
Supplementary Materials

Red and Blue-Black Tin Monoxide, SnO: Pitfalls, Challenges, and Helpful Tools in Crystal Structure Determination of Low-Intensity Datasets from Microcrystals

Hans Reuter

Chemistry, Department of Biology/Chemistry, Osnabrück University, Barbarastr. 7,
49069 Osnabrück, Germany; hreuter@uos.de

Table of contents

Table S1. Atomic coordinates of black-blue SnO at ambient temperature.....	3
Table S2. Bond lengths and angles of black-blue SnO at ambient temperature.....	4
Table S3. Anisotropic thermal displacement parameters of black-blue SnO at ambient temperature.	5
Table S4. Atomic coordinates of black-blue SnO at 100 K.....	6
Table S5. Bond lengths and angles of black-blue SnO at 100 K.....	7
Table S6. Anisotropic thermal displacement parameters of black-blue SnO at 100 K.....	8
Figure S1. Unit cell of red SnO in <i>Cmc2₁</i>	9
Table S7. Crystallographic data or red SnO in <i>Cmc2₁</i>	10
Table S8. Atomic coordinates of red SnO in <i>Cmc2₁</i>	11
Table S9. Bond lengths and angles of red SnO in <i>Cmc2₁</i>	12
Table S10. Anisotropic thermal displacement parameters of red SnO in <i>Cmc2₁</i>	13
Table S11. Atomic coordinates of red SnO in <i>Pbca</i>	14
Table S12. Bond lengths and angles of red SnO in <i>Pbca</i>	15
Table S13. Anisotropic thermal displacement parameters of red SnO in <i>Pbca</i>	16

Table S1. Atomic coordinates of black-blue SnO at ambient temperature.

Atomic coordinates ($x \times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for black-blue SnO at ambient temperature. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	$U(\text{eq})$
Sn(1)	2500	2500	2375(1)	11(1)
O(1)	7500	2500	0	13(1)

Table S2. Bond lengths and angles of black-blue SnO at ambient temperature.

Bond lengths [Å] and angles [°] for black-blue SnO at ambient temperature.

Sn(1)-O(1)#1	2.2213(1)
Sn(1)-O(1)#2	2.2213(1)
Sn(1)-O(1)#3	2.2213(1)
Sn(1)-O(1)	2.2213(1)
O(1)#1-Sn(1)-O(1)#2	74.48(1)
O(1)#1-Sn(1)-O(1)#3	117.71(1)
O(1)#2-Sn(1)-O(1)#3	74.48(1)
O(1)#1-Sn(1)-O(1)	74.48(1)
O(1)#2-Sn(1)-O(1)	117.71(1)
O(1)#3-Sn(1)-O(1)	74.48(1)
Sn(1)#1-O(1)-Sn(1)#4	105.52(1)
Sn(1)#1-O(1)-Sn(1)#3	117.71(1)
Sn(1)#4-O(1)-Sn(1)#3	105.52(1)
Sn(1)#1-O(1)-Sn(1)	105.52(1)
Sn(1)#4-O(1)-Sn(1)	117.71(1)
Sn(1)#3-O(1)-Sn(1)	105.52(1)

Symmetry transformations used to generate equivalent atoms:

#1 -x+1, -y, -z #2 x-1, y, z #3 -x+1, -y+1, -z
#4 x+1, y, z

Table S3. Anisotropic thermal displacement parameters of black-blue SnO at ambient temperature.

Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for black-blue SnO at ambient temperature. The anisotropic displacement factor exponent takes the form:
 $-2\pi^2 [h^2 a^*{}^2 U_{11} + \dots + 2hka^* b^* U_{12}]$

	U11	U22	U33	U23	U13	U12
Sn(1)	11(1)	11(1)	12(1)	0	0	0
O(1)	11(1)	11(1)	18(1)	0	0	0

Table S4. Atomic coordinates of black-blue SnO at 100 K.

Atomic coordinates ($x \times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for black-blue SnO at 100 K. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	$U(\text{eq})$
Sn(1)	2500	2500	2380(1)	4 (1)
O(1)	7500	2500	0	5 (1)

Table S5. Bond lengths and angles of black-blue SnO at 100 K.

Bond lengths [Å] and angles [°] for black-blue SnO at 100 K.

Sn(1)-O(1)#1	2.2213(1)
Sn(1)-O(1)#2	2.2213(1)
Sn(1)-O(1)#3	2.2213(1)
Sn(1)-O(1)	2.2213(1)
O(1)#1-Sn(1)-O(1)#2	74.48(1)
O(1)#1-Sn(1)-O(1)#3	117.71(1)
O(1)#2-Sn(1)-O(1)#3	74.48(1)
O(1)#1-Sn(1)-O(1)	74.48(1)
O(1)#2-Sn(1)-O(1)	117.71(1)
O(1)#3-Sn(1)-O(1)	74.48(1)
Sn(1)#1-O(1)-Sn(1)#4	105.52(1)
Sn(1)#1-O(1)-Sn(1)#3	117.71(1)
Sn(1)#4-O(1)-Sn(1)#3	105.52(1)
Sn(1)#1-O(1)-Sn(1)	105.52(1)
Sn(1)#4-O(1)-Sn(1)	117.71(1)
Sn(1)#3-O(1)-Sn(1)	105.52(1)

Symmetry transformations used to generate equivalent atoms:

#1 -x+1,-y,-z #2 x-1,y,z #3 -x+1,-y+1,-z
#4 x+1,y,z

Table S6. Anisotropic thermal displacement parameters of black-blue SnO at 100 K.

Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for black-blue SnO at 100 K.

The anisotropic displacement factor exponent takes the form:

$$-2\pi^2 [h^2 a^*{}^2 U_{11} + \dots + 2hka^* b^* U_{12}]$$

	U11	U22	U33	U23	U13	U12
Sn(1)	4 (1)	4 (1)	4 (1)	0	0	0
O(1)	4 (1)	4 (1)	8 (1)	0	0	0

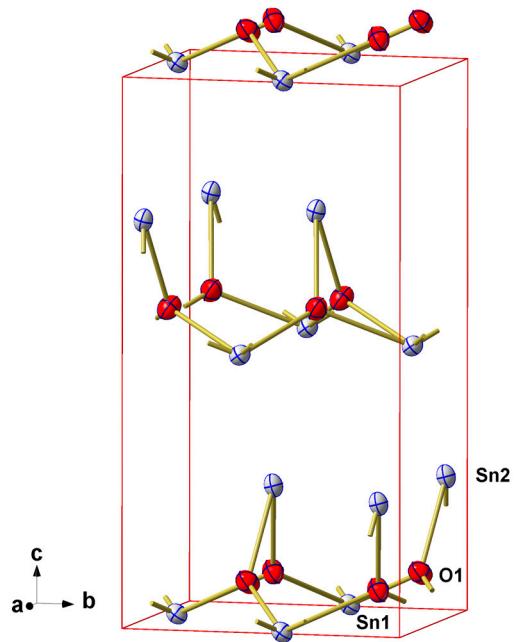


Figure S1. Unit cell of red SnO in $Cmc2_1$.

Table S7. Crystallographic data or red SnO in *Cmc2*₁

Empirical formula	O Sn
Formula weight	134.69
Temperature	296 (2) K
Wavelength	0.71073 Å
Crystal system, space group	Orthorhombic, <i>Cmc2</i> ₁
Unit cell dimensions	a = 5.0116(3) Å b = 5.7445(4) Å c = 11.0600(6) Å
Volume	318.41(3) Å ³
Z, Calculated density	8, 5.619 Mg/m ³
Absorption coefficient	15.433 mm ⁻¹
F(000)	464
Crystal size	0.015 x 0.015 x 0.015 mm
Theta range for data collection	3.684 to 28.979 deg.
Limiting indices	-6<=h<=6, -7<=k<=7, -15<=l<=15
Reflections collected / unique	14367 / 466 [R(int) = 0.0591]
Completeness to theta = 25.242	100.0 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.813 and 0.674
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	466 / 1 / 23
Goodness-of-fit on F ²	1.225
Final R indices [I>2sigma(I)]	R1 = 0.0296, wR2 = 0.0698
R indices (all data)	R1 = 0.0339, wR2 = 0.0758
Absolute structure parameter	0.5(4)
Extinction coefficient	n/a
Largest diff. peak and hole	2.622 and -1.264 e.Å ⁻³

Table S8. Atomic coordinates of red SnO in *Cmc2*₁.

Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for red SnO. U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	U(eq)
Sn(1)	0	5761(3)	27(1)	21(1)
Sn(2)	0	9238(3)	2414(1)	22(1)
O(1)	1870(20)	8770(20)	747(10)	30(2)

Table S9. Bond lengths and angles of red SnO in *Cmc2*₁.

Bond lengths [Å] and angles [°] for red SnO in *Cmc2*₁.

Sn(1)-O(1)#1	2.098(12)
Sn(1)-O(1)#2	2.098(12)
Sn(1)-O(1)#3	2.121(11)
Sn(1)-O(1)	2.121(11)
Sn(2)-O(1)#3	2.087(11)
Sn(2)-O(1)	2.087(11)
O(1)#1-Sn(1)-O(1)#2	96.6(7)
O(1)#1-Sn(1)-O(1)#3	88.4(3)
O(1)#2-Sn(1)-O(1)#3	129.3(6)
O(1)#1-Sn(1)-O(1)	129.3(6)
O(1)#2-Sn(1)-O(1)	88.4(3)
O(1)#3-Sn(1)-O(1)	52.6(7)
O(1)#3-Sn(2)-O(1)	53.5(7)
Sn(2)-O(1)-Sn(1)#4	126.9(6)
Sn(2)-O(1)-Sn(1)	103.8(5)
Sn(1)#4-O(1)-Sn(1)	129.3(6)

Symmetry transformations used to generate equivalent atoms:

#1 x-1/2,y-1/2,z #2 -x+1/2,y-1/2,z
#3 -x,y,z #4 x+1/2,y+1/2,z

Table S10. Anisotropic thermal displacement parameters of red SnO in *Cmc2*₁.

Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for red SnO in *Cmc2*₁.

The anisotropic displacement factor exponent takes the form:

$$-2\pi^2 [h^2 a^*{}^2 U_{11} + \dots + 2hka^*b^* U_{12}]$$

	U11	U22	U33	U23	U13	U12
Sn (1)	4 (1)	4 (1)	4 (1)	0	0	0
O (1)	4 (1)	4 (1)	8 (1)	0	0	0

Table S11. Atomic coordinates of red SnO in *Pbca*.

Atomic coordinates ($x \times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for red SnO in *Pbca*. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	$U(\text{eq})$
Sn(1)	5084 (1)	8262 (1)	3806 (1)	21 (1)
O(1)	8107 (9)	6264 (9)	4519 (5)	25 (1)

Table S12. Bond lengths and angles of red SnO in *Pbca*.

Bond lengths [Å] and angles [°] for red SnO in *Pbca*.

Sn(1)-O(1)	2.057 (5)
Sn(1)-O(1)#1	2.102 (5)
Sn(1)-O(1)#2	2.118 (5)
O(1)-Sn(1)-O(1)#1	89.8 (1)
O(1)-Sn(1)-O(1)#2	94.6 (1)
O(1)#1-Sn(1)-O(1)#2	76.6 (2)
Sn(1)-O(1)-Sn(1)#3	129.1 (2)
Sn(1)-O(1)-Sn(1)#4	127.4 (2)
Sn(1)#3-O(1)-Sn(1)#4	103.4 (2)

Symmetry transformations used to generate equivalent atoms:

#1 -x+3/2,y+1/2,z #2 x-1/2,-y+3/2,-z+1

#3 -x+3/2,y-1/2,z #4 x+1/2,-y+3/2,-z+1

Table S13. Anisotropic thermal displacement parameters of red SnO in *Pbca*.

Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for red SnO in Pbca.

The anisotropic displacement factor exponent takes the form:

$$-2\pi^2 [h^2 a^*{}^2 U_{11} + \dots + 2hka^* b^* U_{12}]$$

	U11	U22	U33	U23	U13	U12
Sn (1)	17 (1)	21 (1)	24 (1)	-2 (1)	-3 (1)	0 (1)
O (1)	19 (2)	23 (2)	33 (3)	-5 (2)	-5 (2)	5 (2)
