed, with a 72-h reaction time to obtain **4l**. Yield: (0.300 g, 98% calculated on benzoyl chloride **3** used); m.p.: 221–223 °C (221–223 °C [35]); as a colorless powder (CHCl $_3$:EtOAc, 1:2, v/v). Yield of the analytically pure sample: (226 mg, 74%). The $^1\text{H-NMR}$, $^{13}\text{C-NMR}$, and HRMS (TOF MS ESI) spectra are in agreement with literature data [35].

Bis[2-(4-methoxyphenylcarbamoyl)]phenyl Diselenide 4m. Preparation according to a modified literature procedure [35]. The general procedure starting from 4-methoxyaniline (129 mg, 1.05 mmol) and bis[(2-chlorocarbonyl)phenyl] diselenide **3** (0.229 g, 0.525 mmol) at RT was employed, with a 48-h reaction time to obtain **4m**. Yield: (0.300 g, 94%); m.p.: 285.5–287.5 °C (290–292 °C [53]); as a colorless powder (CHCl $_3$:EtOAc, 1:2, v/v). Yield of the analytically pure sample: (266 mg, 83%). The $^1\text{H-NMR}$, $^{13}\text{C-NMR}$, and HRMS (TOF MS ESI) spectra are in agreement with literature data [35].

Bis[2-(2,4-dimethoxyphenylcarbamoyl)]phenyl Diselenide 4n. The general procedure starting from 2,4-dimethoxyaniline (161 mg, 1.05 mmol) at RT was employed, with a 168-h reaction time, with separation by acidification with 3.5% HCl to a pH of 2.5, extraction with CH $_2$ Cl $_2$, dry with anhydrous Na $_2$ SO $_4$, filtration, and solvent evaporation to obtain **4n**. Yield: (348 mg, 99% calculated on 2,4-dimethoxyaniline used); m.p.: 211.0–213.5 °C; as a white powder (CH $_2$ Cl $_2$). IR (KBr, cm $^{-1}$) ν 3436 (NH), 1640 (C=O), 1531 (CONH), 1414, 1283 (C–N), 1205, and 1036 (C–O), 831, 732, 578 (C–Se), 453. $^1\text{H-NMR}$ (300 MHz, DMSO- d_6): δ 3.79 (s, 6H, 2OCH $_3$), 3.82 (s, 6H, 2OCH $_3$), 6.57 (dd, $J = 8.8$ Hz, $J = 2.6$ Hz, 2H, Ar-H), 6.69 (d, $J = 2.6$ Hz, 2H, Ar-H), 7.35–7.48 (m, 4H, Ph-H), 7.47 (d, $J = 8.8$ Hz, 2H, Ar-H), 7.75 (d, $J = 7.3$ Hz, 2H, Ph-H), 8.01 (d, $J = 7.2$ Hz, 2H, Ph-H), 9.76 (s, 2H, NH). $^{13}\text{C-NMR}$ (75 MHz, DMSO- d_6): δ 55.33 (2 \times OCH $_3$), 55.69 (2 \times OCH $_3$), 98.94 (2 \times C), 104.25 (2 \times C), 119.06 (2 \times C), 126.28 (2 \times C), 126.79 (2 \times C), 128.39 (2 \times C), 129.98 (2 \times C), 131.88 (2 \times C), 132.30 (2 \times C), 132.93 (2 \times C), 153.74 (2 \times C), 158.28 (2 \times C), 166.20 (2 \times C=O). HRMS (TOF MS ESI): m/z for C $_{30}$ H $_{28}$ N $_2$ O $_6$ Se $_2$ + Na $^+$ calculated: 693.0178; found: 693.0176. Anal. Calcd. for C $_{30}$ H $_{28}$ N $_2$ O $_6$ Se $_2$ (670.47): C, 53.74; H, 4.21; N, 4.18. Found: C, 53.62; H, 4.13; N, 4.27.

Bis[2-(3,4-dimethoxyphenylcarbamoyl)]phenyl Diselenide 4o. The general procedure starting from 3,4-dimethoxyaniline (161 mg, 1.05 mmol) at RT was employed, with a 96-h reaction time to obtain **4o**. Yield: (253 mg, 72%); m.p.: 231.5–234.5 °C (m.p.: 235 °C [53]); as a pale yellow powder (CH $_2$ Cl $_2$). IR (KBr, cm $^{-1}$) ν 3291 (NH), 1641 (C=O), 1513 (CONH), 1452, 1263 (C–N), 1237 (C–O), 1134, 1026 (C–O), 741, 681, 574 (C–Se), 454. $^1\text{H-NMR}$ (300 MHz, DMSO- d_6): δ 3.76 (s, 6H, 2OCH $_3$), 3.77 (s, 6H, 2OCH $_3$), 6.96 (d, $J = 8.8$ Hz, 2H, Ar-H), 7.30–7.45 (m, 6H, Ph-H and Ar-H), 7.47 (s, 2H, Ar-H), 7.83 (s, br., 2H, Ph-H), 7.95 (d, $J = 7.7$ Hz, 2H, Ph-H), 10.51 (s, 2H, NH). $^{13}\text{C-NMR}$ (75 MHz, DMSO- d_6): δ 55.41 (2 \times OCH $_3$), 55.67 (2 \times OCH $_3$), 105.81 (2 \times CH), 111.84, (2 \times CH), 112.72 (2 \times CH), 126.08 (2 \times CH), 128.36 (2 \times CH), 130.10 (2 \times CH), 131.59 (2 \times CH), 132.19 (2 \times C), 132.30 (2 \times C), 133.63 (2 \times C), 145.46 (2 \times C), 148.42 (2 \times C), 165.78 (2 \times C=O). HRMS (TOF MS ESI): m/z for C $_{30}$ H $_{28}$ N $_2$ O $_6$ Se $_2$ + Na $^+$ calculated: 693.0178. Found: 693.0178.

Bis[2-(3,4,5-trimethoxyphenylcarbamoyl)]phenyl Diselenide 4p. The general procedure starting from 3,4,5-trimethoxyaniline (192 mg, 1.05 mmol) at RT was employed, with a 96-h reaction time to obtain **4p**. Yield: (0.291 g, 76%); m.p.: 193–195 °C; as a white powder (CH $_3$ OH, 100 g/L). IR (KBr, cm $^{-1}$) ν 3322 (NH), 1605 (C=O), 1507 (CONH), 1450, 1411, 1313, 1235 (C–N), 1128 (C–O), 1027 (C–O), 1004

(C–O), 738, 623 (C–Se). ¹H-NMR (300 MHz, DMSO-*d*₆): δ 3.65 (s, 6H, 2OCH₃), 3.77 (s, 12H, 4OCH₃), 7.20 (s, 4H, Ar-H), 7.38 (dd, *J* = 7.3 Hz, 2H, Ph-H), 7.44 (dd, *J* = 7.3 Hz, 2H, Ph-H), 7.80–8.00 (m, 4H, Ph-H), 10.56 (s, 2H, NH). ¹H-NMR (600 MHz, DMSO-*d*₆): δ 3.67 (s, 6H, 2OCH₃), 3.80 (s, 12H, 4OCH₃), 7.21 (s, 4H, Ar-H), 7.42 (dd, *J* = 7.3 Hz, *J* = 6.6 Hz, 2H, Ph-H), 7.46 (dd, *J* = 7.9 Hz, *J* = 6.6 Hz, 2H, Ph-H), 7.81 (d, *J* = 7.9 Hz, 2H, Ph-H), 7.95 (d, *J* = 7.3 Hz, 2H, Ph-H), 10.47 (s, 2H, NH). ¹³C-NMR (75 MHz, DMSO-*d*₆): δ 56.33 (4 × OCH₃), 60.64 (2 × OCH₃), 99.21 (4 × CH), 126.6 (2 × CH), 128.9 (2 × CH), 130.5 (2 × CH), 132.1 (2 × CH, 2 × C), 134.4 (2 × C), 134.6 (2 × C), 135.8 (2 × C), 153.2 (4 × C), 166.5 (2 × C=O). ¹³C-NMR (151 MHz, DMSO-*d*₆): δ 55.75 (4 × OCH₃), 60.10 (2 × OCH₃), 98.24 (4 × CH), 126.31 (2 × CH), 128.53 (2 × CH), 130.12 (2 × CH), 131.97 (2 × CH, 2 × C), 133.71 (2 × C), 134.05 (2 × C), 134.75 (2 × C), 152.64 (4 × C), 166.10 (2 × C=O). HRMS (TOF MS ESI): *m/z* for C₃₂H₃₂N₂O₈Se₂ + Na⁺ calculated: 755.0381; found: 755.0381. Anal. Calcd. for C₃₂H₃₂N₂O₈Se₂ (730.53): C, 52.61; H, 4.42; N, 3.83. Found: C, 52.46; H, 4.56; N, 3.72.

Bis[2-(*N*-morpholinecarbamoyl)]phenyl Diselenide **5**. To a stirred solution of morpholine (436 mg, 0.44 mL, 5.0 mmol) in dry CH₂Cl₂ (10 mL), the solution of bis[(2-chlorocarbonyl)phenyl]diselenide **3** (437 mg, 1.0 mmol) in dry CH₂Cl₂ was added and the reaction was continued at RT (+20 °C) for 120 h until no spot of chloride **3** was observed during TLC control (silica gel, eluent CHCl₃:EtOAc, 4:1, *v/v*). When the reaction was completed, the mixture was poured into water (50 mL), acidified with 3.5% HCl to a pH of 1.0, extracted with CH₂Cl₂ (3 × 25 mL), and dried with Na₂SO₄. After evaporation of the solvent, the residue was washed with n-hexane and DEE (25 mL) was added to the oily residue, followed by stirring until the product solidified and was filtered off to obtain **5**. Yield: (458 mg, 85% calculated on diselenide **3** used); m.p.: 149–152 °C; as a pale yellow powder (DEE). IR (KBr, cm⁻¹) ν 3059, 2972, 2959, 2902, 2854 (CH), 1625 (C=O), 1608, 1430, 1279, 1248, 1111 (C–N), 1021 (C–O), 1014 (C–O), 757, 596, 559 (C–Se), 461. ¹H-NMR (300 MHz, DMSO-*d*₆): δ 3.57 (s, 16H, 8CH₂), 7.28–7.37 (m, 4H, Ph-H), 7.39 (ddd, *J* = 7.7 Hz, *J* = 6.4 Hz, *J* = 2.7 Hz, 2H, Ph-H), 7.74 (dd, *J* = 8.2 Hz, *J* = 1.2 Hz, 2H, Ph-H). ¹³C-NMR (75 MHz, DMSO-*d*₆): δ 66.04 (8 × CH₂), 126.84 (2 × CH), 127.36 (2 × CH), 129.05 (2 × C), 130.38 (2 × CH), 131.71 (2 × CH), 136.22 (2 × C), 167.48 (2 × C=O). HRMS (TOF MS ESI): *m/z* for C₂₂H₂₄N₂O₄Se₂ + Na⁺ calculated: 558.9986; 560.9967; 562.9959; found: 559.0029; 560.9987; 562.9982. Anal. Calcd. for C₂₂H₂₄N₂O₄Se₂ (538.36): C, 49.08; H, 4.49; N, 5.20. Found: C, 48.96; H, 4.62; N, 5.29.

Bis[2-(*N*-benzamidocarbamoyl)]phenyl Diselenide **6**. The general procedure starting from benzhydrazide (143 mg, 1.05 mmol) at RT was employed, with a 120-h reaction time to obtain **6**. Yield: (335 mg, 100%); m.p.: 296.5 °C (with decomp.); as a pale yellow wool (AcOH, 1.5 g/L). ¹H-NMR (601 MHz, DMSO-*d*₆): δ 7.42 (ddd, *J* = 7.7 Hz, *J* = 7.3 Hz, *J* = 0.8 Hz, 2H, Ph-H), 7.50 (ddd, *J* = 8.0 Hz, *J* = 7.3 Hz, *J* = 1.1 Hz, 2H, Ph-H), 7.54 (dd, *J* = 7.8 Hz, *J* = 7.4 Hz, 4H, Ar-H), 7.62 (t, *J* = 7.3 Hz, 2H, Ar-H), 7.78 (dd, *J* = 8.0 Hz, *J* = 0.8 Hz, 2H Ph-H), 7.97 (d, *J* = 7.8 Hz, 4H, Ar-H), 8.00 (dd, *J* = 7.7 Hz, *J* = 1.1 Hz, 2H, Ph-H), 10.68 (s, 2H, NH), 10.83 (s, 2H, NH). ¹³C-NMR (151 MHz, DMSO-*d*₆): δ 126.40 (2 × CH), 127.45 (4 × CH), 128.30 (2 × CH), 128.51 (4 × CH), 130.12 (2 × CH), 130.70 (2 × C), 131.94 (2 × CH), 132.30 (2 × C), 132.44 (2 × CH), 132.47 (2 × C), 165.77 (2 × C=O), 166.86 (2 × C=O). HRMS (TOF MS ESI): *m/z* for C₂₈H₂₂N₄O₄Se₂ + Na⁺ calculated: 656.9891; 658.9872; found: 656.9968; 658.9862. Anal. Calcd. for C₂₈H₂₂N₄O₄Se₂ (636.42): C, 52.84; H, 3.48; N, 8.80. Found: C, 53.02; H, 3.29; N, 8.79.

2-Hydroxyacetanilide phenolic ester of bis(2-carboxy)phenyl diselenide **7**. The general procedure starting from 2-hydroxyacetanilide (151 mg, 1.0 mmol) dissolved in dry CH₂Cl₂ (10 mL) at RT was employed, with a 96-h reaction time to obtain **7**. Yield: (119 mg, 36% calculated on 2-hydroxyacetanilide used); m.p.: 178–181 °C; as a pale solid (CH₂Cl₂). IR (KBr, cm⁻¹) ν 3402 (NH), 1760 (OC=O), 1659 (NC=O), 1545 (CONH), 1455, 1370, 1285 (C–N), 1216 (C–O), 1116, 767, 622 (C–Se), 457. ¹H-NMR (300 MHz, DMSO-*d*₆): δ 2.09 (s, 6H, CH₃), 6.75 (dd, *J* = 7.9 Hz, *J* = 7.1 Hz, 2H, Ph-H), 6.85 (d, *J* = 7.0 Hz, 2H, Ph-H), 6.93 (dd, *J* = 7.9 Hz, *J* = 6.6 Hz, 2H, Ph-H), 7.36 (dd, *J* = 7.6 Hz, *J* = 7.0 Hz, 2H, Ph-H), 7.49 (dd, *J* = 7.2 Hz, *J* = 7.0 Hz, 2H, Ph-H), 7.66 (d, *J* = 6.6 Hz, 2H, Ar-H), 7.67 (d, *J* = 7.9 Hz, 2H, Ph-H), 8.33 (d, *J* = 7.2 Hz, 2H, Ph-H), 9.72 (s, 2H, NH). HRMS (TOF MS ESI): *m/z* for C₃₀H₂₄N₂O₆Se₂ + Na⁺ calculated: 688.9865;

690.9857; found: 688.9866; 690.9855. Anal. Calcd. for $C_{30}H_{24}N_2O_6Se_2$ (666.44): C, 54.07; H, 3.63; N, 4.20. Found: C, 53.85; H, 3.90; N, 4.06.

(2*R*,3*S*,6*S*,7*R*,8*R*)-3-(3-Formamido-2-hydroxybenzamido)-8-*n*-alkyl-2,6-dimethyl-4,9-dioxo-1,5-dioxonan-7-yl 3-Methylbutanoate (antimycin A) phenolic ester of bis(2-carboxy)phenyl diselenide **8**. The general procedure starting from naturally occurred antimycin A (0.21 g, 0.40 mmol) dissolved in dry CH_2Cl_2 (5 mL), and Na_2CO_3 (0.16 g, 1.5 mmol), diselenide **3** (87 mg, 0.20 mmol) at RT was employed, with a 10-day reaction time to obtain phenolic ester **8** as a co-crystallizable mixture of different alkyl substituents with $R=C_5H_{11}$ predominantly. Yield: (0.277 g, 97% calculated on antimycin A used); m.p.: >300 °C; as a pale solid. In ^{13}C -NMR, a number of signals is observed and only diagnostic ones are reported ^{13}C -NMR (101 MHz, $DMSO-d_6$): δ 13.88, 17.72, 18.92, 19.00, 19.02, 20.28, 21.85, 21.94, 22.85, 24.43, 25.66, 26.81, 27.18, 28.10 and 30.89 ($C_{aliphatic}$); 58.05, 58.49, 66.77, 66.80 and 77.35 (CHO and CNO); 107.05, 107.47, 117.11, 117.21, 117.28, 118.94, 124.16, 130.59, 131.94 and 136.79 (C_{Ar} , C_{Ar-H}); 151.23 (C_{Ar-CON}), 166.95, 167.16, 167.26, 167.43, 170.42, 174.84, 174.98, 175.09 and 175.11 (C=O). HRMS (TOF MS ESI): m/z for $C_{64}H_{74}N_4O_{20}Se_2 + Na^+$ calculated: 1401.3119; found: 1401.3433; for $C_{65}H_{76}N_4O_{20}Se_2 + Na^+$ calculated: 1411.3291; 1413.3283; found: 1411.3070; 1413.3112; for $C_{66}H_{78}N_4O_{20}Se_2 + Na^+$ calculated: 1427.3440; 1429.3432; found: 1427.3264; 1429.3787; for $C_{67}H_{80}N_4O_{20}Se_2 + Na^+$ calculated: 1441.3596; found: 1441.3926; for $C_{68}H_{82}N_4O_{20}Se_2 + Na^+$ calculated: 1455.3753; 1457.3745; found: 1455.4171; 1457.4158; for $C_{69}H_{84}N_4O_{20}Se_2 + Na^+$ calculated: 1469.3909; 1471.3902; found: 1469.4357; 1471.4440; for $C_{70}H_{86}N_4O_{20}Se_2 + Na^+$ calculated: 1481.4085; 1483.4066; 1485.4058; 1486.4092; found: 1481.3927; 1483.4473; 1485.5210; 1486.4746. Anal. Calcd. for $C_{68}H_{82}N_4O_{20}Se_2$ (1433.31): C, 56.98; H, 5.77; N, 3.91. Found: C, 57.01; H, 5.82; N, 3.77.

2-(Chloroseleno)benzoyl chloride **9**. The compound was prepared from 2,2'-diselenobisbenzoic acid **2** (8.0 g, 20 mmol) using excess of thionyl chloride (6.0 g, 50 mmol) and DMF (1.5 g, 20 mmol) in boiling benzene (150 mL), following the method previously reported [60]. Crude oily product was recrystallized from *n*-hexane (ca. 15 mL/g) and dried under vacuum (20 mmHg) to give **9**. Yield: (8.1 g, 80%); m.p.: 65–67 °C (65–66 °C [60]); as yellow needles (*n*-hexane).

General synthetic procedure for benzisoselenazol-3(2*H*)-ones **10a–g**. To a stirred solution of substituted aniline (13.0 mmol) in anhydrous CH_2Cl_2 (5 mL) a solution of (2-chloroseleno)benzoyl chloride **9** (1.02 g, 4.0 mmol) in anhydrous CH_2Cl_2 (25 mL) was added dropwise. The reaction was continued for 2–24 h. The progress of reaction was controlled by TLC (silica gel, eluent $CHCl_3$:EtOAc, 4:1, *v/v*). After the reaction was finished, the solvent was evaporated, HCl (3.5%, w/w, 40 mL) was added, and the mixture was stirred at RT overnight, then filtered off and left to dry in the air. Crude products were purified by column chromatography on silica gel (70–230 mesh) eluted with $CHCl_3$ to obtain pure **10a–g**.

2-(2-Hydroxyphenyl)-1,2-benzisoselenazol-3(2*H*)-one **10a**. The general procedure starting from 2-hydroxyaniline (0.44 g, 4.0 mmol) and anhydrous Et_3N (1.01 g, 1.4 mL, 10 mmol) at RT was employed, with a 20-h reaction time with separation of the product from the crude reaction mixture by column chromatography using EtOAc as an eluent to obtain **10a**. Yield: (0.46 g, 40% calculated on benzoyl chloride **9** used); m.p.: 195.5–197.5 °C (194–196 °C [54]); as orange flakes (MeCN:H₂O, 84:16, *v/v*). 1H -NMR (400 MHz, $CDCl_3$): δ 6.99 (ddd, $J = 8.0$ Hz, $J = 7.2$ Hz, $J = 1.5$ Hz, 1H, Ar-H), 7.13 (dd, $J = 8.6$ Hz, $J = 1.5$ Hz, 1H, Ar-H), 7.24–7.30 (m, 2H, Ar-H), 7.47–7.54 (m, 1H, Ph-H), 7.65–7.71 (m, 2H, Ph-H), 8.13 (d, $J = 8.1$ Hz, 1H, Ar-H), 8.55 (s, 1H, OH) [57]. ^{13}C -NMR (101 MHz, $DMSO-d_6$): δ 116.94 (2 × CH), 119.22 (2 × CH), 125.73 (2 × CH), 125.81 (2 × CH), 125.97 (C), 127.52 (C), 127.76 (CH), 128.76 (CH), 129.32 (CH), 131.82 (CH), 140.50 (C), 153.25 (C), 165.76 (C=O) [57]. The ^{77}Se NMR, and HRMS are in agreement with literature data [26,57].

2-(3-Hydroxypyridin-2-yl)-1,2-benzisoselenazol-3(2*H*)-one **10b**. The general procedure starting from 2-amino-3-hydroxypyridine (0.44 g, 4.0 mmol) and anhydrous Et_3N (0.95 g, 1.3 mL, 9.4 mmol) at anhydrous MeCN (40 mL) using benzoyl chloride **9** in anhydrous MeCN was employed for a 2-h

reaction time at RT. Solvent was evaporated, water (100 mL) was added portionwise, and the solution was stirred for 1 h. The solid was filtered off, washed with H₂O, and air-dried to obtain **10b**. Yield: (0.76 g, 65%); m.p.: 229 °C (decomp.) with phase transfer at 199 °C (m.p.: 199–201 °C [55]); as a greenish yellow powder (DMSO). ¹H-NMR (DMSO-*d*₆): δ 7.26 (dd, *J* = 8.0 Hz, *J* = 4.6 Hz, 1H, Pyr-H), 7.38 (dd, *J* = 8.0 Hz, *J* = 1.4 Hz, 1H, Pyr-H), 7.50 (ddd, *J* = 7.8 Hz, *J* = 7.2 Hz, *J* = 0.9 Hz, 1H, Ar-H), 7.72 (ddd, *J* = 8.0 Hz, *J* = 7.2 Hz, *J* = 1.3 Hz, 1H, Ar-H), 7.95 (dd, *J* = 7.8 Hz, *J* = 0.9 Hz, 1H Ar-H), 7.98 (dd, *J* = 4.6 Hz, *J* = 1.4 Hz, 1H, Pyr-H), 8.09 (dd, *J* = 8.0 Hz, *J* = 1.3 Hz, 1H, Ar-H), 12.18 (s, 1H, OH). The ¹³C-NMR is in agreement with literature data [55].

2-(2-Methoxyphenyl)-1,2-benzisoselenazol-3(2H)-one 10c. The general procedure starting from 2-methoxyaniline (1.60 g, 13 mmol) at RT was employed, with a 24-h reaction time followed by separation by column chromatography to obtain **10a**. Yield: (0.86 g, 71% calculated on the benzoyl chloride **9**); m.p.: 189.0–191.5 °C (m.p.: 172–174 °C [63]); as yellow prisms (CHCl₃). IR (KBr, cm⁻¹) ν 1621 (C=O), 1270 (C–N), 1108 (C–O), 1020 (C–O). ¹H-NMR (300 MHz, CDCl₃) [63]: δ 3.85 (s, 3H, OCH₃), 6.98–7.07 (m, 2H, Ph-H), 7.36 (ddd, *J* = 8.2 Hz, *J* = 7.6 Hz, *J* = 1.7 Hz, 1H, Ph-H), 7.44 (ddd, *J* = 8.0 Hz, *J* = 6.4 Hz, *J* = 1.7 Hz, 1H, Ar-H), 7.48 (dd, *J* = 8.0 Hz, *J* = 1.7 Hz, 1H, Ph-H), 7.58–7.70 (m, 2H, Ar-H), 8.13 (d, *J* = 7.8 Hz, 1H, Ar-H). ¹³C-NMR (75 MHz, CDCl₃): δ 55.80 (OCH₃), 112.22 (CH), 120.74 (CH), 123.86 (CH), 126.01 (CH), 126.44 (C), 126.74 (C), 129.21 (CH), 129.66 (CH), 129.89 (CH), 132.17 (CH), 139.31 (C), 155.39 (C), 166.70 (C=O).

2-(3-Methoxyphenyl)-1,2-benzisoselenazole-3(2H)-one 10d. Prepared according to a modified procedure [35]. The general procedure starting from 3-methoxyaniline (0.493 g, 4.0 mmol) and dry Et₃N (1.02 g, 1.4 mL, 10 mmol) in anhydrous MeCN (40 mL) at RT with a 2-h reaction time was employed. Water was added portionwise and the solution was left in the refrigerator for crystallization. Then, the solid was filtered off, washed with water and MeCN, and dried in the air to obtain **10d**. Yield: (0.79 g, 65%), m.p.: 166–168 °C (165.5–167.5 °C [35]); as yellow prisms in (MeCN:H₂O, 1:1, *v/v*). The ¹H-NMR and ¹³C-NMR are in agreement with literature data [35].

2-(4-Methoxyphenyl)-1,2-benzisoselenazole-3(2H)-one 10e. Prepared according to a modified procedure [35]. The general procedure starting from 3-methoxyaniline (0.49 g, 4.0 mmol) and Et₃N (1.4 mL, 1.02 g, 10 mmol) in anhydrous MeCN (40 mL) at RT with a 20-h reaction time was employed. Water (40 mL) was added portionwise and the solution was left in the refrigerator for crystallization. Then, the solid was filtered off, washed with water, dried in the air, and the crude product was recrystallized from EtOAc (30 mL/g) to obtain **10e**. Yield: (0.85 g, 70%); m.p.: 180–181 °C (180.5–181.5 °C [35]); as yellow prisms (EtOAc). The ¹H-NMR and ¹³C-NMR data are in agreement with literature data [26,35].

2-(2,4-Dimethoxyphenyl)benzisoselenazol-3(2H)-one 10f. The general procedure starting from 2,4-dimethoxyaniline (1.99 g, 13 mmol) at RT was employed, with a 6-h reaction time with separation by column chromatography using CHCl₃ as an eluent to obtain **10f**. Yield: (0.87 g, 65% calculated on the benzoyl chloride **9**); m.p.: 239–240 °C; as pale yellow prisms (CHCl₃, 75 mL/g). IR (KBr, cm⁻¹) ν 3050, 1601 (C=O), 1588, 1560, 1510, 1442, 1208 (C–N), 1162 (C–O), 1044 (C–O), 747 (C–Se). ¹H-NMR (300 MHz, CDCl₃): δ 3.79 (s, 3H, OCH₃), 3.84 (s, 3H, OCH₃), 6.54 (dd, *J* = 8.6 Hz, *J* = 2.6 Hz, 1H, Ph-H), 6.59 (d, *J* = 2.6 Hz, 1H, Ph-H), 7.25 (d, *J* = 8.6 Hz, 1H, Ph-H), 7.41 (ddd, *J* = 7.8 Hz, *J* = 7.2 Hz, *J* = 1.2 Hz, 1H, Ar-H), 7.60 (ddd, *J* = 7.8 Hz, *J* = 7.2 Hz, *J* = 1.1 Hz, 1H, Ar-H), 7.95 (dd, *J* = 7.2 Hz, *J* = 1.2 Hz, 1H, Ar-H), 7.99 (dd, *J* = 8.0 Hz, *J* = 1.1 Hz, 1H, Ar-H). ¹H-NMR (400 MHz, DMSO-*d*₆): δ 3.75 (s, 3H, OCH₃), 3.81 (s, 3H, OCH₃), 6.58 (dd, *J* = 8.6 Hz, *J* = 2.7 Hz, 1H, Ph-H), 6.70 (d, *J* = 2.7 Hz, 1H, Ph-H), 7.23 (d, *J* = 8.6 Hz, 1H, Ph-H), 7.45 (ddd, *J* = 7.6 Hz, *J* = 7.3 Hz, *J* = 1.1 Hz, 1H, Ar-H), 7.65 (ddd, *J* = 7.9 Hz, *J* = 7.3 Hz, *J* = 1.5 Hz, 1H, Ar-H), 7.85 (ddd, *J* = 7.6 Hz, *J* = 1.5 Hz, *J* = 0.6 Hz, 1H, Ar-H), 8.05 (ddd, *J* = 7.9 Hz, *J* = 1.1 Hz, *J* = 0.6 Hz, 1H, Ar-H). ¹³C-NMR (101 MHz, DMSO-*d*₆): δ 55.45 (OCH₃), 55.68 (OCH₃), 99.39, 104.86, 119.75, 125.74, 125.78, 127.30, 127.74, 130.33, 131.77, 140.29, 156.29, 160.09, 165.76

(C=O). HRMS (TOF MS ESI): m/z for $C_{15}H_{13}NO_3Se + Na^+$ calculated: 357.9953; found: 357.9961. Anal. Calcd. for $C_{15}H_{13}NO_3Se$ (334.2286): C, 53.90; H, 3.92; N, 4.19. Found: C, 53.94; H, 3.75; N, 4.11.

2-(3,4-Dimethoxyphenyl)-1,2-benzoselenazol-3(2H)-one **10g** [53]. The general procedure starting from 3,4-dimethoxyaniline (0.61 g, 4.0 mmol) and anhydrous Et_3N (0.81 g, 1.1 mL, 8 mmol) in anhydrous MeCN (30 mL) at RT was employed, with a 48-h reaction time. Distilled water (60 mL) was added portionwise and left to refrigerator overnight to crystallise. The formed solid was filtered off, washed with water, and dried in the air to obtain **10g**. Yield: (0.91 g, 68%). The analytically pure sample was obtained by column chromatography using $CHCl_3:EtOAc$ in gradient as an eluent with 40% yield; m.p.: 159–161 °C; as colorless crystals (EtOAc). 1H -NMR (300 MHz, DMSO- d_6): δ 3.78 (s, 6H, 2OCH₃), 6.99 (d, J = 8.6 Hz, 1H, Ph-H), 7.03 (dd, J = 8.6 Hz, J = 1.9 Hz, 1H, Ph-H), 7.29 (d, J = 1.9 Hz, 1H, Ph-H), 7.47 (dd, J = 7.8 Hz, J = 7.2 Hz, 1H, Ar-H), 7.67 (dd, J = 8.1 Hz, J = 7.2 Hz, 1H, Ar-H), 7.90 (d, J = 7.8 Hz, 1H, Ar-H), 8.09 (d, J = 8.1 Hz, 1H, Ar-H). ^{13}C -NMR (75 MHz, DMSO- d_6): 55.5 (OCH₃), 55.6 (OCH₃), 109.5 (C-H), 111.8 (C-H), 117.2 (C-H), 125.8 (C-H), 126.1 (C-H), 127.8 (C-H), 128.4 (C), 132.0 (C-H), 132.4 (C), 139.0 (C), 147.1 (C), 148.7 (C), 164.9 (C=O). HRMS (TOF MS ESI): m/z for $C_{15}H_{13}NO_3Se + Na^+$ calculated: 357.9953; found: 357.9951. Anal. Calcd. for $C_{15}H_{13}NO_3Se$ (334.2282): C, 53.90; H, 3.92; N, 4.19. Found: C, 53.79; H, 4.05; N, 4.08.

3.2. Evaluation of Antimicrobial Activity

The antimicrobial activity of compounds **4–6**, **7**, **10** was determined by the 96-well microdilution assay, according to Clinical and Laboratory Standards Institute (2008), 3rd ed. M27-A3 [73] with slight modifications, using Gram-positive bacterial strains: *Staphylococcus aureus* ATCC 6538, *Staphylococcus epidermidis* ATCC 14990, *Enterococcus faecalis* ATCC 29212, *Enterococcus hirae* ATCC 10541; Gram-negative bacterial strains: *Escherichia coli* ATCC 10536, *Pseudomonas aeruginosa* ATCC 15442; yeast-like fungi: *Candida albicans* ATCC 10231, *Candida glabrata* ATCC 90030; and filamentous fungus *Aspergillus niger*. Standard strains were supplied by American Type Culture Collection (ATCC), while *A. niger* was obtained from the Collection of Microorganisms of Wroclaw Medical University. Bacterial strains were cultured in Lysogeny broth (LB) at 37 °C and yeast strains in Sabouraud dextrose broth at 30 °C.

Briefly, stock solutions in DMSO were serially diluted in LB or Sabouraud dextrose broth to obtain a final range of 0.125–128 $\mu\text{g/mL}$ concentration in microdilution sterile plates (Sarstedt, Stare Babice, Poland). Microbial inocula were prepared to obtain a final optical density at $\lambda = 600$ nm (OD_{600}) = $0.1 \pm 10\%$ in a well. The plates were cultivated for 24 h at 37 °C (bacteria) or 48 h at 30 °C (fungi). Afterwards, OD_{600} was measured using a microplate reader ASYS UVM 340 (Biogenet, Józefów, Poland). The concentration which resulted in $\geq 90\%$ growth inhibition was determined as the minimal inhibitory concentration (MIC_{90} ; $\mu\text{g/mL}$).

3.3. Evaluation of Antiviral Activity and Cytotoxicity

The compounds **4–5**, **7**, **10** at various concentrations were incubated with the following viruses: HHV-1 (human herpes virus type 1), EMCV (encephalomyocarditis virus), and VSV (vesicular stomatitis virus) for 1 h at RT. HHV-1 was used at the dose of 10^6 TCID₅₀/mL, whereas EMCV and VSV were used at the dose of 10^8 TCID₅₀/mL. The TCID₅₀ (tissue culture infectious dose) determines the cytopathic effect caused by the virus in about 50% of infected cells. The virus titer was measured in human cell line A549 (human lung adenocarcinoma cell line, ATCC 185) which was observed for cytopathic effect after 48 h. The minimal concentration that caused 1000-fold decrease of virus titer was taken as the minimal inhibitory concentration, MIC ($\mu\text{g/mL}$).

In parallel, the cytotoxicity of the compounds was determined in the A549 cell line (cell line used in antiviral activity test) by incubation at various concentrations for 48 h at 37 °C in the atmosphere of 5% CO₂ in air. The minimal concentration which was toxic to approximately 50% of the cells was taken as the TCCD₅₀ (Tissue Culture Cytotoxic Dose).

Both, cytotoxicity and virus titers were determined using MTT assay. To ensure that colourful compounds did not interfere with MTT reagents, the extent of cell damage was evaluated by microscopic examination prior to MTT assay.

4. Conclusions

The reaction of bis[(2-chlorocarbonyl)phenyl] diselenide **3** with various aminophenols and their derivatives is a convenient synthetic route to bis[2-(hydroxyphenylcarbamoyl)]phenyl diselenides **4** (reduced forms of ebselenols) and other compounds containing protected hydroxyl groups. This general approach allows the production of ebselenol-derived diselenides with higher yields than the alternative method based on the reductive benzenoselenazolone ring opening of ebselenols. The secondary amine such as morpholine and benzohydrazide also reacted on their nitrogen atoms to form bis[2-(*N*-morpholinecarbamoyl)]phenyl and bis[2-(*N*-benzamidecarbamoyl)]phenyl diselenides, respectively, while reaction with aminophenol derivatives containing a blocked amine group such a 2-hydroxyacetanilide and natural antimycin A led to corresponding phenolic esters of bis(2-carboxy)phenyl diselenide. As a result of our studies, 15 new structures have been developed, including 13 diselenides and two benzenoselenazol-3(*2H*)-ones.

The investigation of the antimicrobial and antiviral activities of bis[2-(hydroxyphenylcarbamoyl)]phenyl diselenides and their derivatives as well as cyclic analogues have revealed new biologically active structures. The compounds, particularly bis[2-(hydroxyphenylcarbamoyl)]phenyl diselenides and reference benzenoselenazol-3(*2H*)-ones, exhibited high antimicrobial activity against Gram-positive bacterial species (*Enterococcus* spp., *Staphylococcus* spp.), and some compounds were also active against Gram-negative *E. coli* and fungi (*Candida* spp., *A. niger*). The antiviral activity evaluation allowed the identification of substances with high anti-HHV-1 and moderate anti-EMCV activities. The chemotherapeutic indices for some of these compounds were significantly higher (concentration active against virus was lower than the one showing a cytotoxic effect) than for other organoselenium compounds previously reported in the literature.

As novel molecules with the ability to inhibit microbial and viral functions are in high demand, the results of our studies are encouraging for further exploration of organoselenium compounds as anti-infective agents.

Supplementary Materials: Supplementary materials are available online.

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Sample Availability: Samples of the compounds 4–10 are available from the authors.



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