

Fe-Doped TiO₂ Supported on HY Zeolite for Solar Photocatalytic Treatment of Dye Pollutants

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Table S-1. Comparison of the efficiency in the removal of pollutant dyes from aqueous media catalyzed by different photocatalytic systems.

Catalyst	Dye	Dye removal (%)	Irradiation	Time of degradation (min)	Ref.
Fe-TiO ₂ /HY	Methylene blue	>98	Visible light	60	Present Study
TiO ₂ /Fe-Pc*	Methylene blue	97	Visible light	90	[S1]
Fe-TiO ₂ /(HT-DS)**	Methylene blue	96	Visible light	120	[S2]
Fe-Pc/TiO ₂	Methyl orange	94	Visible light + H ₂ O ₂	180	[S3]
Fe-TiO ₂ /Pc	Methyl orange	49	Visible light	180	[S4]
Fe-TiO ₂ /ZSM-5	Yellow GX	99	Visible light	20	[S5]
Fe-TiO ₂ /graphene	Crystal violet	74	Visible light	35	[S6]
Ni-TiO ₂ /ZSM-5	Yellow GX	99	Visible light	60	[S5]
Pt-TiO ₂ /natural zeolite	Methyl orange	86	Visible light + H ₂ O ₂	30	[S7]
Pt-TiO ₂ /graphene oxide	Tartrazine	96	Visible light	180	[S8]
Fe-TiO ₂ /natural clay	Malachite green	>98	UV light + H ₂ O ₂ or K ₂ S ₂ O ₈	30	[S9]
CuO/nano-X-zeolite	Methylene blue	55	UV light	120	[S10]

* Pc: Phthalocyanine; **HT: hydrotalcite, DS: dodecylsulfonate.

References to Supplementary Material

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