

Article

Capacity of Aromatic Compound Degradation by Bacteria from Amazon Dark Earth

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Abstract: Amazon dark earth (ADE) is known for its high organic matter content, biochar concentration and microbial diversity. The biochar amount suggests the existence of microorganisms capable of degrading aromatic hydrocarbons (AHs). In an effort to investigate the influence of bacteria on the resilience and fertility of these soils, we enriched five ADE soils with naphthalene and phenanthrene, and biodegradation assays with phenanthrene and diesel oil were carried out, as well. After DNA extraction, amplification and sequencing of the 16S rRNA bacterial gene, we identified 148 isolates as the *Proteobacteria*, *Firmicutes* and *Actinobacteria* phyla comprising genera closely related to AHs biodegradation. We obtained 128 isolates that degrade diesel oil and 115 isolates that degrade phenanthrene. Some isolates were successful in degrading both substrates within 2 h. In conclusion, the obtained isolates from ADE have degrading aromatic compound activity, and perhaps, the biochar content has a high influence on this.

Keywords: Terra Preta de Índio; microbial ecology; genetic diversity; phenanthrene; diesel oil; biodegradation assay

1. Introduction

Amazon dark earth (ADE) has a stable micro-ecosystem that sustains itself [1] (There is a great presence of vegetable, animal, ceramic, biochar and ash residues and existing high aromatic hydrocarbon concentrations. Biochar, as the black material of carbon associated with agriculture is named, is the most recalcitrant portion of organic matter from partial lignin burning, hemicellulose and cellulose and is rich in polyaromatic hydrocarbon-condense units [2]. Because of its porosity, biochar fragments can retain water, can be protective structures [3] and can be carbon sources for microorganisms [4]. All of these characteristics indicate that the biological activity strongly influences the nutrient cycles of this environment.

Aromatic hydrocarbons (AHs) are a class of hydrophobic organic compounds made of casted aromatic rings with linear and angular molecular arrangements, and they are found in the environment due to the processing of wood, char, creosote, petroleum mud, asphalt and pesticides [5]. These substances can be bioactive, but persist in ecosystems for years, due to the low solubility in water and to the adsorption on solid particles [6].

Biodegradation by microorganisms is an adaptation mechanism for the survival of the microbial community, since gene transference and mutation are essential for the metabolism of different carbon compounds [7,8]. Thus, much research has focused on the biological degradation of AHs through metabolism and co-metabolism [9]. Sayler *et al.* [10] demonstrated the positive correlation between the AH mineralization rates on oil-contaminated sediments and bacterial colony number with its own plasmids with genes that code for the degradation of AH enzymes.

AH degradation by bacteria is due to genes from plasmids or nuclear DNA and sometimes both [11]. These genes jointly act to produce degrader phenotypes for a determined substrate and allow its utilization as a carbon and energy source [12]. The degradation process involves enzyme machinery: dehalogenases, dehydrogenases and hydrolases system [13].

Amongst the several microorganisms that are found on ADE, the most abundant bacteria are from the phyla, *Firmicutes*, *Proteobacteria*, *Verrucomicrobia*, *Acidobacteria* and *Actinobacteria* [14,15]. Some members have an infinite number of beneficial characteristics for ecology and the economy, being biodegraders, as well. Thus, detailed studies with these genera have become necessary to discover new strategies for biotechnological, agricultural and pharmaceutical uses.

Studies by isolation in association with molecular analyses comprise the best way to analyze the physiology and phylogeny of an organism. Furthermore, bioassays are crucial to observing the activity efficiency by means of controlled conditions with laboratory evaluations [16].

2. Experimental Section

The study areas were from Central Amazon with distinct land uses and vegetal covering. The Balbina area (BAL) (01°30'26,4" S–60°05'34" W), located at Presidente Figueiredo City (Amazonas State, AM), has medium secondary rainforest (capoeira) coverage and is considered a model for the original ADE, since it has not been cultivated or deforested for more than 20 years. The area within the Caldeirão Experimental Station (CAL) (Embrapa Centro de Pesquisa Agroflorestal da Amazônia Ocidental, CPAA) (03°15'11" S–60°13'43" W) is located at Iranduba City (AM), also under *capoeira*.

The Barro Branco area (BB) (03°18'24.76" S–60°32'05.10" W) is located next to Iranduba City (AM), under *Citrus* culture. The Costa do Açutuba (ACU) (03°05'53.92"S–60°21'19.90"W) and Hatahara (HATA) (03°16'28.45" S–60°12'17.14" W) areas are ADE sites that have been extensively analyzed by pedo-archaeological expeditions, being utilized for intentional anthropogenic activities based on intensive management on bare places [17].

2.1. Sampling

Samples were collected from depths of 0–10 cm at a georeferenced central point and two other points 1.5 m away, and also, 3 subsamples 30 cm from each other were collected from each point. The sampling was performed in plastic tubes of 5 × 10 cm and stored in thermal boxes with dry ice, then sent to the Laboratório de Biologia Celular e Molecular (BCM) of the Center of Nuclear Energy in Agriculture, the University of São Paulo, Piracicaba-São Paulo State, Brazil. One part was stored below 4 °C for microbial and chemical analyses, while another part was stored at –20 °C for molecular analyses.

2.2. Enrichment Assay

One gram of soil in triplicate was added to mineral medium Bushnell Haas (Difco, Detroit, MI, USA) with naphthalene 99% and phenanthrene 98% (Sigma Aldrich, St. Louis, MO, USA) (diluted in ethanol 100%) at concentrations of 0.05% (w/v) [18] as carbon and energy sources. The negative control was composed by mineral medium and AHs. The positive controls were standards from the Leibniz Institute DSMZ - German Collection of Microorganisms and Cell Cultures, Braunschweig, Germany (DSMZ) culture collection within the mineral medium (Table 1). The mesocosms were incubated at 30 °C with 300 rpm shaking for 15 days. A subculture was carried out under the same conditions. Finally, the cultures were grown on solid medium previously treated with 200 µL of alcoholic solution of AHs at a concentration of 50 ppm on its surface for the attainment of bacterial colonies from serial dilutions (10³ to 10⁶). The satisfactory quantity was between 30 and 300 colonies. Later, the purified colonies were grown in liquid tryptic soy medium (Difco, Detroit, MI, USA), and the genomic DNA extraction was carried out as described by Doyle [19], followed by storage in 40% glycerol in an ultra-freezer.

Table 1. Positive standards utilized on the enrichment assay. AH, aromatic hydrocarbon.

DSMZ Standard	AH Metabolism
<i>P. putida</i> (DSM 6899)	toluene dioxygenase
<i>P. putida</i> (DSM 291)	
<i>P. fluorescens</i> (DSM 8369)	naphthalene dioxygenase
<i>P. abietaniphila</i> (DSM 6506)	

2.3. Isolates Identification

2.3.1. 16S rRNA Gene of *Bacteria* Amplification

To identify the isolates, a conventional PCR was performed with external primers fD1 (5'-AGAGT TTGATCCTGGCTCAG-3') and rD1 (5'-AAGGAGGTGATCCAGCC-3') [20]. The conditions were: 5

pmol of primers, 200 μ M of each dNTP (Life Technologies of Brazil, São Paulo, Brazil), 1 x Taq buffer (Life Technologies of Brazil, São Paulo, Brazil), 1.5 mM MgCl₂ (Life Technologies of Brazil, São Paulo, Brazil), 2 U Platinum Taq DNA polymerase (Life Technologies of Brazil, São Paulo, Brazil) and 10 ng of DNA at a final volume of 25 μ L. The reaction was initiated with 3 min of denaturation at 94 °C, followed by 30 cycles with denaturation at 94 °C for 1 min, annealing at 55 °C for 30 s, extension at 72 °C for 30 s and a final extension of 72 °C for 10 min.

2.3.2. Sequencing Reaction

To generate information about the main degrading bacterial genera from ADE, 16S rRNA gene sequencing was done. The precipitation of PCR products was achieved by the addition of isopropanol, followed by incubation, centrifugation and resuspension in 30 μ L of autoclaved ultrapure water. The reaction was performed using the DYEnamic ET Terminator Cycle Sequencing Kit (GE Healthcare, São Paulo, Brazil) in a final volume of 10 μ L, with one reaction for each primer: 200 ng of PCR product; 5 pmol of primer; 0.5X of buffer; and 2.0 μ L of DYEnamic ET Terminator Cycle. The conditions were: 2 min at 96 °C, 30 cycles at 95 °C for 20 s, annealing at 55 °C for 15 s and extension at 60 °C for 1 min. Afterwards, precipitation was carried out to eliminate the dNTPs not used with 1 μ L of sodium acetate/EDTA, 40 μ L of ethanol 95% and centrifugation at 12,000 rpm for 15 min; the supernatant was discarded, and 500 μ L of ethanol 70% was added, followed by centrifugation for 5 min, discarding of the supernatant, drying in a DNA concentrator for 5 min at 40 °C and resuspension in 10 μ L of formamide (Life Technologies of Brazil, São Paulo, Brazil). The microplate was placed in the thermocycler to denature the DNA at 96 °C for 5 min and immediately placed on ice for 2 min. The reading of the marked bases were done by the automatic sequencer, ABI Prism 3130 Genetic Analyzer (Life Technologies, Carlsbad, NM, USA), which utilizes capillary electrophoresis for the separation and detection of amplified fragments of \leq 1500 bp in length [21].

2.3.3. Phylogenetic Analysis

Reads were edited by the removal of low quality bases (<20 Phred quality) and the formation of contigs through the Phred/Phrap/Consed software [22–24]. The partial contigs of the 16S rRNA gene were analyzed on the Ribosomal Database Project (RDP) Classifier server [25] for the isolates' genera and on the EzTaxon server [26] for the isolates' species to corroborate the data. The nucleotide sequences described in this study have been submitted to the GenBank database of the National Center for Biological Information under the accession numbers, KJ524106–KJ524135 and KJ524136–KJ524271.

2.4. Biodegradation Assays

2.4.1. Cell Density

To study the equal efficiency of biodegradation by isolates, there is a need to use the same and known amount of cells on the inoculum. In this case, 10⁸–10⁹ cells per mL (cells/mL), measured by cell counting under microscope (Bel Photonics, Osasco, Brazil) with a Neubauer Chamber (Boeco, Hamburg, Germany) and spectrophotometry with Nanodrop 2000c (Thermo Scientific, West Palm Beach, FL, USA) at a wavelength of 600 nm. Due to different morphology, size and color, the isolates

were classified under these characteristics, and 20 types were obtained. One isolate of each type was utilized as a standard for the optical density with dilutions and calculations as described by Lelliott and Stead [27], a graphic relating cell number and OD for each of the 20 types of models being constructed. Then, the inoculants were prepared in nutrient broth medium (Oxoid, Basingstoke, UK) at 170 rpm for 12 h at 32 °C. Every evaluation of each OD of isolates was plotted to its correspondent graphic type beyond each graphic formula. The final volume of inoculants varied between 4 to 50 µL.

2.4.2. The Assays

To evaluate the isolates efficiency of AH biodegradation, the assay was performed as described by Hanson *et al.* [28] with the 148 isolates in triplicate in polystyrene 96-well microplates, together with positive and negative controls, also in triplicate. This assay is based on the redox indicator 2,6-dichlorofenol-indofenol (DCPIP) principle of discoloration when electrons are released in the environment by the bacterial cells when they use the substrates. The molecular conformation of the indicator, then, changes and reflects the light differently, turning its coloration from blue (oxidized form) to colorless (reduced form). The redox indicator used was 2,6-dichlorofenol-indofenol (DCPIP) (Sigma Aldrich, St Louis, MO, USA) (diluted in ultrapure water and filtered through a 0.22-µm membrane) at 0.2% in the final volume. The substrates used were phenanthrene (diluted in acetone and filtered through a 0.22-µm membrane), as a strictly controlled assay, and commercial diesel oil (Shell, Piracicaba, Brazil) (filtered through a 0.22-µm membrane), as a possible practical use. The positive control used was DSM11192 *Gordonia* sp., as described in Table 2. The negative controls consisted only of mineral medium, substrates, the redox indicator and Nutriente Broth (NB) medium without inoculants.

Table 2. Positive standard utilized in the bioassays.

No. DSMZ	Microorganism	AH metabolism
DSM 11192	<i>Gordonia</i> sp.	n-alkane, naphthalene, toluene, m-xylene, p-xylene

A pretest was done with the positive control with phenanthrene concentrations of 0.05%, 0.1%, 0.2%, 0.4%, 0.8% and 1% in the final volume, and with diesel oil concentrations of 0.5%, 1%, 2%, 4% 8% and 16% in the final volume. The best substrates concentrations, or the minimum inhibition, for the microorganism were 0.05% of phenanthrene and 8% of diesel oil, and the degradation activity was observed to begin by 8 h for diesel oil and 24 h for phenanthrene by DSM11192. The concentrations of 0.1% and 0.2% of phenanthrene inhibited DSM11192; although, the maximum concentration of diesel tolerated was 16%.

Finally, after the volume of cells, substrate and redox indicator, a mineral medium Bushnell Haas (BH) (Difco, Detroit, MI, USA) was added to complete the 200 µL. The microplates were maintained in dark at room temperature (18 °C) and orbital shaking of 170 rpm, and measurements were made at 2 h, 4 h, 8 h, 16 h, 24 h and 40 h on the spectrophotometer, Multiskan FC, (Thermo Scientific, West palm Beach, FL, USA), with shake mode before measurement at 620 nm, which captures the blue DCPIP as a chromosphere. Photos were also taken. As the absorbances are in inverted relation to the degradation rate, we normalized the data, subtracting from 1, and the rates were obtained based on the negative control absorbance. Negative rates indicate the degradation of substrates and refer to low chromosphere

absorbance for each time of measurement, indicating degradation activity by isolates when enzymes act on aromatic substrates molecules, releasing oxygen atoms to the medium and leading to a change of DCPIP color.

3. Results and Discussion

From the enrichment assay with the five ADE soils, we could isolate, sequence and study the biodegradation capacity of a total of 148 bacteria: 48 from Costa do Açutuba, 71 from Hatahara, 14 from Caldeirão, nine from Barro Branco and from Balbina. The 16S rRNA gene sequences with a length ≥ 400 bp were identified from the RDP database, and all names are in the Supplementary Material. It was possible to obtain bacterial representatives of the phyla, *Proteobacteria*, *Actinobacteria* and *Firmicutes*.

From the identification at EzTaxon-e, it was possible to observe that the majority of the isolated 16S rRNA sequences were similar to those in the database from contaminated areas, as well as the isolates from degradation bioassays with aromatic hydrocarbons.

The isolates' identification corroborates Navarrete *et al.* [14] and Germano *et al.* [15], who studied the bacterial community of ADE by strict molecular analysis and found the following bacterial phyla: *Firmicutes*, *Proteobacteria*, *Verrucomicrobia*, *Acidobacteria*, *Gemmatimonadetes*, *Actinobacteria* and *Nitrospira*. We had 78% of positive isolates for the degradation of phenanthrene and 86% of positive isolates for the degradation of diesel oil, but there were degradation capacity variations, which can be seen in Figures 1 and 2, for phenanthrene and diesel oil, respectively. In the Supplementary Material we provide the full visualization of all isolates for each substrate degradation. However, after observing small absorbance, high percentages since 24 h or less were observed for the pH change as a result of cell growth. Therefore, this changed the DCPIP light reflection from colorless to light purple, now being another chromosphere captured by the spectrophotometer, and the amount of cells was an additional material that was captured by 620 nm. However, this affected our results, once we had already registered the previous absorbance.

It was possible to observe a clear decay on the absorbance data beginning at 2 h for 24.35% of the positive isolates for phenanthrene degradation and 27% of the positive isolates for diesel degradation. After 4 h, 0.87% degraded phenanthrene and 6% of the positive isolates degraded diesel. After 8 h, 5.22% of the positive isolates degraded phenanthrene and 30% of the positive isolates degraded diesel oil. After 16 h, 45.22% degraded phenanthrene and 29% of the positive isolates degraded diesel oil. After 24 h, 19% of the positive isolates degraded phenanthrene and 9% of the positive isolates degraded diesel oil. Finally, after 40 h, 5% of the positive isolates for phenanthrene degradation degraded phenanthrene and 5% of the positive isolates for diesel oil degraded diesel. For the positive control, similar degradation rates to the pretest were observed, and degradation began at 8 h for diesel and 24 h for phenanthrene.

Figure 1. Visualization of the best isolates rates for the degradation of 0.05% of phenanthrene after (a) 2 h; (b) 4 h; (c) 8 h; (d) 16 h; (e) 24 h; and (f) 40 h; measured at 620 nm. Negative data indicate degradation. The positive control is at the top of the figure, and the genus of each isolate is shown.

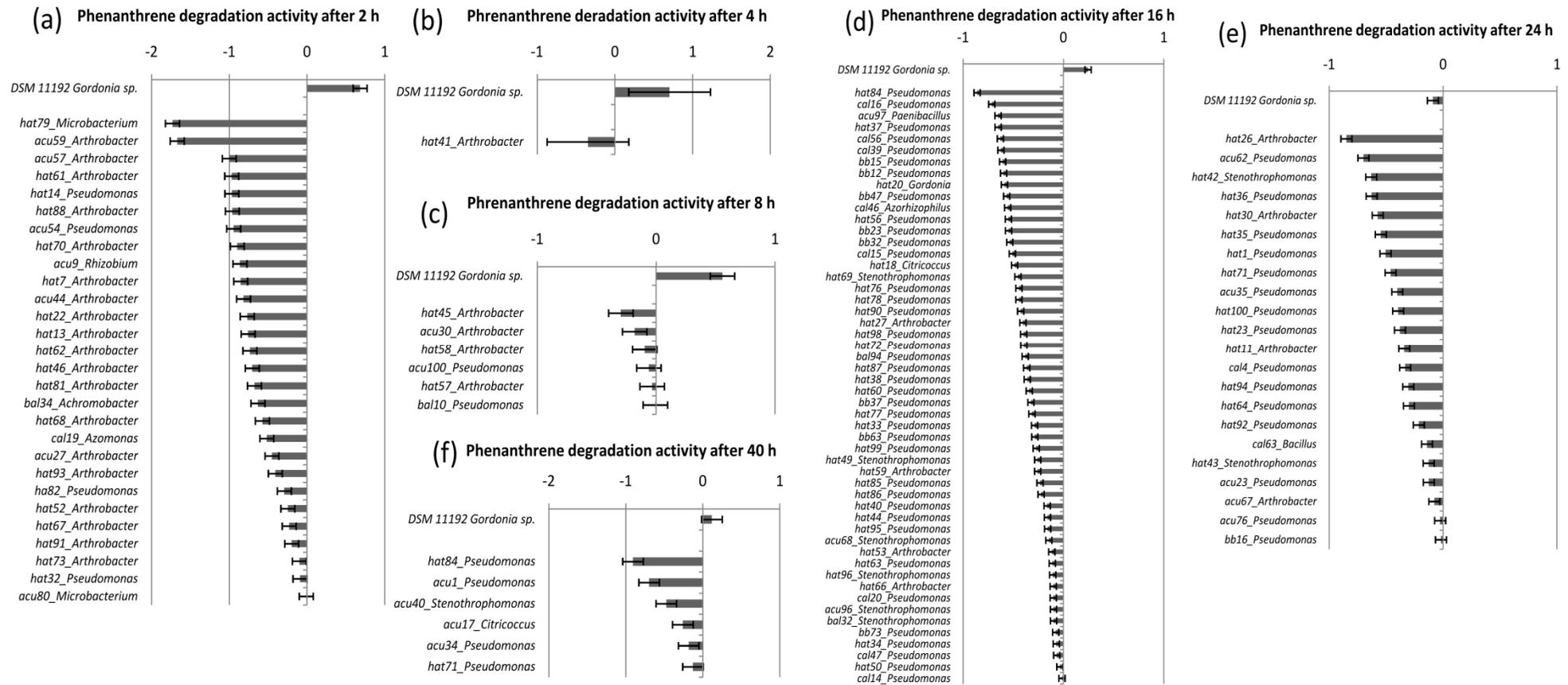
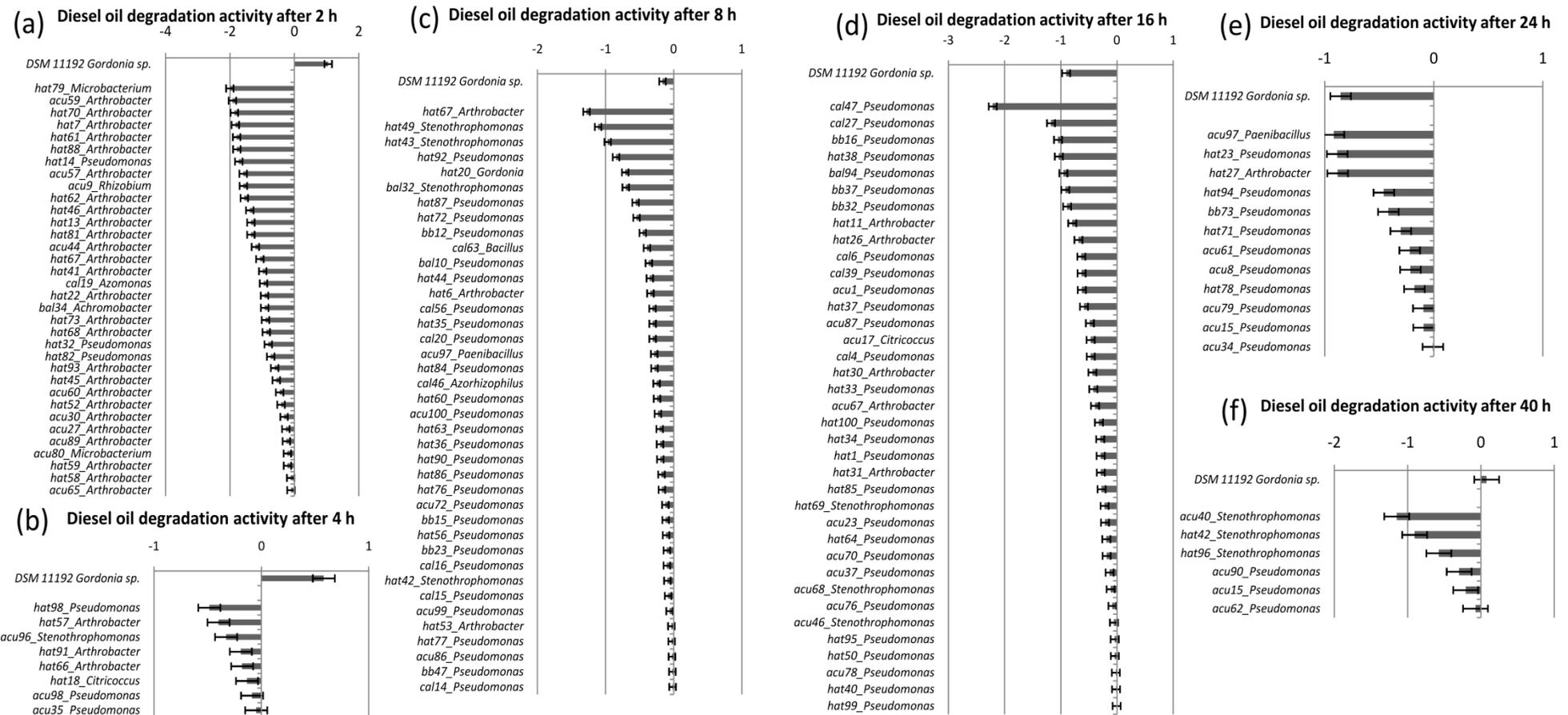


Figure 2. Visualization of the best isolate rates for the degradation of 8% of diesel after (a) 2 h; (b) 4 h; (c) 8 h; (d) 16 h; (e) 24 h; and (f) 40 h; measured at A620 nm. Negative data indicate degradation. The positive control is at the top of the figure, and the genus of each isolate is shown.



We could observe that many isolates developed a faster degradation activity of both substrates than the positive control. Furthermore, 24 isolates degraded both substrates within 2 h from the experiment start, and they can be seen in (a) from Figures 1 and 2 near the positive control.

Achromobacter bal 34 represents 4.16% of the best isolates for both substrates; *Azomonas* cal 19 represents 4.16%; *Rhizobium* acu9 represents 4.16%; *Pseudomonas* hat14 and hat32 represent 8.33%; and *Arthrobacter* acu27, acu44, acu57, acu59, hat13, hat46, hat52, hat61, hat62, hat67, hat68, hat7, hat81, hat70, hat88 and hat93 dominate, representing 62.5% of the best isolates for both substrates.

In Figures 3 and 4, we show the degradative capacity for phenanthrene and diesel oil, respectively, based on the isolates' genus.

Pseudomonas was the dominant genus in all of the ADE soils studied. They are highly adapted and distributed in all environments [29] and strongly related to the biodegradation of hydrocarbons [30]. *Achromobacter*, *Azomonas* and *Stenotrophomonas* also are related with aromatic hydrocarbons of diverse complexities [30–33] and were identified within the best degradative isolates for diesel oil. Isolates of *Achromobacter*, *Azomonas* and *Rhizobium* were found to degrade both substrates used in our bioassays. Furthermore, our isolate of *Azorhizophilus* degraded both phenanthrene and diesel oil. This genus has only two species, which have been documented as nitrogen-fixing bacteria [34], but interestingly, they have not been widely associated with AHs until now.

The second most dominant active group was represented by the *Actinobacteria* phylum, the organisms of which have versatile metabolism and are able to use diverse substrates, such as carbon, and energy sources. Most genera have great industrial interest, such as *Arthrobacter*, *Rhodococcus* and *Gordonia*, which are associated with saturated aliphatic and aromatic hydrocarbons [35–37], in addition to *Citricoccus* [38] and *Microbacterium*. *Methylobacterium* is methane utilizing and degrades explosive compounds in association with plants [39]. Our bioassays strongly evidenced the degradative activity of *Arthrobacter*, *Microbacterium* and *Gordonia* isolates, found within the best degradative isolates for both substrates.

Bacillus and *Paenibacillus* are from the *Firmicutes* phylum, have also been found between the isolates from the enrichment assay and were active in the biodegradation assay, being described as aromatic hydrocarbon-degrading bacteria [30,31,40].

The disparity of the degradation of phenanthrene and diesel oil by isolates is due to phenanthrene being a pure substance, unlike diesel oil, which is a complex mixture of hydrocarbons, such as alkanes, cycloalkanes, polyaromatic hydrocarbons and sulfur [41]. Therefore, diesel oil provides a diverse carbon source, being more utilized by diverse degradation genes, differently from phenanthrene, with only certain genes, such as the *pnhA* genes, that can produce the phenanthrene dioxygenases to degrade the substrate.

Thereby, our bioassay results show that the bacterial populations of the ADE are likely to be activated for the degradation of AHs as a survival characteristic.

The existence of this bacterial population, which consumes aromatic hydrocarbons in ADE, may be due to a great biochar content in ADE. Therefore, the existence of high amounts of biochar is perhaps a selective factor for the bacterial community, and horizontal gene transfer can be a cause for the existence of such a capacity in ADE bacteria.

Figure 3. Visualization of the isolates based on their identification of the genus within phyla for the degradation of 0.05% of phenanthrene after (a) 2 h; (b) 4 h; (c) 8 h; (d) 16 h; (e) 24 h; and (f) 40 h; measured at 620 nm. Negative data indicate degradation. The positive control is at the top of the figure, and the genus of each isolate is shown. Each colored bar indicates the time of degradation for a determined degradation rate.

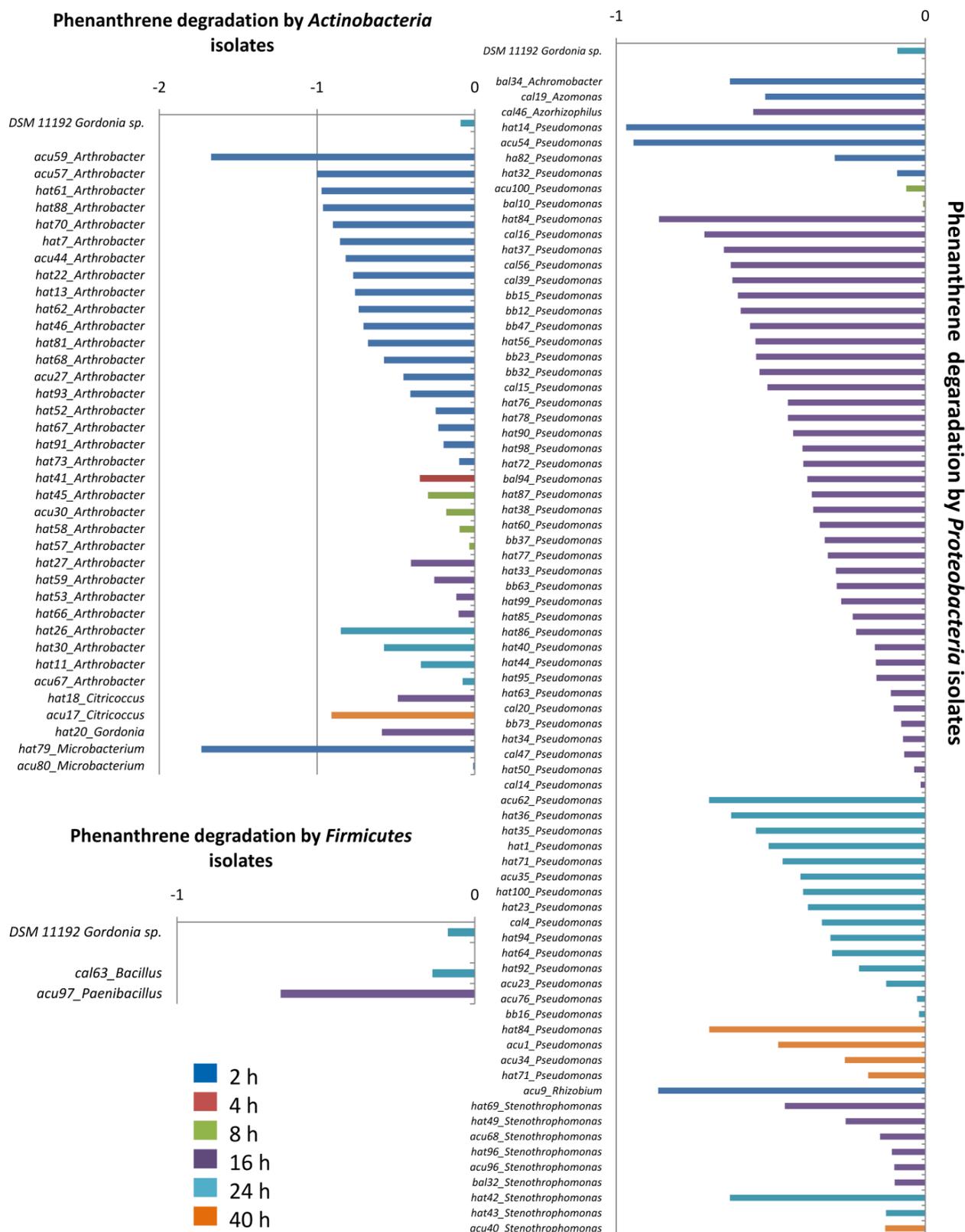
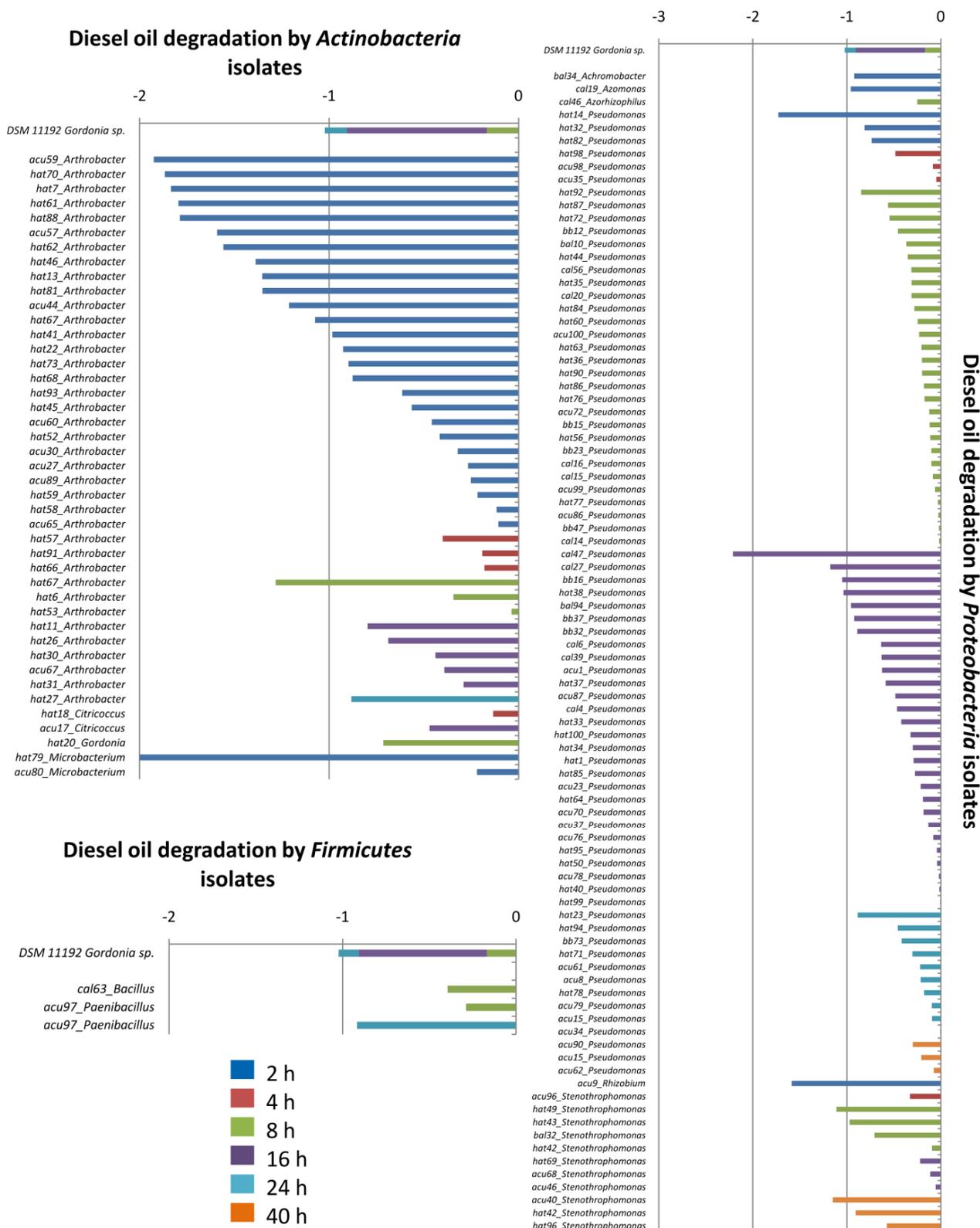


Figure 4. Visualization of the isolates based on their identification of the genus within the phyla for the degradation of 8% of diesel after 2 h, 4 h, 8 h, 16 h, 24 h and 40 h measured at 620 nm. Negative data indicate degradation. The positive control is at the top of the figure, and the genus of each isolate is shown. Each colored bar indicates the time of degradation for a determined degradation rate.



4. Conclusions

This work proved isolates that are aromatic hydrocarbon degraders and that can use complex sources of carbon, like phenanthrene and the compost of diesel oil, as carbon and energy sources. Thus, these isolates present great biotechnological potential for the generation of products and process of interest for bioremediation and bioprospection, both for molecules and genes linked to the metabolic pathways of biodegradation. The generated data from this study also provide some information about the role of those microorganisms in nutrient cycling, the process of which directly influences the resilience and fertility associated with ADE.

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Author Contributions

Siu Mui Tsai was the project planner and higher supervisor. Mariana Gomes Germano was the sub-supervisor and author who performed part of the collection and analysis of the data. Finally, Fernanda Mancini Nakamura performed part of the collection, as well as the analysis of the data and organized the article, and so, is the contact for this paper.

Conflicts of Interest

The authors declare no conflict of interest.

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