





# Reaction of Aldoximes with Sodium Chloride and Oxone under Ball-Milling Conditions

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#### 1. <sup>1</sup>H and <sup>13</sup>C NMR spectra of 2 and 3a



Figure S1. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) spectrum of compound 2a.



Figure S2. <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) spectrum of compound 2a.





**Figure S4.** <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) spectrum of compound 2b.





Figure S6. <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) spectrum of compound 2c.



Figure S8. <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) spectrum of compound 2d.





Figure S10. <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) spectrum of compound 2e.

10



0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 -210 Figure S11. <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>) spectrum of compound 2e.







-54 -55 -56 -57 -58 -59 -60 -61 -62 -63 -64 -65 -66 -67 -68 -69 -70 -71 -72 -73 -74 -75 **Figure S14.** <sup>19</sup>F NMR (471 MHz, CDCl<sub>3</sub>) spectrum of compound 2f.









110 100 -1 Figure S20. <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) spectrum of compound 2i.





10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 -210 Figure S22. <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>) spectrum of compound 2i.





Figure S24. <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) spectrum of compound 2j.





Figure S26. <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) spectrum of compound 2k.







Figure S28. <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) spectrum of compound 2l.





Figure S30. <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) spectrum of compound 2m.





Figure S32. <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) spectrum of compound 2n.







Figure S34. 1H NMR (500 MHz, CDCl3) spectrum of compound 20.



Figure S36. 1H NMR (500 MHz, CDCl3) spectrum of compound 3a.



### 2. Single-Crystal X-ray Crystallography of 2a

Single crystals of **2a** were obtained by slow evaporation from a mixture of dichloromethane/*n*-hexane at 4 °C. Single-crystal X-ray diffraction data were collected on a diffractometer (Gemini S Ultra, Agilent Technologies) equipped with a CCD area detector using graphite-monochromated Cu K $\alpha$  radiation ( $\lambda$  = 1.54184 Å) in the scan range 9.254° < 2 $\theta$  < 146.750°. The structure was solved with direct methods using SHELXS-97 and refined with full-matrix least-squares refinement using the SHELXL-97 program within OLEX2. Crystallographic data have been deposited in the Cambridge Crystallographic Data Centre as deposition number CCDC 2003914.



Figure S38. ORTEP Diagrams of 2a with 30% Thermal Ellipsoids.

Identification code	2003914
Empirical formula	C <sub>16</sub> H <sub>14</sub> ClNO <sub>2</sub>
Formula weight	287.73
Temperature/K	293(2)
Crystal system	monoclinic
Space group	P21/c
a/Å	7.3283(3)
b/Å	17.5340(8)
c/Å	11.3990(5)
α/°	90
β/°	90.530(4)
γ/°	90
Volume/Å <sup>3</sup>	1464.65(11)
Ζ	4
$\rho_{calc}g/cm^3$	1.305
$\mu/mm^{-1}$	2.313
F(000)	600.0
Crystal size/mm <sup>3</sup>	$0.250 \times 0.220 \times 0.150$
Radiation	$CuK\alpha$ ( $\lambda = 1.54184$ )
20 range for data collection/°	9.254 to 146.750
Index ranges	$-7 \le h \le 8, -21 \le k \le 21, -14 \le l \le 11$
Reflections collected	5918
Independent reflections	2866 [ $R_{int} = 0.0196$ , $R_{sigma} = 0.0209$ ]
Data/restraints/parameters	2866/0/183
Goodness-of-fit on F <sup>2</sup>	1.059
Final R indexes $[I \ge 2\sigma(I)]$	$R_1 = 0.0496, wR_2 = 0.1441$
Final R indexes [all data]	$R_1 = 0.0568, wR_2 = 0.1551$
Largest diff. peak/hole/e Å <sup>-3</sup>	0.33/-0.36

Table S1. Crystal Data and Structure Refinement for 2a.