Supplementary Material

Catalytic oxidation of dimethyl disulfide (CH₃SSCH₃, DMDS) over bimetallic Cu-Au and Pt-Au catalysts supported on γ -Al₂O₃, CeO₂, and CeO₂-Al₂O₃

Tuomas K. Nevanperä ¹, Satu Ojala ^{1,*}, Tiina Laitinen ¹, Satu Pitkäaho ¹, Sami Saukko ² and Riitta L. Keiski ¹

- ¹ University of Oulu, Faculty of Technology, Environmental and Chemical Engineering, FI-90014 University of Oulu, P.O.Box 4300, Finland
- ² University of Oulu, Center of Microscopy and Nanotechnology, FI-90014 University of Oulu, P.O.Box 7150, Finland;
- * Correspondence: satu.ojala@oulu.fi; Tel.: +358 50 3506098



Figure S1. N2 adsorption isotherms of the prepared bimetallic catalysts.



Figure S2. Pore size distributions of the prepared bimetallic catalysts.



Figure S3. HR-TEM images of Pt-Au/Al catalyst showing examples of particles with their corresponding EDX data.





Figure S4. HR-TEM images of Cu-Au/Al catalyst showing examples of particles with their corresponding EDX data.



Figure S5. HR-TEM images of Pt-Au/Ce catalyst showing examples of particles.



 $\label{eq:Figure S6.} Figure \ S6. \ \text{HR-TEM} \ \text{image of Cu-Au/Ce catalyst with corresponding EDX \ data.}$



Figure S7. HR-TEM images of Pt-Au/Ce-Al catalyst showing examples of particles.





Figure S8. HR-TEM images of Pt-Au/Ce-Al catalyst showing examples of particles with their corresponding EDX data.



Figure S9. HAADF-STEM images of Pt-Au/Ce catalyst.



Figure S10. DMDS conversion, product yields (SO₂, CO₂, CO, and CH₂O), oxygen exchange rate, H₂ uptake, and NH₃ desorption results for the Pt-Au/Al and Pt-Au/Ce catalysts (Figure 15 shows corresponding graph for the Pt-Au/Ce-Al catalyst).



Figure S11. DMDS conversion, product yields (SO₂, CO₂, CO, and CH₂O), oxygen exchange rate, H₂ uptake, and NH₃ desorption results for the Cu-Au/Al, Cu-Au/Ce, and Cu-Au/Ce-Al catalysts.