

Phosphine-Free Palladium Catalyzed Heck reaction with using natural compounds

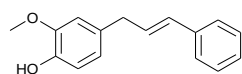
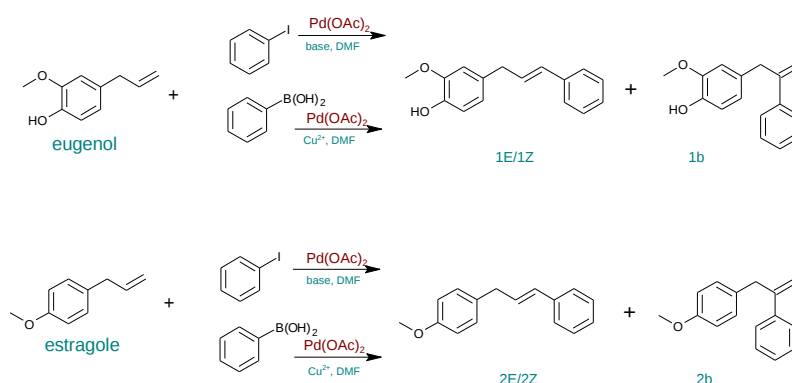
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General Remarks: Products have been characterized by the GC-MS (HP II 5890 + Mass Selective Detector HP 5971A) and by ^1H NMR and ^{13}C NMR. NMR spectra have been measured for the post-reaction mixtures. Unless noted, ^1H NMR spectra were recorded on Bruker 500 MHz in CDCl_3 , ^{13}C NMR spectra were recorded on 125 MHz in CDCl_3 .

Control experiment 1:



1E/1Z (2-methoxy-4-(3-phenyl-2-propen-1-yl)-phenol)

(**1E/1Z**) MS: m/z (%) = 91 (30), 115 (37), 131 (16), 165 (12), 179 (17), 207 (23), 240 (100) [M^+]

^1H -NMR (500 MHz, CDCl_3): δ = 7.37–7.33 (m, 2,6H, H_{Ar}), 7.31–7.27 (m, 2,4H, H_{Ar}), 6.91 (d, J = 12.4Hz, 3H, H_{Ar}), 6.42 (d, J = 15.8Hz, 1H, CH), 6.28–6.22 (m, 1H, CH), 3.86 (s, 3H, CH_3), 3.56 (d, J = 7.1Hz, 2H, CH_2) ppm.

^{13}C -NMR (125 MHz, CDCl_3): δ = 146.82, 145.28, 140.50, 130.89, 130.12, 128.75, 128.54, 126.97, 126.20, 119.82, 114.59, 108.27, 55.90, 39.36 ppm.

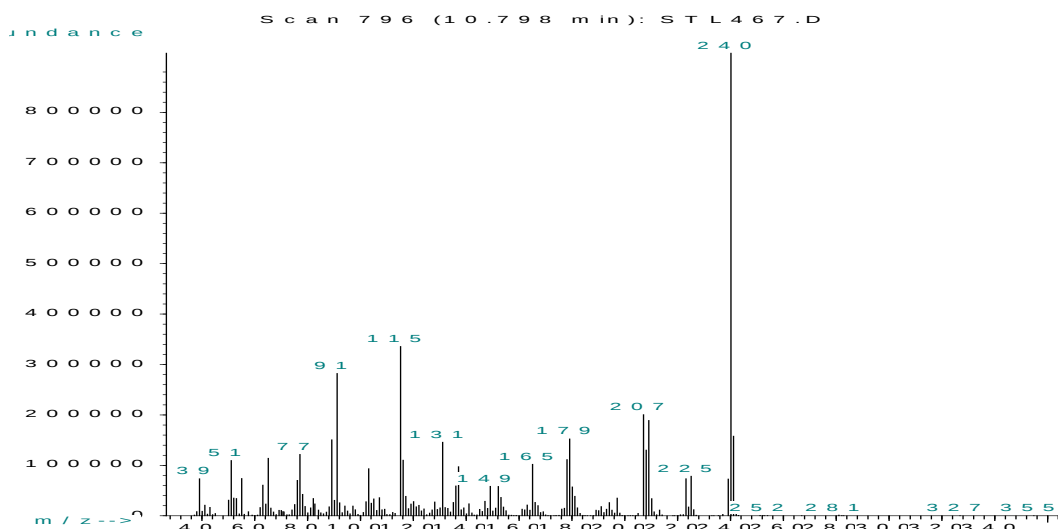
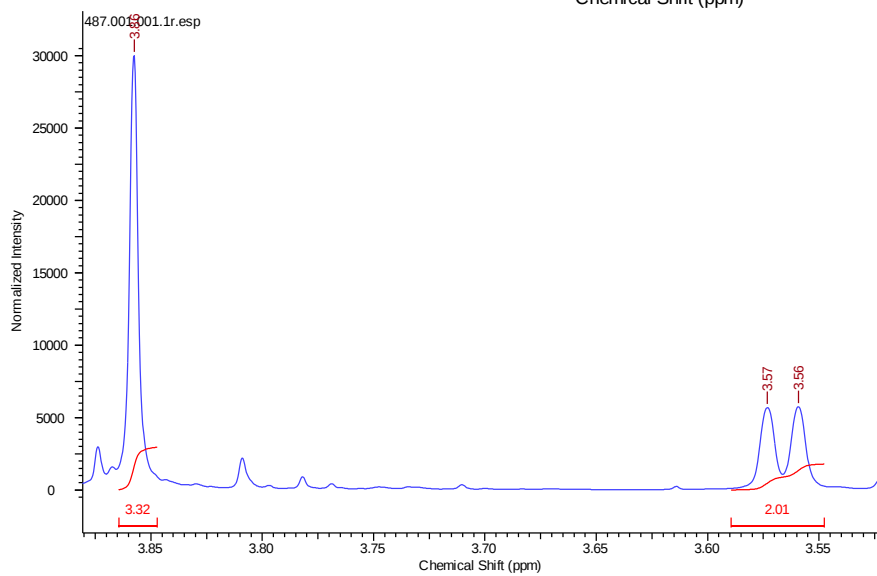
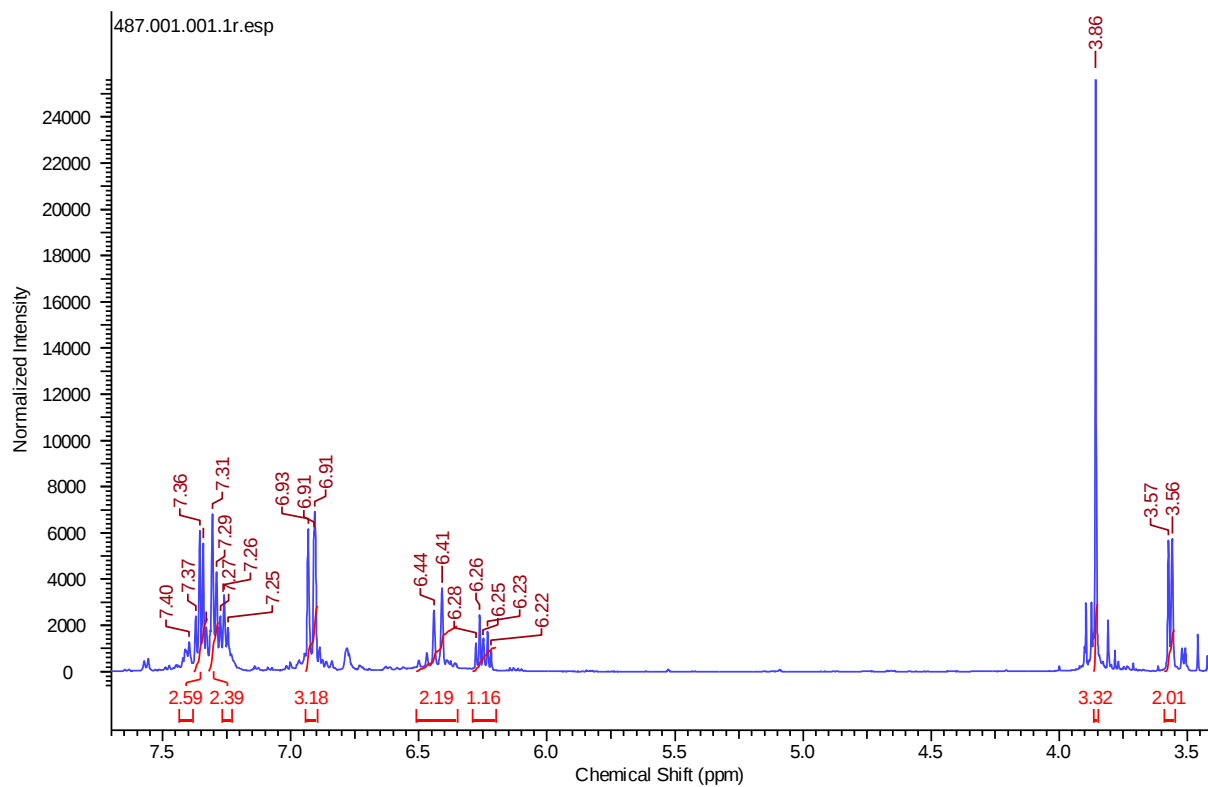


Figure S1. Mass spectrum of (2-methoxy-4-(3-phenyl-2-propen-1-yl)-phenol) (**1E**) and (**1Z**).



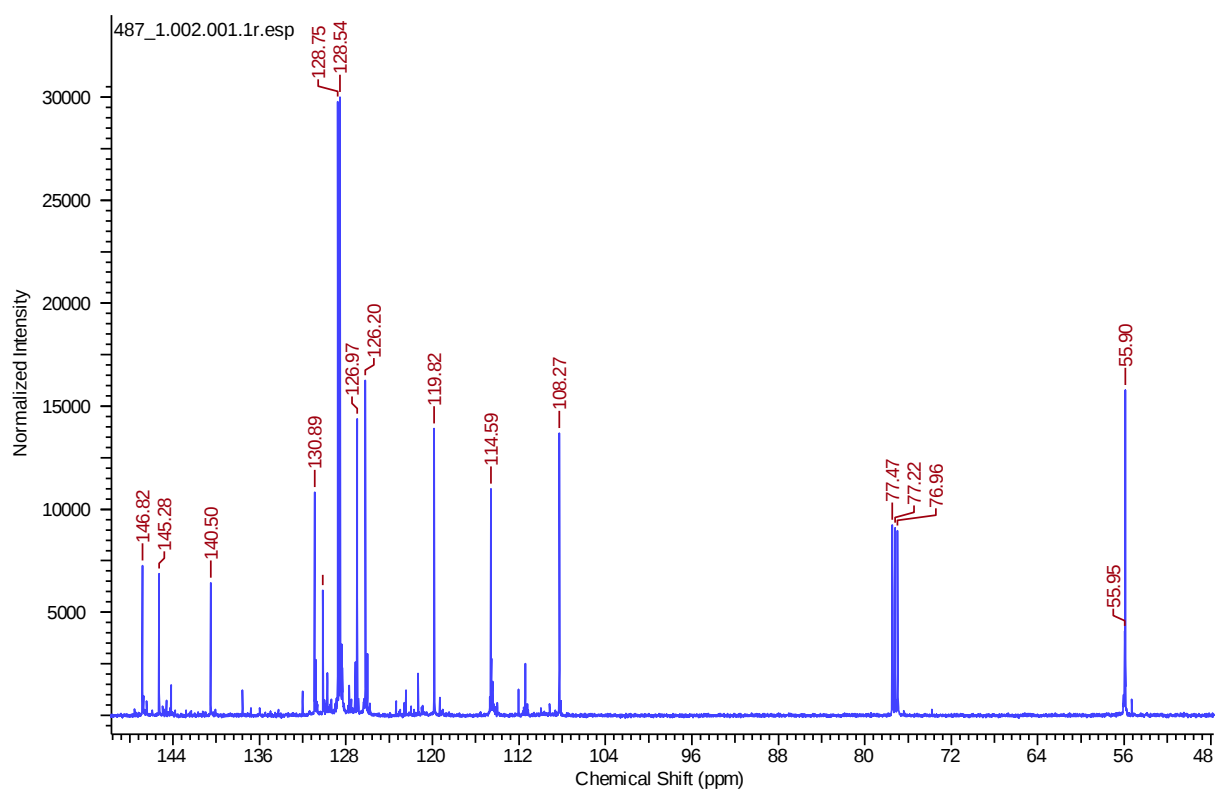
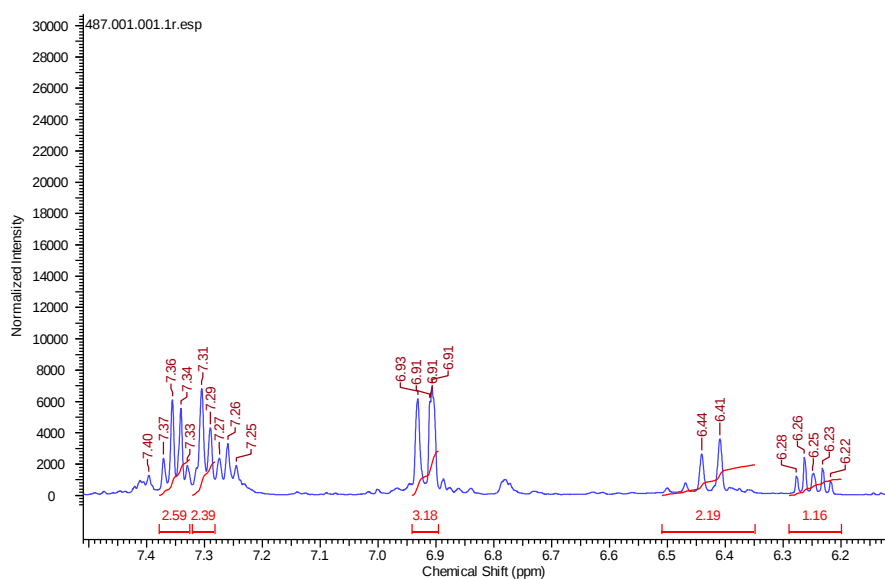
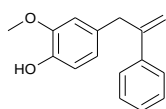


Figure S2. ^1H and ^{13}C NMR spectra of (2-methoxy-4-(3-phenyl-2-propen-1-yl)-phenol) (**1E**).
Reaction conditions (method A): PhI (1 mmol), eugenol (1 mmol), K_2CO_3 (2 mmol), $\text{Pd}(\text{OAc})_2$ (1×10^{-5} mol),
DMF: H_2O (4:1), 3h, 100°C



1b (2-methoxy-4-(2-phenylprop-2-en-1-yl)-phenol)

MS: m/z (%) = 77 (24), 103 (24), 137 (100), 225 (17), 240 (85) [M^+]

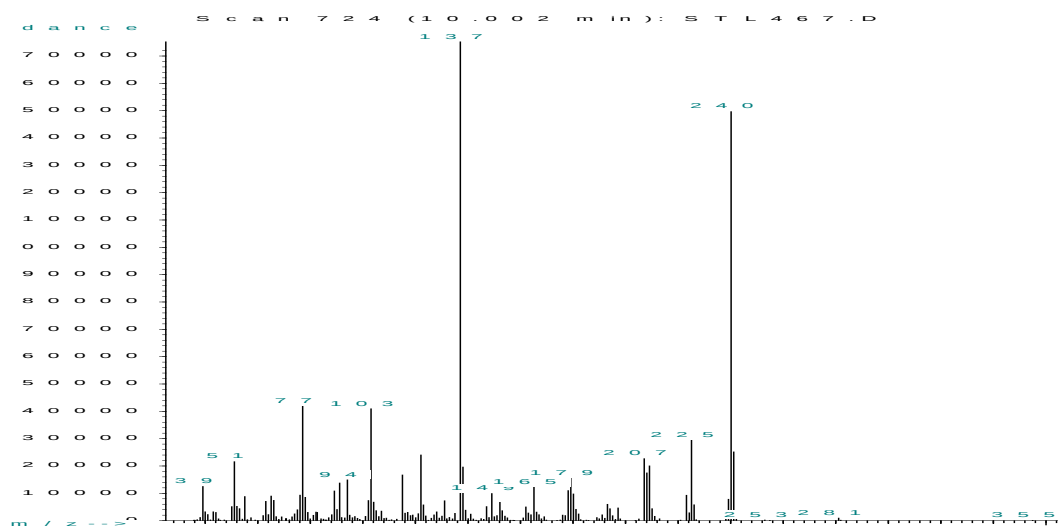
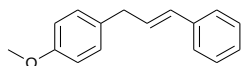


Figure S3. Mass spectrum of 1b (2-methoxy-4-(2-phenylprop-2-en-1-yl)-phenol) (1b).



2E/2Z (1-Phenyl-3-(4-methoxyphenyl)-1-propene)

MS: m/z (%) = 77 (21), 91 (25), 115 (61), 121 (31), 165 (14), 178 (14), 193 (27), 209 (21), 224 (100) [M^+] [1]

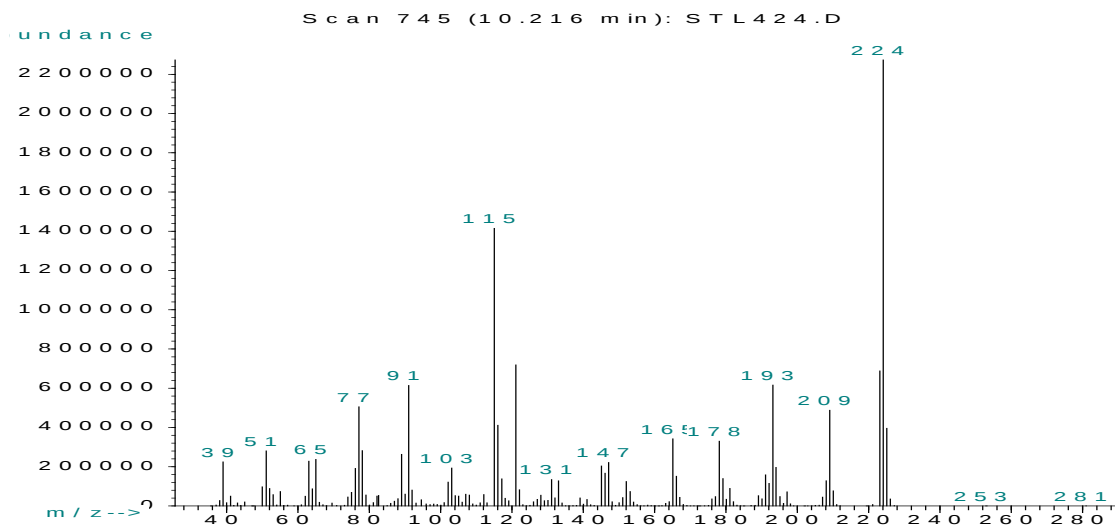
$^1\text{H-NMR}$ (500 MHz, CDCl_3): δ = 7.49–7.32 (m, 12H, H_{Ar}), 7.29 (d, J = 8.8, 2H, H_{Ar}), 6.99 (d, 8.8 Hz, 2H, H_{Ar}), 6.96 (d, 8.8 Hz, 2H, H_{Ar}), 6.55 (m, 2H, CH, CH), 6.47 (td, J = 15.6, 6.7 Hz, 1H, CH), 6.35 (td, J = 15.6, 6.7 Hz, 1H, CH), 3.89 (s, 3H, OCH_3), 3.88 (s, 3H, OCH_3), 3.65 (d, J = 6.9, 2H, CH_2), 3.62 (d, J = 6.7, 2H, CH_2) ppm [2].

2E ((E)-1-Phenyl-3-(4-methoxyphenyl)-1-propene)

$^{13}\text{C-NMR}$ (125 MHz, CDCl_3): δ = 158.29, 137.73, 130.93, 130.65, 130.03, 129.78, 128.67, 127.42, 126.30, 113.95, 55.36, 38.62 ppm [2].

2Z ((Z)-1-Phenyl-3-(4-methoxyphenyl)-1-propene)

$^{13}\text{C-NMR}$ (125 MHz, CDCl_3): δ = 159.06, 140.63, 132.32, 130.49, 129.85, 128.84, 128.64, 127.22, 127.20, 114.11, 55.28, 39.51 ppm.



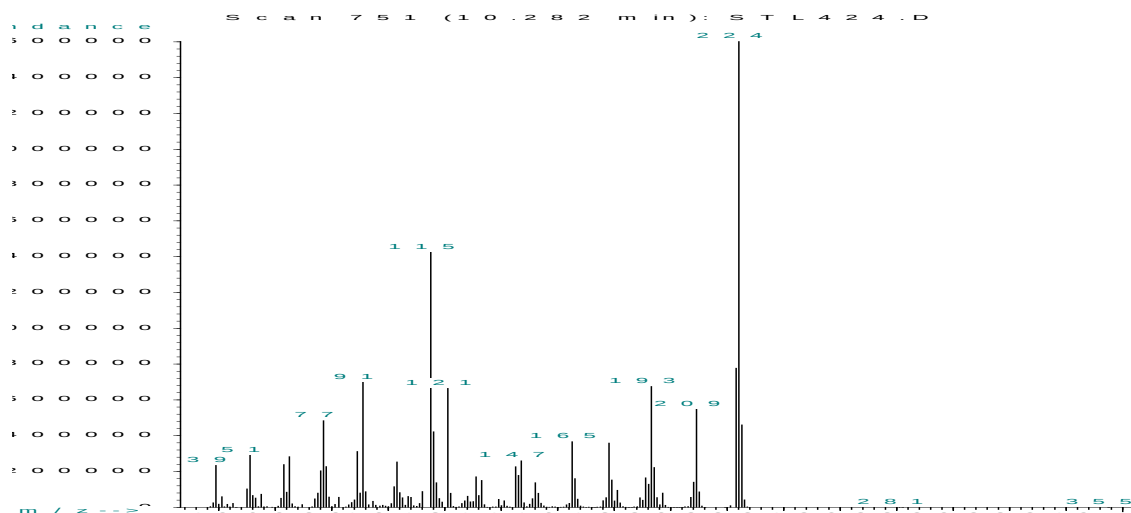
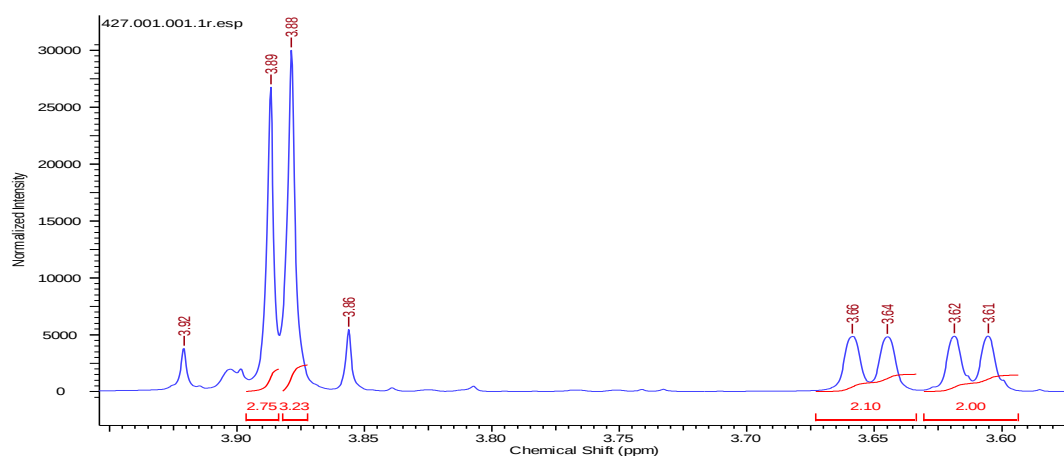
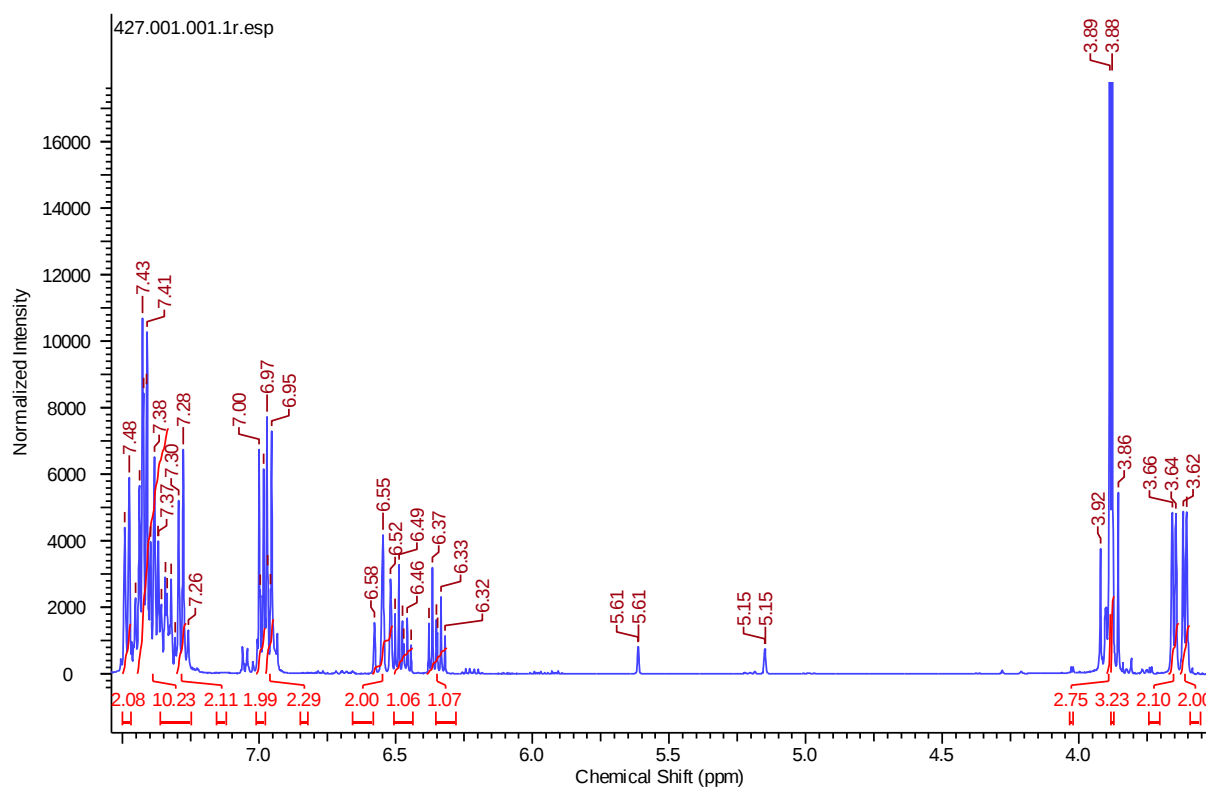
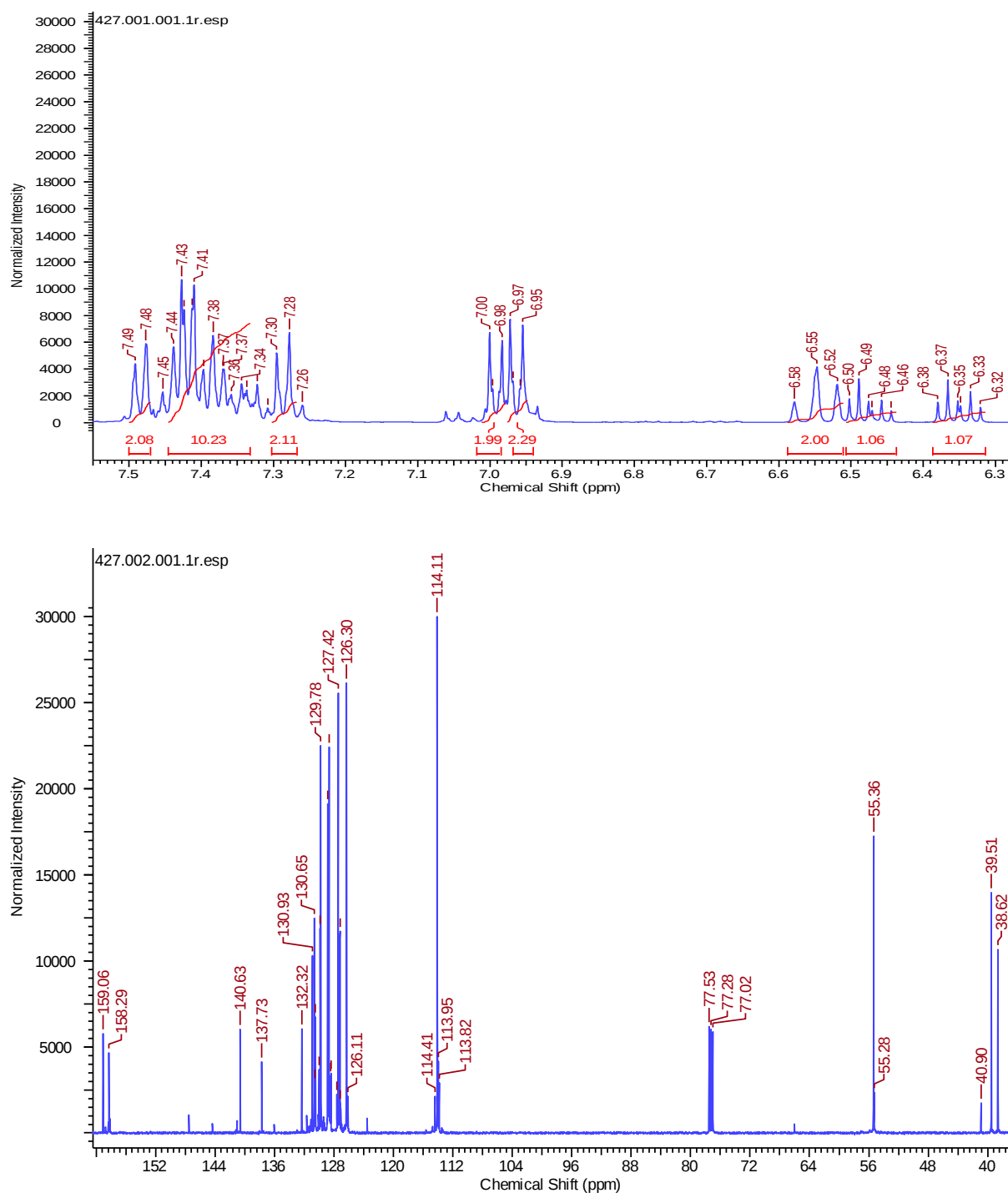


Figure S4. Mass spectrum of 1-Phenyl-3-(4-methoxyphenyl)-1-propene (**2E**) and (**2Z**).





2b (1-methoxy-4-(2-phenyl-2-propen-1-yl)-benzene)

MS: m/z (%) = 77 (23), 103 (13), 121 (100), 193 (10), 209 (10), 224 (45) [M^+]

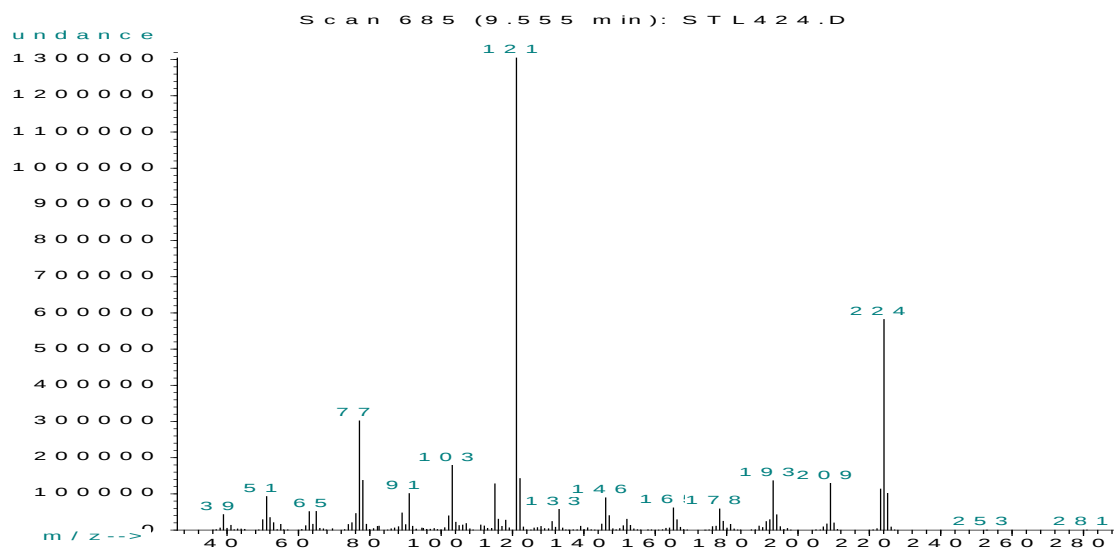
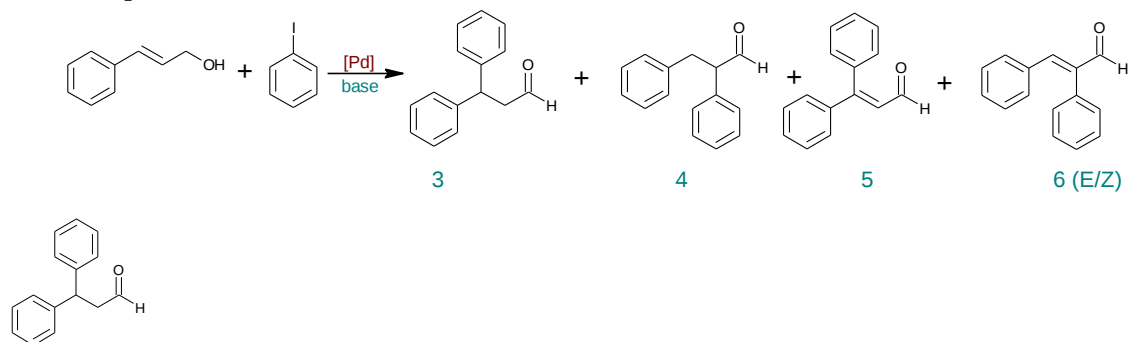


Figure S6. Mass spectrum of (1-methoxy-4-(2-phenyl-2-propen-1-yl)-benzene) (2b).

Control experiment 2:



3 (3,3-diphenylpropanal)

MS: m/z (%) = 77 (20), 105 (23), 167 (100), 192 (20), 210 (56) [M^+] [3].

$^1\text{H-NMR}$ (500 MHz, CDCl_3): δ = 9.74 (t, J = 1.91, 1H, CHO), 7.25–7.17 (m, 10H, H_{Ar}), 4.63 (t, J = 7.82 Hz, 1H, CH), 3.17 (dd, J = 7.82, 1.91 Hz, 2H, CH_2) ppm [3,5].

$^{13}\text{C-NMR}$ (125 MHz, CDCl_3): δ = 198.33, 142.20, 128.26, 127.24, 126.42, 46.95, 42.91 [3].

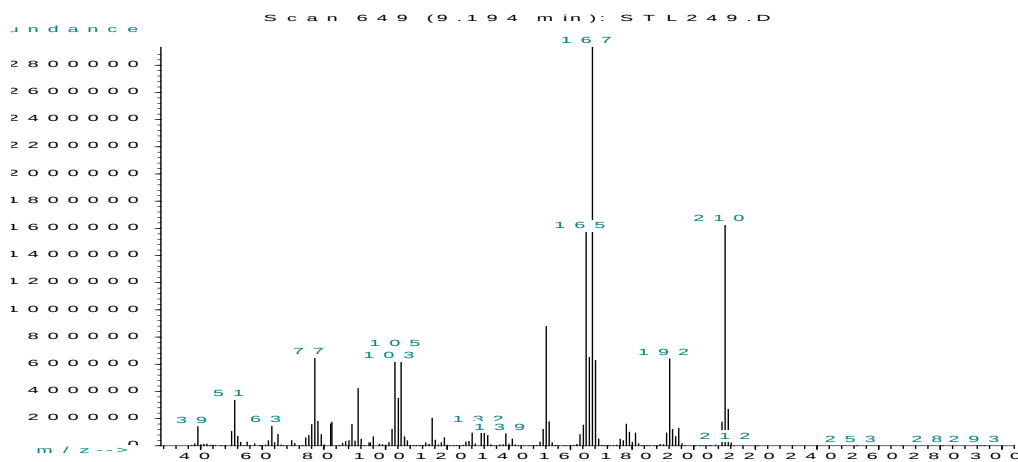


Figure S7. Mass spectrum of 3,3-diphenylpropanal (3).

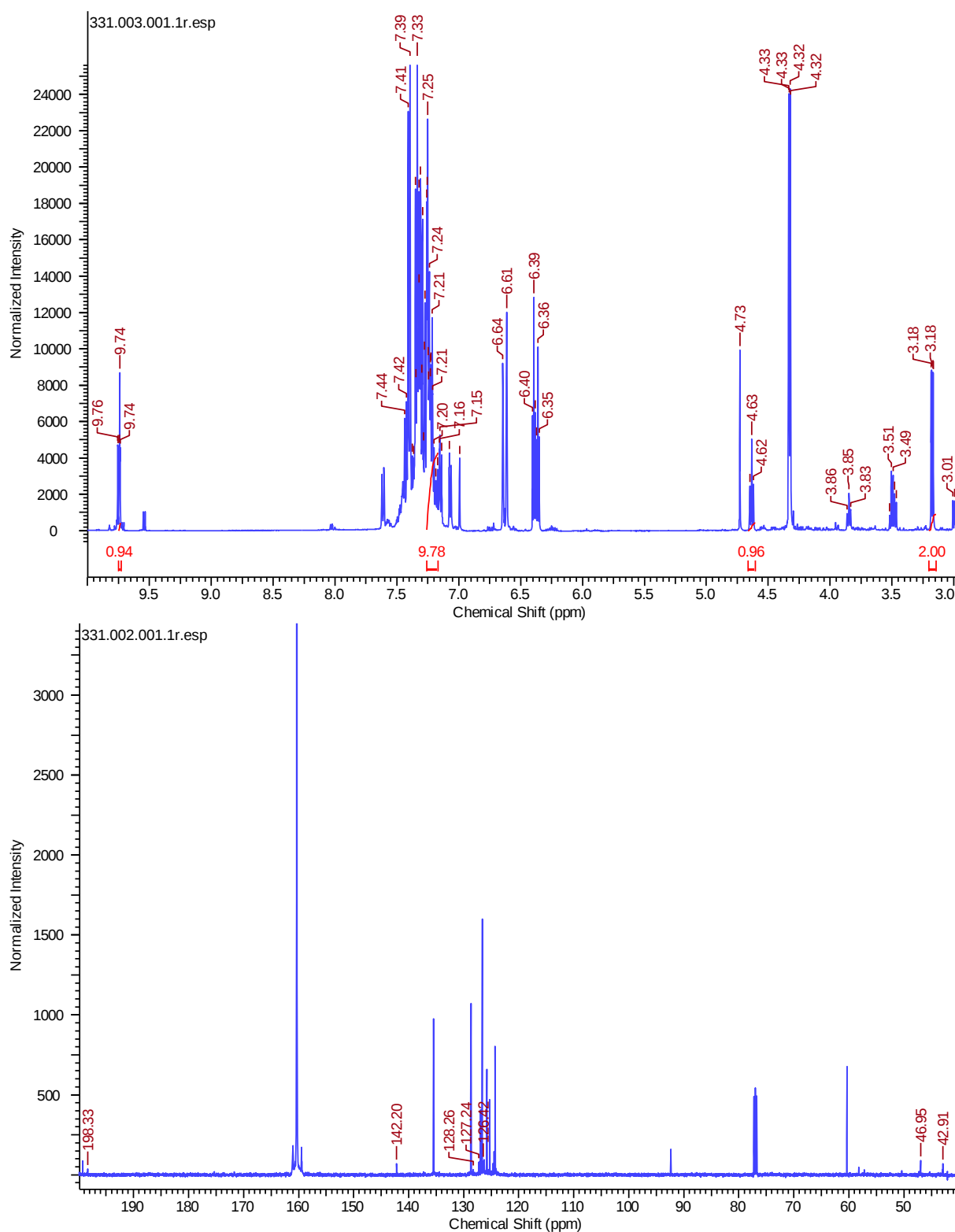
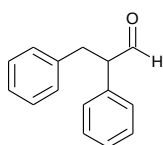


Figure S8. ¹H and ¹³C NMR spectra of 3,3-diphenylpropanal (**3**).

Reaction conditions (method C): PhI (1 mmol), cinnamyl alcohol (1 mmol), K₂HPO₄ (2 mmol), Pd(OAc)₂ (1×10⁻⁵ mol), DMF, 6h, 100°C



4 (2,3-diphenylpropanal)

MS: m/z (%) = 77 (12), 91 (100), 103 (9), 165 (9), 181 (18), 210 (32) [M⁺] [4]

^1H -NMR (500 MHz, CDCl_3): δ = 9.76 (d, J = 1.53, 1H, CHO), 7.31–7.26 (m, 10H, H_{Ar}), 3.85 (t, J = 6.68 Hz, 1H, CH), 3.48 (dd, J = 13.44, 6.68 Hz, 1H, CH_2), 2.98 (dd, J = 14.11, 7.82 Hz, 1H, CH_2) [5] ppm.
 ^{13}C -NMR (125 MHz, CDCl_3): δ = 199.24, 135.47, 128.67, 126.96, 126.63, 125.85, 125.79, 125.26, 124.29, 33.83, 28.68

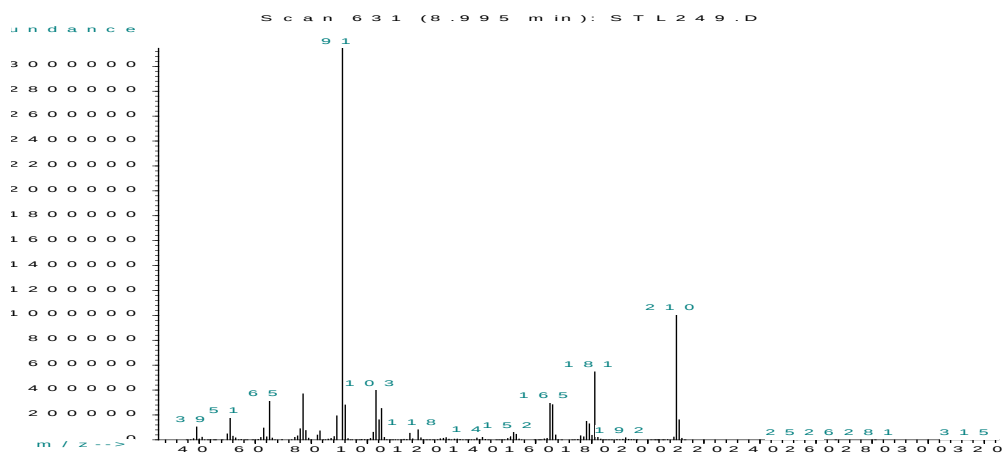
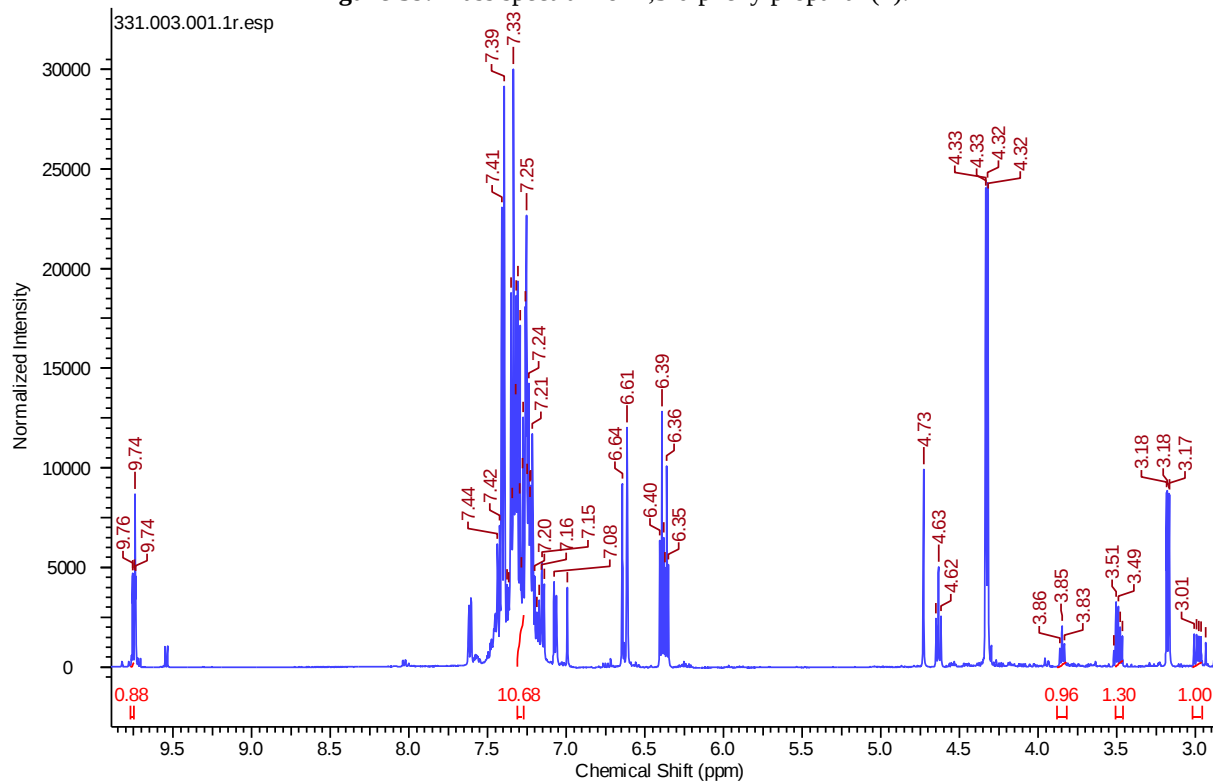
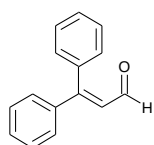
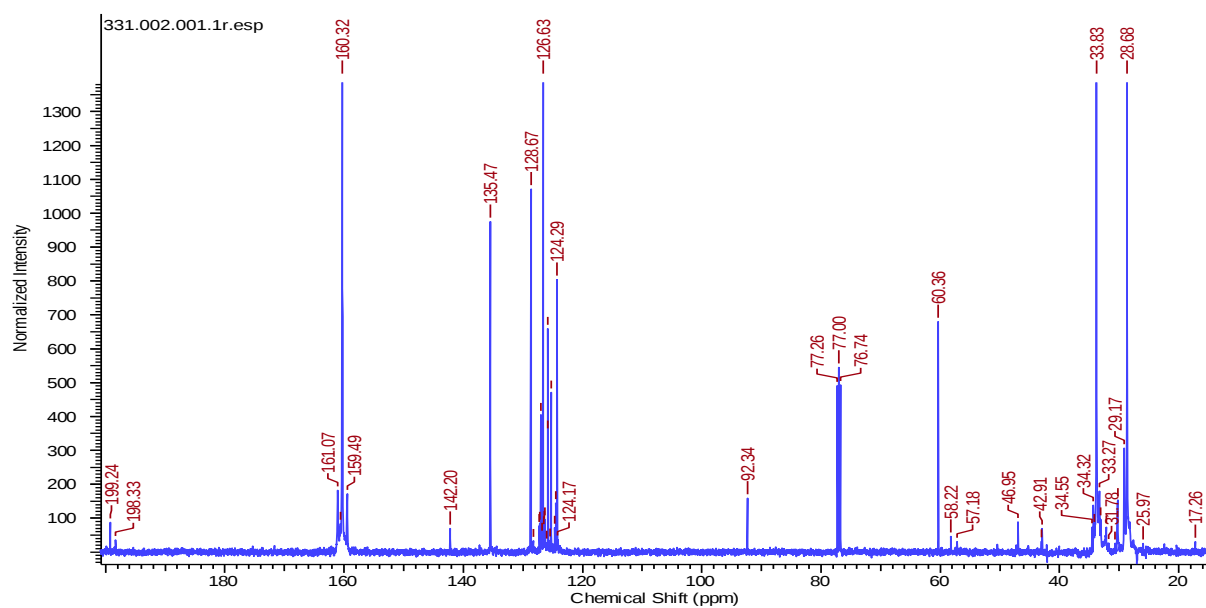


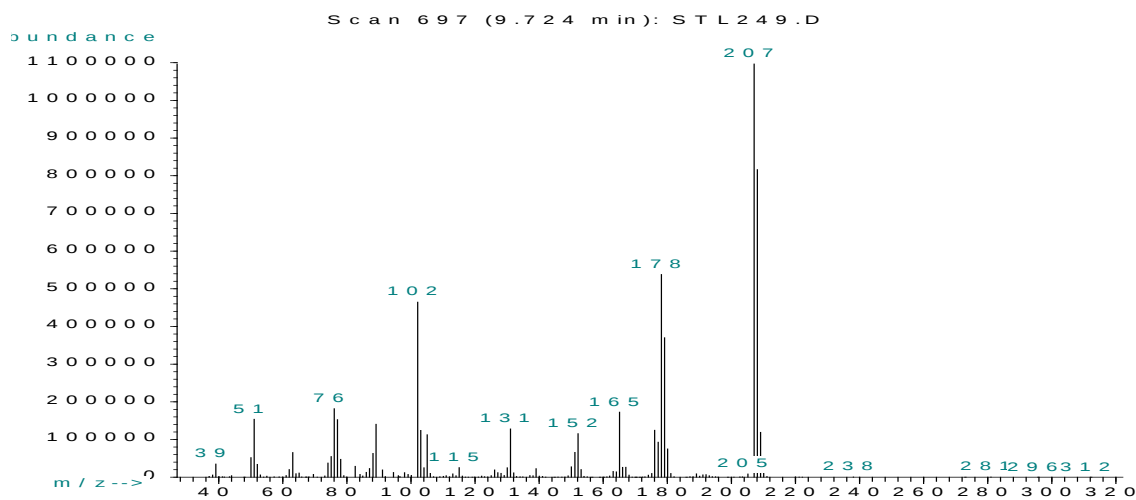
Figure S9. Mass spectrum of 2,3-diphenylpropanal (4).





5 (3,3-diphenylprop-2-enal)

MS: m/z (%) = 77 (18), 89 (14), 102 (44), 152 (12), 165 (15), 178 (50), 207 (100), 208 (73) [M^+] [6]



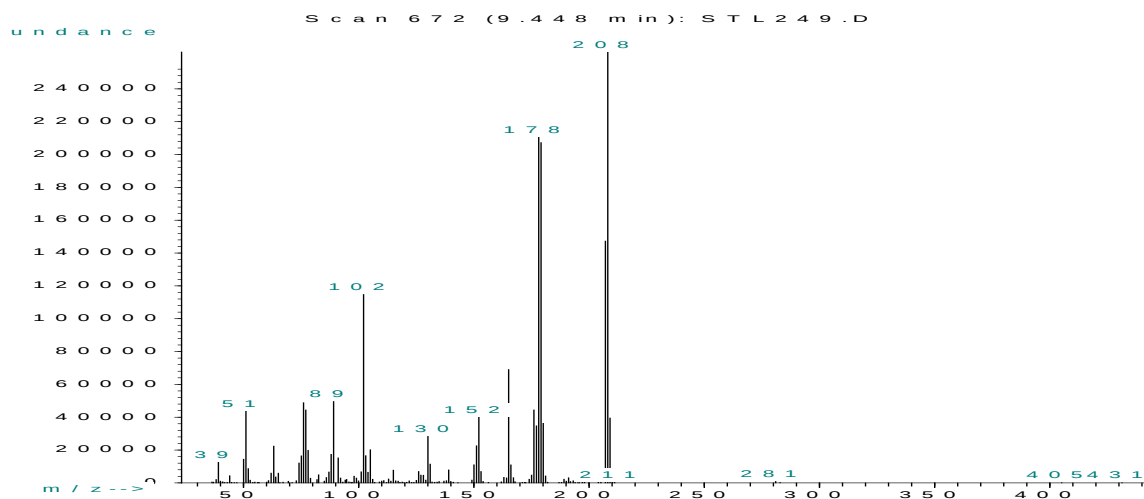
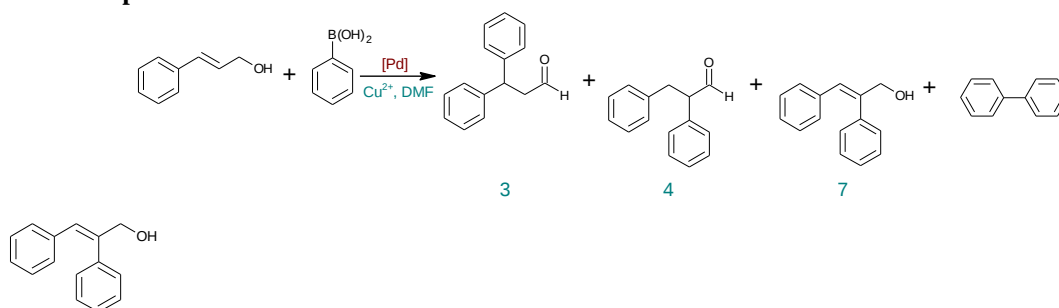


Figure S12. Mass spectrum of (2E)-2,3-diphenylprop-2-enal (6).

Control experiment 3:



7 (2,3-diphenyl-2-propen-1-ol)

MS: m/z (%) = 77 (18), 91 (54), 105 (100), 165 (114), 178 (25), 210 (46).

$^1\text{H-NMR}$ (500 MHz, CDCl_3): δ = 7.45–7.42 (m, 5H, H_{Ar}), 7.33–7.10 (m, 5H, H_{Ar}), 7.02 (s, 1H, CH), 4.73 (s, 2H, CH_2), 1.86 (s, 1H, OH) ppm.

$^{13}\text{C-NMR}$ (125 MHz, CDCl_3): δ = 140.59, 138.71, 136.86, 128.89, 128.58, 128.32, 127.62, 127.29, 126.54, 126.23, 68.30 ppm.

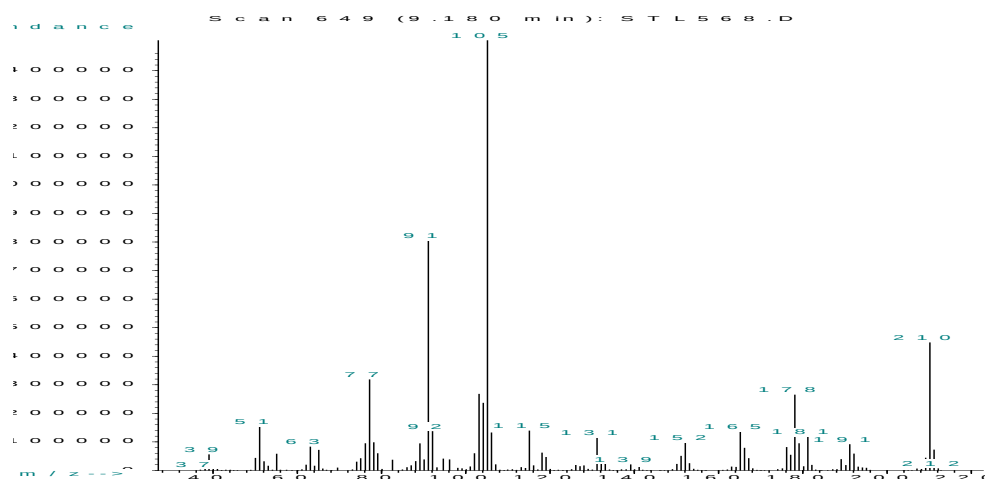


Figure S13. Mass spectrum of 2,3-diphenyl-2-propen-1-ol (7).

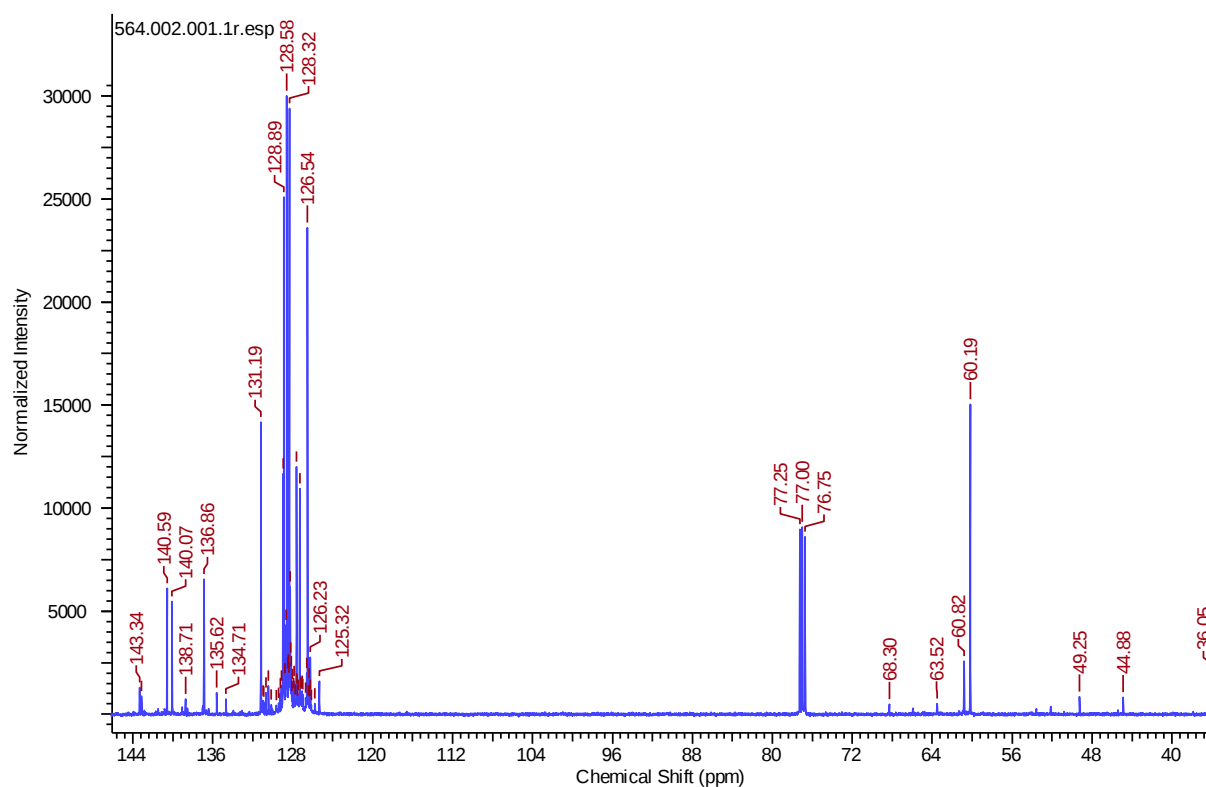
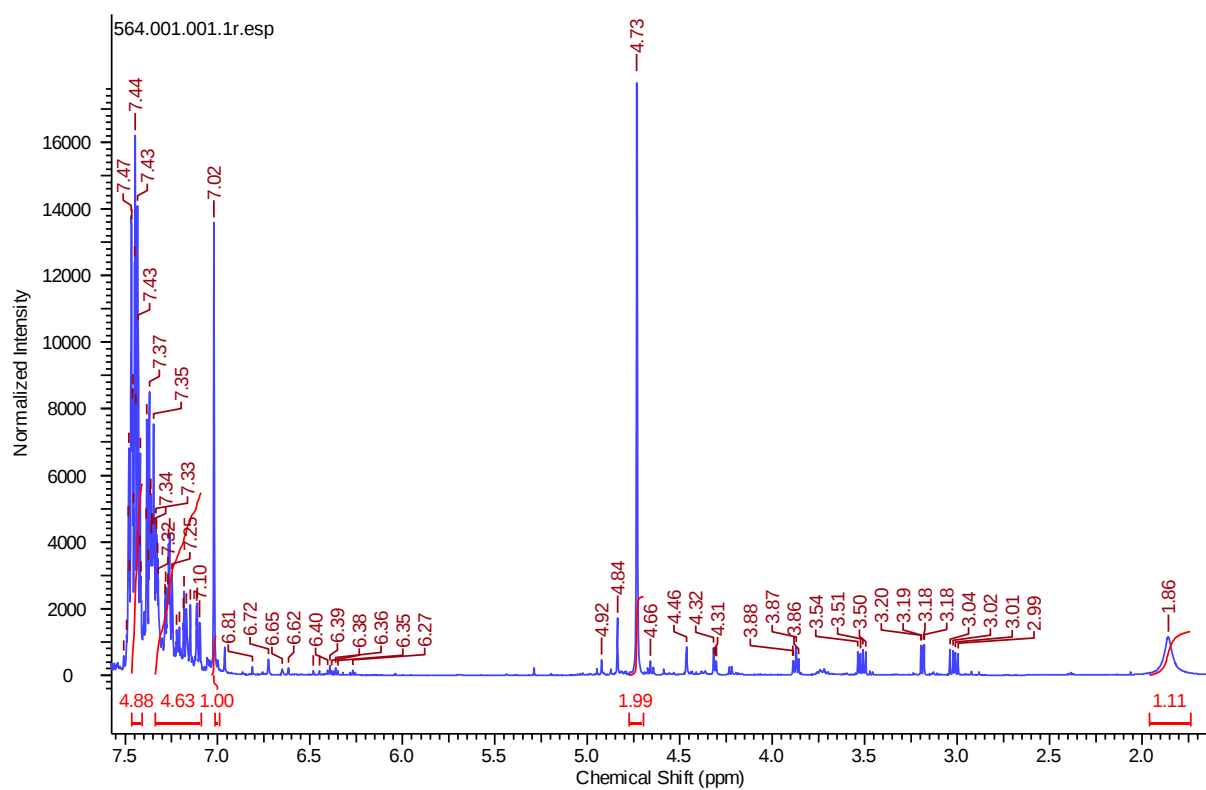
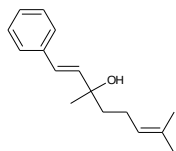


Figure S14. ^1H and ^{13}C NMR spectra of 2,3-diphenyl-2-propen-1-ol (7).

Control experiment 4:



MS: m/z (%) = 43 (81), 69 (24), 77 (15), 91 (20), 105 (11), 129 (46), 147 (100), 172 (46), 197 (4), 230 (2) [M^+]

¹³C-NMR (125 MHz, CDCl₃): δ = 137.05, 136.61, 123.02, 128.50, 127.26, 127.09, 126.33, 124.30, 73.43, 42.51, 28.24, 25.66, 22.93, 17.70.

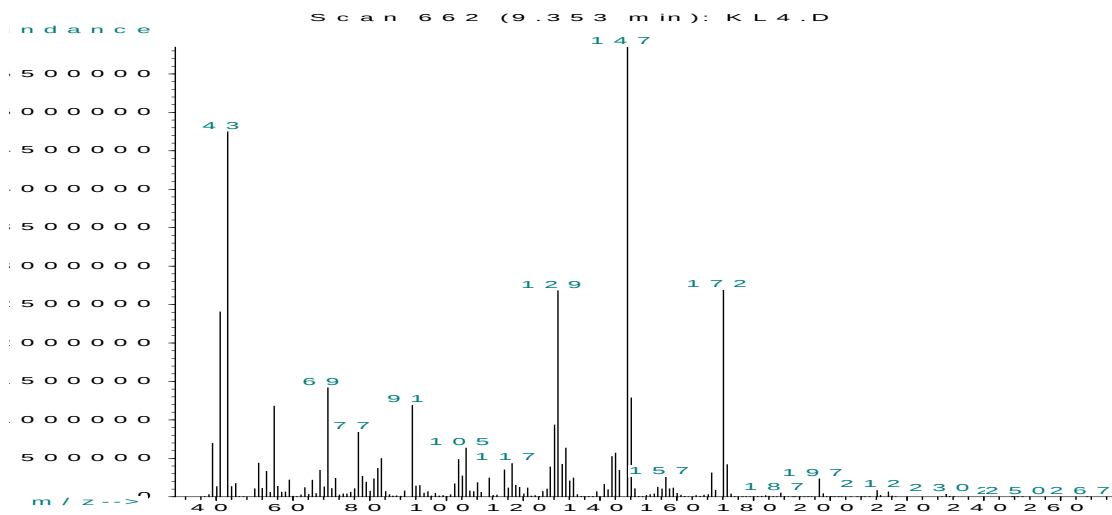


Figure S15. Mass spectrum of (1E)-3,7-dimethyl-1-phenylocta-1,6-dien-3-ol (**8**).

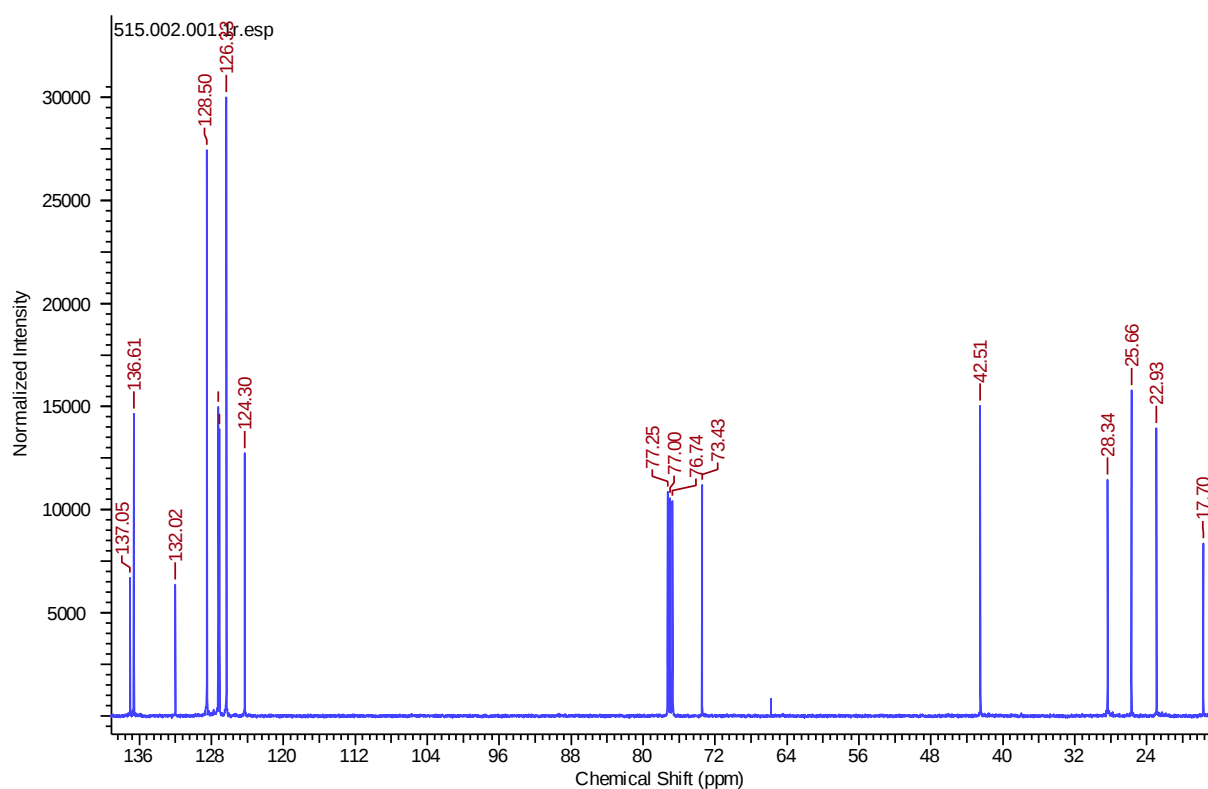
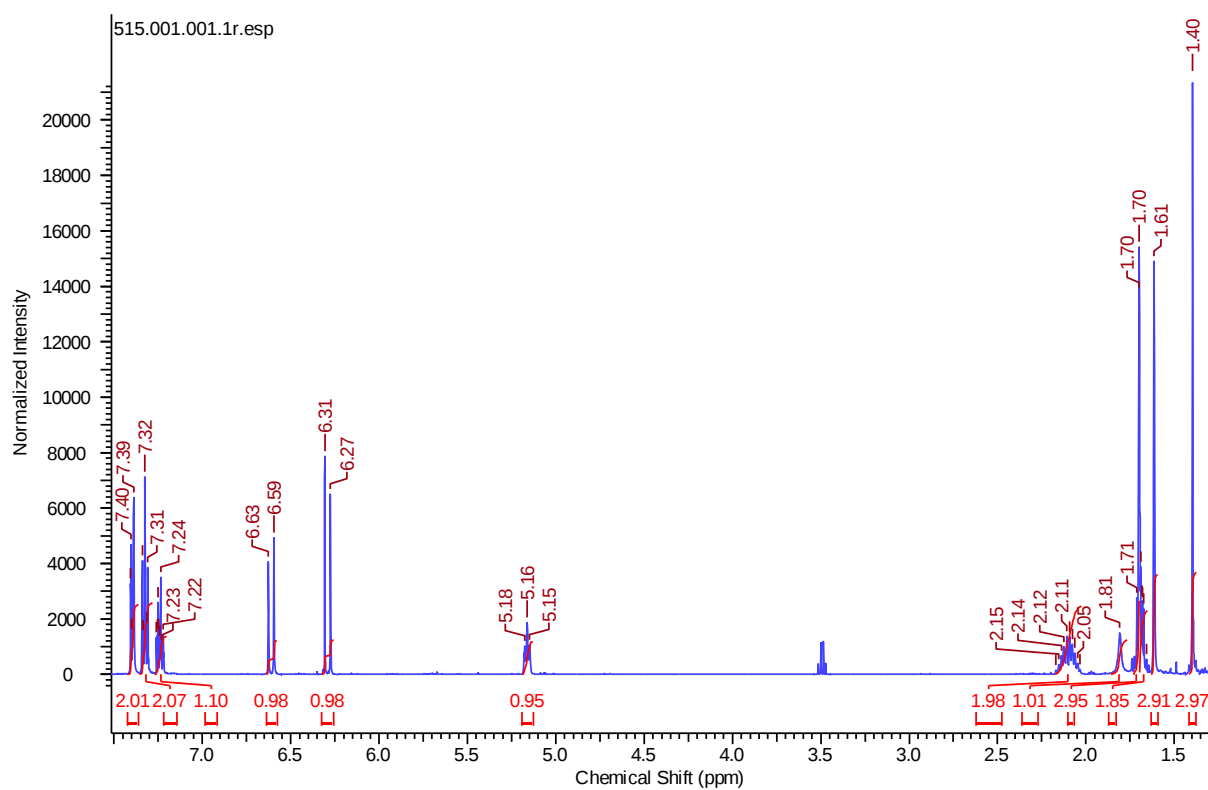
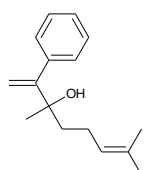


Figure S16. ^1H and ^{13}C NMR spectra of (1E)-3,7-dimethyl-1-phenylocta-1,6-dien-3-ol (**8**).
 Reaction conditions (method C): PhI (1 mmol), linalool (1 mmol), Et_3N (2 mmol), $\text{Pd}(\text{OAc})_2$ (1×10^{-5} mol), DMF, 5h, 100°C



9 (3,7-dimethyl-2-phenylocta-1,6-dien-3-ol)

MS: m/z (%) = 43 (33), 77 (6), 91 (8), 105 (5), 129 (24), 147 (100), 172 (6), 212 (5), 230 (2) [M⁺]

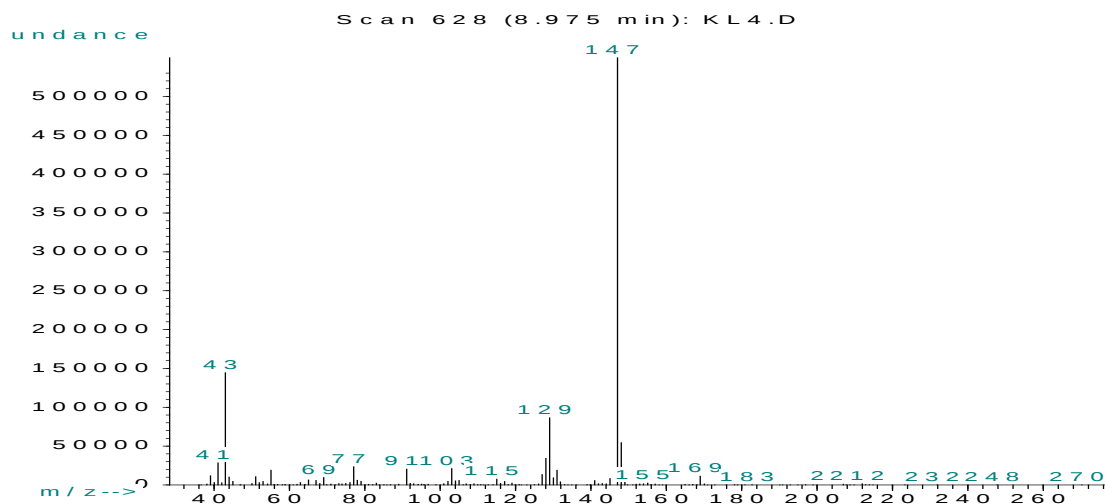
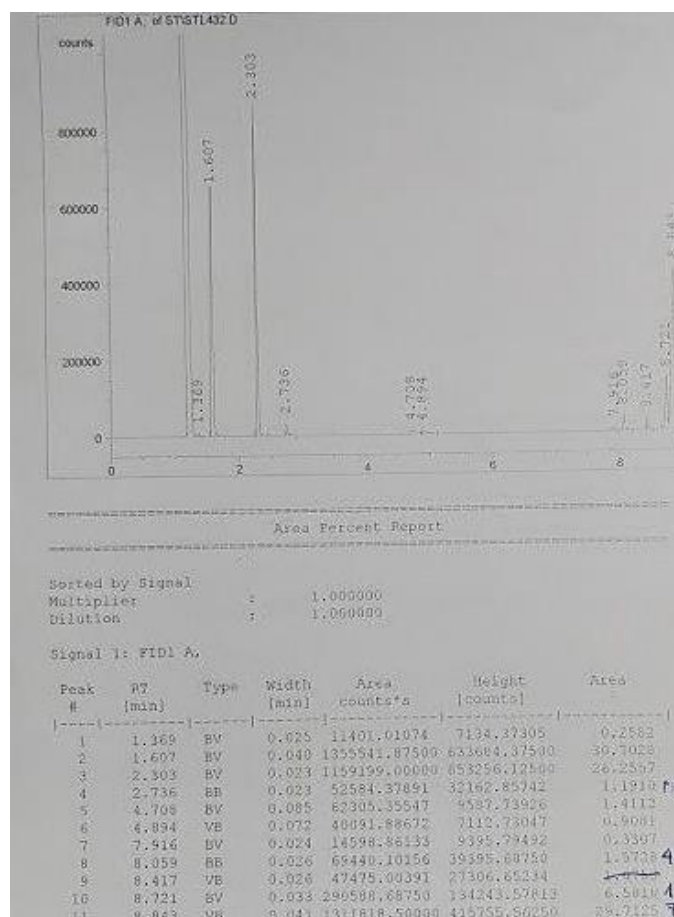


Figure S17. Mass spectrum of (3,7-dimethyl-2-phenylocta-1,6-dien-3-ol) (**9**).

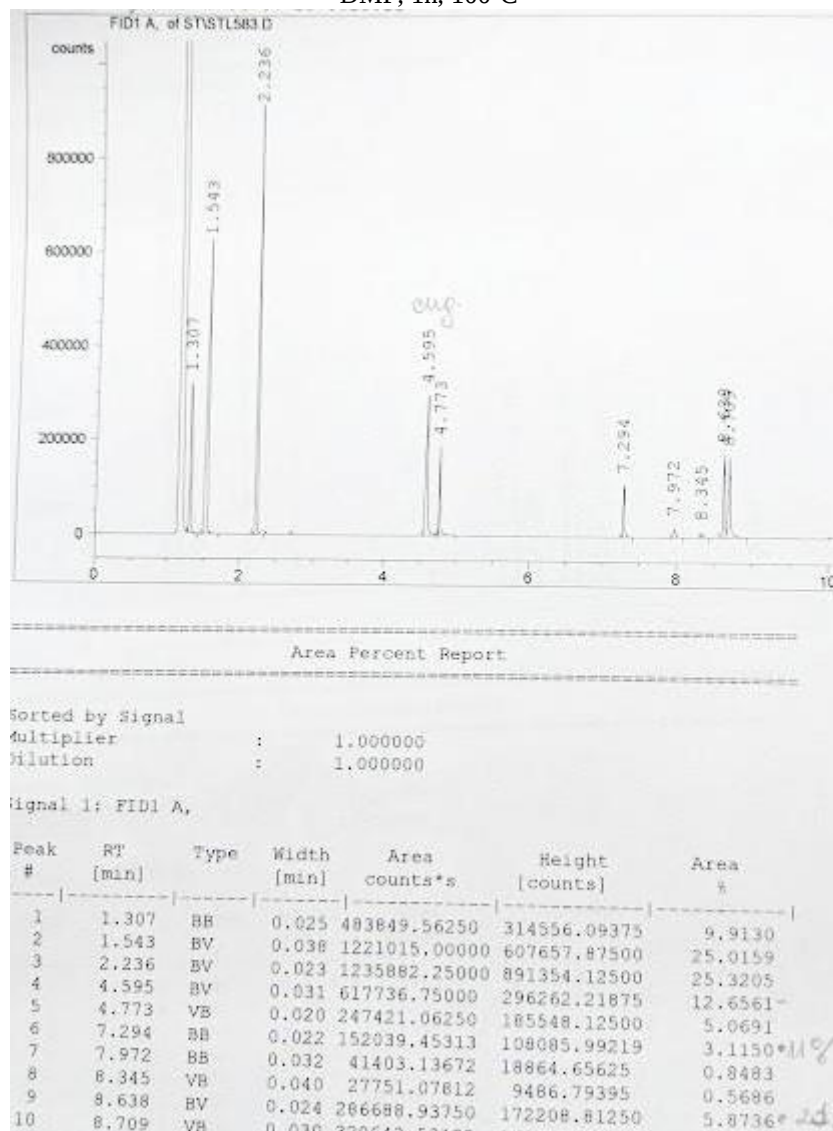
GC examples (preparation of samples for GC was described in the Experimental part)



Peak [min]	Compound
1.607	DMF
2.303	mesitylene
2.736	PhI

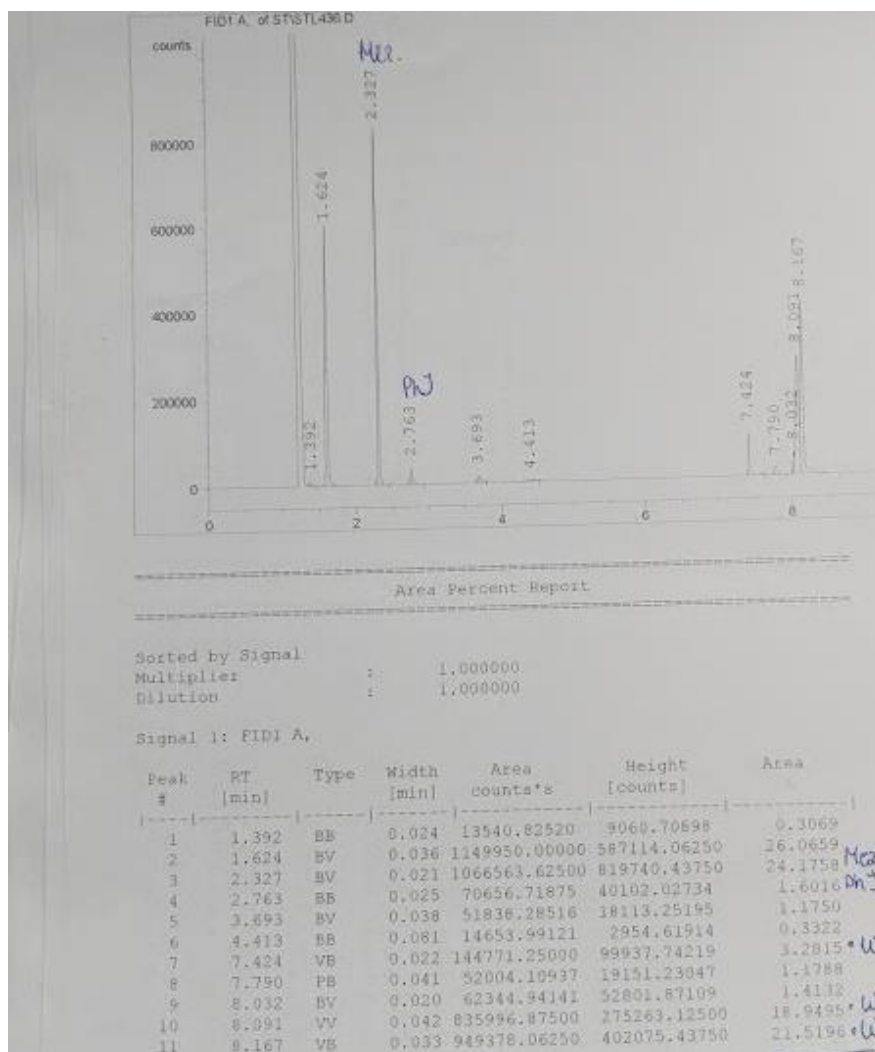
4.708	eugenol
4.894	biphenyl
8.059	1b
8.721	1Z
8.843	1E

Figure S18. GC profile for the Heck reaction of PhI with eugenol.
Reaction conditions (method A): PhI (1 mmol), eugenol (1 mmol), K₂CO₃ (2 mmol), Pd(OAc)₂ (1×10⁻⁵ mol), DMF, 1h, 100°C



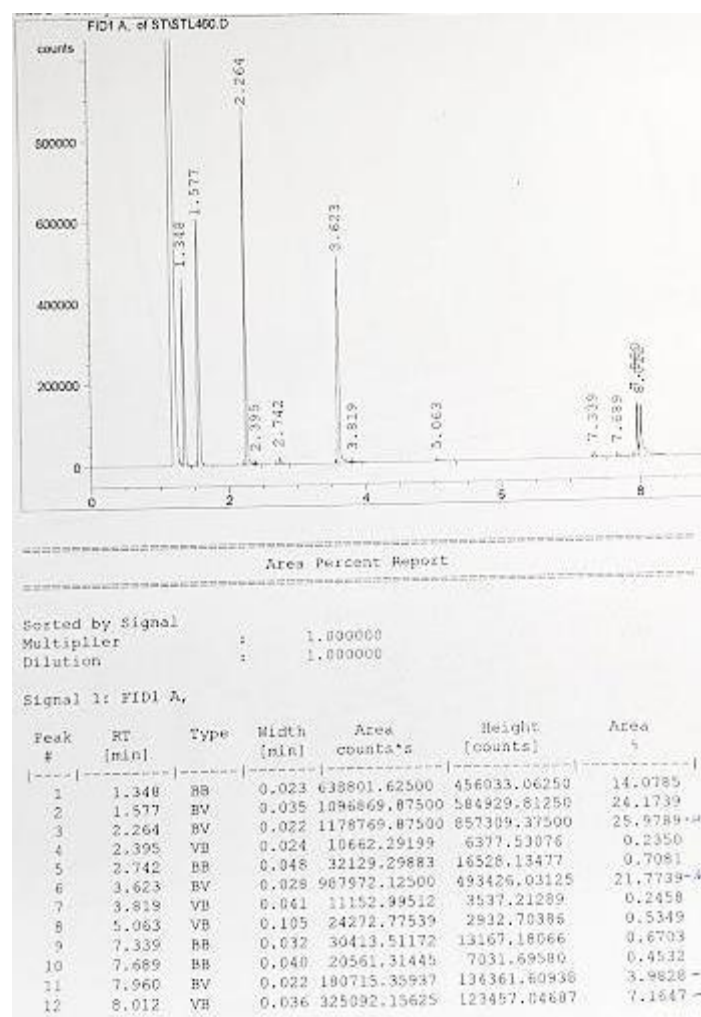
Peak [min]	Compound
1.543	DMF
2.236	mesitylene
4.595	eugenol
4.773	biphenyl
7.294	1b
8.638	1Z
8.709	1E

Figure S19. GC profile for the Heck-type reaction of eugenol with PhB(OH)₂.
Reaction conditions (method B): eugenol (1 mmol), phenylboronic acid (1.5 mmol), Cu²⁺ salt (2 mmol), Pd(OAc)₂ (1×10⁻⁶ mol), DMF (5 cm³), 6h, 100°C



Peak [min]	Coumpound
1.624	DMF
2.327	mesitylene
2.763	PhI
3.693	estragole
7.424	2b
8.091	2Z
8.167	2E

Figure S20. GC profile for the Heck reaction of PhI with estragole.
Reaction conditions (method A): PhI (1 mmol), estragole (1 mmol), K₂CO₃ (2 mmol), Pd(OAc)₂ (1×10⁻⁵mol), DMF (5 cm³), 3h, 100°C



Peak [min]	Compound
1.577	DMF
2.264	mesitylene
3.623	estragole
7.339	2b
7.960	2Z
8.012	2E

Figure S21. GC profile for the Heck-type reaction of estragole with PhB(OH)₂. Reaction conditions (method B): estragole (1 mmol), phenylboronic acid (1.5 mmol), Cu²⁺ salt (2 mmol), PdCl₂cod (1×10⁻⁵ mol), DMF (5 cm³), 4h, 100°C

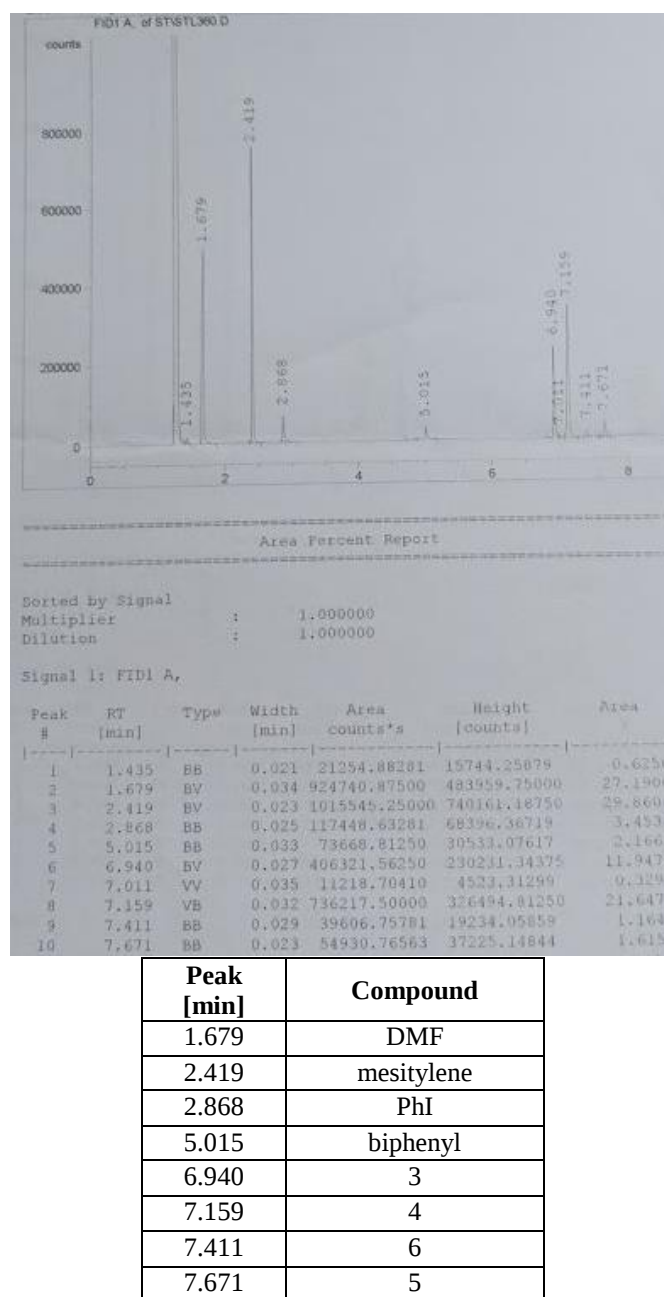
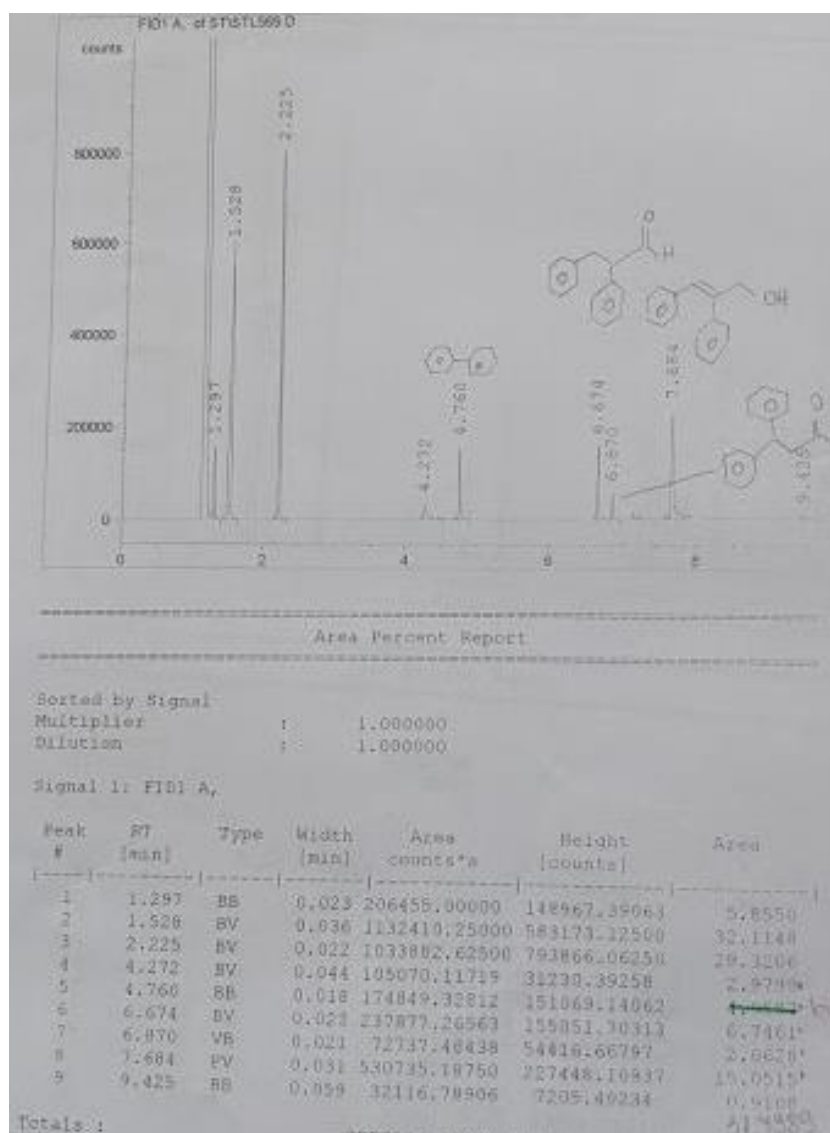


Figure S22. GC profile for the Heck reaction of cinnamyl alcohol with PhI. Reaction conditions (method C): PhI (1 mmol), cinnamyl alcohol (1 mmol), NaOAc (2 mmol), PdCl₂cod (1×10⁻⁵ mol), DMF (5 cm³), TBAB (1g), 6h, 100 °C



Peak [min]	Compound
1.528	DMF
2.225	mesitylene
4.272	cinnamyl alcohol
4.760	biphenyl
6.674	4
6.870	3
7.784	7

Figure S23. GC profile for the Heck-type reaction of cinnamyl alcohol with PhB(OH)_2 . Reaction conditions (method B): cinnamyl alcohol (1 mmol), phenylboronic acid (1.5 mmol), Cu^{2+} salt (2 mmol), Pd_2dba_3 (1×10^{-5} mol), DMF (5 cm^3), 0.5h, 50°C

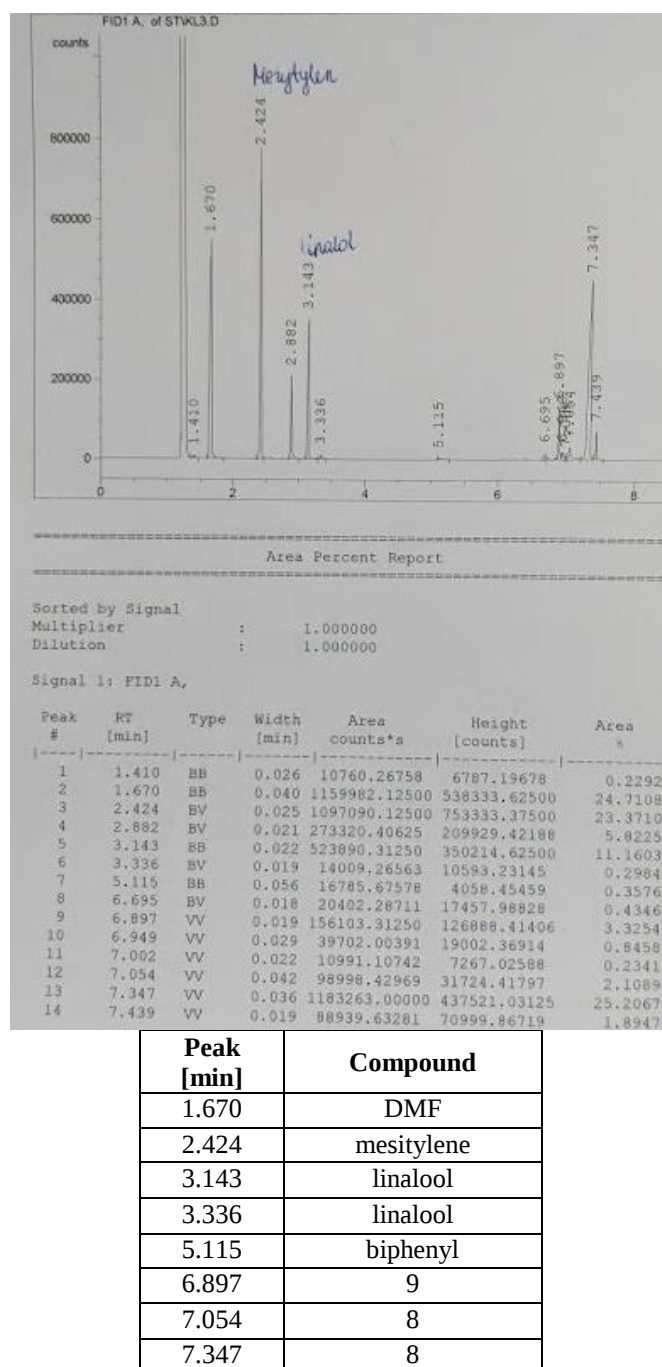


Figure S24. GC profile for the Heck reaction of linalool with PhI.

Reaction conditions (method C): PhI (1 mmol), linalool (1 mmol), NaOAc (2 mmol), PdCl₂cod (1×10⁻⁵ mol), DMF (5 cm³), 3h, 140°C

References

- Wang, J.; Huang, W.; Zhang, Z.; Xiang, X.; Liu, R.; Zhou X. FeCl₃·6H₂O Catalyzed Disproportionation of Allylic Alcohols and Selective Allylic Reduction of Allylic Alcohols and Their Derivatives with Benzyl Alcohol. *J. Org. Chem.* **2009**, 74, 3299–3304, DOI: 10.1021/jo900070q
- Ortar, G. Palladium-catalyzed cross-coupling reaction of allyl acetates with pinacol aryl- and vinylboronates. *Tetrahedron Lett.* **2003**, 44, 4311–4314, DOI: 10.1016/S0040-4039(03)00980-8
- Lerebours, R.; Wolf, C. Palladium(II)-Catalyzed Conjugate Addition of Arylsiloxanes in Water. *Org. Lett.* **2007**, 9(14), 2737–2740, DOI: 10.1021/ol071067v

4. Calo, V.; Nacci, A.; Monopoli, A.; Cotugno, P. Palladium-Nanoparticle-Catalysed Ullmann Reactions in Ionic Liquids with Aldehydes as the Reductants: Scope and Mechanism. *Chem. Eur. J.* **2009**, 15, 1272–1279, DOI: 10.1002/chem.200801621
5. Calo, V.; Nacci, A.; Monopoli, A.; Ferola, V. Palladium-Catalyzed Heck Arylations of Allyl Alcohols in Ionic Liquids: Remarkable Base Effect on the Selectivity. *J. Org. Chem.* **2007**, 72, 2596–2601, DOI: 10.1021/jo070005f
6. Cadierno, V.; Garcia-Garrido, S. E.; Gimeno, J. Isomerization of Propargylic Alcohols into α,β -Unsaturated Carbonyl Compounds Catalyzed by the Sixteen-Electron Allyl-Ruthenium(II) Complex $[\text{Ru}(\text{h}^3\text{-2-C}_3\text{H}_4\text{Me})(\text{CO})(\text{dppf})][\text{SbF}_6]$. *Adv. Synth. Catal.* **2006**, 348, 101–110, DOI: 10.1002/adsc.200505294