

# *Electronic Supplementary Information*

## *Development of hydrogen-bonded dimer-type photoluminescent liquid crystals of fluorinated tolanecarboxylic acid*

Shigeyuki Yamada <sup>1,\*</sup>, Mitsuki Kataoka <sup>1</sup>, Keigo Yoshida <sup>1</sup>, Masakazu Nagata <sup>2,3</sup>,  
Tomohiro Agou <sup>2</sup>, Hiroki Fukumoto <sup>2</sup>, ant Tsutomu Konno <sup>1</sup>

<sup>1</sup> Faculty of Molecular Chemistry and Engineering, Kyoto Institute of Technology, Matsugasaki, Sakyo-ku, Kyoto 606-8585, Japan

<sup>2</sup> Department of Quantum Beam Science, Graduate School of Science and Engineering, Ibaraki University, 4-12-1 Nakanarusawa, Hitachi, Ibaraki 316-8511, Japan

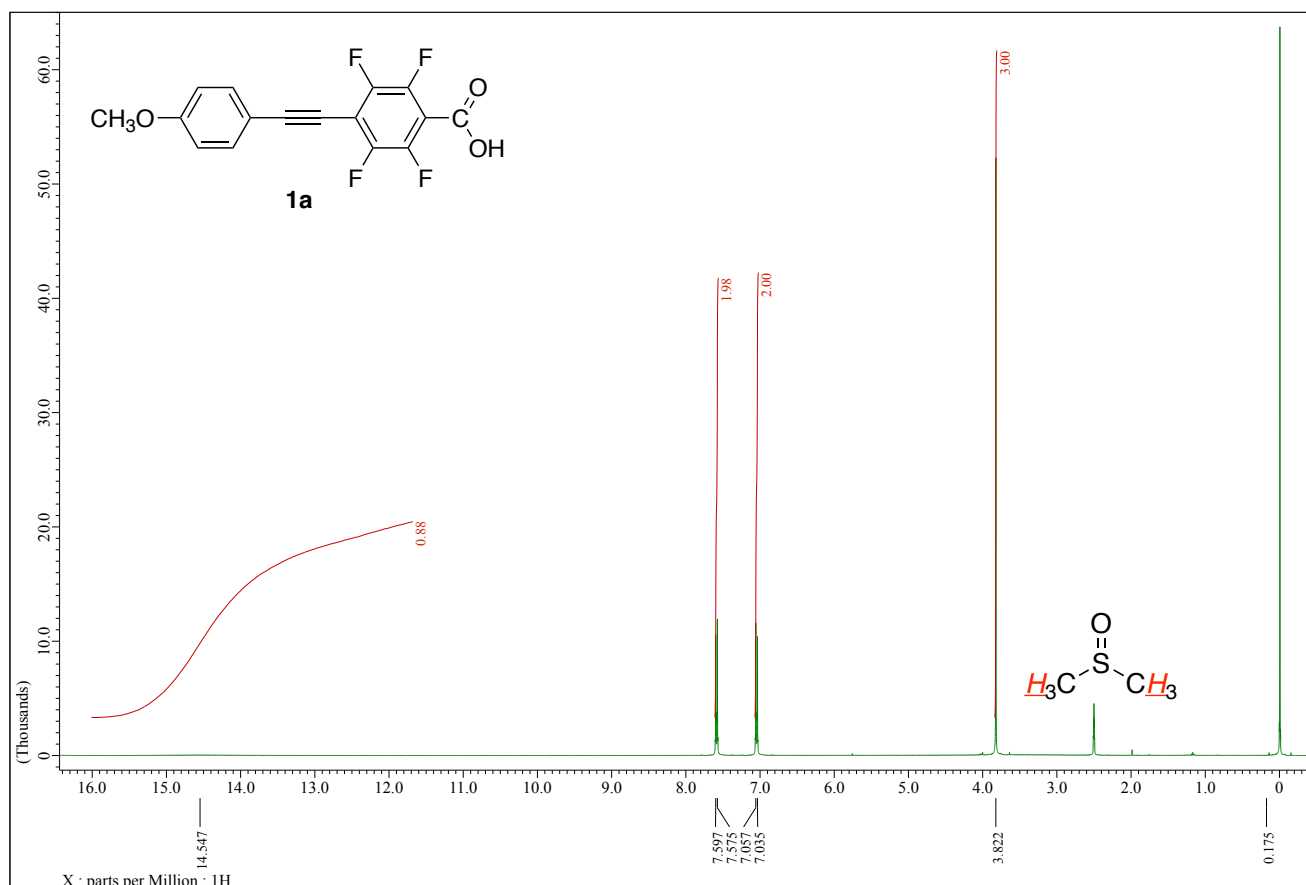
<sup>3</sup> Present address: Organic Materials Chemistry Group, Sagami Chemical Research Institute, Hayakawa 2743-1, Ayase, Kanagawa 252-1193, Japan

\* Correspondence: syamada@kit.ac.jp (S.Y.); Tel: +81-75-724-7517

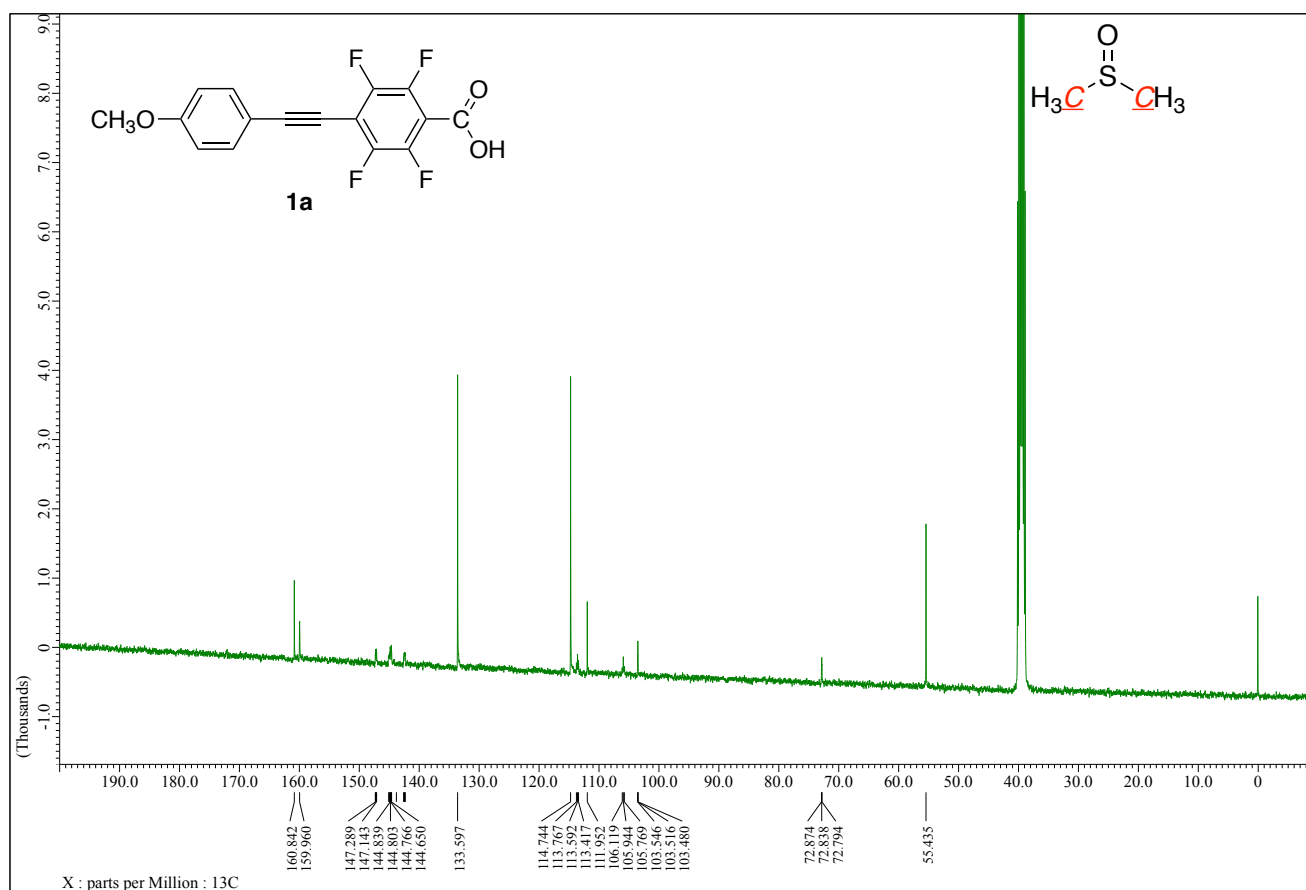
### *Table of Contents*

NMR spectrum	••••••••••	S-2
Crystallographic analysis	••••••••••	S-18
Phase transition behavior		
DSC thermograms	••••••••••	S-19
VT-PXRD patterns	••••••~•••••	S-22
Photophysical behavior		
CH <sub>2</sub> Cl <sub>2</sub> solution phase	••••••~•••••	S-23
Solvatochromic PL behavior	••••••~•••••	S-24
Cry phase	••••••~•••••	S-25
Nematic LC phase	••••••~•••••	S-26
Quantum chemical calculation	••••••~•••••	S-27
Cartesian coordinate	••••••~•••••	S-29

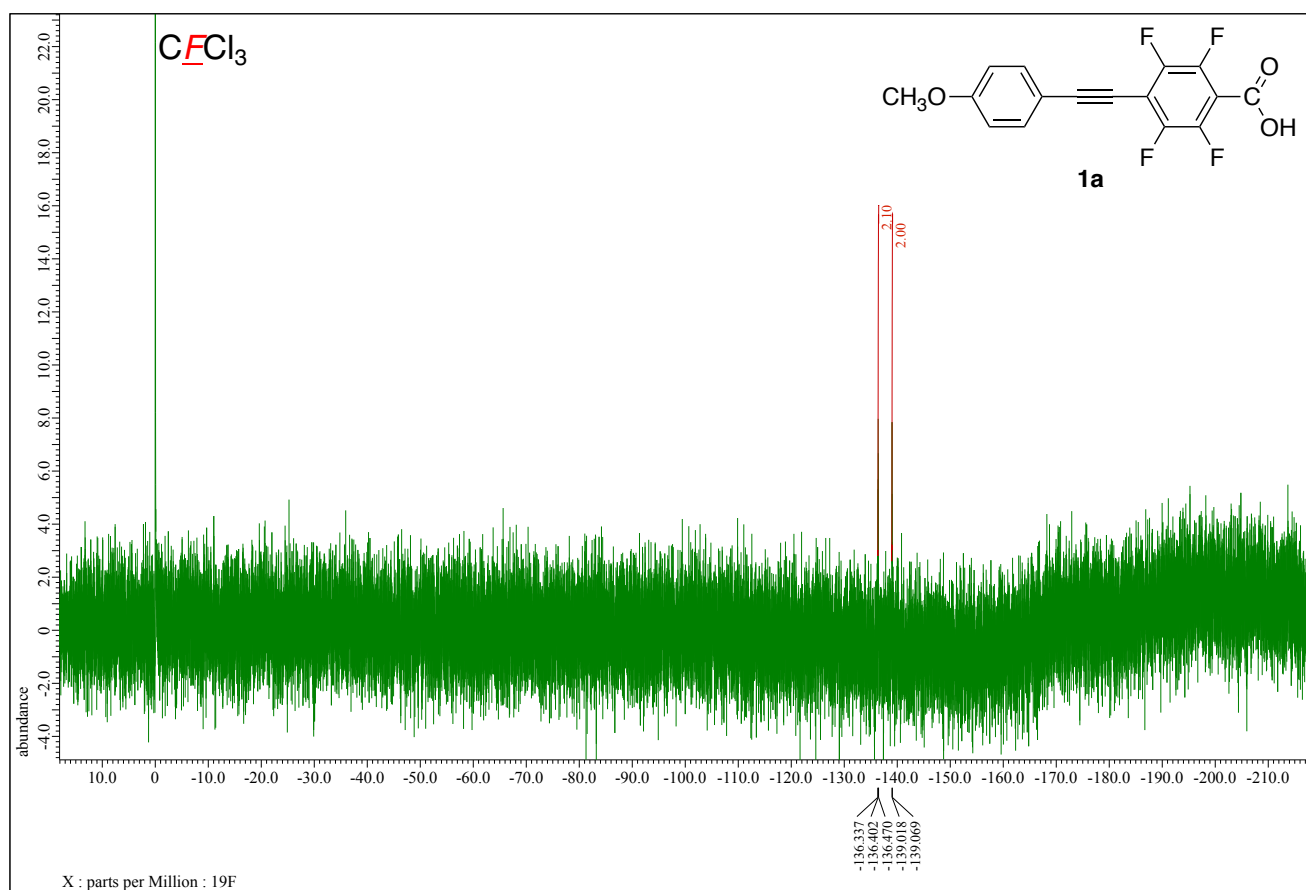
## NMR spectrum



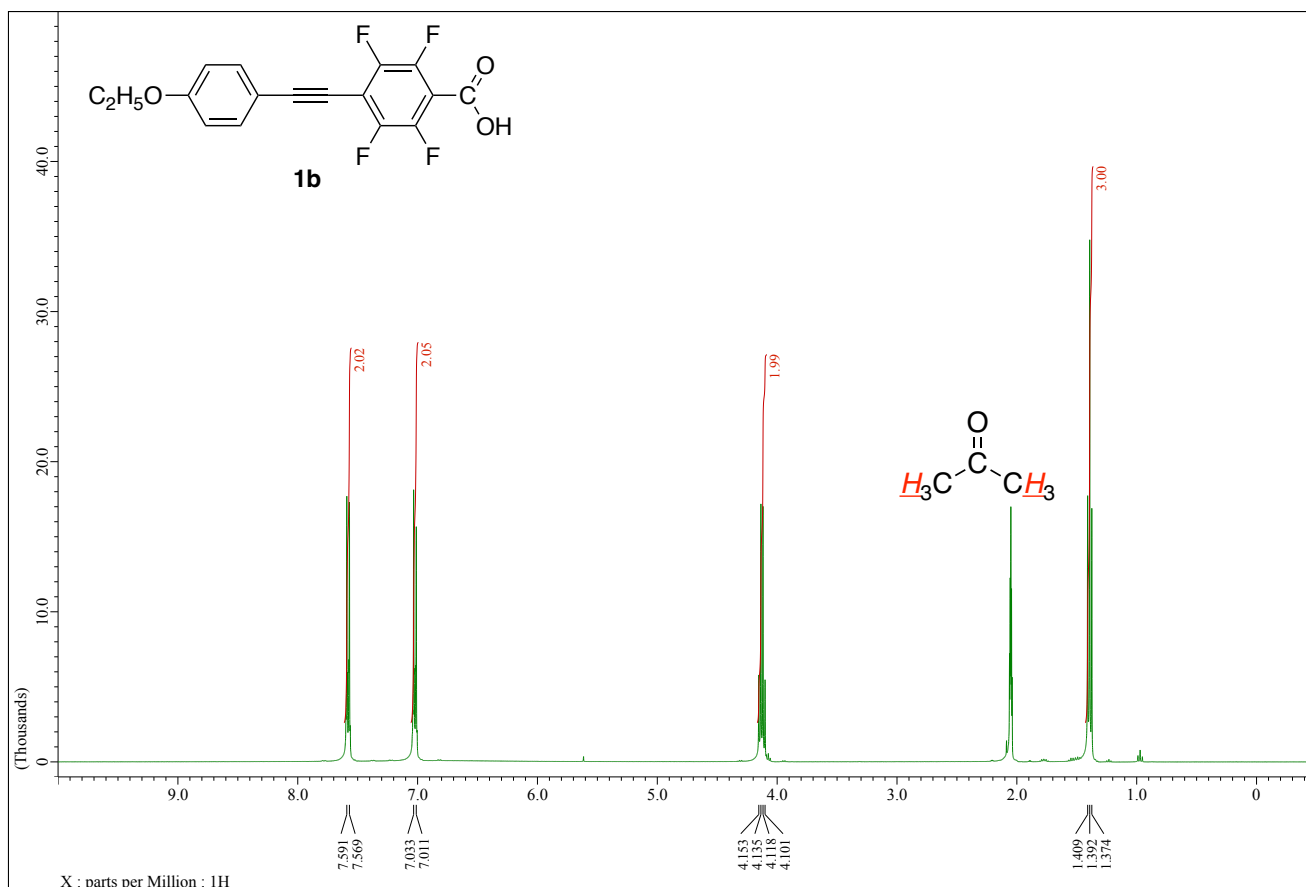
**Figure S1.** <sup>1</sup>H NMR spectrum of **1a** (DMSO-*d*<sub>6</sub>, 400 MHz).



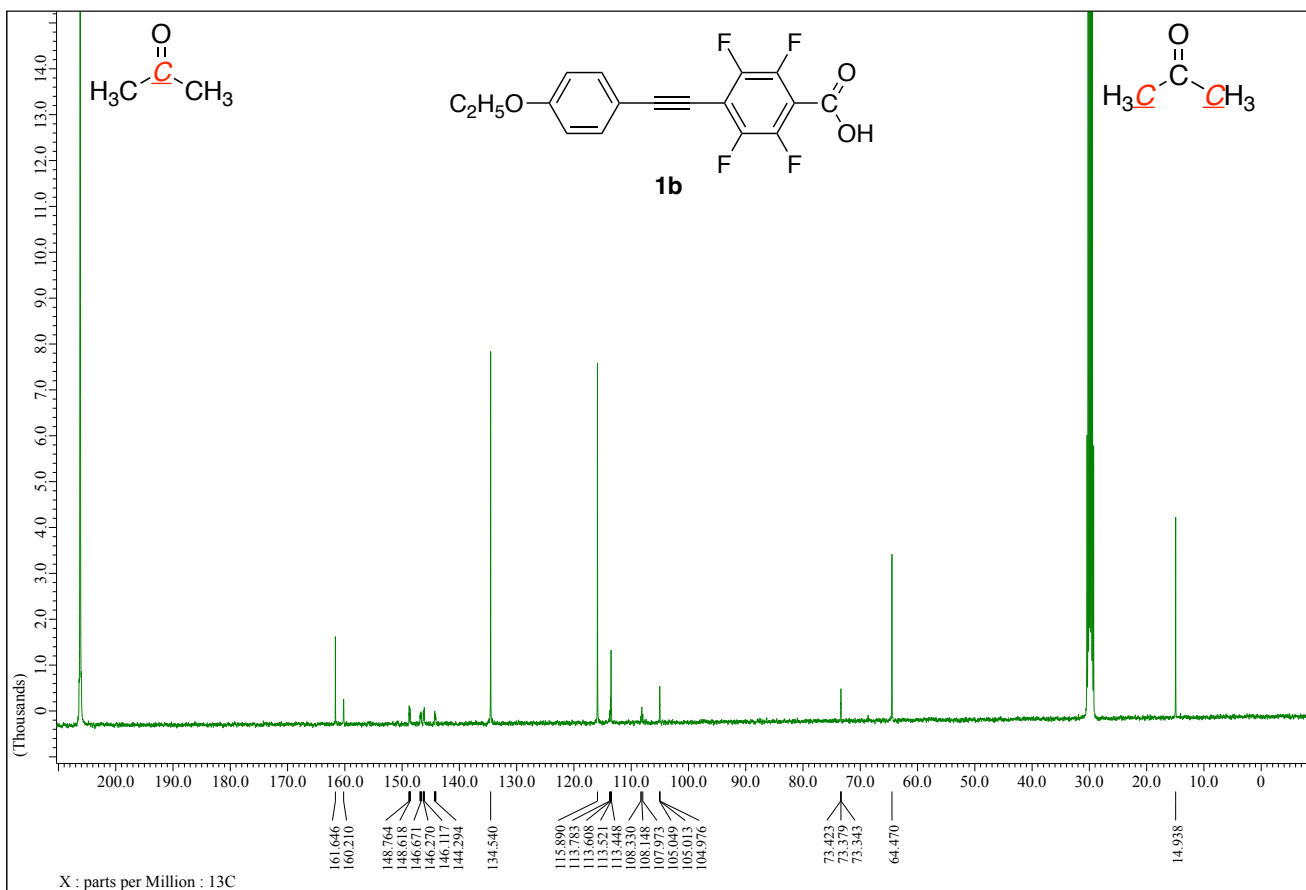
**Figure S2.** <sup>13</sup>C NMR spectrum of **1a** (DMSO-*d*<sub>6</sub>, 100 MHz).



**Figure S3.**  $^{19}\text{F}$  NMR spectrum of **1a** ( $\text{DMSO-}d_6$ ,  $\text{CFCl}_3$ , 376 MHz).

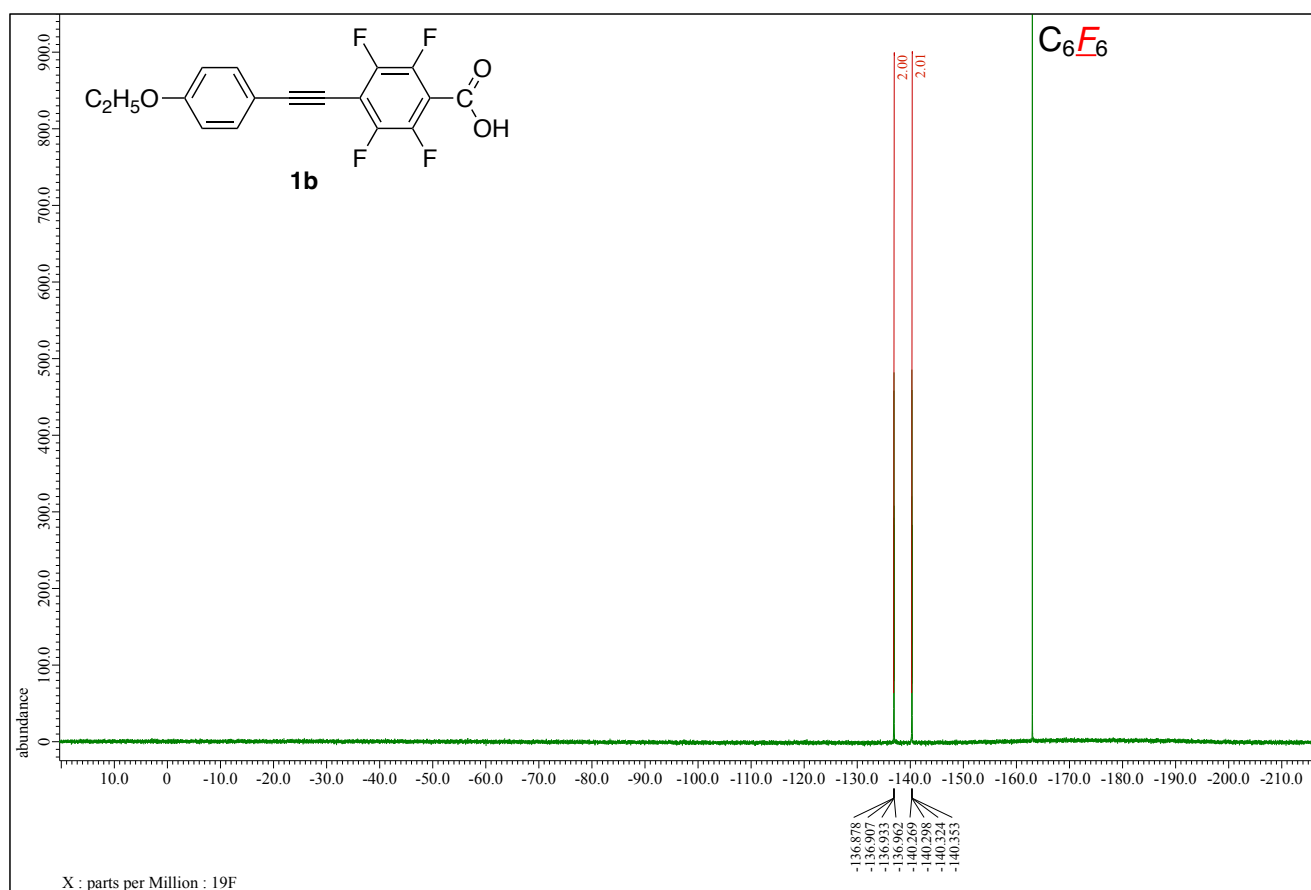


**Figure S4.** <sup>1</sup>H NMR spectrum of **1b** (acetone-*d*<sub>6</sub>, 400 MHz).



**Figure S5.** <sup>13</sup>C NMR spectrum of **1b** (acetone-*d*<sub>6</sub>, 100 MHz).





**Figure S6.**  $^{19}\text{F}$  NMR spectrum of **1b** (acetone- $d_6$ ,  $\text{C}_6\text{F}_6$ , 376 MHz).

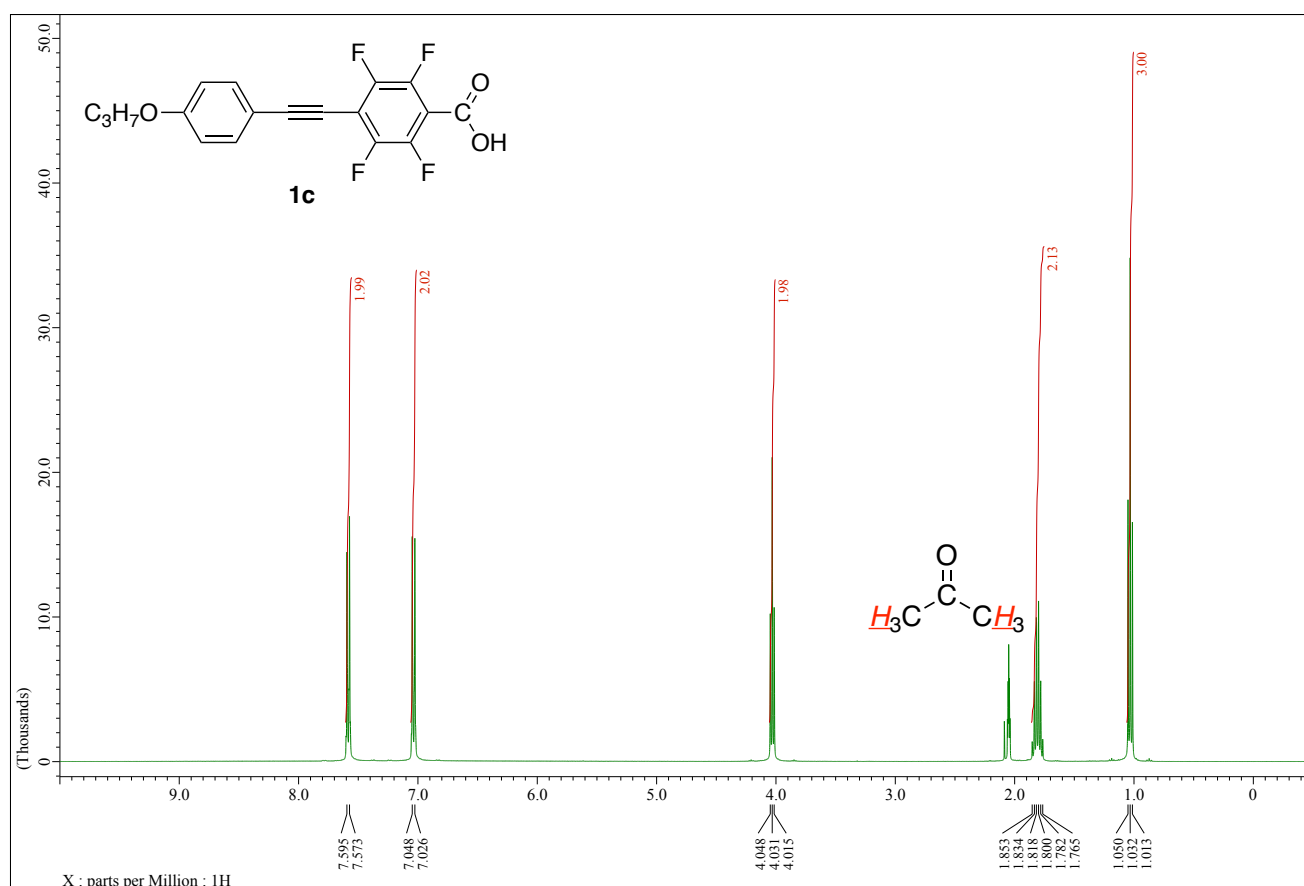


Figure S7. <sup>1</sup>H NMR spectrum of **1c** (acetone-*d*<sub>6</sub>, 400 MHz).

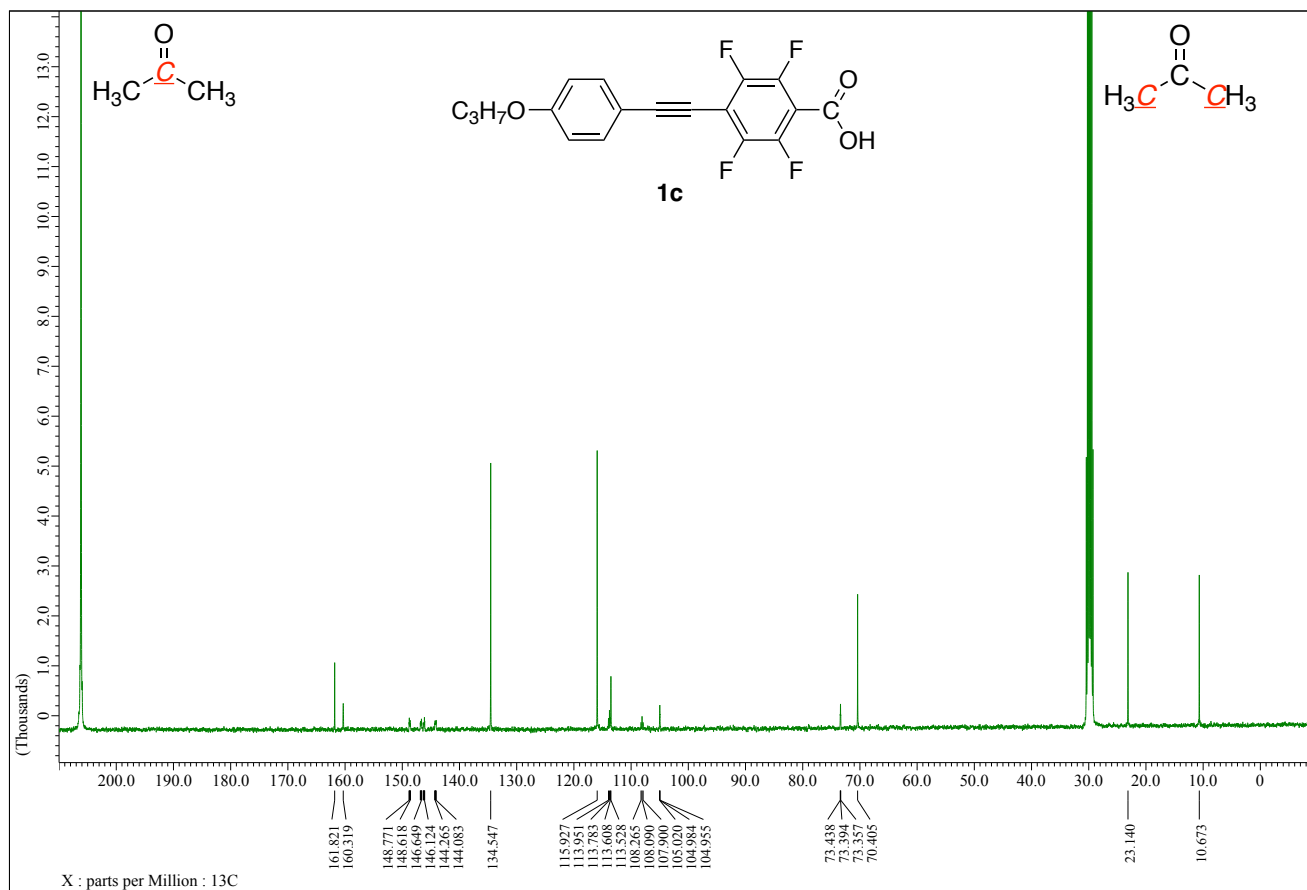
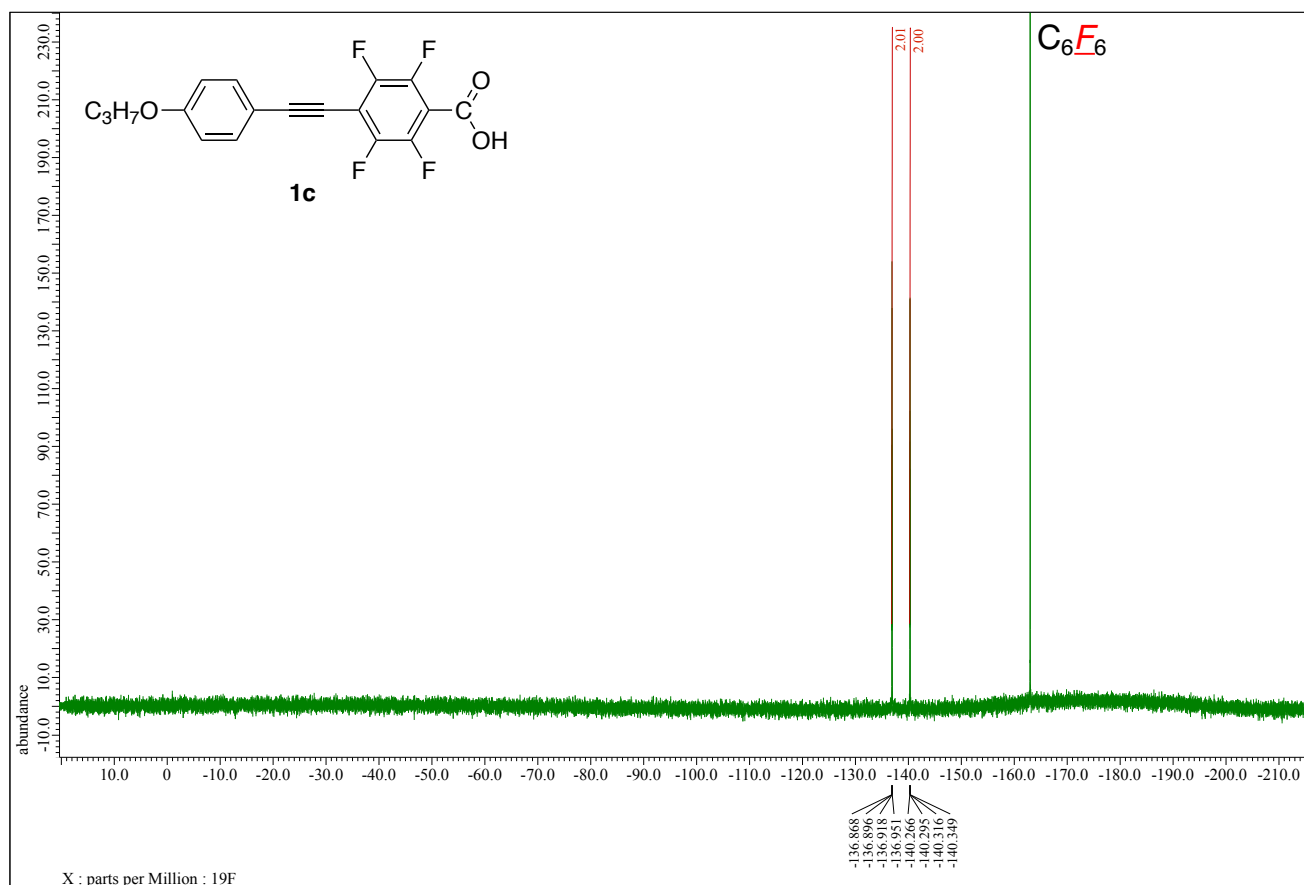


Figure S8. <sup>13</sup>C NMR spectrum of **1c** (acetone-*d*<sub>6</sub>, 100 MHz).



**Figure S9.** <sup>19</sup>F NMR spectrum of **1c** (acetone-*d*<sub>6</sub>, C<sub>6</sub>F<sub>6</sub>, 376 MHz).

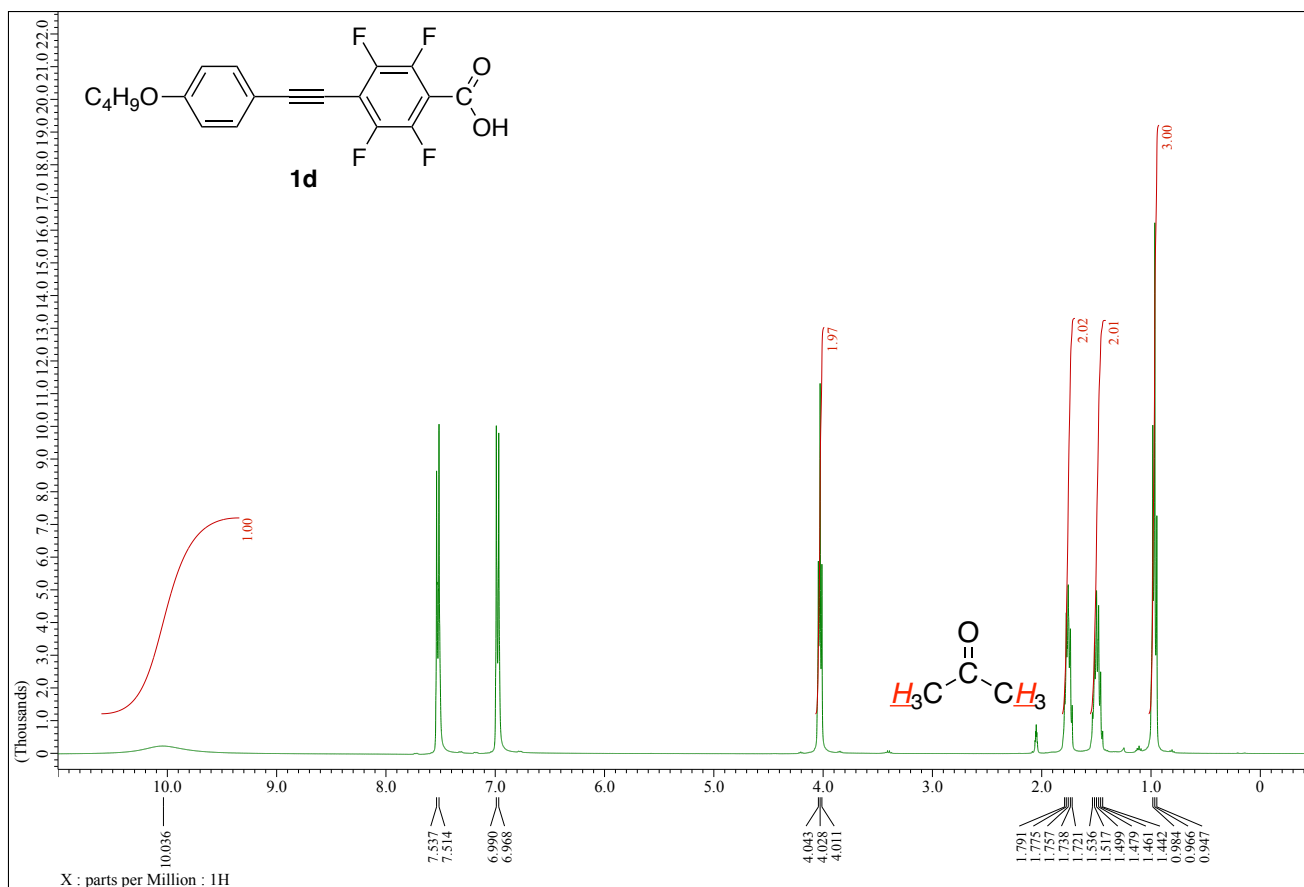


Figure S10. <sup>1</sup>H NMR spectrum of **1d** (acetone-*d*<sub>6</sub>, 400 MHz).

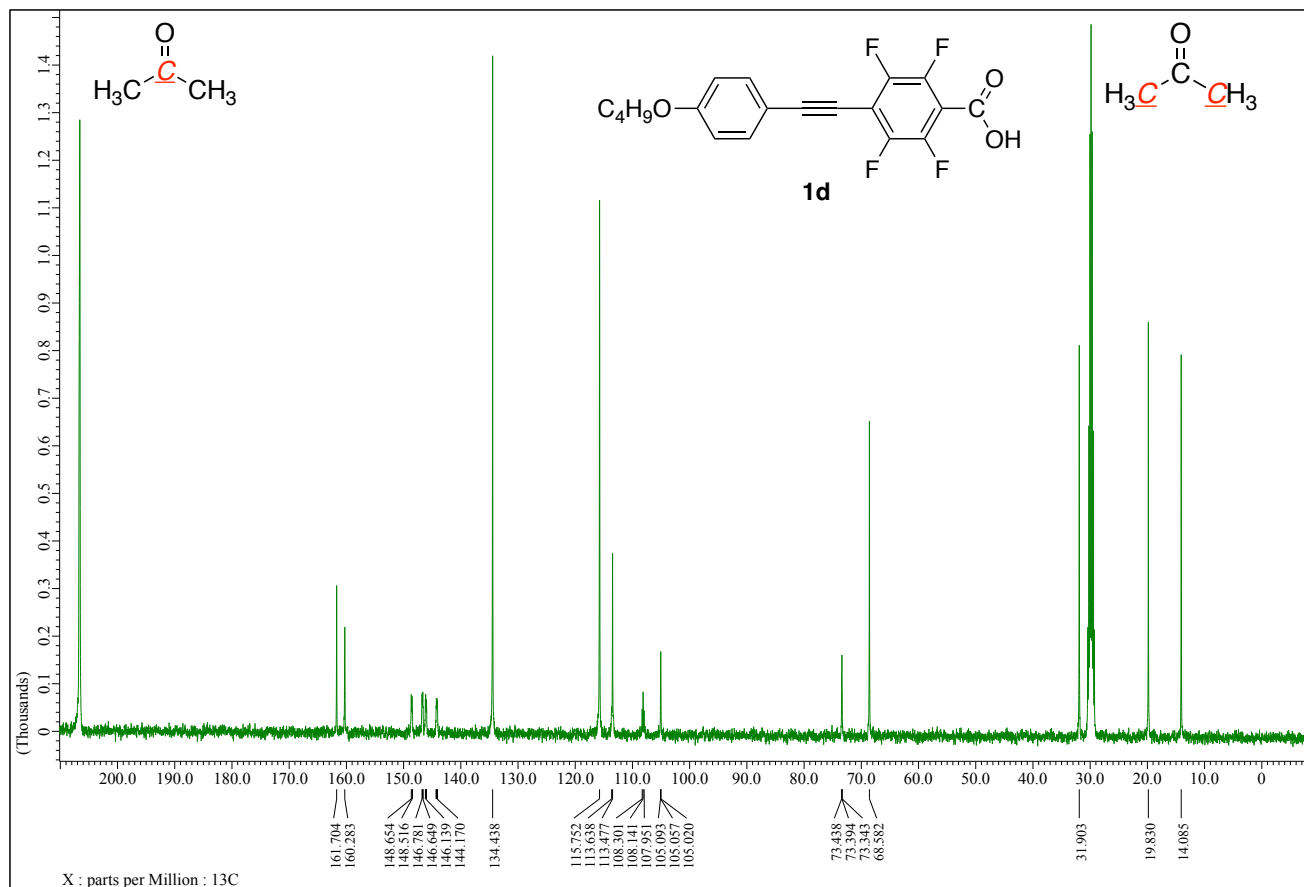
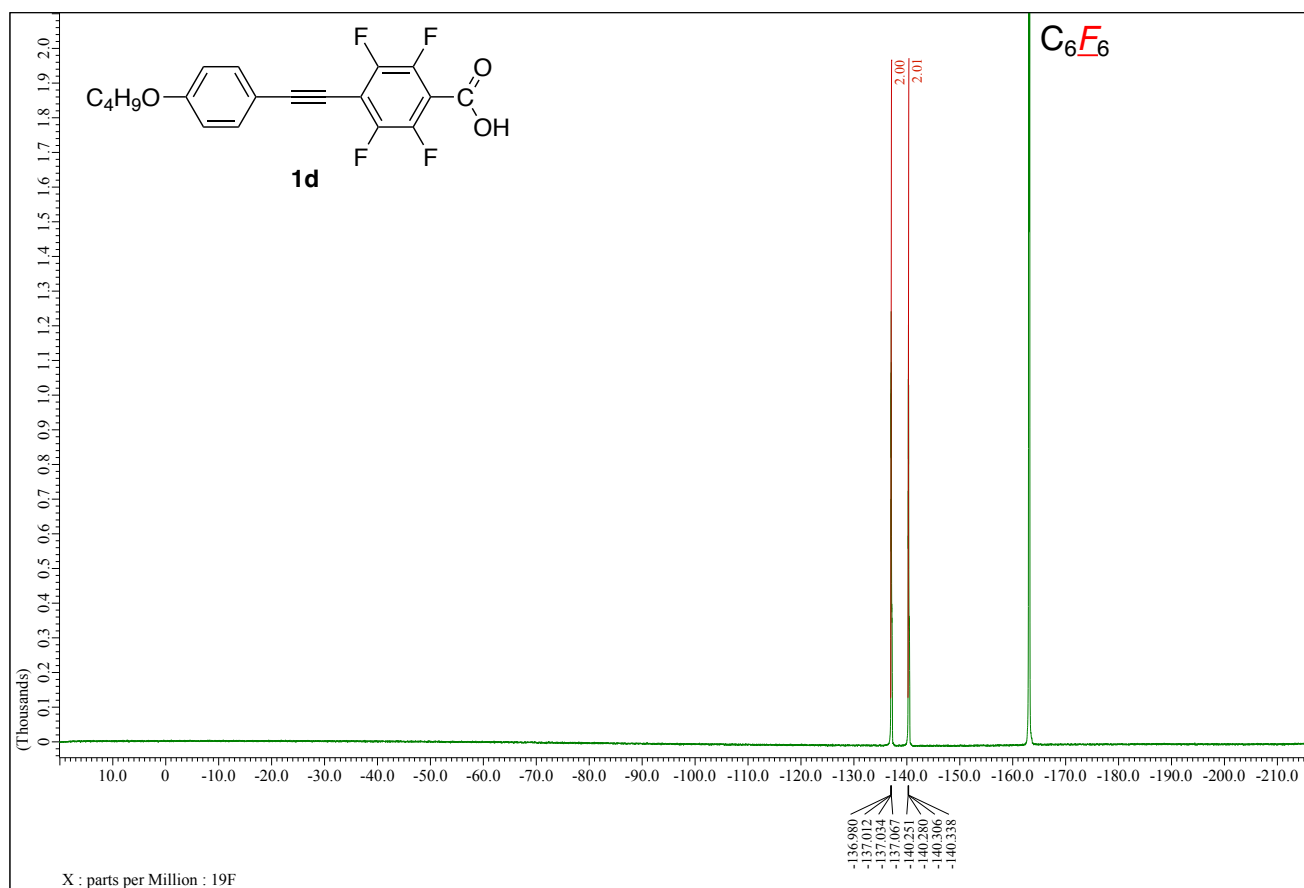
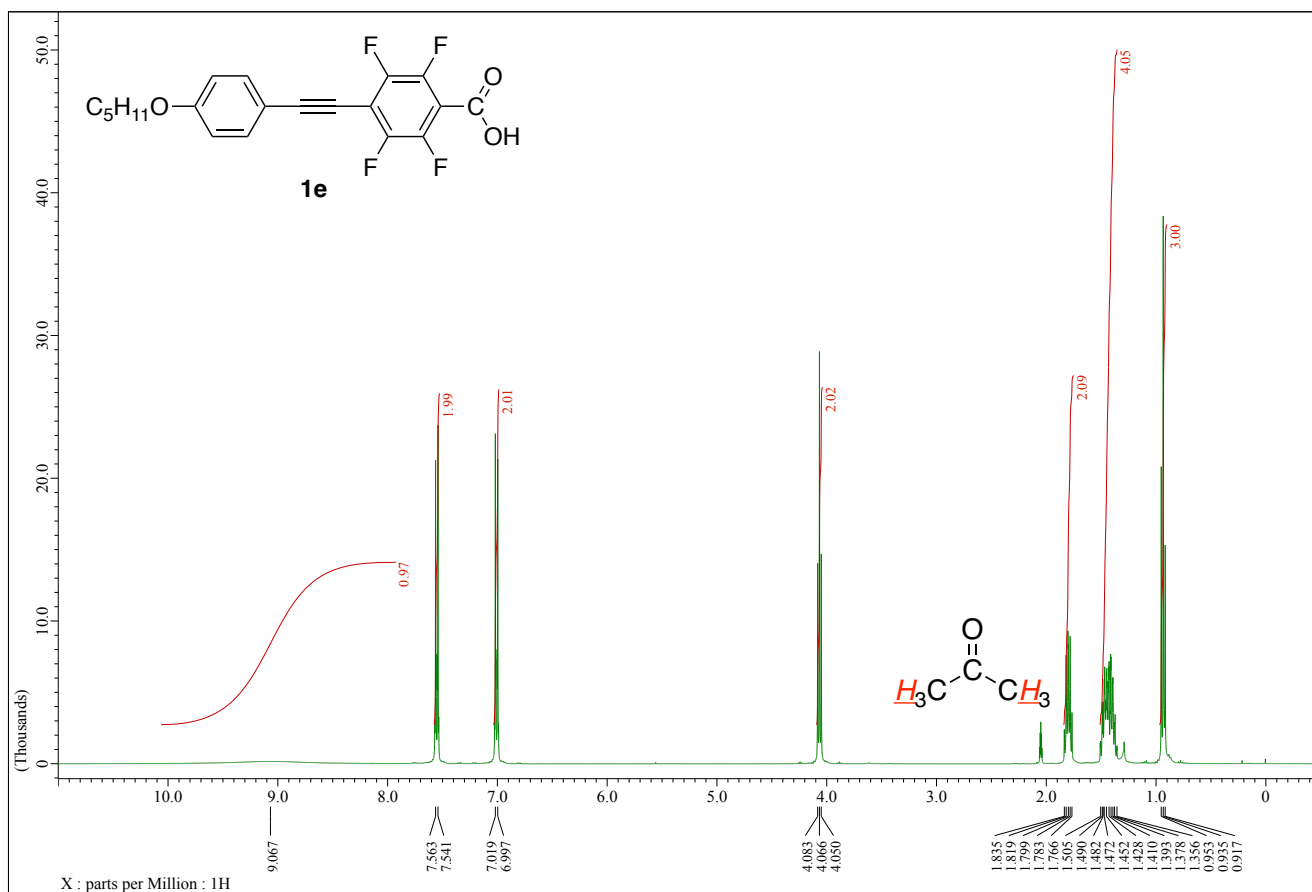


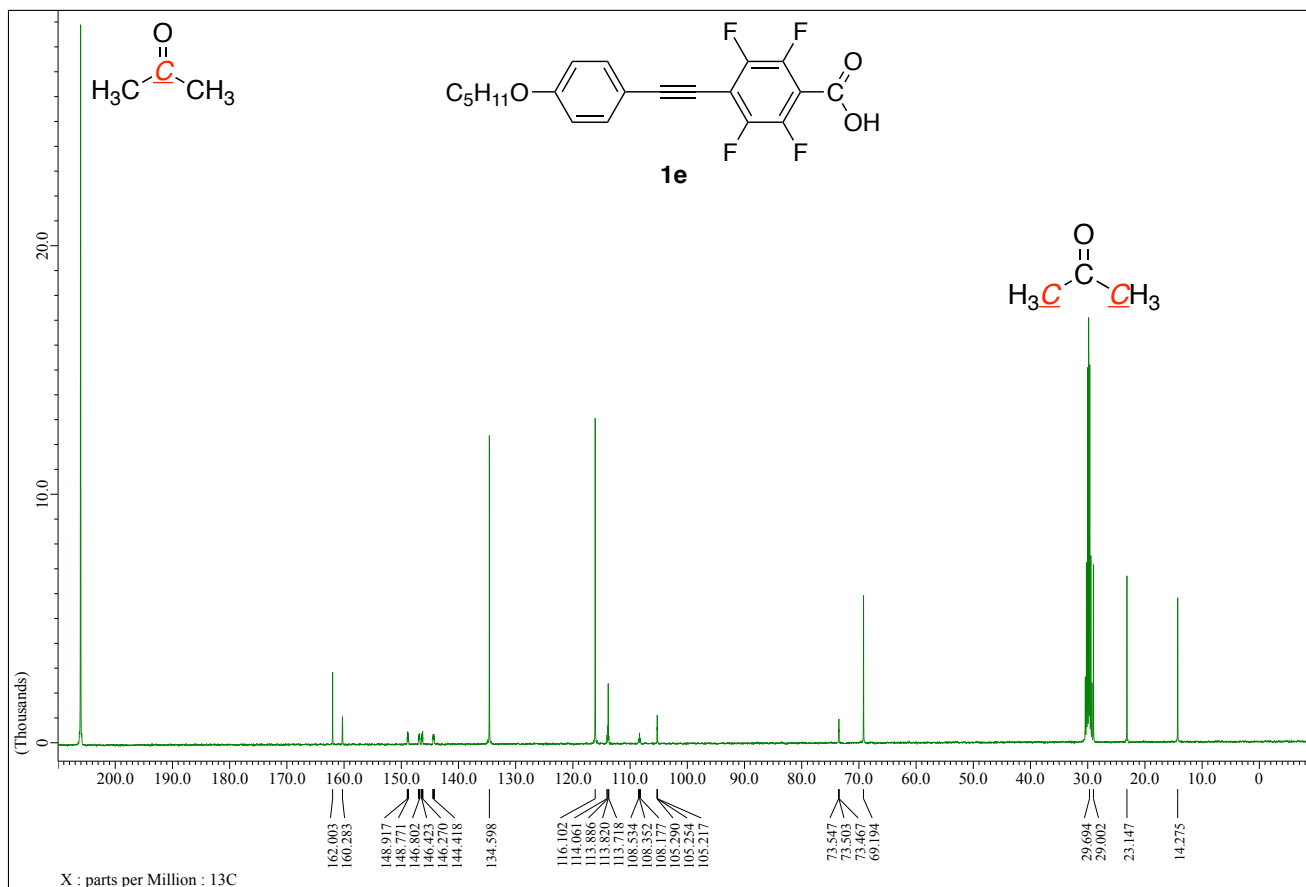
Figure S11. <sup>13</sup>C NMR spectrum of **1d** (acetone-*d*<sub>6</sub>, 100 MHz).



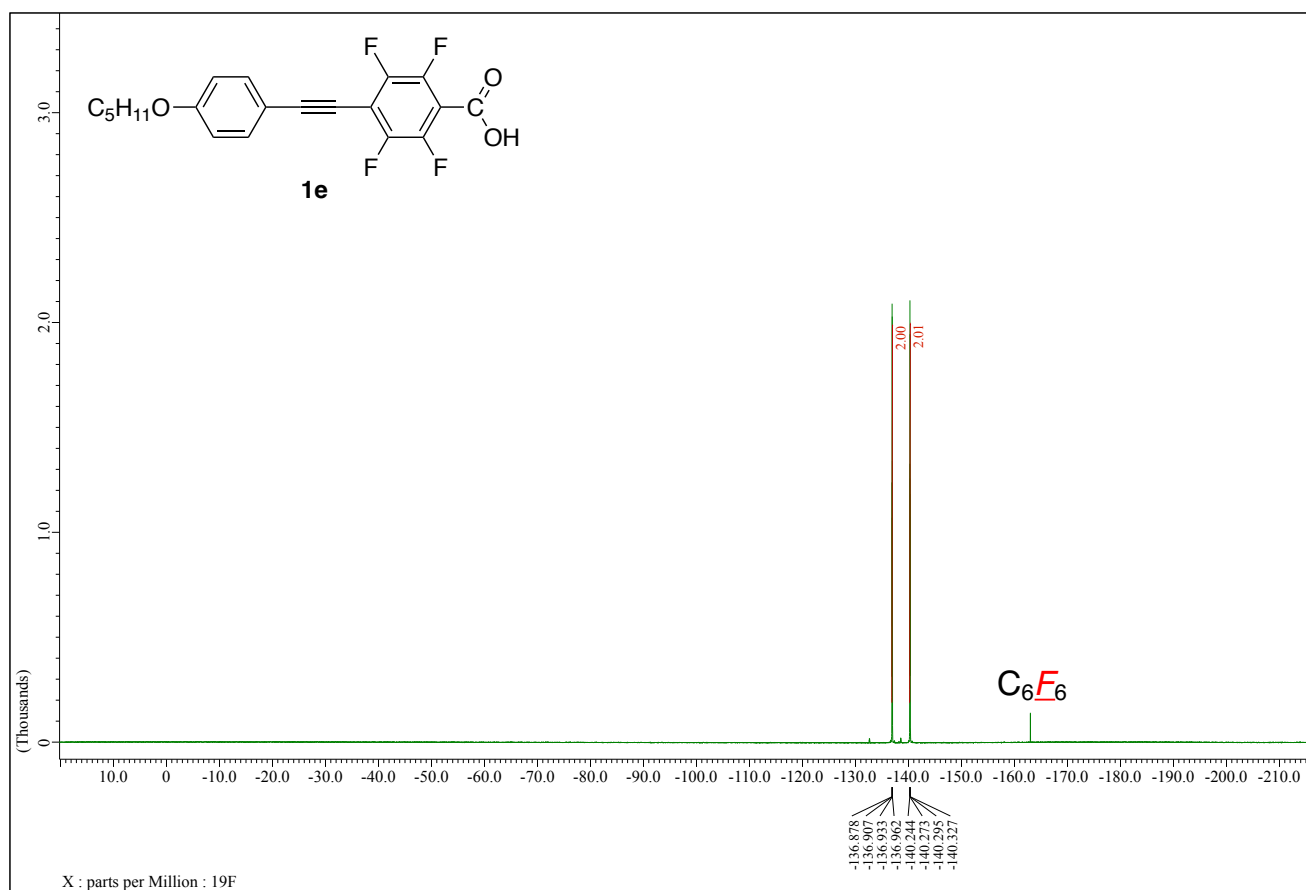
**Figure S12.**  $^{19}F$  NMR spectrum of **1d** (acetone- $d_6$ ,  $C_6F_6$ , 376 MHz).



**Figure S13.** <sup>1</sup>H NMR spectrum of **1e** (acetone-*d*<sub>6</sub>, 400 MHz).



**Figure S14.** <sup>13</sup>C NMR spectrum of **1e** (acetone-*d*<sub>6</sub>, 100 MHz).



**Figure S15.** <sup>19</sup>F NMR spectrum of **1e** (acetone-*d*<sub>6</sub>, C<sub>6</sub>F<sub>6</sub>, 376 MHz).

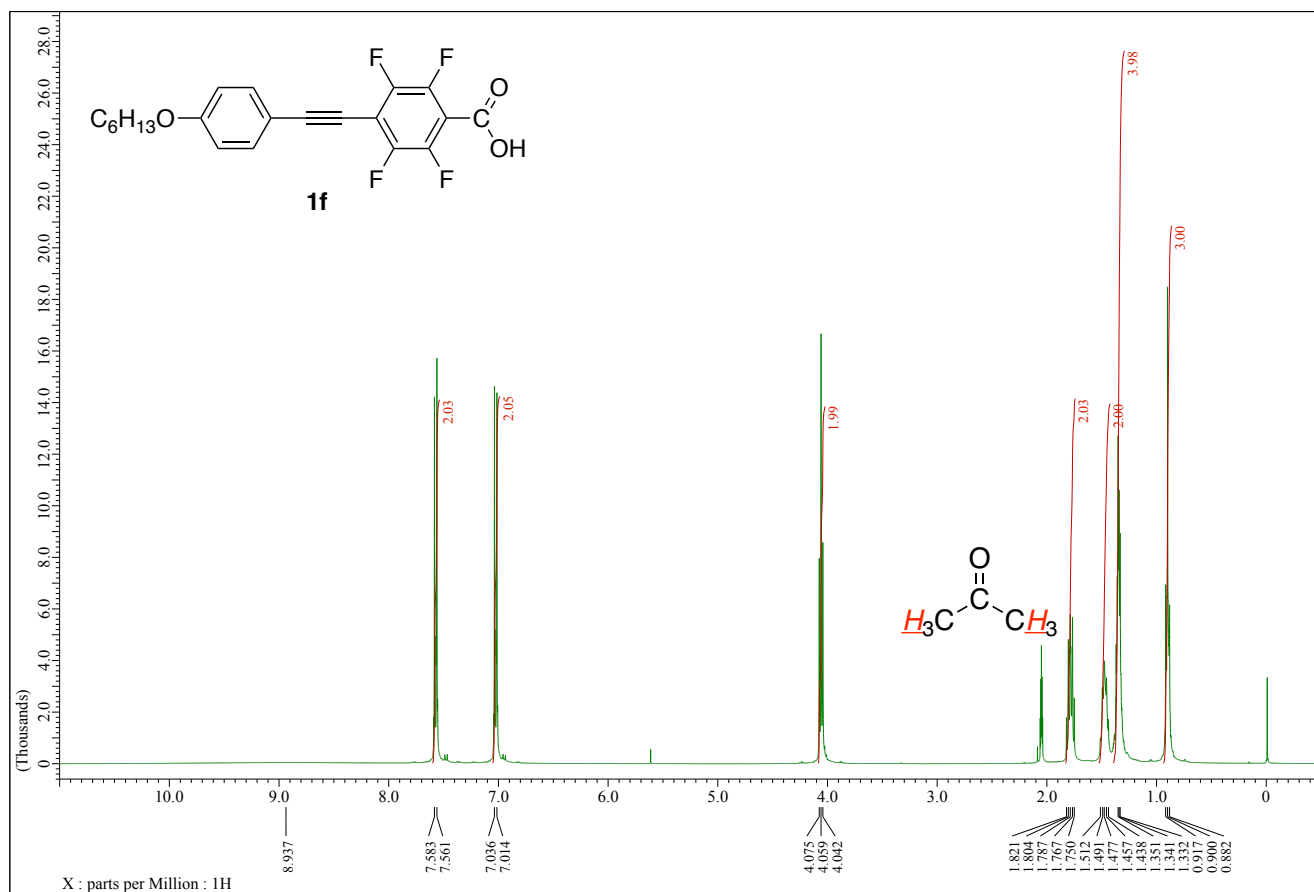


Figure S16. <sup>1</sup>H NMR spectrum of **1f** (acetone-*d*<sub>6</sub>, 400 MHz).

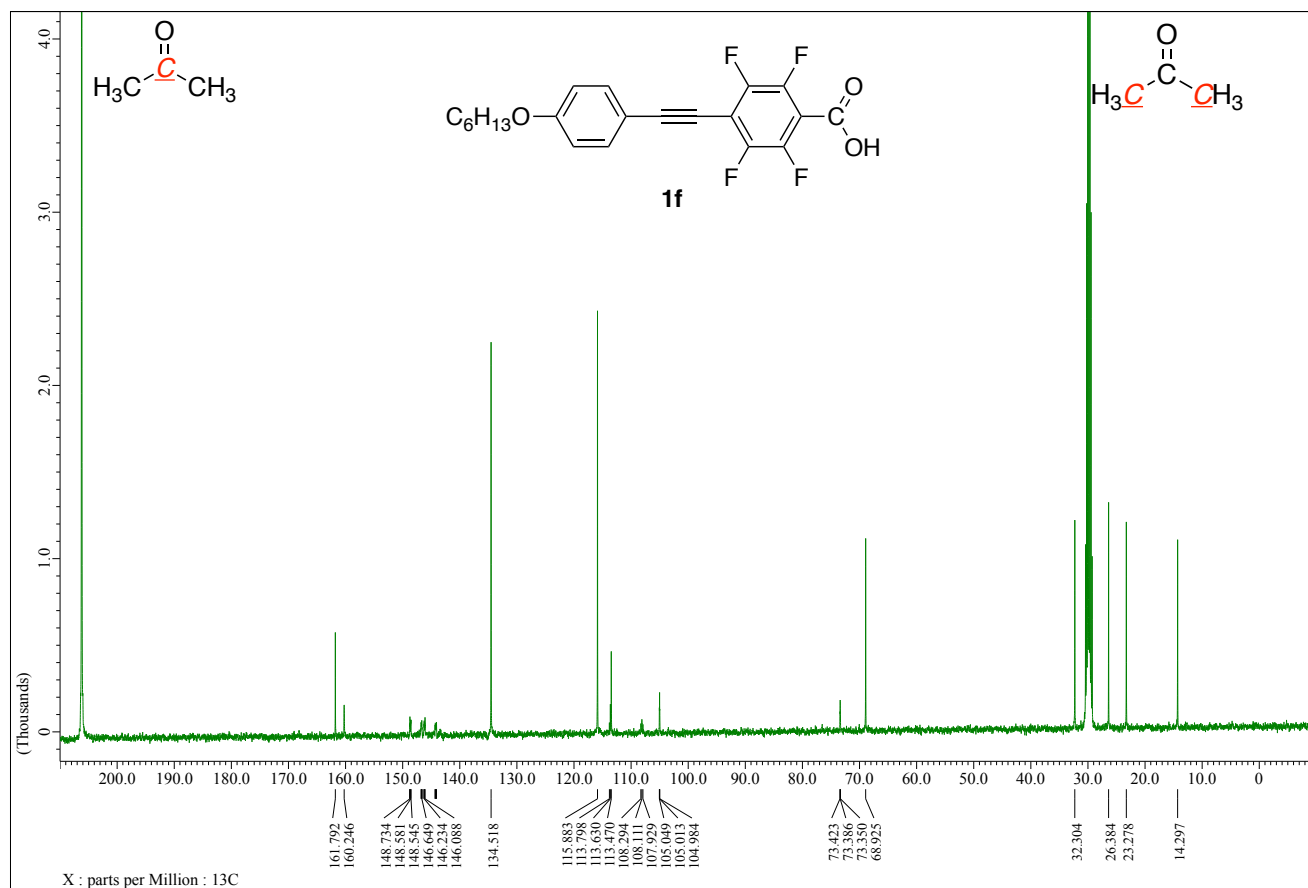
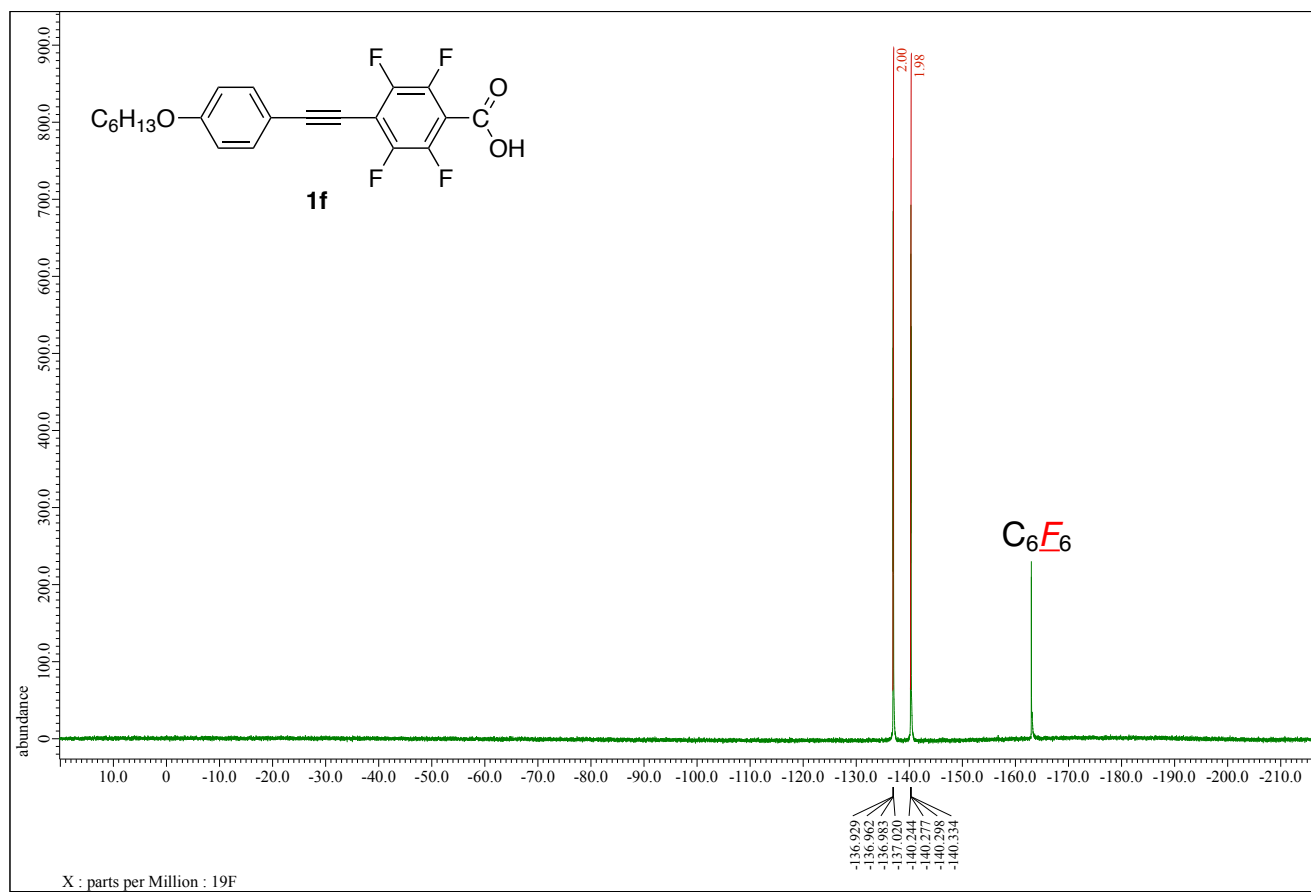
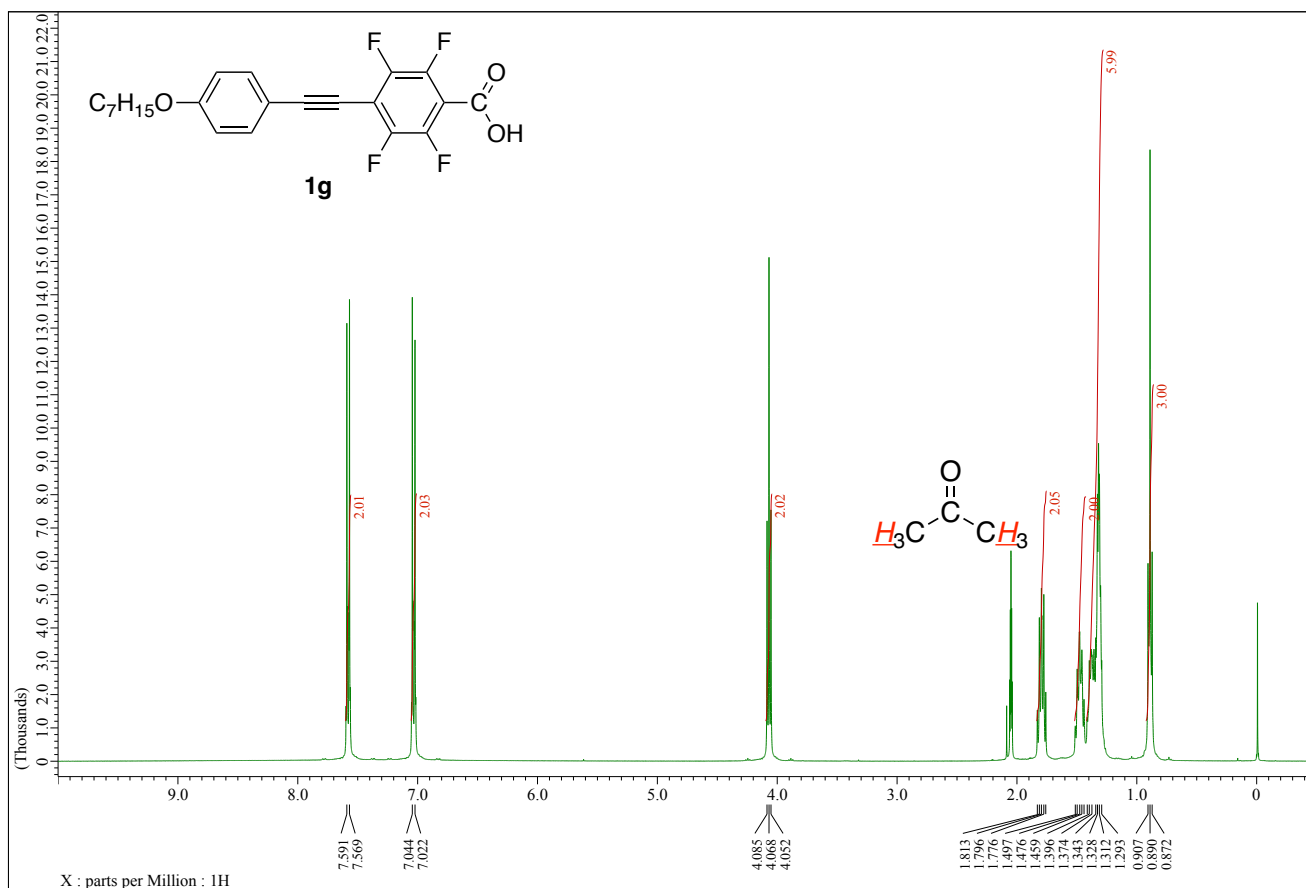


Figure S17. <sup>13</sup>C NMR spectrum of **1f** (acetone-*d*<sub>6</sub>, 100 MHz).

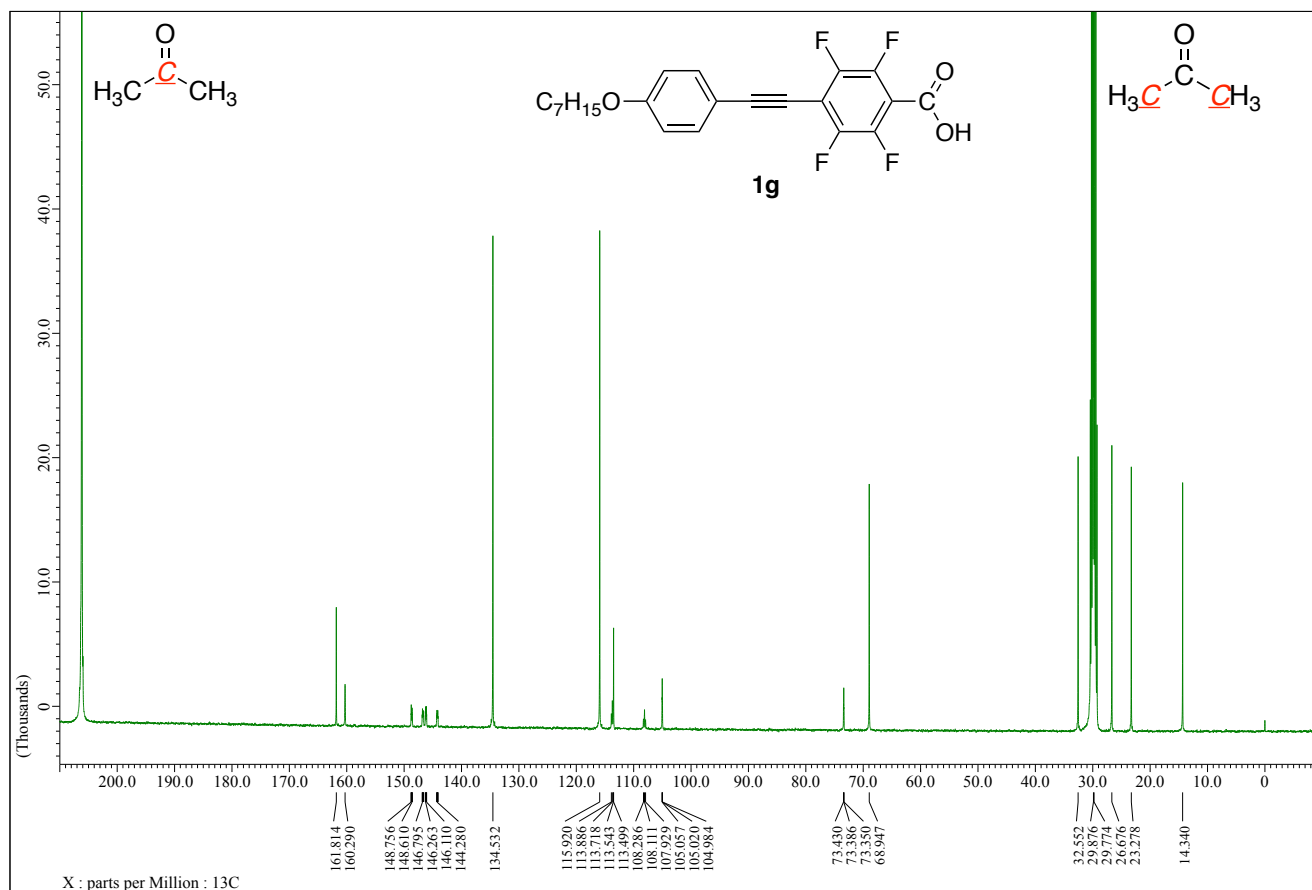




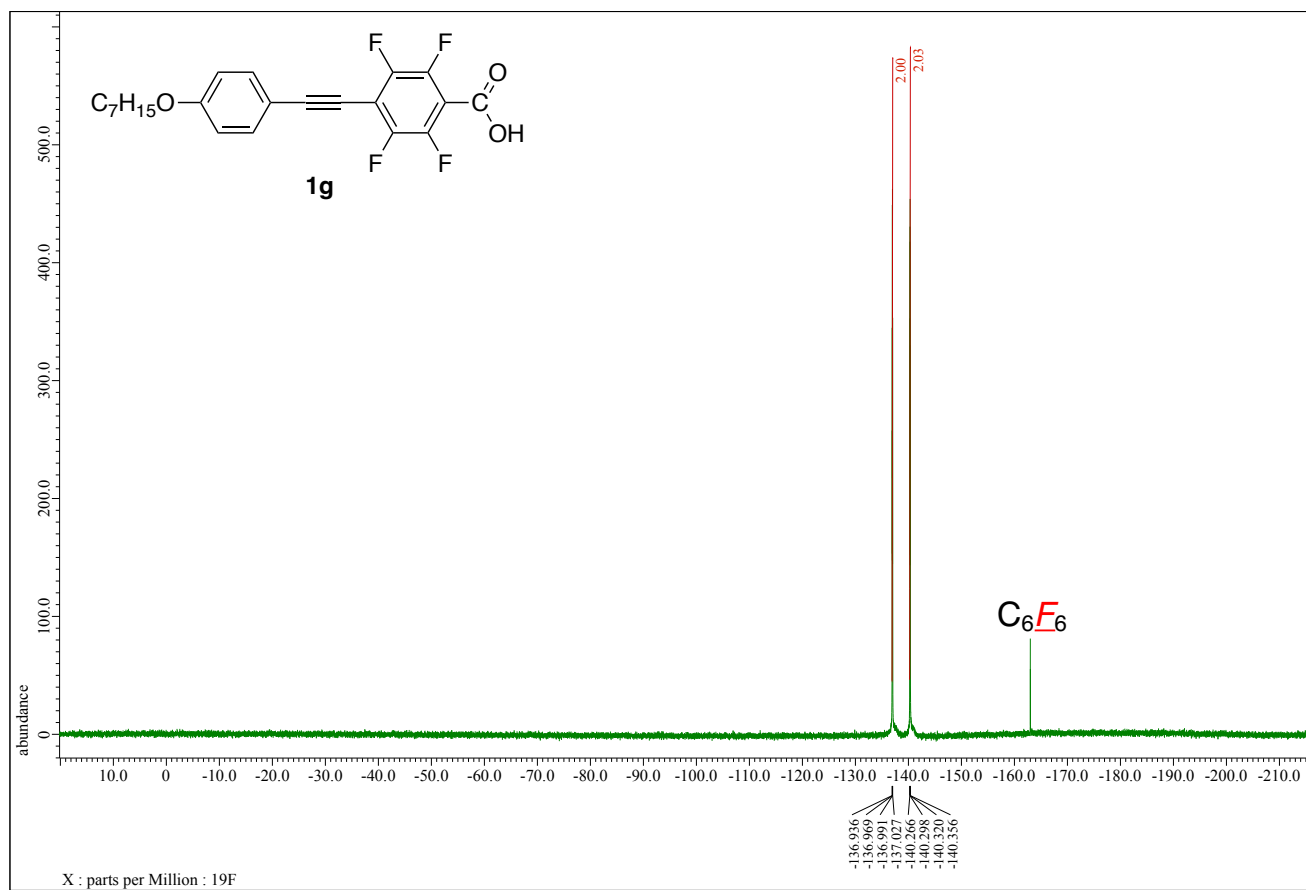
**Figure S18.**  $^{19}F$  NMR spectrum of **1f** (acetone- $d_6$ ,  $C_6F_6$ , 376 MHz).



**Figure S19.**  $^1\text{H}$  NMR spectrum of **1g** (acetone- $d_6$ , 400 MHz).



**Figure S20.**  $^{13}\text{C}$  NMR spectrum of **1g** (acetone- $d_6$ , 100 MHz).



**Figure S21.** <sup>19</sup>F NMR spectrum of **1g** (acetone-*d*<sub>6</sub>, C<sub>6</sub>F<sub>6</sub>, 376 MHz).

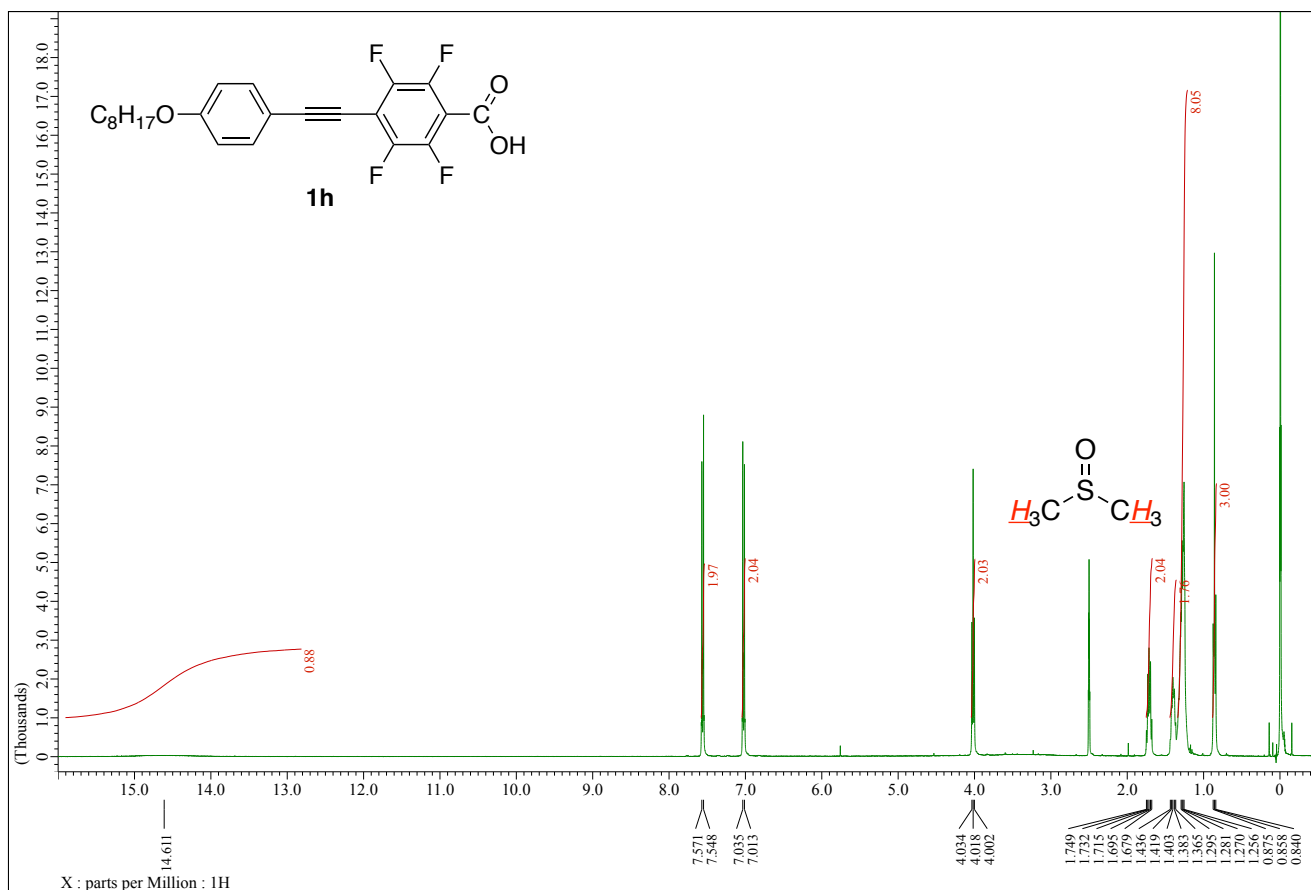


Figure S22. <sup>1</sup>H NMR spectrum of **1h** (DMSO-*d*<sub>6</sub>, 400 MHz).

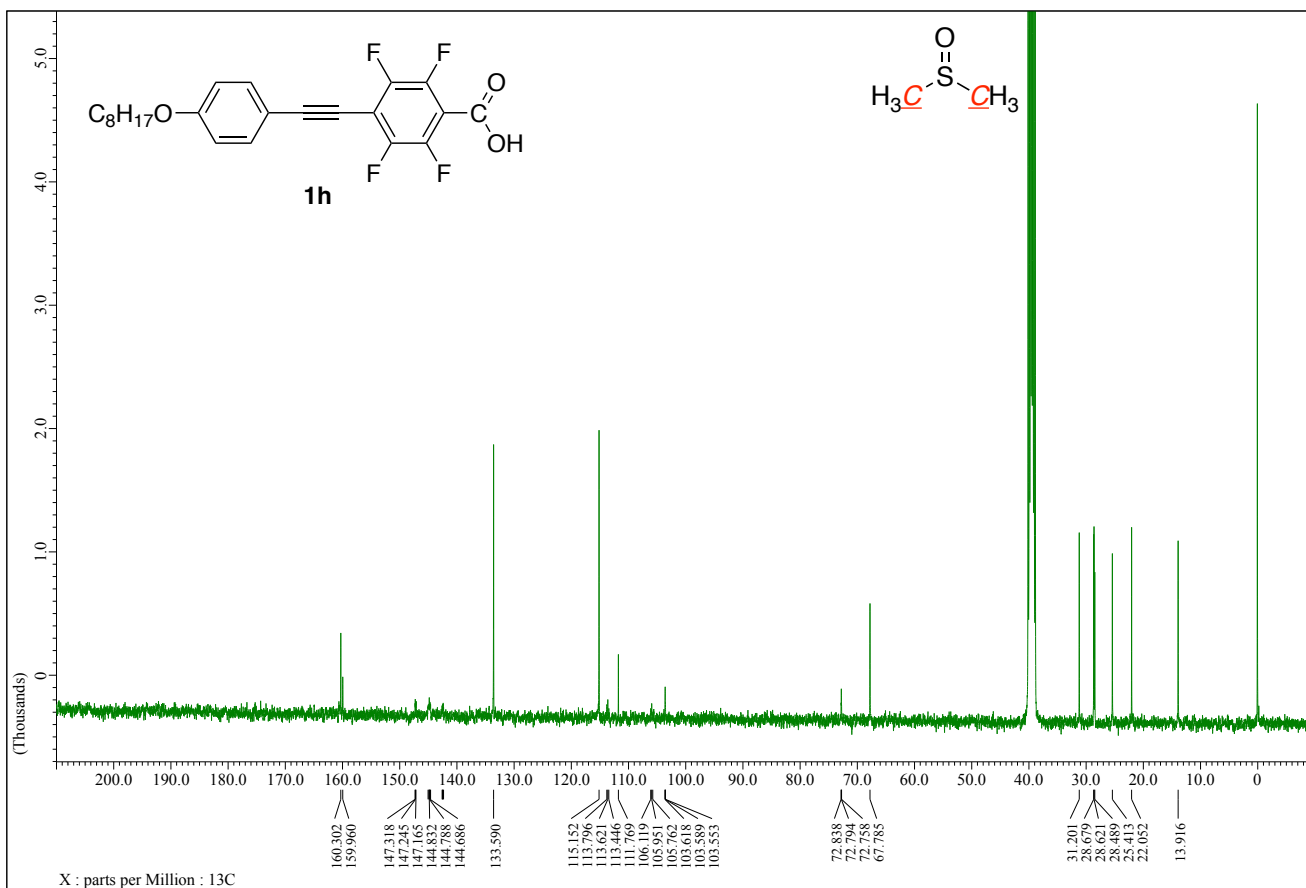
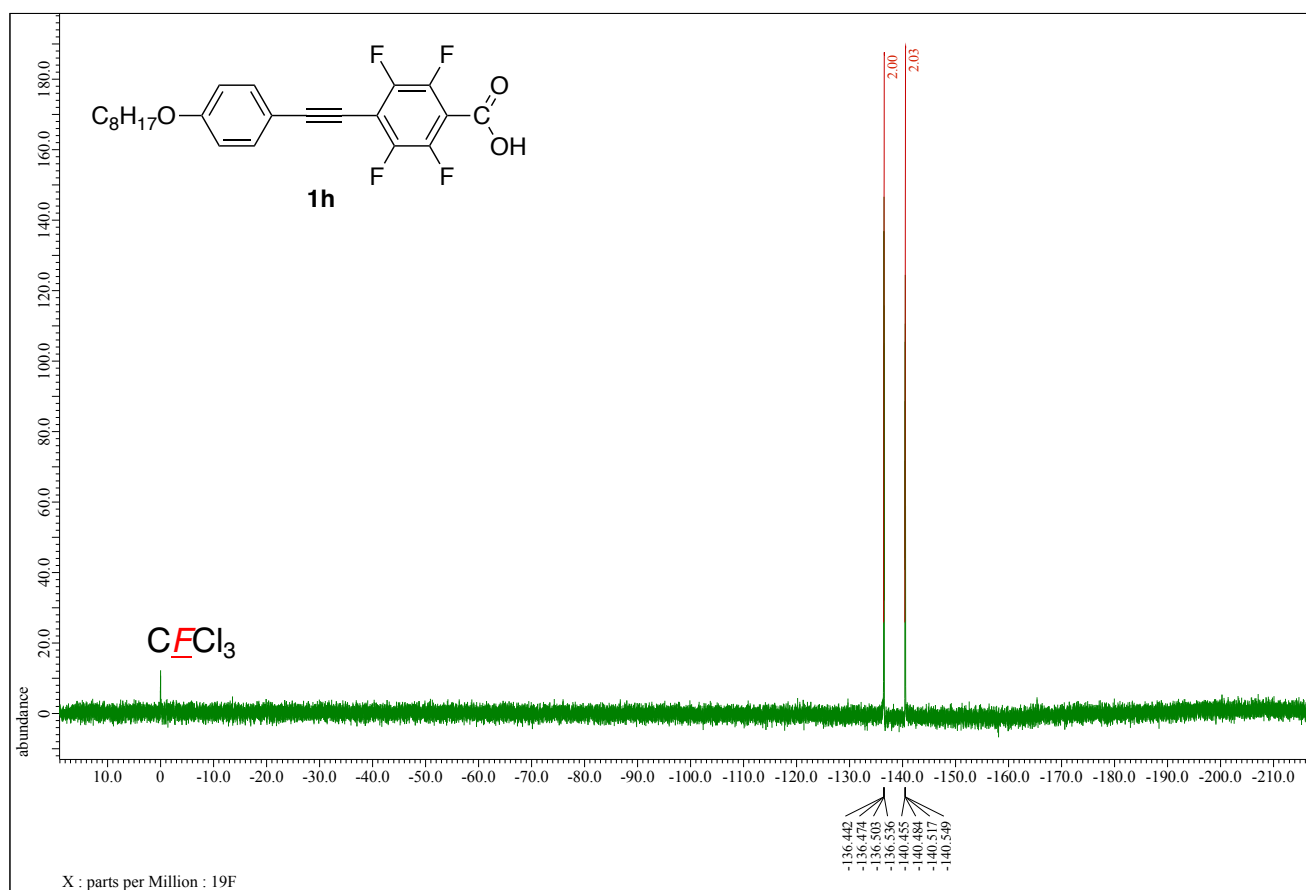


Figure S23. <sup>13</sup>C NMR spectrum of **1h** (DMSO-*d*<sub>6</sub>, 100 MHz).



**Figure S24.**  $^{19}\text{F}$  NMR spectrum of **1h** (DMSO- $d_6$ ,  $\text{CFCl}_3$ , 376 MHz).

## Crystallographic analysis

Single crystal X-ray diffractions were recorded on an XtaLab AFC11 diffractometer (Rigaku). The reflection data were integrated, scaled, and averaged using CrysAlisPro (ver. 1.171.39.43a, Rigaku). Empirical absorption corrections were applied using the SCALE 3 ABSPACK scaling algorithm (CrysAlisPro). The structure were identified by a direct method (SHELXT-2018/2) and refined using a full matrix least square method (SHELXL-2014/7) visualized by Olex2. The crystallographic data were deposited into the Cambridge Crystallographic Data Centre (CCDC) database. These data can be obtained free of charge from the CCDC via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif).

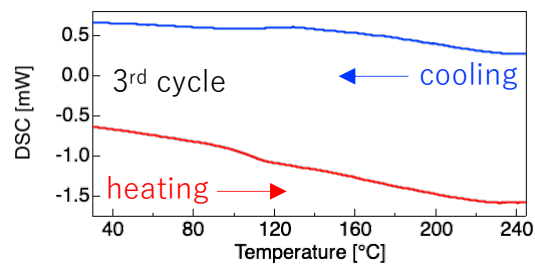
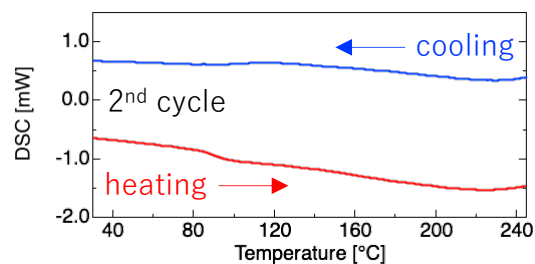
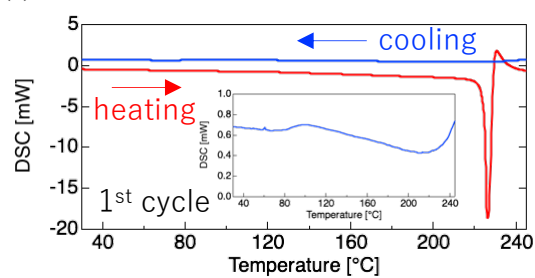
**Table S1.** Crystallographic data of **1a** and **1e**

	<b>1a</b>	<b>1e</b>
CCDC #	2193549	2193550
Empirical Formula	C <sub>16</sub> H <sub>8</sub> F <sub>4</sub> O <sub>3</sub>	C <sub>20</sub> H <sub>16</sub> F <sub>4</sub> O <sub>3</sub>
Formula weight	324.22	380.33
Temperature [K]	293	293
Crystal Color / Habit	Colourless / Block	Colourless / Block
Crystal Size [mm]	0.542 x 0.293 x 0.231	0.550 x 0.330 x 0.190
Crystal System	Triclinic	Monoclinic
Space Group	<i>P</i> −1	<i>C</i> 2/ <i>c</i>
<i>a</i> [Å]	6.4440(3)	23.399(4)
<i>b</i> [Å]	7.329(3)	8.7158(7)
<i>c</i> [Å]	15.3823(7)	19.649(2)
$\alpha$ [°]	95.185(3)	90
$\beta$ [°]	100.896(4)	115.880(14)
$\gamma$ [°]	99.639(3)	90
<i>V</i> [Å <sup>3</sup> ]	701.88(6)	3605.3(8)
<i>Z</i>	2	8
<i>R</i> [ <i>F</i> <sup>2</sup> > 2 <i>s</i> ( <i>F</i> <sup>2</sup> )] [a]	0.0480	0.0559
<i>wR</i> <sub>2</sub> ( <i>F</i> <sup>2</sup> ) [b]	0.1528	0.1888

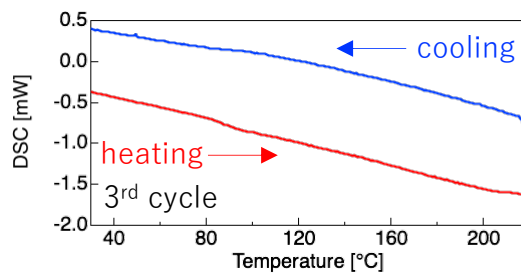
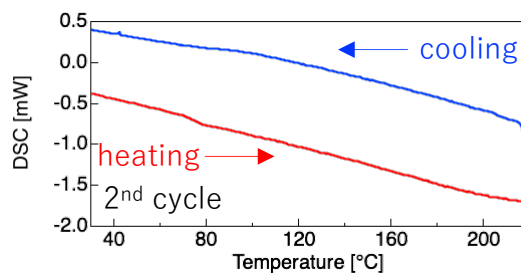
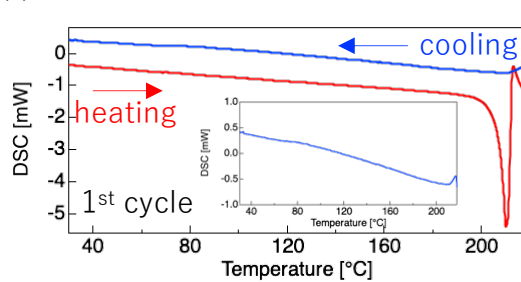
$$[a] R = \Sigma ||F_o| - |F_c|| / \Sigma |F_o|. [b] wR = \{[\Sigma w(|F_o| - |F_c|)] / \Sigma w|F_o|\}^{1/2}.$$

## Phase transition behavior

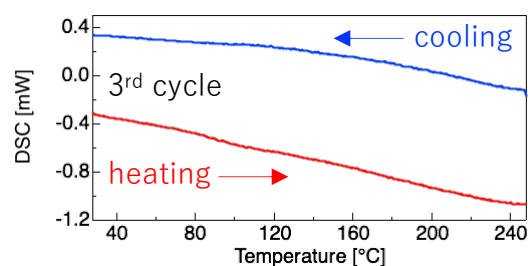
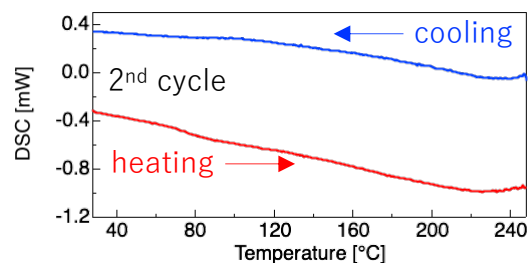
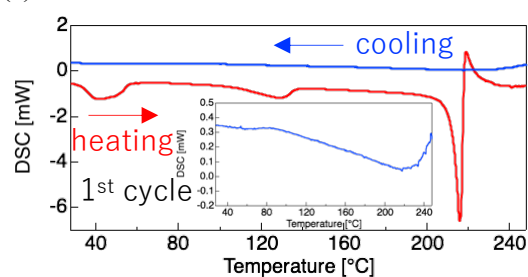
(a) 1a



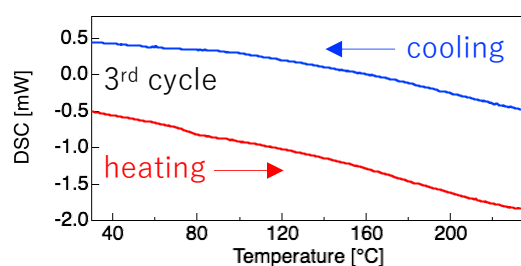
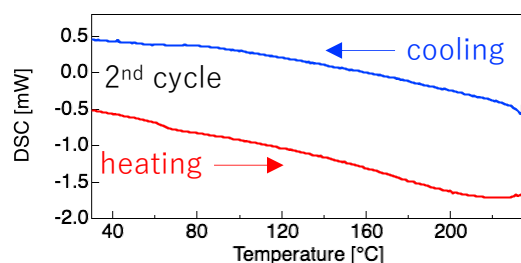
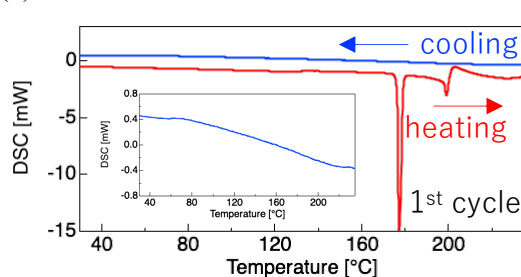
(b) 1b



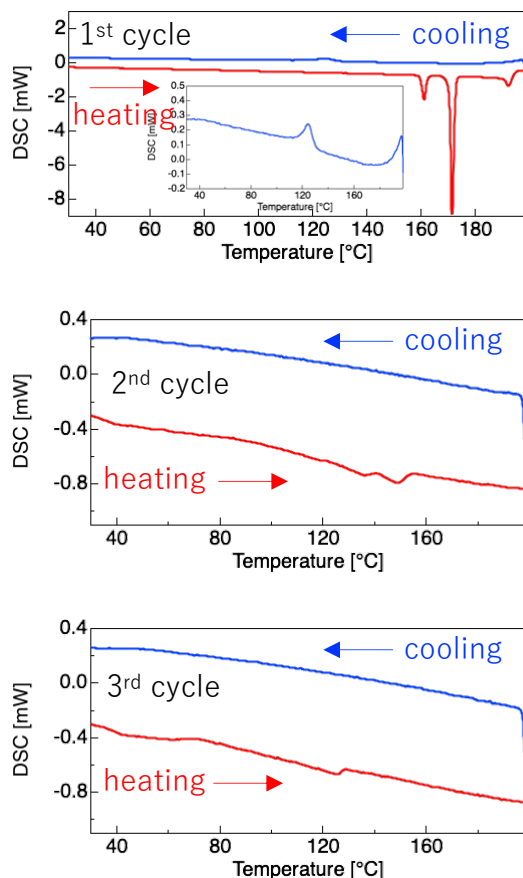
(c) 1c



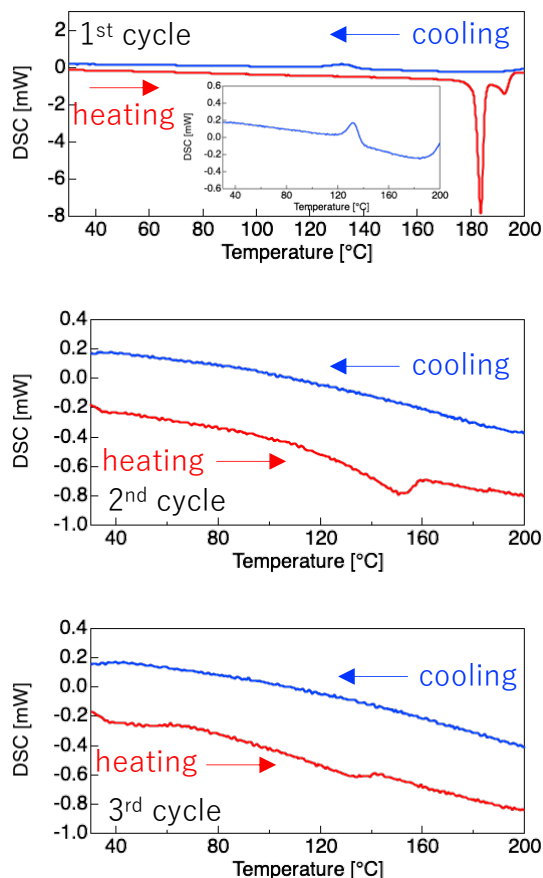
(d) 1d



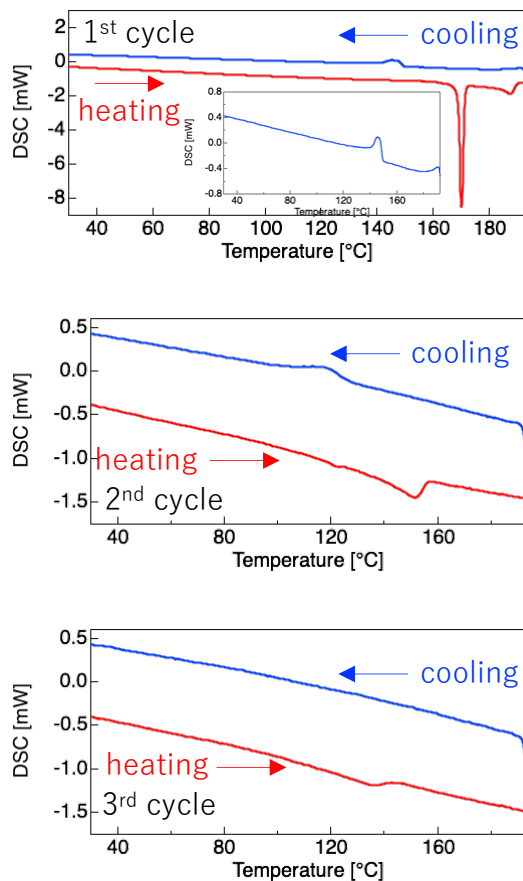
(e) **1e**



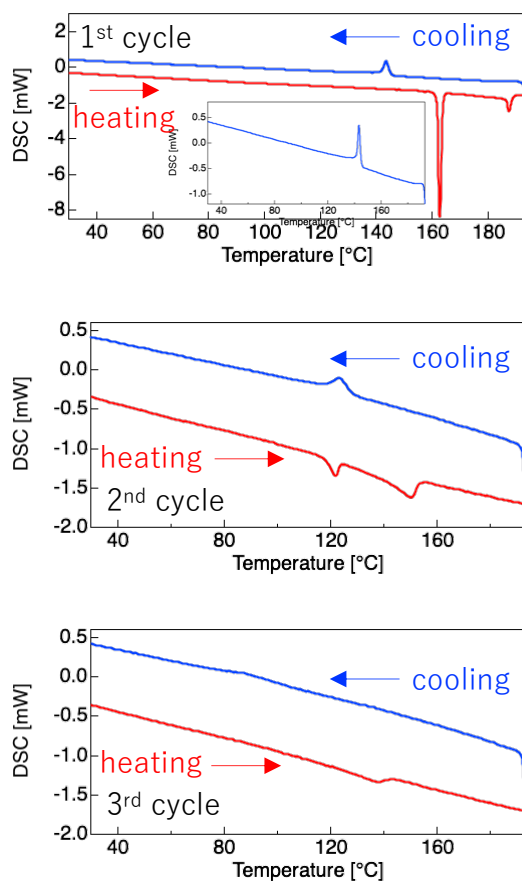
(f) **1f**



(g) **1g**



(h) **1h**



**Figure S25.** DSC Thermograms of **1a–h** under N<sub>2</sub> atmosphere. Scan rate: 5.0 °C min<sup>-1</sup> or 10 °C min<sup>-1</sup>.



**Table S2.** Phase transition behavior of **1a–h** observed by DSC measurement.

Molecule	Phase sequence		Temperature [°C]	Enthalpy ( $\Delta H$ ) [kJ mol <sup>-1</sup> ]	Entropy ( $\Delta S$ ) [J mol <sup>-1</sup> K <sup>-1</sup> ]
<b>1a</b>	1 <sup>st</sup> Heating	Cry-Iso	224	47.4	96.0
	1 <sup>st</sup> Cooling	Iso-G	165	— a	— a
	2 <sup>nd</sup> Heating	G-Iso	158	— a	— a
	2 <sup>nd</sup> Cooling	Iso-G	168	— a	— a
<b>1b</b>	1 <sup>st</sup> Heating	Cry-Iso	207	44.8	91.0
	1 <sup>st</sup> Cooling	Iso-G	140	—	—
	2 <sup>nd</sup> Heating	G-Iso	137	— a	— a
	2 <sup>nd</sup> Cooling	Iso-G	146	— a	— a
<b>1c</b>	1 <sup>st</sup> Heating	Cry-Iso	211	49.6	102.0
	1 <sup>st</sup> Cooling	Iso–Cry	158	— a	— a
	2 <sup>nd</sup> Heating	G-Iso	142	— a	— a
	2 <sup>nd</sup> Cooling	Iso-G	137	— a	— a
<b>1d</b>	1 <sup>st</sup> Heating	Cry-N	176	29.9	66.7
		N-Iso	198	7.27	15.4
	1 <sup>st</sup> Cooling	Iso-G	124	— a	— a
	2 <sup>nd</sup> Heating	G-Iso	111	— a	— a
	2 <sup>nd</sup> Cooling	Iso-G	105	— a	— a
<b>1e</b>	1 <sup>st</sup> Heating	Cry <sup>1</sup> -Cry <sup>2</sup>	160	8.9	20.4
		Cry <sup>2</sup> -N	170	39.3	88.6
		N-Iso	191	5.0	10.8
	1 <sup>st</sup> Cooling	Iso-G	132	— a	— a
	2 <sup>nd</sup> Heating	G-Iso	137	— a	— a
	2 <sup>nd</sup> Cooling	Iso-G	134	— a	— a
<b>1f</b>	1 <sup>st</sup> Heating	Cry-N	181	27.2	60.0
		N-Iso	190	3.77	8.14
	1 <sup>st</sup> Cooling	Iso-G	141	— a	— a
	2 <sup>nd</sup> Heating	G-Iso	122	— a	— a
	2 <sup>nd</sup> Cooling	Iso-G	121	— a	— a
<b>1g</b>	1 <sup>st</sup> Heating	Cry-N	169	29.2	66.1
		N-Iso	184	7.63	16.7
	1 <sup>st</sup> Cooling	Iso-G	150	−5.03	−11.9
	2 <sup>nd</sup> Heating	G-Iso	131	— a	— a
	2 <sup>nd</sup> Cooling	Iso-G	127	— a	— a
<b>1h</b>	1 <sup>st</sup> Heating	Cry-N	161	31.2	71.6
		N-Iso	186	5.53	12.0
	1 <sup>st</sup> Cooling	Iso-N	146	−7.01	−16.8
	2 <sup>nd</sup> Heating	G-N	118	— a	— a

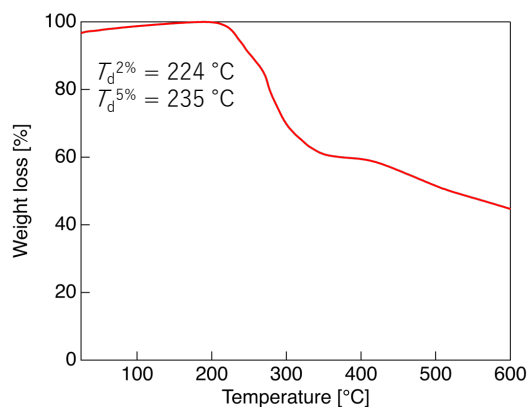
	N-Iso	137	— <sup>a</sup>	— <sup>a</sup>
2 <sup>nd</sup> Cooling	Iso-G	131	— <sup>a</sup>	— <sup>a</sup>

---

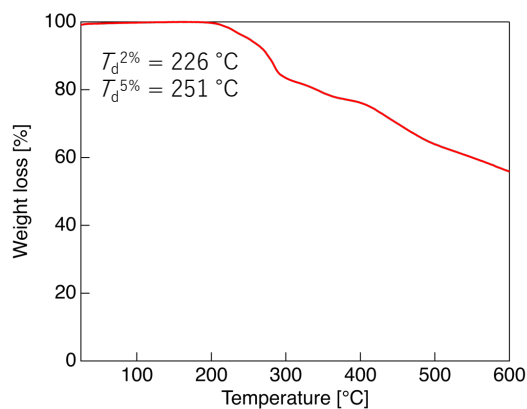
<sup>a</sup> Determined by POM observation.

## TG Analysis

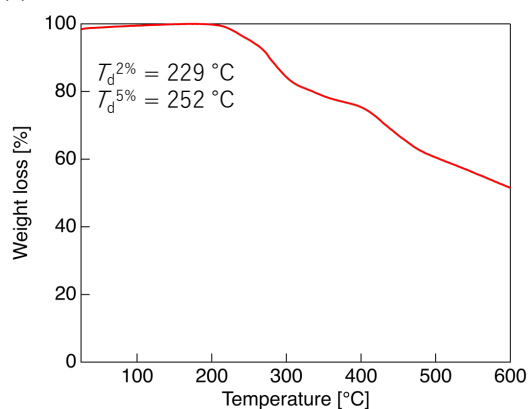
(a) **1a**



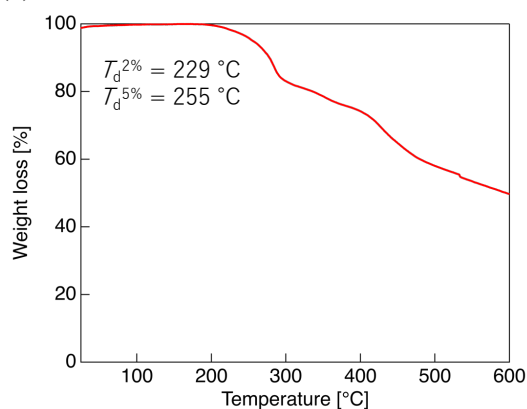
(b) **1b**



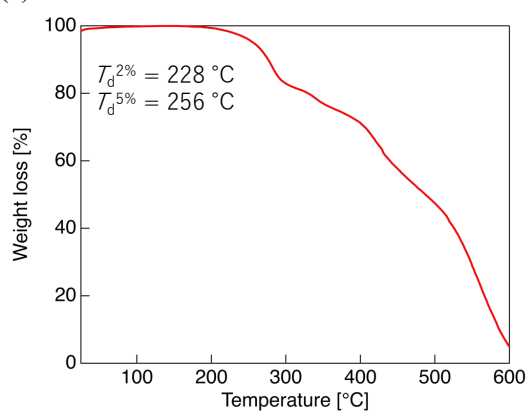
(c) **1c**



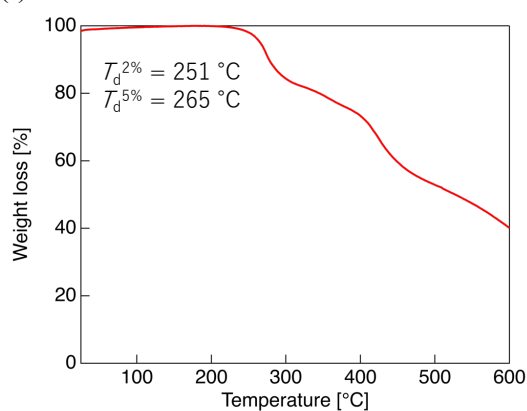
(d) **1d**



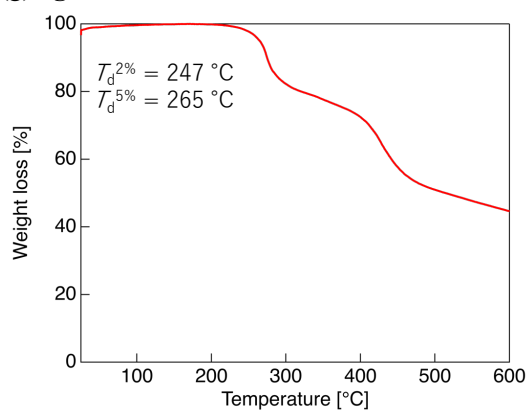
(e) **1e**



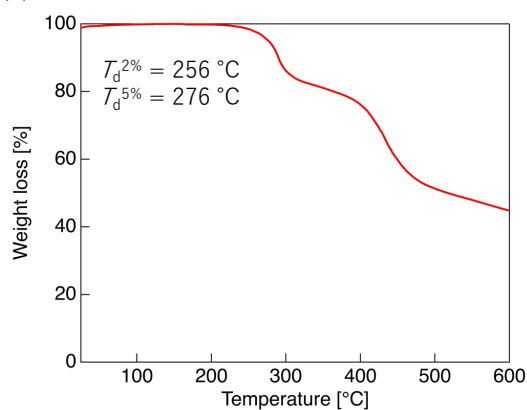
(f) **1f**



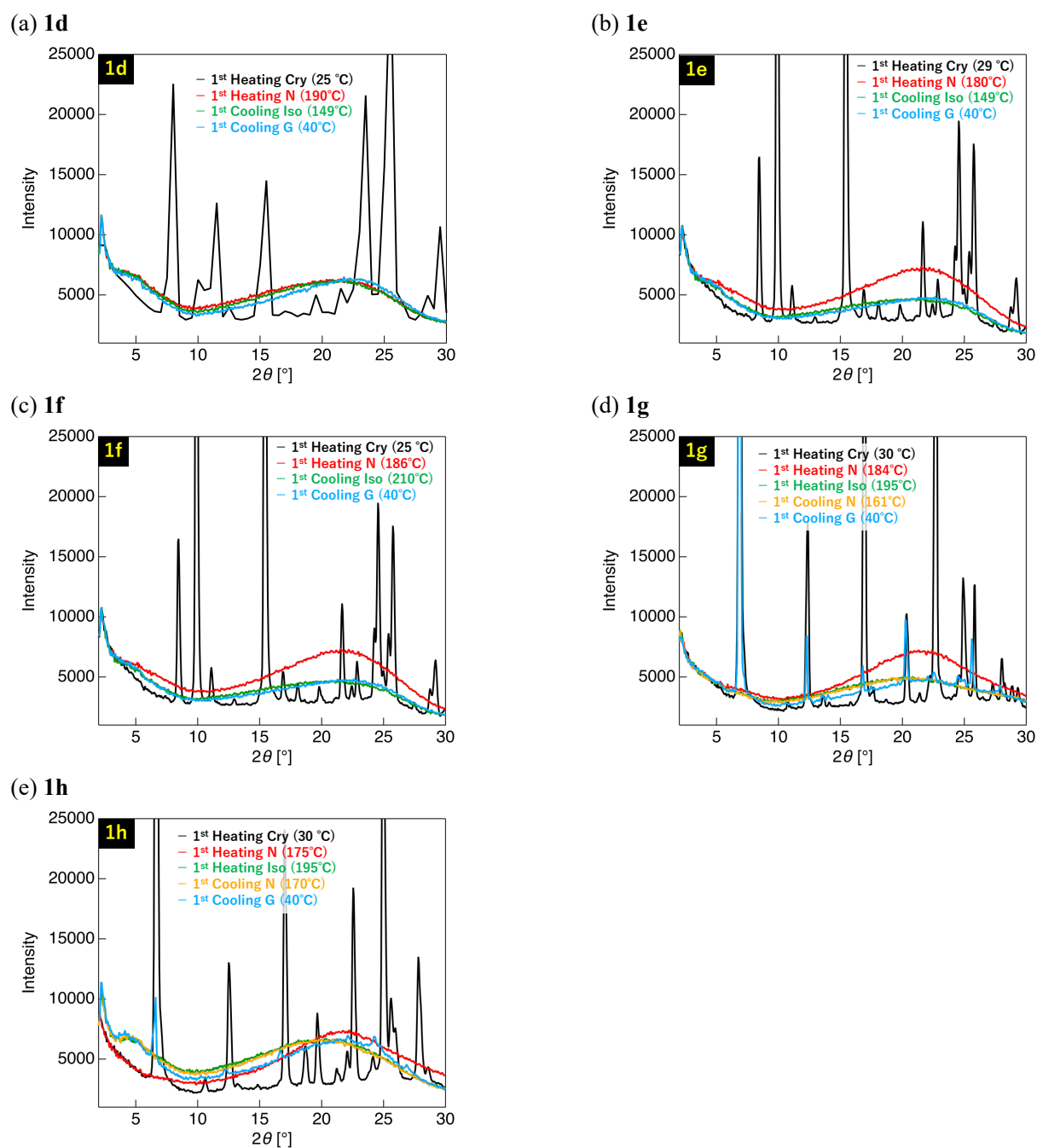
(g) **1g**



(h) **1h**



**Figure S26.** TGA Thermograms of **1a–h** under N<sub>2</sub> atmosphere. Scan rate: 10 °C min<sup>-1</sup>.

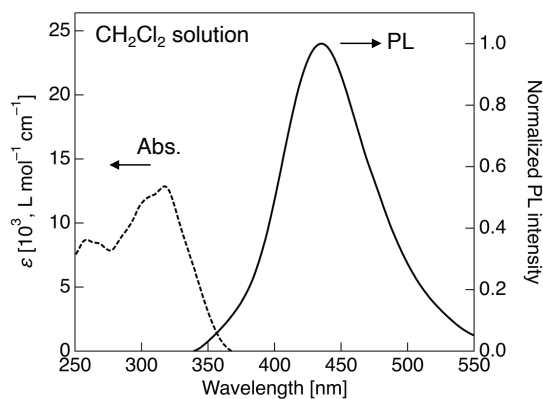


**Figure S27.** VT-PXRD pattern of **1d–h** during the 1<sup>st</sup> heating and cooling processes.

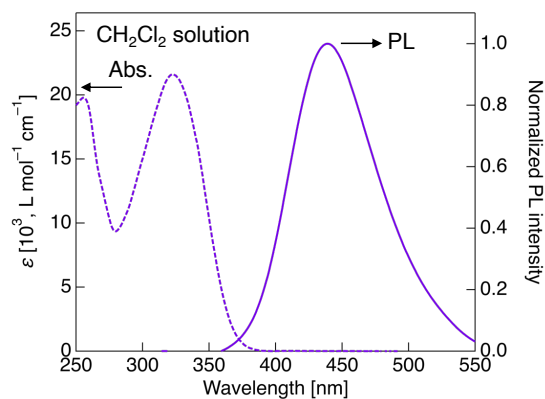
## Photophysical behavior

### CH<sub>2</sub>Cl<sub>2</sub> solution phase

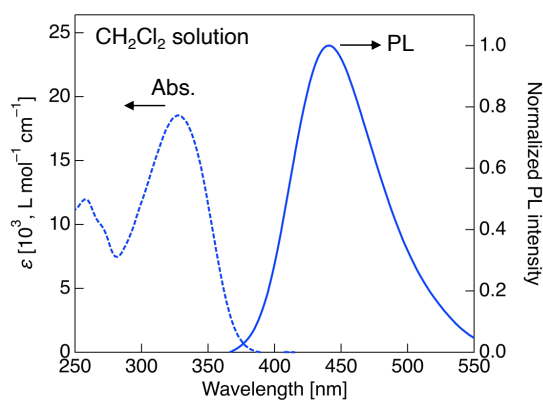
(a) **1a**



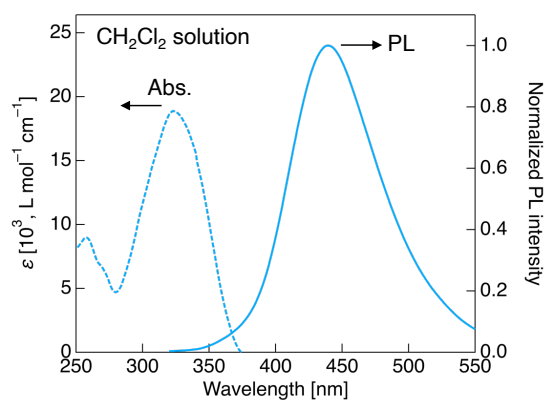
(b) **1b**



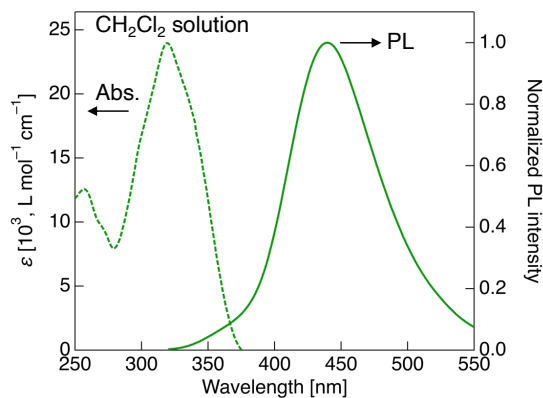
(c) **1c**



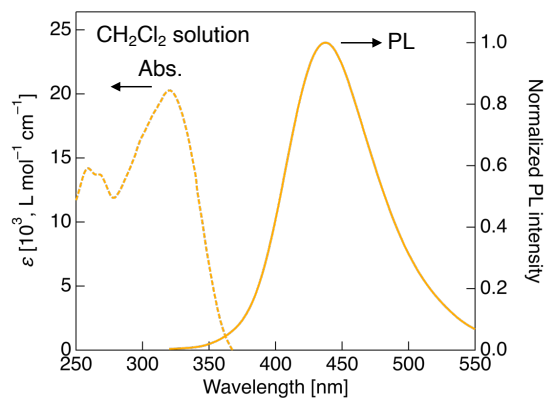
(d) **1d**



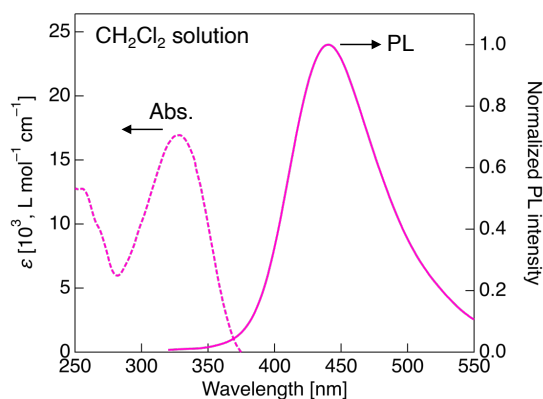
(e) **1e**



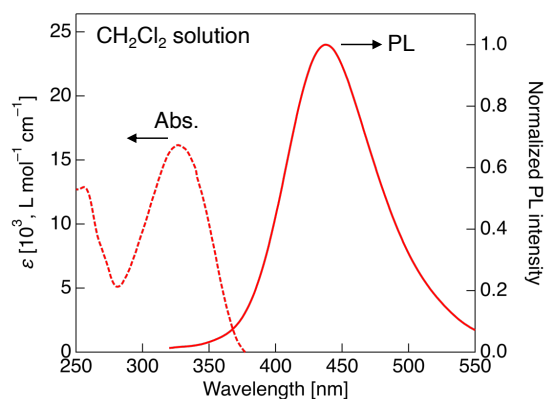
(f) **1f**



(g) **1g**

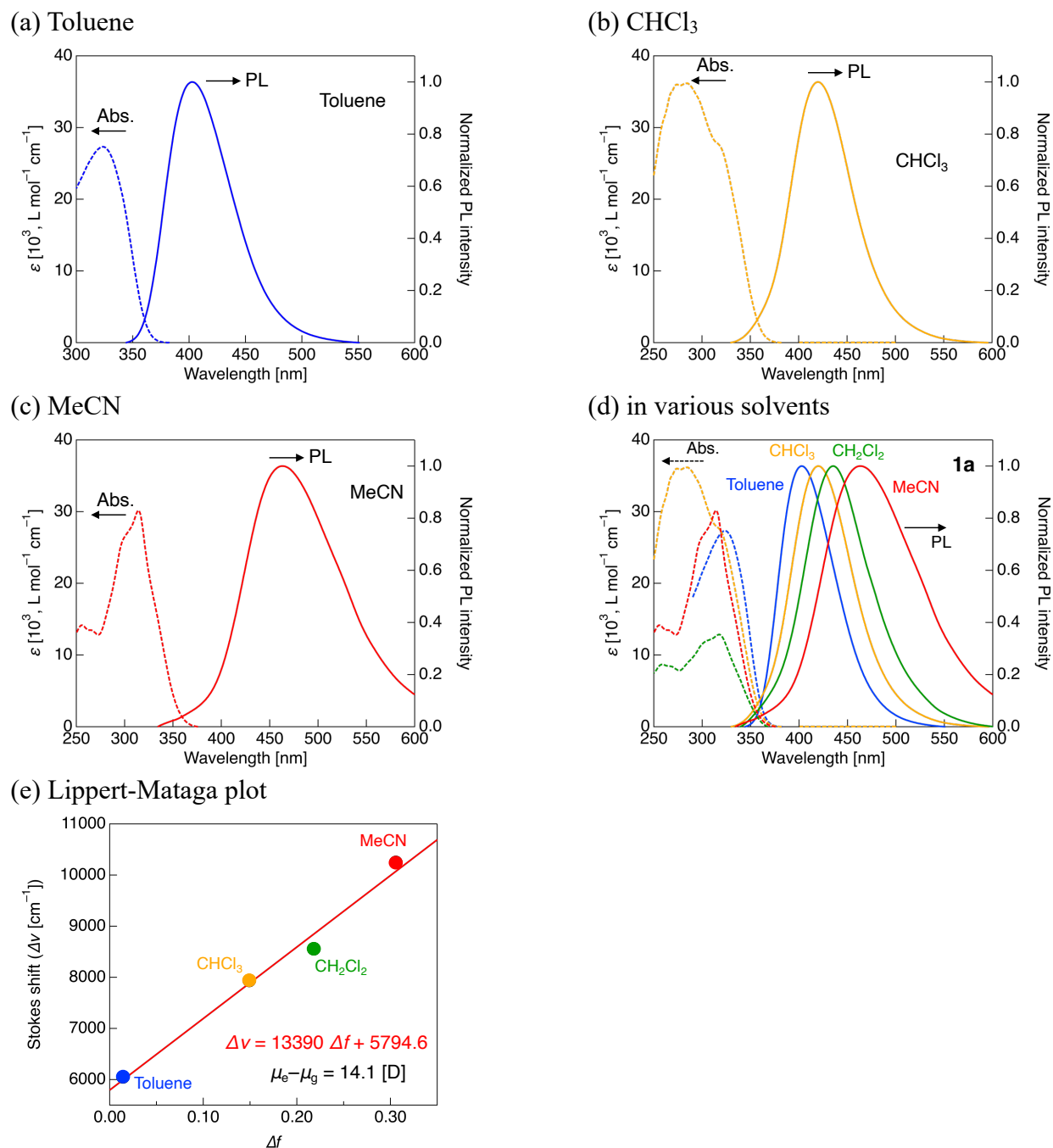


(h) **1h**



**Figure S28.** UV-vis absorption and PL spectrum of **1a–h** in CH<sub>2</sub>Cl<sub>2</sub> solution ( $1.0 \times 10^{-5}$  mol L<sup>-1</sup>).

## Solvatochromic PL behavior



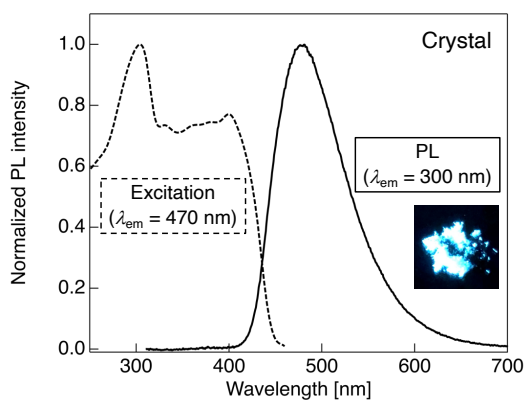
**Figure S29.** UV-vis absorption and PL spectra of **1a** in a different solvent: (a) toluene, (b) CHCl<sub>3</sub>, (c) MeCN. (d) in various solvents, (e) Lippert-Mataga plot of **1a**.

**Table S3.** Solvent effect on the photophysical behavior.

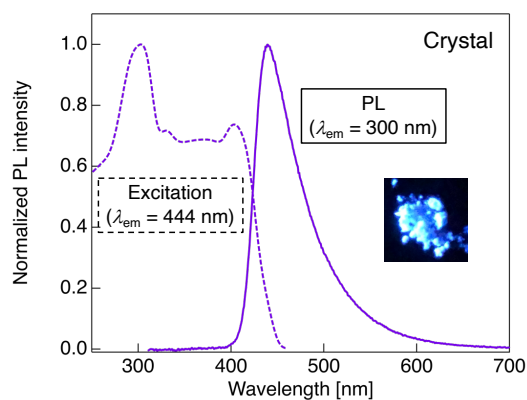
	$\epsilon$ (25 °C)	$n$	$\Delta f$	$\nu_{\text{Abs}}$	$\nu_{\text{F}}$	$\Delta\nu$
Toluene	2.3807	1.494	0.01416715	30864	24814	6050
CHCl <sub>3</sub>	4.806	1.443	0.14909508	31746	23810	7936
CH <sub>2</sub> Cl <sub>2</sub>	8.93	1.421	0.21817531	31546	22989	8557
MeCN	35.94	1.341	0.30574908	31847	21598	10249

## Cry phase

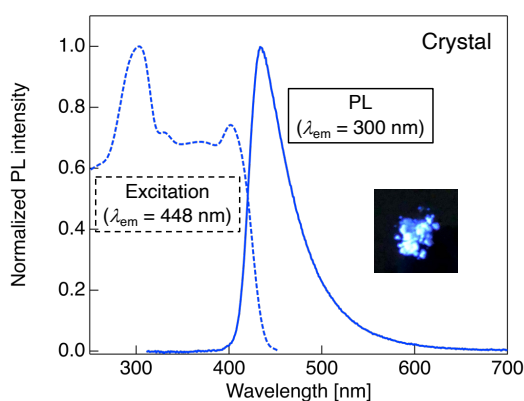
(a) **1a**



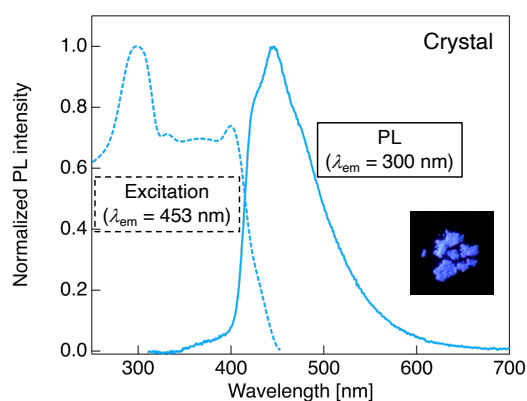
(b) **1b**



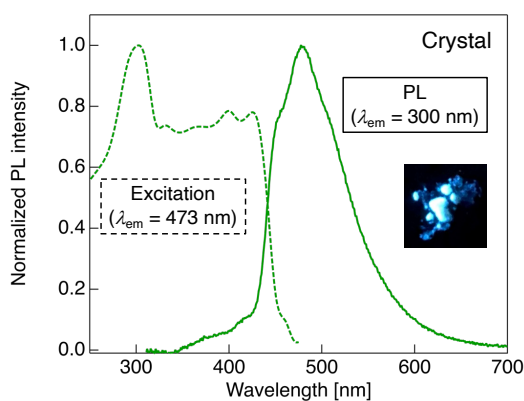
(c) **1c**



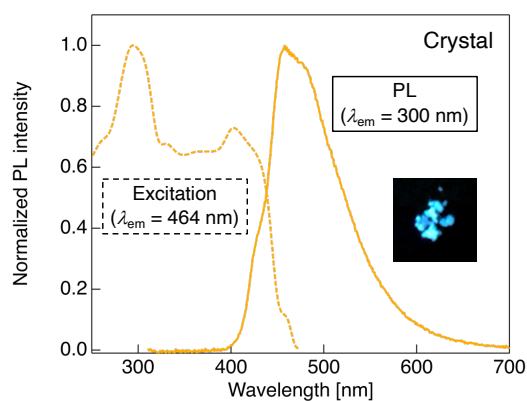
(d) **1d**



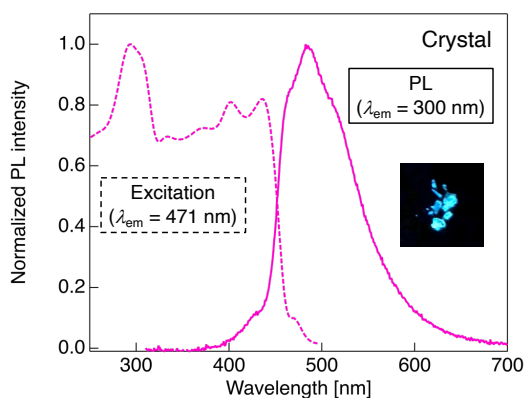
(e) **1e**



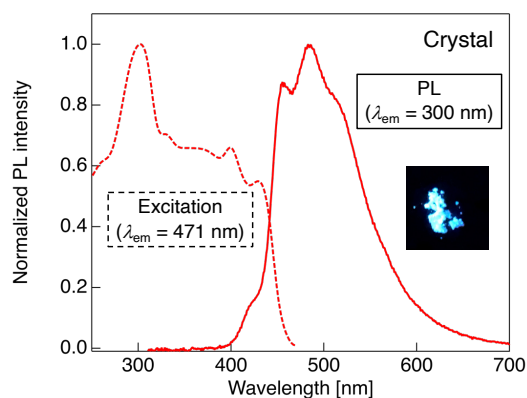
(f) **1f**



(g) **1g**



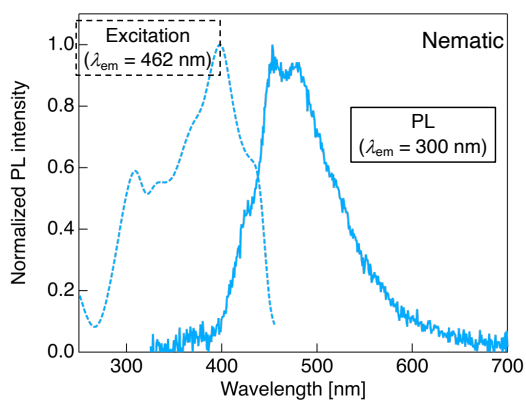
(h) **1h**



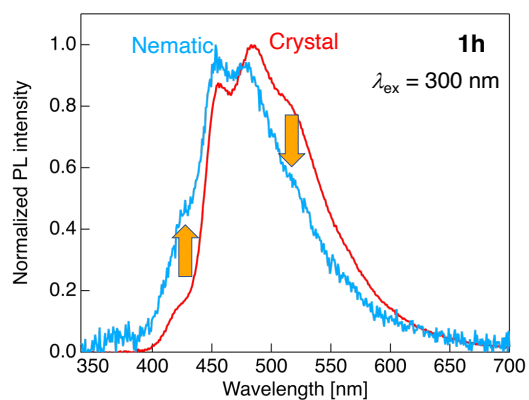
**Figure S30.** Excitation and PL spectra of **1a–h** in Cry phase.

## Nematic LC phase

(a)



(b)



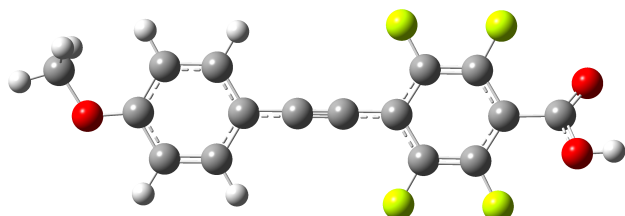
**Figure S31.** (a) Excitation and PL spectrum of **1h** in the aggregated structure of nematic phase. (b) Difference in PL spectrum between Cry and N phases.



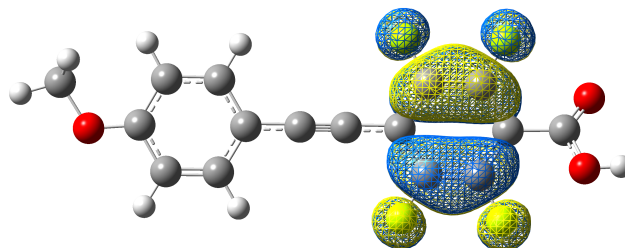
## Quantum chemical calculation

Quantum chemical calculation was performed using the Gaussian 16 program set. The initial structure for quantum chemical calculations was optimized by the structural optimization calculation with the M06-2X functional and the 6-31+G(d,p) basis set. Theoretical vertical electronic transitions were determined by time-dependent self-consistent field (TD-SCF) approximation.

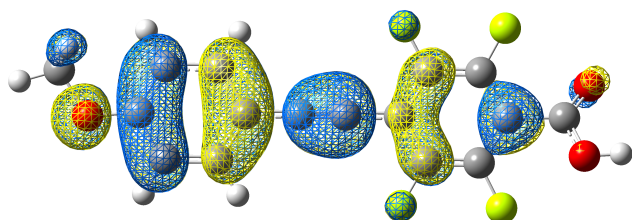
(a) Optimized structure



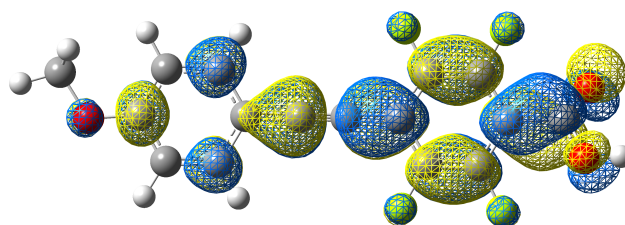
(b) HOMO-1 (-8.84 eV)



(c) HOMO (-7.45 eV)



(d) LUMO (-1.75 eV)



(e) Theoretical data

<Optimization>

SCF Done: E(RM062X) = -1239.07820687 Hartree A.U. after 10 cycles

Dipole moment (field-independent basis, Debye):

X= 4.6107 Y= 0.0184 Z= -1.0269 Tot= 4.7237

<TD-SCF>

Excited State	1:	Singlet-A	3.8827 eV	319.33 nm	f=1.3882	<S**2>=0.000
	79 → 83	0.11368				
	82(H) → 83(L)	0.67893				

This state for optimization and/or second-order correction.

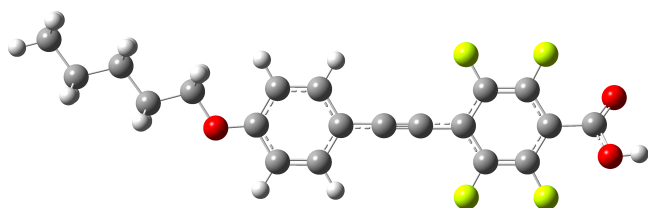
Total Energy, E(TD-HF/TD-DFT) = -1238.93552070

Copying the excited state density for this state as the 1-particle RhoCI density.

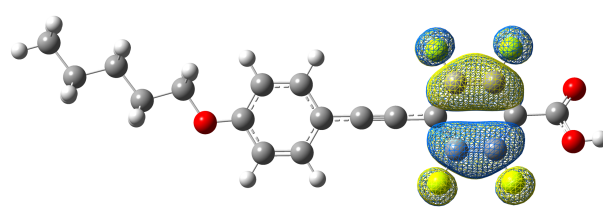
Excited State	2:	Singlet-A	4.6453 eV	266.90 nm	f=0.0644	<S**2>=0.000
	79 → 87	0.14571				
	81(H-1) → 83(L)	0.65560				
	82 → 87	0.16620				

**Figure S32.** (a) Optimized structure of **1a** and its orbital distributions for (b) HOMO-1, (c) HOMO, and (d) LUMO. (e) Theoretical data calculated by optimization and TD-SCF calculation.

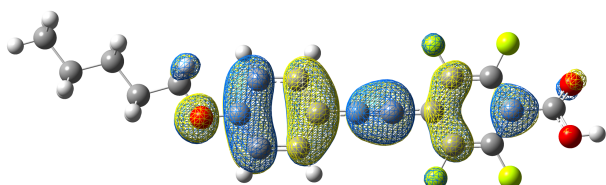
(a) Optimized structure



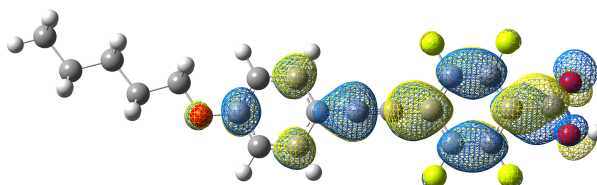
(b) HOMO-1 (-8.84 eV)



(c) HOMO (-7.44 eV)



(d) LUMO (-1.75 eV)



(e) Theoretical data

<Optimization>

SCF Done: E(RM062X) = -1396.26476274 Hartree A.U. after 11 cycles

Dipole moment (field-independent basis, Debye):

X= -5.0394 Y= -0.3780 Z= 0.9308 Tot= 5.1385

<TD-SCF>

Excited State	1:	Singlet-A	3.7717 eV	328.73 nm	f=1.5811	<S**2>=0.000
	95 → 99	-0.12014				
	98(H) → 99(L)	0.67727				

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-DFT) = -1396.12615684

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State	2:	Singlet-A	4.6185 eV	268.45 nm	f=0.0897	<S**2>=0.000
	95 → 103	0.13470				
	97(H-1) → 99(L)	0.66107				
	98 → 103	-0.15373				

**Figure S33.** (a) Optimized structure of **1e** and its orbital distributions for (b) HOMO-1, (c) HOMO, and (d) LUMO. (e) Theoretical data calculated by optimization and TD-SCF calculation.

**Table S4.** Cartesian coordinate for **1a**.

No.	Atom	Type	Coordinates (Angstroms)								
	No.		x	y	z						
1	8	0	7.119914	-0.378629	0.015764	16	6	0	-3.164992	1.235515	-0.005529
2	6	0	5.775929	-0.230893	0.008445	17	6	0	-3.925545	0.063407	0.008872
3	6	0	5.121218	1.003920	-0.068737	18	6	0	-3.228813	-1.147279	0.024796
4	1	0	5.676208	1.931716	-0.129124	19	6	0	-1.844563	-1.181829	0.031048
5	6	0	3.730007	1.041741	-0.068187	20	9	0	-1.108028	2.349768	-0.046825
6	1	0	3.220313	1.997825	-0.127723	21	9	0	-3.753003	2.430068	-0.046816
7	6	0	2.978192	-0.137572	0.007627	22	9	0	-3.873721	-2.313079	0.069227
8	6	0	3.649655	-1.373275	0.084078	23	9	0	-1.232809	-2.366883	0.064482
9	1	0	3.075449	-2.291986	0.143189	24	6	0	-5.417349	0.146047	0.034096
10	6	0	5.031173	-1.418495	0.084775	25	8	0	-5.994684	-0.876323	-0.598847
11	1	0	5.561379	-2.362943	0.143874	26	8	0	-6.028067	1.044897	0.560405
12	6	0	1.551074	-0.089010	0.007713	27	1	0	-6.958948	-0.772991	-0.532548
13	6	0	0.338712	-0.050904	0.008275	28	6	0	7.922526	0.792457	-0.059149
14	6	0	-1.080599	-0.011156	0.008293	29	1	0	8.954696	0.448067	-0.037302
15	6	0	-1.781305	1.198748	-0.013885	30	1	0	7.734698	1.334591	-0.991283
						31	1	0	7.735042	1.448763	0.796563

**Table S5.** Cartesian coordinate for **1e**.

No.	Atom		Coordinates (Angstroms)			No.	Atom		Coordinates (Angstroms)		
	No.	Type	x	y	z		No.	Type	x	y	z
1	8	0	-5.233268	-0.983555	-0.008581	22	9	0	5.890712	-2.116862	-0.092038
2	6	0	-3.902604	-0.745975	-0.008320	23	9	0	3.261753	-2.381238	-0.089416
3	6	0	-3.331962	0.532708	0.004708	24	6	0	7.233299	0.455578	-0.029042
4	1	0	-3.948396	1.422687	0.015511	25	8	0	7.888700	-0.520543	0.600679
5	6	0	-1.946680	0.664341	0.003831	26	8	0	7.772067	1.402214	-0.550215
6	1	0	-1.502746	1.654443	0.013921	27	1	0	8.841756	-0.340782	0.536138
7	6	0	-1.116417	-0.463791	-0.009976	28	6	0	-6.124062	0.132049	0.005280
8	6	0	-1.702733	-1.744165	-0.023071	29	1	0	-5.938293	0.736128	0.903137
9	1	0	-1.067721	-2.623880	-0.033705	30	1	0	-5.941128	0.755990	-0.879490
10	6	0	-3.078150	-1.882536	-0.022219	31	6	0	-7.541874	-0.403214	0.001478
11	1	0	-3.542977	-2.862578	-0.032097	32	1	0	-7.683306	-1.028491	-0.888146
12	6	0	0.303959	-0.317835	-0.010398	33	1	0	-7.681200	-1.047095	0.878068
13	6	0	1.510726	-0.194876	-0.010593	34	6	0	-8.571119	0.725976	0.014566
14	6	0	2.922726	-0.045697	-0.009669	35	1	0	-8.417268	1.373825	-0.859242
15	6	0	3.524969	1.215884	0.024526	36	1	0	-8.415469	1.355217	0.901547
16	6	0	4.901259	1.362937	0.018146	37	6	0	-10.009892	0.211462	0.010578
17	6	0	5.752622	0.255129	-0.006810	38	1	0	-10.163943	-0.417849	-0.874986
18	6	0	5.154552	-1.006935	-0.035423	39	1	0	-10.162434	-0.435915	0.883293
19	6	0	3.777374	-1.151703	-0.043126	40	6	0	-11.034755	1.343638	0.023096
20	9	0	2.762230	2.309320	0.068483	41	1	0	-10.914713	1.986918	-0.854947
21	9	0	5.392150	2.600109	0.072594	42	1	0	-12.057953	0.957893	0.020008
						43	1	0	-10.913225	1.968767	0.913952