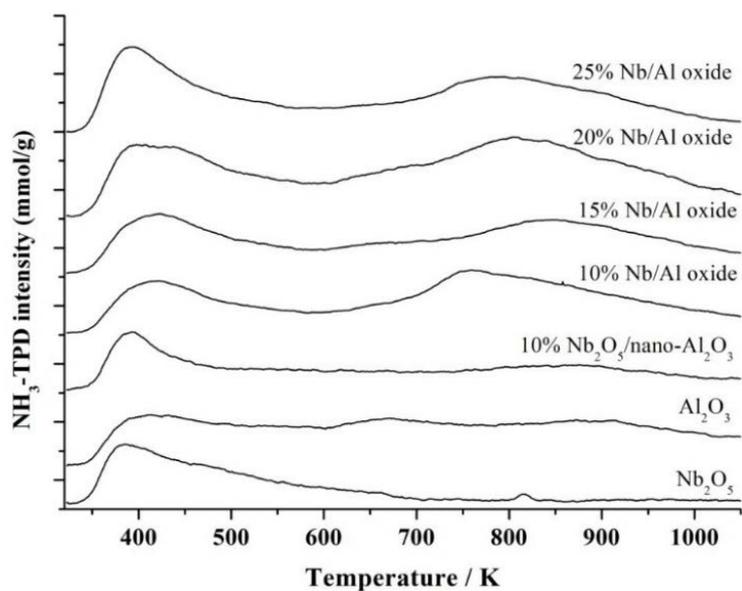
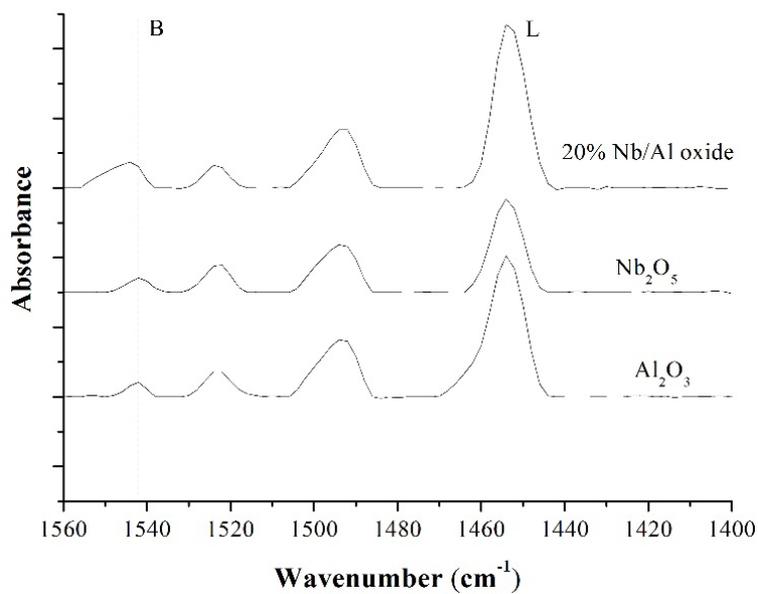


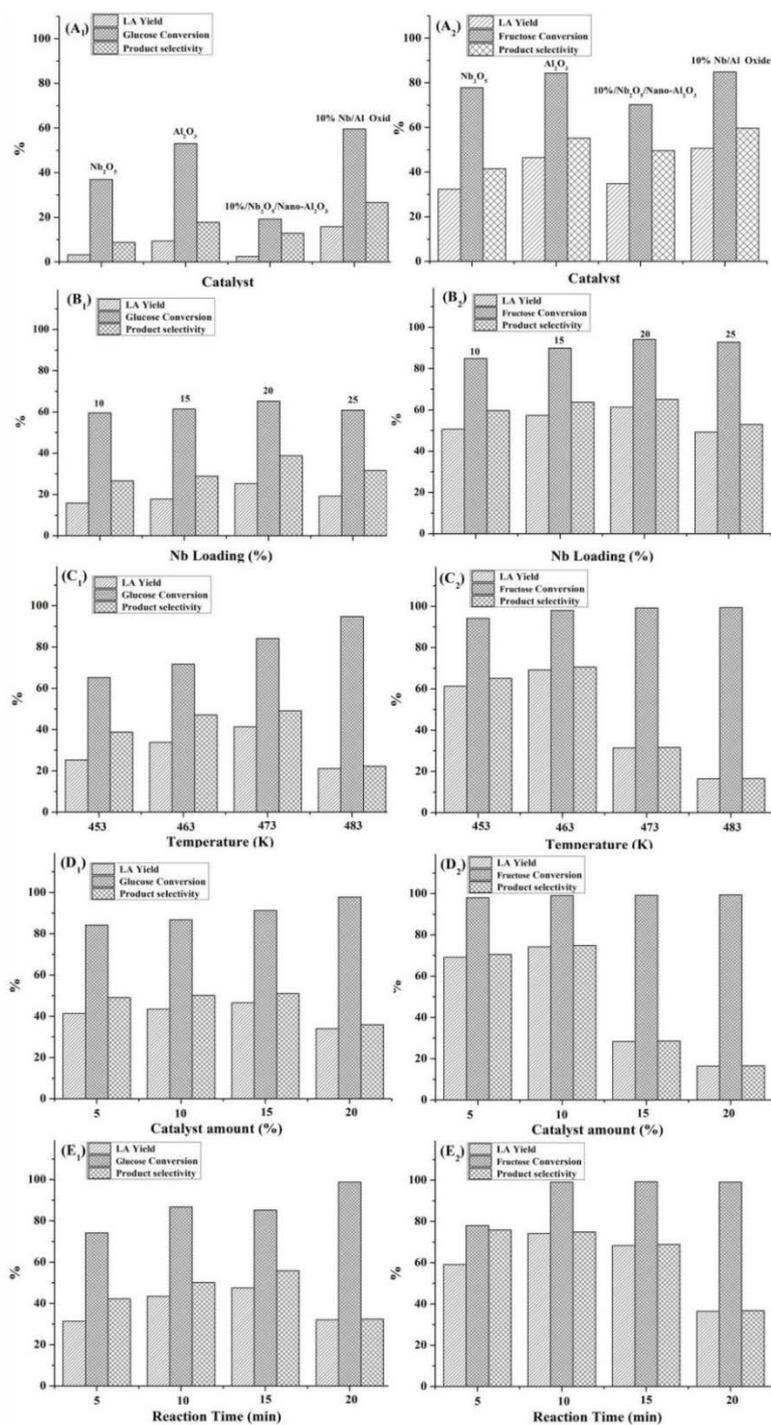
## Supplementary Information



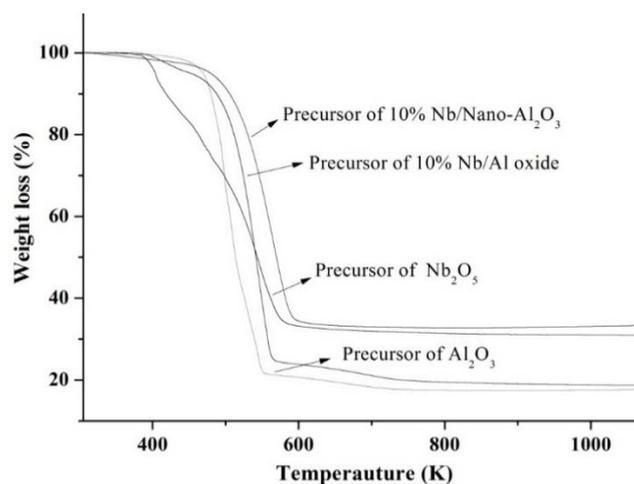
**Figure S1.** NH<sub>3</sub>-TPD curves of Al<sub>2</sub>O<sub>3</sub>, 10% Nb<sub>2</sub>O<sub>5</sub>/nano-Al<sub>2</sub>O<sub>3</sub>, Nb<sub>2</sub>O<sub>5</sub>, and a series of Nb/Al oxide catalysts.



**Figure S2.** FTIR spectra of pyridine adsorption of Al<sub>2</sub>O<sub>3</sub>, Nb<sub>2</sub>O<sub>5</sub>, and 20% Nb/Al oxide.



**Figure S3.** (A<sub>1</sub> and A<sub>2</sub>) Catalytic conversion of glucose and fructose to LA with different metal oxide catalysts (2.0 g sugar, 30 mL H<sub>2</sub>O, 5 wt% catalyst, 453 K, 10 min); (B<sub>1</sub> and B<sub>2</sub>) Catalytic conversion of glucose and fructose to LA over Nb/Al oxides with different Nb loadings (2.0 g sugar, 30 mL H<sub>2</sub>O, 5 wt% catalyst, 453 K, 10 min); (C<sub>1</sub> and C<sub>2</sub>) Effect of reaction temperature on catalytic conversion of glucose and fructose to LA over 20% Nb/Al oxide (2.0 g sugar, 30 mL H<sub>2</sub>O, 5 wt. % catalyst, 10 min); (D) Effect of 20% Nb/Al oxide dosage on catalytic conversion of glucose and fructose to LA (2.0 g sugar, 30 mL H<sub>2</sub>O, 473 K (D<sub>1</sub>) and 463K (D<sub>2</sub>), 10 min); (E) Effect of reaction time on catalytic conversion of glucose and fructose to LA over 20%Nb/Al oxide (2.0 g sugar, 30 mL H<sub>2</sub>O, 473K (E<sub>1</sub>) and 463 K (E<sub>2</sub>)).



**Figure S4.** Thermogravimetric analysis of 10%Nb/Nano- $\text{Al}_2\text{O}_3$ , 10%Nb/Al,  $\text{Al}_2\text{O}_3$ , and  $\text{Nb}_2\text{O}_5$  precursors.

**Table S1.** Distributing of Brønsted and Lewis acid concentrations (mmol/g) of catalyst samples.

Samples	Concentration	Brønsted Lewis		Ration of Brønsted to Lewis
		(mol/g)		
20% Nb/Al		0.0071	0.0293	0.24
10% Nb/Al		0.0051	0.0243	0.21
$\text{Nb}_2\text{O}_5$		0.0030	0.0180	0.17
$\text{Al}_2\text{O}_3$		0.0022	0.0234	0.09

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