

# Comparison of the Effect of the Amino Acids on Spontaneous Formation and Transformation of Calcium Phosphates

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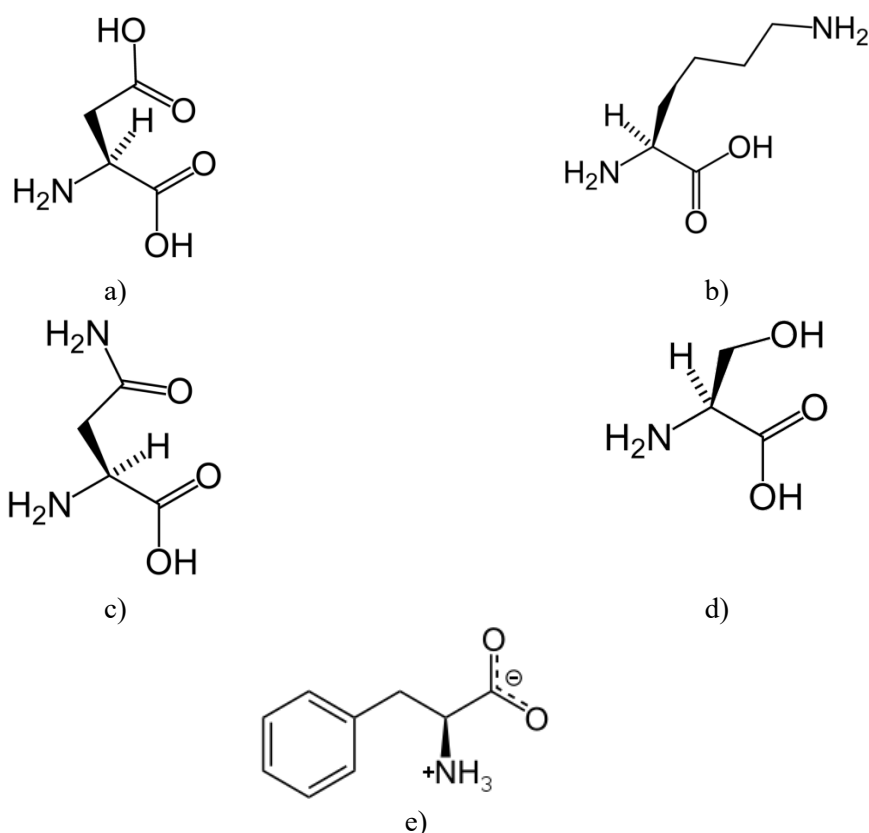
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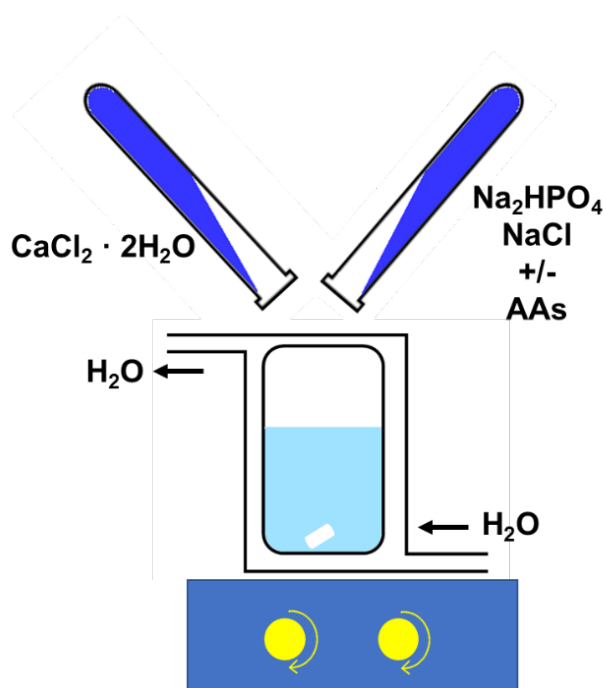
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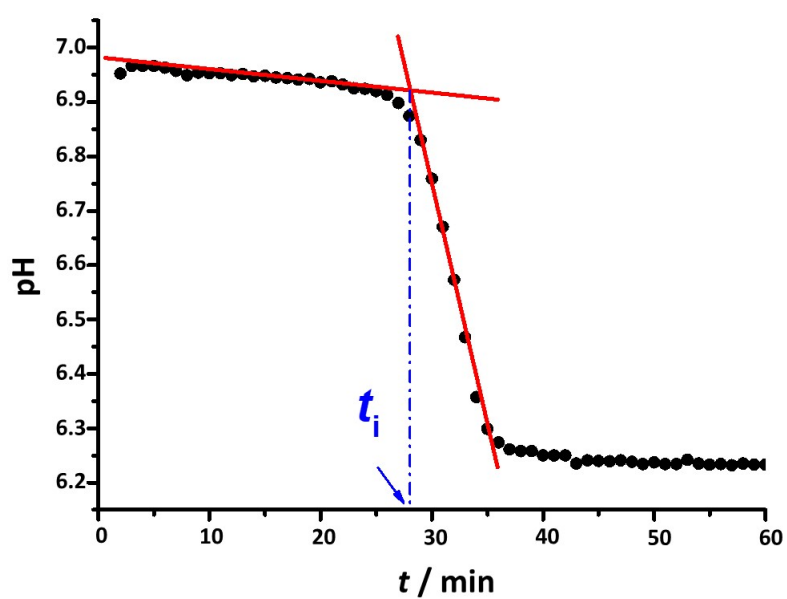
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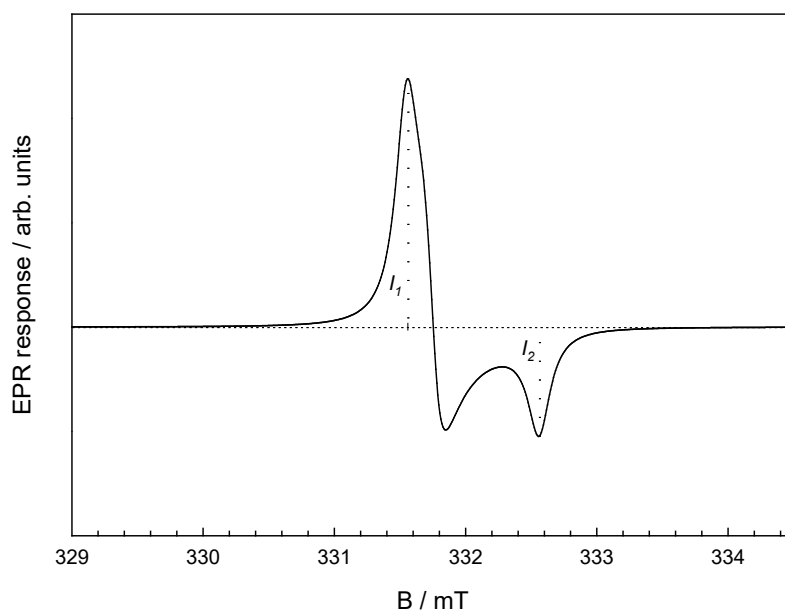
**Figure S11.** Molecular structure of investigated amino acids: a) Asp, b) Lys, c) Asn, d) Ser and e) Phe.



**Figure SI2.** Schematic illustration of precipitation experiments.



**Figure SI3.** Determination of induction time ( $t_i$ ) as the intercept between two tangents drawn on the first two parts of the pH vs time curve.



**Figure SI4.** Definition of  $R$ -value:  $R=I_2/I_1$ .

**Table SI1.** Experimental conditions for precipitation experiments.

$c(\text{CaCl}_2 \cdot 2\text{H}_2\text{O})$	$5 \text{ mmol dm}^{-3}$
$c(\text{Na}_2\text{HPO}_4)$	$5 \text{ mmol dm}^{-3}$
$c(\text{NaCl})$	$0.15 \text{ mol dm}^{-3}$
$c(\text{amino acid})$	$1, 2.5, \text{ and } 5 \text{ mmol dm}^{-3}$
pH	7.4
$V$	40 mL
$\vartheta$	$(25 \pm 0.1) ^\circ\text{C}$
Mixing mode	magnetic stirring