

Supporting information

Reversible Aggregation of Molecular-Like Fluorophores Driven by Extreme PH in Carbon Dots

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Received: 1 July 2020; Accepted: 13 August 2020; Published: date

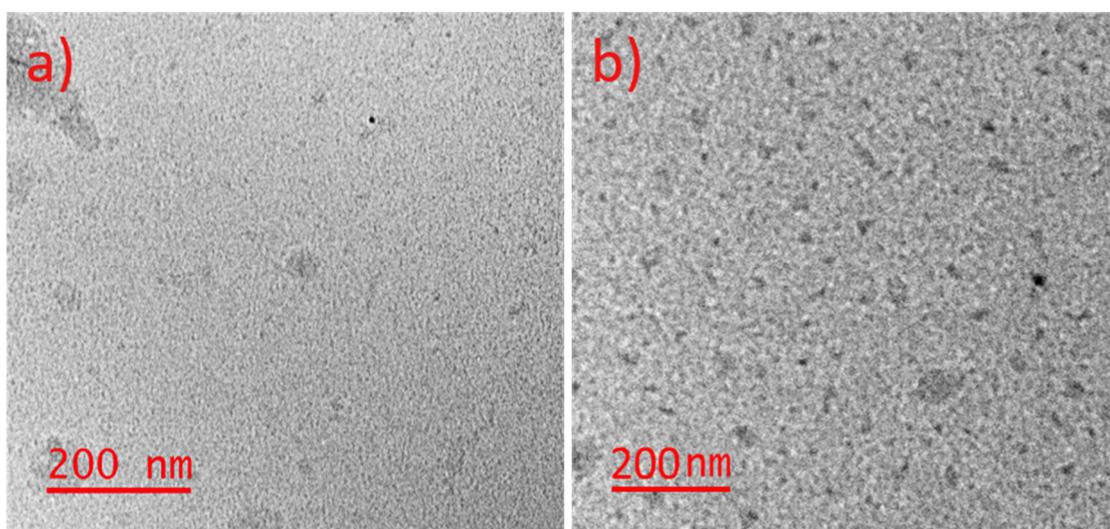


Figure S1. Representative TEM images of (a) CU2 and (b) CU25.

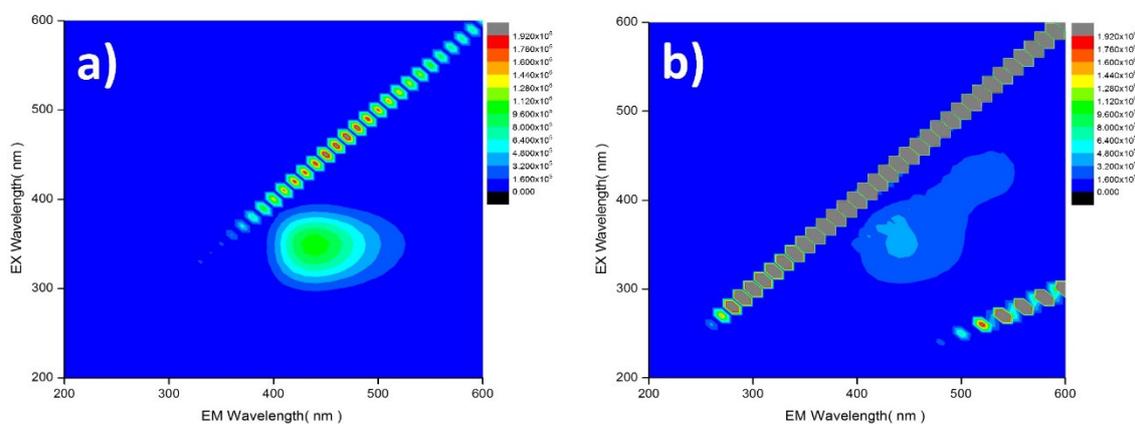


Figure S2. 3D photoluminescence spectra (excitation (y-axis), emission (x-axis), intensity (false colors scale)) of citrazinic acid (a) in water and (b) in H₂SO₄ 10% at concentration of 10 mg L⁻¹.

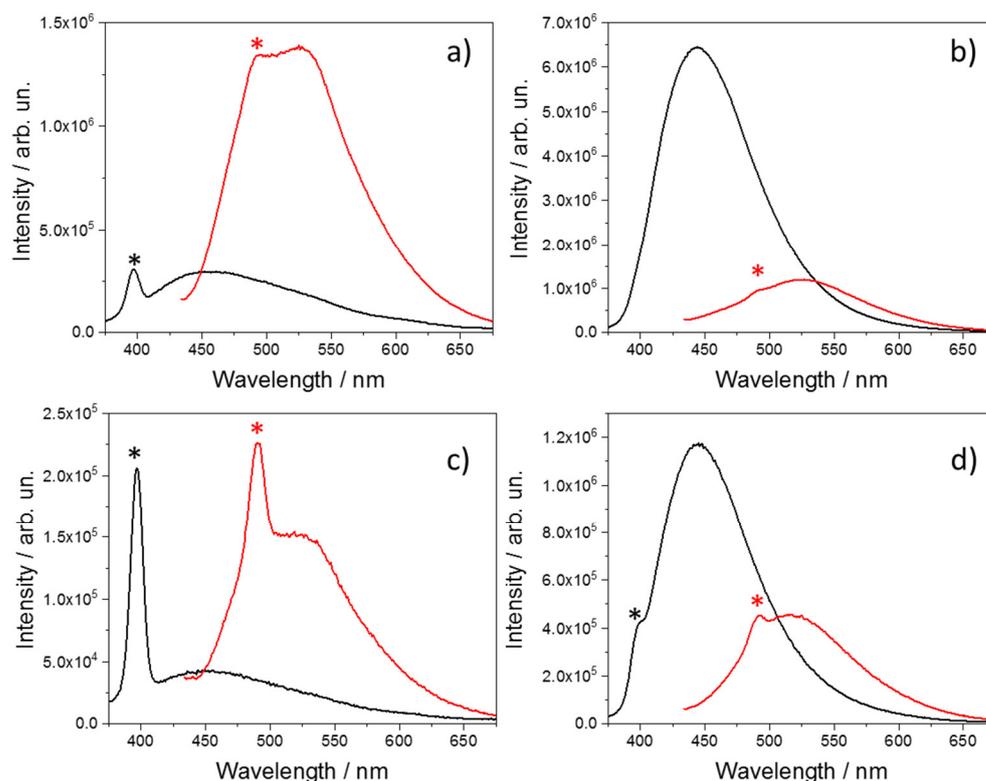
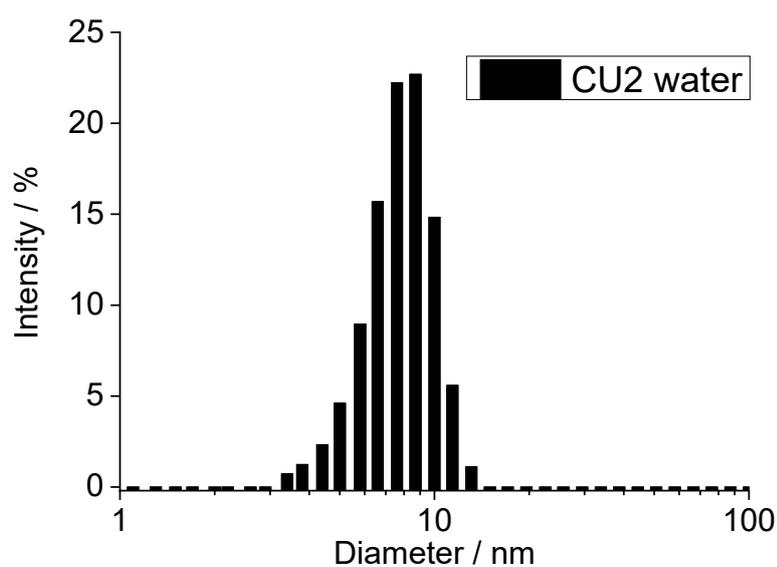
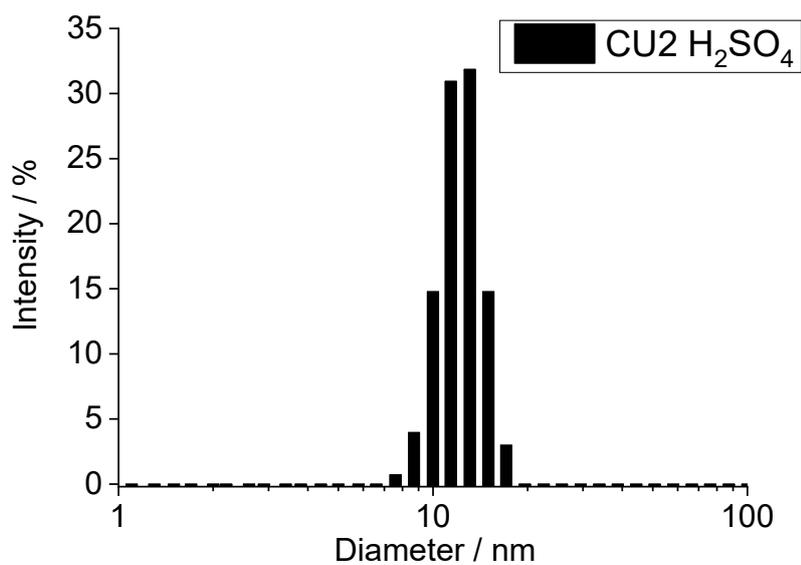


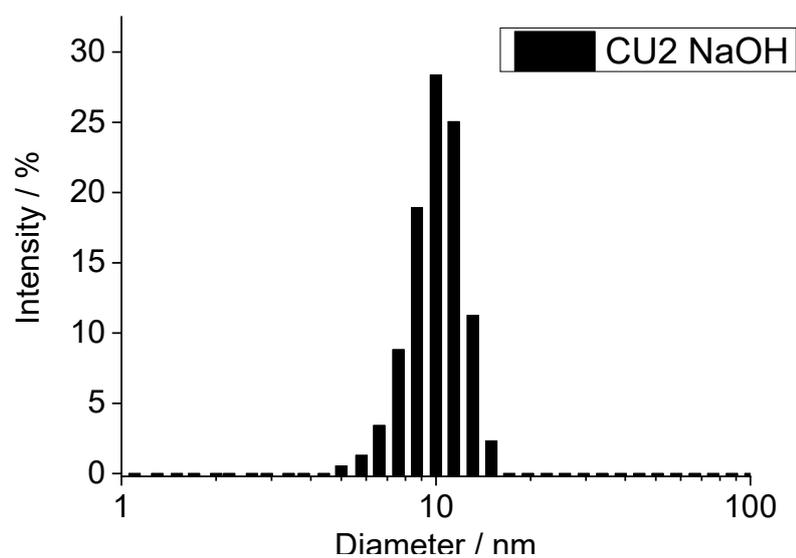
Figure S3. PL emissions of CU2 C-dots in sulfuric acid (10%) with excitation at 350 nm (black) and 420 nm (red) at the C-dots concentrations of (a) 1 mg L⁻¹ and (c) 0.1 mg L⁻¹ and after neutralization with NaOH pellets with excitation at 350 nm (black) and 420 nm (red) at the C-dots concentrations of (b) 1 mg L⁻¹ and (d) 0.1 mg L⁻¹. The asterisks indicate Raman vibrational modes of water; Figure S4: Light scattering analysis of CU2 and CU25 C-dots in the aqueous solutions (10 mg L⁻¹) at different pH values (water = 7, H₂SO₄ = 1, NaOH = 14); * The asterisks indicate Raman vibrational modes of water.



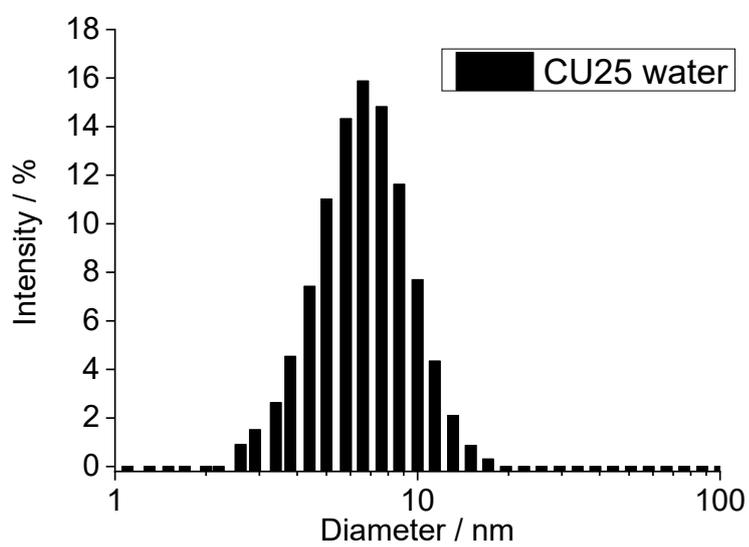
(a)



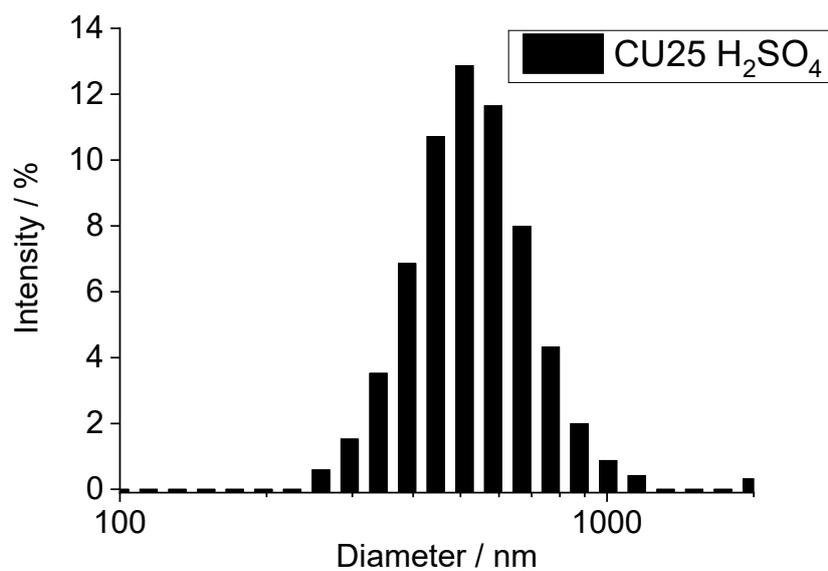
(b)



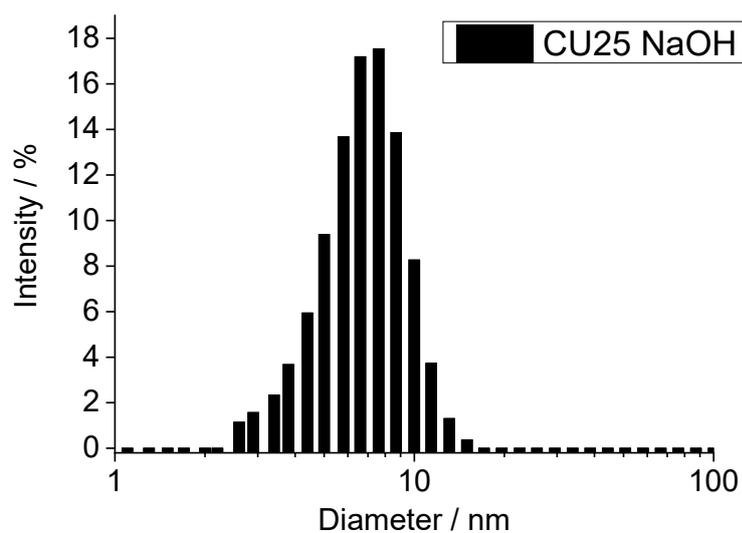
(c)



(d)



(e)



(f)

Figure S4. Light scattering analysis of CU2 and CU25 C-dots in the aqueous solutions (10 mg L⁻¹) at different pH values (water = 7, H₂SO₄ = 1, NaOH = 14). CU2 in (a) water; (b) H₂SO₄ and (c) NaOH; CU25 in (d) water, (e) H₂SO₄ and (f) NaOH.

Table 1. Decay lifetimes under excitations at 340 and 405 nm.

SAMPLE	τ_1	τ_2
CU2/water ($\lambda_{ex} = 340$ nm; $\lambda_{em} = 420$ nm)	4.7 ns	10.0 ns
CU2/H ₂ SO ₄ ($\lambda_{ex} = 340$ nm; $\lambda_{em} = 420$ nm)	2.1 ns	7.7 ns
CU2/H ₂ SO ₄ ($\lambda_{ex} = 405$ nm; $\lambda_{em} = 510$ nm)	2.5 ns	6.2 ns
CU2/NaOH ($\lambda_{ex} = 340$ nm; $\lambda_{em} = 420$ nm)	5.3 ns	9.1 ns
CU25/water ($\lambda_{ex} = 405$ nm; $\lambda_{em} = 510$ nm)	4.4 ns	9.1 ns
CU25/H ₂ SO ₄ ($\lambda_{ex} = 405$ nm; $\lambda_{em} = 510$ nm)	3.2 ns	6.9 ns
CU25/NaOH ($\lambda_{ex} = 340$ nm; $\lambda_{em} = 420$ nm)	2.3 ns	7.8 ns
CU25/NaOH ($\lambda_{ex} = 405$ nm; $\lambda_{em} = 510$ nm)	2.8 ns	9.1 ns