

## **Supporting Information**

### **Fast Assembly of Metal Organic Framework UiO-66 in Acid-Base Tunable Deep Eutectic Solvent for the Acetalization of Benzaldehyde and Methanol**

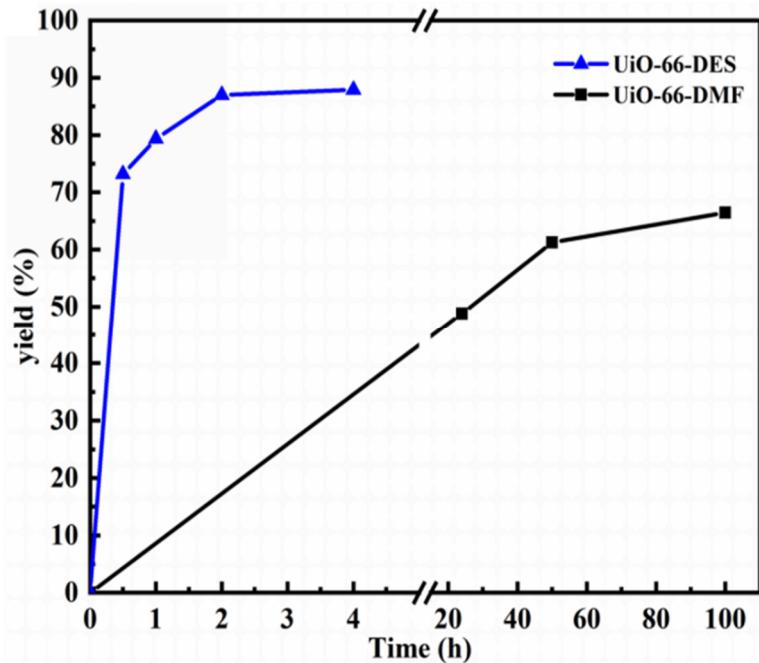
Lifang Chen \*, Xiangzhu Ye, Ting Zhang, Hao Qin, Hongye Cheng and Zhiwen Qi \*

State Key Laboratory of Chemical Engineering, School of Chemical Engineering,  
East China University of Science and Technology, Shanghai, 200237, P.R. China.

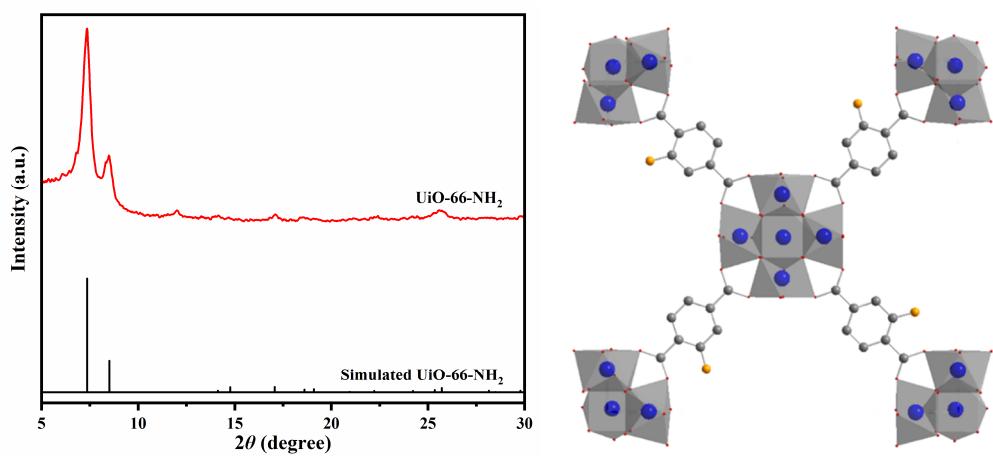
\*E-mail: lchen@ecust.edu.cn; zwqi@ecust.edu.cn

**Table S1.** pH for the aqueous solutions (0.01 mol/L) of MIm-PTSA mixtures with different molar ratios (20 °C) and corresponding melting point.

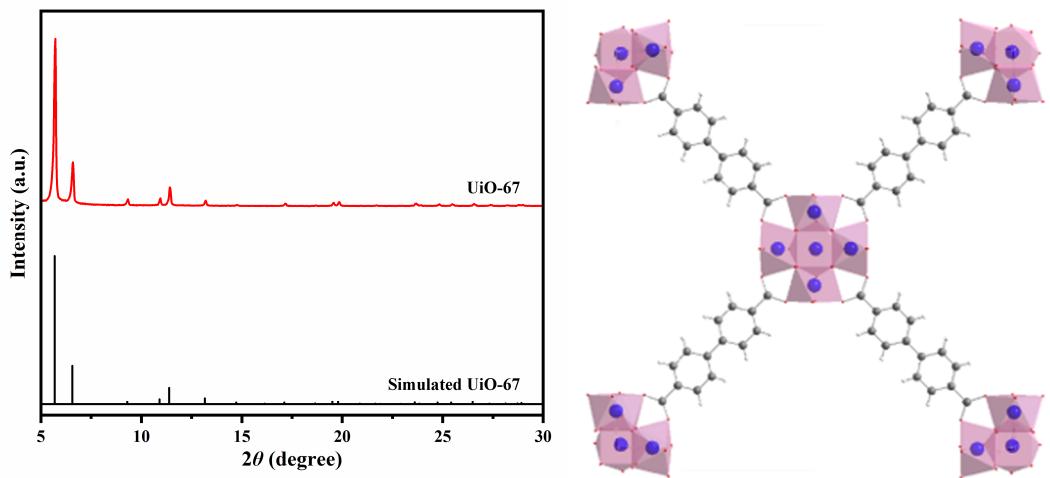
entry	MIm:pTSA	pH	melting point (°C)
1	1:0	10.13	142-143
2	3:1	7.41	39
3	4:3	6.91	9
4	1:2	1.82	16
5	0:1	1.32	103



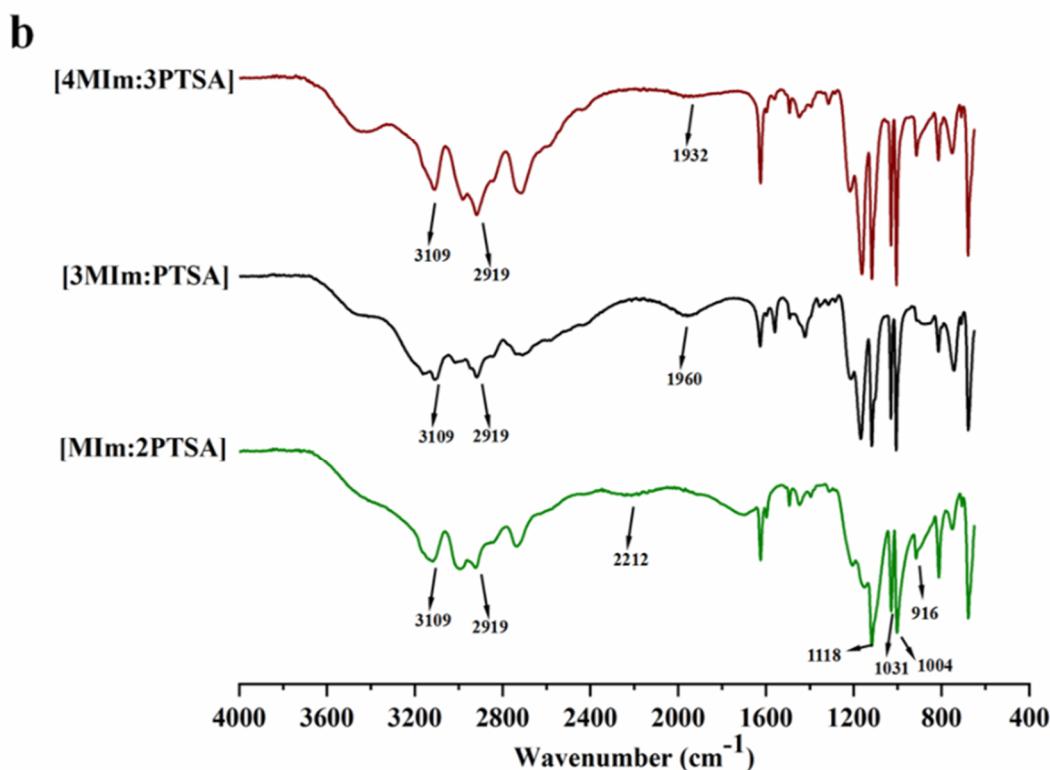
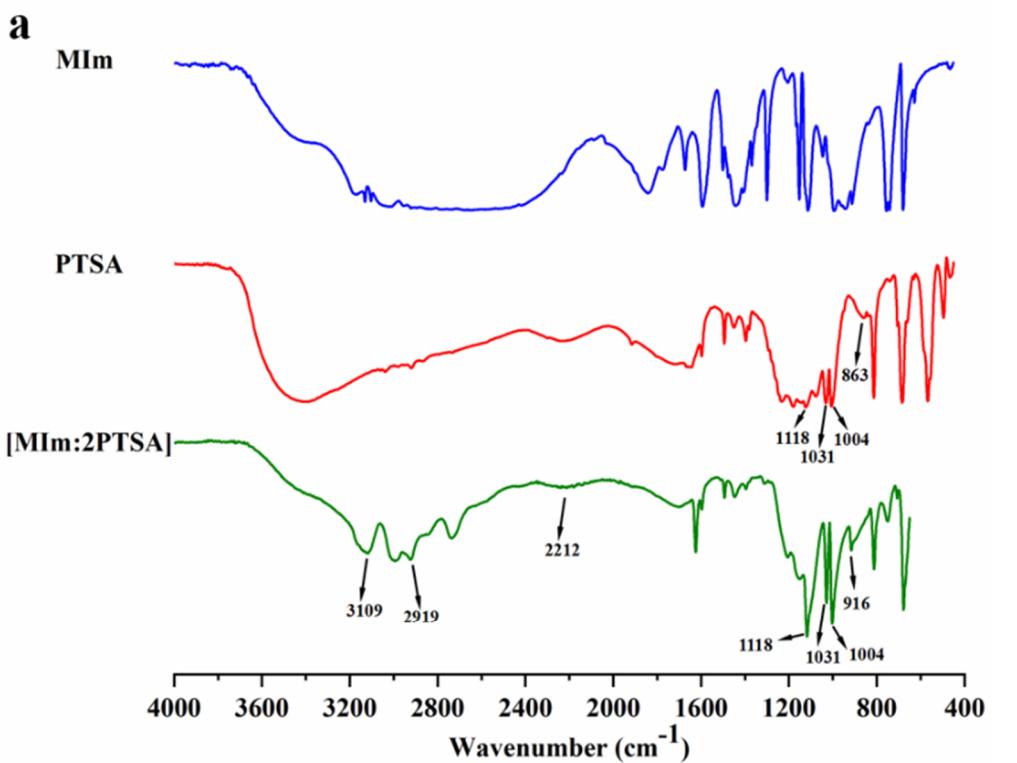
**Figure S1.** Formation rate of UiO-66 in DES and DMF.



**Figure S2.** XRD patterns of simulated and synthesized UiO-66-NH<sub>2</sub>, and the crystal structure of UiO-66-NH<sub>2</sub>.



**Figure S3.** XRD patterns of simulated and synthesized UiO-67, and the crystal structure of UiO-67.



**Figure S4.** (a) FT-IR spectra of MIm, PTSA, and [Mim:2PTSA]; (b) FT-IR spectra of [4Mim:3PTSA], [3Mim:PTSA], and [Mim:2PTSA].

**Table S2.** The catalytic performance of UiO-66-DES in the acetalization of furfural and 5-hydroxymethylfurfural with methanol.

entry	Reactant	Time (h)	Conversion (%)
1	furfural	1	78.4
2	5-hydroxymethylfurfural	1	93.1