

Photocatalytic Degradation of Chlorpyrifos with Mn-WO₃/SnS₂ Heterostructure

Charlie M. Kgoetlana, Soraya P. Malinga and Langelihle N. Dlamini *

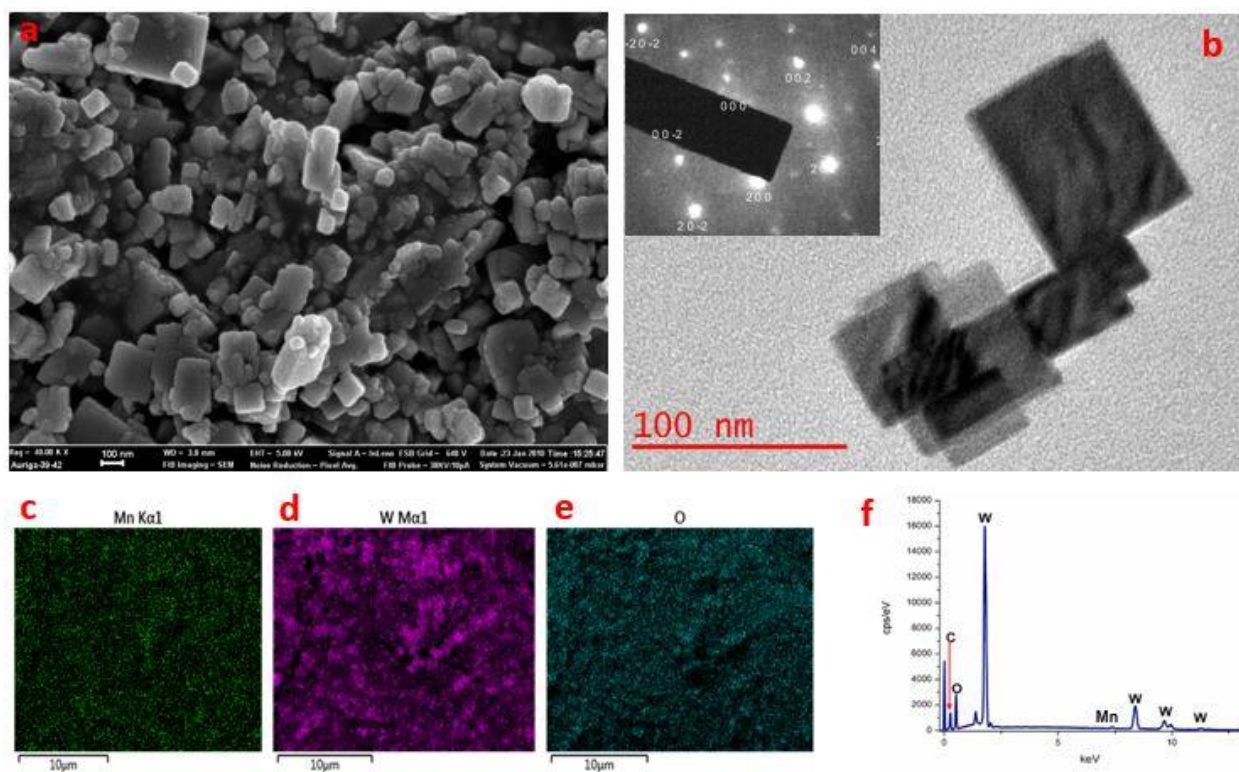
Department of Chemical Sciences, University of Johannesburg, Doornfontein Campus, P.O. Box 17011, Doornfontein, Johannesburg 2028, South Africa; kgoetlanacm@gmail.com (C.M.K.); smalinga@uj.ac.za (S.P.M.)

* Correspondence: lndlamini@uj.ac.za; Tel.: +011-559-6945

Received: 21 May 2020; Accepted: 04 June 2020; Published: date

Table S1: Average crystallite sizes of nanomaterials

Material	Average crystallite size (nm)
Pristine WO ₃	43
Mn-WO ₃	40
Mn-WO ₃ /SnS ₂	40
WO ₃ /SnS ₂	40
Pristine SnS ₂	17



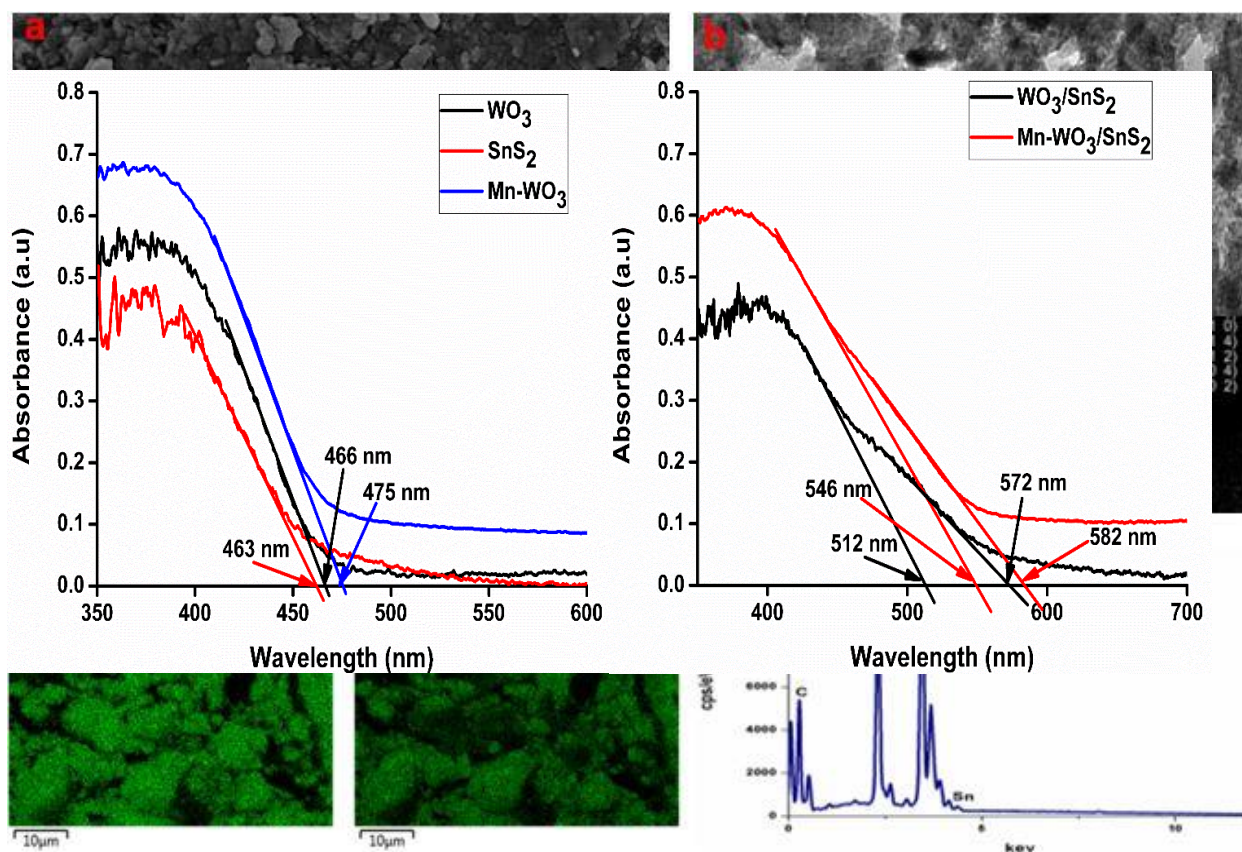


Figure S3: (a) FESEM image of pristine SnS₂, (b) TEM image of SnS₂ (inset is the corresponding SAED image), (c)-(d) elemental mapping, and (e) EDX spectrum of SnS₂ nanoparticles

Figure S4: Absorption spectra of WO_3 , SnS_2 , Mn-WO_3 , WO_3/SnS_2 , and $\text{Mn-WO}_3/\text{SnS}_2$

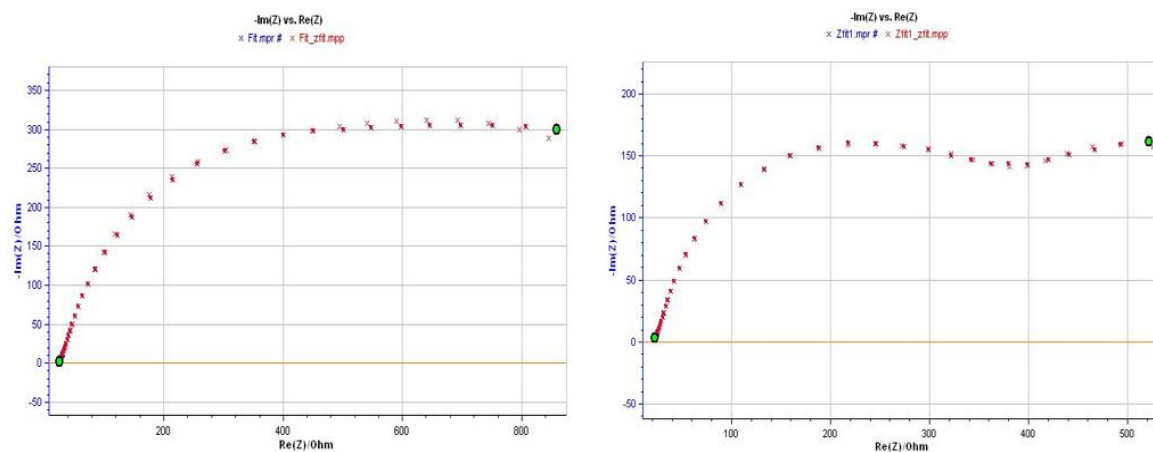


Figure S5: EIS spectra showing the fitted spectra when obtaining the Randles circuit for (a) WO_3 and Mn-WO_3 , and (b) SnS_2 , WO_3/SnS_2 and $\text{Mn-WO}_3/\text{SnS}_2$

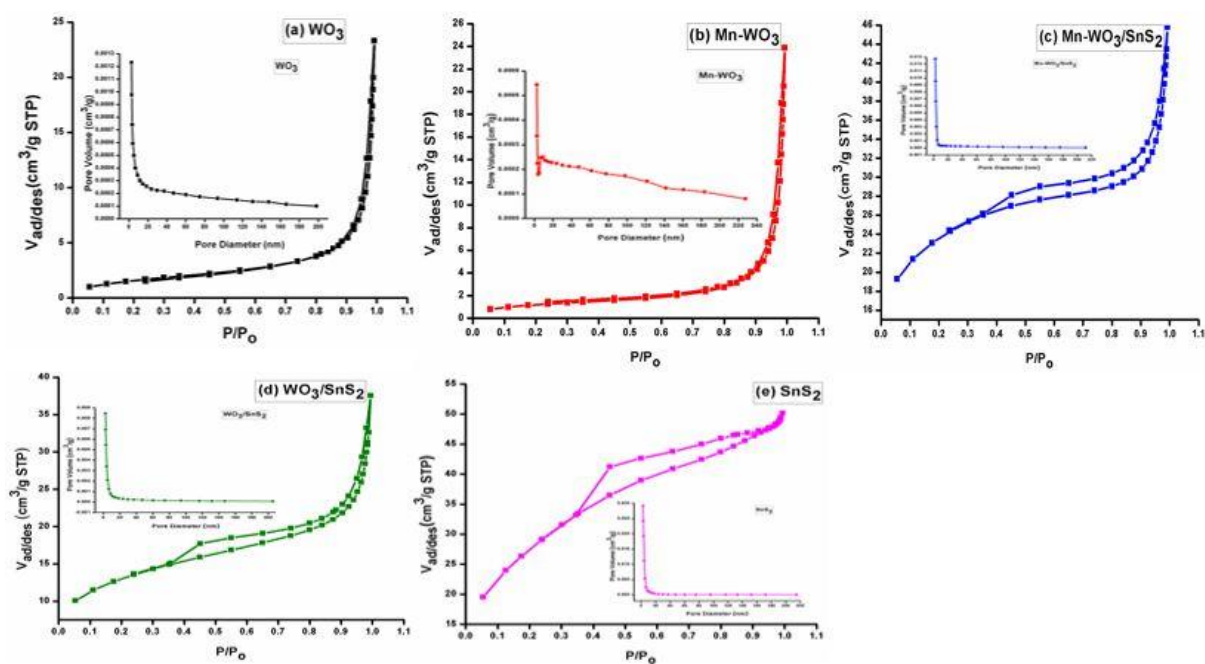


Figure S6: (a-e) N₂ adsorption-desorption isotherm of (a) WO₃, (b) Mn-WO₃, (c) Mn-WO₃/SnS₂, (d) WO₃/SnS₂, and (e) SnS₂ (**insets** are pore volume graphs)

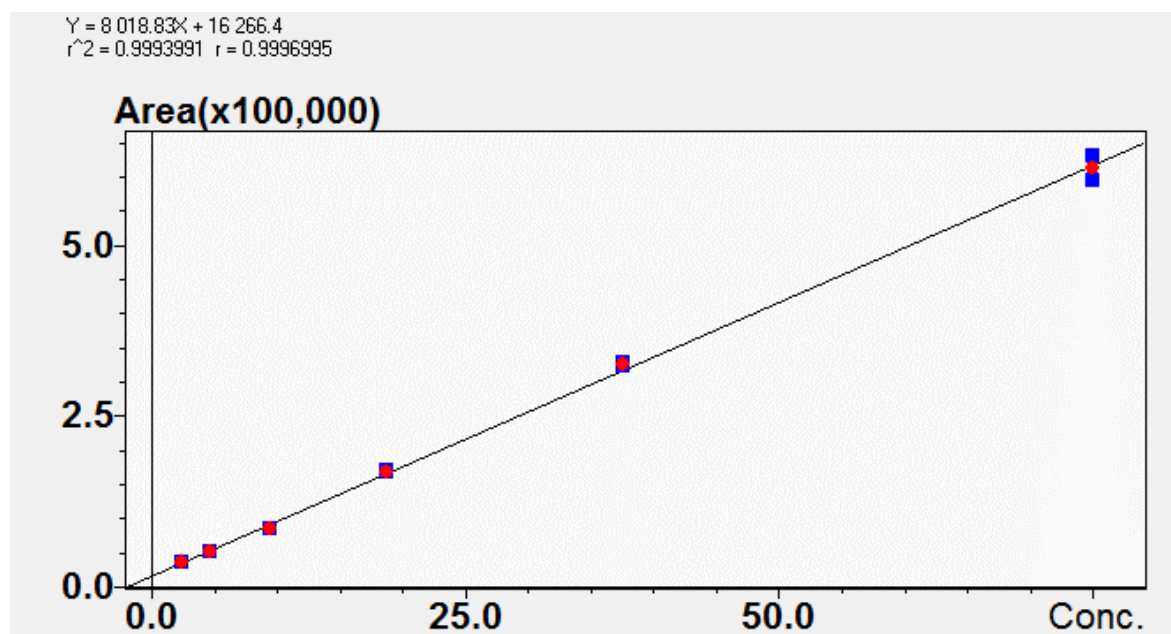


Figure S7: Calibration curve of chlorpyrifos standards from 3.125 to 75 ppb

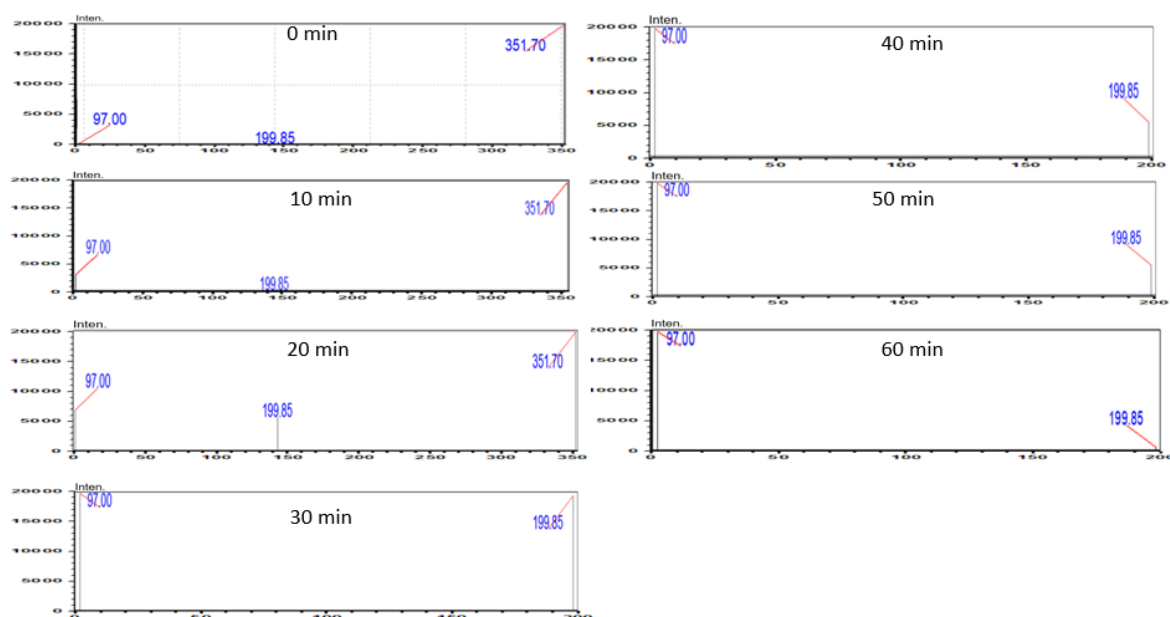


Figure S8: Mass spectra showing m/z ratios from 0 to 60 minutes

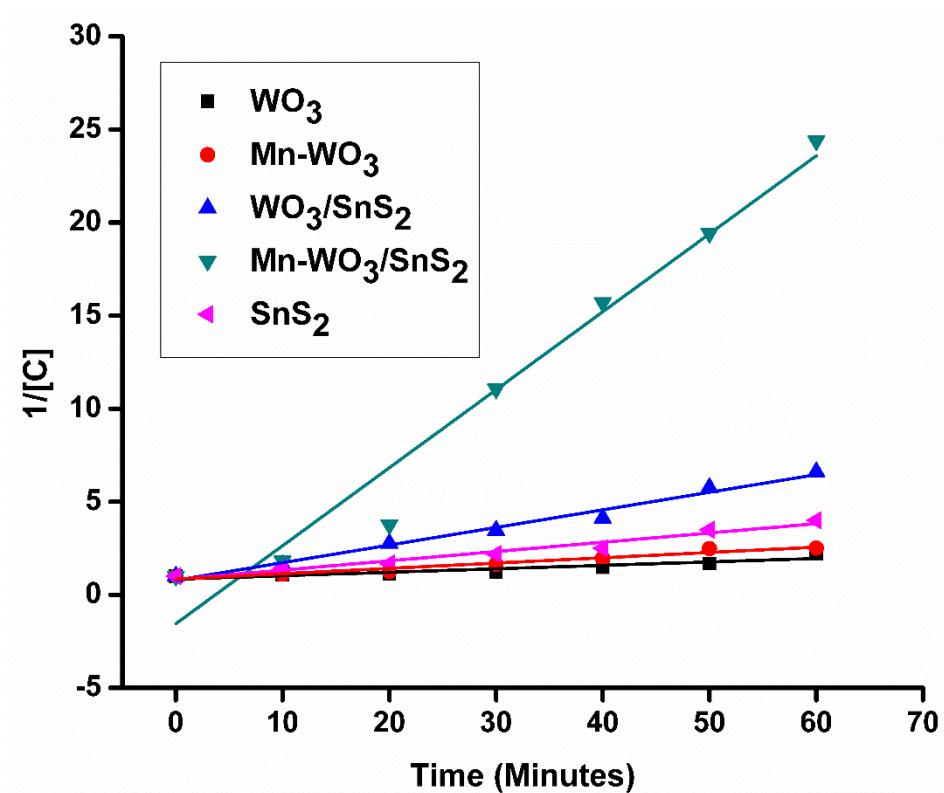


Figure S9: Fitted second order reaction kinetics graphs of the nanoparticles