Supplementary Materials

for

Effect of Copper-Modification of g-C_3N_4 on the Visible-Light-Driven Photocatalytic Oxidation of Nitrophenols

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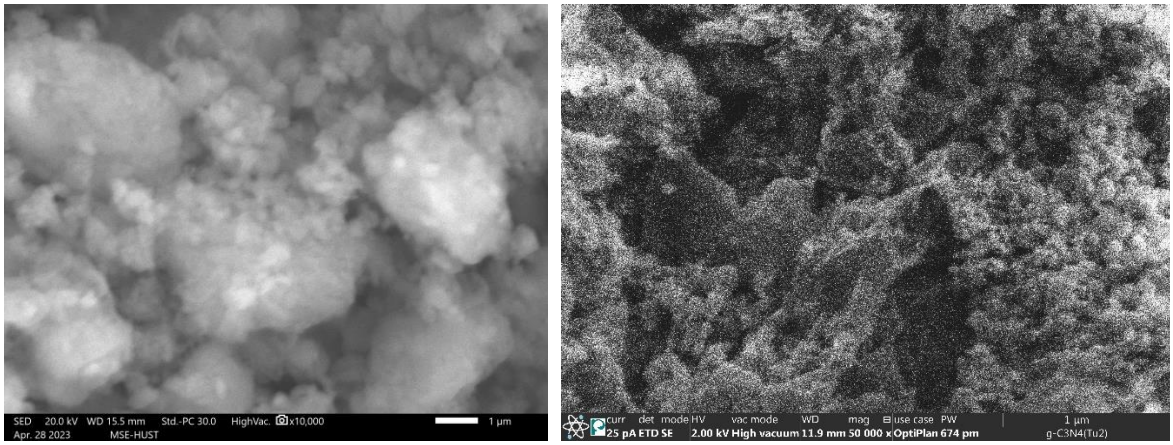

(**a**) (**b**)

Figure S1. SEM images of catalyst materials 3% Cu/g-C_3N_4 (**a**) and g-C_3N_4 (**b**) at 10 000× magnification.

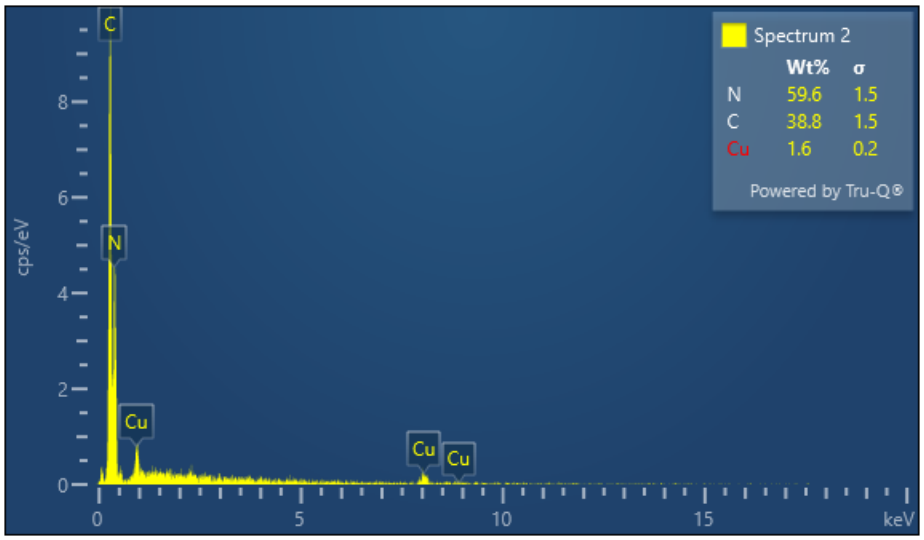

Figure S2. EDS spectrum of the 3% Cu/g-C_3N_4 sample (SEM).

Table S1. XPS surface composition (at.%) of the 3% Cu/g-C_3N_4 sample.

Element	Surface ratios (at.%)
O 1s	39.9
N 1s	51.8
C 1s	39.9
Cu 2p	3.7
N/Cu	14.0

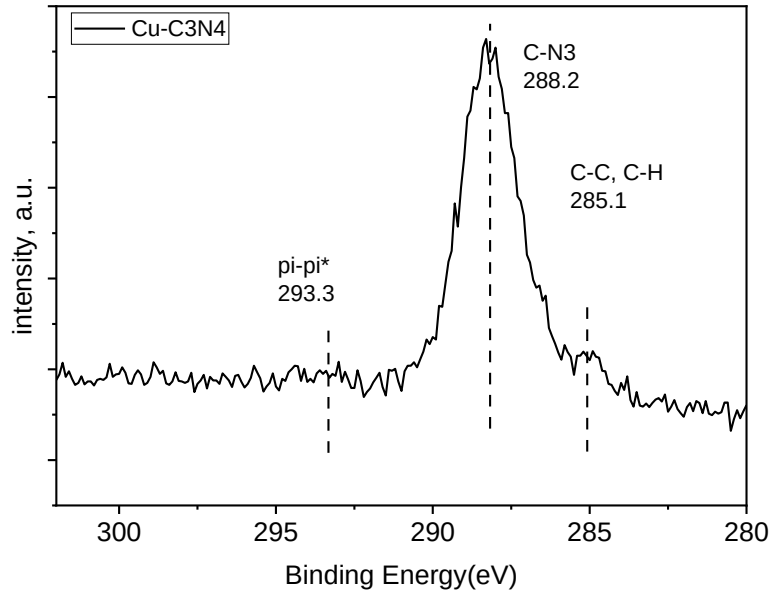

Figure S3. The C 1s binding energy region in the XPS of the 3% Cu/g-C_3N_4 sample.

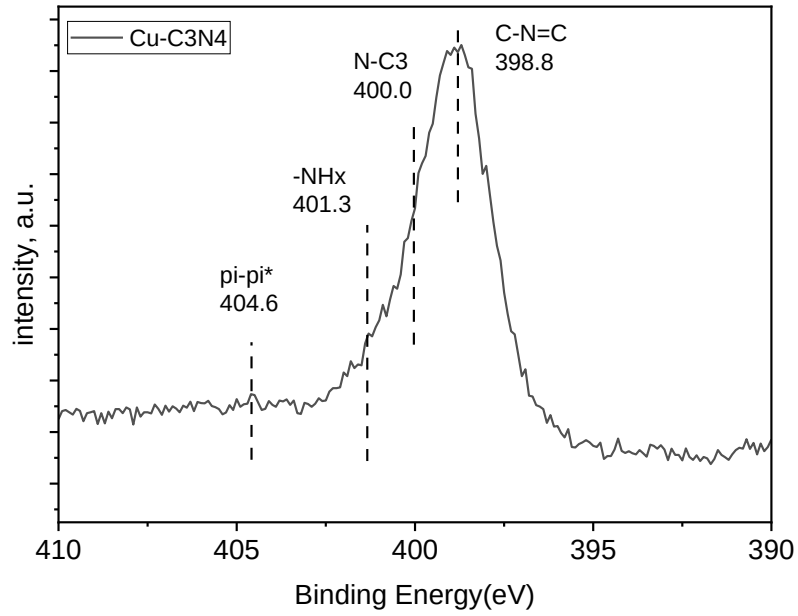

Figure S4. The N 1s binding energy region in the XPS of the 3% Cu/g-C_3N_4 sample.

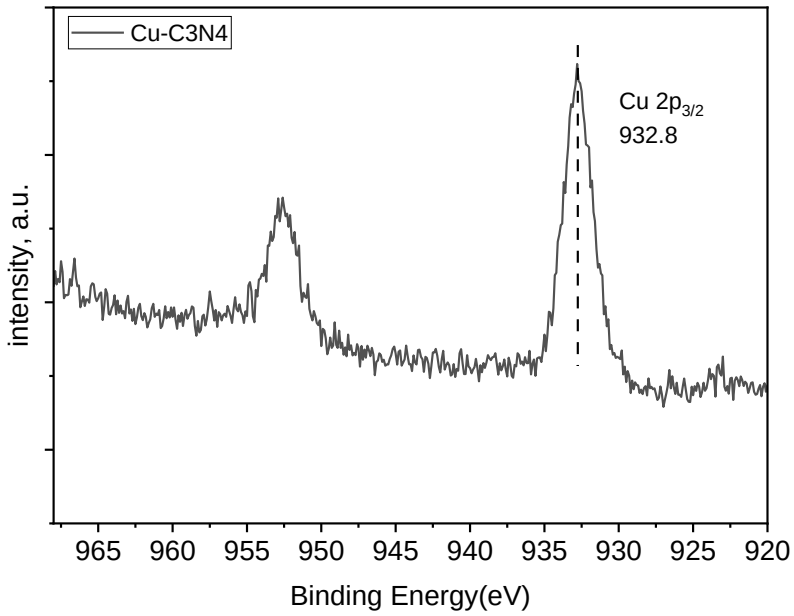

Figure S5. The Cu 2p binding energy region in the XPS of the 3% Cu/g-C_3N_4 sample.

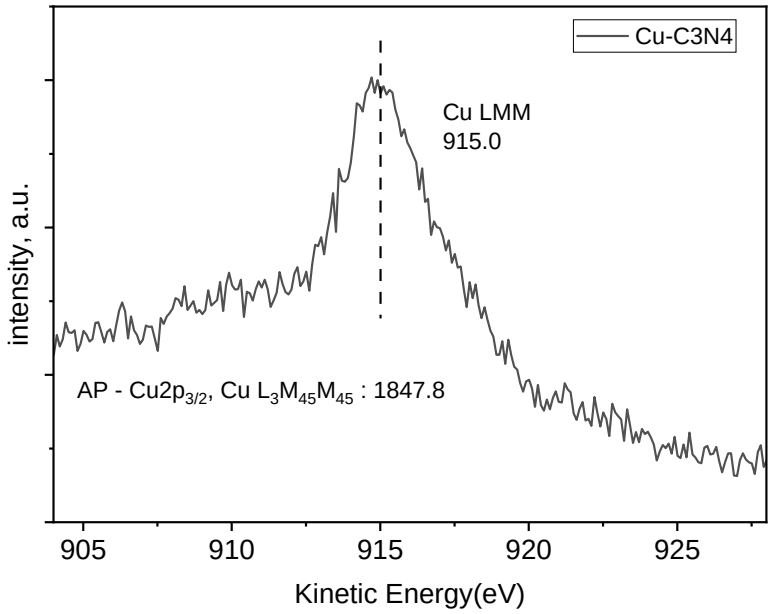

Figure S6. Cu LMM Auger peak and modified Auger parameter of the 3% Cu/g-C_3N_4 sample.

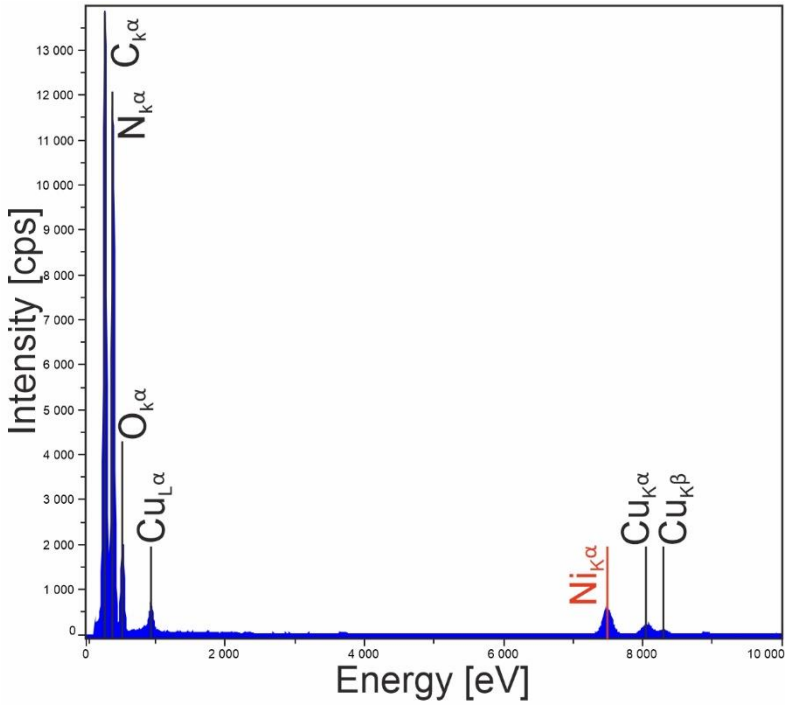

Figure S7. EDS spectrum of the 3% Cu/g-C_3N_4 sample (TEM).

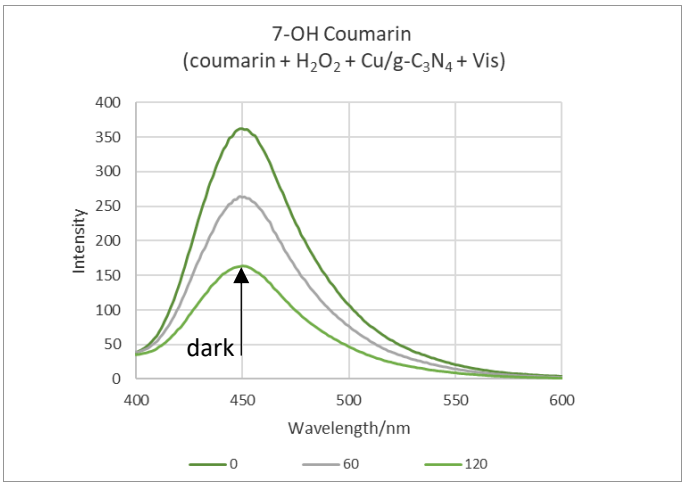

Figure S8. Formation of 7-hydroxy-coumarin upon visible-light irradiation (for 120 min) in the presence of 3% Cu/g-C_3N_4 catalyst. Before starting the irradiation, the reaction mixture was kept in dark for 5 hours.