

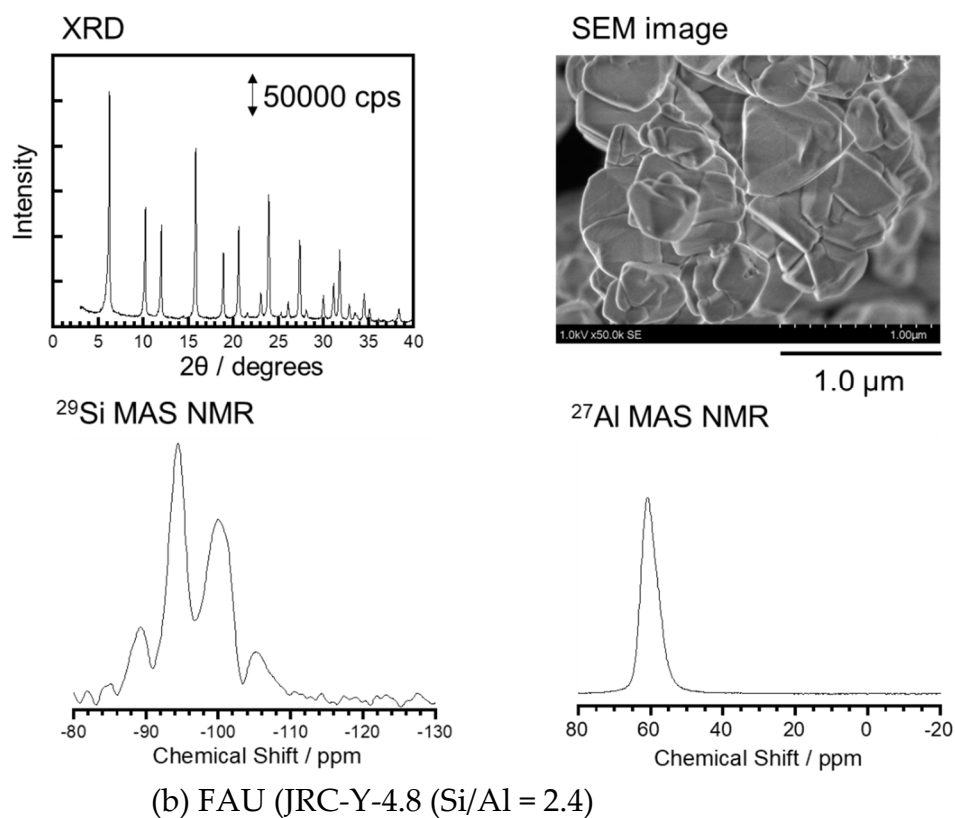
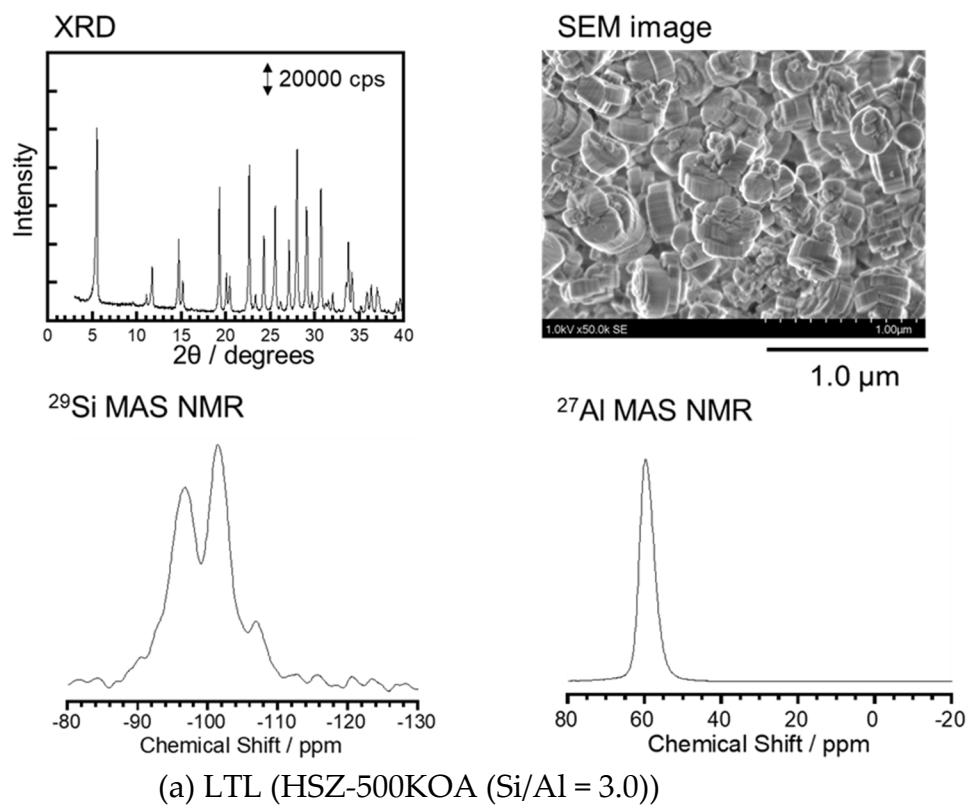
CHA-type zeolite prepared by interzeolite conversion method using FAU and LTL-type zeolite: effect of the raw materials on the crystallization mechanism, and physicochemical and catalytic properties

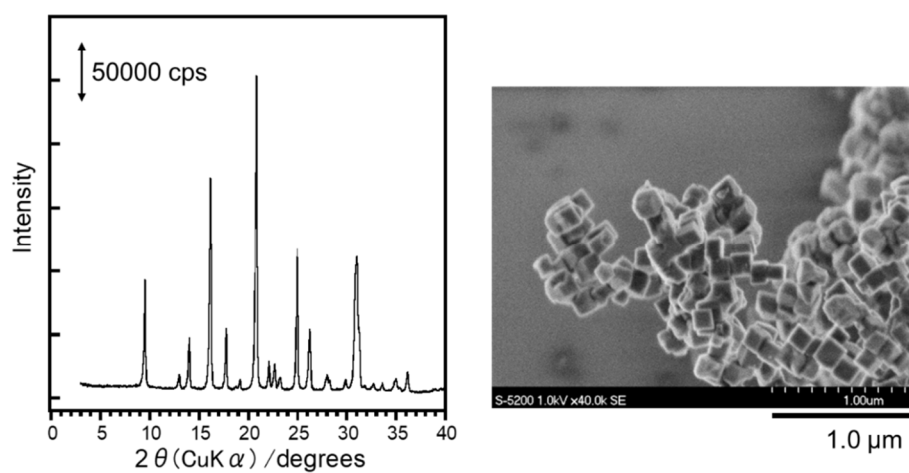
Toshiki Nishitoba ¹, Takuya Nozaki ¹, Sungsik Park ¹, Yong Wang ¹, Junko N. Kondo ¹, Hermann Gies ^{1,2}, Toshiyuki Yokoi ^{*1}

¹ Institute of Innovative Research, Tokyo Institute of Technology, 4259 Nagatsuta, Midori-ku, Yokohama 226-8503, Japan;

² Institute of Geology, Mineralogy und Geophysics, Ruhr-University Bochum, 44780, Germany

* Correspondence: yokoi@cat.res.titech.ac.jp; Tel.: +81-45-924-5430, Fax: +81-45-924-5431





(c) Seed crystal (CHA-type zeolite)

Figure S1. Physicochemical properties of (a) LTL (HSZ-500KOA (Si/Al = 3.0)), (b) FAU (JRC-Y-4.8 (Si/Al = 2.4) and (c) seed crystal (CHA-type zeolite) used as parent zeolite for the synthesis of CHA.

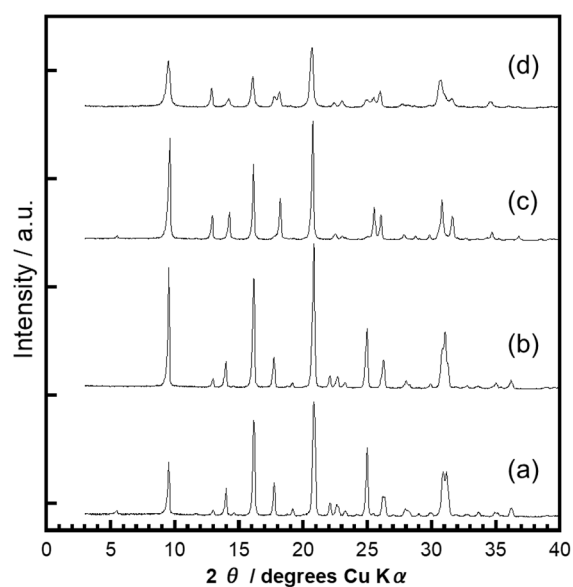


Figure S2. XRD patterns of (a) CHA-LTL-TMAda, (b) CHA-FAU-TMAda, (c) CHA-LTL-TEA, (d) CHA-FAU-TEA.

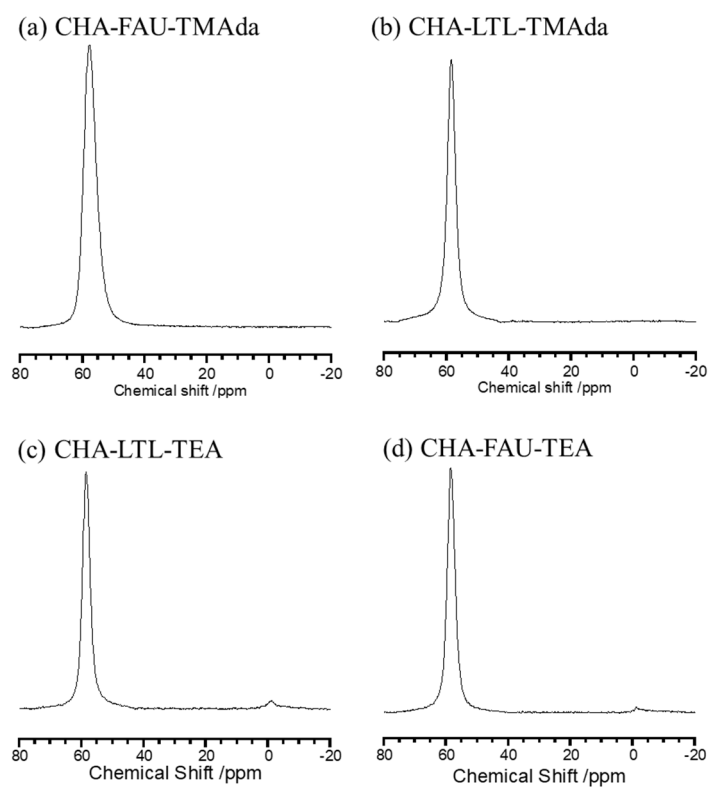


Figure S3. ^{27}Al MAS NMR spectra of the calcined Na-type products: (a) CHA-FAU-TMAda, (b) CHA-LTL-TMAda, (c) CHA-LTL-TEA, (d) CHA-FAU-TEA.

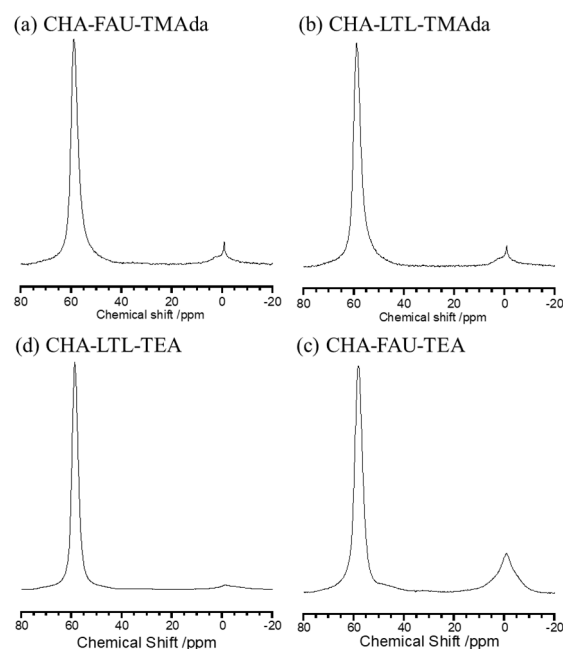


Figure S4. ^{27}Al MAS NMR spectra of the H^+ - type products: (a) CHA-FAU-TMAda, (b) CHA-LTL-TMAda, (c) CHA-FAU-TEA, (d) CHA-LTL-TEA.

Table S1. The products' selectivities in the MTO reaction over CHA-LTL-TMAda, CHA-FAU-TMAda, CHA-LTL-TEA, and CHA-FAU-TEA.

Catalyst	Acid amount ^a / mmol g ⁻¹	TOS ₉₅ ^b / min	Product selectivity (C-atom %) ^b					
			C2=	C3=	C4=	Paraffins (C1-C4)	DME	Over C5
CHA-LTL-TMAda	0.46	180	56.7	28.8	6.4	6.0	0.3	1.9
CHA-FAU-TMAda	0.47	240	55.9	28.9	7.7	4.7	2.0	0.9
CHA-LTL-TEA	0.80	180	51.6	33.4	6.3	4.6	2.8	1.3
CHA-FAU-TEA	1.17	120	48.0	30.8	5.9	5.2	8.9	1.1

a; Estimated by the NH_3 -TPD, b; TOS₉₅ indicates TOS (time on Stream) required to achieve methanol conversion drop below 95%.