

Supplementary Materials

Synthesis of Magnetic α -Fe₂O₃/Rutile TiO₂ Hollow Spheres for Visible-Light Photocatalytic Activity

Zhongli Zhou ¹, Hang Yin ¹, Yuling Zhao ^{1,2}, Jianmin Zhang ^{3,4}, Yahui Li ¹, Jinshi Yuan ¹, Jie Tang ^{5,*} and Fengyun Wang ^{1,2*}

¹ College of Physics; Qingdao University, No. 308 Ningxia Road, Qingdao 266071, China; zhouzl1019@163.com (Z.Z.); hangyinccc@gmail.com (H.Y.); zhaoyuling@qdu.edu.cn (Y.Z.); lyh_kitou@163.com (Y.L.); yuanjinshi@foxmail.com (J.Y.)

² State Key Laboratory of Bio Fibers and Eco Textiles; Qingdao University, No. 308 Ningxia Road, Qingdao 266071, China; zhaoyuling@qdu.edu.cn (Y.Z.)

³ National Engineering Research Center for Intelligent Electrical Vehicle Power System, Qingdao University, No. 308 Ningxia Road, Qingdao 266071, , China; jamiede@163.com

⁴ School of Mechanical and Electrical Engineering, Qingdao University, No. 308 Ningxia Road, Qingdao 266071, China;

⁵ National Institute for Materials Science, 1-2-1 Sengen, Tsukuba 3050047, Japan

* Correspondence: tang.jie@nims.go.jp (J.T.); fywang@qdu.edu.cn (F.W.)

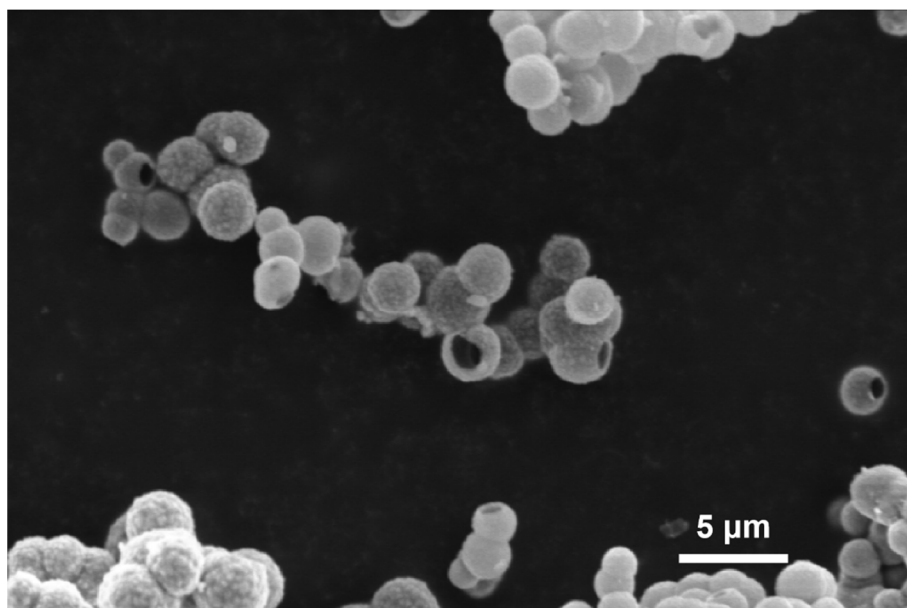


Figure S1. SEM image of α -Fe₂O₃/Rutile TiO₂ (0.025 mM) placed after 3 months. The morphology and structure of the as synthesized sample is almost no change.

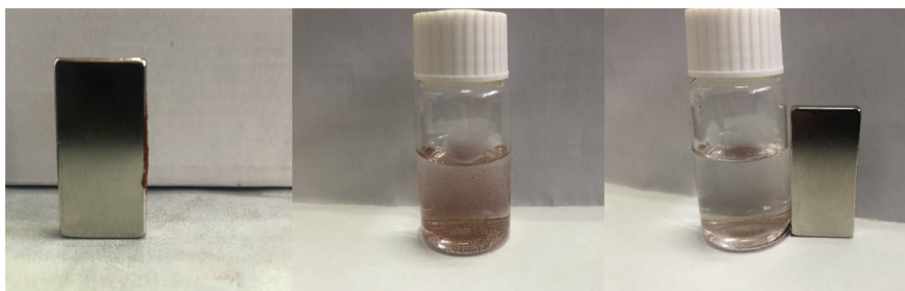


Figure S2. Image of Magnet adsorbs $\alpha\text{-Fe}_2\text{O}_3/\text{Rutile TiO}_2$ (0.025 mM).

Table S1. The values of remnant magnetization (Mr) and coercivity (Hc) of the samples.

Samples	Time (h)	Mr (emug ⁻¹)	Hc (Oe)
α -Fe ₂ O ₃ /RT (0 mM)	48 h	0.005	75
α -Fe ₂ O ₃ /RT (0.025 mM)	48 h	0.042	565.74
α -Fe ₂ O ₃ /RT (0.042 mM)	48 h	0.029	669.65
α -Fe ₂ O ₃	48 h	0.37	372
RT	6 h	0.09	174.9

Table S2. Comparison of the photocatalytic performances of other materials.

Materials	Light source	Time (min)	Photodegradation Amount (%)	Ref
Rutile TiO ₂	Artificial solar light	300 min	70% RhB	[1]
α -Fe ₂ O ₃ /Anatase TiO ₂	Visible light	100 min	86% RhB	[2]
Fe ₂ O ₃ /Anatase TiO ₂	Visible light	100 min	85% RhB	[3]
1D Fe ₂ O ₃ @Anatase TiO ₂	Visible light	360 min	86.53% RhB	[4]
α -Fe ₂ O ₃ (Al)@Anatase TiO ₂	UV-light	100 min	90% RhB	[5]
α -Fe ₂ O ₃ /Rutile TiO ₂	Visible light	100 min	93% RhB	This work

Reference

- [1] Wang, Y. W.; Zhang, L. Z.; Deng, K. J.; Chen, X.-Y.; Zou, Z. G. Low Temperature Synthesis and Photocatalytic Activity of Rutile TiO₂ Nanorod Superstructures. *J. Phys. Chem. C* **2007**, 111, 2709-2714.
- [2] Tang, H.; Zhang, D.; Tang, G. G.; Yang, X. F. Hydrothermal synthesis and visible-light photocatalytic activity of α -Fe₂O₃/TiO₂ composite hollow microspheres. *Ceramics International* **2013**, 39, 8633-8640.
- [3] Wang, T.; Yang, G. D.; Liu, J.; Yang, B. L. Orthogonal synthesis, structural characteristics, and enhanced visible-light photocatalysis of mesoporous Fe₂O₃/TiO₂ heterostructured microspheres. *APPL SURF SCI* **2014**, 311, 314-323.
- [4] Tao, Q. Q.; Huang, X.; Bi, J. T.; Yu, L.; Hao, H. X. Aerobic Oil-Phase Cyclic Magnetic Adsorption to Synthesize 1D Fe₂O₃@TiO₂ Nanotube Composites for Enhanced Visible-Light Photocatalytic Degradation. *Nanomaterials* **2020**, 10, 1345.
- [5] Liu, J.; Yang, S. L.; Wu, W.; Tian, Q. Y.; Cui, S. Y.; Dai, Z. G.; Ren, F. 3D Flowerlike α -Fe₂O₃ @TiO₂ Core-Shell Nanostructures: General Synthesis and Enhanced Photocatalytic Performance. *ACS Sustain. Chem. Eng* **2015**, 3, 2975-2984.