

Supporting Information

Table S1. The crystallographic data of $\text{Ce}_{1-x}\text{Fe}_x\text{VO}_4$ samples, some of which were obtained using the Rietveld refinements.

Samples	CeVO_4	$\text{Ce}_{0.75}\text{Fe}_{0.25}\text{VO}_4$	$\text{Ce}_{0.50}\text{Fe}_{0.50}\text{VO}_4$	$\text{Ce}_{0.25}\text{Fe}_{0.75}\text{VO}_4$	FeVO_4
crystal system	tetragonal	tetragonal	tetragonal	tetragonal	triclinic
space group	I41/amd	I41/amd	I41/amd	I41/amd	P-1
cell formula units Z	4	4	4	4	6
a/[Å]	7.39921	7.378008	7.354506	7.349089	6.71924
a/[Å]	7.39921	7.378008	7.354506	7.349089	8.06432
c/[Å]	6.496023	6.481832	6.478912	6.469042	9.25428
R_p	19.9	27.5	20.3	31.0	30.6
R_{wp}	19.8	22.7	17.4	26.7	27.8
R_{exp}	23.79	33.30	30.82	31.41	32.83
χ^2	0.693	0.467	0.320	0.722	0.7188

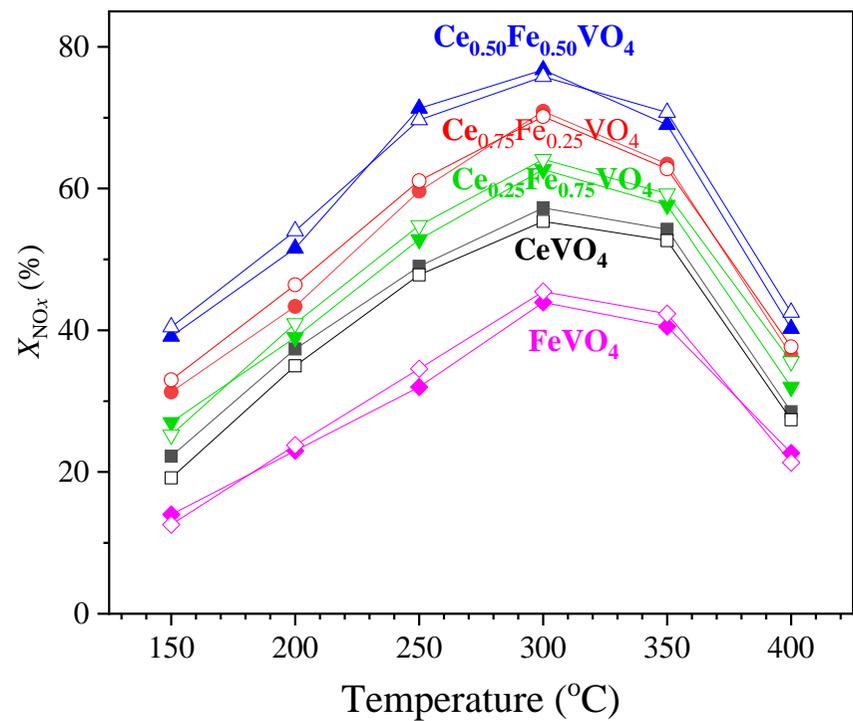


Figure S1. X_{NO_x} over $\text{Ce}_{1-x}\text{Fe}_x\text{VO}_4$ catalysts prepared by a hydrothermal method. Reaction conditions: $[\text{NO}] = 500 \text{ ppm}$, $[\text{NH}_3] = 500 \text{ ppm}$, $[\text{O}_2] = 3 \text{ vol. \%}$, balance N_2 ; catalyst weight: 200 mg; total flow rate: $1000 \text{ mL}\cdot\text{min}^{-1}$.

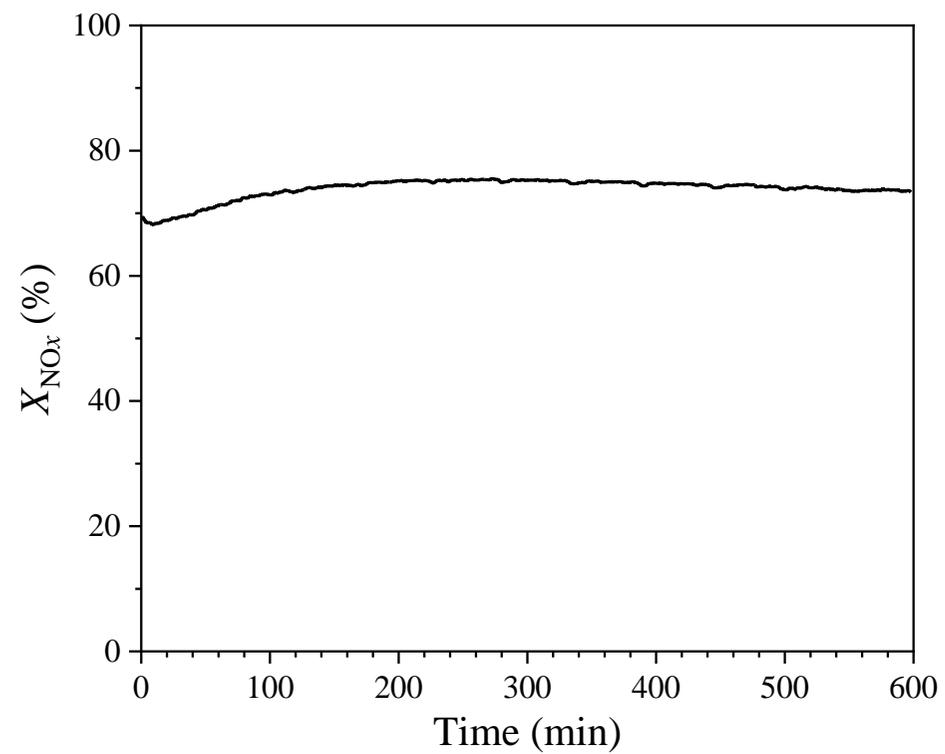


Figure S2. X_{NOx} over $Ce_{0.5}Fe_{0.5}VO_4$ at 300 °C as a function of time on stream. Reaction conditions: $[NO] = 500$ ppm, $[NH_3] = 500$ ppm, $[O_2] = 3$ vol. %, balance N_2 ; catalyst weight: 200 mg; total flow rate: $1000 \text{ mL} \cdot \text{min}^{-1}$.

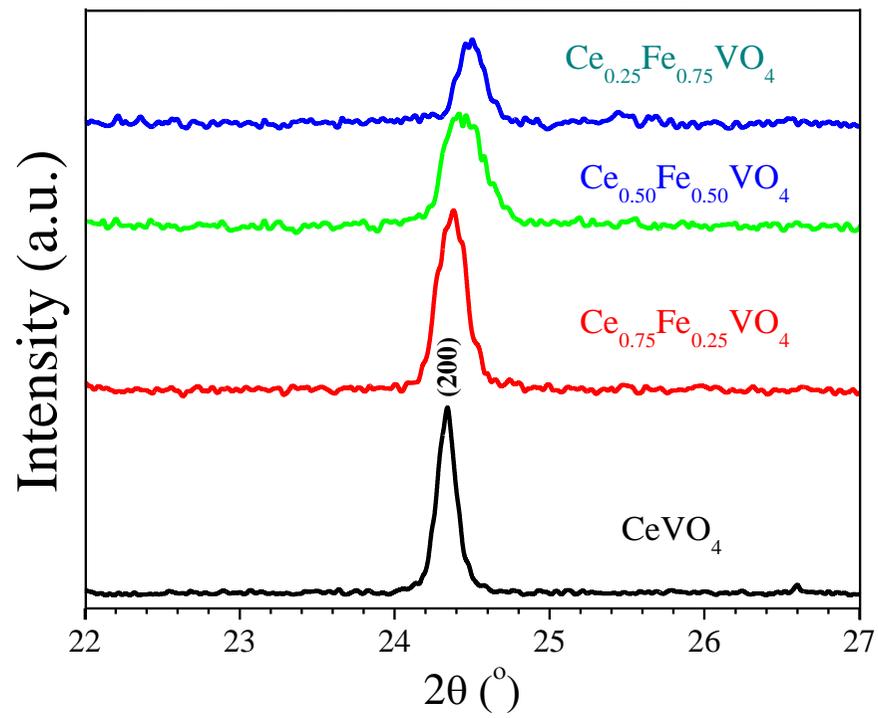


Figure S3. XRD patterns ($2\theta = 22\text{--}27^\circ$) of $\text{Ce}_{1-x}\text{Fe}_x\text{VO}_4$ catalysts.