

Article



Supplementary Materials: Testing Metal–Organic Framework Catalysts in a Microreactor for Ethyl Paraoxon Hydrolysis

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Figure S1. TEM images of 1.

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Figure S2. TEM-EDS-elemental mapping and EDX-elemental distributions of 1.



Figure S3. SEM images of 1.

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Figure S4. TEM images of 1a.



Figure S5. TEM-EDS-elemental mapping and EDX-elemental distributions of 1a.

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Figure S6. SEM images of 1a.



Figure S7. TEM images of 4.



Figure S8. TEM-EDS-elemental mapping and EDX-elemental distributions of 4.

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Figure S9. TEM-EDS-elemental mapping and EDX-elemental distributions of 4a.



Figure S10. SEM images of 4a.





Figure S11. TEM images of 3.





Figure S12. TEM images of 3a.



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Figure S13. TEM-EDS-elemental mapping and EDX-elemental distributions of 3a.



Figure S14. HR-TEM and EDS elemental mapping images with EDX pattern (left) and FE-SEM images (right) of **3a**.



Figure S15. PXRD patterns of 1, 1a and 1a^{AR} (left) and 4 and 4a (right).

ARAfter catalytic run.



Figure S16. PXRD patterns of 2 and 2a (left), and 3 and 3a (right).

MOF	BET surface area (m ² g ⁻¹)	Pore volume (cm ³ g ⁻¹)
1	222	0.26
1a	180	0.23
1a ^{AR}	216	0.20
2a	116	0.09
3a	34	0.06
4	217	0.26
4a	204	0.24

Table S1. BET surface areas and pore volume for prepared MOFs.

ARAfter catalytic run.



Figure S17. N2 adsorption-desorption isotherms.



Figure S18. UV-Vis calibration curve of p-nitrophenol.



Figure S19. Reaction yield using MOF catalysts **1** at varied flow rates and catalyst particle sizes of reactor 1 (**1a**) and reactor 2 (**1b**).



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