

Supplementary Material

Step-by-Step Growth of HKUST-1 on Functionalized TiO₂ Surface: An Efficient Material for CO₂ Capture and Solar Photoreduction

Barbara Di Credico ^{1,*}, Matteo Redaelli ¹, Marianna Bellardita ², Massimo Calamante ³, Cinzia Cepek ⁴, Elkid Cobani ¹, Massimiliano D'Arienzo ¹, Claudio Evangelisti ⁵, Marcello Marelli ⁵, Massimo Moret ¹, Leonardo Palmisano ² and Roberto Scotti ¹

¹ Department of Materials Science, INSTM, University of Milano-Bicocca, Via R. Cozzi, 55, 20125 Milano, Italy; mat.redaelli@gmail.com (M.R.); e.cobani@campus.unimib.it (E.C.); massimiliano.darienzo@unimib.it (M.D.A.); massimo.moret@unimib.it (M.M.); roberto.scotti@unimib.it (R.S.)

² Dipartimento di Energia, Ingegneria dell'Informazione e Modelli Matematici DEIM - Università degli Studi di Palermo, Viale delle Scienze (Ed. 6), 90128 Palermo, Italy; marianna.bellardita@unipa.it (M.B.); leonardo.palmisano@unipa.it (L.P.)

³ Istituto di Chimica dei Composti Organometallici ICCOM-CNR, Via Madonna del Piano, 10, 50019 Sesto Fiorentino (Firenze), Italy; mcalamante@iccom.cnr.it

⁴ Istituto Officina dei Materiali IOM-CNR, Laboratorio TASC, Area Science Park–Basovizza, Edificio MM, Strada Statale 14, km 163,5 I-34149 Trieste, Italy; cepek@iom.cnr.it

⁵ Istituto di Scienze e Tecnologie Molecolari ISTM-CNR, via G. Fantoli, 16-15, 20138 Milano, Italy; claudio.evangelisti@istm.cnr.it (C.E.); m.marelli@istm.cnr.it (M.M.)

* Correspondence: barbara.dicredico@unimib.it; Tel.: +39-02-64485023

1.1 Structural and Adsorption Analysis

Table S1. Structural parameters and porosity of TiO₂ RE NPs.

Sample	L ₍₁₀₁₎ (nm)	L ₍₁₀₁₎ TEM (nm)	Pore volume (DCPV, cm ³ g ⁻¹)	SSA _{BET} (m ² g ⁻¹)
RE TiO ₂	16.5	45-60	0.21	170.5
HKUST-1*	/	/	0.33	692.2

Average crystallite sizes (L_{XRD}), calculated by the Scherrer equation, the crystal width estimated from the TEM images (L_{(101) TEM}), the specific surface areas (SSA_{BET}) and BJH pore volumes (DCPV).

* In the case of HKUST-1 we have reported the data from the literature (Ref 31) since the production of HKUST-1 alone, under the synthesis conditions, does not give a fair comparison.

1.2 TEM Analysis

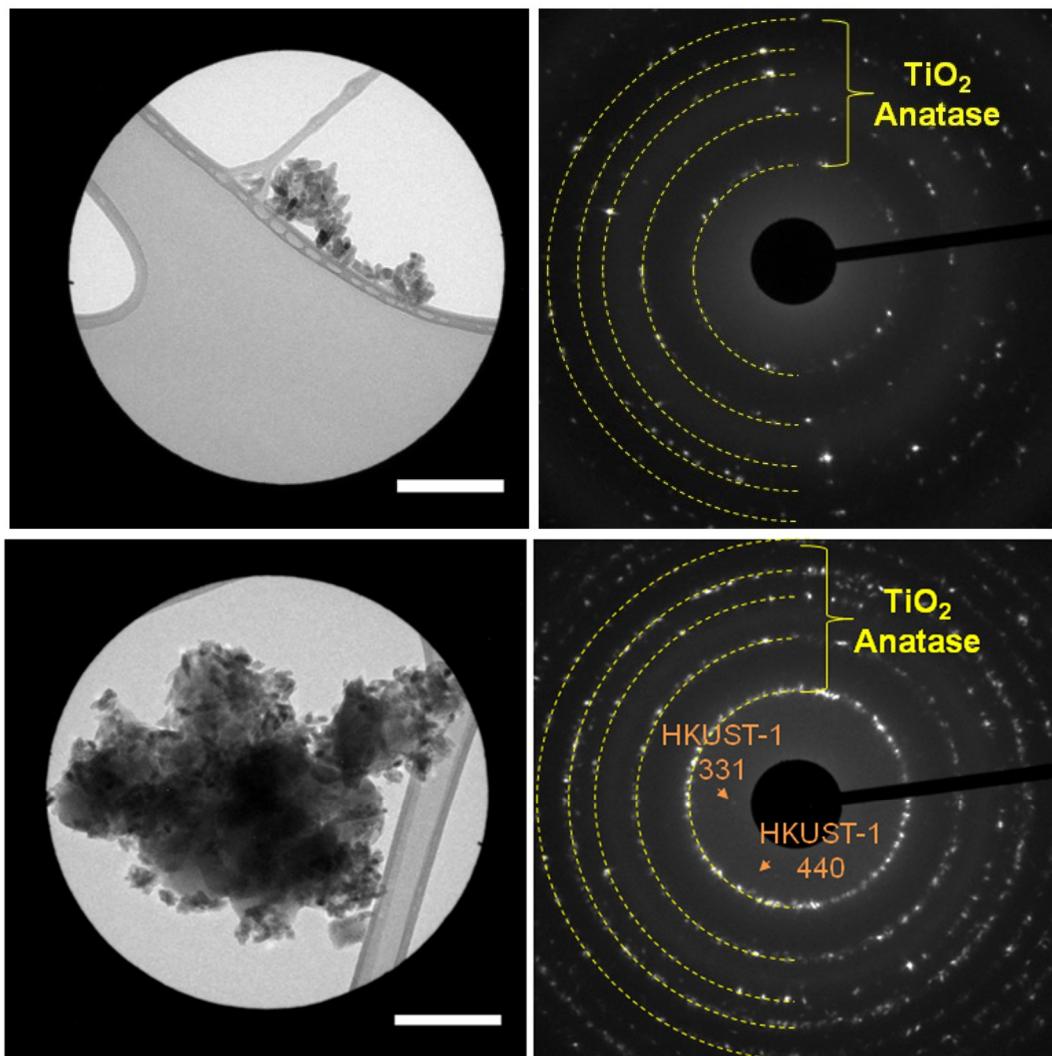


Figure S1. TEM micrographs and related SAED analysis of: above TiO₂ anatase crystals and below the TiO₂/HKUST-1 hybrid system. the latter shows some faint reflection at d-spacing > 3.5 Å indexed as HKUST-1. All scale bars are 200 nm, TiO₂ anatase reflexes are indexed as 101, 103, 200, 105 and 213 planes (from the centre to outside).

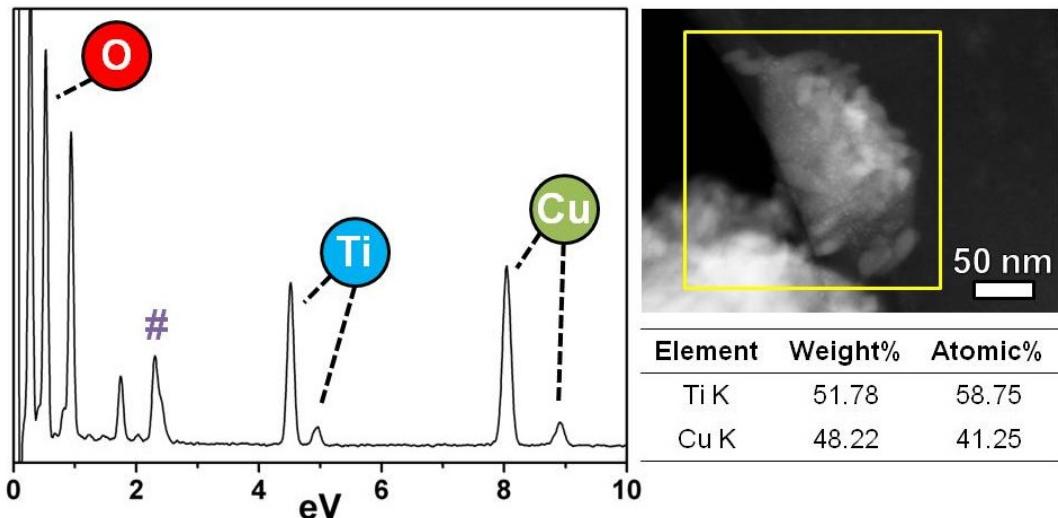


Figure S2. STEM-EDS spectra (on the left side) of selected and representative grain (on the top-right side) and related quantitative analysis (on the bottom-right side) of $\text{TiO}_2/\text{HKUST}-1$. On the spectra, the peak # is related to the Mo signal, from supported holey carbon Molybdenum TEM grid.

1.3 DR-UV-Vis Spectra

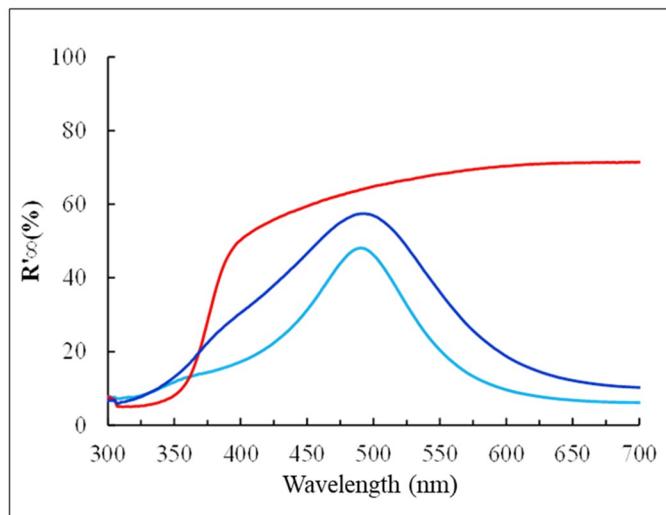


Figure S3. Kubelka-Munk function of TiO_2 -PHA (red line), pure HKUST-1 (sky blue line) and $\text{TiO}_2/\text{HKUST}-1$ (blue line).

1.4 XPS Analysis

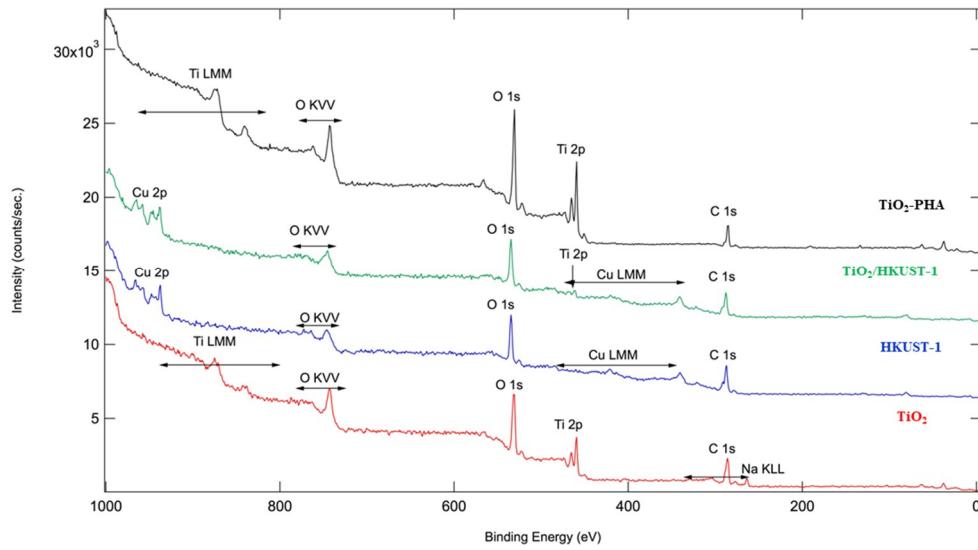


Figure S4. XPs survey spectra of TiO_2 , $\text{TiO}_2\text{-PHA}$, HKUST-1 and $\text{TiO}_2/\text{HKUST-1}$.

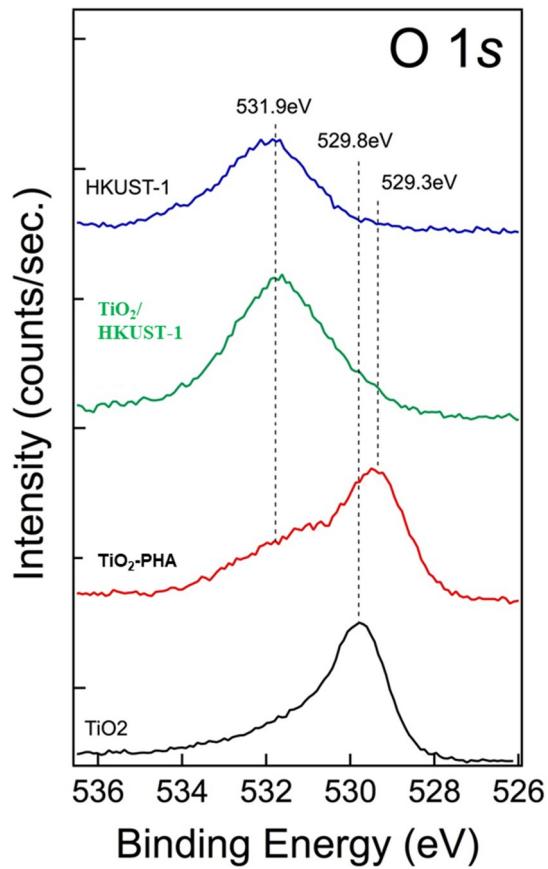


Figure S5. High resolution XPS spectra of O1s in HKUST-1, TiO_2 , $\text{TiO}_2\text{-PHA}$ and $\text{TiO}_2/\text{HKUST-1}$.

1.5 Photocatalytic Activity

Table S2. Results of the photocatalytic runs in the presence of the bare TiO₂-PHA sample.

TiO ₂ -PHA			
	1 st run		
Time (h)	CH ₄ (μM)	CH ₃ CHO (μM)	CO (μM)
1	1.65	0.85	0.00
2	2.45	0.74	0.01
3	3.50	0.89	0.01
4	4.93	1.16	0.01
5	6.20	1.24	0.02
6	8.97	1.48	0.02
2 nd run			
1	0.20	-	-
2	0.36	-	-
3	0.50	-	-
4	0.45	-	-
5	0.66	-	-
6	0.69	-	-

Table S3. Results of the photocatalytic runs in the presence of the bare HKUST-1 sample.

HKUST-1			
	1 st run		
Time (h)	CH ₄ (μM)	CH ₃ CHO (μM)	CH ₃ CH ₂ OH (μM)
1	0.57	16.17	-
2	1.52	70.04	331.33
3	1.93	82.33	407.90
4	2.23	87.56	439.00
5	2.57	115.71	476.45
6	2.80	107.27	536.23
2 nd run			
1	0.05	-	-
2	0.08	-	-
3	0.13	-	-
4	0.23	-	-
5	0.37	-	-
6	0.42	-	-

Table S4. Results of the photocatalytic runs in the presence of the sample TiO₂/HKUST-1.

TiO ₂ /HKUST-1		
Time (h)	1 st run	
	CH ₄ (μM)	CH ₃ CHO (μM)
	1 nd run	
1	0.53	-
2	0.98	-
3	1.14	-
4	1.30	-
5	1.56	-
6	2.19	-
2 nd run		
1	0.24	-
2	0.67	-
3	1.00	-
4	1.32	-
5	2.07	-
6	2.63	-

1.6 Hybrid Catalyst Characterization after Photocatalysis

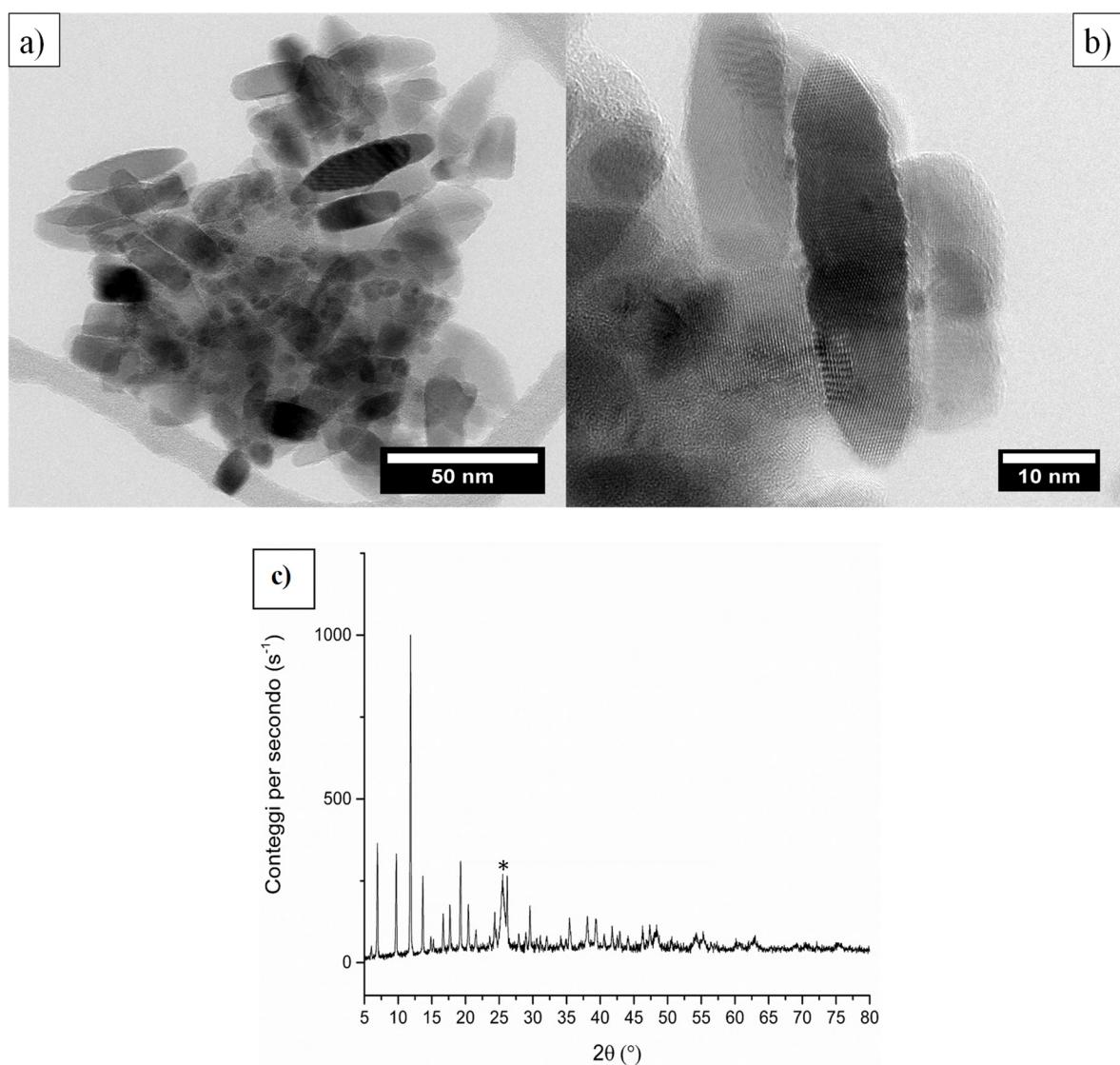


Figure S6. (a,b) TEM image and (c) XRPD pattern of $\text{TiO}_2/\text{HKUST}$ (the asterisks refer to anatase) after the second photocatalytic run. There are no morphological, phase and composition changes after the reactions.