

A Second Generation Mn-Porphyrin Dimer with a Twisted Linker as a Potential Blood Pool Agent for MRI: Tuning the Geometry and Binding with Serum Albumin

Supporting Information

Table of Contents

Fig. S.1. ¹ H-NMR of 1	2
Fig. S.2. ¹ H-NMR of <i>m</i> -P2	3
Fig. S.3. Mass spectrum of 1	4
Fig. S.4. Mass spectrum of <i>m</i> -P2	5
Fig. S.5. Mass spectrum of <i>m</i> -MnP2.	6
Fig. S.6. UV-Visible spectra of <i>m</i> -P2 and <i>m</i> -MnP2.	7
Fig. S.7. Molecular dynamics calculations for both MnP2 dimers	8
Fig. S.8. Mass spectrum of oversulfonated <i>m</i> -P2	9
Dissociation constant determination	10

Characterization

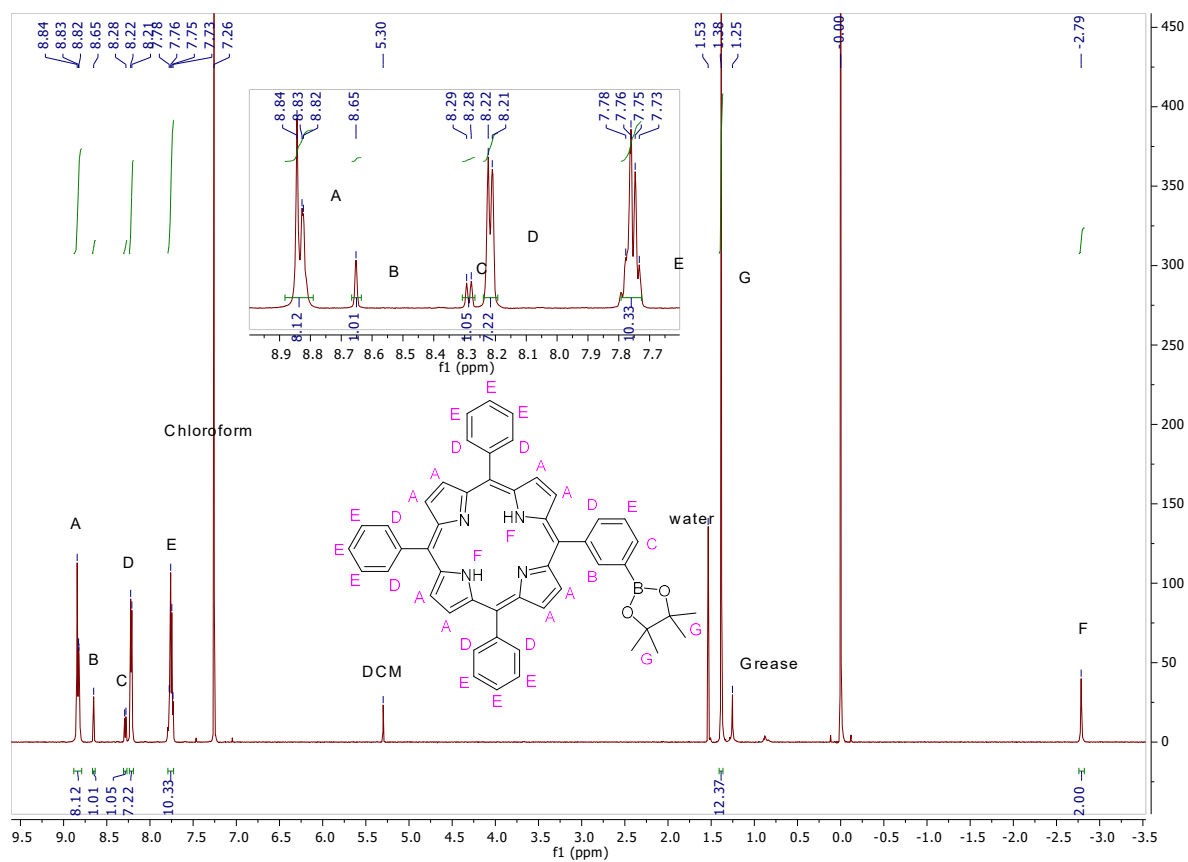


Figure S1. ^1H -NMR of **1** acquired in CDCl_3 with 0.1% TMS. Residual solvent peaks are CHCl_3 , CH_2Cl_2 , grease, and water.

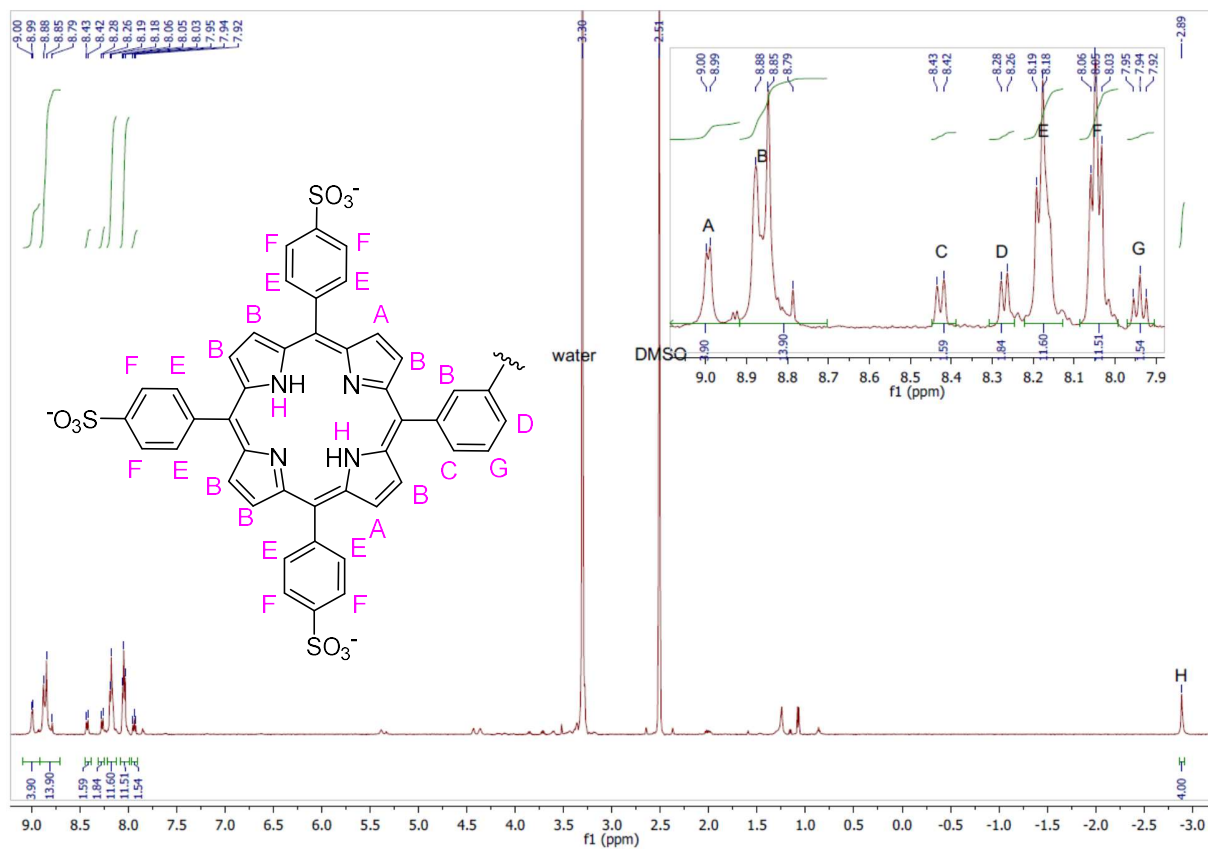


Figure S2. ^1H -NMR of *m*-P2 acquired in $\text{DMSO-}d_6$. Residual solvent peaks are DMSO and water.

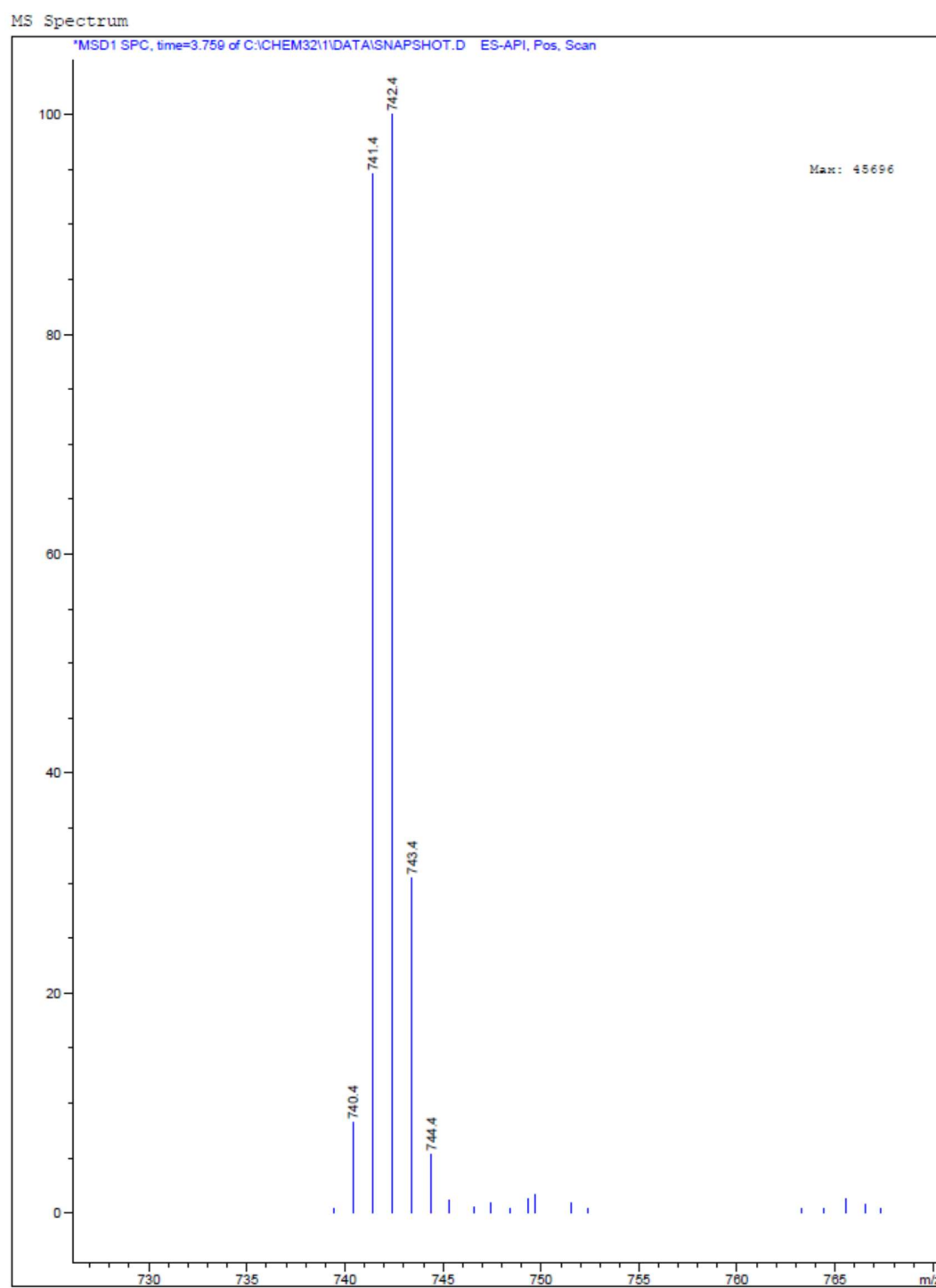


Figure S3. Positive mode ESI-MS of **1**.

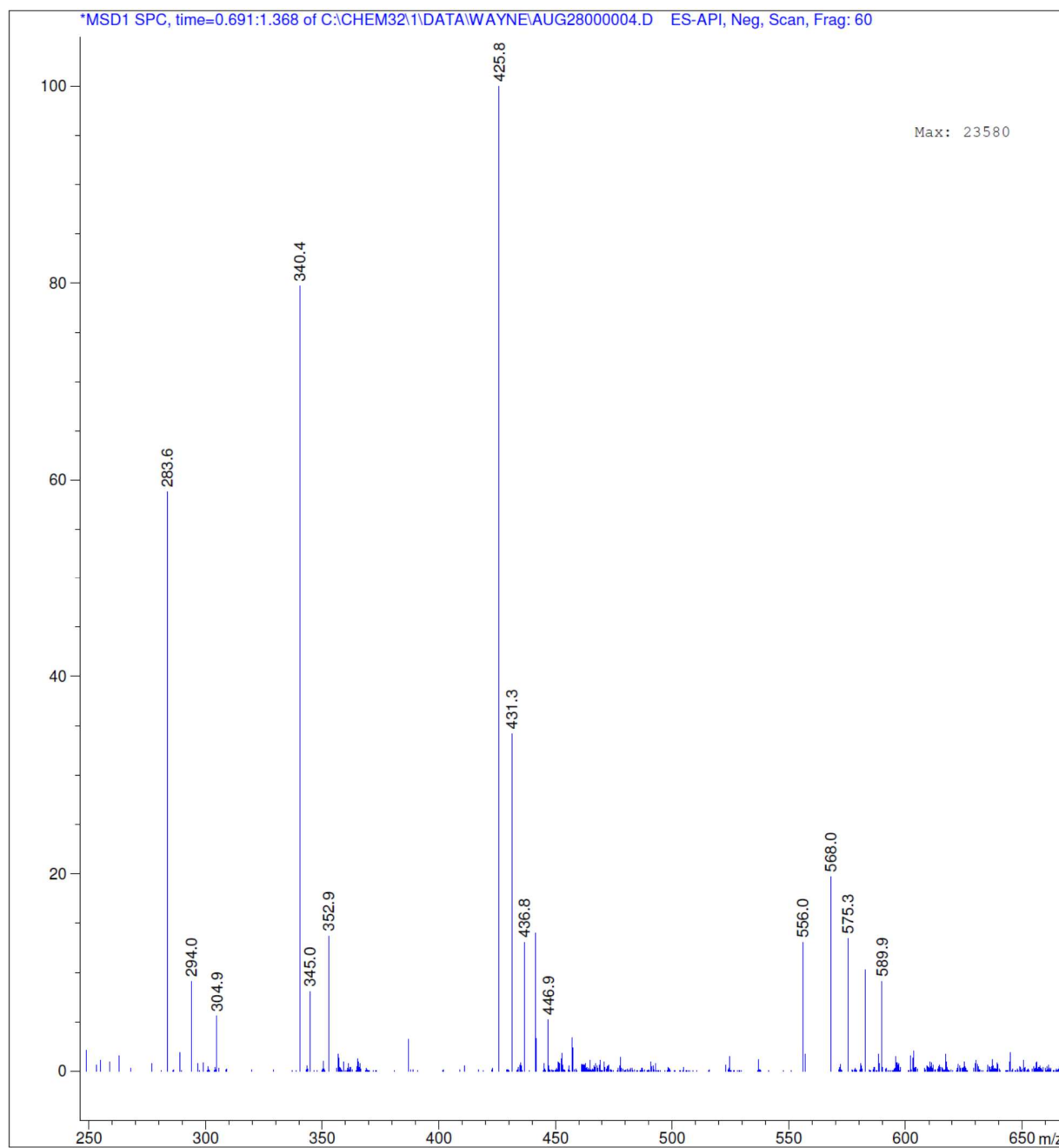


Figure S4. Negative mode ESI-MS of *m*-P2.

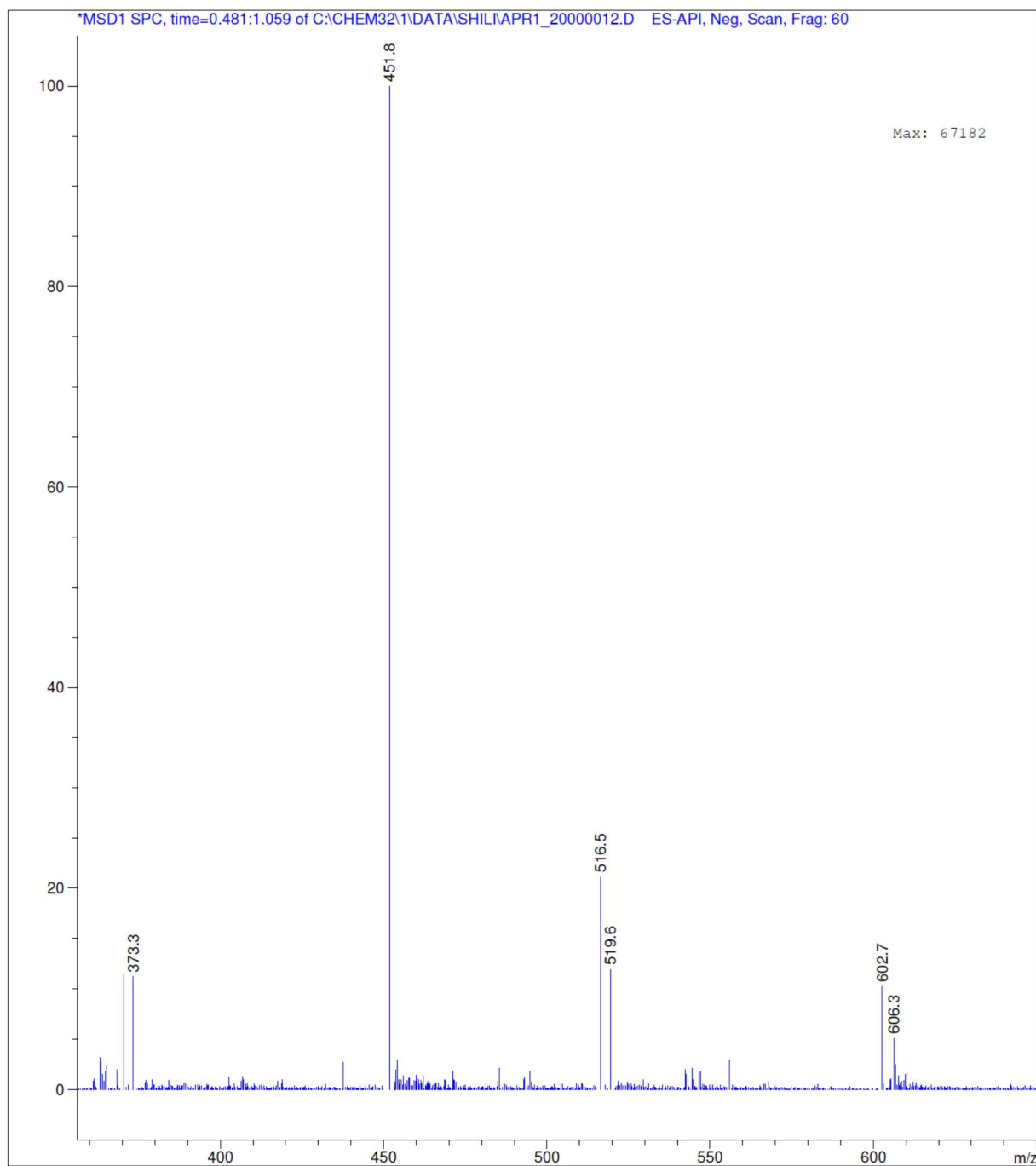


Figure S5. Negative mode ESI-MS of *m*-MnP2.

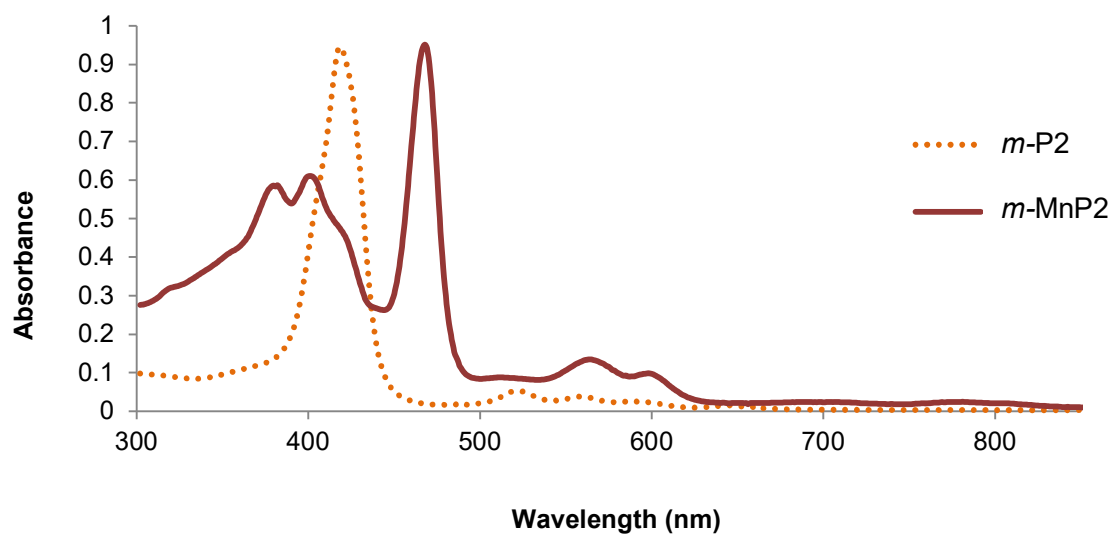


Figure S6. UV-Visible spectra of *m*-P2 and *m*-MnP2 measured in 25 mM pH 7 HEPES buffer. λ_{max} of *m*-P2 = 422 nm, λ_{max} of *m*-MnP2 = 468 nm.

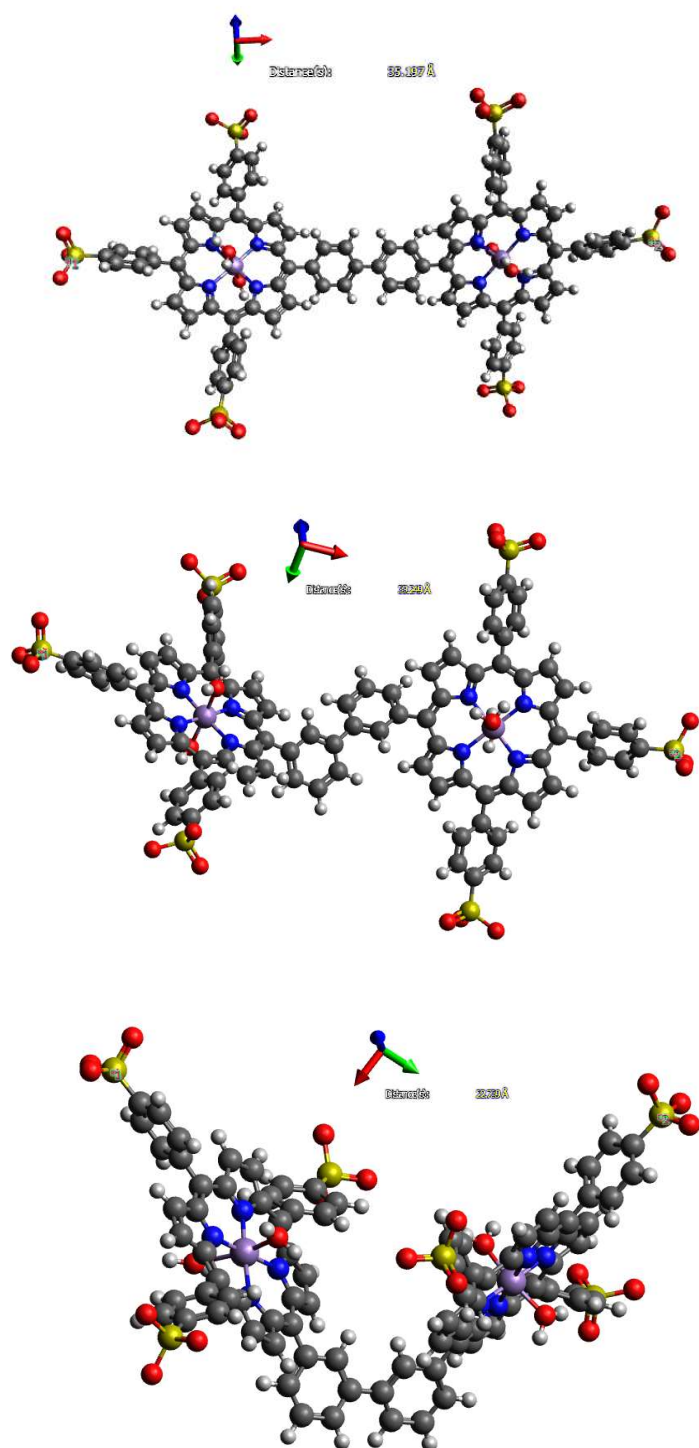


Figure S7. Molecular modeling of MnP2 (top) and two conformers of *m*-MnP2 (middle and bottom). The distance between two distal S-atoms are labeled. (35.197, 33.249 and 22.739 Å, respectively).

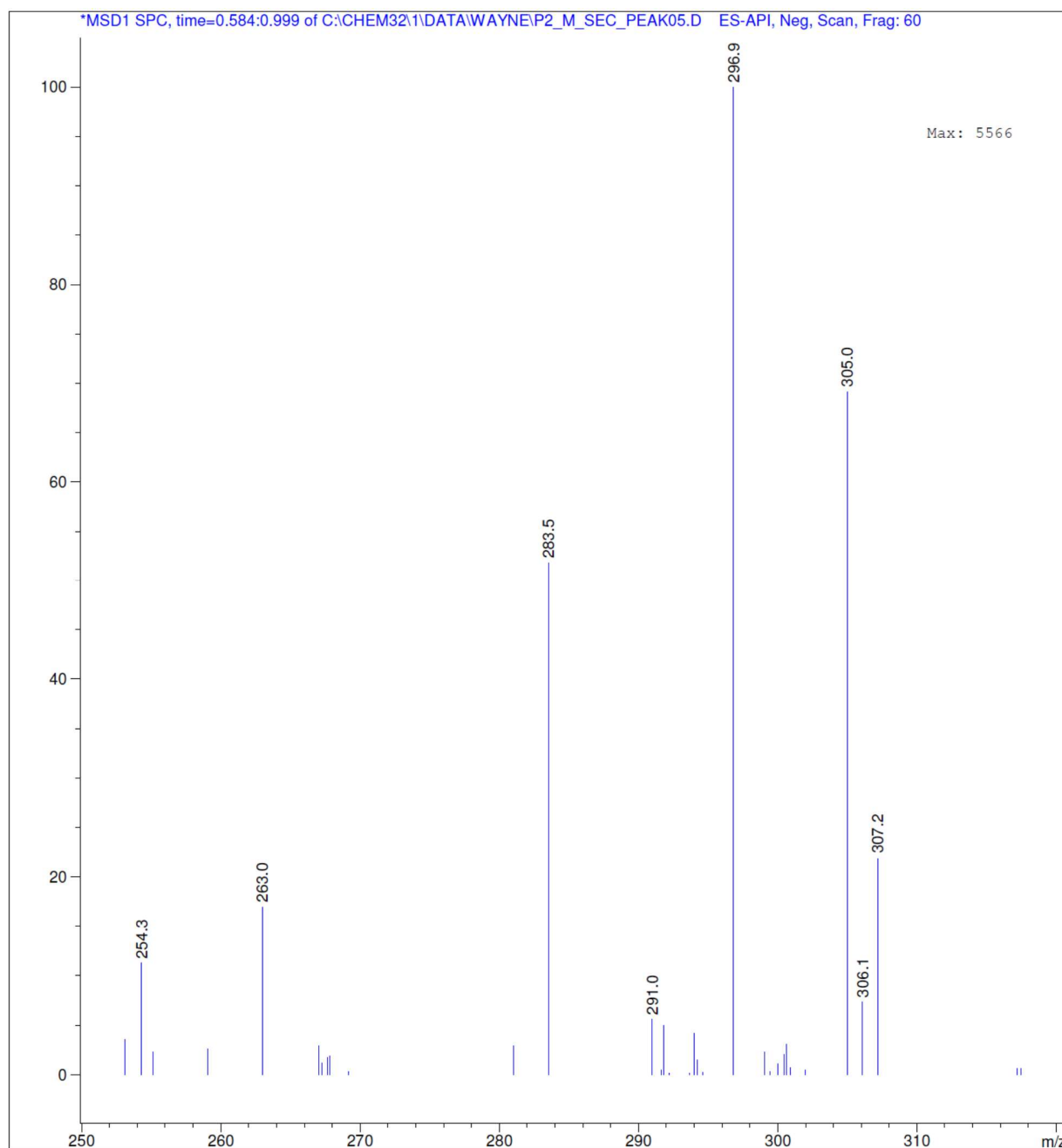


Fig. S8. Negative mode ESI-MS of oversulfonated *m*-P2 found $m/z = 296.80$ ($[M]^{6-}$), calculated for $C_{88}H_{49}N_8O_{21}S_7^{6-}$ ($m/z = 296.69$).

Dissociation Constant Determination

The dissociation constant, K_d , was obtained using the GraphPad Prism/OriginLab Pro 9.0 software by fitting the experimental data to the following equations:

$$LR \xrightleftharpoons{K_d} L + R \quad [1]$$

$$LR = \frac{(x+L_0+K_d) - \sqrt{(x+L_0+K_d)^2 - 4x \times L_0}}{2} \quad [2]$$

$$L = L_0 - LR \quad [3]$$

$$Y - Y_0 = MLR \times LR \quad [4]$$

These equations are based on the assumption that the porphyrin, L, and the HSA, R, are bound to form a 1:1 LR complex. L_0 is the total concentration of porphyrin and x denotes the total concentration of HSA in the solution. The MLR is the molar absorbance of the LR complex. Y and Y_0 are the observed absorbance and the initial absorbance respectively.