

Supporting Information

Sustainable Electrochemical NO Capture and Storage System Based on the Reversible Fe²⁺/Fe³⁺-EDTA Redox Reaction

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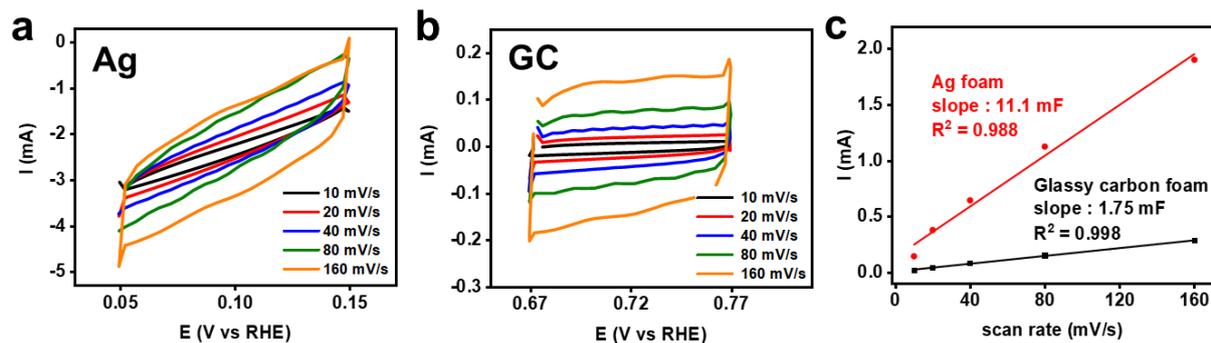


Figure S1. Electrochemical active surface area (ECSA) analysis foam electrodes. Cyclic voltammety curves of (a) Ag and (b) GC foam electrodes at varied scan rate from 10 to 160 mV s^{-1} in a 0.1 M Na_2SO_4 solution (pH 2.7). The geometric surface area of Ag and GC foam was $18 \text{ cm}^2_{\text{geo}}$. The ECSA was determined by dividing (c) double layer capacitance by intrinsic areal capacitance of Ag (33 uF/cm^2) and GC (30 uF/cm^2). The ESCA of Ag and GC foam was 370 cm^2 and 53 cm^2 , respectively

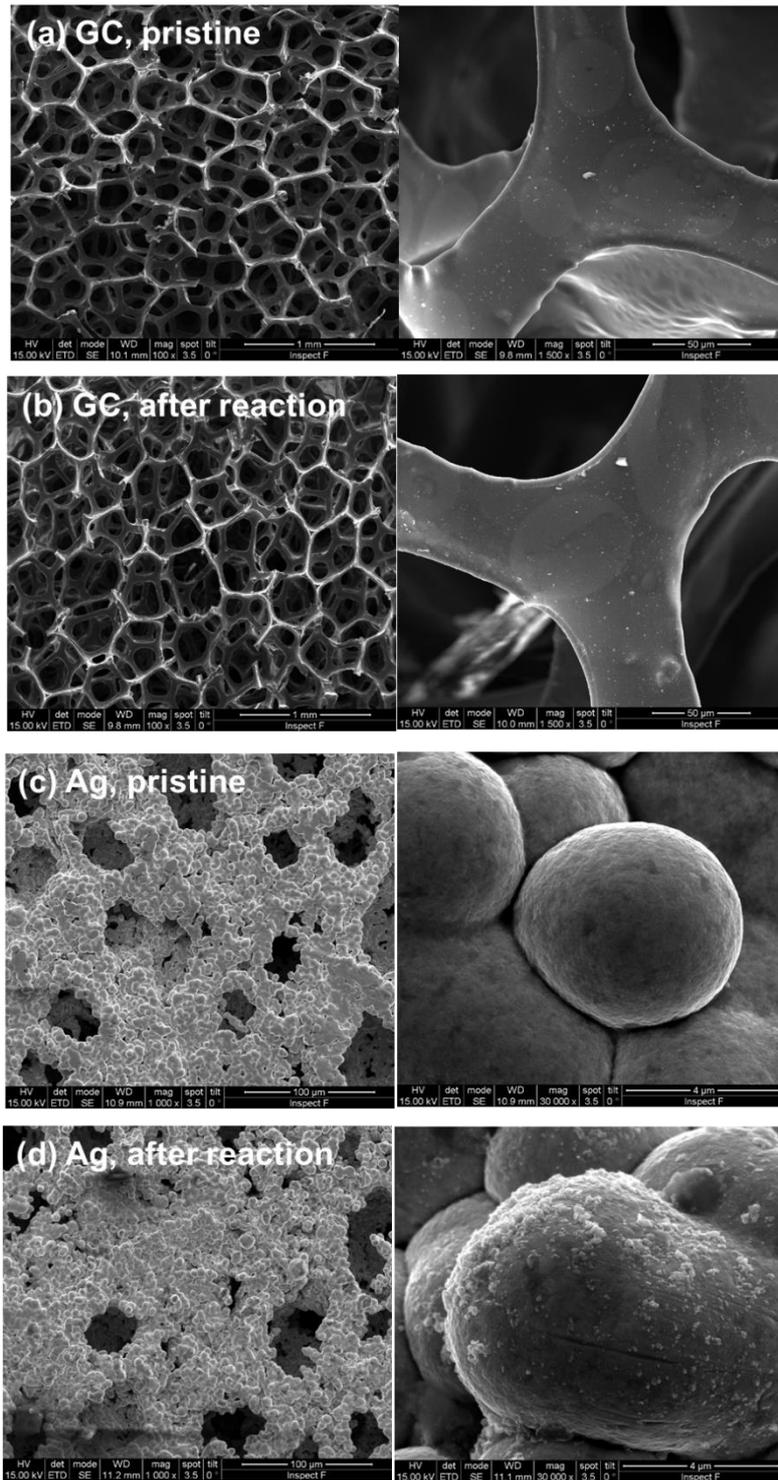


Figure S2. SEM images of (a, b) GC and (c, d) Ag foam electrodes before and after reaction of full-cell system (Figure 4).

Table S1. Elemental analysis of GC and AG foam electrode before and after reaction of full-cell (Figure 4)

	C (at%)	Ag (at%)	Fe (at%)	O (at%)	N (at%)
GC, pristine	89.0	0.0	0.0	4.4	6.7
GC, after reaction	88.8	0.0	0.0	4.6	6.6
Ag, pristine	21.2	52.7	0.0	11.8	14.3
Ag, after reaction	20.1	50.2	0.0	17.1	12.6

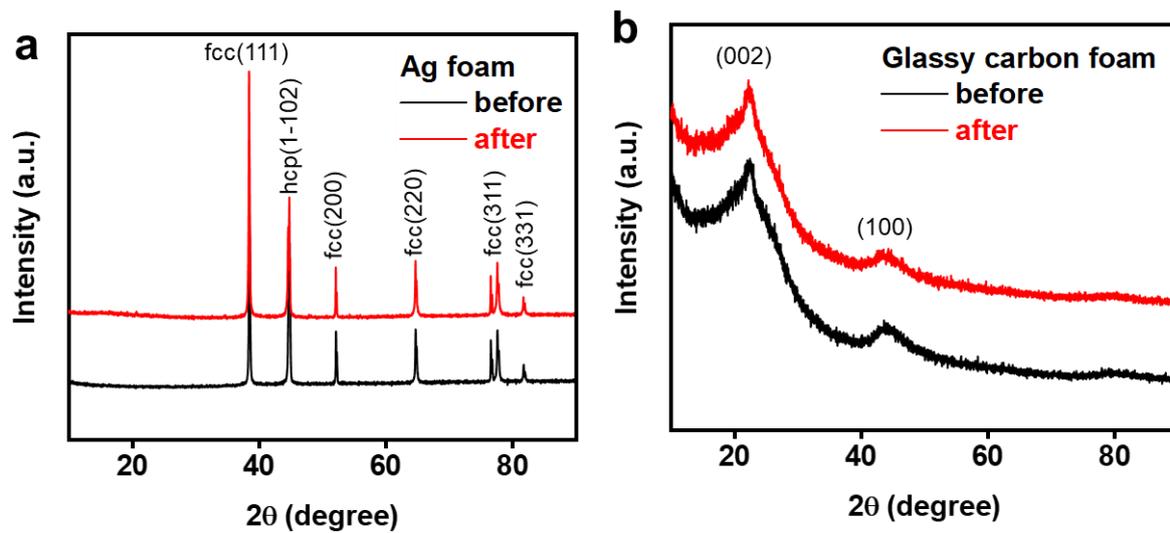


Figure S3. XRD patterns of (a) Ag and (b) GC foam electrodes before and after reaction of full-cell (Figure 4).