

Article

## Discovery of a Potent Anti-Yeast Triterpenoid Saponin, Clematoside-S from *Urena lobata* L.

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**Abstract:** *Urena lobata* has been used as a traditional medicinal plant in India and China. In this study, we investigated the antimicrobial activity and isolated the active compound from the leaves of *U. lobata*. The 80% ethanol extract from *U. lobata* leaves showed an effective anti-yeast activity against *Saccharomyces cerevisiae* (*S. cerevisiae*) strains. Using a combination of chromatographic methods, (–)-trachelogenin (**1**) and clematoside-S (**2**) were isolated from this plant for the first time, and their chemical structure was identified by mass spectrometry (MS) and extensive nuclear magnetic resonance (NMR) data analysis. In addition, **1** was found to be inactive against all of the test microorganisms in the antimicrobial assay, whereas **2** exhibits a specific anti-yeast activity against *S. cerevisiae* strains with diameter of inhibition zones in the range from 11 to 20 mm. Furthermore, the MIC (minimum inhibitory concentration) and MBC (minimum bactericidal concentration)

values of **2** against *S. cerevisiae* strains were detected to be in the ranges of 0.61 to 9.8 µg/mL and 2.42 to 9.8 µg/mL, respectively. This is the first report of **2** with a specific anti-yeast activity. The above result suggests the potential application of *U. lobata* to be used as a natural anti-yeast agent in food preservation.

**Keywords:** *Urena lobata*; triterpene saponin; clematoside-S; anti-yeast activity

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## 1. Introduction

Food spoilage by the food-related yeast causes the deterioration of a wide range of foodstuffs such as wines, milk, fruit and vegetable juices, soft drink or meat. Spoilage yeasts not only significantly influence the cost and availability of foods and beverages, but also lead to economic losses in food industry [1–3]. Notably, *Saccharomyces cerevisiae* is listed on one of the most significant spoilage yeasts for fruit juices and soft drinks [4]. Although chemical preservatives can exclude yeast spoilage, there is a strong consumer demand to avoid or diminish the use of artificial preservatives. Therefore, much effort has been expended in the search for effective natural compounds from herbs and spices in order to control food spoilage caused by yeast and replace existing synthetic antibiotics in foodstuffs [5,6].

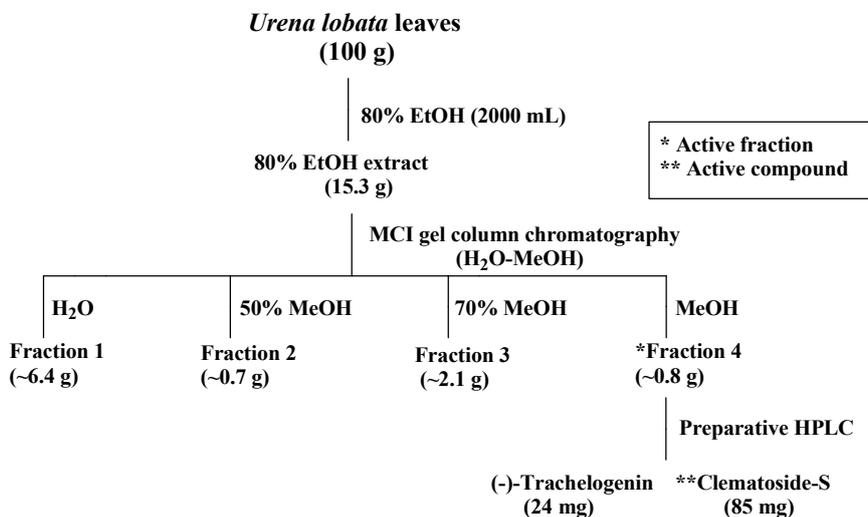
*Urena lobata* L., indigenous in China, is a member of the Malvaceae family. The plant, commonly known as Ye-Mian-Hua in China, is a popular folk medicine as diuretic, febrifuge, and also as a remedy for dysentery, cough, dropsy and rheumatism to exhibit a variety of biological activities, including antioxidant, anti-inflammatory, anti-proliferative, and antibacterial activities [7–9]. Recently, it has been reported that the methanolic extract from *U. lobata* leaves showed antibacterial activity against *Micrococcus roseus* and *Mycobacterium smegmatis* [10]. Some phytochemical compounds such as flavonoids, triglycerides and lignans have been isolated from this plant [11–13]. In the course of our ongoing program on identifying antimicrobial principles from natural materials, we found that the aqueous ethanolic extract from the leaves of *U. lobata* showed significant anti-yeast activity using *S. cerevisiae* as an indicator. In the present study, we have attempted to isolate the anti-yeast substance(s) from the leaves of *U. lobata*. On bioassay guided fractionation of aqueous ethanolic extract, further work led to the isolation of (–)-trachelogenin (**1**) and clematoside-S (**2**) (Figure 1). Herein, we describe the isolation and structural elucidation of compounds **1–2**, together with evaluating their antibacterial and antifungal activities.

## 2. Results and Discussion

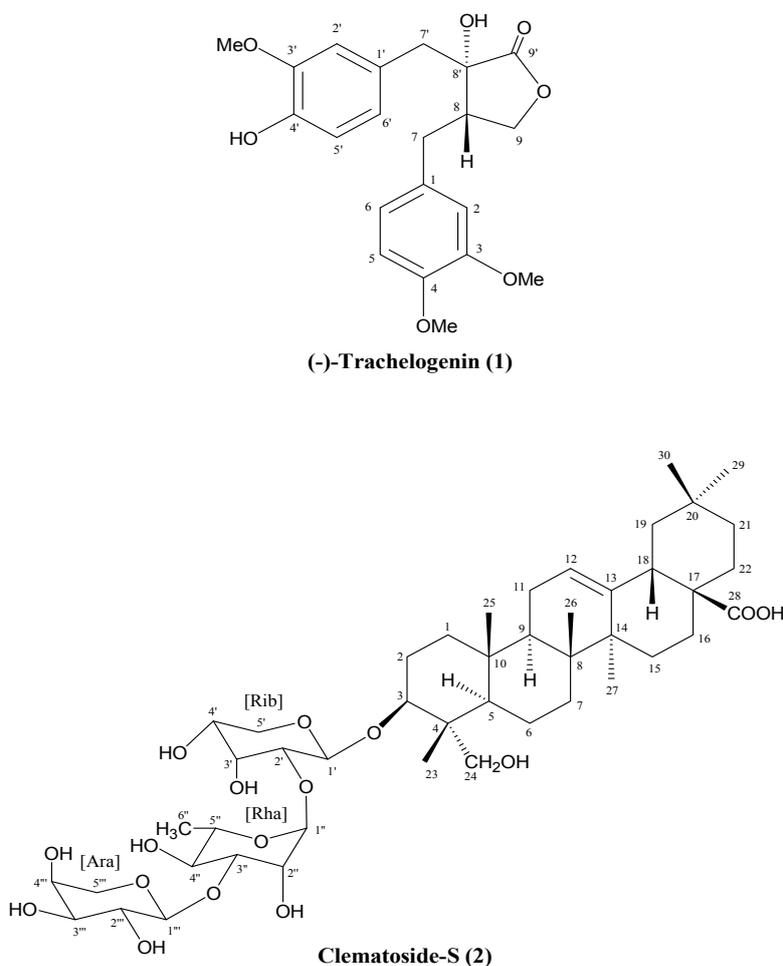
### 2.1. Isolation of Compounds **1–2** from the Leaves of *Urena lobata*

Scheme 1 shows the extract and isolation scheme of compounds **1–2** from the leaves of *U. lobata*. Powdered leaves of *U. lobata* were extracted with 80% ethanol and further fractionation was performed with a guidance of inhibitory zone diameter against *S. cerevisiae* (ATCC 204508). At a concentration of 10 mg/mL, the crude extract exhibited 15 mm of inhibitory zone diameter against *S. cerevisiae*. The extract was subjected to MCI gel column chromatography to give four fractions with a step

gradient elution of water-methanol. The active fraction 4 (16 mm of inhibitory zone diameter against *S. cerevisiae* at a concentration of 10 mg/mL) eluted with 100% methanol, was further purified by preparative HPLC to afford compounds **1** and **2**. This is the first report of compounds **1–2** isolated from *U. lobata*.



**Scheme 1.** Isolation scheme of (-)-trachelogenin (**1**) and clematoside-S (**2**) from *U. lobata*.



**Figure 1.** Chemical structure of (-)-trachelogenin (**1**) and clematoside-S (**2**).

## 2.2. Identification of Isolated Compounds 1–2

The chemical structures of the isolated compounds **1** and **2** were identified by spectroscopic analyses consisting of MS,  $^1\text{H-NMR}$ ,  $^{13}\text{C-NMR}$  and 2D-NMR data analyses. By comparison with literature data [14–16], compound **1** was identified as (–)-trachelogenin: pale-yellow gum; electron spray ionization-mass spectrometry (ESI-MS) (negative),  $m/z$  387.25  $[\text{M} - \text{H}]^-$ ; electron spray ionization-high resolution mass spectrometry (ESI-HRMS)  $m/z$  411.1420 (calcd. for  $\text{C}_{21}\text{H}_{24}\text{O}_7\text{Na}$ , 411.1420);  $[\alpha]_{\text{D}}^{25} -32.2^\circ$  ( $c$  0.43,  $\text{CHCl}_3$ );  $^1\text{H-NMR}$   $\delta$  (methanol- $d_4$ ) ppm ( $J$  in Hz): 2.45 (1H, m, H-8), 2.50 (1H, dd,  $J = 13.6, 4.1$ , H<sub>a</sub>-7), 2.80 (1H, dd,  $J = 13.6, 5.0$ , H<sub>b</sub>-7), 2.85 (1H, d,  $J = 13.6$ , H<sub>a</sub>-7'), 3.12 (1H, d,  $J = 13.6$ , H<sub>b</sub>-7'), 3.78 (3H, s, 3'-OMe), 3.79 (6H, s, 3 and 4-OMe), 3.97 (1H, d,  $J = 8.8$ , H<sub>a</sub>-9), 3.98 (1H, d,  $J = 8.8$ , H<sub>b</sub>-9), 6.57 (1H, dd,  $J = 8.1, 1.9$ , H-6'), 6.70 (4H, m, H-2, 2', 5' and 6), 6.85 (1H, d,  $J = 8.1$ , H-5).  $^{13}\text{C-NMR}$   $\delta$  (150 MHz, methanol- $d_4$ ) ppm: 30.82 (C-7), 40.50 (C-7'), 43.14 (C-8), 55.04 (3- and 4-OMe), 55.17 (3'-OMe), 70.40 (C-9), 76.01 (C-8'), 111.86 (C-5), 112.53 (C-2), 113.59 (C-2'), 114.69 (C-5'), 120.81 (C-6), 122.68 (C-6'), 126.82 (C-1'), 132.05 (C-1), 145.34 (C-4'), 147.46 (C-3'), 147.76 (C-4), 149.18 (C-3), 179.17 (C-9').

Compound **2** was obtained as white powder. In the FAB-MS spectra (negative mode), deprotonated molecular ion peak of **2** was observed at  $m/z$  881  $[\text{M} - \text{H}]^-$ , while positive fast atom bombardment-mass spectrometry (FAB-MS) showed a highest and base ion peak at  $m/z$  883  $[\text{M} + \text{H}]^+$ . So, these data show that the molecular weight of the active principle is 882 Da. The linked-scan spectrum in negative mode showed the fragmentations from the quasi molecular ion of  $m/z$  881  $[\text{M} - \text{H}]^-$  to  $m/z$  750  $[\text{M} + \text{H} - 131]^-$ , 732  $[\text{M} - \text{H} - 149]^-$ , 604  $[\text{M} - \text{H} - 277]^-$ , 585  $[\text{M} - \text{H} - 295]^-$ , and 471  $[\text{M} - \text{H} - 410]^-$ , which suggests the active compound is a glycoside containing a sugar moiety of the sequence of pentose-deoxyhexose-pentose. The molecular weight of the aglycone is deduced to be 472 Da.

The  $^1\text{H-NMR}$  spectrum apparently showed six tertiary and one secondary methyl signals. The  $^{13}\text{C-NMR}$  spectrum in  $\text{CD}_3\text{OD}$  showed 47 signals except for those overlapping with  $\text{CD}_3\text{OD}$  signals, two of which were assigned to those of acetic acid ( $\delta_{\text{C}}$  29.09 ppm,  $\delta_{\text{C}}$  175.76 ppm). The polarization transfer (DEPT)-135 and DEPT-90 NMR spectra enclosed one methine carbon at  $\delta_{\text{C}}$  49.17 ppm among the signals overlapping with those of  $\text{CD}_3\text{OD}$ , and showed 7 methyl, 13 methylene, 18 methine, 7 quaternary, and one carbonyl carbon signals to be assigned to the active principle. Since the active principle contains a pentose-deoxyhexose-pentose chain, 30 carbon signals comes from an aglycone, which is expected to be a triterpene. The heteronuclear single quantum coherence (HSQC) spectrum showed the signals of six tertiary and one secondary methyl carbons, the latter of which was assigned to a deoxyhexose by heteronuclear multiple-bond correlation (HMBC) spectrum analysis. The assignments of the protons and carbons are discussed in the following sessions and summarized in Table 1.

The HMQC spectrum showed that a carbinol carbon at  $\delta_{\text{C}}$  82.28 ppm, which was assigned to C-3 of an aglycone, carries a proton at  $\delta_{\text{H}}$  3.61 (H-3), and these showed long range couplings to the anomeric proton and carbon of a sugar, suggesting the linkage between the aglycone and a sugar chain. The proton at  $\delta_{\text{H}}$  3.61 ppm (H-3) showed long range couplings to methyl carbon (C-23) at  $\delta_{\text{C}}$  13.79 ppm and a quaternary carbon (C-4) at  $\delta_{\text{C}}$  43.95 ppm. The methyl proton on C-23 long-range coupled to C-4 and a primary carbinyl carbon (C-24) at  $\delta_{\text{C}}$  64.58 ppm in addition to the carbinol carbon, C-3, and a methine carbon (C-5) at  $\delta_{\text{C}}$  48.15 ppm. The methine proton at  $\delta_{\text{H}}$  1.26 ppm on C-5 showed coupling to C-6 at  $\delta_{\text{C}}$  18.80 ppm, C-10 at  $\delta_{\text{C}}$  37.61 ppm, and C-25 at  $\delta_{\text{C}}$  16.40 ppm in addition to C-4, C-23, and C-24.

The methyl proton on C-25 showed a coupling to a methylene carbon at  $\delta_C$  39.67 ppm, which was finally assigned to C-1, in addition to two quaternary carbons at  $\delta_C$  37.61 ppm (C-10) and  $\delta_C$  49.0 (C-9). The carbonyl proton at  $\delta_H$  3.61 ppm (H-3) on C-3 showed vicinal couplings to methylene protons at  $\delta_H$  1.74 ppm (H-2a,  $J = 4.5$  Hz) and  $\delta_H$  1.85 ppm (H-2b,  $J = 12.2$  Hz). These two protons are on the carbon at  $\delta_C$  26.58 ppm (C-2), showed a geminal coupling ( $J = 17.4$  Hz), and vicinal couplings to a proton at  $\delta_H$  1.60 ppm (H-1a) on the carbon at  $\delta_C$  39.67 ppm (C-1). H-1 shows a geminal coupling to H-1b at  $\delta_H$  0.98 ppm, which shows vicinal couplings to H-2a and H-2b. Taking all the information mentioned above, a partial structure (ring A) of the aglycone was determined as shown in Figure 1. Of the two olefinic carbon signals at  $\delta_C$  123.60 ppm and  $\delta_C$  145.21 ppm which are later assigned to C-12 and C-13, respectively, C-12 carries an olefinic proton at  $\delta_H$  5.23 ppm (H-12) which showed long-range couplings to carbons at  $\delta_C$  24.52 ppm (secondary),  $\delta_C$  49.0 ppm (C-9), and  $\delta_C$  42.72 ppm (tertiary, C-18). The HMQC spectrum showed that the carbon at  $\delta_C$  24.52 ppm carries protons at  $\delta_H$  1.87 ppm and  $\delta_H$  1.89 ppm, both of which showed couplings to the olefinic proton (H-12) at  $\delta_H$  5.23 ppm in the HMBC spectrum. In addition, one or both of the protons appearing at  $\delta_H$  1.87 ppm and  $\delta_H$  1.89 ppm showed couplings to both olefinic carbons. These H-H and C-H couplings suggest that the carbon at  $\delta_C$  24.52 ppm is vicinal to the carbon at  $\delta_C$  123.60 ppm (C-12) to be assigned to C-11. A long range coupling from the proton at  $\delta_H$  1.62 ppm (H-9) on the methine carbon at  $\delta_C$  49.00 ppm (C-9) to C-11 was observed. To C-9, were observed long range couplings from H-12 at  $\delta_H$  5.23 ppm and methyl protons at  $\delta_H$  0.96 ppm (H<sub>3</sub>-25) and  $\delta_H$  0.80 ppm (H<sub>3</sub>-26). H-9 couples to C-11 in addition to the carbons at  $\delta_C$  37.61 ppm (C-10),  $\delta_C$  48.15 ppm (C-5), and  $\delta_C$  40.50 ppm (C-8). These couplings enclosed the sequence of C-10/C-9/C-11/C-12/C-13, and the attachment of C-25 on C-10. The methyl protons (H<sub>3</sub>-26) shows a long range coupling to a quaternary carbon at  $\delta_C$  42.96 ppm, which was assigned to C-14, in addition to C-9, C-10, and the methylene carbon at  $\delta_C$  33.39 ppm assigned to C-8 through HMQC and  $H^1-H^1$  COSY spectrum analyses. Based on the above information, the partial structure constructing rings A, B, and C appeared, and hederagenin came out as the candidate for the aglycone. Comparing our NMR data with those in literatures [17–19], the aglycone was unambiguously identified to be hederagenin.

The linked scan MS suggests the sequence of pentosyl-deoxyhexosyl-pentosyl chain. The D configurations of the arabinose and ribose and the L configuration of rhamnose were established after hydrolysis of **2** followed by GC analysis [20–22]. GC-MS analysis of the trimethylsilylates of *N*-methoximes of the sugars obtained by acid hydrolysis of **2** showed the major peaks at  $t_{RS}$  of 15.29, 15.81, and 17.18 min, respectively. The  $t_{RS}$  of these peaks were identical to those of standard silylated samples, showing that the sugar components of the active principle were D-arabinose, D-ribose, and L-rhamnose. The HMBC showed that the carbonyl proton (H-3) at  $\delta_H$  3.61 ppm long-range coupled to an anomeric carbon (C-1') at  $\delta_C$  104.63 ppm, and the anomeric proton (1'-H) at  $\delta_H$  4.51 ppm to C-3 of hederagenin at  $\delta_C$  82.28 ppm. The  $J$  value between H-1' and H-2' ( $\delta_H$  3.69 ppm) on the carbon (C-2') at  $\delta_C$  76.33 ppm was 6.0 Hz, suggesting the axial-equatorial relation of the two protons, and all signals of the protons and the carbons of the pentose directly attached to hederagenin were assigned as shown in Figure 1 and Table 1. The C-2' and C-3' were distinguished by the long-range coupling from H-5a' and H-5b' to the carbon at  $\delta_C$  74.21 ppm (C-3'). Based on the chemical shifts of all signals and the  $J$  value (6.0 Hz) between H-1' and H-2', the pentose was determined to be  $\alpha$ -D-ribose (Figure 1). Similarly, all the signals of  $\alpha$ -L-rhamnose were assigned as shown in Table 1 and the ether linkage from C-2' of

ribose to C-1' of rhamnose was determined by the long-range coupling from H-2' of ribose to C-1'' of rhamnose and that from H-1'' of rhamnose to C-2' of ribose. The third sugar, arabinose, was determined to be  $\beta$ -anomer based on the  $J$  value (4.3 Hz) between H-1''' and H-2''', and all signals of  $\beta$ -arabinose were assigned as shown Table 1. The long-range coupling from H-3'' to C-1''' and that H-1''' to C-3'' showed the ether linkage between C-3'' of rhamnose and C-1''' of arabinose. Finally, the chemical structure of **2** was determined to be  $\beta$ -D-arabinosyl-(1-3)- $\alpha$ -L-rhamnosyl-(1-2)- $\alpha$ -D-riboseyl-(1-3 $\beta$ )-hederagenin, namely clematoside-S.

**Table 1.** The NMR spectroscopic data for clematoside-S (**2**) in MeOH- $d_4$ .

Position	$\delta_C$	$\delta_H$ (Mult., $J$ in Hz)	HMBC
1	39.67	0.98 (overlap), 1.60 (m)	-
2	26.58	1.74 (m), 1.85 (m)	-
3	82.28	3.61 (m)	C-4, C-23, C-1'
4	43.95	-	-
5	48.15	1.26 (m)	C-4, C-10, C-23, C-25
6	18.80	1.36 (m), 1.51 (m)	-
7	33.39	1.26 (m), 1.63 (m)	-
8	40.50	-	-
9	49.00	1.62 (m)	C-5, C-8, C-10, C-11
10	37.61	-	-
11	24.52	1.87 (m), 1.89 (m)	C-12, C-13
12	123.60	5.23 (m)	-
13	145.21	-	-
14	42.96	-	-
15	28.82	1.07 (m), 1.76 (m)	-
16	20.05	1.61 (m), 2.01 (m)	C-28
17	47.63	-	-
18	42.72	2.84 (m)	C-12, C-13, C-17, C-28
19	47.22	1.12 (m), 1.69 (m)	-
20	31.59	-	-
21	34.88	1.20 (m), 1.39 (m)	-
22	33.81	1.53 (m), 1.74 (m)	-
23	13.79	0.70 (s)	C-3, C-4, C-5, C-24
24	43.95	3.37 (m), 3.53 (m)	C-4, C-23
25	16.40	0.96 (s)	C-1, C-9, C-10
26	17.75	0.80 (s)	C-7, C-8, C-9, C-14
27	26.46	1.17 (s)	C-13, C-14, C-15
28	181.88	-	-
29	33.57	0.90 (s)	C-19, C-20, C-21, C-30
30	23.97	0.93 (s)	C-19, C-20, C-21, C-29
1'	104.63	4.51 (d, $J = 6.0$ )	C-3, C-5'
2'	76.33	3.69 (m)	C-1', C-3'
3'	74.21	3.69 (m)	C-1', C-2'
4'	69.64	3.75 (dd, $J = 13.8, 6.6$ )	-
5'	65.44	3.5 (d, $J = 6.6$ ), 3.83 (d, $J = 6.6$ )	C-1', C-3'

Table 1. Cont.

Position	$\delta_C$	$\delta_H$ (Mult., $J$ in Hz)	HMBC
1"	101.52	5.21 (d, $J = 1.6$ )	C-2", C-2"', C-3", C-5"
2"	71.73	4.05 (d, $J = 2.4$ )	C-3", C-4"
3"	80.70	3.83 (d, $J = 13.8$ )	C-4", C-1'''
4"	72.96	3.83 (d, $J = 13.8$ )	C-3", C-5"
5"	70.33	3.90 (d, $J = 7.2$ )	-
6"	17.99	1.23 (m)	C-5"
1'''	104.18	4.99 (d, $J = 4.3$ )	C-3", C-3'''
2'''	72.60	3.67 (d, $J = 3.2$ )	C-1''', C-3'''
3'''	68.66	3.96 (d, $J = 3.2$ )	C-1'''
4'''	70.17	3.76 (dd, $J = 13.2, 6.6$ )	C-1''', C-4'''
5'''	65.13	3.67 (d, $J = 13.2$ ), 3.89 (d, $J = 13.2$ )	-

### 2.3. Antimicrobial Activity of Isolated Active Compound

The antimicrobial activity of the 80% ethanol extract from *U. lobata* leaves was evaluated using Oxford plate method against five strains of food-borne bacteria (*Escherichia coli*, *Salmonella typhimurium*, *Staphylococcus aureus*, *Bacillus subtilis*, and *Bacillus laterosporus*), four strains of fungi (*Aspergillus flavus*, *Aspergillus niger*, *Rhizopus oryzae*, and *Penicillium citrinum*), and six yeast strains (*Candida albicans*, *Saccharomyces cerevisiae* ATCC 204505, *Saccharomyces cerevisiae* AY529515.1, *Saccharomyces cerevisiae* AJ746340.1, *Saccharomyces cerevisiae* JX103178.1, and *Saccharomyces boulardii* KG254081.1) as shown in Table 2. The solvent (95% methanol) used as the negative control did not show any activity. At a concentration of 10 mg/mL, the crude extract showed no antibacterial activity. Meanwhile, the extract was inactive against three species of fungi (*A. flavus*, *R. oryzae*, and *P. citrinum*) even at 10 mg/mL. Among the test fungi, the most sensitive strain was *A. niger* with the diameter of inhibition zone of 9 mm. The extract showed no activity (10 mg/mL) against *C. albicans*. It was worth noting that the extract exhibited remarkable anti-yeast activities against *S. cerevisiae* ATCC 204505, *S. cerevisiae* AY529515.1, *S. cerevisiae* AJ746340.1, *S. cerevisiae* JX103178.1, and *S. boulardii* KG254081.1 with the diameter of the inhibition zone in the range from 14 to 17 mm. The result suggest that *U. lobata* leaves inhibits selectively the growth of some yeast strains. So, we selected out *S. cerevisiae* ATCC 204505 as the indicator for detecting the main anti-yeast substance(s) in *U. lobata* leaves. Using a combination of chromatographic methods, (-)-trachelogenin (**1**) and clematoside-S (**2**) were isolated from the 80% ethanol extract of *U. lobata* leaves. It was found that **1** showed no activity against all of the selected microorganisms in the antimicrobial assay (Table 2). However, **2** showed promising activity against *S. cerevisiae* ATCC 204505 as compared to standard anti-yeast reagent, streptomycin. In addition, **2** exhibited a potent inhibitory effect against the five test yeast strains, except for *C. albicans*, with diameter of inhibition zones in the range from 11 to 20 mm.

Further study was carried out to investigate the anti-yeast effect of **2** against *S. cerevisiae* ATCC 204505, *S. cerevisiae* AY529515.1, *S. cerevisiae* AJ746340.1, *S. cerevisiae* JX103178.1, and *S. boulardii* KG254081.1 by measuring the minimum inhibitory concentration (MIC) and the minimum bactericidal concentration (MBC). The MICs and MBCs of **2** to the five test yeast strains

were shown in Table 3. These results demonstrated that **2** had certain antibacterial and bactericidal property. In general, the MICs of **2** against the test yeasts were in range from 0.61 to 9.80 µg/mL, and MBCs from 2.42 to 9.80 µg/mL, respectively. Associated with the results of disc diameter of inhibition zone in Table 2, it was clearly indicated that **2** showed the strongest activity against *S. cerevisiae* ATCC 204505, whereas it showed the moderate activity against the other yeast strains (Table 3). clematoside-S (**2**) isolated from clematotic species has been reported to exhibit cytotoxic activity against several cancer cells [23]. However, compound **2** has never been studied for antimicrobial activity. Our finding is the first report on the isolation of **2** with anti-yeast activity from *U. lobata*. Therefore, it may be proposed that *U. lobata* can be used as a natural anti-yeast agent to control food spoilage caused by yeast. However, further study is warranted to provide clear evidence for toxicity profile.

**Table 2.** Antimicrobial activity of *U. lobata* extract, Fra. 4, (–)-trachelogenin (**1**) and clematoside-S (**2**).

Microorganisms Strain	Inhibition Zone (mm)					
	Extract <sup>a</sup>	Fra. 4 <sup>b</sup>	1 <sup>c</sup>	2 <sup>c</sup>	Negative Control <sup>d</sup>	Positive Control <sup>e</sup>
<b>Gram negative bacteria</b>						
<i>Escherichia coli</i> ATCC 25922	0	NT <sup>f</sup>	0	0	0	35
<i>Salmonella typhimurium</i> ATCC 14028	0	NT	0	0	0	28
<b>Gram positive bacteria</b>						
<i>Staphylococcus aureus</i> ATCC 25923	0	NT	0	0	0	28
<i>Bacillus subtilis</i> ATCC 21216	0	NT	0	0	0	18
<i>Bacillus cereus</i> ATCC 10231	0	NT	0	0	0	15
<i>Bacillus laterosporus</i> ATCC 64	0	NT	0	0	0	21
<b>Fungi</b>						
<i>Aspergillus flavus</i> ATCC 204304	0	NT	0	0	0	25
<i>Aspergillus niger</i> ATCC 16404	9	NT	0	9	0	20
<i>Rhizopus oryzae</i> ATCC 9363	0	NT	0	0	0	28
<i>Penicillium citrinum</i> ATCC 14994	0	NT	0	0	0	26
<b>Yeasts <sup>g</sup></b>						
<i>Candida albicans</i> ATCC 50013	0	NT	0	0	0	0
<i>Saccharomyces cerevisiae</i> ATCC 204508	15	16	0	20	0	21
<i>Saccharomyces cerevisiae</i> AY529515.1	17	NT	0	11	0	21
<i>Saccharomyces cerevisiae</i> AJ746340.1	14	NT	0	12	0	21
<i>Saccharomyces cerevisiae</i> JX103178.1	14	NT	0	12	0	21
<i>Saccharomyces boulardii</i> KG254081.1	17	NT	0	13	0	21

<sup>a</sup> The concentration of the 80% ethanol extract from *U. lobata* leaves was 2 mg/disk; <sup>b</sup> Fra. 4 (fraction 4) was obtained due to Scheme 1; <sup>c</sup> The concentration of compounds **1–2** was 2 mg/disk; <sup>d</sup> 95% methanol as the negative control; <sup>e</sup> Penicillin (2 mg/disk) as the positive control against bacteria and streptomycin (2 mg/disk) as the positive control against fungi and yeasts; <sup>f</sup> Not tested; <sup>g</sup> *Saccharomyces cerevisiae* AY529515.1, *Saccharomyces cerevisiae* AJ746340.1, *Saccharomyces cerevisiae* JX103178.1, and *Saccharomyces boulardii* KG254081.1 were isolated from spoiled grapes.

**Table 3.** MIC<sup>a</sup> and MBC<sup>b</sup> of clematoside-S (2) for different yeast strains.

Yeast Strains	MIC (µg/mL)	MBC (µg/mL)
<i>Saccharomyces cerevisiae</i> ATCC 204505	0.61	2.42
<i>Saccharomyces cerevisiae</i> \AY529515.1	1.21	4.84
<i>Saccharomyces cerevisiae</i> \AJ746340.1	9.80	9.80
<i>Saccharomyces cerevisiae</i> \JX103178.1	1.21	4.84
<i>Saccharomyces boulardii</i> \KG254081.1	2.42	9.80

<sup>a</sup> Minimum inhibitory concentration (MIC); <sup>b</sup> Minimum bactericidal concentration (MBC).

### 3. Experimental

#### 3.1. General Procedure

ESIMS spectra were acquired on a Thermo Finnigan TSQ Quantum Ultra AM mass spectrometer system with an electrospray source operating in both positive and negative ion modes (Thermo Electron, San Jose, CA, USA). Fast atom bombardment (FAB) MS was obtained with a Jeol JMS-SX102A instrument (JEOL, Tokyo, Japan). Optical rotation values were measured on JASCO P-1020 polarimeter (JASCO, Tokyo, Japan). NMR spectra were recorded with a Bruker AV II-600 instrument (<sup>1</sup>H, 600 MHz; <sup>13</sup>C, 125 MHz) (Bruker Co., Karlsruhe, Germany). Column chromatography was performed with MCI gel 100A (75–150 µm, Sci-Bio-Chem Co., Ltd., Chengdu, China). The HPLC system (Agilent 1200 Series Purification System, Agilent, Santa Clara, CA, USA) consisted of an injector (G1328B), a column oven (35 °C), a pump (G1311A), a diode array detector (G1315D), and an Inertsil PREP-ODS column (6 × 250 mm i.d. with a particle size of 5 µm, GL-Science, Tokyo, Japan). The solvents used were all of HPLC-grade for HPLC analysis.

#### 3.2. Plant Material and Regents

The leaves of *U. lobata* used in this study were purchased from Chengdu Medicinal Materials (Chengdu, China) and properly identified at the Department of Pharmacology, Hua Xi Medicinal Center of Sichuan University, China. A voucher specimen is deposited in the Rice Research Institute, Sichuan Agricultural University (No. 20090603). Penicillin and streptomycin were obtained from Sichuan Changwei Pharmaceutical Co., Ltd. (Chengdu, China). Agar, beef extract, sucrose and peptone were purchased from Chengdu Best Reagents Co., Ltd. (Chengdu, China). All other reagents used were of analytical grade.

#### 3.3. Extraction and Isolation of the Active Compound

The dried leaves of *U. lobata* (100 g) was crushed into powders with a mixer, followed by extraction with 80% ethanol (2000 mL) in total for 3 day at room temperature and filtered. Evaporation of the solvent under reduced pressure yielded the 80% ethanol extract (15.3 g), which was applied on a MCI gel column (3.5 × 20 cm) with a water-methanol gradient to give four fractions (fraction 1, water (800 mL, ~6.4 g); fraction 2, 50% methanol in water (800 mL, 0.7 g); fraction 3, 70% methanol in water (800 mL, 2.1 g); fraction 4, 100% methanol (600 mL, 0.8 g)). Consequently, fraction 4 was further purified by preparative HPLC (column, Inertsil ODS 6 × 250 mm i.d. with a particle size of

5  $\mu\text{m}$ , GL-Science; mobile phase, water-methanol = 20:80; flow rate, 1 mL/min; detection, UV 220 nm) to yield 24 mg of (–)-trachelogenin (**1**,  $t_{\text{R}}$  12.8 min) and 85 mg of clematoside S (**2**,  $t_{\text{R}}$  15.4 min).

### 3.4. Acid Hydrolysis and GC-MS of Sugars

A part (2.1 mg) of **2** was dissolved in 2 M HCl (5% MeOH, 1 mL) and heated at 95 °C for 90 min. After the hydrolysis, EtOAc (2 mL) was added to the solution and vortexed to remove aglycone. This extraction was repeated twice. The aqueous residue was divided into two microtubes, and subjected to a centrifuge-concentration *in vacuo*. The dried concentrate was converted to *N*-methoximes in 50  $\mu\text{L}$  of pyridine containing methoxyamine hydrochloride (15 mg/mL) overnight at room temperature. To this reaction mixture, 30 mL of MSTFA (*N*-methyl-*N*-trimethylsilyltrifluoroacetamide) was added and heated at 45 °C for 20 min for trimethylsilylation. The trimethylsilylated *N*-methoxime sugars were submitted to GC-MS analysis under the following condition: column, DB-1 (0.25 mm i.d.  $\times$  25 m, 0.25  $\mu\text{m}$ ); injector temperature, 250 °C; carrier gas, He (0.9 mL/min); oven temperature program, 70 °C (1 min), 70–150 °C (20 °C/min), 150–180 °C (2 °C/min); ionization, EI (70 eV).

### 3.5. Antimicrobial Activity

The tested microorganisms contained Gram-negative bacteria (*Escherichia coli* ATCC 25922, *Salmonella typhimurium* ATCC 14028), Gram-positive bacteria (*Staphylococcus aureus* ATCC 25923, *Bacillus subtilis* ATCC 21216, *Bacillus cereus* ATCC 10231, *Bacillus laterosporus* ATCC 64), fungi (*Aspergillus niger* ATCC 16404, *Aspergillus flavus* ATCC 204304, *Penicillium citrinum* ATCC 14994, and *Rhizopus oryzae* ATCC 9363) and yeasts (*Candida albicans* ATCC 50013, *Saccharomyces cerevisiae* ATCC 204508, *Saccharomyces cerevisiae* AY529515.1, *Saccharomyces cerevisiae* AJ746340.1, *Saccharomyces cerevisiae* JX103178.1, and *Saccharomyces boulardii* KG254081.1), and were obtained from the Key Laboratory of Food Science and Technology of Sichuan Province, Sichuan University. The yeasts (*Saccharomyces cerevisiae* AY529515.1, *Saccharomyces cerevisiae* AJ746340.1, *Saccharomyces cerevisiae* JX103178.1, and *Saccharomyces boulardii* KG254081.1) were isolated from spoiled grapes and identified by morphology, biochemical tests and ITS sequence analysis. All the microorganisms were maintained on nutrient agar at 4 °C and were sub-cultured every month in our laboratory. In the present test, nutrient agar culture medium was for bacteria and the medium of potatoes was for fungi and yeasts strains.

Antimicrobial activity of the test sample was determined by Oxford plate method [24]. In short, bacterial cultures were diluted to obtain a bacterial suspension of  $10^6$  CFU/mL with sterile water. Petri plates containing 20 mL of nutrient agar were inoculated with 0.2 mL of bacterial culture and were allowed to dry in sterile chamber. The Oxford plates (6 mm in diameter) were impregnated with 0.1 mL of test sample in 95% methanol and placed on the inoculated agar. Penicillin and streptomycin were used as the positive control for bacteria and fungus, respectively. The inoculated plates of bacteria were incubated at 37 °C for 24 h, and fungi were incubated at 35–37 °C for 48 h. The antimicrobial activity was evaluated by measuring the zone of inhibition against the test organisms.

The minimum inhibitory concentration (MIC) of the test sample was evaluated for the yeast strains which were determined by the method of broth dilution [25]. An aliquot of 2 mL of the medium of potatoes was placed into each tube, and all tubes were autoclaved at 121 °C. Test sample (filtered,

0.22 µm) was added to the tubes to keep the final concentrations ranging from 0.06 to 40 µg/mL. The test yeast suspension was added into to the inoculum size of 10<sup>6</sup> CFU/mL. Then, the inoculated tubes were incubated at 37 °C for 18–24 h. The MIC was evaluated by measuring the turbidity of inoculated culture media. Another liquid medium without adding any yeast was prepared as the negative control. The minimum inhibitory concentration at which no microorganism grew in the culture media was defined as the value of MIC. The minimum bactericidal concentration (MBC) of the test sample was determined according to the MIC values. The sample showing no increases in turbidity were streaked on potatoes medium and incubated at 37 °C for 18–24 h. The lowest concentration of the test sample where was no viable yeasts was identified as the value of MBC.

#### 4. Conclusions

In the present study, (–)-trachelogenin (**1**) and clematoside-S (**2**) were isolated from the leaves of *U. lobata* for the first time by a combination of chromatographic methods and their chemical structure was identified by MS and extensive NMR data analysis. Furthermore, to the best of our knowledge, this is the first study demonstrating the antimicrobial activity of clematoside-S. In addition, clematoside-S shows a specific anti-yeast activity against *S. cerevisiae* strains, which are one of the most significant spoilage yeasts for juice and soft drinks.

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#### Author Contributions

Xiao-Ling Gao, Ying Liao, Jie Wang, and Xiao-Yan Liu performed the experiments. Kai Zhong, Yi-Na Huang, and Hong Gao analyzed the data. Bo Gao and Zheng-Jun Xu designed the experiments.

#### Conflicts of Interest

The authors declare no conflict of interest.

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