

Microbial Removal of Heavy Metals from Contaminated Environments Using Metal-Resistant Indigenous Strains

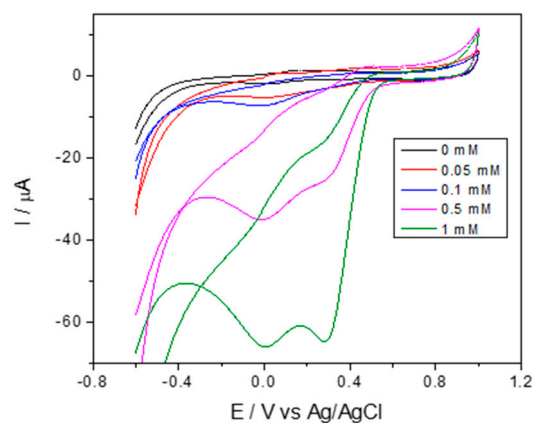


Figure S1. Cyclic voltammograms recorded on the surface of the PB/SPE sensor: in the absence and in the presence of Cr^{6+} at different concentrations (0.1 M KCl + 0.1 M HCl, $v = 0.1$ V/s).

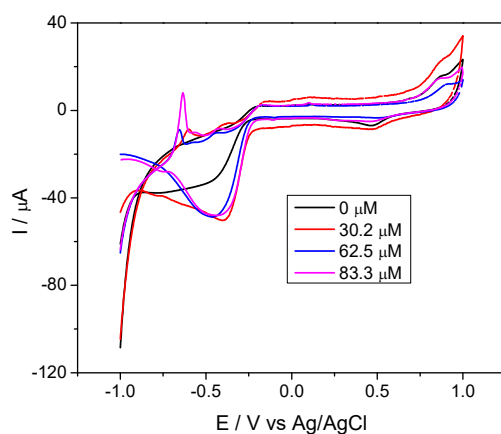


Figure S2. Cyclic voltammograms recorded at the surface of the AuNPs/CS/SPEs sensor in the absence and in the presence of Pb^{2+} at different concentrations (Tris-HCl buffer 0.1 M, pH 5, $v = 0.1$ V/s).

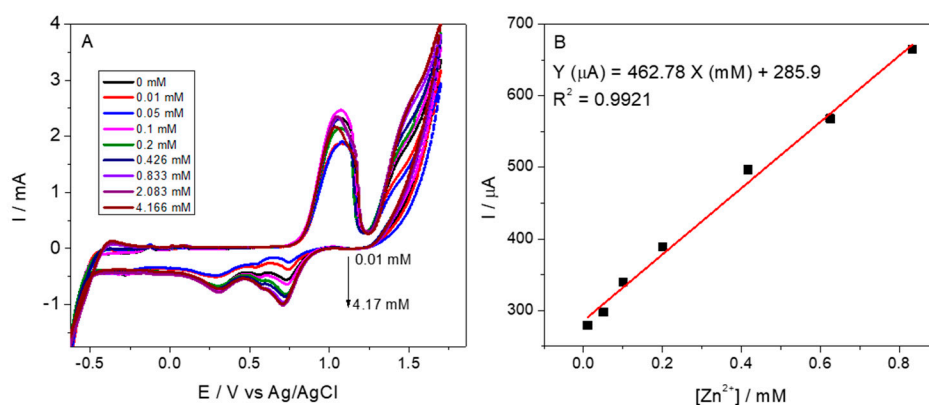


Figure S3. Detection of Zn^{2+} using the MWCNTs-CS/PB/AuSPE sensor in cyclic voltammetry: (A) Voltammograms obtained in the electrolyte solution in the absence and in the presence of different concentrations of Zn^{2+} ($v = 0.1 \text{ V/s}$); (B) Calibration curve for the variation of the reduction peak current in cyclic voltammetry vs. concentration of Zn^{2+} .

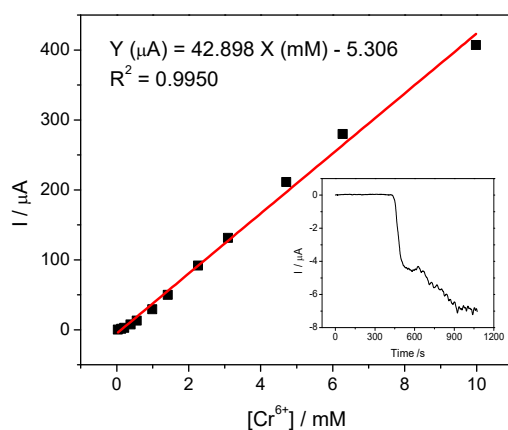


Figure S4. Calibration curve obtained for Cr^{6+} ion detection using the PB/SPE sensor (applied potential 0.3 V vs. Ag/AgCl, 0.1 M KCl + 0.1 M HCl). Inset: Representation of current intensity dependence on Cr^{6+} concentration.

Table S1. Optimization of the working potential and the performance parameters for the amperometric detection of Cr^{6+} using Prussian Blue modified SPEs (0.1 M KCl + 0.1 M HCl).

E (V)	Linear Range (mM)	R^2	Sensitivity ($\mu\text{A}/\text{mM}$)	LOD (μM)	Specific sensitivity ($\text{mA}\cdot\text{M}^{-1}\cdot\text{cm}^{-2}$)
0.28		0.998	39.10	0.61	311.32
0.3	0.035-13.413	0.995	42.90	0.56	341.54
0.33		0.999	36.66	0.82	291.89

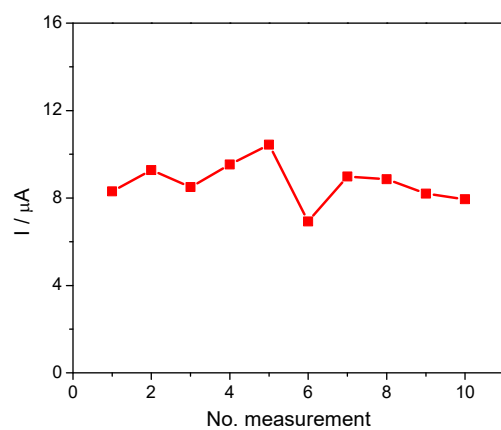


Figure S5. Stability of sensors modified with Prussian Blue for the determination of 1 mM Cr^{6+} .

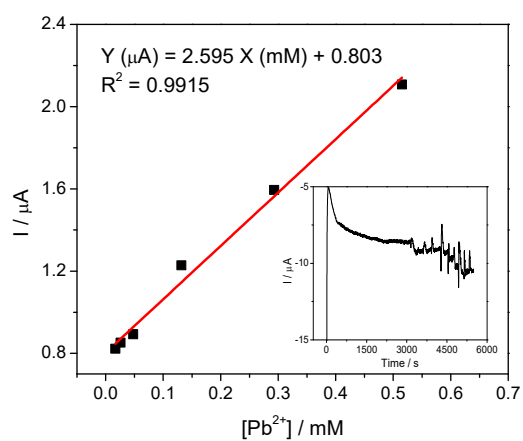


Figure S6. Amperometric detection of Pb^{2+} ion using AuNPs-CS/SPE sensor ($E = -0.45 \text{ V}$ vs. Ag/AgCl , 0.1 M Tris-HCl, pH 5). Inset: Representation of current intensity dependence on Pb^{2+} concentration.

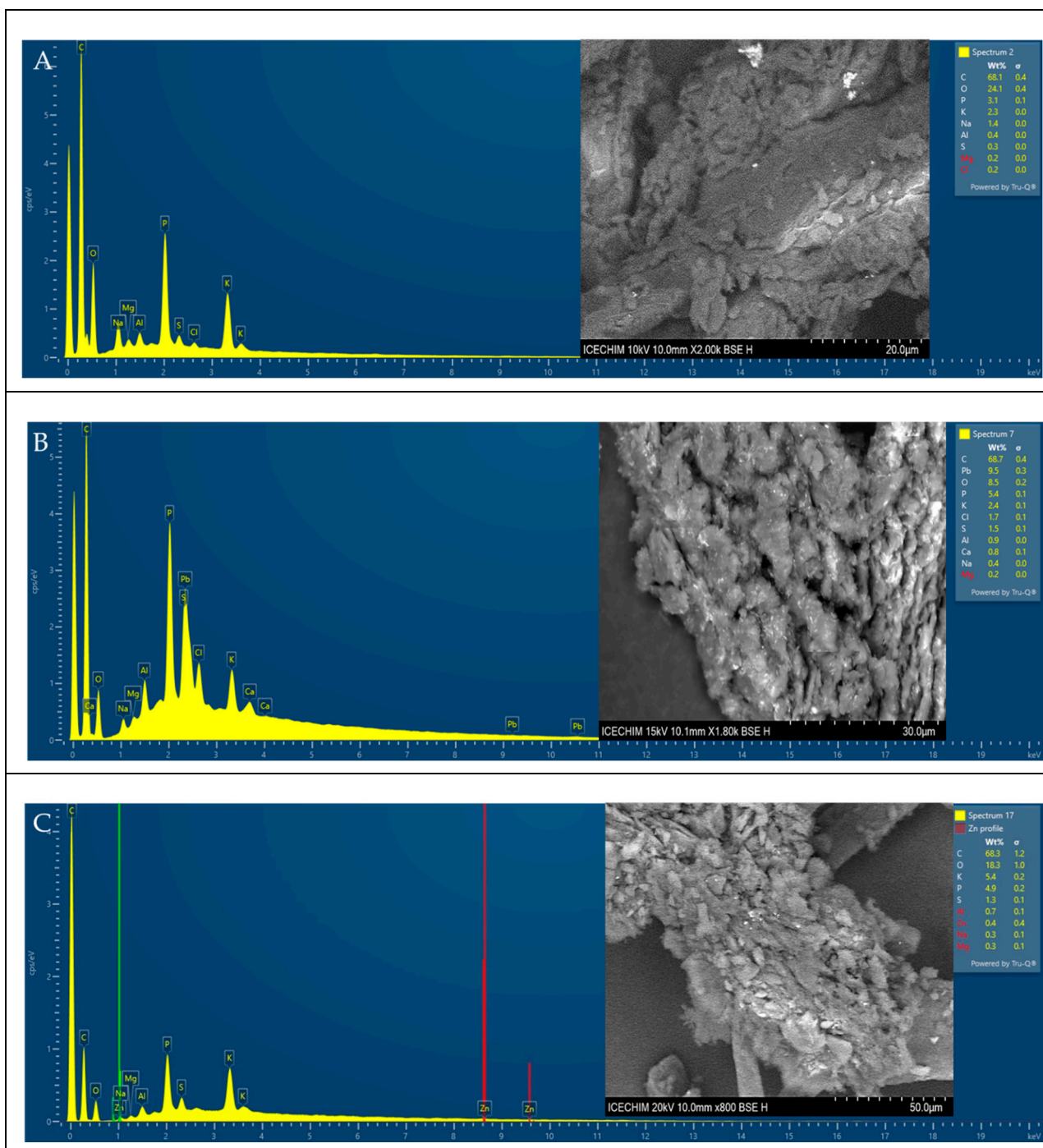


Figure S7. Scanning electron micrographs (SEM) and EDX spectra of *Bacillus marisflavi* biomass A) control, B) Pb treated, C) Zn treated.

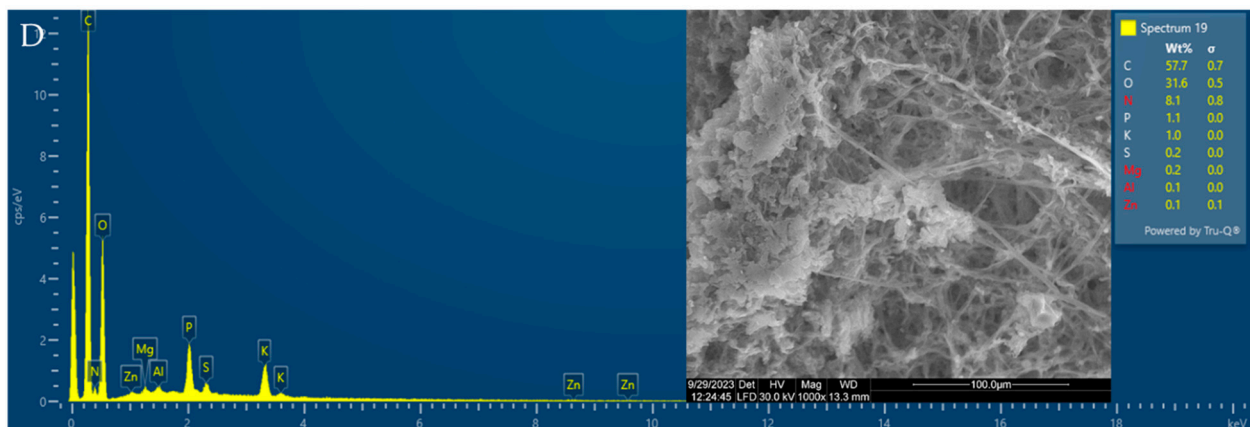
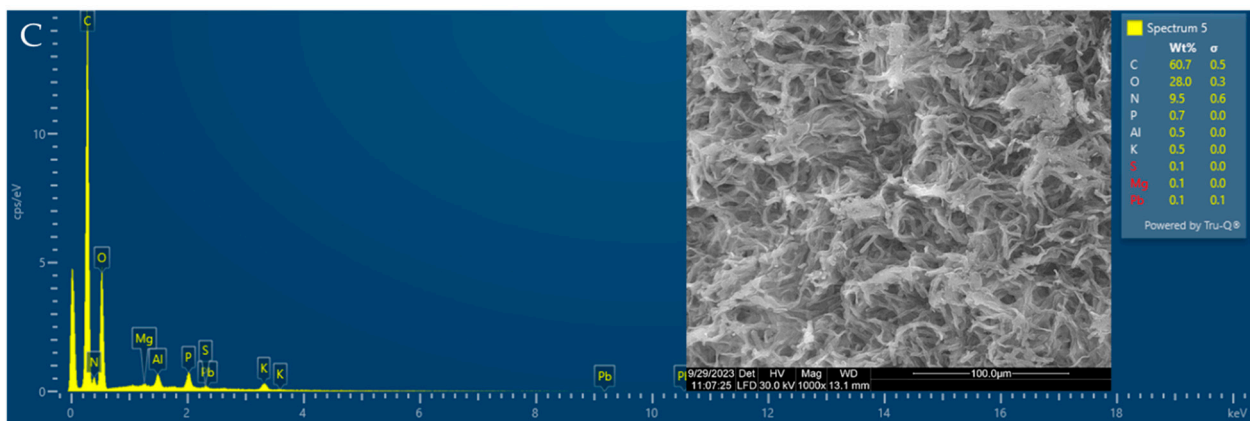
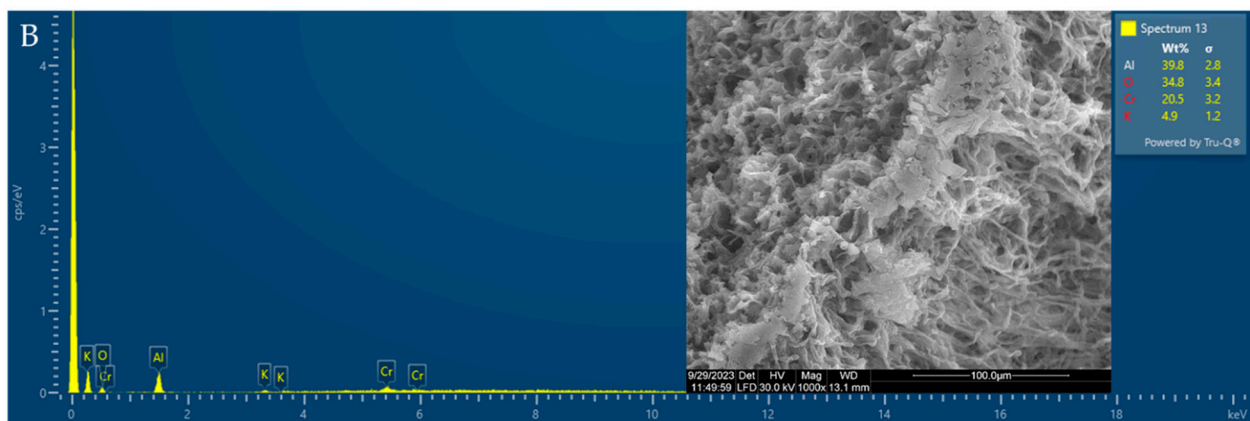
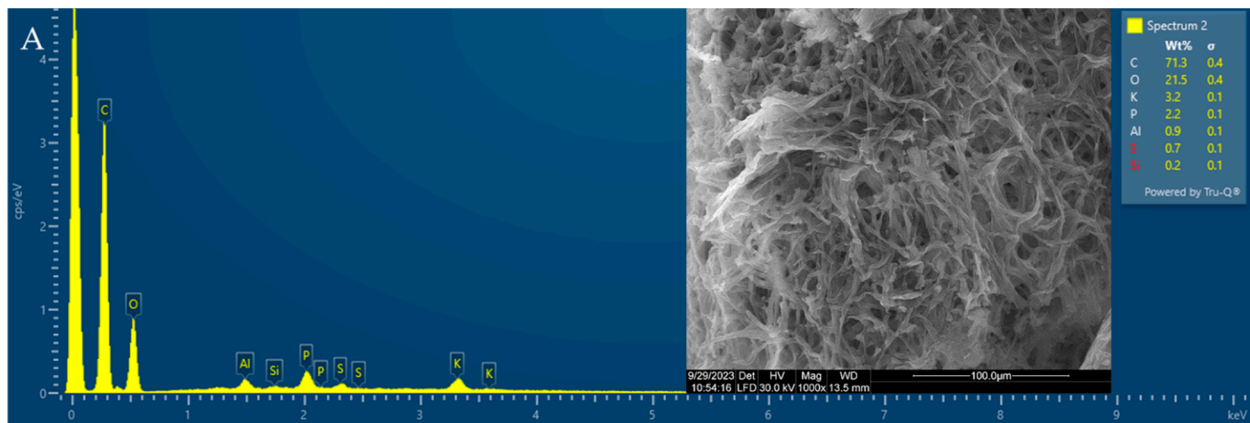


Figure S8. Scanning electron micrographs (SEM) and EDX spectra of *Trichoderma longibrachiatum* biomass A) control, B) Cr treated, C) Pb treated, D) Zn treated.

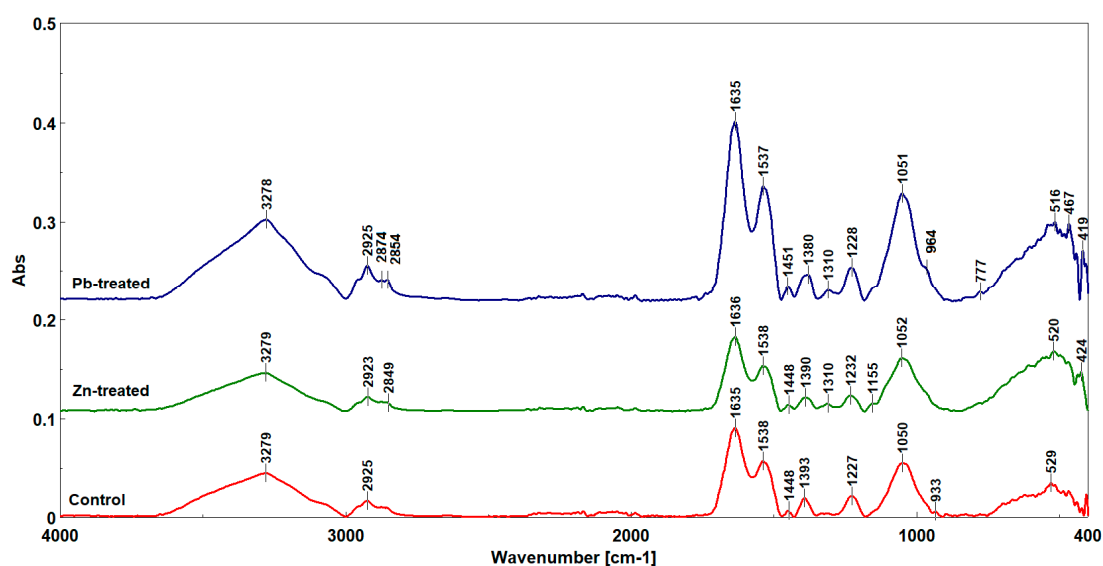


Figure S9. Fourier transform infrared spectroscopy spectra of *Bacillus marisflavi* biomass before and after treatment with Pb (NO₃)₂ and ZnSO₄.

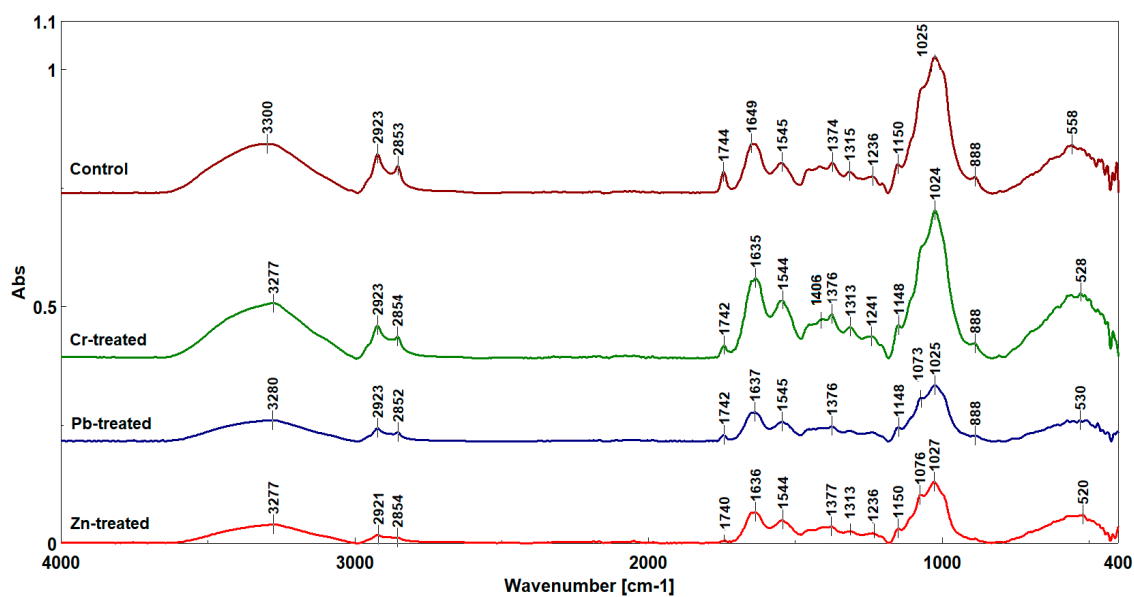


Figure S10. Fourier transform infrared spectroscopy spectra of *Trichoderma longibrachiatum* biomass before and after treatment with K₂Cr₂O₇, Pb (NO₃)₂ and ZnSO₄.