

Supporting Information

Island-type hybrid catalysts applied for anion exchange membrane water electrolysis

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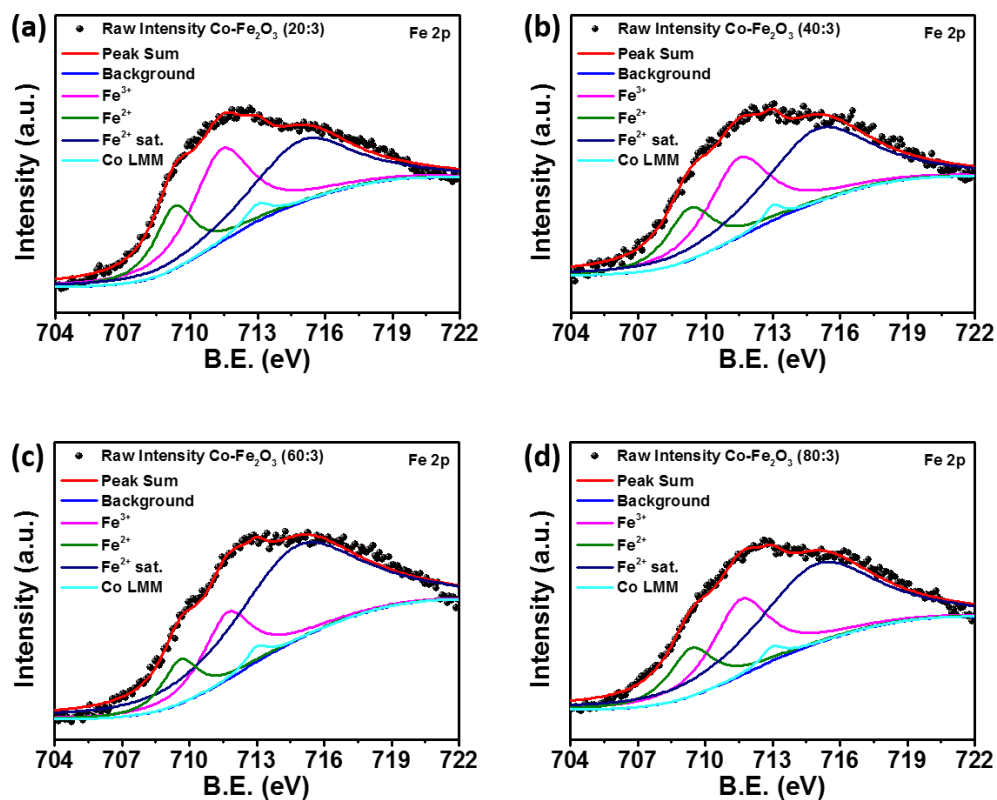


Figure S1. The XPS of Fe 2p spectra for (a) Co-Fe₂O₃ (20:3), (b) Co-Fe₂O₃ (40:3), (c) Co-Fe₂O₃ (60:3), (d) Co-Fe₂O₃ (80:3).

Table S1. The XPS of integrated value for Fe³⁺, Fe²⁺, and Co LMM.

	Fe ³⁺	Fe ²⁺	Co LMM
Co-Fe ₂ O ₃ (20:3)	64.6	30.09	5.31
Co-Fe ₂ O ₃ (40:3)	64.05	32.17	3.78
Co-Fe ₂ O ₃ (60:3)	62.94	30.8	6.26
Co-Fe ₂ O ₃ (80:3)	63.32	31.86	4.82

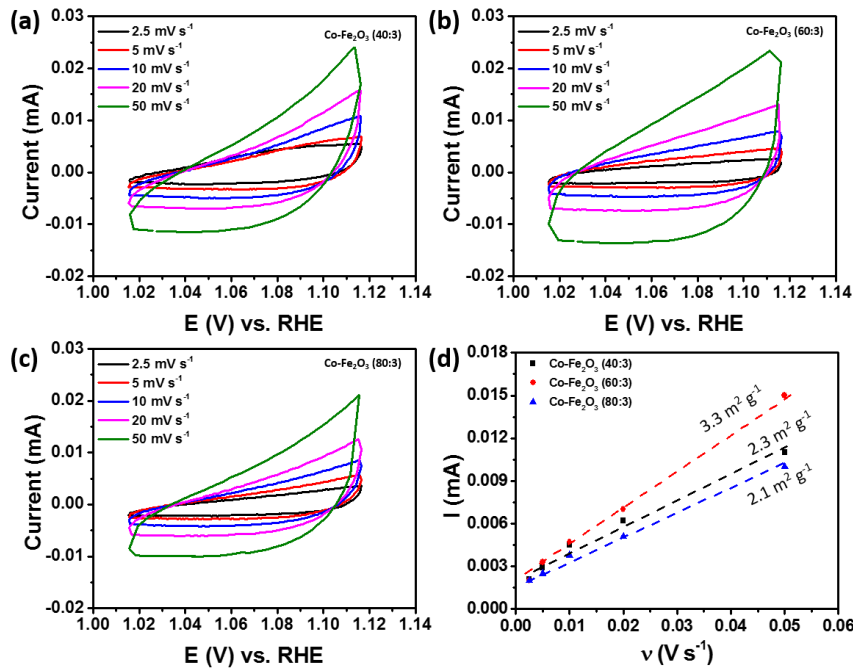


Figure S2. The CV method near the non-faradic potential window: (a) Co-Fe₂O₃ (40:3), (b) Co-Fe₂O₃ (60:3), (c) Co-Fe₂O₃ (80:3), and (d) the scan rates vs. the nonfaradic currents for all samples.

Cycle voltammetry (CV) was performed to calculate the electrochemical surface area (ECSA) with various scan rates from 2.5 to 50 mV s⁻¹ using the capacitive method. For calculation of double-layer charge capacitance (C_{DL}), CV with various scan rates is conducted, and the difference between anodic (j_a) and cathodic (j_c) currents was determined to obtain C_{DL} , which is the slope of cathodic and anodic current. The value of ECSA is calculated by:

$$ECSA = \frac{C_{DL}}{mC_s} \quad (S1)$$

where m means the loading of the active catalysts and C_s means the specific capacitance, which is 4×10^{-5} F cm⁻² [1].

Reference

1. Cao, X.; Johnson, E.; Nath, M. Identifying high-efficiency oxygen evolution electrocatalysts from Co–Ni–Cu based selenides through combinatorial

electrodeposition. *J. Mater. Chem. A* **2019**, 7, 9877–9889,
doi:10.1039/C9TA00863B.