

## Supporting Information

### Solvent effects on the spin crossover properties of iron(II) imidazolyimine complexes

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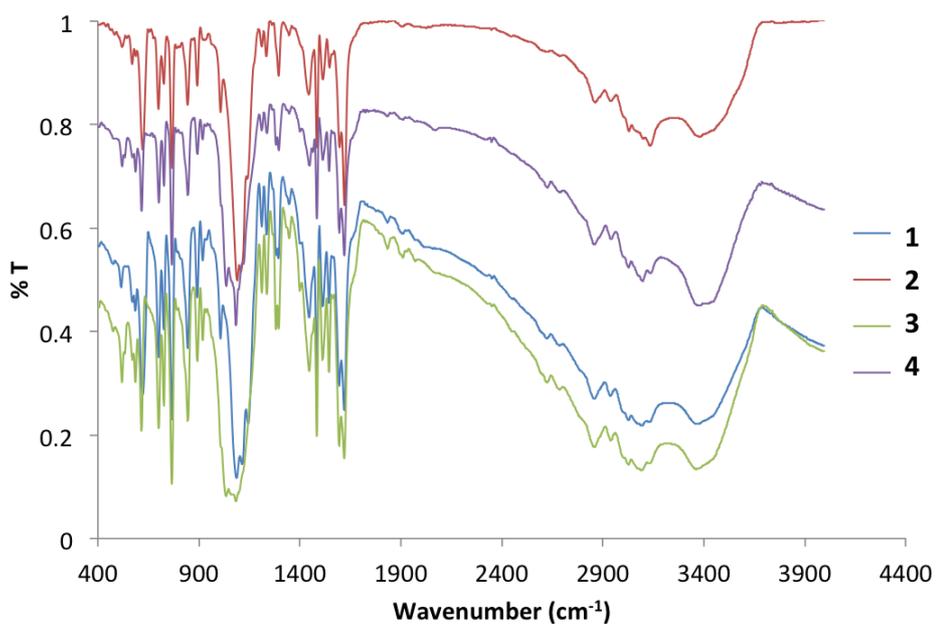
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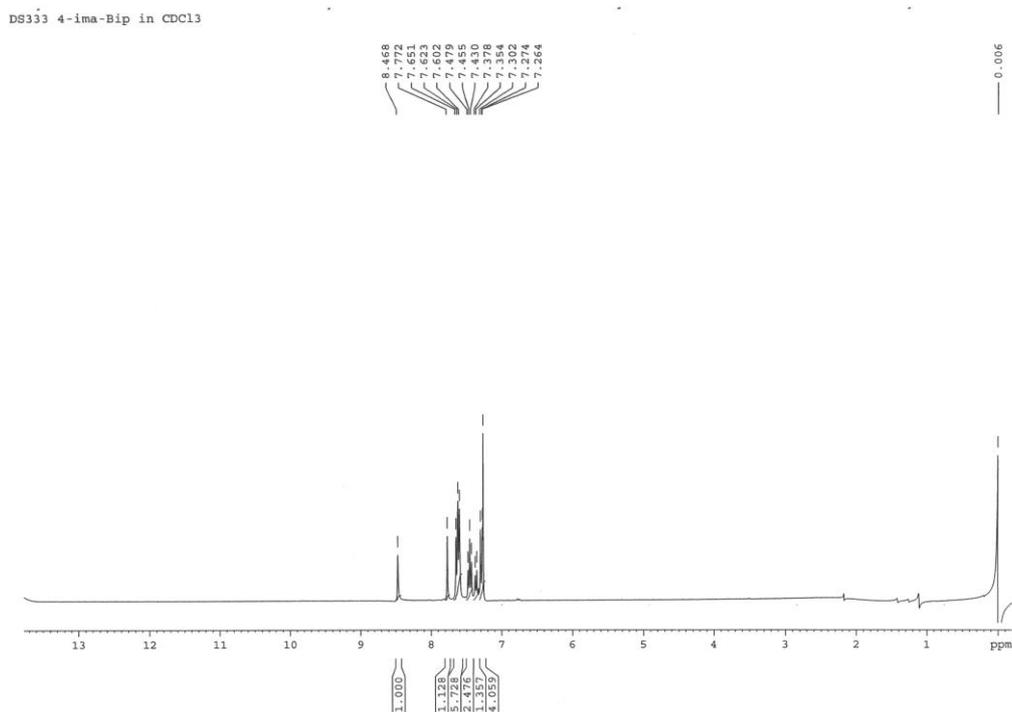
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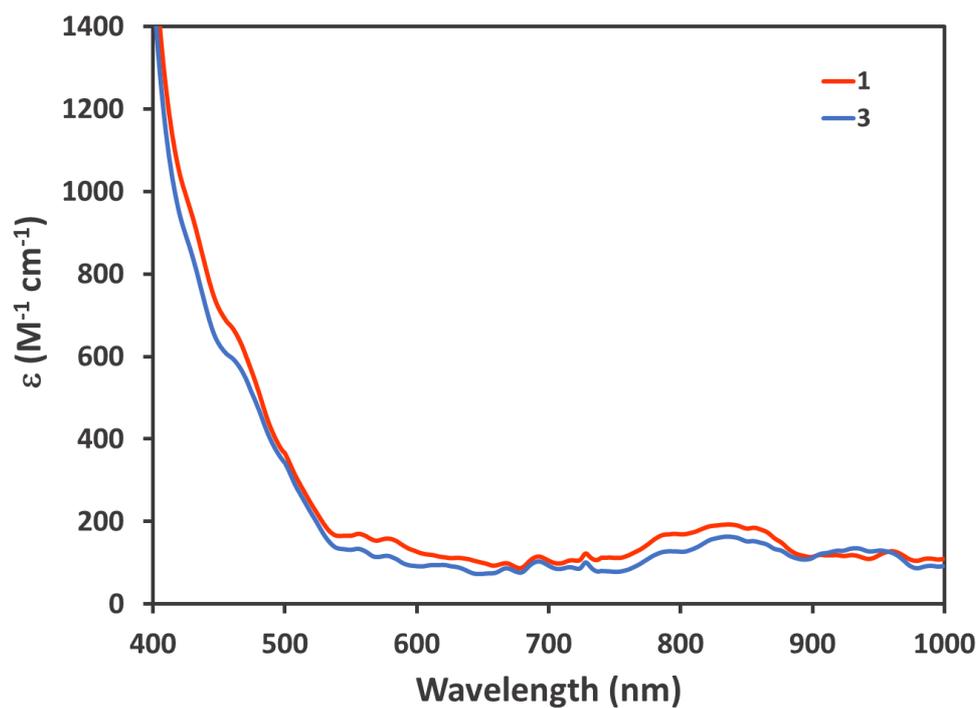
E-mail: [hdavid@mail.wu.ac.th](mailto:h david@mail.wu.ac.th)



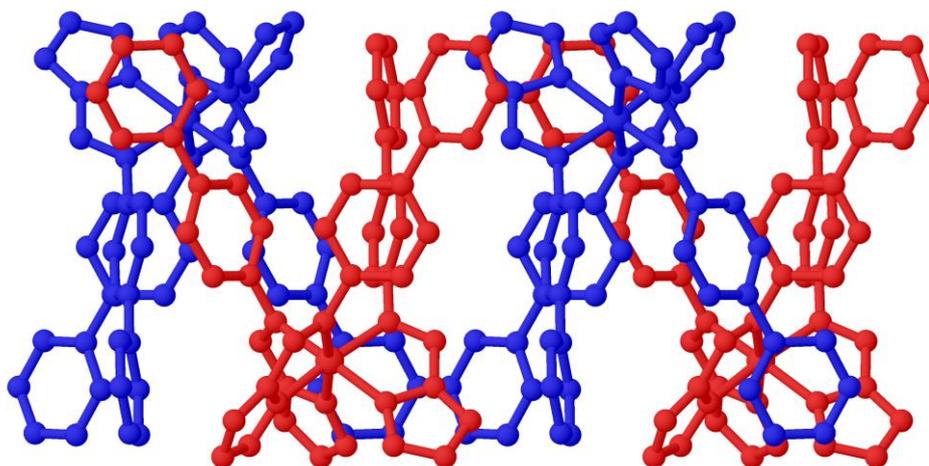
**Figure S1** IR spectra of **1- 4**.



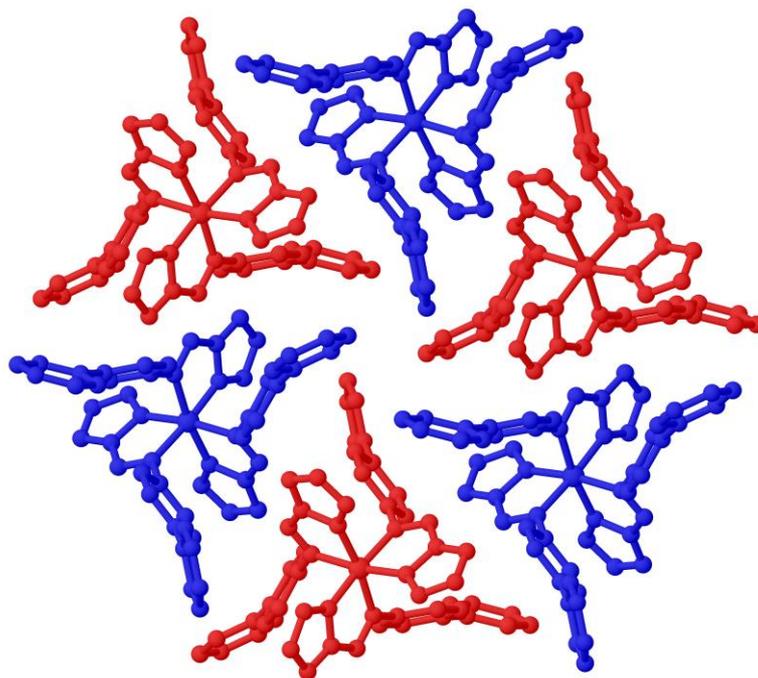
**Figure S2**  $^1\text{H-NMR}$  spectrum of **4-ima-Bp**.



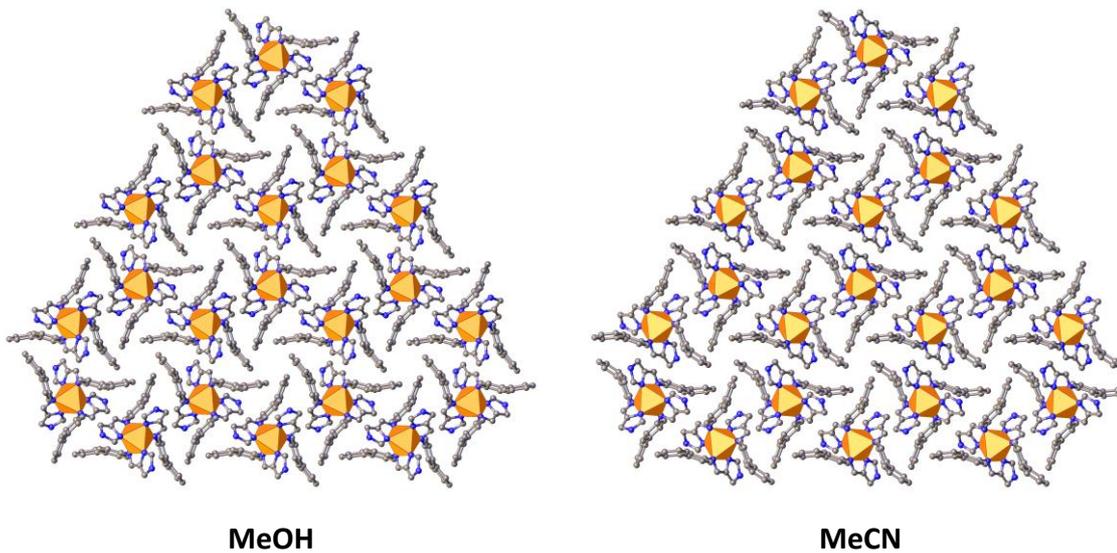
**Figure S3** UV-Vis of  $[\text{Fe}(4\text{-ima-Bp})_3](\text{ClO}_4)_2$  **1** and  $[\text{Fe}(4\text{-ima-Bp})_3](\text{BF}_4)_2$  **3** in MeCN in a 0.1 M solution.



**Figure S4** Side-on view of the packing in *fac*- $[\text{Fe}(4\text{-ima-Bp})_3](\text{ClO}_4)_2 \cdot 3\text{MeOH}$  **2**.



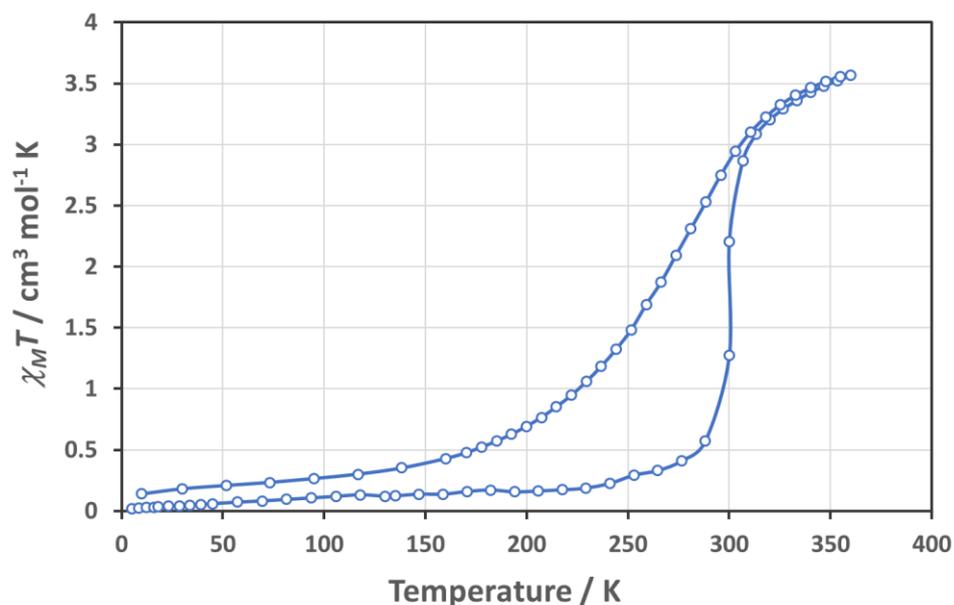
**Figure S5** View of the *pseudo*-hexagonal packing motif in *fac*-[Fe(4-ima-Bp)<sub>3</sub>](BF<sub>4</sub>)<sub>2</sub>·3.5MeCN **5**.



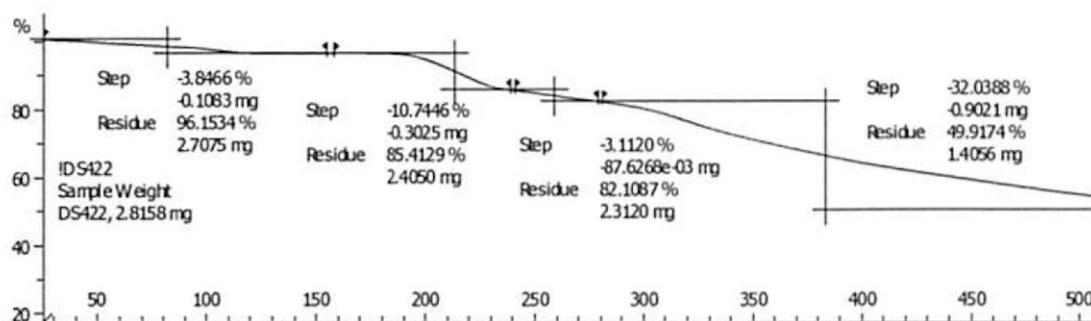
**Figure S6** Comparative view of the hexagonal and *pseudo*-hexagonal packing motifs found in **2** and **5**.

**Table S1** Geometric parameters of C-H $\cdots\pi$  and  $\pi$ - $\pi$  interactions in **1-2** and **5**.

<b>Complex</b>		<b>Type</b>	<b>Distance</b>
<b>1</b>	C10-H10 $\cdots\pi$ (C5-C10)	<i>Intramolecular</i>	2.681 Å
	C7-H7 $\cdots\pi$ (C5-C10)	<i>Intermolecular</i>	2.689 Å
	C13-H13 $\cdots\pi$ (C11-C16)	<i>Intermolecular</i>	2.905 Å
<b>2</b>	C10-H10 $\cdots\pi$ (C5-C10)	<i>Intramolecular</i>	2.635 Å
	C7-H7 $\cdots\pi$ (C5-C10)	<i>Intermolecular</i>	2.630 Å
	C13-H13 $\cdots\pi$ (C11-C16)	<i>Intermolecular</i>	2.808 Å
<b>5</b>	C42-H42 $\cdots\pi$ (C5-C10)	<i>Intramolecular</i>	2.654 Å
	C10-H10 $\cdots\pi$ (C21-C26)	<i>Intramolecular</i>	2.561 Å
	C22-H22 $\cdots\pi$ (C37-C42)	<i>Intramolecular</i>	2.657 Å
	C25-H25 $\cdots\pi$ (C5-C10)	<i>Intermolecular</i>	2.716 Å
	C39-H39 $\cdots\pi$ (C21-26)	<i>Intermolecular</i>	2.776 Å
	C45-H45 $\cdots\pi$ (C11-C16)	<i>Intermolecular</i>	3.055 Å
	$\pi$ - $\pi$	-	3.690 Å



**Figure S7** SQUID profile of *fac*-[Fe(4-ima-Bp)<sub>3</sub>](ClO<sub>4</sub>)<sub>2</sub>·3EtOH **1**.



**Figure S8** TGA of *fac*-[Fe(4-ima-Bp)<sub>3</sub>](ClO<sub>4</sub>)<sub>2</sub>·3EtOH **1**.

The first mass loss of 3.85% occurs at *ca.* 80 °C and is consistent with one equivalent of EtOH (calculated 4.06%). The second mass loss of 10.74% occurs at *ca.* 210 °C and is broadly suggestive of loss of the two remaining EtOH molecules (calculated 8.12%).