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Minimizing the Health Risks from Hydrocarbon Contaminated Soils by Using Electric Field-Based Treatment for Soil Remediation

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Abstract: The present work addresses the assessment of human health risk from soil contaminated with total petroleum hydrocarbons (TPHs) due to crude oil pollution, with a particular focus on the polycyclic aromatic hydrocarbon (PAH) group of carcinogenic and toxic substances. Given that the measured risk for human health exceeded the accepted level, the study considered an electrochemical remediation method. The laboratory-scale experiments were conducted by using an electric field-based treatment as a possible solution for the remediation of contaminated soil. After 20 days of treatment, while the voltage applied was 15 V (specific voltage of 1 V/cm), the hydrocarbon content was significantly reduced. The parameters measured to determine the overall remediation efficiency were pH, redox potential, ionic strength, soil characteristics, voltage gradient, and zeta potential. The remediation degree observed during the experiments was around 50% for TPHs and 46% for PAHs. The applied remediation method resulted in significant removal efficiency of the tested contaminants from the soil. Consequently, the human health risk assessment for the new degree of contaminants in the soil was achieved. This data demonstrated to what extent the application of the remediation applied technology ensured an acceptable risk under the same exposure conditions for the industrial workers.

Keywords: risk assessment; PAHs; electrochemical remediation; contaminated soil

1. Introduction

In recent decades, the issue of soil pollution has increased due to rapid urbanization and industrialization. Unfortunately, contaminated soil is causing significant environmental problems due to the persistence of pollutants, which accumulate in soil and prove difficult to remove [1]. Moreover, contaminated soils are a key environmental concern for all industrialized countries, and, in particular, Europe and Romania. In Europe, according to data extrapolated to the whole of Europe and provided by the European Environment Agency [2], the estimate of the total number of potentially contaminated sites stands at around 2.5 million, of which about 14% (350,000 sites) are considered to be heavily contaminated and, hence, in need of immediate remediation measures. In Romania, the National Environmental Protection Agency drafted the National Inventory in 2014, which lists 1183 potentially contaminated sites and 210 contaminated sites, to give a total number of 1393 affected sites. The industries having significantly contributed to soil pollution in Romania are the mining and metallurgical industry, with 160 potentially contaminated sites; the petroleum industry, with 861 potentially contaminated sites; and the chemical industry, with 37 potentially contaminated

sites [3]. On the other hand, as the proposal for a Soil Framework Directive was withdrawn by the European Commission in May 2014, Europe still lacks a common soil legislative instrument and European Union (EU) Member States have their own laws on soil protection. If soil contamination is the presence of a minor and unwanted contaminant in the natural environment (in our case study, the presence of polycyclic aromatic hydrocarbons in soil), soil pollution represents the introduction of contaminants into the natural environment that cause adverse change; so, specifically, the present work is about soil pollution with crude oil during oil exploration. Regarding soil contamination and risk assessment, some of the European Union member countries have already established procedures to assess and manage the risks posed by contaminated sites, while others are currently in the process of doing so. In this regard, the first significant European project was promoted by the German Environment Ministry and coordinated by the Federal Environment Agency (Umweltbundesamt, Dessau-Roßlau, Germany), and involved scientists from 16 European countries: Austria, Belgium, Denmark, Finland, France, Germany, Greece, Ireland, Italy, Netherlands, Norway, Portugal, Spain, Sweden, Switzerland, and the United Kingdom. The project was called “Concerted Action on Risk Assessment for Contaminated Sites in the European Union/CARACAS” (1996–1998) and was funded by the European Commission under the Environment and Climate Programme [4]. Other initiatives linked to CARACAS are NICOLE (Network for Industrially Contaminated Land in Europe), an industry-led concerted action programme; RACE (Risk Abatement Centre for Contaminated Sites in Central and Eastern European Countries); the EEA’s European Topic Centre on Soil; the International Standards Organisation Technical Committee TC190 dealing with soil quality [4]; and TIMBRE (Tailored Improvement of Brownfield Regeneration in Europe), a European research project financed under the EU 7th Framework Programme, which Romania, through the National Agency for Environmental Protection and Technical University of Civil Engineering Bucharest, was part of.

As mentioned above, the sector that is mainly responsible for the organic contamination of soil in Romania is the petroleum industry, with all of its related activities (extraction, transportation, and processing of petroleum products). Crude petroleum is a complex mixture containing more than 17,000 identified chemical compounds [5]. Saturated and aromatic hydrocarbons represent the majority of non-polar fractions [5,6]. Polycyclic aromatic hydrocarbons (PAHs) are persistent in the environment due to their low water solubility and low volatility [7]. An overview of the main contaminants affecting soil in Europe indicated that, in 2012, PAHs accounted for 13.3% of the contaminants affecting soil [8]. Over the past decade, the treatment of these compounds has become a matter of interest for numerous authors [7]. In this context, various technologies such as bioremediation, chemical treatments, and physical treatments [7–10], have been applied to remediate soils, sediments, and groundwater. Despite efforts made over the last two decades, there is still no reliable technology for the remediation of contaminated soils [11,12]. Many technologies have been developed over the past years, but each handles a single type of contaminant or site condition [12]. Moreover, usually the sites are polluted simultaneously with different types of contaminants, which makes the remediation of these sites a complicated process.

This paper focuses on the application of a physical-chemical treatment that is based on an electric field applied to soils contaminated with PAHs due to crude oil pollution over time. The main variables that could influence overall treatment efficiency and were monitored during the scientific research were pH, redox potential, ionic strength, soil characteristics, voltage gradient and zeta potential. The human health risk assessment related to the concentrations of contaminants before and after the application of the treatment, was also considered in this study. Therefore, the main objectives of this study were to (1) obtain information about the soil contamination level of Total Petroleum Hydrocarbons (TPHs) and PAHs; (2) apply the risk assessment procedure in accordance with the Romanian regulations and guidelines; (3) evaluate the remediation degree with regard to the electrochemical technique used for soil decontamination; (4) obtain information about the risk level once the remediation goals are achieved; and (5) evaluate the results obtained to provide useful information for the decision-making factors involved in the proper management of contaminated sites.

2. Working Methodology

The working methodology has been divided into five phases. The first phase is the collection of samples of contaminated soil. The second phase consists of comparing the concentration of TPHs and PAHs found in the contaminated soil with those provided by the applicable Romanian regulations. The third phase involves measuring the incremental lifetime cancer risk (ILCR) resulting from PAH soil contamination by oil pollution. The next phase is the experimental research addressing the remediation of the contaminated soil, comparing new TPH and PAH concentrations in soil with allowed thresholds and determining the efficiency of the proposed treatment solution. In the final phase, the study conducted risk estimations for different remedial efficiency scenarios for the considered industrial area. In this manner, the study determined the influence of the electrochemical method, in respect to risk reduction, for different scenarios concerning the applied remediation technology: 1. no remediation; 2. electrochemical remediation for 10 days and 3. electrochemical remediation for 20 days.

2.1. Site Description, Sampling and Analysis

The soil samples were collected from an industrial area located in southeastern Romania. The contaminated soil was subjected to oil drillings, production, or storage activities. The samples of soil were taken according to the national standard procedure illustrated by State standard (STAS) 7184/1-84 [13] and following the Romanian regulation in force for the investigation of potentially contaminated sites: Order No. 184/1997 for the approval of the Environmental Balance Sheet Procedure [14]. A diagonal system for the soil sampling was considered and basic soil sampling tools, such as soil probes, were used. The number of sampling points was established according to the regulations in force, where it was specified that for an area of 1000 m², a minimum number of four sampling points is required. In the present work, five soil samples were collected from an area of 1000 m², from five different points. For the chemical analysis, the contaminated soil samples were physically combined and mixed in order to form a single homogeneous sample (according composite sampling protocol). The collected soil was stored at 4 °C, transported to the laboratory, and further chemically analyzed.

By using the standard methods, SR ISO 13877:1999 [15] and SR EN ISO 16703:2011 [16], the Polycyclic Aromatic Hydrocarbons (ΣPAHs) and the Total Petroleum Hydrocarbons (TPHs), were determined. The soil sample was passed through a sieve with approximately 2 mm openings. For the sample extraction, 6 benches SOXHLET equipment (Heidolph, Schwabach, Germany) was used. An approximate 20 g portion of contaminated soil was extracted with 250 mL of high-performance liquid chromatography (HPLC) grade petroleum ether solvent. The extract was concentrated to a low solvent volume using a Heidolph rotary evaporator and eluted with hexane. After the extraction stage, the soil sample was transferred to a capped and sealed flask for gas-chromatographic analysis [17–19]. For the identification and quantification of PAHs existing in the extracted sample, a HPLC system, model Agilent 1200 series (Agilent Technologies, Waldbronn, Germany) equipped with photodiode array detector, was used. The considered PAHs in the present work are naphthalene, acenaphthene, fluorene, phenanthrene, anthracene, fluoranthene, pyrene, benzo-a-anthracene, chrysene, benzo-b-fluoranthene, benzo-k-fluoranthene, benzo-a-pyrene, dibenzo-a, h-anthracene, benzo-ghi-perylene and indeno(1,2,3-cd)pyrene. These are found in the U.S. EPA list of 16 PAHs, issued by the U.S. Environmental Protection Agency (EPA, Washington, DC, USA) in 1976, known for their negative influence on humans [20,21].

2.2. Human Health Risk Assessment Methodology

Quantification of the Incremental Lifetime Cancer Risk (ILCR) was performed to estimate the exposure risk of soil contamination with carcinogenic chemicals. In this manner, it was possible to evaluate the risk of the investigated contaminated site and show how to use the risk-based approach to achieve the remediation goals on sites.

The methodology used for the assessment of risks on human health was the classical method with multiple pathways. This methodology is already widely used for the environmental assessment of industrial facilities and has been used in Europe. The methodology includes four key phases: 1. Hazard identification (review key research to identify any potential health problems that a chemical can cause); 2. Exposure assessment (determine the amount, duration, and pattern of exposure to the chemical); 3. Dose–response assessment (estimation of the relationship between dose, or level of exposure to a substance, and the incidence and severity of an effect); and 4. Risk characterization (assess the risk for the chemical to cause cancer or other illnesses in the general population) [22]. All of these four steps could be influenced by different factors, such as the degree of pollutant concentration in the soil, the pollutant type, and the exposure scenario. The assessment of total exposure within the framework of the risk assessment requires complex knowledge from various disciplines.

Throughout the present study, the human health risk was assessed by considering the industrial use of the soil, based on the different electrochemical solution scenarios proposed for soil remediation. Specifically, it was presumed that the electrochemically treated soil in different experimental conditions was placed to their initial location for industrial use. In these conditions, the human exposure was quantified according to U.S. EPA equations [23], taking into account exposure through accidentally ingested soil and dermal contact, as follows:

$$\text{Dose}_{\text{s.i.}} = (C_{\text{cont}} \times IR \times FI) \times (BW^{-1}) \times ((ED \times EF) \times AT^{-1}) \times CF \quad (1)$$

Therefore, exposure through accidental ingestion of contaminated soil ($\text{Dose}_{\text{s.i.}}$ —mg/kg/day) was calculated considering the concentration of the contaminant in the soil (C_{cont} —mg/kg_{dw}), the ingestion rate (IR —mg soil/day), and the fraction ingested from the contaminated source (FI —unitless) [23].

$$\text{Dose}_{\text{d.c.}} = ((C_{\text{cont}} \times SA \times AF \times ABS) \times BW^{-1}) \times ((EF \times ED) \times AT^{-1}) \times CF \quad (2)$$

For the exposure through dermal contact ($\text{Dose}_{\text{d.c.}}$ —mg/kg/day), the parameters considered for dose calculation are the concentration of the contaminant in the soil (C_{cont} —mg/kg_{dw}), the surface area of skin available for contact (SA —cm²/day), the soil-to-skin adherence factor (AF —mg/cm²), and the fraction absorbed across the skin (ABS —unitless) [23].

From both equations, BW represents body weight, CF represents the conversion factor (10^{-6} kg/mg), EF represents exposure frequency (days/year), ED represents exposure duration (years), and AT represents average time (days). Incremental lifetime cancer risk (ILCR) can be expressed as:

$$\text{ILCR} = D \times SF \quad (3)$$

where ILCR represents Individual Cancer Risk (unitless), D represents the daily dose (mg/kg/day), and SF represents the slope factor ((mg/kg/day)⁻¹).

2.3. Electrochemical Treatment

As part of the electrochemical (electrokinetic) treatment, a low-density direct current was applied by two electrodes inserted into the polluted soil. A current promotes the movement of pollutants in the pore fluid towards the electrode chambers, where they are finally collected and treated. Therefore, this technique is recommended for application with fine and low-permeability soils [24–27] and has been mainly applied to the treatment of soils polluted by metals, due to the property to move the ionic species [28]. This treatment has emerged as an alternative to conventional soil treatments due to its peculiar advantages, namely the capability of treating fine and low-permeability materials, as well as achieving a high yield in the removal of salt content and inorganic and organic pollutants [29]. The application of this electric field-based treatment involves the use of a constant voltage applied to two electrodes (one is the anode and the other is the cathode). Several physical and chemical processes take place during the treatment implementation. The physical processes, which result in the mobilization and transportation of contaminants towards the anode or the cathode electrodes

for subsequent removal out of the contaminated soil [30], are electromigration and electroosmosis. Electromigration is the movement of ionic species in an electric field towards the electrode of opposite charge [27,30]. The electroosmotic flux represents the movement of water mass towards the cathode, under the influence of the electric field [27,30,31]. Regarding the chemical processes that take place, we can mention the electrolysis of water, adsorption/desorption of contaminants on the soil particle surfaces, redox reactions, and acid/basic reactions [27,30].

In order to develop the electrokinetic research, an experimental setup was used, consisting of an electrochemical cell, with dimensions 150 mm × 150 mm × 150 mm (L × l × h), with electrodes inside connected to a power supply. The overall diagram of the experimental system is shown in Figure 1.

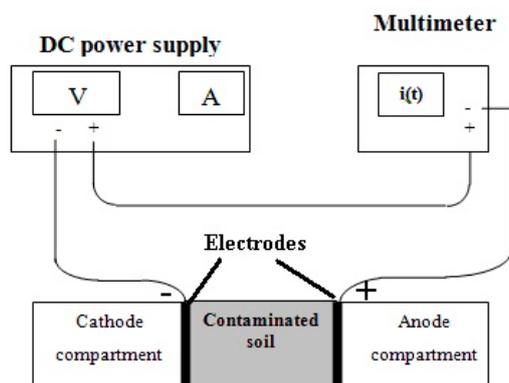


Figure 1. Diagram of the experimental setup.

The contaminated sample was inserted into the electrochemical cell and was further compacted so that no empty space could exist. In order to ensure the occurrence of all physical and chemical processes, the humidity of the soil was maintained at a minimum of 20–30% by adding water when necessary. The experiment lasted for 20 days. Bulk soil samples were collected from the anode and cathode areas, over the entire depth of the cell. The samples were then mixed in order to obtain a homogenous probe. An intermediate soil was created in the same manner, by sampling 10 days after the experiment was started. The results from this experiment, and the final results, are illustrated in the following paragraphs.

3. Results and Discussion

3.1. Electrochemical Removal of Hydrocarbons from the Contaminated Soil

The main characteristics of the investigated soil were that the pH was around 5.87 (there is a tendency for the acidification of the soil sample) and the humidity was around 14.82%. The concentration level of TPHs and PAHs identified in the soil are presented in Table 1.

Table 1. Initial characterization of soil sample [mg/kg_{dw}].

Analyzed Parameter	Industrial Areas		Determined Values
	Alert Threshold	Intervention Threshold	
Total Petroleum Hydrocarbons (TPHs)	1000	2000	14,725
Polycyclic Aromatic Hydrocarbons (ΣPAHs)	25	150	103.16

As illustrated in Table 1, the results from the analytical procedures, compared with the regulation in force, revealed that pollution levels exceeded the alert thresholds for the industrial use of soil and were below the intervention thresholds. As the hazard level was not given by the concentration level of contaminants (TPHs concentration in soil = 14,725 mg/kg_{dw} with respect to ΣPAHs = 103.16 mg/kg_{dw}),

but by the degree of toxicity of the analyzed compounds, Benzo[a]pyrene (BaP) was the most toxic of the parent PAHs with a Slope Factor of $1.2 \times 10 \text{ (mg/kg/day)}^{-1}$, and deeper investigation was necessary. A remediation solution for the contamination problem was proposed and risk assessment before and after the remediation process, was considered. The electrochemical treatment was tested in laboratory-scale experiments.

The experimental research was conducted for 20 days with a soil quantity of 6 kg. A voltage of approximately 15 V was applied. During the experiments, a series of parameters were monitored: pH, redox potential (ORP), current intensity, energy consumption, and treatment efficiency.

The values for the pH ranged between 3.93 and 10.32, but the main aim was to obtain values close to 7 in order to create a more suitable environment in the soil after the application of the electrochemical remediation. For this purpose, the ORP was monitored and when the values went near or below 0, the polarity of the experiment was changed. ORP was assessed to have better control of the oxidation reactions that took place during the experiment (see Figure 2). The influence of ORP on pH was a secondary consequence.

To ensure proper development of the electrochemical processes (improving the electrical conductivity and consequently stimulating the oxidation and reduction reactions), 20–25% of soil moisture content was obtained by adding a suitable amount of deionized water. The higher the soil moisture content, the better the electrochemical processes are expected to function. Additionally, the oxidation reactions occurred near the anode area and, as a result, the ORP values suggested that the oxidation reactions occurred more rapidly than the reduction reactions taking place near the cathode area. A uniform decontamination of the soil requires a change in the polarity of the electrodes to increase the treatment efficiency. A consequence of this approach is also the non-uniformity of the pH in the entire soil sample (usually, in the anode area there is an acid front and in the cathode area there is a basic front). The same behavior is also identified with regard to the ORP parameter.

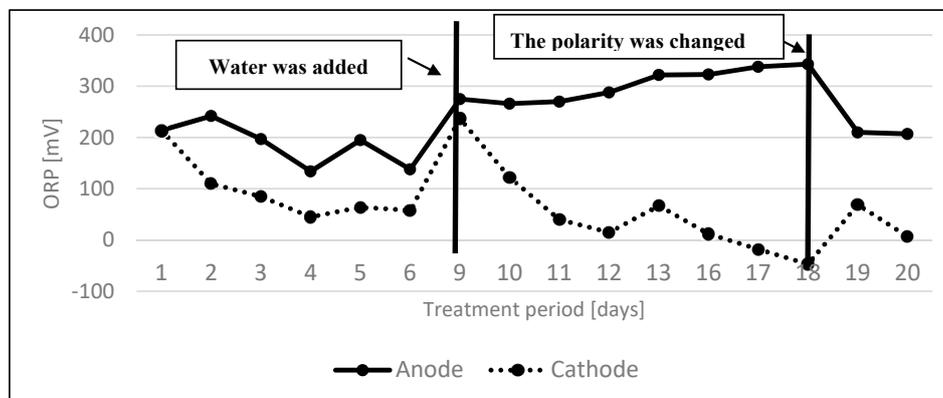


Figure 2. Redox potential (ORP) trend in the experimental research.

During the experiment, the current intensity was measured. The maximum value was 0.19 A, recorded at the beginning of the experiment, while the minimum value was 0.03 A. Energy consumption plays an important role in the development of sustainable technologies for the remediation of contaminated soils, with real consequences on cost-effectiveness. Therefore, another parameter that was measured was energy consumption, using an electronic power meter—type KGS02-01/1109. After 10 days, 6.5 kWh were consumed, while at the end of the experimental study, this parameter reached a value of approximately 14.7 kWh. Obviously, as shown below, an increase in the envisaged remediation time (at the same voltage) allowed for better removal of the contaminant, but this improvement was counterbalanced by the fact that a higher time of remediation (e.g., 20 days or longer) results in higher energy expenditure and a higher remediation strategy cost.

The treatment efficiency was the parameter that had real importance in determining the success of the conducted experiment. Figure 3 shows that the degradation efficiency of the electrochemical

solution for Σ PAHs removal had the same order of magnitude as for the TPHs. On the other hand, the degradation efficiency was higher for both Σ PAHs and TPHs when the time of soil electrical treatment was increased from 10 to 20 days. It was demonstrated that the removal of PAHs could reach from 16% (after 10 days of the remediation process) to 46% (after 20 days of the remediation process), while removal of the heavy TPHs could reach from 17% (after 10 days of the remediation process) to 50% (after 20 days of the remediation process). Concerning the trend of the remediation degree of the tested method, it was observed that even by increasing the period, the results were better at the end of the treatment period (the initial concentration of PAHs in the soil was 103.6 mg/kg_{dw}, decreasing to 87.06 mg/kg_{dw} after 10 days (ElectroChemical treatment after 10 days—EC10) and to 55.49 mg/kg_{dw} after 20 days (EleCtrochemical treatment after 20 days—EC20); and the initial concentration of TPHs in the soil was 14,725 mg/kg_{dw}, decreasing to 12,242 mg/kg_{dw} in EC10 and to 7324.50 mg/kg_{dw} in EC20) and the remediation degree from one stage of the process to another dropped significantly. An overview of the tendency of the remediation degree to correlate with soil remediation time, is shown in Figure 4. The trend for both TPHs and Σ PAHs was observed where the remediation degree was higher in the second part of the experiment and increased as the time of remediation increased. These results are maintained by different authors in the literature [29–31], which indicates that electrochemical treatment efficiency increases along with an increasing treatment period.

Nationally, accordingly to the National Strategy and the National Action Plan for the Management of Contaminated Sites in Romania [32], the highest number of contaminated sites is found in the oil extraction industry (738 sites), with an additional 178 sites in the extractive industry and 61 sites in the petroleum industry. These numbers show that the decontamination of polluted sites is, and will remain in the coming years, a major issue for Romania. Even in the absence of accurate information on the contamination degree of the contaminated sites mentioned above, the results obtained during the present research could be considered useful for the future management of this environmental issue.

As mentioned above, our choice was to assess the degradation efficiency of TPHs and PAHs in a pilot-scale electrochemical cell, with lab-scale experiments, in order to analyze and evaluate the effects and optimal design of an electrochemical solution for the remediation of contaminated sites. For this purpose, we collected typical soil samples and remediated them in the laboratory to obtain clear information with better control of the process. Consequently, the interferences that may occur during field testing were minimized, and we obtained significant results prior to applying the electrochemical solution to the in-field remediation application. Of course, a soil map indicating the PAHs in excess would be better to clearly describe the situation of the contaminated sites in the framework of real-scale applications. Additionally, it must be specified that, during previous research [10,25,27], soil samples were usually collected from areas near the two electrodes.

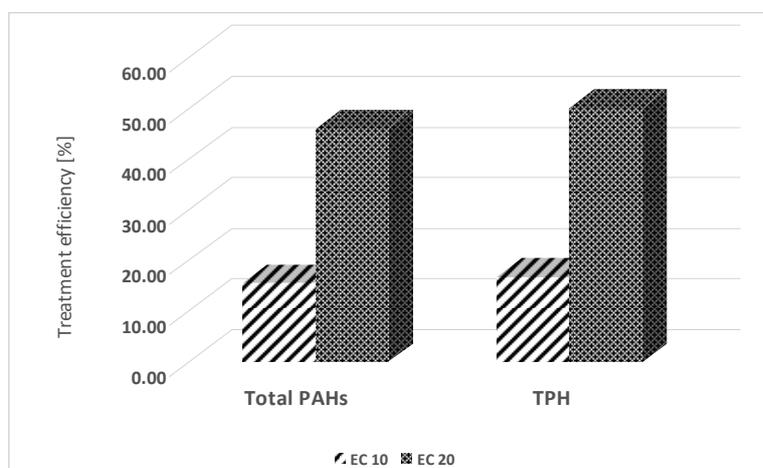


Figure 3. Degradation efficiency of PAHs and TPHs in soil across the electrochemical remediation.

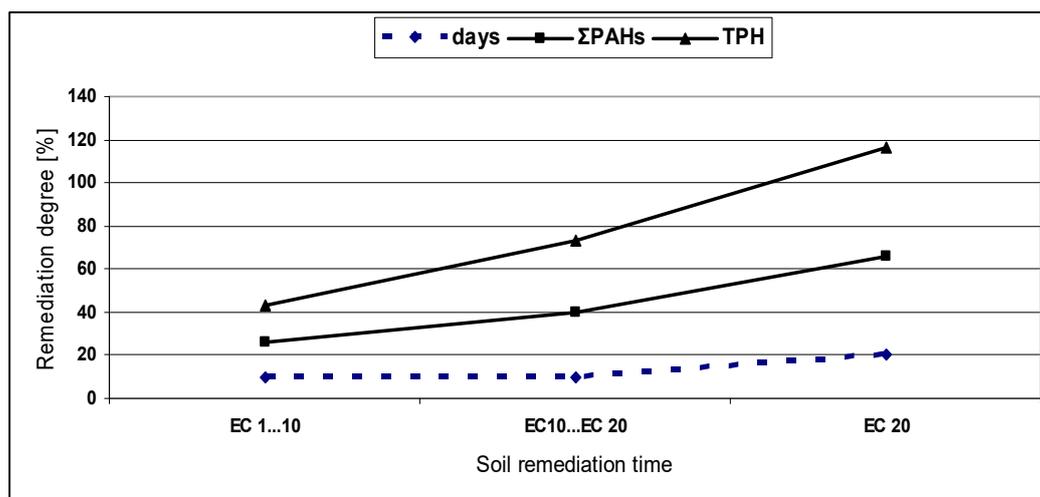


Figure 4. Remediation degree of PAHs and TPHs in soil as a function of time.

3.2. Risk Assessment of PAH-Contaminated Soil

Further, health risk assessment was performed for each considered scenario for the industrial use of soil. The exposure to PAHs in soil was assessed since it has been demonstrated that exposure to some of these contaminants can cause an increase in the incidence of cancer. The circumstances of exposure that were taken into account, in the industrial use of soil, were the accidental ingestion of contaminated soil and dermal penetration, correlated with the exposure conditions in the framework of the industrial use of soil. The exposure parameters were selected considering the equations provided by the U.S. EPA [33], as illustrated in Table 2. The Incremental Lifetime Cancer Risk was calculated for industrial workers for each exposure scenario. For the assessment of ILCR, the Slope Factor was equal to $1.2 \times 10 \text{ (mg/kg/day)}^{-1}$, defining the Benzo[a]pyrene (BaP) compound, the most toxic PAH according to the Californian Environmental Protection Agency, Office of Environmental Health Hazard Assessment (OEHHA) [33]. Considering this, we selected the most conservative approach for the assessed scenarios.

Table 2. Exposure parameters [34,35].

Parameter Category	Units	Industrial Workers
Common parameters		
Body weight (BW)	kg	80
Average time (AT)	days	9125
Lifetime (LT)	years	70
Exposure frequency (EF)	days/year	250
Exposure through accidental soil ingestion—specific parameters		
Soil ingestion rate (IR)	mg/day	50
Fraction ingested from contaminated source (adults)	unitless	0.3
Exposure through dermal contact—specific parameters		
Skin surface area available to contact (SA)	cm ² /d	3527
Soil-to-skin adherence factor (AF)	mg/cm ²	0.2
Absorption factor (ABS)	unitless	0.13

The accepted level of cancer risk has been a subject widely discussed in the literature [36,37]. If, initially and generally, an additional lifetime cancer risk of one in a million (10^{-6}) was considered most frequently as the acceptable value for individual cancer risk, the management of risks posed by soil contamination later changed this consideration. In the United States, the National Oil and

Hazardous Substances Pollution Contingency Plan (NCP) [38] provides information on cancer risk indices related to Superfund site remediation and indicates an acceptable cancer risk range of 10^{-4} to 10^{-6} . This typical range is indicated also by other authorities and authors in different countries [39,40].

The majority of PAHs have been classified by the U.S. EPA as probable human carcinogens, being commonly believed and demonstrated through different research studies, to have a real contribution to the development of human cancers [41]. Evidence indicates that mixtures of PAHs are carcinogenic to humans. Initially, evidence arose from occupational studies on workers exposed to mixtures containing PAHs, and further long-term studies have demonstrated an increased risk of, predominantly, skin and lung, but also bladder and gastrointestinal, cancers [42].

In the present study, the carcinogenic risk for industrial workers' receptors showed that the worst scenario is where no remediation method is applied. In this scenario, the Incremental Lifetime Cancer Risk is 1.13×10^{-3} (as illustrated in Table 3). Consequently, the assessed ILCR exceeds the acceptable cancer risk range of 10^{-4} to 10^{-6} .

Table 3. Health risk assessment based on different experimental scenarios of the electrochemical experiments.

Experimental Conditions	Contaminated Soil without Remediation	Electrochemical Treatment (10 Days)	Electrochemical Treatment (20 Days)
Scenarios	Scenario 1	Scenario 2	Scenario 3
Incremental Lifetime Cancer Risk	1.13×10^{-3}	9.54×10^{-4}	6.08×10^{-4}

The remediation method of the electrochemical treatment for 10 days showed a decrease in risk from 1.13×10^{-3} to 9.54×10^{-4} . This is in line with the acceptable range of increased cancer incidence risk, but afterwards the electrochemical treatment duration was extended to 20 days in order to assess to what extent the experimental conditions have an influence on ILCR. The increase in the treatment period to 20 days led to a reduction in ILCR of 46% (Table 3).

The results showed that, in the considered exposure scenario, ILCR is strictly related to the concentration of PAHs in the soil. The application of the proposed remediation method in the present research led to a 15.61% reduction in PAHs in the soil after 10 days and a 46.21% reduction after 20 days. The same trend was observed with regard to ILCR. Therefore, in the case of *no treatment*, EC10: EC20 corresponded to PAH concentrations of 103.16 mg/kg_{d,w}: 87.06 mg/kg_{d,w}: 55.49 mg/kg_{d,w}, and ILCR values of 1.13×10^{-3} : 9.54×10^{-4} : 6.08×10^{-4} .

3.3. Influence of Electrochemical Treatment of Soil on Hydrocarbon Degradation Efficiency and Risk to Humans

The risk assessment performed for the investigated site, if no remedying method was applied and for different scenarios of the proposed remediation method, calculated various levels of risk for the same exposure conditions and the use of soil (industrial use of soil). The remedying methods, with the soil concentration observed and related risk levels, are presented in Table 4.

Table 4. Level of risk according to the concentration of PAHs in the soil.

Test	Concentration in Soil [mg/kg _{d,w}]	Risk Level
Without remediation	103.16	1.13×10^{-3}
EC10	87.06	9.54×10^{-4}
EC20	55.49	6.08×10^{-4}

Figures 5 and 6 indicate that the degradation of TPHs and PAHs shows similar behavior throughout the experimental research, while the ILCR from PAH-contaminated soil decreases when the TPH concentration in the soil is reduced. As a general trend from the experiments performed,

a higher period of electroremediation gave a lower ILCR value. It is true that a longer electrochemical remediation time results in a lower concentration of contaminants in soil with a positive influence on the ILCR, but, in this case, deeper reflection on the cost–benefit analysis is required.

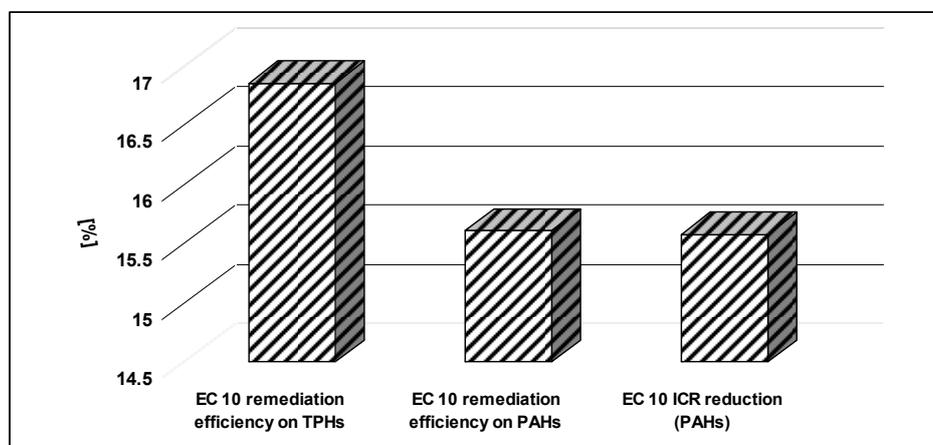


Figure 5. Degradation efficiency of PAHs and TPHs in soil during the electrochemical remediation.

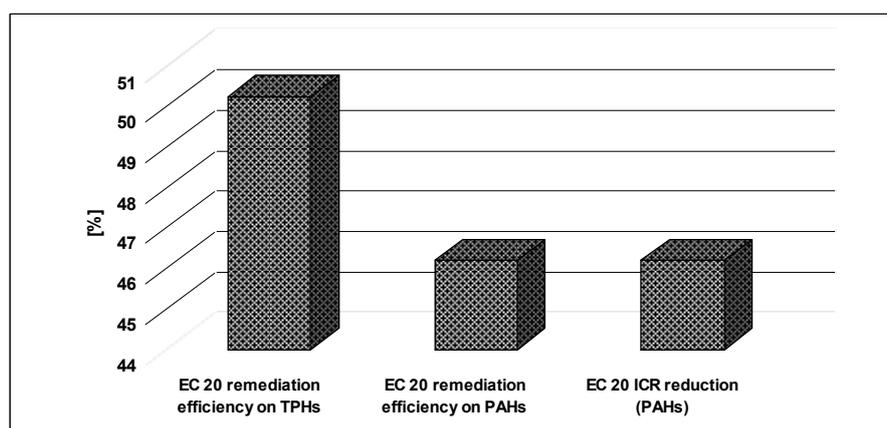


Figure 6. Degradation efficiency of PAHs and TPHs in soil during the electrochemical remediation.

The evaluation of the efficiency of hydrocarbon degradation in soil based on different experimental scenarios of the electrochemical method, showed, over time, the important relationship between the removal of total petroleum hydrocarbons and the reduction of the considered polycyclic aromatic hydrocarbons from the soil (Table 5).

Table 5. Efficiency of hydrocarbon degradation in soil based on different experimental scenarios of the electrochemical experiments.

Experimental Conditions Efficiency on Hydrocarbon Degradation in Soil	Electrochemical Treatment (10 Days)	Contaminated Soil without Remediation after Electrochemical Treatment (20 Days)
ΣPAHs	15.61 [%]	46.21 [%]
TPHs	16.87 [%]	50.26 [%]

4. Conclusions

The assessment of risk to humans from our studied contaminated site indicated that a contamination of 14,275 mg/kg_{dw}, corresponding to a concentration of PAHs equal to 103.16 mg/kg_{dw},

caused a risk level of 1.13×10^{-3} . A removal rate of 16.87% for TPHs and 15.61% for PAHs was obtained when the electrochemical remedial method was applied for a test period of 10 days, and when doubling the treatment period, rates of 46.21% for TPHs and 50.26% for PAHs, were obtained. The reduction of TPHs in soil, and as a consequence the decrease of PAH concentration levels, resulted in an acceptable value of incremental lifetime cancer risk (ILCR). The results obtained showed that ILCR depends on the considered scenario and is directly related to the remediation rate of PAHs in the soil: a PAH reduction of 16.61% (Scenario 2) led to a 15.58% decrease in ILCR, while a PAH reduction of 50.26% (Scenario 2) led to a 46.20% decrease in ILCR.

The results of the risk assessment obtained for the considered scenarios can be useful in making decisions with respect to the management of contaminated sites, but also in improving the assessment of soil quality by including hazardous contaminants, such as carcinogenic PAHs. In this manner, it is possible to identify risk-based strategies to promote high-efficiency remediation methods with a low impact on the environment.

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