

Revealing the synergetic effects between reactants in oxidative coupling of methane on stepped MgO(100) catalyst

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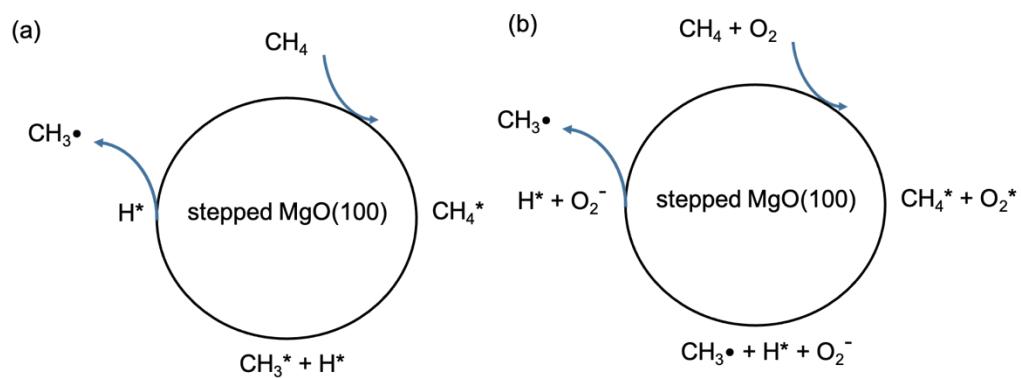
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Table S1 The optimized bond distance and adsorption energies of CH₄ and O₂ molecular adsorbed on the Stepped MgO(100).

	O ₂ *	CH ₄ *	CH ₄ * + O ₂ *
E/eV	-0.25	-0.25	-0.38
R(C-H)/Å		1.10	1.10
R(O _s -H)/Å		2.50	2.30
R(C-Mg)/Å		2.86	
R(O-O)/Å	1.26	-	1.26
R(O-Mg)/Å	2.30	-	2.31
	2.86		2.92
R(O _s -Mg)/Å	2.13	2.13	2.14
	2.14	2.13	2.14

Table S2 The optimized bond distance and bader charge along the reaction pathway of the first C-H bond breaking on the Stepped MgO(100).

	Pathway I			Pathway II		
	CH_4^*	TS	$\text{H}^* + \text{CH}_3^*$	$\text{CH}_4^* + \text{O}_2^*$	TS	$\text{O}_2^- + \text{H}^* + \text{CH}_3\bullet$
$R(\text{C-H})/\text{\AA}$	1.10	1.40		1.10	1.39	
$R(\text{O}_s\text{-H})/\text{\AA}$	2.50	1.28	0.99	2.30	1.21	0.98
$R(\text{C-Mg})/\text{\AA}$	2.86	2.26	2.18			
$R(\text{O-O})/\text{\AA}$	-	-	-	1.26	1.31	1.37
$R(\text{O-Mg})/\text{\AA}$	-	-	-	2.31 2.92	2.13 2.32	2.03 2.23
$R(\text{O}_s\text{-Mg})/\text{\AA}$	2.13 2.13	2.58 1.98	2.78 1.95	2.14 2.14	2.46 2.02	2.79 1.93
$Q(\text{Os})/e$	-1.67	-1.65	-1.65	-1.56	-1.39	-1.63
$Q(\text{Mg})/e$	1.63	1.63	1.64	1.64(-Os) 1.65 1.64	1.64(-Os) 1.66 1.64	1.65(-Os) 1.64 1.64
$Q(\text{O})/e$				-0.16 -0.08	-0.28 -0.31	-0.48 -0.41
$Q(\text{H})/e$	0.06	0.38	0.75	0.03	0.38	0.76
$Q(\text{CH}_3)/e$	-0.10	-0.55	-0.76	-0.05	-0.19	0.00



Scheme S1. The schematic illustration OCM reaction mechanism. (a) conventional MvK (b) proposed L-H.