

Short Note

## 2-(2-Imino-1-methylimidazolidin-4-ylidene)hydrazinecarbothioamide

Ahmed A. Al-Amiery <sup>1,2,\*</sup>, Abdul Amir H. Kadhum <sup>1</sup> and Abu Bakar Mohamad <sup>1</sup>

<sup>1</sup> Department of Chemical & Process Engineering, Universiti Kebangsaan Malaysia, 43600 Bangi Selangor, Malaysia

<sup>2</sup> Applied Chemistry division, Applied Science Department, University of Technology (UOT), Baghdad, Iraq

\* Author to whom correspondence should be addressed; E-Mail: dr.ahmed1975@gmail.com; Tel.: +6-0-192-903-670.

Received: 21 May 2012 / Accepted: 19 June 2012 / Published: 3 July 2012

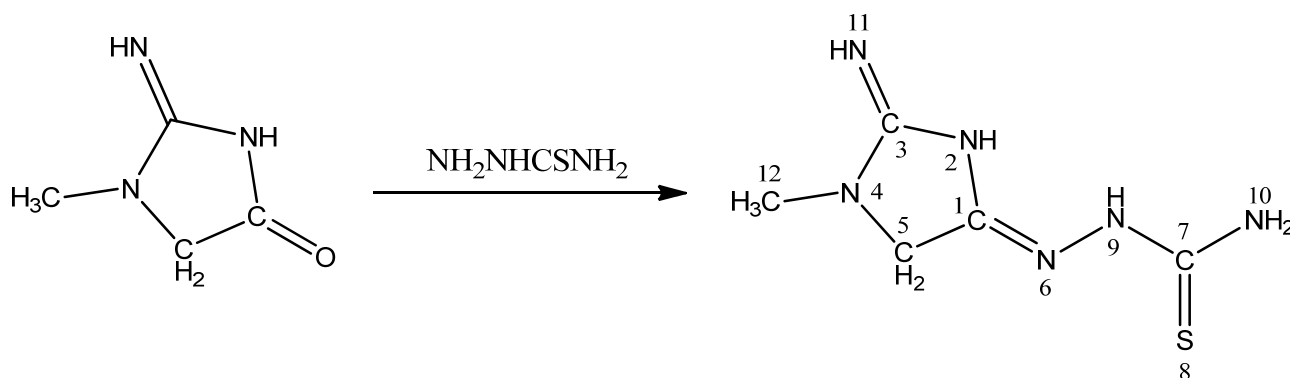
---

**Abstract:** A new thiosemicarbazone, 1-methyl-2-imino-N-(methanethialdiamine)-yl-4-iminoimidazolidin was synthesized and its UV-VIS, IR, and NMR spectroscopic data and CHN analysis are presented.

**Keywords:** 2-(2-imino-1-methylimidazolidin-4-ylidene)hydrazinecarbothioamide; thiosemicarbazide; thiosemicarbazones

---

The chemistry of thiosemicarbazones has received considerable attention because of their variable bonding modes, promising biological implications, structural diversity, and ion-sensing ability [1,2]. They have been used as drugs and are reported to possess a wide variety of biological activities against bacteria, fungi, and certain type of tumors, and they are also a useful model for bioinorganic processes [3]. In continuation of previous studies [4–14], we have focused on synthesis of new heterocyclic compounds, and herein we are reporting the synthesis of 1-methyl-2-imino-N-(methanethialdiamine)-yl-4-iminoimidazolidin as new molecule (Scheme 1).

**Scheme 1.** Synthesis of 1-methyl-2-imino-N-(methanethialdiamine)-yl-4-iminoimidazolidin.

## Experimental

### Synthesis of 2-(2-Imino-1-methylimidazolidin-4-ylidene)hydrazinecarbothioamide

A mixture of 1-methyl-4-oxo-2-iminoimidazolidin (2.26 g, 0.02 mol) and thiosemicarbazide (1.82 g, 0.02 mol) in 100 mL of ethanol was refluxed for 3 h. The solvent was evaporated on a rotary evaporator. The title compound was washed with cold ethanol, and dried under vacuum over P<sub>4</sub>O<sub>10</sub>. Yield 70%, (light brown) [7].

Melting point: 153 °C.

UV-VIS in DMF 255 and 322 nm.

FT-IR spectroscopy; 3421 cm<sup>-1</sup> (N-H stretching vibrations, NH<sub>2</sub>); 1631 cm<sup>-1</sup> (C=N) and 1618 cm<sup>-1</sup> (C=N).

<sup>1</sup>H-NMR (300 MHz, DMSO-*d*<sub>6</sub>): 1.80 (s, 1H, NH), 2.20 (s, 3H, CH<sub>3</sub>), 2.70 (s, 2H, CH<sub>2</sub>), 8.00 (s, 1H, NH), 9.10 (s, 1H, NH) and 10.90 (s, 1H, NH).

Elemental analysis: C, 32.25 (31.91); H, 5.41 (5.11); N, 45.13 (44.74).

## Acknowledgements

This study was supported by University of Technology, Baghdad, Iraq and Universiti Kebangsaan Malaysia under the DIP-2012-02 grant.

## References

- Casas, J.S.; García-Tasende, M.S.; Sordo, J. Corrigendum to Main group metal complexes of semicarbazones and thiosemicarbazones. A structural review. *Coord. Chem. Rev.* **2000**, 209, 197–261.
- Mishra, D.; Naskar, S.; Drew, M.G.B.; Chattopadhyay, S.K. Synthesis, spectroscopic and redox properties of some ruthenium(II) thiosemicarbazone complexes: Structural description of four of these complexes. *Inorganica Chim. Acta* **2006**, 359, 585–592.

3. Al-Amiery, A.A.; Al-Majedy, Y.K.; Abdulreazak, H.; Abood, H. Synthesis, Characterization, Theoretical Crystal Structure, and Antibacterial Activities of Some Transition Metal Complexes of the Thiosemicarbazone. *Bioinorg. Chem. Appl.* **2011**, *2011*, doi:10.1155/2011/483101.
4. Kadhum, A.A.H.; Mohamad, A.B.; Al-Amiery, A.A.; Takriff, M.S. Antimicrobial and antioxidant activities of new metal complexes derived from 3-aminocoumarin. *Molecules* **2011**, *16*, 6969–6984.
5. Al-Amiery, A.A.; Al-Bayati, R.I.H.; Saour, K.Y.; Radi, M.F. Cytotoxicity, antioxidant and antimicrobial activities of novel 2-quinolone derivatives derived from coumarins. *Res. Chem. Intermediat.* **2011**, *38*, 559–569.
6. Al-Amiery, A.A.; Musa, A.Y.; Kadhum, A.H.; Mohamad, A.B. The use of umbelliferone in the synthesis of new heterocyclic compounds. *Molecules*, **2011**, *16*, 6833–6843.
7. Kadhum, A.A.H.; Al-Amiery, A.A.; Musa, A.Y.; Mohamad, A.B. The Antioxidant Activity of New Coumarin Derivatives. *Int. J. Mol. Sci.* **2011**, *12*, 5747–5761.
8. Kadhum, A.A.H.; Al-Amiery, A.A.; Shikara, M.; Mohamad, A. Synthesis, Structure elucidation and DFT studies of New thiadiazoles. *Int. J. Phys. Sci.* **2011**, *6*, 6692–6697.
9. Al-Amiery, A.A.; Al-Majedy, Y.K.; Ibrahim, H.H.; Al-Tamimi, A.A. Antioxidant, antimicrobial, and theoretical studies of the thiosemicarbazone derivative Schiff base 2-(2-imino-1-methylimidazolidin-4-ylidene) hydrazinecarbothioamide (IMHC). *Org. Med. Chem. Lett.* **2012**, *2*, doi:10.1186/2191-2858-2-4.
10. Al-Amiery, A.A.; Kadhum, A.A.H.; Mohamad, A.A. Antifungal and Antioxidant Activities of Pyrrolidonethiosemicarbazone Complexes. *Bioinorg. Chem. Appl.* **2012**, *2012*, 1–5.
11. Al-Amiery, A.A. Synthesis and antioxidant, antimicrobial evaluation, DFT studies of novel metal complexes derivate from Schiff base. *Res. Chem. Intermediat.* **2012**, *38*, 745–759.
12. Al-Amiery, A.A. Antimicrobial and Antioxidant Activities of New Metal Complexes Derived from (E)-3-((5-phenyl-1,3,4-oxadiazol-2-ylimino)methyl)naphthalen-2-ol. *Med. Chem. Res.* **2011**, doi:10.1007/s00044-011-9880-1.
13. Al-Amiery, A.A.; Kadhum, A.A.H.; Mohamad, A.A. Antifungal Activities of New Coumarins. *Molecules* **2012**, *17*, 5713–5723.
14. Junaedi, S.; Kadhum, A.A.H.; Al-Amiery, A.A.; Mohamad, A.A.; Takriff, M.S. Synthesis and Characterization of Novel Corrosion Inhibitor Derived from Oleic Acid: 2-Amino 5-Oleyl-1,3,4-Thiadiazol (AOT). *Int. J. Electrochem. Sci.* **2012**, *7*, 3543–3554.