

Supplementary Material

Complexes of formaldehyde and α -dicarbonyls with hydroxylamine: FTIR matrix isolation and theoretical study

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This file contains:

Figure S1. The optimized structures of the HCHO-NH₂OH complexes. The $\Delta E^{\text{CP}}(\text{ZPE})$ binding energies in kJ mol⁻¹ are given in parentheses. The intermolecular distances are given in Å.

Figure S2. The optimized structures of the CHOCHO-NH₂OH complexes. The $\Delta E^{\text{CP}}(\text{ZPE})$ binding energies in kJ mol⁻¹ are given in parentheses. The intermolecular distances are given in Å.

Figure S3. The spectra of the CHOCHO/Ar (a), ND₂OD/Ar (b) and CHOCHO/ND₂OD/Ar (c) matrices recorded after matrix deposition at 11 K. The bands of CHOCHO-ND₂OD complexes are indicated by the arrows.

Figure S4. The spectra of the CHOCHO/N₂ (a), ND₂OD/N₂ (b) and CHOCHO/ND₂OD/N₂ (c) matrices recorded after matrix deposition at 11 K. The bands of CHOCHO-ND₂OD complexes are indicated by the arrows.

Figure S5. The optimized structures of the CH₃COCHO-NH₂OH complexes. The $\Delta E^{\text{CP}}(\text{ZPE})$ binding energies in kJ mol⁻¹ are given in parentheses. The intermolecular distances are given in Å.

Figure S6. The spectra of the CH₃COCHO/Ar (a), ND₂OD/Ar (b) and CH₃COCHO/ND₂OD/Ar (c) matrices recorded after matrix deposition at 11 K. The bands of CH₃COCHO-ND₂OD complexes are indicated by the arrows.

Figure S7. The location of the bond (3,-1) and ring (3,1) critical points in the MP2/6-311++G(2d,2p) optimized structures of the HCHO-NH₂OH complexes.

Figure S8. The location of the bond (3,-1) and ring (3,1) critical points in the MP2/6-311++G(2d,2p) optimized structures of the CHOCHO -NH₂OH complexes.

Figure S9. The MP2 optimized structures of the formaldehyde, glyoxal and methylglyoxal complexes with hydroxylamine assigned to the structures isolated in argon matrix. The intermolecular distances are given in Å. The binding energies in kJ mol⁻¹ are given in parentheses.

Figure S10. The MP2 optimized structures of the formaldehyde, glyoxal and methylglyoxal complexes with hydroxylamine assigned to the structures isolated in nitrogen matrix. The intermolecular distances are given in Å. The binding energies in kJ mol⁻¹ are given in parentheses.

Table S1. Selected geometrical parameters of the hydroxylamine and formaldehyde subunits in their binary complexes. For comparison the corresponding parameters of the monomers (M) are also given. The complexes are numbered in the same way as presented in Fig. S1. Bond distances are given in Å, angles in °.

Table S2. Selected geometrical parameters of the hydroxylamine and glyoxal subunits in their binary complexes. For comparison the corresponding parameters of the monomers (M) are also given. The complexes are numbered in the same way as presented in Fig. S2. Bond distances are given in Å, angles in °.

Table S3. The comparison of the observed wavenumbers (cm^{-1}) and wavenumber shifts ($\Delta\nu = \nu_{\text{GH}} - \nu_{\text{M}}$) for the CHOCHO-NH₂OH (GH) complexes present in the Ar and N₂ matrices with the corresponding calculated values for the complexes **I_{GH}** - **IV_{GH}**.

Table S4. The comparison of the observed wavenumbers (cm^{-1}) and wavenumber shifts ($\Delta\nu = \nu_{\text{GH}} - \nu_{\text{M}}$) for the CHOCHO-ND₂OD (GH) complexes present in the Ar and N₂ matrices with the corresponding calculated values for the complexes **I_{GH}** - **IV_{GH}**.

Table S5. Selected geometrical parameters of the hydroxylamine and methylglyoxal subunits in their binary complexes. For comparison the corresponding parameters of the monomers (M) are also given. The complexes are numbered in the same way as presented in Fig. S3. Bond distances are given in Å, angles in °.

Table S6. The comparison of the observed wavenumbers (cm^{-1}) and wavenumber shifts ($\Delta\nu = \nu_{\text{MH}} - \nu_{\text{M}}$) for the CH₃COCHO-NH₂OH (MH) complexes present in the Ar and N₂ matrices with the corresponding calculated values for the complexes **I_{MHk}** - **IV_{MHk}** and **I_{MHa}** - **IV_{MHa}**.

Table S7. The comparison of the observed wavenumbers (cm^{-1}) and wavenumber shifts ($\Delta\nu = \nu_{\text{MH}} - \nu_{\text{M}}$) for the CH₃COCHO-ND₂OD (MH) complexes present in the Ar and N₂ matrices with the corresponding calculated values for the complexes **I_{MHk}** - **IV_{MHk}** and **I_{MHa}** - **IV_{MHa}**.

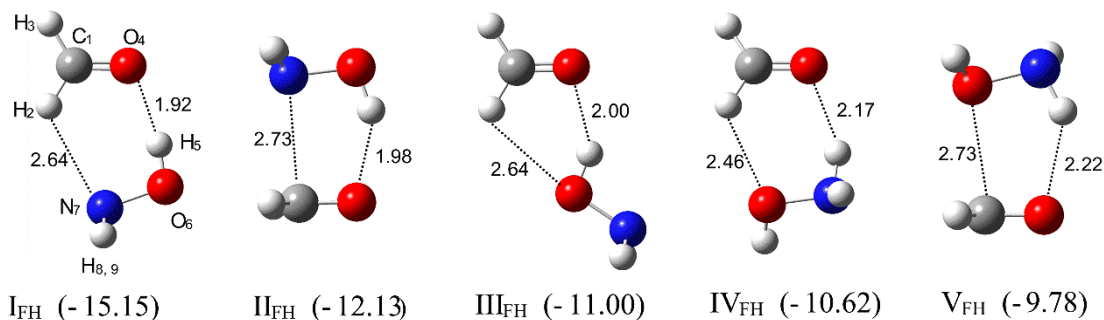


Figure S1. The optimized structures of the HCHO-NH₂OH complexes. The $\Delta E^{\text{CP}}(\text{ZPE})$ binding energies in kJ mol⁻¹ are given in parentheses. The intermolecular distances are given in Å.

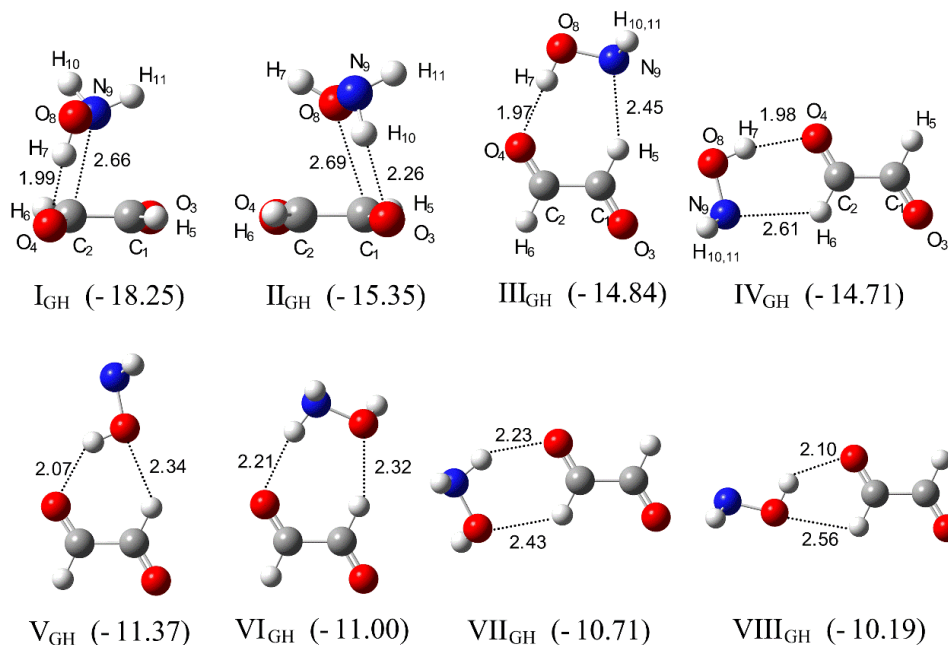


Figure S2. The optimized structures of the CHOCHO-NH₂OH complexes. The $\Delta E^{\text{CP}}(\text{ZPE})$ binding energies in kJ mol⁻¹ are given in parentheses. The intermolecular distances are given in Å.

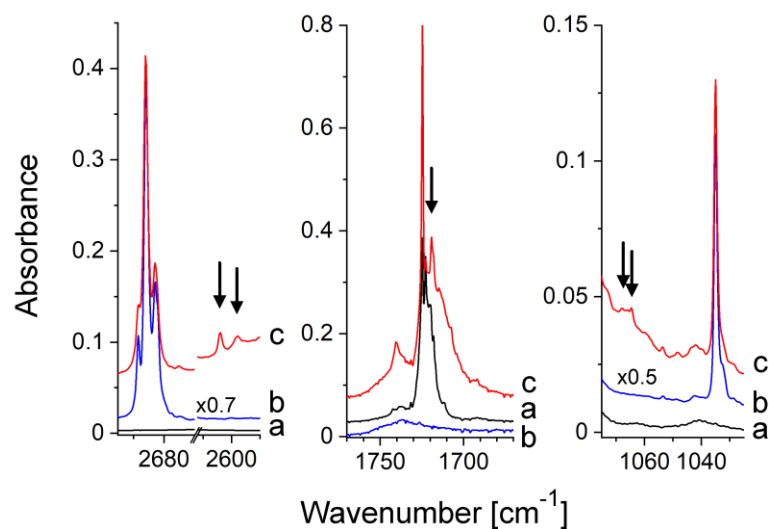


Figure S3. The spectra of the CHOCHO/Ar (a), ND₂OD/Ar (b) and CHOCHO/ND₂OD/Ar (c) matrices recorded after matrix deposition at 11 K. The bands of CHOCHO-ND₂OD complexes are indicated by the arrows..

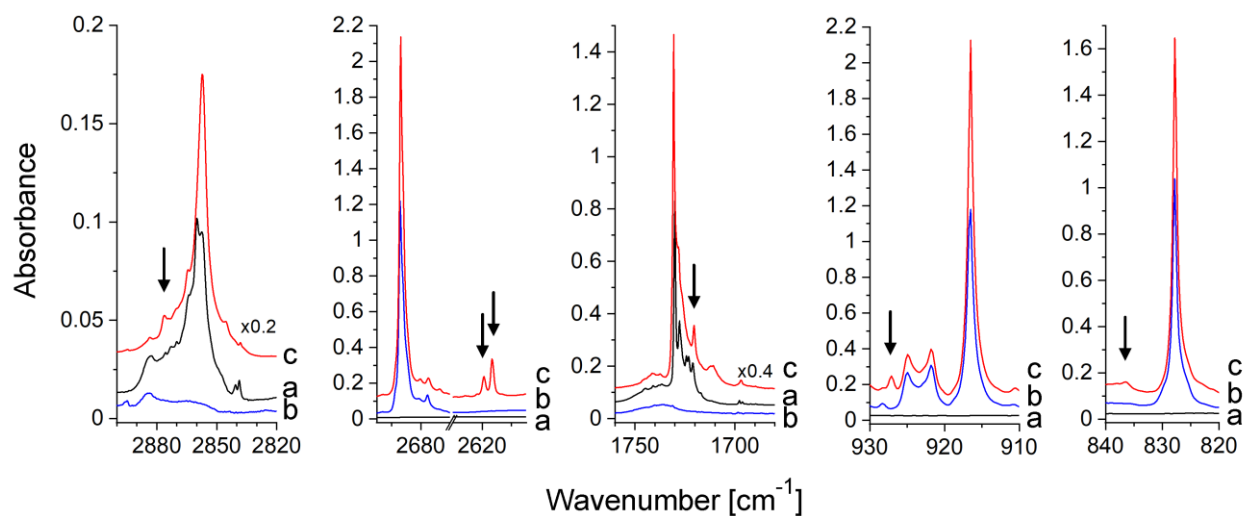


Figure S4. The spectra of the CHOCHO/N₂ (a), ND₂OD/N₂ (b) and CHOCHO/ND₂OD/N₂ (c) matrices recorded after matrix deposition at 11 K. The bands of CHOCHO-ND₂OD complexes are indicated by the arrows..

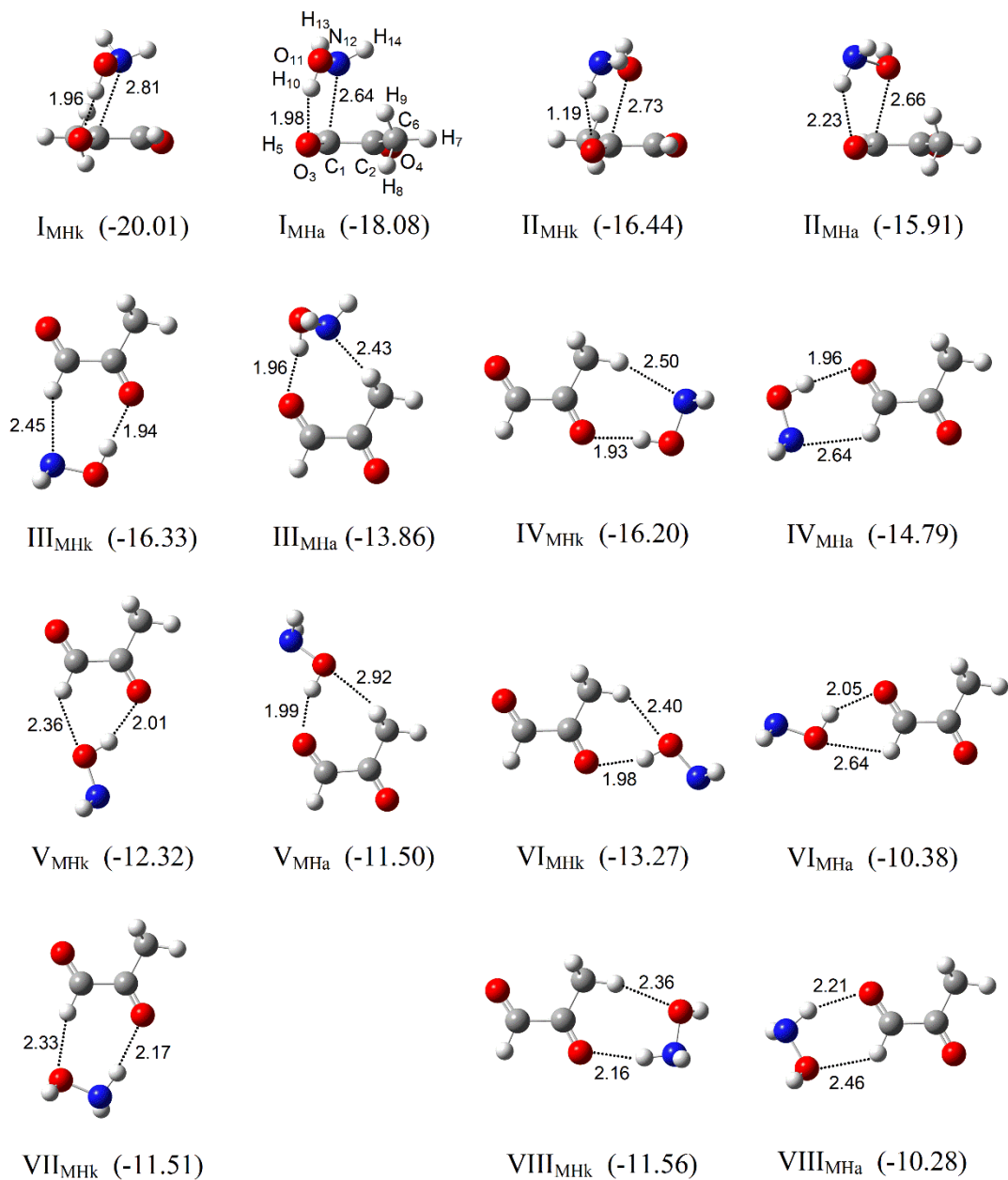


Figure S5. The optimized structures of the $\text{CH}_3\text{COCHO-NH}_2\text{OH}$ complexes. The $\Delta E^{\text{CP}}(\text{ZPE})$ binding energies in kJ mol^{-1} are given in parentheses. The intermolecular distances are given in Å.

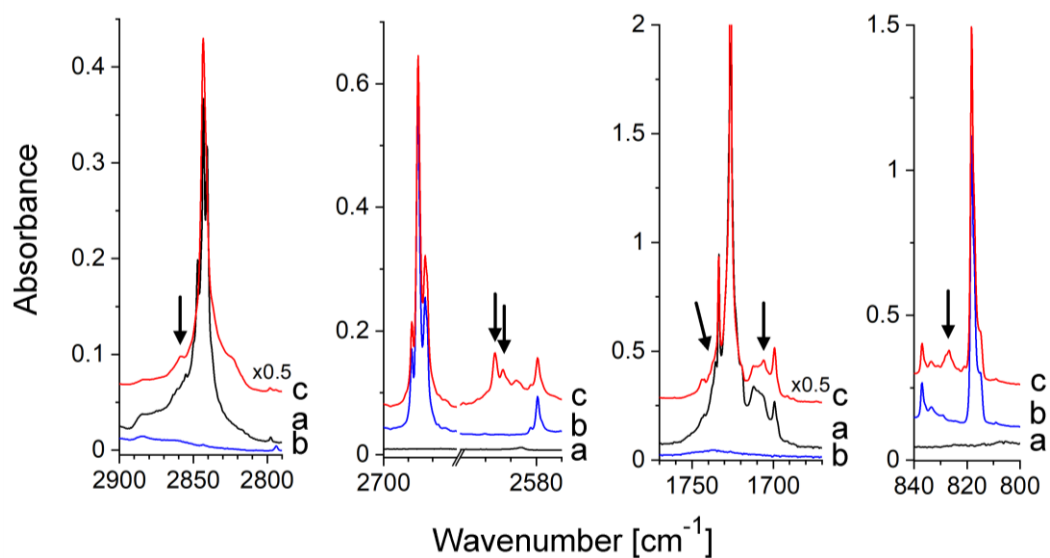


Figure S6. The spectra of the CH₃COCHO/Ar (a), ND₂OD/Ar (b) and CH₃COCHO/ND₂OD/Ar (c) matrices recorded after matrix deposition at 11 K. The bands of CH₃COCHO-ND₂OD complexes are indicated by the arrows.

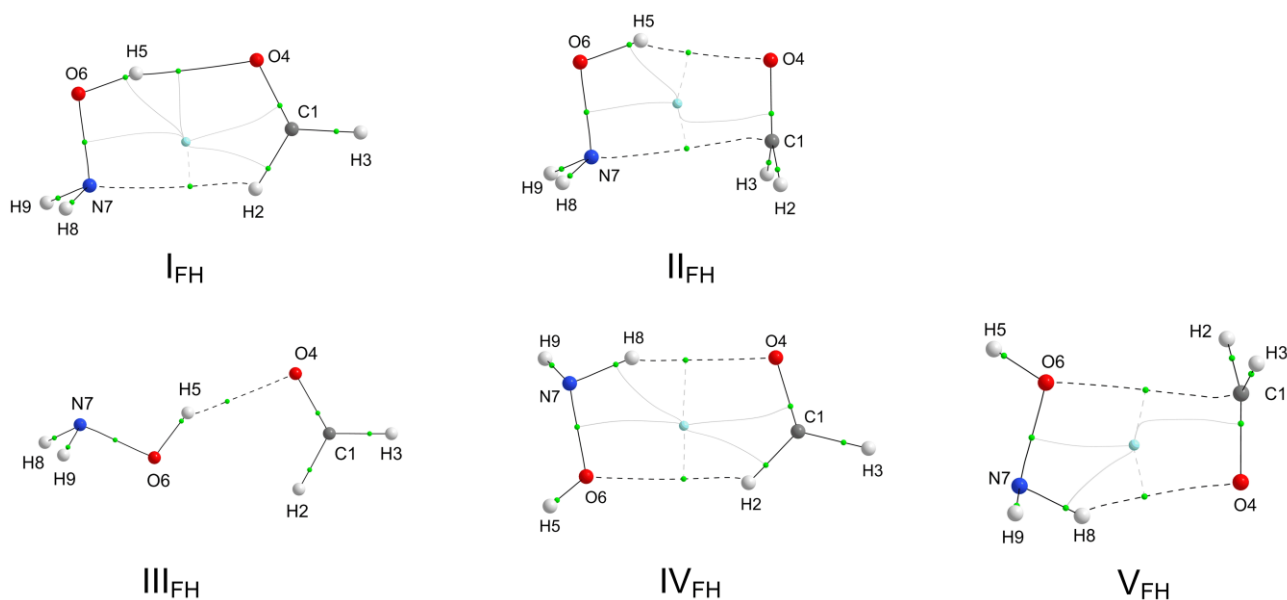


Figure S7. The location of the bond (3,-1) and ring (3,1) critical points in the MP2/6-311++G(2d,2p) optimized structures of the HCHO-NH₂OH complexes.

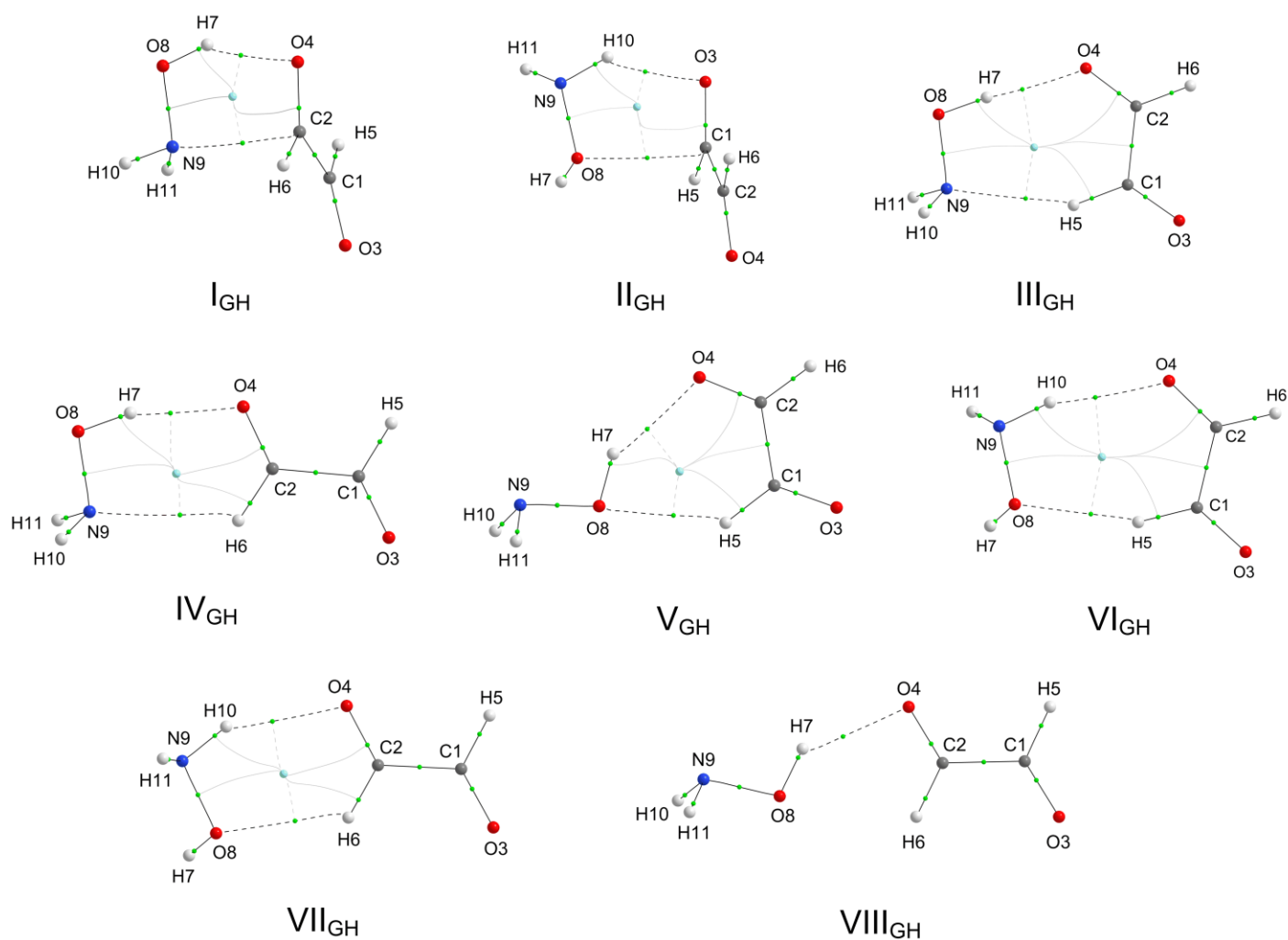


Figure S8. The location of the bond (3,-1) and ring (3,1) critical points in the MP2/6-311++G(2d,2p) optimized structures of the CHOCHO-NH₂OH complexes.

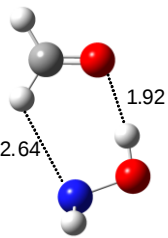
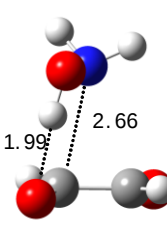
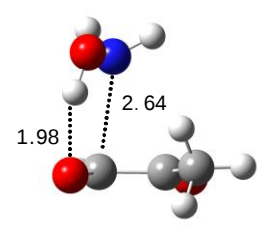
<i>Argon matrix</i>		
FA-HA	Gly-HA	MGly-HA
		
I_{FH} (-15.15)	I_{GH} (-18.25)	I_{MHa} (-18.08)

Figure S9. The MP2 optimized structures of the formaldehyde, glyoxal and methylglyoxal complexes with hydroxylamine assigned to the structures isolated in argon matrix. The intermolecular distances are given in Å. The binding energies in kJ mol⁻¹ are given in parentheses.

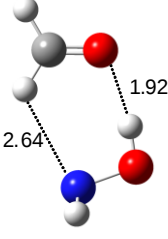
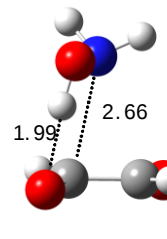
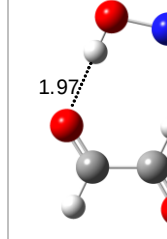
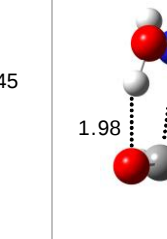
<i>Nitrogen matrix</i>			
FA-HA	Gly-HA		MGly-HA
			
I_{FH} (-15.15)	I_{GH} (-18.25)	III_{GH} (-14.84)	I_{MHa} (-18.08)

Figure S10. The MP2 optimized structures of the formaldehyde, glyoxal and methylglyoxal complexes with hydroxylamine assigned to the structures isolated in nitrogen matrix. The intermolecular distances are given in Å. The binding energies in kJ mol⁻¹ are given in parentheses.

Table S1. Selected geometrical parameters of the hydroxylamine and formaldehyde subunits in their binary complexes. For comparison the corresponding parameters of the monomers (M) are also given. The complexes are numbered in the same way as presented in Fig. S1. Bond distances are given in Å, angles in °.

<i>Parameter</i>	M	I _{FH}	II _{FH}	III _{FH}	IV _{FH}	V _{FH}
r C ₁ -O ₄	1.213	1.219	1.218	1.217	1.218	1.217
r C ₁ -H ₂	1.098	1.094	1.096	1.096	1.095	1.096
r C ₁ -H ₃	1.098	1.097	1.096	1.096	1.098	1.097
r O ₆ -H ₅	0.959	0.967	0.967	0.965	0.959	0.960
r O ₆ -N ₇	1.448	1.447	1.439	1.446	1.455	1.451
r N ₇ -H ₈	1.012	1.013	1.013	1.012	1.015	1.015
r N ₇ -H ₉	1.012	1.013	1.013	1.012	1.012	1.012
R O ₄ ...H ₅		1.928	1.985	1.997		
R H ₂ ...N ₇		2.640				
R H ₂ ...O ₆				2.637	2.461	
R H ₈ ...O ₄					2.169	2.218
θ H ₂ -C ₁ -O ₄	121.6	120.9	121.6	121.3	121.1	121.7
θ H ₃ -C ₁ -O ₄	121.6	120.6	121.6	121.1	120.9	121.6
θ H ₂ -C ₁ -H ₃	116.7	118.4	116.7	117.6	118.0	116.7
θ H ₅ -O ₆ -N ₇	101.7	101.5	102.4	102.5	101.6	101.6
θ O ₆ -N ₇ -H ₈	103.6	103.8	104.1	103.8	103.0	103.4
θ O ₆ -N ₇ -H ₉	103.6	103.8	104.1	103.8	103.5	103.4
θ H ₈ -N ₇ -H ₉	105.8	105.2	105.9	105.5	105.9	105.5
θ O ₄ -H ₅ -O ₆		166.9	151.7	146.9		
θ C ₁ -H ₂ -N ₇		120.0				
θ C ₁ -H ₂ -O ₆					128.7	
θ O ₄ -H ₈ -N ₇					156.7	141.3
φ H ₅ -O ₆ -N ₇ -H ₈	124.8	125.1	124.6	124.9	128.2	125.5
φ H ₅ -O ₆ -N ₇ -H ₉	-124.8	-125.1	-124.6	-124.9	-121.6	-124.6
φ C ₁ -O ₄ -H ₅ -O ₆		-0.0	0.4	0.1		
φ C ₁ -H ₂ -N ₇ -O ₆		-0.0				
φ C ₁ -O ₄ -H ₈ -N ₇					0.0	-13.9
φ C ₁ -H ₂ -O ₆ -N ₇					9.8	

Table S2. Selected geometrical parameters of the hydroxylamine and glyoxal subunits in their binary complexes. For comparison the corresponding parameters of the monomers (M) are also given. The complexes are numbered in the same way as presented in Fig. S2. Bond distances are given in Å, angles in °.

Parameter	M	I _{GH}	II _{GH}	III _{GH}	IV _{GH}	V _{GH}	VI _{GH}	VII _{GH}	VIII _{GH}
r C ₁ -C ₂	1.518	1.514	1.515	1.519	1.517	1.520	1.519	1.517	1.517
r C ₁ -O ₃	1.215	1.215	1.217	1.218	1.215	1.217	1.217	1.215	1.215
r C ₂ -O ₄	1.215	1.220	1.215	1.219	1.220	1.217	1.217	1.219	1.219
r C ₁ -H ₅	1.100	1.100	1.098	1.098	1.100	1.097	1.098	1.100	1.100
r C ₂ -H ₆	1.100	1.099	1.100	1.099	1.096	1.099	1.100	1.097	1.098
r O ₈ -H ₇	0.959	0.968	0.961	0.965	0.965	0.964	0.959	0.959	0.963
r O ₈ -N ₉	1.448	1.436	1.451	1.446	1.447	1.447	1.453	1.454	1.447
r N ₉ -H ₁₀	1.012	1.013	1.015	1.013	1.013	1.013	1.014	1.014	1.012
r N ₉ -H ₁₁	1.012	1.013	1.013	1.013	1.013	1.013	1.013	1.013	1.012
R O ₄ ...H ₇		1.988		1.970	1.980	2.070	2.209		2.096
R O ₃ ...H ₁₀			2.264						
R H ₅ ...N ₉				2.458	2.609				
R O ₈ ...H ₅						2.337	2.321		
R O ₄ ...H ₁₀								2.227	
R O ₈ ...H ₆								2.428	2.559
θ C ₂ -C ₁ -O ₃	121.3	121.7	121.3	119.9	121.3	119.7	119.9	121.5	121.3
θ C ₁ -C ₂ -O ₄	121.3	121.6	121.7	122.0	120.3	122.4	122.0	120.4	120.7
θ C ₂ -C ₁ -H ₅	115.4	115.0	115.5	115.9	115.4	116.5	116.1	115.3	115.3
θ C ₁ -C ₂ -H ₆	115.4	115.3	115.0	115.6	117.3	115.0	115.3	117.0	116.4
θ O ₃ -C ₁ -H ₅	123.3	123.3	123.1	124.3	123.3	123.7	124.0	123.2	123.3
θ O ₄ -C ₂ -H ₆	123.3	123.0	123.3	122.4	122.4	122.6	122.7	122.6	122.8
θ H ₇ -O ₈ -N ₉	101.7	102.5	102.0	102.2	101.5	102.9	129.1	101.7	102.6
θ O ₈ -N ₉ -H ₁₀	103.6	104.8	103.2	103.8	103.8	103.8	103.2	102.8	103.8
θ O ₈ -N ₉ -H ₁₁	103.6	104.4	103.4	103.8	103.8	103.8	103.2	103.5	103.8
θ H ₁₀ -N ₉ -H ₁₁	105.8	106.2	105.3	105.4	105.3	105.6	105.9	105.9	105.6
θ C ₂ -O ₄ -H ₇		100.8	100.7	126.6	112.0	117.1	125.8	100.3	
θ C ₁ -H ₅ -N ₉			112.1	151.5					
θ C ₂ -O ₄ -N ₉		93.6							
θ O ₄ -H ₇ -O ₈		148.2		176.9	163.3	144.20	164.9	136.6	
θ O ₃ -H ₉ -N ₉			135.5						
θ H ₅ -N ₉ -O ₈				104.9					
θ H ₅ -N ₉ -H ₁₀				118.5					
θ H ₅ -N ₉ -H ₁₁				118.5					
θ C ₂ -H ₆ -N ₉					118.9				
θ C ₁ -H ₅ -O ₈						135.2	156.3		
θ O ₄ -H ₁₀ -N ₉								151.6	
θ C ₂ -H ₆ -O ₈									106.7
φ O ₃ -C ₁ -C ₂ -O ₄	-180.0	166.0	-171.5	-180.0	180.0	-180.0	179.6	-180.0	180.0
φ O ₃ -C ₁ -C ₂ -H ₆	0.0	-10.4	7.0	0.0	0.0	0.0	-0.4	0.0	0.0
φ H ₅ -C ₁ -C ₂ -O ₄	0.0	-11.7	7.1	0.0	0.0	0.0	-0.6	0.0	0.0
φ H ₇ -O ₈ -N ₉ -H ₁₀	124.8		130.1	125.0	125.1	124.9	128.4	-128.7	-124.9
φ H ₇ -O ₈ -N ₉ -H ₁₁	-124.8		-120.3	-125.0	-125.1	-124.9	-121.3	121.0	124.8
φ C ₂ -O ₄ -H ₇ -O ₈		10.9		0.0	-0.1	0.4	-1.1		-0.3
φ C ₁ -O ₃ -H ₁₀ -N ₉			-26.0						
φ C ₁ -O ₃ -N ₉ -O ₈			-3.5						
φ C ₁ -H ₅ -N ₉ -O ₈				-0.1					
φ C ₂ -H ₆ -N ₉ -O ₈					-0.1				
φ C ₂ -O ₄ -O ₈ -N ₉		2.6							
φ C ₁ -H ₅ -O ₈ -H ₇						-0.1			
φ C ₁ -H ₅ -O ₈ -N ₉							18.3		
φ C ₂ -H ₆ -O ₈ -N ₉								-8.1	-12.6
φ C ₂ -O ₄ -H ₁₀ -N ₉								-1.7	

Table S3. The comparison of the observed wavenumbers (cm⁻¹) and wavenumber shifts ($\Delta\nu = \nu_{\text{GH}} - \nu_{\text{M}}$) for the CHOCHO-NH₂OH (GH) complexes present in the Ar and N₂ matrices with the corresponding calculated values for the complexes **I_{GH}** - **IV_{GH}**.

Approximate description	Experimental						Calculated			
	Ar			N ₂			$\Delta\nu$			
	ν M	ν GH	$\Delta\nu^1$	ν M	ν GH ²	$\Delta\nu^1$	I _{GH}	II _{GH}	III _{GH}	IV _{GH}
NH ₂ OH										
$\nu(\text{OH})$	3635.5	3521.0 3512.4	-118.8	3637.6	3541.1 3520.9 3515.8	-96.5 <i>-119.2</i>	-145(77)	-19(66)	-106(383)	-103(356)
$\delta(\text{NOH})$	1351.2	1412.1 1410.2	+60.9	1367.4	1416.6 1399.7	+49.2 +32.3	+61(63)	-12(13)	+72(29)	+71(34)
$\omega(\text{NH}_2)$	1118.3	1129.0 1125.8	+9.1	1133.0	1142.6	+9.6	+14(111)	+19(147)	+24(112)	+22(122)
$\nu(\text{NO})$	895.6			895.3	898.8	+3.5	+14(3)	-4(10)	+9(8)	+6(8)
CHOCHO										
$\nu(\text{CH})$	2860.1 2854.9	2857.6	-0.4	2857.1	2875.6	+18.5	-9(60) +12(39)	-6(64) +12(34)	+15(53) +20(1)	+1(52) +51(8)
$\nu(\text{C=O})$	1724.5	1719.0	-5.5	1730.1	1720.2 1723.1	-9.9 -7.0	-4(122) +4(23)	+3(115) +5(9)	-11(126) -6(20)	-13(89) -6(33)
$\gamma(\text{CH})$	812.1 807.8			807.4	820.8	+13.4	+37(0)	+19(1)	+23(8)	+16(2)

¹ In the case when the splitting of the band was observed the average of the two wavenumbers at which the two peaks appear was taken into account to calculate $\Delta\nu$ value.
² The wavenumbers in italic are due to complex of different structure (see text).

Table S4. The comparison of the observed wavenumbers (cm⁻¹) and wavenumber shifts ($\Delta\nu = \nu_{\text{GH}} - \nu_{\text{M}}$) for the CHOCHO-ND₂OD (GH) complexes present in the Ar and N₂ matrices with the corresponding calculated values for the complexes **I_{GH}** - **IV_{GH}**.

Approximate description	Experimental						Calculated			
	Ar			N ₂			$\Delta\nu$			
	ν M	ν GH	$\Delta\nu^1$	ν M	ν GH	$\Delta\nu^1$	I_{GH}	II_{GH}	III_{GH}	IV_{GH}
ND ₂ OD										
$\nu(\text{OD})$	2685.1	2604.0 2598.0	-84.1	2686.9	2619.2 2616.7	-68.9	-106(42)	-14(36)	-77(202)	-75(184)
$\delta(\text{NOD})$	1034.5	1068.0 1064.5	+33.5	1043.2			+38(6)	-7(3)	+38(7)	+41(7)
$\omega(\text{ND}_2)$	915.0			916.5	927.1	+10.6	+0(29)	+12(44)	+14(40)	+14(42)
$\nu(\text{NO})$	818.4			827.6	838.0 836.6	+9.7	+28(41)	-10(52)	+21(40)	+16(44)
CHOCHO										
$\nu(\text{CH})$	2860.1 2854.9			2857.1	2876.3	+19.2	-10(59) +12(36)	-6(64) +12(34)	+15(48) +20(1)	+1(51) +51(9)
$\nu(\text{C=O})$	1724.5	1719.6	-4.9	1730.1	1720.5	-9.6	-5(125) +4(25)	+3(127) +5(9)	-11(128) -6(20)	-13(93) -6(33)
$\gamma(\text{CH})$	812.1 807.8			807.4	820.9	+13.5	+41(5)	+19(2)	+22(4)	+16(3)

¹ In the case when the splitting of the band was observed the average of the two wavenumbers at which the two peaks appear was taken into account to calculate $\Delta\nu$ value.

Table S5. Selected geometrical parameters of the hydroxylamine and methylglyoxal subunits in their binary complexes. For comparison the corresponding parameters of the monomers (M) are also given. The complexes are numbered in the same way as presented in Fig. S3. Bond distances are given in Å, angles in °.

Parameter	M	I _{MHk}	I _{MHa}	II _{MHk}	II _{MHa}	III _{MHk}	III _{MHa}	IV _{MHk}	IV _{MHa}
r C ₁ -C ₂	1.528	1.524	1.524	1.524	1.524	1.528	1.528	1.527	1.526
r C ₁ -O ₃	1.215	1.215	1.221	1.215	1.217	1.217	1.217	1.214	1.220
r C ₂ -O ₄	1.221	1.225	1.221	1.223	1.221	1.225	1.222	1.225	1.097
r C ₁ -H ₅	1.101	1.101	1.099	1.100	1.100	1.099	1.100	1.100	1.097
r C ₂ -C ₆	1.499	1.495	1.499	1.497	1.497	1.496	1.494	1.492	1.498
r C ₆ -H ₇	1.083	1.083	1.084	1.084	1.083	1.083	1.083	1.084	1.083
r C ₆ -H ₈	1.088	1.088	1.088	1.088	1.087	1.088	1.090	1.089	1.088
r C ₆ -H ₉	1.088	1.087	1.088	1.087	1.088	1.088	1.089	1.088	1.088
r O ₁₁ -H ₁₀	0.959	0.968	0.968	0.960	0.961	0.966	0.965	0.967	0.965
r O ₁₁ -N ₁₂	1.448	1.440	1.437	1.452	1.450	1.445	1.446	1.445	1.447
r N ₁₂ -H ₁₃	1.012	1.013	1.013	1.016	1.016	1.013	1.013	1.013	1.013
r N ₁₂ -H ₁₄	1.012	1.013	1.013	1.012	1.013	1.013	1.013	1.013	1.013
R O ₄ -H ₁₀		1.956				1.936		1.930	
R O ₃ -H ₁₀			1.980				1.961		1.964
R O ₄ -H ₁₃				2.190					
R O ₃ -H ₁₃					2.236				
R N ₉ -H ₅						2.447			2.637
R N ₉ -H ₈							2.432		
R N ₉ -H ₇								2.501	
θ C ₂ -C ₁ -O ₃	122.8	122.8	122.8	122.6	123.2	121.5	123.5	122.7	121.8
θ C ₁ -C ₂ -O ₄	117.7	118.1	118.1	118.0	117.6	118.1	116.7	116.8	117.5
θ C ₂ -C ₁ -H ₅	114.1	114.0	114.2	114.1	113.9	121.5	114.2	114.2	116.0
θ O ₃ -C ₁ -H ₅	123.1	123.2	122.9	123.2	122.9	123.8	122.3	123.2	122.2
θ C ₁ -C ₂ -C ₆	117.3	117.3	117.1	117.2	117.4	117.7	117.6	118.0	117.5
θ O ₄ -C ₂ -C ₆	125.0	124.5	124.8	124.6	125.0	124.2	125.6	125.2	125.0
θ H ₇ -C ₆ -H ₈	110.5	110.5	110.1	110.2	110.7	110.7	111.0	111.0	110.5
θ H ₇ -C ₆ -H ₉	110.6	110.8	110.8	110.9	110.2	110.7	110.3	111.0	110.5
θ H ₈ -C ₆ -H ₉	106.6	106.7	106.7	106.7	107.2	106.4	106.6	106.2	106.6
θ H ₁₀ -O ₁₁ -N ₁₂	101.7	102.0	102.3	101.9	102.0	102.2	102.0	102.2	101.4
θ O ₁₁ -N ₁₂ -H ₁₃	103.6	104.5	104.9	103.1	103.4	103.9	103.3	103.8	103.8
θ O ₁₁ -N ₁₂ -H ₁₄	103.6	104.0	104.2	103.4	103.5	103.9	104.2	103.8	103.8
θ H ₁₃ -N ₁₂ -H ₁₄	105.8	105.8	106.2	105.4	105.7	105.4	105.3	105.3	105.3
θ O ₄ -H ₁₀ -O ₁₁		151.5				178.7		179.7	
θ O ₃ -H ₁₀ -O ₁₁			148.6				171.7		164.5
θ C ₁ -O ₃ -H ₁₀			99.9						112.2
θ O ₄ -H ₁₃ -N ₁₂				139.7					
θ O ₃ -H ₁₃ -N ₁₂					138.2				
θ C ₁ -H ₅ -N ₉						155.2			118.5
θ C ₆ -H ₈ -N ₉							153.38		
θ C ₆ -H ₇ -N ₉								156.6	
φ O ₃ -C ₁ -C ₂ -O ₄	180.0	169.3	169.1	170.2	172.3	-180.0	178.0	-180.0	-180.0
φ O ₄ -C ₁ -C ₂ -H ₅	0.0	-8.1	-7.2	-8.0	-6.0	0.0	-1.9	0.0	0.0
φ H ₅ -C ₁ -C ₂ -C ₆	-180.0	174.3	174.6	173.8	174.7	-180.0	178.5	-180.0	-180.0
φ O ₃ -C ₁ -C ₂ -C ₆	0.0	-8.4	-9.1	-7.8	-7.0	0.0	-1.6	0.0	0.0
φ O ₄ -C ₂ -C ₆ -H ₇	0.1	3.9	4.4	6.7	-3.1	0.0	1.4	0.3	0.0
φ O ₄ -C ₂ -C ₆ -H ₈	-121.6	-117.6	-116.5	-114.3	-125.1	-121.8	-120.5	-121.8	-121.7
φ O ₄ -C ₂ -C ₆ -H ₉	121.7	125.9	126.6	129.0	117.6	121.8	122.8	122.4	121.7
φ H ₁₀ -O ₁₁ -N ₁₂ -H ₁₃	124.8	119.3	116.9	129.2	126.8	125.0	135.1	125.2	125.1
φ H ₁₀ -O ₁₁ -N ₁₂ -H ₁₄	-124.8	-129.9	-131.7	-121.2	-123.1	-125.0	-115.1	-124.9	-125.0
φ O ₄ -H ₁₀ -O ₁₁ -N ₁₂		-11.3						-54.3	
φ C ₂ -O ₄ -H ₁₀ -O ₁₁		20.4				0.1		53.5	
φ O ₃ -H ₁₀ -O ₁₁ -N ₁₂			3.4						-0.2
φ C ₁ -O ₃ -H ₁₀ -O ₁₁			3.6				-96.8		0.2
φ O ₄ -H ₁₃ -N ₁₂ -O ₁₁				8.9					
φ C ₂ -O ₄ -H ₁₃ -N ₁₂				-26.1					
φ C ₁ -O ₃ -H ₁₃ -N ₁₂					-26.2				-0.1
φ C ₁ -C ₆ -H ₈ -N ₁₂							-95.8		
φ C ₁ -C ₆ -H ₇ -N ₁₂								-0.4	

Table S5. - Continuation

Parameter	M	V _{MHk}	V _{MHa}	VI _{MHk}	VI _{MHa}	VII _{MHk}	VIII _{MHk}	VIII _{MHa}
r C ₁ -C ₂	1.528	1.529	1.528	1.528	1.526	1.529	1.528	1.527
r C ₁ -O ₃	1.215	1.216	1.217	1.214	1.219	1.218	1.215	1.219
r C ₂ -O ₄	1.221	1.224	1.221	1.225	1.221	1.224	1.224	1.221
r C ₁ -H ₅	1.101	1.098	1.100	1.100	1.099	1.099	1.101	1.097
r C ₂ -C ₆	1.499	1.496	1.496	1.495	1.498	1.497	1.495	1.499
r C ₆ -H ₇	1.083	1.083	1.083	1.084	1.083	1.083	1.084	1.083
r C ₆ -H ₈	1.088	1.088	1.088	1.088	1.088	1.088	1.089	1.088
r C ₆ -H ₉	1.088	1.088	1.088	1.088	1.088	1.088	1.088	1.088
r O ₁₁ -H ₁₀	0.959	0.965	0.964	0.965	0.964	0.959	0.959	0.959
r O ₁₁ -N ₁₂	1.448	1.446	1.447	1.446	1.447	1.454	1.454	1.454
r N ₁₂ -H ₁₃	1.012	1.013	1.013	1.013	1.012	1.012	1.013	1.014
r N ₁₂ -H ₁₄	1.012	1.013	1.013	1.013	1.012	1.015	1.015	1.013
R O ₄ -H ₁₀		2.005		1.978				
R O ₁₁ -H ₅		2.359			2.642	2.331		2.460
R O ₃ -H ₁₀			1.990		2.047			
R O ₁₁ -H ₉			2.539					
R O ₁₁ -H ₇				2.401			2.362	
R O ₄ -H ₁₄						2.166	2.157	
R O ₃ -H ₁₃								2.207
θ C ₂ -C ₁ -O ₃	122.8	121.6	123.6	122.7	122.2	121.5	122.8	122.0
θ C ₁ -C ₂ -O ₄	117.7	118.4	116.7	116.9	117.5	118.1	117.0	117.7
θ C ₂ -C ₁ -H ₅	114.1	115.1	114.0	114.1	115.1	114.9	114.1	115.6
θ O ₃ -C ₁ -H ₅	123.1	123.3	122.4	123.2	122.6	123.6	123.1	122.4
θ C ₁ -C ₂ -C ₆	117.3	117.4	118.0	117.7	117.3	117.5	117.7	117.4
θ O ₄ -C ₂ -C ₆	125.0	124.2	125.3	125.4	125.1	124.4	125.3	124.9
θ H ₇ -C ₆ -H ₈	110.5	110.6	110.4	110.7	110.5	110.6	110.8	110.5
θ H ₇ -C ₆ -H ₉	110.6	110.7	110.5	110.7	110.5	110.7	110.9	110.5
θ H ₈ -C ₆ -H ₉	106.6	106.5	106.1	106.3	106.6	106.4	106.2	106.5
θ H ₁₀ -O ₁₁ -N ₁₂	101.7	102.9	102.3	102.7	102.6	101.7	101.6	101.7
θ O ₁₁ -N ₁₂ -H ₁₃	103.6	103.8	103.8	103.8	103.8	103.5	103.5	102.9
θ O ₁₁ -N ₁₂ -H ₁₄	103.6	103.8	103.8	103.8	103.8	103.2	103.2	103.5
θ H ₁₃ -N ₁₂ -H ₁₄	105.8	105.5	105.5	105.5	105.5	105.9	105.9	106.0
θ O ₄ -H ₁₀ -O ₁₁		151.0		155.0				
θ C ₁ -O ₃ -H ₁₀		138.1			101.3			
θ O ₃ -H ₁₀ -O ₁₁			162.9		142.4			
θ C ₂ -H ₉ -O ₁₁			119.2					
θ C ₂ -H ₇ -O ₁₁				139.6			163.2	
θ C ₁ -H ₅ -O ₁₁					104.8			125.7
θ O ₄ -H ₁₄ -N ₁₂						167.4	169.5	
θ O ₃ -H ₁₃ -N ₁₂								153.2
φ O ₃ -C ₁ -C ₂ -O ₄	180.0	180.0	-178.1	180.0	180.0	-180.0	179.8	179.8
φ O ₄ -C ₁ -C ₂ -H ₅	0.0	0.0	1.8	0.0	0.0		-0.1	0.0
φ H ₅ -C ₁ -C ₂ -C ₆	-180.0	180.0	-178.2	180.0	180.0	-180.0	-180.0	179.9
φ O ₃ -C ₁ -C ₂ -C ₆	0.0	0.0	1.9	0.0	0.0	0.0	0.0	-0.1
φ O ₄ -C ₂ -C ₆ -H ₇	0.1	0.0	-0.7	-0.1	0.0	0.2	2.0	0.0
φ O ₄ -C ₂ -C ₆ -H ₈	-121.6	-121.8	-122.2	-122.1	-121.6	-121.6	-120.0	-121.6
φ O ₄ -C ₂ -C ₆ -H ₉	121.7	121.8	121.2	121.9	121.6	122.0	124.2	121.6
φ H ₁₀ -O ₁₁ -N ₁₂ -H ₁₃	124.8	124.9	124.9	124.9	124.9	121.4	121.8	128.5
φ H ₁₀ -O ₁₁ -N ₁₂ -H ₁₄	-124.8	-124.9	-125.0	-124.9	-124.9	-128.4	-128.0	-121.3
φ O ₄ -H ₁₀ -O ₁₁ -N ₁₂		179.9		-179.9				
φ C ₂ -O ₄ -H ₁₀ -O ₁₁		0.0		0.0				
φ O ₃ -H ₁₀ -O ₁₁ -N ₁₂			164.4					
φ C ₁ -O ₃ -H ₁₀ -O ₁₁			2.9		-0.3			
φ C ₂ -C ₆ -H ₇ -O ₁₁				0.3			-0.7	
φ C ₁ -H ₅ -O ₁₁ -H ₁₀					-0.3			
φ C ₂ -O ₄ -H ₁₄ -N ₁₂						1.5	3.0	
φ C ₂ -C ₁ -H ₅ -O ₁₁						12.2		
φ C ₁ -O ₃ -H ₁₃ -N ₁₂								2.7
φ O ₃ -C ₁ -H ₅ -O ₁₁								-4.8

Table S6. The comparison of the observed wavenumbers (cm^{-1}) and wavenumber shifts ($\Delta\nu = \nu_{\text{MH}} - \nu_{\text{M}}$) for the $\text{CH}_3\text{COCHO-NH}_2\text{OH}$ (MH) complexes present in the Ar and N_2 matrices with the corresponding calculated values for the complexes $\text{I}_{\text{MHk}} - \text{IV}_{\text{MHk}}$ and $\text{I}_{\text{MHa}} - \text{IV}_{\text{MHa}}$.

Approximate description	Experimental						Calculated							
	Ar			N_2			$\Delta\nu$							
	ν_{M}	ν_{MH}	$\Delta\nu^1$	ν_{M}	ν_{MH}	$\Delta\nu^1$	I_{MHk}	II_{MHk}	III_{MHk}	IV_{MHk}	I_{MHa}	II_{MHa}	III_{MHa}	IV_{MHa}
NH_2OH														
$\nu(\text{OH})$	3635.5	3511.7 3507.6	-125.8	3637.6	3505.7 3495.2	-137.1	-148(138)	-14(62)	-127(494)	-132(520)	-156(73)	-17(68)	-92(367)	-112(402)
$\delta(\text{NOH})^2$	1351.2	1404.1	+52.9	1367.4	1402.7 1400.3	+34.1	+64(58)	-13(12)	+81(28)	+86(27)	+62(51)	-17(17)	+64(31)	+74(33)
$\omega(\text{NH}_2)$	1118.3	1134.1 1131.2	+14.4	1133.0	1139.3 1140.6	+7.0	+17(117)	+19(148)	+23(109)	+21(105)	+11(109)	+13(152)	+22(110)	+21(120)
$\nu(\text{NO})$	895.6	904.8 902.9	+8.3	895.3			+9(5)	-4(7)	+8(8)	+6(7)	+13(6)	-2(7)	+7(7)	+5(8)
CH_3COCHO														
$\nu(\text{CH})$	2843.1 2840.7	2858.2	+16.3	2844.7 2840.2 2835.9			-1(60)	+8(60)	+20(1)	+4(60)	+24(51)	+5(57)	+17(57)	+57(11)
$\nu_{\text{ket}}(\text{C=O})$	1733.5	1737.7	+4.2	1739.0 1737.2 1735.9	1741.3	+3.9	-5(132)	0(134)	-3(64)	-2(18)	+4(61)	0(53)	-4(38)	-3(80)
$\nu_{\text{ald}}(\text{C=O})$	1726.4	1705.4	-21.0	1730.1 1727.9	1710.8	-18.2	+6(27)	+7(11)	-14(126)	-4(145)	-10(97)	+2(88)	0(137)	-17(65)
$\delta(\text{CH}_3)^2$	1420.0	1416.1	-3.9	1423.2 1420.8			0(8) +3(24) -3(36)	-1(8) -1(18) -3(32)	-2(10) -2(20) +2(36)	+3(8) -2(10) +7(32)	+5(6) +3(25) -2(31)	+4(8) 0(11) -5(29)	+13(9) +6(10) +10(25)	+1(10) 0(15) 0(30)
$\nu_{\text{as}}(\text{C-C})$	1228.3	1233.7	+5.4	1234.5 1229.9	1240.5 1237.1	+6.6	+5(14)	+9(16)	+8(13)	+15(20)	+5(15)	+6(15)	+3(16)	+1(16)
$\nu_{\text{s}}(\text{C-C})$	777.1	784.7	+7.6	780.9 779.5 777.6	785.8 782.3	+4.8	+6(15)	+6(13)	+4(13)	+6(13)	+6(15)	+5(11)	+5(14)	+2(20)

¹ In the case when the splitting of the band was observed the average of the two wavenumbers at which the two peaks appear was taken into account to calculate $\Delta\nu$ value.

² The 1416.1 and 1404.1 cm^{-1} bands observed in the spectra of the complex in Ar matrix are assigned to the coupled $\delta(\text{NOH})+\delta(\text{CH}_3)$ vibrations.

Table S7. The comparison of the observed wavenumbers (cm⁻¹) and wavenumber shifts ($\Delta\nu = \nu_{\text{MH}} - \nu_{\text{M}}$) for the CH₃COCHO-ND₂OD (MH) complexes present in the Ar and N₂ matrices with the corresponding calculated values for the complexes **I_{MHk}** – **IV_{MHk}** and **I_{MHa}** – **IV_{MHa}**.

Approximate description	Experimental						Calculated							
	Ar			N ₂			$\Delta\nu$							
	ν_{M}	ν_{MH}	$\Delta\nu^1$	ν_{M}	ν_{MH}	$\Delta\nu^1$	I_{MHk}	II_{MHk}	III_{MHk}	IV_{MHk}	I_{MHa}	II_{MHa}	III_{MHa}	IV_{MHa}
ND ₂ OD														
$\nu(\text{OD})$	2685.1	2596.4 2593.2	-90.3	2686.9	2595.6 2593.5	-92.3	-109(73)	-10(34)	-93(253)	-96(264)	-114(40)	-12(37)	-67(189)	-61(207)
$\nu(\text{NO})$	818.4	821.1	+2.7	827.6			+22(44)	-10(50)	+22(38)	+23(37)	+23(33)	-8(50)	+14(33)	+16(44)
CH ₃ COCHO														
$\nu(\text{CH})$	2843.1 2840.7	2858.6	+16.7				-2(58)	+8(59)	+20(1)	+4(56)	+24(51)	+6(57)	+17(54)	+57(12)
$\nu_{\text{ket}}(\text{C=O})$	1733.5	1737.0	+3.5	1739.0 1737.2 1735.9			-6(137)	0(144)	-3(63)	-7(144)	+3(59)	+3(52)	-4(37)	-3(79)
$\nu_{\text{ald}}(\text{C=O})$	1726.4	1705.9	-20.5	1730.1 1727.9			+6(30)	+7(12)	-14(128)	+1(19)	-10(103)	+1(102)	0(139)	-17(70)
$\delta(\text{CH}_3)$	1420.0			1366.1 1364.5 1368.2	1369.2	+2.9	+6(1)	+4(1)	+27(3)	+27(3)	+4(7) +3(14) -2(33)	+4(8) 0(12) -5(34)	0(16) +6(21) +10(23)	+1(10) 0(16) 0(30)
$\nu_{\text{as}}(\text{C-C})$	1228.3			1234.5 1229.9	1241.4 1238.8	+7.9	+13(18)	+9(16)	+8(14)	+16(22)	+6(17)	+6(16)	+3(15)	+1(16)
$\nu_{\text{s}}(\text{C-C})$	777.1			780.9 779.5 777.6	785.6 783.7	+5.4	+8(11)	+5(12)	+4(13)	+5(13)	+5(12)	+3(11)	+5(13)	+2(21)

¹ In the case when the splitting of the band was observed the average of the two wavenumbers at which the two peaks appear was taken into account to calculate $\Delta\nu$ value.