

Supporting Information:

Evaluating the effect of iron(III) in the preparation of a conductive porous composite using a biomass waste-based starch template

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Table S1. Physical properties of the porous obtained materials at each stage of synthesis.

Sample	ρ_{skel} $\text{g}\cdot\text{cm}^{-3}$	ρ_{bulk} $\text{g}\cdot\text{cm}^{-3}$	ϵ %	V_p
St	1.470 ± 0.010	0.140 ± 0.010	90.476 ± 5.849	6.460 ± 0.417
St/iron(III)	1.500 ± 0.006	0.195 ± 0.003	87.000 ± 1.382	4.460 ± 0.070
St/iron(III)/PEDOT	1.337 ± 0.008	0.201 ± 0.010	84.966 ± 4.257	4.227 ± 0.211

Table S2. Physisorption Nitrogen physisorption data of cryogel template and conductive porous composites at different soaking times.

Sample	V_p $\text{Cc}\cdot\text{g}^{-1}$	Pore Radius	Surface area
		nm	$\text{m}^2\cdot\text{g}^{-1}$
St- 0 h	0.003	2.01	6.899
PEDOT-St 0.5 h	0.004	2.01	5.096
PEDOT-St 8 h	0.002	2.02	0.357
PEDOT-St 24 h	0.002	1.63	0.225
PEDOT-St 48 h	0.000	1.17	0.316

Table S3. Elemental composition of porous structures synthesized with PEDOT at different immersion times in iron(III) solution.

Soaking time in iron(III) solution (hours)	Element (%)			
	N	C	H	S
0.5	0.06	43.09	5.61	10.28
8	0.06	47.20	4.45	18.07
24	0.04	44.81	4.76	14.86
48	0.06	41.65	5.49	9.70

Table S4. Fitting of electrochemical impedance spectroscopy data using the R(RQ)(RQ) circuit model.

Circuit element	Immersion time in iron(III) (h)			
	0.5	8	24	48
Rs1 (Ω)	2502.2	31137	9.4186E^5	1.5582E^5
Rp1 (Ω)	3949.7	1.482E^7	3.5438E^7	1.993E^5
CPE1 (F)	5.636E^{-12}	1.125E^{-10}	1.277E^{-10}	6.516E^{-10}
CPE1.N	0.84	0.83731	0.52077	0.82964
Rs2 (Ω)	-	-	1355.9	2627.4
Rp2 (Ω)	-	-	5.2739E^6	1.9695E^5
CPE2 (F)	-	-	5.8335E^{-12}	1.0885E^{-11}
CPE2.N	-	-	0.92928	0.81716

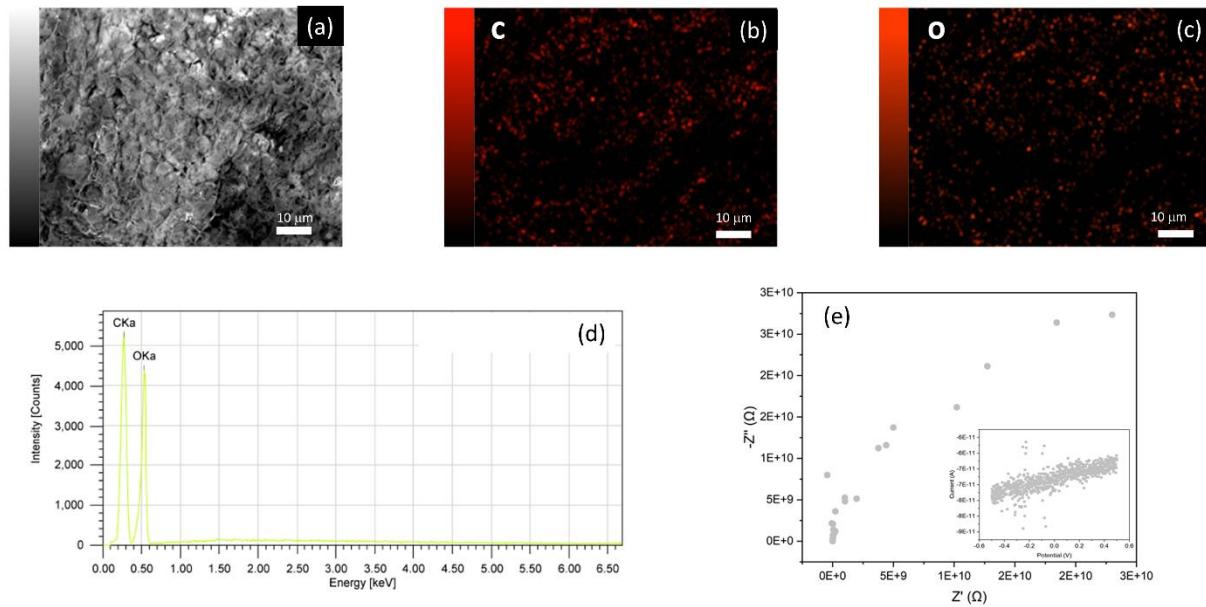


Figure S1. Starch template chemical and electrical measurements: elemental mapping (a) SEM image, (b) carbon, (c) oxygen (d) energy-dispersive spectroscopy spectra and (e) Nyquist plot (inset shows cyclic voltammetry curve).

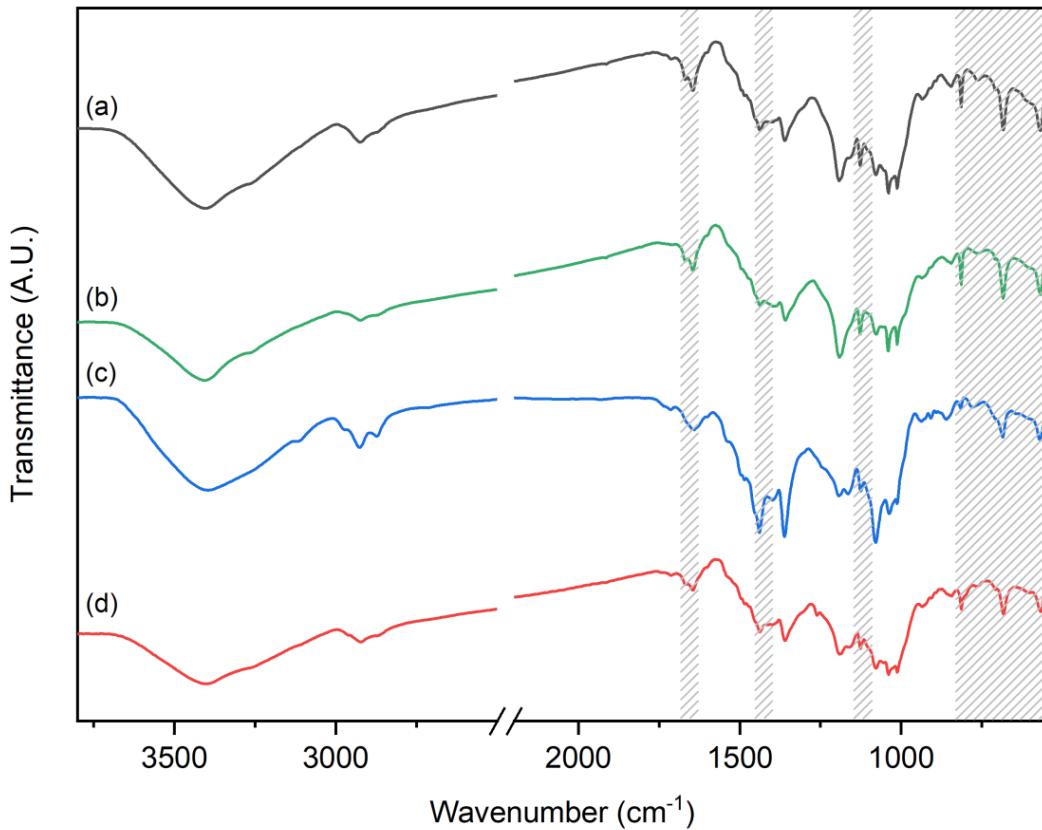


Figure S2. Infrared spectroscopy of the different samples polymerized after being soaked for (a) 0.5, (b) 8, (c) 24, and (d) 48 hours in an iron(III) solution.