

Supplementary Material

Synthesis and Characterization of Newly Designed and Highly Solvatochromic Double Squaraine Dye for Sensitive and Selective Recognition towards Cu²⁺

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1. NMR spectral analysis:

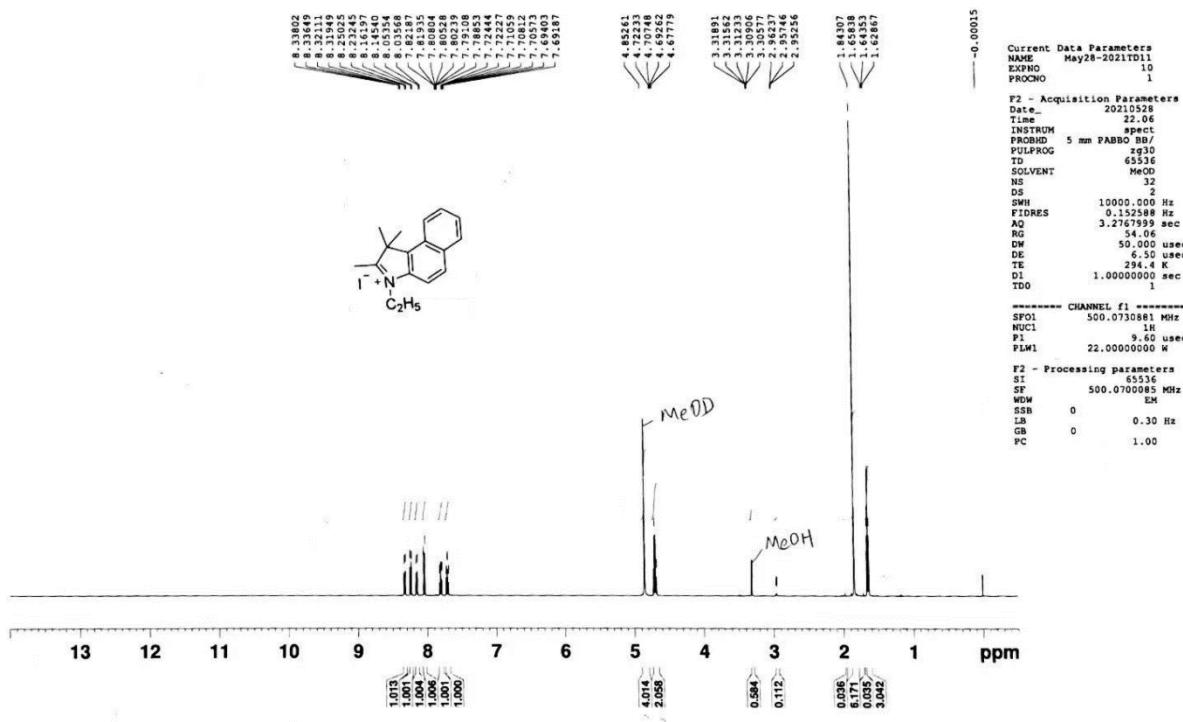


Figure S1. ^1H -NMR spectrum of **3** in MeOD.

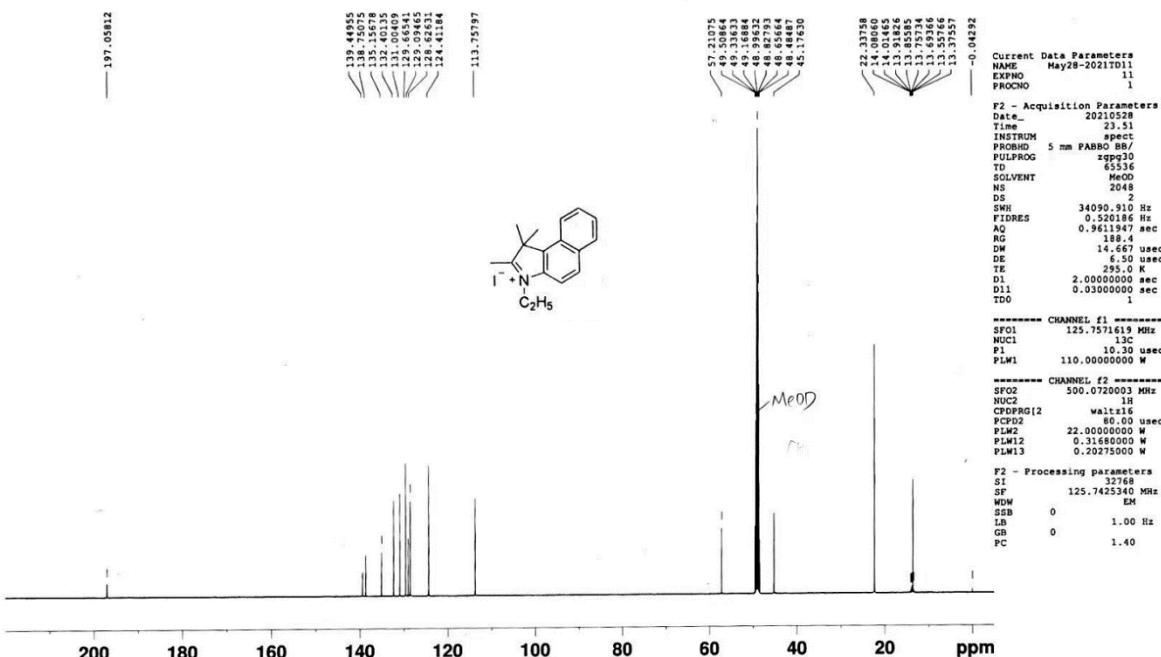


Figure S2. ^{13}C -NMR spectrum of **3** in MeOD.

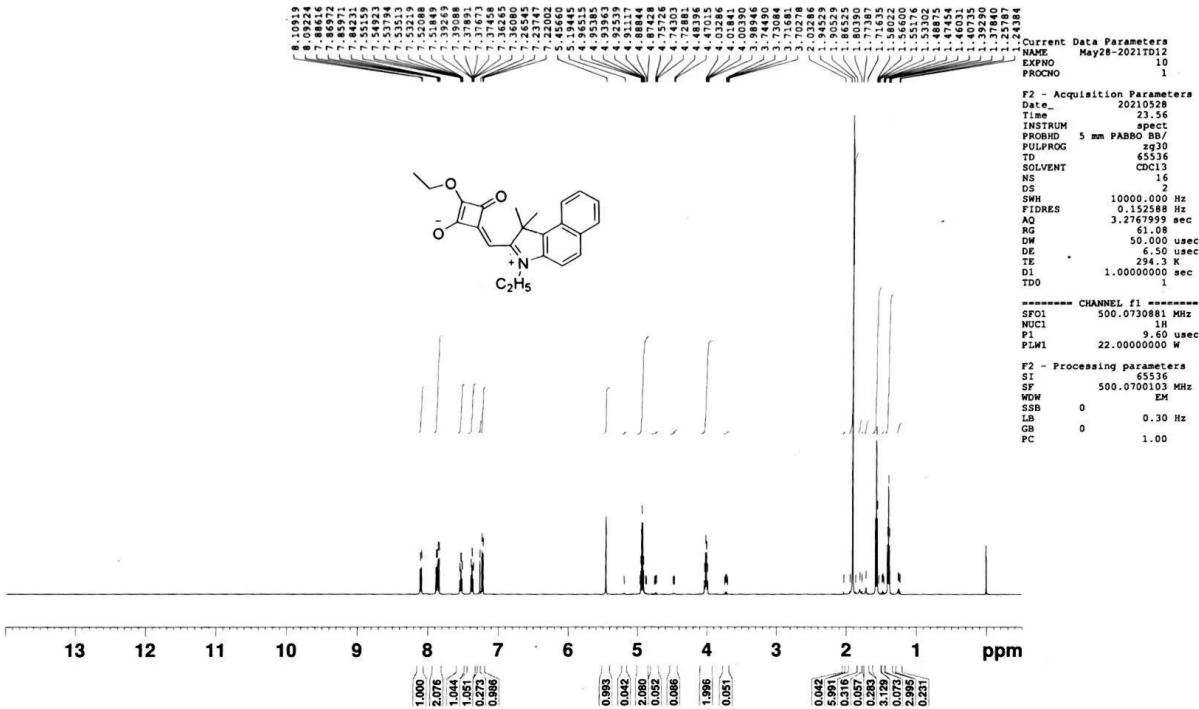


Figure S3. ¹H-NMR spectrum of 4 in CDCl₃.

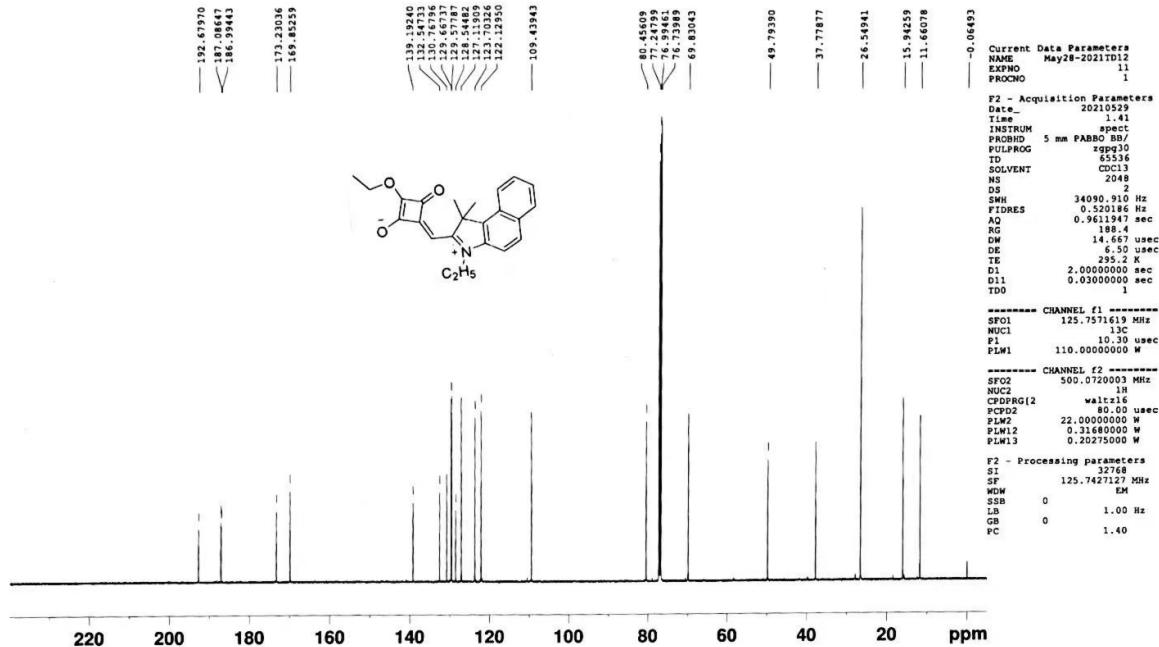


Figure S4. ¹³C-NMR spectrum of 4 in CDCl₃.

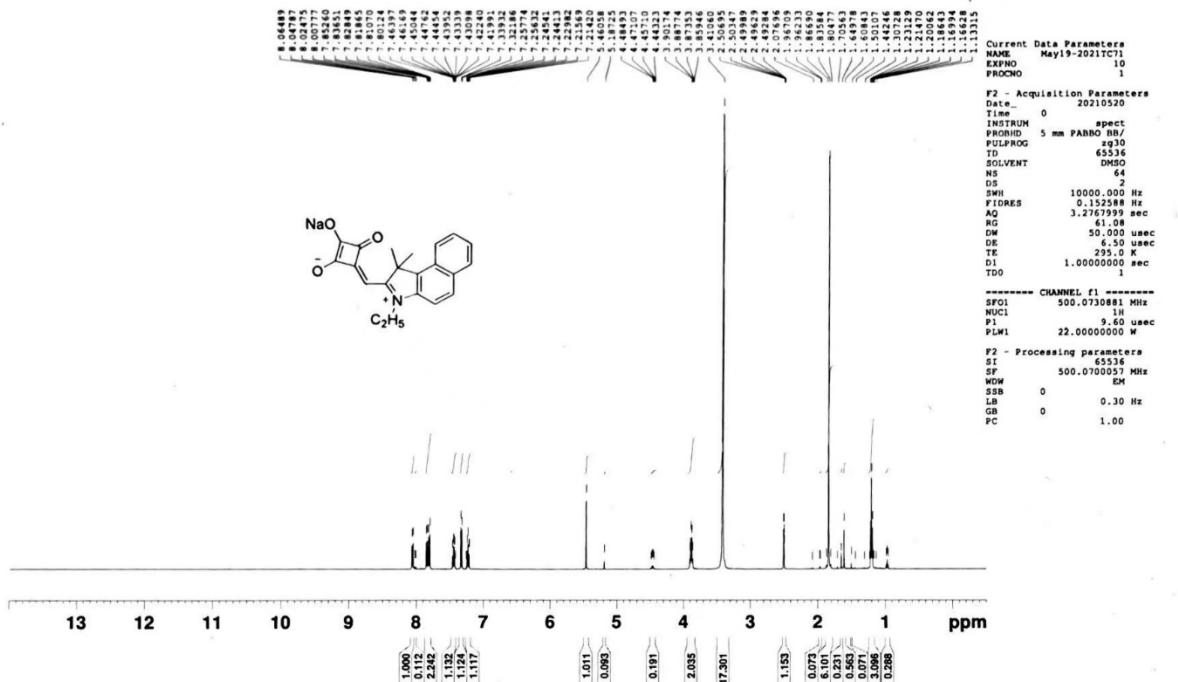


Figure S5. ^1H -NMR spectrum of **5** in DMSO.

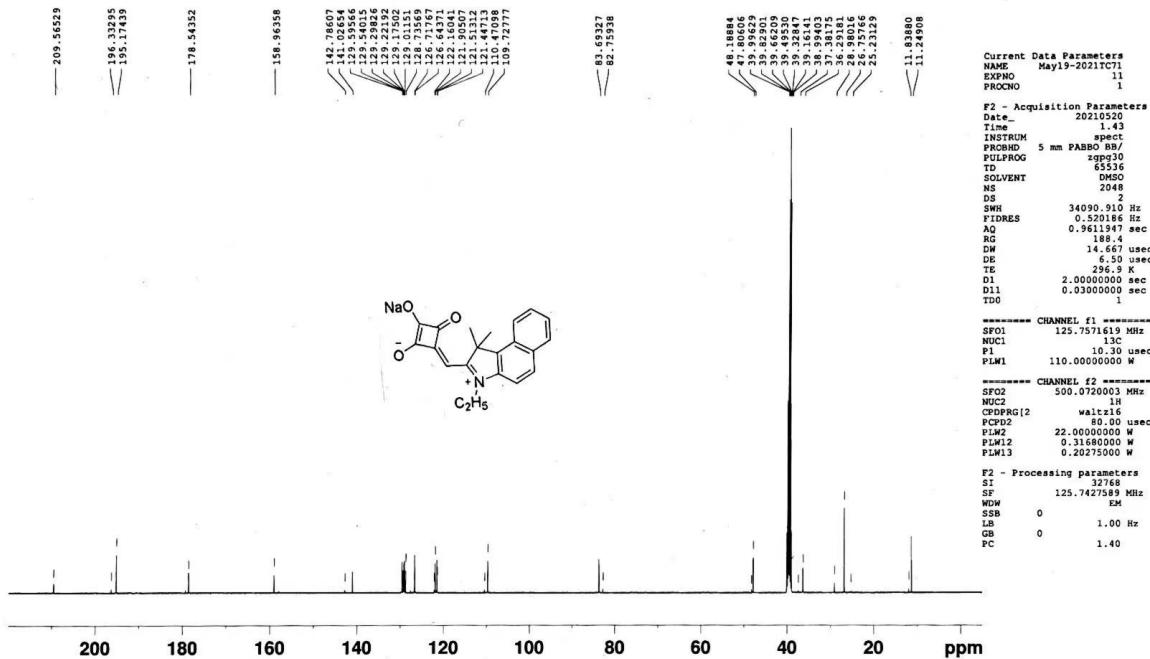


Figure S6. ^{13}C -NMR spectrum of **5** in DMSO.

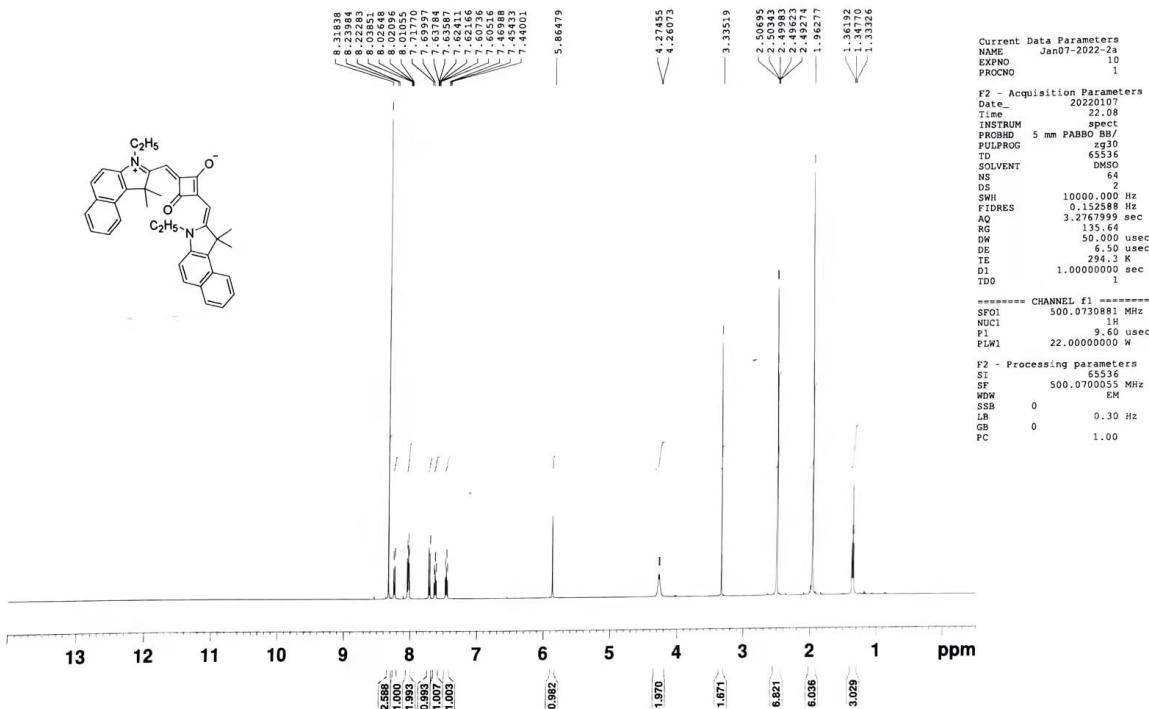


Figure S7. ^1H -NMR spectrum of **6** in DMSO.

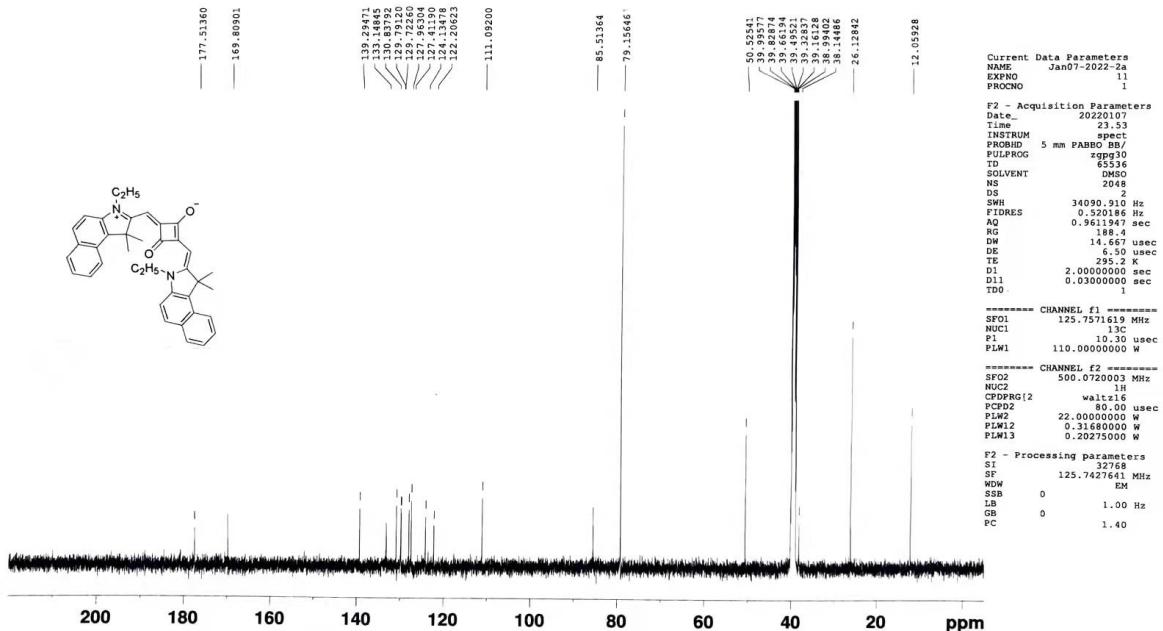
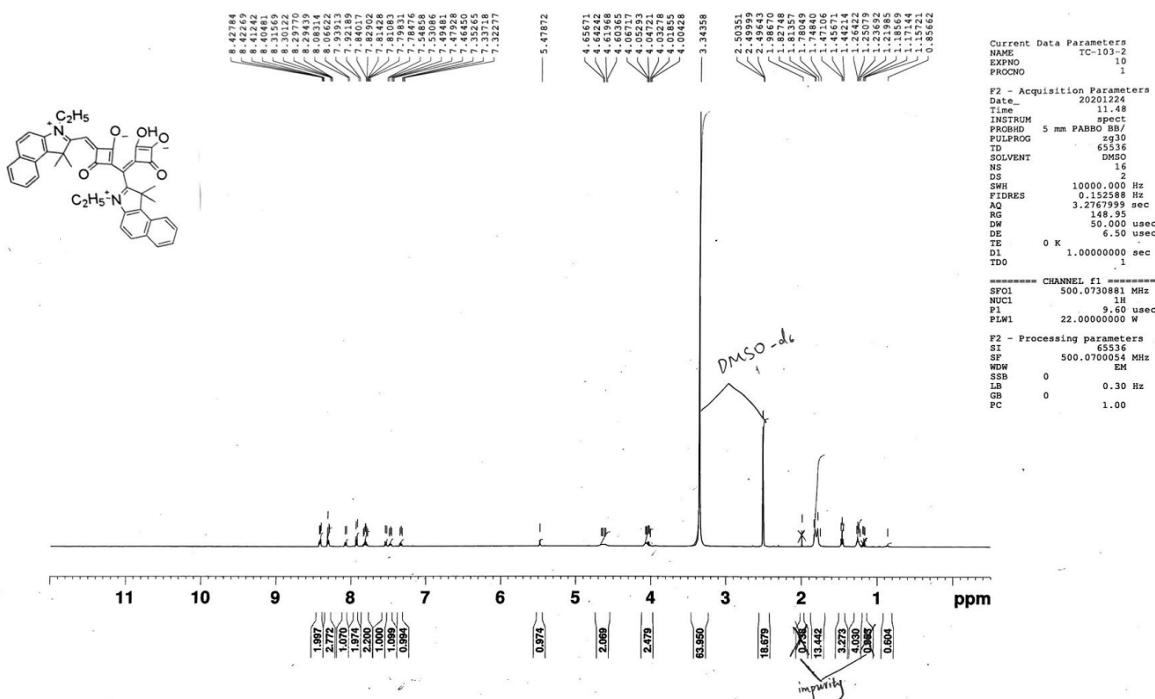
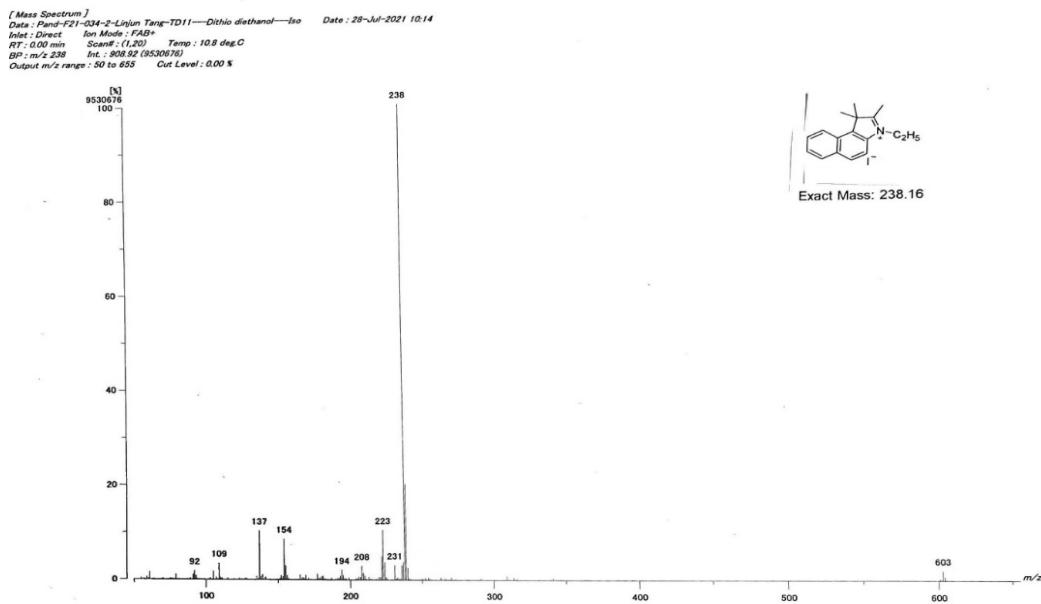


Figure S8. ^{13}C -NMR spectrum of **6** in DMSO.



2. Mass spectral analysis:



Data : Pand-F21-035-2-Liqun Tang-TD11—NR—Dithio diethanol—Iso Date : 28-Jul-2021 11:12
 Sample : -
 RT : 3.05 min Scan# : 25
 Elements : C 17.0, H 20.0, N 1.0
 Mass Tolerance : 1000ppm, 50mmu if m/z < 5, 50mmu if m/z > 50
 Unsaturation (U.S.) : -0.5 - 10.0
 Observed m/z Int% Err(ppm / mmu) U.S. Composition
 f 238.1597 100.00 +0.5 / +0.1 8.5 C17 H20 N

Figure S11. HR-MS spectrum of 3.

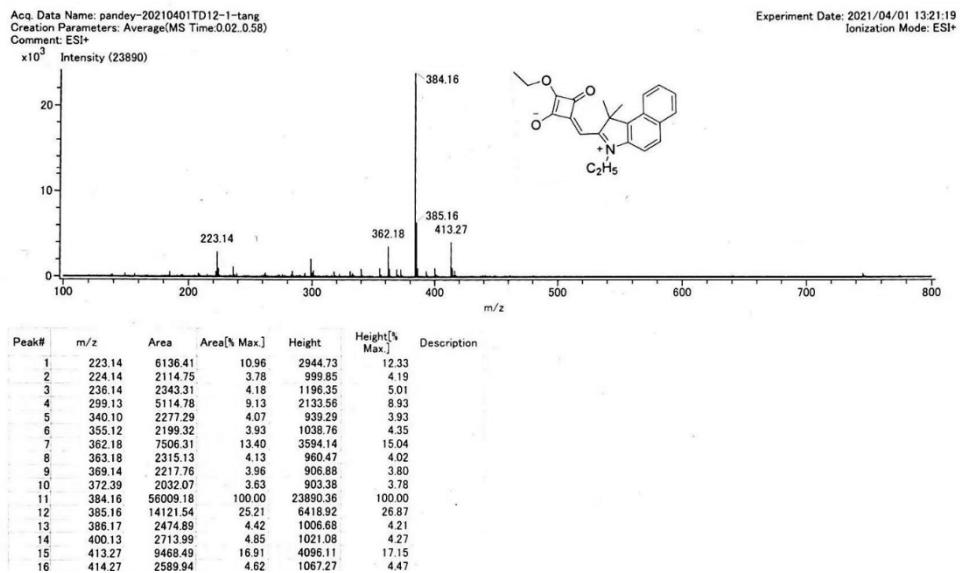
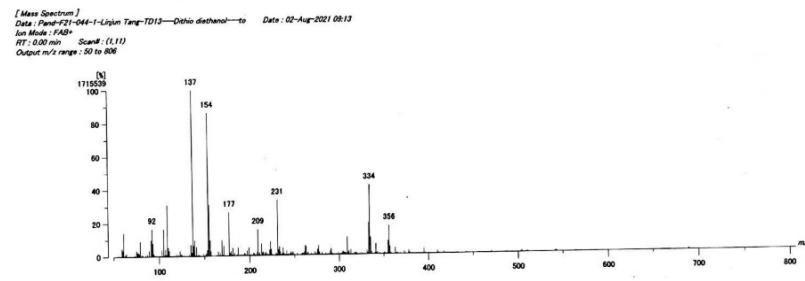
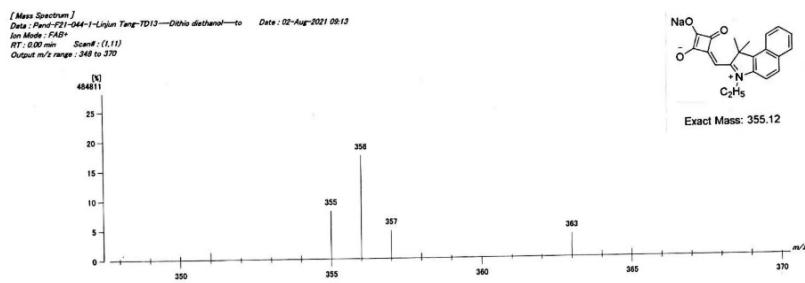
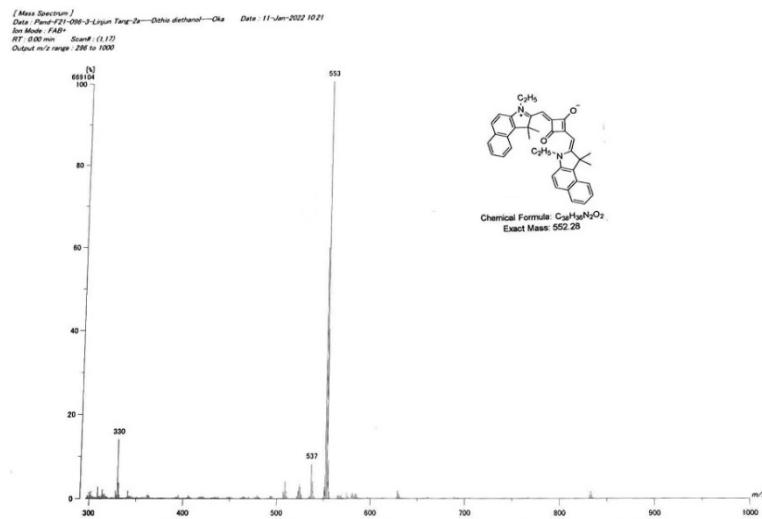


Figure S12. TOF-MS spectrum of 4.



Data : Pard-F21-045-2-Lijun Tang-TD13==HR—Othio diethanol—to Date : 02-Aug-2021 14:04
Sample :
Note :
RT : 1.22 min Scan# : 32
Elements : C 2.1/2, H 18/2, N 1/2, O 3/2, Na 1/2
Mass Tolerance : 1000ppm, 5mmu if m/z < 50mmu if m/z > 50
Unsaturation (U.S.) : -0.5 - 0.0
Observed m/z Int% Err(ppm / mmu) U.S. Composition
1 355.1178 60.14 -1.8 / -0.8 13.0 C21 H18 N O2 Na

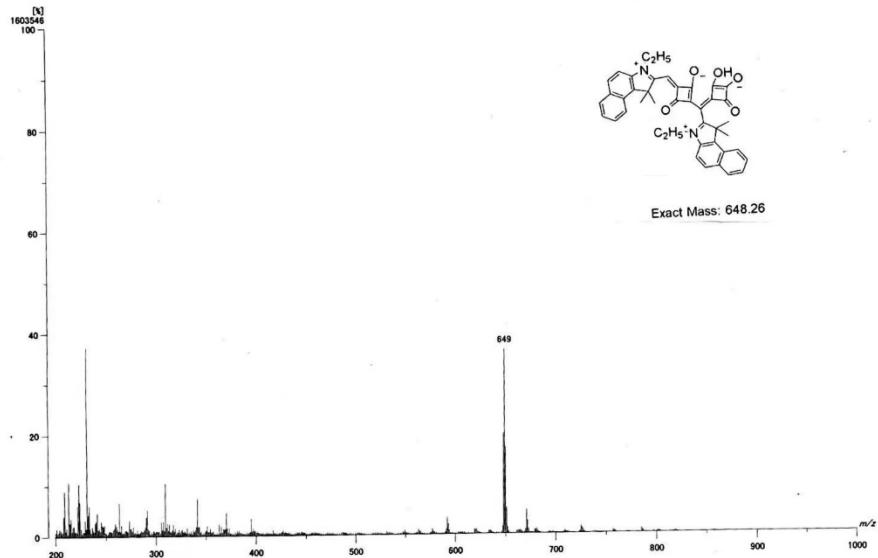
Figure S13. HR-MS spectrum of 5.



Data : Pard-F21-097-4-Lijun Tang-Td==HR—Othio diethanol—Oka Date : 11-Jan-2022 16:38
Sample :
Note :
RT : 6.51 min Scan# : 27
Elements : C 12.36/2, H 37/2, N 2/2, O 2/2
Mass Tolerance : 1000ppm, 5mmu if m/z < 50mmu if m/z > 50
Unsaturation (U.S.) : -0.5 - 2.0
Observed m/z Int% Err(ppm / mmu) U.S. Composition
1 552.2778 100.00 -14.5 / -8.0 22.0 12C37 13C H37 N 2 O2
2 552.2778 100.00 +8.3 / +4.6 22.5 12C37 13C H35 N2 O2
3 552.2778 100.00 +0.2 / +0.1 22.0 12C38 H36 N2 O2
4 553.2808 78.63 -0.2 / -0.1 22.0 12C37 13C H38 N2 O2
5 553.2808 78.63 -8.3 / -4.6 21.5 12C38 H37 N2 O2
6 554.2885 28.19 -4.3 / -2.4 21.5 12C37 13C H37 N2 O2

Figure S14. HR-MS spectrum of 6.

[Mass Spectrum]
 Data : Pand-F22-059-3-Jijun Tang---TMI---DithiodiEtOH---Ola Date : 04-Jul-2022 10:59
 Ion Mode: FAB+
 RT : 0.00 min Scan#: (1.16)
 Output m/z range : 199 to 1000



Data : Pand-F22-060-2-Jijun Tang---TMI---HR-DithiodiEtOH---Ola Date : 04-Jul-2022 11:23

Sample : -
 Note : -
 RT : 1.72 min Scan#: 8
 Elements : 12C 42.0, 13C 2.0, N 2.2, O 5.5
 Mass Tolerance : 1000ppm 5mmu if m/z < 5, 50mmu if m/z > 50
 Unsaturation (U.S.) : -0.5 - 30.0

Observed m/z	Int%	Err(ppm / mmu)	U.S.	Composition
1 648.2639	64.93	+16.1 / +10.4	27.0	12C40 13C2 H34 N2 O5
2		+9.2 / +5.9	26.5	12C41 13C2 H35 N2 O5
3		+2.3 / +1.5	26.0	12C42 H39 N2 O5
4 649.2712	100.00	+15.2 / +9.9	26.5	12C40 13C2 H35 N2 O5
5		+8.4 / +5.4	26.0	12C41 13C2 H36 N2 O5
6		+1.5 / +1.0	25.5	12C41 H37 N2 O5
7 650.2757	44.24	+10.1 / +6.6	26.0	12C41 13C2 H36 N2 O5
8		+3.2 / +2.1	25.5	12C41 13C2 H37 N2 O5
9 651.2834	.12.68	+9.9 / +6.4	25.5	12C40 13C2 H37 N2 O5

Figure S15. HR-MS spectrum of 1.

3. UV-Vis Spectroscopic Studies:

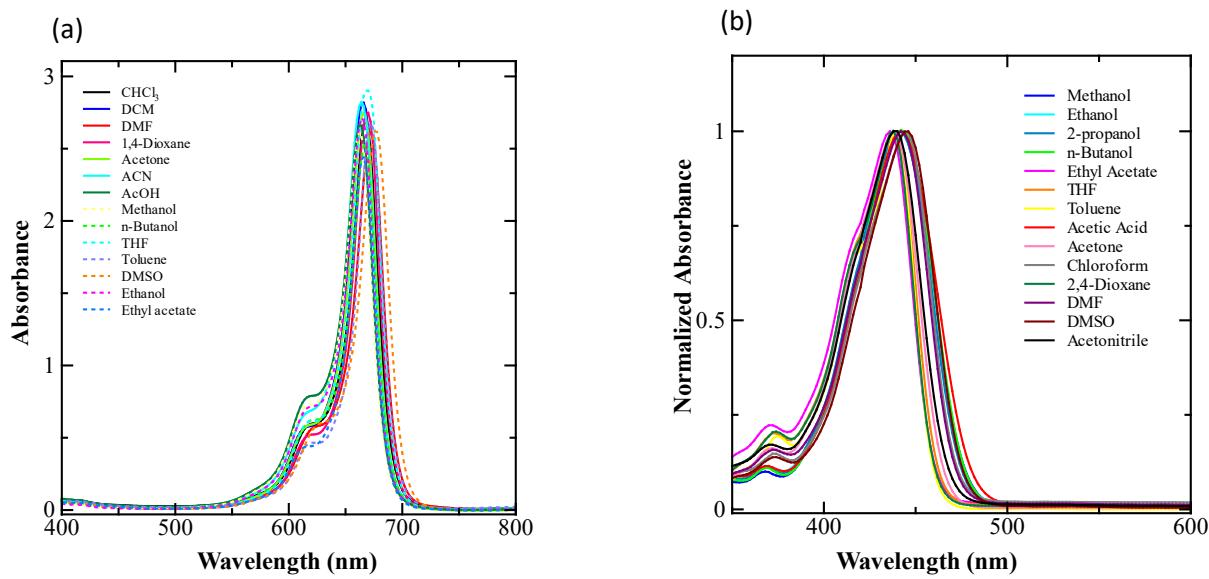


Figure S16. (a) UV-Vis spectra of **6** dissolved in different super dehydrated solvent (25 μ M) and (b) UV-Vis spectra of **4** dissolved in different super dehydrated solvents (10 μ M).

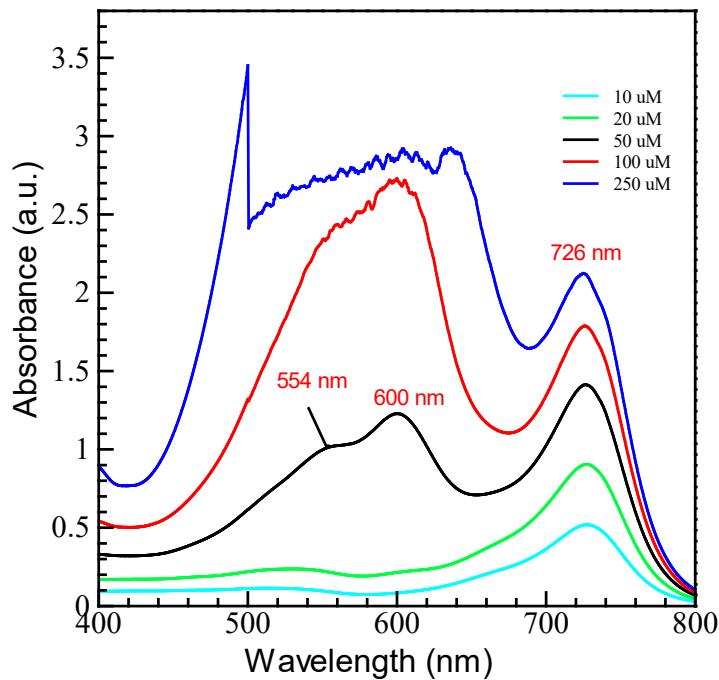


Figure S17. UV-Vis absorption of different concentration of DSQ (**1**) dissolved in DMF

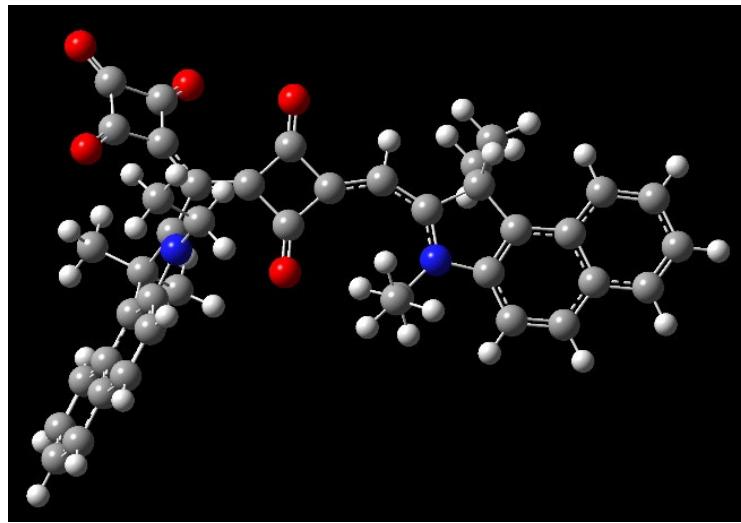


Figure S18. The three-dimensional structure of **1** utilizing Theoretical MO calculation (Gaussian G09 program)

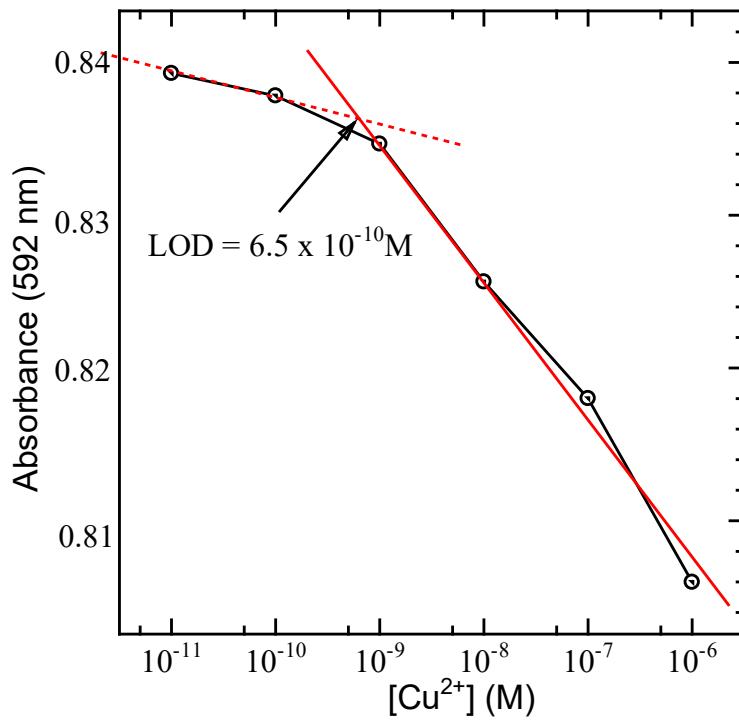


Figure S19. Change in absorption at 592 nm upon addition of Cu^{2+} ions showing limit of detection

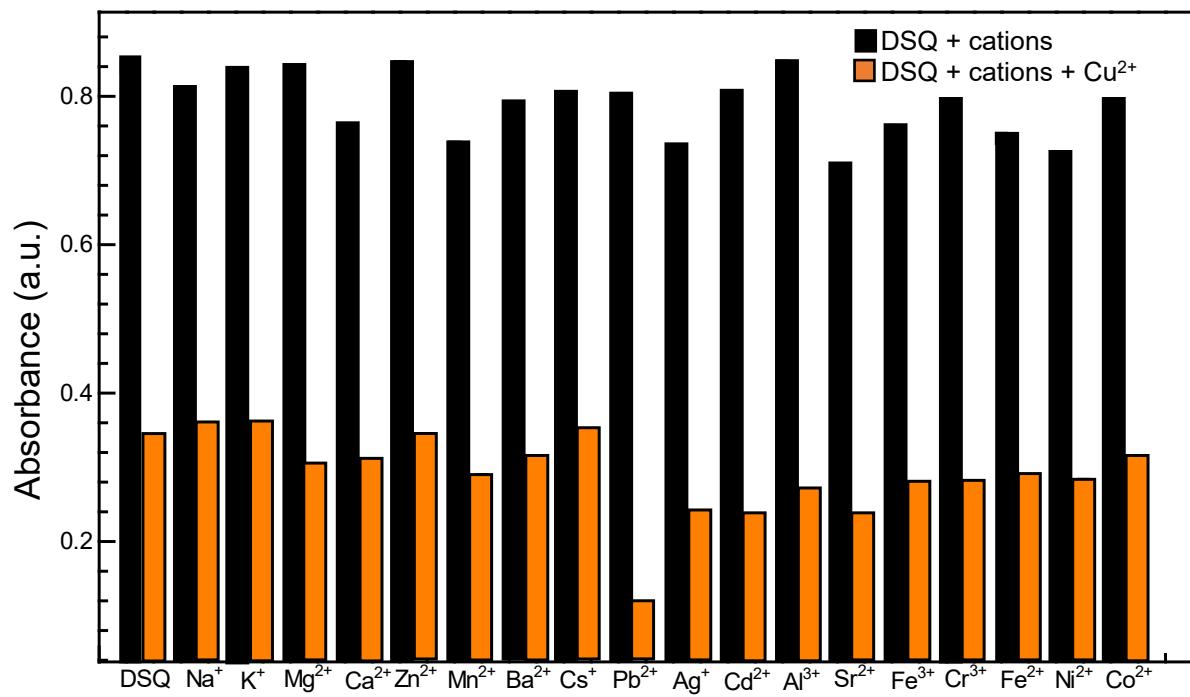


Figure S20. UV-Vis absorption at 592 nm of 10 μM DSQ (**1**) (dissolved in DMF/ACN, 1:99, v/v) in different ion solutions after addition of Cu^{2+}