

Hierarchical Core/Shell Structured Ag@Ni(OH)₂ Nanospheres as Binder-Free Electrodes for High Performance Supercapacitors

Sa Lv *, Xuefeng Chu, Fan Yang, Huan Wang, Jia Yang, Yaodan Chi and Xiaotian Yang *

Jilin Provincial Key Laboratory of Architectural Electricity & Comprehensive Energy Saving, School of Electrical Engineering and Computer, Jilin Jianzhu University, Changchun 130118, China; stone2009@126.com; ctpnxxn@163.com; whuan@ciac.ac.cn; yangjia@jlju.edu.cn; chiyaodan@jlju.edu.cn

* Correspondence: lvsa82@163.com (S.L.); hanyxt@163.com (X.Y.); Tel.: +86-0431-8456-6181 (S.L.)

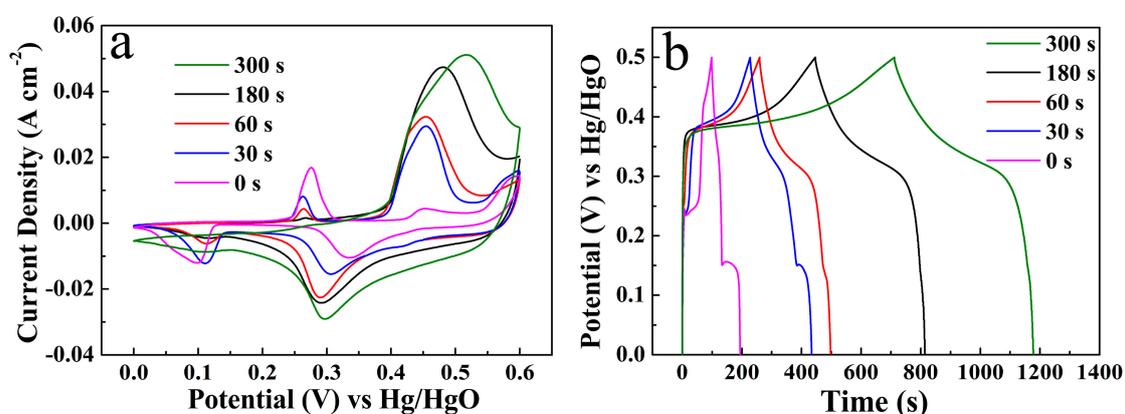


Figure S1. Ag@Ni(OH)₂ electrode corresponding to different Ni(OH)₂ deposition time: (a) CV curves; (b) GCD curves.

The specific capacitance of Ag@Ni(OH)₂ electrode can be obtained by the following equation:

$$C_s = \frac{\int I(V)dV}{2vS\Delta V}$$

where C_s (F cm⁻²) is the specific capacitance, $I(V)$ (A) is the response current, V (V) is the potential vs. Hg/HgO, v (V s⁻¹) is the scan rate, S (cm²) is the effective area of the electrode and ΔV (V) is the working potential. So the specific capacitance decreases with the increase of scan rate.

The specific capacitance of Ag@Ni(OH)₂ electrode can be obtained by the following equation:

$$C_s = \frac{I \times \Delta t}{S\Delta V}$$

where C_s (F cm⁻²) is the specific capacitance, I (A) is the charge–discharge current, Δt (s) is the discharging time, S (cm²) is the effective area of the electrode and ΔV (V) represents the potential drop during discharge.

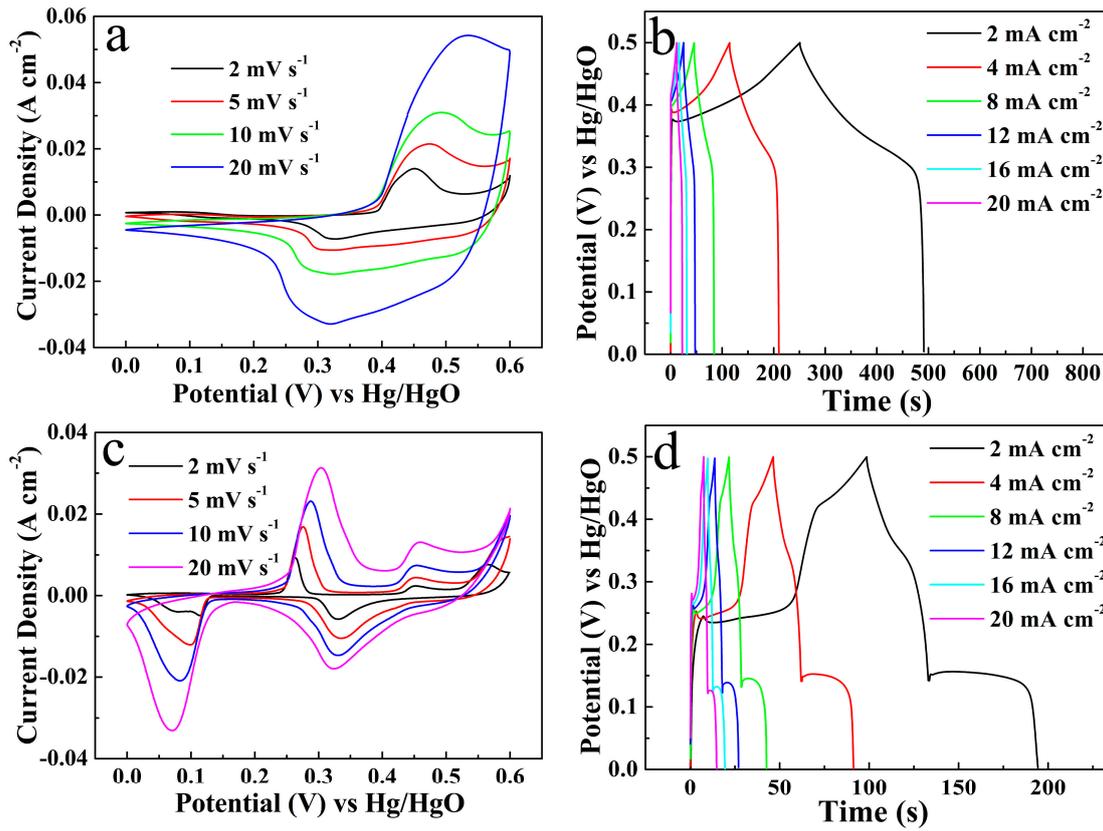


Figure S2. (a,c) The CV curves and (b,d) GCD curves of Ni(OH)₂ and the Ag electrode, respectively.

Table S1. The specific capacitance versus current density of the three electrodes.

Electrode	2 mA	4 mA	8 mA	12 mA	16 mA	20 mA
C_s (F cm ⁻²)						
Ag@Ni(OH) ₂	1.864	1.395	1.098	0.946	0.838	0.760
Ni(OH) ₂	0.962	0.766	0.619	0.540	0.480	0.436
Ag	0.388	0.359	0.336	0.319	0.310	0.296

The average equivalent series resistance (R_{ESR}) are derived from the equation:

$$R_{ESR} = \frac{V_{drop}}{2I}$$

where, V_{drop} (V) is the abrupt voltage drop at the beginning of the discharging curve and I (A) is the corresponding current.

Table S2. The R_{ESR} and voltage drop versus current density of the Ag@Ni(OH)₂ electrode.

I (mA)	2 mA	4 mA	8 mA	12 mA	16 mA	20 mA
V_{drop} (V)	0.0043	0.0081	0.0156	0.0230	0.0305	0.0380
R_{ESR} (Ω cm ⁻²)	1.075	1.015	0.975	0.958	0.953	0.950
Average R_{ESR} (Ω cm⁻²)						0.988

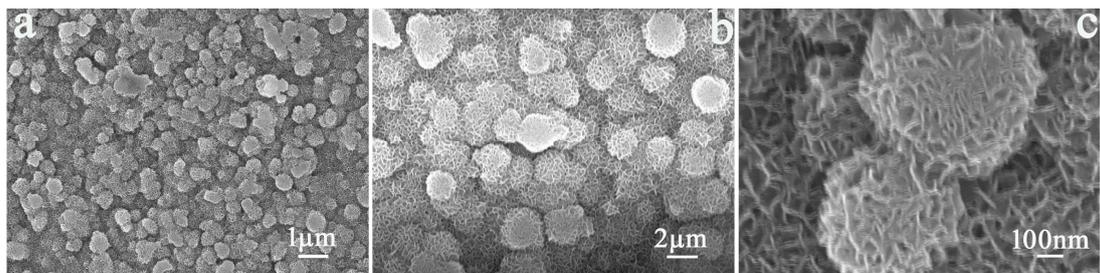


Figure S3. FE-SEM images of the Ag@Ni(OH)₂ composite at different magnifications after 3000 cycles.